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Gold(I)-Catalyzed Highly Diastereo- and Enantioselective Cyclization/[4+3] Annulation Cascades between 2-(1-Alkynyl)-2alken-1-ones and Anthranils

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Abstract: This work reports gold-catalyzed [4+3]-annulations of 2-(1-alkynyl)-2-alken-1-ones with anthranils to yield epoxybenzoazepine products with excellent *exo*-diastereoselectivity (dr > 25:1). The utility of this new gold catalysis is manifested by applicable substrates over a broad scope. More importantly, the enantioselective versions of these [4+3]-cycloadditions have been developed satisfactorily with chiral gold catalysts under ambient conditions (DCM, 0 °C); the ee levels range from 88.0-99.9%. With DFT calculations, we postulate a stepwise pathway to rationalize the preferable *exo*-stereoselection.

Nitroxy (N-O) functionalities commonly exist as the structural cores in numerous synthetic drugs or naturally occurring compounds.^[1,2] Among them, tetrahydro-1*H*-benzo[*b*]azepines bearing a hydroxyl group, such as species I-IV,^[3] are representatives of seven-membered alkaloids, which show strong biological activity in antiparastic disease,^[3a] antidiuretic hormone receptors^[3b] and β_2 -adrenergic agonists^[3e-f]. One convenient and easy construction of this seven-membered azacyclic core is to realize catalytic [4+3]-annulations between



Figure 1: Selected bioactive molecules and a short synthetic route

anthranils with suitable all carbon 1,3-dipoles, according to the protocol in eq 1.^[4,5] Despite this practicability, there are only very few all carbon 1,3-dipoles to react with anthranils in catalytic [4+3]-annulations. Luo et. al.^[4a] reported the first examples to employ donor-acceptor cyclopropanes to generate Sc(III)-enolate-containing carbocations, further furnishing the [4+3]-annulations with anthranils (eq 2). The enantioselective versions of such [4+3]-cyclopropane/anthranil annulations are recently

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achieved by You^[4b] and coworkers, using chiral palladium catalysts. We recently achieved gold-catalyzed [4+3]annulations between 2-alkenyl-1-ethynylbenzenes and anthranils in racemic versions;^[5] this process involves a trap of gold-containing isobenzofulvene cation (I) with anthranils (eq 3). To highlight this 1,3-dipole approach (eq 1), we report new



gold-catalyzed [4+3]-annulations^[6] using 2-(1-alkynyl)-2-alken-1ones to generate gold-containing 3-furyl methyl cations (II),^[7] further leading to furan-fused epoxybenzoazepines with excellent diastereoselectivity (dr > 25:1, eq 4). Notably, this exclusive *exo*-stereoselection is distinct from the *endo*-selectivity in the cyclopropane system.^[4] Apart from diastereoselectivity, we also developed highly enantioselective [4+3]-cycloadditions using chiral gold catalysts,^[8] further manifesting the synthetic significance. The work reports the first success to use anthranils in gold-catalyzed asymmetric reactions.^[9]

up to 99.9% er

We optimized the reactions of 2-(1-alkynyl)-2-alken-1-one 1a with anthranil 2a (1.5 equiv.) using various gold catalysts; the results are shown in Table 1. Initially, the reaction of species 1a and anthranil 2a was investigated with Ph₃PAuCl/AgNTf₂ (5 mol %) in dry dichloromethane (4Å MS, DCM, 0.5 h) at room temperature; this condition afforded a [4+3]-annulation product epoxybenzo[b]furo[3,4-e]azepine 3a as one diastereomer in 80% yield (dr > 25:1, entry 1). The use of (PhO)₃PAuCl/AgNTf₂ (5 mol%) provided 3a in 68% yield (entry 2). Other gold catalysts L'AuCl/AgNTf₂ (L'= IPr and P(t-Bu)₂(o-biphenyl)) gave the desired product 3a in 40-41% yields whereas unreacted 1a was recovered in 30-35% (entries 3-4). The variation of silver salts as in Ph₃PAuCl/AgX, (X = SbF₆ and OTf) maintained high efficiency to yield compound 3a in 73-75% yields (entries 5-6). For Ph₃PAuCl/AgNTf₂ in different solvents (entries 7-9), the yields of compound 3a were as follows: dichloroethane (77%), toluene (30%) and nitromethane (0%). AgNTf2 alone in DCM was completely inactive (entry 10). The molecular structure of compound **3a** was confirmed by its X-ray diffraction^[10] to reveal

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Table 1. Optimization of [4+3]-Annulation

Me Ph O Ph 1a ^[a] + (Au] 5 mol% solvent, time rt, 4Å MS (1.5 equiv.)				Ph- Me O Ph 3a ^[b]	
Entry	Catalyst (5 mol%)	Time (h)	Solvent	Yields(%) ^[b]	
				3a	1a
1	Ph ₃ PAuCl/AgNTf ₂	0.5	DCM	80	
2	(PhO) ₃ PAuCl/AgNTf ₂	1	DCM	68	
3	IPrAuCl/AgNTf ₂	9	DCM	40	35
4	LAuCI/AgNTf ₂	15	DCM	41	30
5	Ph ₃ PAuCl/AgSbF ₆	1	DCM	73	
6	Ph₃PAuCl/AgOTf	2	DCM	75	
7	Ph ₃ PAuCl/AgNTf ₂	1	DCE	77	
8	Ph ₃ PAuCl/AgNTf ₂	10	toluene	30	45
9	Ph ₃ PAuCl/AgNTf ₂	24	MeNO ₂		77
10	AgNTf ₂	24	DCM		90

[a] [1a] = 0.1 M. [b] Product yields are obtained after purification from a silica column. IPr = 1,3-bis(diisopropylphenyl) imidazol-2-ylidene. L = $P(t-Bu)_2(o-biphenyl)$.

an exo-configuration.

We assessed the substrate scope of these new annulations using various envnones 1 and Ph₃PAuCl/AgNTf₂ under the standard condition; the resulting products 3 were present exclusively as one diastereomer (dr > 25:1). We examined the reactions of enynones 1b-1d bearing various alkynyl substituents $4-XC_6H_4$ (X = Me, OMe and Br), further affording desired products 3b-3d in satisfactory yields (74-80%, entries 1-3). For these alkynyl substituents, we also tested an enynone 1e bearing a 3-chlorophenyl derivative, furnishing compound 3e in 78% yield (entry 4). We prepared also envnone 1f possessing a 3-thienylalkyne moiety; the reaction proceeded smoothly to deliver compound 3f in 76% yield (entry 5). We prepared also alkyl-substituted alkynes 1g-1h (R² = cyclopropyl and *n*-butyl) that provided the desired products 3g-3h in 77-81 % yields (entries 6-7). We next varied the alkenyl substituents (R1) of enynones 1i-1j with $R^2 = 4-XC_6H_4$ (X = Me and Cl); their resulting products 3i and 3j were obtained in 75-79% yields (entries 8-9). For methyl- and 3-thienyl-substituted alkene deivatives 1k and 1l, these gold catalyzed reactions furnished compounds 3k and 3l in 71% and 77% yields (entries 10-11). Our variations of the ketone substituents were also suitable to this gold catalysis. As shown in entries 12-13, alkylketone derivatives 1m and 1n (R^3 = ethyl and *n*-butyl) were applicable substrates delivering their products 3m-3n in 76 and 78% yields respectively. Envnones 10-1p bearing phenyl and 2-thienyl ketones were also applicable substrates to produce compounds 30-3p in 83% and 69% yields (entry 14-15). Aldehyde substrate 1q (R₃ = H) was less efficient, affording compound 3q in 55% yield (entry 16). We also prepared one enynone 1r with





[a] [1] = 0.1 M. [b] Product yields are obtained after separation from silica column

 R^1 = n-butyl and R^2 , R^3 = Ph that gave a complicated mixture of products. For the key 1,3-dipole intermediate (eq 4), the presence of R^1 = aryl in crucial to stabilize its cation stability.

Table 3 summarizes the catalytic [4+3]-annulations of various substituted anthranils (**2b-2i**) with model enynone **1a**; only one diastereomeric product was obtained exclusively. For anthranils **2b-2c** bearing X = Me and Br at the C(6) carbon, their corresponding epoxybenzoazepines **4b** and **4c** were obtained in82% and 72% yields respectively (entries 1 and 2). This gold catalysis worked well with additional anthranils **2d-2h** bearing

Table 3. Reactions on Various Anthranils



[a] [1] = 0.1 M. [b] Product yields are obtained after separation from silica column

various substituents at the C(5) carbon (Y = Me, OMe, CI, Br and OCO₂Me), delivering cyclic nitroxy products **4d-4h** with the yields exceeding 73% (entries 3-7). We tested the reaction of C(3)-substituted anthranil **2i** with Z = Me, further producing compound **4i** in 70% yield (entry 8). A wide scope of applicable enynones and anthranils manifests the practicability of these

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Scheme 1: Gram scale synthesis and chemical functionalizations

[4+3]-nitroxy annulations.

A gram-scale synthesis and chemical fumctionalizations are depicted in scheme 1. Treatment of compound **1a** (1.0 g, 4.06 mmol) with anthranil **2a** (1.5 equiv.) and PPh₃AuCl/AgNTf₂ (5 mmol) in DCM furnished epoxybenzoazepine **3a** in 76% yield (eq 5). Treatment of species **3a** with Zn powder (30 equiv.)/HOAc (5 equiv.)^[11a] in DCM at room temperature for 3 hour led to a N-O bond cleavage to deliver benzoazepine derivative **5** in 90% yield. Compound **5** was subjected to *m*-CPBA-oxidation (1.1 equiv.) in DCM to yield a *N*-hydroxyamine derivative **6** that was characterized by X-ray diffraction.^[10] Alternatively, alkylation of compound **5** with Mel/K₂CO₃ in DMF^[11b] furnished *N*-methylated benzoazepine derivative **7** in excellent yield (95%). Hydrolytic opening of the furan ring of compound **5** with H₂SO₄ in HOAc:H₂O^[11c] (3:1) at 50 °C afforded 3,4-diketodehydrobenzoazepine derivative **8** in 47% yield.

Our next task was to realize these reactions with high enantioselectivity. Initial test with $L_1AuCl/AqNTf_2$ [L₁ = (R)-SEGPHOS] and L₂AuCl/AqNTf₂ [L₂= (R)-BINAP] under standard condition gave a complicated mixture of products, from which we were unable to obtain compound 3a in pure form (entries 1-2). To our pleasure, the use of L₃AuCl/AqNTf₂, $[L_3 = (R)$ -BIPOL-A1] in DCM (25 °C, 1 h) greatly improved the enantioselectivity of compound (+)-3a with 84.6% ee (entry 3). Running the reaction at 0 °C enabled the ee level of compound (+)-3a up to 99.5% ee (entry 4). A bulky chiral phosphoramide L₄, the Feringa ligand, also yielded compound (+)-3a with 99.5% ee, but the yield was slightly decreased to 75% (entry 5). For $L_3 = (R)$ -BIPOL-A1, we altered the silver salts with AgSbF₆ to, afford compound(+)-3a in 68% yield with 92.9% ee (entry 6). The performance of L₃AuCl/AgNTf₂ in different solvents (entries 7-8) at 0 °C were also very effective: DCE (80% yield, 99.4% ee) and toluene (83% yield, 91.1% ee). The absolute configuration of compound (+)-3a was inferred from the X-ray diffraction of its bromo relative (+)-3d.^[10]

The reaction generality of these enantioselective annulations were further assessed with various 2-(1-alkynyl)-2-alken-1-ones 1 and anthranils 2 with variations of the R¹ to R⁶ substituents. In nearly all cases, the use of L₃AuCl/AgNTf₂ enabled high enantioselectivity with ee > 90% whereas compound (+)-4i was obtained in 88% ee using L₄ ligand (entry 12). Most reactions were operated at 0 °C, but in entries 3-4, 6, 9, 11 and 12 the temperatures were kept -10 °C. We first tested the reactions with

Table 4. Enantioselectivity with chiral gold catalysts



[a] [1a] = 0.03M. [b] Product yields are obtained after purification from a silica column. [c] The ee values were determined by chiral HPLC analysis on a chiral stationary phase. [d] Na[$3,5-(CF_3)_2C_6H_3$]4B] (10 mol %) was added [e] Only starting species were recovered.



substrates 1b and 1d bearing various alkynyl substituents with $R^2 = 4-XC_6H_4$ (X = Me and Br); the resulting products (+)-3b and (+)-3d were obtained with ee = 98.5% and 94.0% respectively (entries 1-2). The molecular structure of compound (+)-3d was confirmed with X-ray diffraction.^[10] For meta-chlorophenyl substrate 1e ($R^2 = 3$ -CIC₆H₄) its resulting compound (+)-3e was produced in 74% yield and 95.6% ee (entry 3). A 3-thiophenecontaining 2-(1-alkynyl)-2-alken-1-one 1f delivered the desired product (+)-3f with 92.4% ee (entry 4). For species 1i and 1j bearing variable alkenyl substituents $R^1 = 4-XC_6H_4$ (X = Me and CI) the resulting products (+)-3i and (+)-3i were formed with 99.9% and 91.6% ee respectively (entries 5-6). The variations of ketone substituents (R^3 = ethyl and n-butyl) were amenable to these enantioselective reactions, vielding the desired products(+)-3m and (+)-3n with high ee levels, ca. 91% and 96% (entries 7-8). For anthranils 2d and 2e bearing C(5)substituents (R^4 = Me and OMe), their resulting compounds(+)-4d and (+)-4e were produced with 92% and 98.5% ee (entries 9-10). The enantioselectivity maintained high efficiency for C(6)substituted anthranils 2i (R⁵ = Cl), yielding compound (+)-4i with 90.4% ee. In the case of C(3)-substituted anthranil **2k** ($R^6 = Me$). high enantioselectivity (88.0% ee) was obtained for its corresponding product 4i, despite the generation of a tertiary carbon (entry 12). We also performed this enantioselective annulation on cyclopropyl-substituted alkyne species

Table 5. Enantioselective [4+3]-Annulations



[a] [1] = 0.03 M. [b] Product yields are obtained after separation from silica column. The ee values were determined by HPLC on a chiral stationary phase. [c] These values correspond to reaction temperatures at -10 °C. [d] L_4 ligand was employed in entry 12.

 $(R^2 = cyclopropyl, 1g)$, yielding compound (+)-3g in 69% yield and 91.2% ee.

A gram-scale synthesis is depicted in eq 6. Treatment of compound **1a** (1.0 g, 4.06 mmol) with anthranil **2a** (1.5 equiv.) and chiral catalyst L₃AuCl/AgNTf₂ (5 mol %) in DCM furnished epoxybenzoazepine (+)-**3a** in 75% yield and 98.4 % ee. Treatment of species **3a** with Zn powder (30 equiv.)/HOAc (5 equiv.)^[11a] in DCM (25 °C, 3 h) led to cleavage of a N-O bond to deliver a benzoazepine derivative (+)-**5** in 90% yield and 98.4% ee.



We employ DFT-calculations to rationalize the exostereochemistry of this catalytic [4+3]-nitroxy annulation. Initially, enynone 1 forms gold π -alkyne species to undergo cyclization to generate gold-containing 3-furyl methyl cations A that serve as 1,3-all carbon dipoles; this process is well documented in literatures.^[6] Our DFT calculation indicates a release of 19.9 kcal/mol; its corresponding transition state is depicted in Supporting Information. We were unable to locate the transtion state for a concerted pathway, leading to product 3a. In a stepwise mechanism, an intial N-attack of anthranil to 1,3dipole^[12] A forms an intermediate B via a small barrier, ca. +8.6 kcal/mol; this process is exothermic to release 1.7 kcal/mol. A ring closure of intermediate B forms an endo annulation product C that is more favourable than the exo-isomer C' on both kinetic and thermodymaic aspects. State TsBC is lower than TsBC' by 6.6 kcal/mol whereas intermediate C is more stable than C' by



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Scheme 2. A Plausible Mechanism: the black line indicates the most plausibl pathway to give the observed exo-annulation product.

0.6 kcal/mol. Accordingly, a stepwise mechanism involving an endo route is a viable route based on DFT calculations.

In summary, we report gold-catalyzed [4+3]-annulations^[13] of anthranils^[9] 2-(1-alkynyl)-2-alken-1-ones with to yield epoxybezoazepine products with excellent exo-facial diastereoselectivity (dr > 25:1). This new gold catalysis has a broad scope of applicable substrates. Notably. the enantioselective versions of these [4+3]-cycloadditions have been developed satisfactorily with chiral phosphoramidite gold catalysts^[14] under ambient conditions; their ee levels are up to 99.9%. We perform DFT calculations to support a stepwise mechanism to rationalize the exo diastereoselectivity.

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This work reports gold-catalyzed [4+3]-annulations of 2-(1-alkynyl)-2-alken-1-ones with anthranils to yield epoxybezoazepine products with excellent *exo*-diastereoselectivity (dr > 25:1). More importantly, the enantioselective versions of these [4+3]-cycloadditions have been developed satisfactorily with chiral gold catalysts under ambient conditions (DCM, 0 °C); the ee levels from 88.0-99.9%, further highlighting the synthetic values.

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Gold(I)-Catalyzed Highly Diastereoand Enantioselective Cyclization/[4+3] Annulation Cascades between 2-(1-Alkynyl)-2alken-1-ones and Anthranils