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Exploration of the Molecular Architecture of the Orthosteric Binding Site in the $\alpha 4 \beta 2$ Nicotinic Acetylcholine Receptor with Analogs of 3-(Dimethylamino)butyl Dimethylcarbamate (DMABC) and 1-(Pyridin-3-yl)-1,4-diazepane

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Graphical abstract


# Orthosteric Binding Site in the $\alpha 4 \beta 2$ Nicotinic 

## Acetylcholine Receptor with Analogs of 3-

## (Dimethylamino)butyl Dimethylcarbamate (DMABC) and

1-(Pyridin-3-yl)-1,4-diazepane

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ABSTRACT
X-ray crystal structures of Acetylcholine binding proteins (AChBPs) have revealed two different possible extensions to the classical ligand binding pocket known to accommodate various nicotinic agonists. One of the pockets is limited in size while the other is of considerable dimensions and protrudes along the interfacial cleft between subunits. To probe these putative extensions in functional nicotinic acetylcholine receptors (nAChRs), elongated analogs of 3-(dimethylamino)butyl dimethylcarbamate (DMABC) and 1-(pyridine-3-yl)-1,4-diazepane were prepared and characterized pharmacologically at neuronal heteromeric nAChRs. Although the new analogs, relative to parent compounds, displayed lower binding affinities, functional characterization of selected compounds revealed that they had retained partial $\alpha 4 \beta 2 \mathrm{nAChR}$ agonist activity. The structure-activity relationship data did not indicate an upper limit to the size of substituents as would have been expected if the bound in the smaller pocket. The data were better in agreement with a binding mode in which substituents protrude along the interfacial cleft of the receptor. This was further supported by docking into a homology model of the $\alpha 4-\beta 2$ nAChR interface and by surface plasmon resonance biosensor analysis of binding of the compounds to acetylcholine-binding proteins, where they exhibit preference for Lymnaea stagnalis ACh binding protein (Ls-AChBP) over the Aplysia california ACh binding protein (Ac-AChBP). These results suggest new opportunities for expanding chemical space in the development of partial agonist and may be of interest in relation to development of novel smoking cessation aids.

Keywords: nACh receptor agonists, structure-activity studies, $L s$-AChBP, $A c$-AChBP, Lobeline, DMABC analogs

## 1. Introduction

The neurotransmitter acetylcholine (ACh, 1, Figure 1) mediates its physiological effects in the central and peripheral nervous systems through two classes of receptors: the muscarinic and nicotinic ACh receptors (mAChRs and nAChRs, respectively) [1-4]. Central nAChRs are involved in regulation of release of other neurotransmitters, including dopamine, norepinephrine, serotonin, glutamate, and $\gamma$-aminobutyric acid [58]. The fact that imbalances in the cholinergic system have been proposed to be involved in devastating diseases such as schizophrenia, depression, epilepsy, and Alzheimer's and Parkinson's diseases has spawned the interest in nAChRs as putative drug targets in these disorders [1, 3].

Neuronal nAChRs consist of five subunits that can be either identical (homomeric, $\alpha$ only) or different (heteromeric, combination of $\alpha$ and $\beta$ subunits) [1][12]. The $\alpha 4 \beta 2$ and $\alpha 7$ receptors are the most abundant nAChRs in the brain while $\alpha 3 \mathrm{~b} 4 \mathrm{nAChRs}$ dominate in the autonomic ganglia. [1, 13]. The orthosteric binding sites in the heteromeric receptors are located at the extracellular $\alpha^{(+)} / \beta^{(-)}$interface ( $\alpha$ and $\beta$ contributing with the principal and complementary binding component, respectively).

ACh binding proteins (AChBPs) have proven valuable as surrogate proteins for structural studies of orthosteric ligand binding to nAChRs [16-20]. The AChBP is a soluble homopentameric protein complex with a structure and fold similar to that of the extracellular domain of the nAChR. Although the overall sequence identities between AChBPs and nAChRs are low (20-24\%), the interfacial ACh binding site in the AChBP is highly conserved including the "aromatic box" which consists of five aromatic residues. These residues, $\mathrm{TyrA}, \mathrm{TrpB}, \mathrm{TyrC1}, \mathrm{TyrC} 2$ and TrpD , named according to the loop in the orthosteric nAChR site on which they reside [21] (Figure 2A), are strictly conserved among all nAChRs as well as $L s$-AChBP. In $C t$ - and $A c$-AChBP TrpD in the nAChR is replaced by an isoleucine and a tyrosine, respectively. From more than 50

AChBP co-crystal structures, it is clear that the binding site is flexible, and that the aromatic box residues play an important role in defining its shape and size. When an agonist binds, as illustrated by ( $S$ )-nicotine bound to $L s$-AChBP [24] (Figure 2A), the aromatic residues pack around the ligand to form a closed binding site. In response to more bulky ligands, including antagonists methyllycaconitine [17], $\alpha$-cobratoxin [25], and $\alpha$-conotoxin ImI [26], TyrC1 and TyrC2 on the flexible loop C which forms a lid on the binding site can easily be displaced outwards. Two other aromatic box residues observed to adopt alternative conformation in AChBP structures are $\operatorname{TyrA}$ and $\operatorname{TrpD}$ (TyrD in Ac-AChBP). Alternative conformations of these residues have revealed two different extensions to the binding pocket that can accommodate non-classical ligands, including lobeline [17, 20] (2, Figure 1 and Figure 2B) and the synthetic compound, $3 \alpha-$ (benzoyloxy)-8ß-((R)-2-hydroxy-2-phenylethyl)-8 $\alpha$-methyl-8
azoniabicyclo[3.2.1]octane methiodide (3, Figure 1 and Figure 2C) [27]. From crystallographic studies of $A c$ - AChBP it is known that a sub-pocket is made accessible by rotameric shift of TyrA upon binding of 2 (Figure 2B) [17]. However, mutational studies indicate that the flip is dependent on specific residues on Loop F on the complementary subunit interface to stabilize the tyrosine in its alternative conformation [27], and it remains unknown if the pocket exists in nAChRs. The second sub-pocket is an extension of the binding site along the dimer interface towards the ion channel and has been observed by co-crystallization of $\mathbf{3}$ with $A c$-AChBP [27]. Simultaneous repositioning of TyrA and TyrD in $A c$-AChBP exposes a narrow cleft where the ligand is wedged between the two aromatic residues (Figure 2C). The existence of this interfacial pocket is also evident from co-crystal structures with MLA and from covalent trapping experiments with a maleimide analog of MLA in both $\alpha 7$ and $\alpha 4 \beta 2$ receptors [28, 29], it is evident that the pocket is not an AChBP artifact. The alternative
sub-pockets as extensions to the classical agonist binding site have so far received little attention.

In previous studies, we have explored the medicinal chemistry of 3(dimethylamino)butyl dimethylcarbamate (4, DMABC) and its desmethyl analog 5 [3133], and 1-(pyridin-3-yl)-1,4-diazepane (6) [34] as potent nAChR agonists (Figure 1). Co-crystal structures with $L s$-AChBP have revealed that both 4 and 6 form direct contacts to TyrA, the residue observed to adopt a flipped state in the $A c$ - AChBP cocrystal structure with 2 [35, 36] (SI Figure 1A,B). Inspired by these co-crystal structures, we have in the present study elaborated on the scaffolds of $\mathbf{4}$ and $\mathbf{6}$ to explore possible extensions of the agonist binding pocket and to investigate the structureactivity relationships (SARs) of elongated analogs of these structures at the $\alpha 4 \beta 2$ nAChR.

## 2. Results and discussion

### 2.1 Ligand design

Based on overlays of AChBP co-crystal structures with $\mathbf{4}$ and $\mathbf{6}$ on co-crystal structures with 2 and 3 (Figures 3A-D), we designed a series of ligands with substituents that could protrude towards the two cavities (illustrated in Figures 2B and 2C). In the first series, one of the methyl groups on the basic amine of $\mathbf{4}$ was replaced with 2-hydroxyphenethyl and phenylpropyl substituents to give compounds $\mathbf{7}$ and $\mathbf{8}$ (Table 1). These compounds were designed to challenge the position of $\operatorname{TyrA}$ potentially allowing access to the sub-pocket associated with the binding of 2 (Figure 2B). Along these lines, a series of compounds based on bioisosteric replacements of the dimethylcarbamate moiety with methylcarbamate to give compounds 9-12 (Table 1) was also planned. In a second series, the 3-methyl group of $\mathbf{4}$ was replaced with longer arylalkyl substituents from phenethyl to phenylpentyl to give compounds 13-16 and aryl ethers 17-21 (Table
1). These compounds were inspired from the overlay of $\mathbf{4}$ and $\mathbf{3}$ (Figure 3B) indicating that large substituents in this position could protrude along the interfacial cleft (Figure 2C) without compromising the binding mode of the core agonist fragment. Again, bioisosteric replacements of the dimethylcarbamate with methylcarbamate and an imidazolyloxy moiety were explored resulting in C3 substituted analogs 22-27 (Table 1). In a third series, we explored the 1,4 -diazepane scaffold of 6 . As $N-4$ forms a hydrogen bond to TyrA in the X-ray structure, substitution in this position could lead to either tyrosine flip exposing the subpocket associated with binding of $\mathbf{2}$ in $A c-A C h B P$ or alternatively a minor change in binding mode allowing the substituent to exit the binding site via the cavity along the interfacial cleft (Figure 3C and 3D). Thus benzyl, phenethyl, phenylpropyl, and phenpropenyl substituents were introduced on the $N-4$ of the 1,4-diazepane to challenge these areas in extension to the orthosteric binding site and resulted in compounds 28-33 (Table 2). In the analog series of $\mathbf{6}$, we also explored a series of piperazine analogs, 34-36, with benzyl, phenethyl and phenylpropyl substituents on $N-4$ of the piperazine ring (Table 2).

### 2.2 Chemistry

The syntheses of $N$-arylalkyl analogs of $\mathbf{4}$ (compounds $\mathbf{7 - 1 2}$ ) are illustrated in Scheme 1. The synthesis of $\mathbf{9}$ and $\mathbf{1 0}$ was achieved by a three-step procedure starting from methyl acetoacetate and either methyl benzylamine or methyl phenethylamine to yield 38 and 39. Alcohols 40 and 41 were obtained by reduction with sodium which after treatment with $1,1^{\prime}$-carbonyldiimidazole (CDI) and either methyl- or dimethylamine (in either $\mathrm{H}_{2} \mathrm{O}$ or ethanol) gave 9, $\mathbf{1 0}$ and 37. Products 9 and $\mathbf{3 7}$ were exposed to hydrogenolysis that afforded $\mathbf{4 2}$ and $\mathbf{4 3}$ in good yields which subsequently were reacted with either the appropriate epoxide to give $\mathbf{7}$ and $\mathbf{1 1}$ or subjected to reductive amination to afford $\mathbf{8}$ and $\mathbf{1 2}$.

The synthesis approaches to C-3-arylalkyl analogs of 4 are presented in Schemes 2 and 3. Target compounds 13-15 and 22-25 were synthesized starting from 44-47 (Scheme 2), which have been previously described [37, 38]. Subsequent Michael addition afforded compounds 48-51 that were treated with $\mathrm{LiAlH}_{4}$ to give the alcohols 52-55. Carbamate formations were performed as described above to yield the final compounds. Compound 27 was obtained by reacting the alkoxide of 54 with 2,4,5-tribromo-1-methyl- $1 H$-imidazole and subsequent treatment of the product with $n$-BuLi (Scheme 2 ).

Target compounds $\mathbf{1 7}$ and $\mathbf{1 8}$ were synthesized from 56[31] by an Ullmann like reaction, whereas $\mathbf{1 9 - 2 1}$ were obtained by nucleophilic aromatic substitution with $\mathbf{5 6}$ as the nucleophile (Scheme 3). A different approach was applied for the synthesis of the C3 -arylalkyl analogs $\mathbf{1 6}$ and $\mathbf{2 6}$ to introduce the hydroxyl group in the $\alpha$ position to the phenyl ring (Scheme 3). Alcohol 57[39] was protected as the tetrahydropyranyl (THP) ether using our previously published procedure [31]. The procedures for the following steps to afford $\mathbf{1 6}$ and $\mathbf{2 6}$ are described above, and the final target compounds $\mathbf{1 6}$ and $\mathbf{2 6}$ were obtained after THP-deprotection with $p$-toluenesulfonic acid monohydrate.[31]

The syntheses of the $N$-arylalkyl 3-pyridyl diazacycle analogs are presented in Scheme 4. Analog $\mathbf{2 8}$ was synthesized in two steps from 58[40]. First, intermediate $\mathbf{5 9}$ was obtained by a selective bromination in the 6-position of $\mathbf{5 8}$ using NBS. Subsequent Bocdeprotection followed by reductive amination afforded the analog. Compounds 29-36 were synthesized as described for 28. All target compounds were characterized pharmacologically as the oxalate salts.

### 2.3 Pharmacology and SAR discussion

The binding affinities displayed by the target compounds and reference compounds $\mathbf{1 , 2}$, and 4-6 in a $\left.{ }^{3} \mathrm{H}\right]$-epibatidine competition binding assays to membranes from HEK293 cells stably expressing recombinant rat $\alpha 4 \beta 2, \alpha 4 \beta 4$, and $\alpha 3 \beta 4$ nAChRs are presented in Tables 1 and 2. Functional data for selected compounds and reference compounds determined in a fluorescence-based functional assay the FLIPR Membrane Potential Blue assay and in two-electrode-voltage-clamp (TEVC) electrophysiology recordings at nAChR expressed in Xenopus oocytes are given in Tables 3 and 4. Dissociation constants determined for $\mathbf{3 0}$ and $\mathbf{3 2}$ at $A c$ - and $L s$-AChBP by surface plasmon resonance (SPR) biosensor analyses, respectively, are shown in Table 5.

In the following, SARs have been classified into groups: $N$-arylalkyl analogs of $\mathbf{4}$ (compounds 7-12); C-3-arylalkyl analogs of 4 (compounds 13-26); and $N$-arylalkyl 3pyridyl diazacycle analogs (compounds 28-36).

### 2.3.1 N-arylalkyl analogs of DMABC (4)

Earlier reported SAR studies for analogs of $\mathbf{4}$ showed that incorporating the amino group into a piperidine or pyrrolidine ring decreased binding affinities significantly relative to $\mathbf{4}$ [33]. Based on overlay of the X-ray structure of AChBP co-crystallised with structures 2 and 4 (Figure 3A) this can be rationalized by either a loss of affinity due to the energetic penalty associated with moving TyrA away to allow access to the sub-pocket associated with binding of $\mathbf{2}$ or, alternatively, by an altered binding mode of the core agonist fragment required to position the increased steric bulk in the sub-pocket associated with binding of $\mathbf{3}$. We hypothesized that if the first scenario was in play we should be able to alter the size of the substituent on the basic nitrogen to optimally fit the sub-pocket illustrated in Figure 2B. This should then be reflected in improved
binding affinities up until a certain point after which it should drop drastically. For the second scenario a more flat SAR is to be expected. The resulting $N$-arylalkyl analogs 712 exhibited binding affinities in the mid-to-high micromolar range corresponding to a reduction in affinity at $\alpha 4 \beta 2$ nAChRs of more than 900 -fold compared to 4 . No significant recovery of affinity was observed with increasing size of substituents, nor upon incorporation of a hydroxyl group ( $\mathbf{7}$ and $\mathbf{1 1}$ ) to mimic the hydroxyl group in 2. Together, this SAR does not suggest that substituents access the subpocket analogous to the one observed upon binding of $\mathbf{2}$ in $A c-\mathrm{AChBP}$.

### 2.3.2 C3-arylalkyl analogs of DMABC (4)

We have previously reported that presence of a small hydrophobic group, preferably a methyl group, at the C-3 carbon of $\mathbf{4}$ was optimal for nAChR receptor binding and have found that the $(R)$-enantiomer of $\mathbf{4}$ is significantly more potent than the $(S)$-enantiomer [33]. Introduction of larger hydrophobic substituents, including isopropyl or cyclohexyl, in this position induced a shift from nanomolar to micromolar affinity at the heteromeric nAChRs [33]. Introduction of the more planar cyclopropyl, 3-thiophenyl, and phenyl substituents also led to significant decreases in binding affinity compared to the parent compound, $\mathbf{4}$ [31]. However, overlay of the X-ray structures of $\mathbf{3}$ and $\mathbf{4}$ (Figure 3B) suggested that even larger substituents in the $\mathrm{C}-3$ position of $\mathbf{4}$ could protrude along the interfacial cleft basically without limitation to length of the substituents. Therefore, arylalkyl chains with varying length were introduced in the C-3 position to probe the effect of larger substituents. The resulting compounds (13-15) exhibited affinities in the low micromolar range corresponding to $45-95$ fold loss in affinity relative to 4 . For the monomethylcarbamates (22-25) a decrease in affinities of more than 800 -fold relative to 5 was observed. Interestingly, elongation of the alkyl chain from two to five carbons did not affect binding affinity significantly as evident comparing the phenethyl
substituted analog $22(7.1 \mu \mathrm{M}, \alpha 4 \beta 2 \mathrm{nAChR})$ to the phenylpentyl substituent of $\mathbf{2 5}$ (2.7 $\mu \mathrm{M}, \alpha 4 \beta 2 \mathrm{nAChR})$. A similar trend was observed for the dimethylcarbamate analogs 13-15. This indicates that the binding pocket can accommodate substituents with a significant difference in length and is compatible with the flat SAR that was expected for compounds protruding into an interfacial cleft (Figure 2C).

In a previous study, imidazolyloxy moities were shown to be carbamate bioisosteres at the $\alpha 4 \beta 2$ nAChRs [35]. Exchange of the carbamate in compound 24 for a 1-methyl 1 H imidazoloxy moiety (27) yielded the most potent compound of the series with an affinity at the $\alpha 4 \beta 2$ nAChR of $0.76 \mu \mathrm{M}$ and a selectivity for the $\alpha 4 \beta 2 \mathrm{nAChRs}$ over the $\alpha 4 \beta 4$ and $\alpha 3 \beta 4$ nAChRs, in concordance with data for the parent compounds [35].

To illustrate compatibility of the elongated ligands with a structural model of the receptor, compounds $\mathbf{1 3 - 1 5}, \mathbf{2 2}-\mathbf{2 5}$, and $\mathbf{2 7}$ were docked into a homology model of the $\alpha 4 \beta 2$ nAChR interface constructed using the co-crystal structure of AChBP with $\mathbf{3}$ to ensure a model with an accessible interfacial cleft. The suggested binding modes of the $R$-enantiomers of these compounds only differed with respect to how far along the interfacial cleft the substituent protruded (Figure 4A). In addition, the binding mode was found not to compromise the binding mode of the parent agonist fragment, $\mathbf{4}$, which in a co-crystal structure was identified as the ( $R$ )-enantiomer [35]. Analogs of 4 with aryl ethers in the C3-position (17-21) were found to have decreased affinities relative to 4 and corresponding alkyl phenyl analogs with substituents of comparable size as evident by the 14 - and 3 -fold lower affinity ( $\alpha 4 \beta 2 \mathrm{nAChR}$ ) of $\mathbf{1 7}$ and $\mathbf{2 0}$ relative to $\mathbf{2 2}$ and 23. These data are in concordance with observations made from previous $\mathrm{C}-3$ substituted analogs of 4 in which introduction of a hydroxymethyl group in the C-3 position led to a significantly larger decrease in nAChR binding affinity compared to 4,[31] whereas the binding affinity of the corresponding ethyl analog was much less affected.[41] These data are consistent with a high internal energy of $17(\sim 10 \mathrm{~kJ} / \mathrm{mol})$
for the conformation resembling the bioactive conformation of 4 [35]. Internal hydrogen bonding from the protonated amine to the ether oxygen is preferred as opposed to the internal hydrogen bond to the carbamate oxygen observed in the bound conformation of 4 (SI Figure 1). As the compounds originally were inspired by $\mathbf{2}$, we investigated the effect of a hydroxyl group in the $\alpha$-position relative to the phenyl ring. In $\mathbf{2}$, this is highly favorable and leads to a 25 -fold increase in affinity [42]. In the present series the effect was opposite and led to ca. 10 -fold decrease in affinity at $\alpha 4 \beta 2$ nAChRs (compare 13 and 23 to 16 and 26).

### 2.3.3 N-arylalkyl 3-pyridyl diazacyle analogs

3-Pyridyl substituted diazacycles and diazabicycles are privileged structures and have been the prime scaffolds for synthesis of high affinity $\alpha 4 \beta 2 \mathrm{nAChR}$ agonists. Among these, the scaffold of $\mathbf{6}$ has been extensively applied in the study of nAChRs in which introduction of various substituents in the 5- and 6-positions on the pyridine ring afforded high-affinity compounds. Especially, rather large substituents including methoxyalkyl groups were allowed in the 5-position but not in the 6-position. Furthermore, analogs of $\mathbf{6}$ have been reported in which the 1,4-diazepane ring was exchanged for the piperazine ring which generally led to a reduction in $\alpha 4 \beta 2$ affinities by up to two orders of magnitude [34]. To our knowledge, the nAChRs have not systematically been challenged by large substituents in the $N-4$ position of the scaffold of 6 and the corresponding piperazine scaffold.

The overlay of X-ray structures of $\mathbf{2}$ and $\mathbf{6}$ (Figure 3C) suggested an arylalkyl chain length of approximately 1-2 carbons in the N-4 position of 6 to be optimal for reaching into the subpocket associated to the binding of $\mathbf{2}$. On the other hand, a chain length of three should be too long. Alternatively, if accessing the interfacial cleft pocket, arylalkyl
substituents could in principle be of any length but would require a slight change in binding mode of the core agonist fragment to allow the substituent to escape the cavity. Introduction of phenylalkyl substituents (benzyl, phenethyl, and phenylpropyl) led to compounds 30-32 which displayed 100-150 fold reduced binding affinities at $\alpha 4 \beta 2$ nAChRs compared to the parent compound, $\mathbf{6}$. Since $\mathbf{6}$ is a highly potent $\alpha 4 \beta 2 \mathrm{nAChR}$ agonist (Table 2), the $N-4$ substituted analogs still displayed binding affinities in the low to medium nanomolar range. The equivalent series of 1-(pyridin-3-yl)piperazine analogs (34-36) displayed 5 - to 10 -fold lower affinity at the three heteromeric nAChRs compared to the 1,4 -diazepane containing analogs 30-32. However, a similar SAR pattern was observed for analogs of 6 and the corresponding piperazine analogs indicating a similar binding mode of the analogs in which the benzyl, phenethyl, and phenylpropyl substituents in the $N-4$ positions are tolerated with a flat SAR in agreement with binding along the interfacial cleft as illustrated by docking in Figure 4B.

Based on docking experiments, the analogs 30-32 are positioned less optimal for interaction of the core agonist fragment and in particular the protonated $N-4$ with the receptor. This may explain the relatively larger decrease in affinities (>100-fold) compared to the C-3 substituted analogs of 4 which only suffered a ca. 35 -fold reduction in affinity and which upon docking had its core agonist part optimally positioned as in the X-ray structure. To probe the importance of reduction of entropic penalty of binding, an alkenyl analog of 32, 29, was evaluated. However, 32 and 29 were found equipotent at the receptors indicating that this specific reduction in conformational flexibility is not important for receptor binding. In an attempt to increase binding affinity, a bromine atom was inserted in the 6-position of the pyridine ring of $\mathbf{6}$, which previously was found to improve affinity and potency [34, 43]. However, the analog (28) exhibited a 10 -fold decrease in affinity at the $\alpha 4 \beta 2$ nAChRs
compared to $\mathbf{3 2}$ which again may be indicative of a slightly shifted binding mode of the core agonist fragment. Introduction of bromine in the meta-position of the phenyl ring of $\mathbf{3 1}$ gave compound $\mathbf{3 3}$ which was equipotent to $\mathbf{3 1}$ at the $\alpha 4 \beta 2$ receptors but more than 4 -fold more potent at the $\alpha 3 \beta 4 \mathrm{nAChRs}$ compared to 31 indicating room for improvement of affinity in this specific position.

### 2.3.4 Functional characterization

Selected compounds of the C-3-arylalkyl analogs of $\mathbf{4}$ and the $N$-arylalkyl 3-pyridyl diazacycle analogs of 6 (Table 3) were characterized functionally at recombinant heteromeric mouse $\alpha 4 \beta 2$ and rat $\alpha 3 \beta 4$ nAChRs in the FLIPR Membrane Potential Blue assay. Compounds $\mathbf{8}, \mathbf{1 2}, \mathbf{1 7}$, and $\mathbf{2 0}$ were devoid of activity at concentrations up to 300 $\mu \mathrm{M}$ (not shown). The results revealed all remaining compounds to be weak agonists except for 29, which exhibited weak antagonistic properties (Table 3). The pharmacological activity of compounds $\mathbf{3 0 - 3 3}$ could not be determined in the FLIPR Membrane Potential Blue assay. One of these, 30, was characterized functionally by two-electrode voltage clamp electrophysiological recordings on $\alpha 4 \beta 2$ receptors expressed in Xenopus laevis oocytes. On oocytes injected with $\alpha 4$ and $\beta 2$ mRNA in a 1:4 ratio to favor assembly of $(\alpha 4)_{2}(\beta 2)_{3}$ receptors[15] a partial agonist response corresponding to ca $16 \%$ of the maximal response of ACh and an $\mathrm{EC}_{50}$ value of $1.8 \mu \mathrm{M}$ were observed. On oocytes injected with $\alpha 4$ and $\beta 2$ mRNA in a 10:1 ratio favoring expression of $(\alpha 4)_{3}(\beta 2)_{2}$ receptors the response was $6 \%$ of the maximal ACh elicited current and a slightly higher $\mathrm{EC}_{50}$ of $7.9 \mu \mathrm{M}$. (Table 4 and Figure 5).

The agonistic properties of $\mathbf{1 3}-\mathbf{1 5}, \mathbf{2 3}, \mathbf{2 4}, \mathbf{2 7}, \mathbf{3 0}, \mathbf{3 5}$, and $\mathbf{3 6}$ are highly interesting and support the docking studies suggesting that the substituted analogs have a binding mode
in which the core agonist fragment of 4 and $\mathbf{6}$ at least to some degree is conserved because the compounds are still able to act as agonists.

### 2.3.5 Characterization of compounds $\mathbf{3 0}$ and $\mathbf{3 2}$ at $A c$ - and $L s$-AChBP

Since the target compounds were hypothesized to address cavities based on studies of the $A c$-AChBP and $L s$-AChBP, we included the two AChBPs in the profiling of benzyl and phenylpropyl $N-4$ substituted analogs of 6 ( $\mathbf{3 0}$ and 32) using surface plasmon resonance (SPR) biosensor analyses (Table 5). The reference compounds ( $S$ )-nicotine and $\mathbf{2}$ were included in the study, and the data for these compounds were in agreement with data previously reported [27]. The results revealed $\mathbf{3 0}$ and $\mathbf{3 2}$ to have $\sim 6-$ and $\sim 50-$ fold higher affinity for $L s$ - over $A c$ - AChBP , respectively. Being equipotent at the $\alpha 4 \beta 2$ nAChRs, $32\left(K_{d}=0.020 \mu \mathrm{M}\right)$ showed an 8 -fold higher affinity at $L s$-AChBP compared to $30\left(K_{d}=0.16 \mu \mathrm{M}\right)$ but despite this deviation the binding affinities obtained for compounds $\mathbf{3 0}$ and $\mathbf{3 2}$ to $L s$-AChBP reflected the binding data at $\alpha 4 \beta 2$ nAChRs (Table 5). This correlation corresponds well to previous observations made from analogs of $4[35]$ and $\mathbf{6}[36]$ in which high-affinity compounds at the $\alpha 4 \beta 2$ receptors were found to hold good affinity at $L s$-AChBP. Further, as the hydrophobic cavity associated with binding of 2 in $A c$-AChBP has been suggested not to be present in $L s$-AChBP it supports our conclusion of a binding mode of elongated nAChR agonists along the interfacial cleft of $\alpha 4 \beta 2$ nAChRs.

### 2.4 Conclusion

$N$ - and C-3-arylalkyl analogs of 4 and $N$-arylalkyl analogs of 6 were designed and synthesized to explore possible extensions to the nAChR agonist binding pocket. Based on X-ray structures of AChBP with 2 and 3, two such pockets were potentially accessible but the flat SAR for elongated analogs of the compounds investigated here is
best in agreement with binding in one of these pockets that protrudes along the interfacial cleft as observed in the co-crystal structure of $\mathbf{6}$ with $A c$-AChBP. To be compatible with binding in the hydrophobic lobeline pocket, a clear optimum in terms of binding affinities for compounds with intermediate size substituents was expected but not observed. The suggested binding mode is further supported by docking and SPR analysis on AChBPs which show preference for $L s$-AChBP over $A c$-AChBP. These findings propose a new opportunity for design of partial agonists with properties like $\mathbf{2}$, and the fact that agonistic properties were obtained with using both $\mathbf{4}$ and $\mathbf{6}$ as starting point suggests that this type of ligands could be generally accessible from agonist scaffolds.

## 3. Experimental section

### 3.1 Computational chemistry

### 3.1.1 Homology modeling

A homology model of the $\alpha 4 \beta 2$ nAChR interface was constructed as described by Harpsøe et al.[15] with the following exceptions: The Ac-AChBP crystal structure used as template was $2 \mathrm{Y} 58[27]$ instead of $2 \mathrm{BYQ}[17]$. This change necessitated changes in the alignment to accommodate minor differences in sequence between the two templates. These changes included the removal of H 1 and S 2 in the beginning of the sequence from the alignment (said residues are not present in 2Y58), and changing A43 and A138 to valine (these residues are mutated to valine in 2Y58).

### 3.1.2 Ligand docking

Initally, the $N-4$ benzyl substituted 6 was docked to the model using the induced fit docking protocol implemented in Schrodinger's Maestro v9.3. Default settings were
used except ligand scaling was set to 0.8 and amino acid residues within $8 \AA$ of the docked ligand were sampled. The highest ranking pose with an agonist-like binding mode, i.e. hydrogen bond from pyridine nitrogen to water molecule and hydrogen atom of the protonated $N$-4 of 1,4-diazepane pointing towards the backbone carbonyl oxygen of $\operatorname{Trp} 143$, was selected. In subsequent docking runs the Glide v5.8[44] program was used together with the Glide extra precission (XP) scoring function. The binding modes and docking scores were selected based on ranking according to the Glide E-model score.

### 3.2. Chemistry

### 3.2.1. General procedures

All reactions involving air sensitive reagents were performed under a nitrogen atmosphere using syringe-septum cap techniques. When using dry solvents, diethyl ether $\left(\mathrm{Et}_{2} \mathrm{O}\right)$ was stored over sodium wires, and THF was distilled or obtained from a solvent purification system. Furthermore, dry DCM and DMF were either obtained by drying over molecular sieves ( $4 \AA$ ) or from a solvent purification system. All other solvents, stated dry, were purchased from a commercial source and stored over molecular sieves $(4 \AA)$. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra were obtained on a Varian Gemini 2000 ( 300 MHz ), Varian Mercury Plus ( 300 MHz ), or Bruker Avance (400 MHz ). Data are tabulated in the following order for ${ }^{1} \mathrm{H}$ NMR: chemical shift ( $\delta$ ) (multiplicity (bs, broad singlet; s, singlet; d, doublet; dd, doublet of doublets; dq, dublet of quartets; ddd, doublet of doublet of doublets; dt, doublet of triplets; ddt, doublet of doublet of triplets; dtd, doublet of triplet of doublets; $p$, pentet; $q$, quartet; qd, quartet of doublets; t , triplet; td , triplet of doublets; tt , triplet of triplets; m , multiplet; sxt, sextet; sep, septet), coupling constant(s) $J(\mathrm{~Hz})$, number of protons). The solvent residual peak was used as internal reference. Carbon-fluorine coupling constants observed in ${ }^{13} \mathrm{C}$

NMR of fluorine containing compounds are stated. Analytical thin-layer chromatography (TLC) was carried out using Merck silica gel $60 \mathrm{~F}_{254}$ plates, and the compounds were visualized with UV light ( 254 nm ) and/or $\mathrm{KMnO}_{4}$ spraying reagent. Flash column chromatography (FCC) was performed using Merck silica gel 60 (0.0400.063 mm ). Dry column vacuum chromatography (DCVC) was performed using Merck silica gel $60(0.015-0.040 \mathrm{~mm})$. Reverse phase column chromatography was performed using Merck LiChroprep RP-18 (0.040-0.063 mm). Liquid chromatography-mass spectrometry (LC-MS) was performed using an Agilent 1200 series system connected to a Bruker Esquire 3000Plus Iontrap MS system with a gradient of water $+0.2 \%$ formic $\operatorname{acid}($ Buffer A) and $\mathrm{MeCN}+0.2 \%$ formic acid (Buffer B). All microwave reactions were carried out in a glass vial using a Biotage Initiator instrument. Elemental analyses were performed by Johannes Theiner, Department of Physical Chemistry, University of Vienna, Austria, and are within $\pm 0.4 \%$ of the theoretical values, unless otherwise stated. Melting points (mp) were determined by OptiMelt from Stanford Research Systems in open capillary tubes and are uncorrected.

### 3.2.2. General procedure for oxalate salt formation

The free amine ( 1.0 eq ) was dissolved in a small amount of warm acetone to which a solution of oxalic acid (1.1 eq) in warm acetone was added. The mixture was stored at 5 ${ }^{\circ} \mathrm{C}$ until crystallization. If crystallization did not occur a few drops of $\mathrm{Et}_{2} \mathrm{O}$ was added.

### 3.2.3. 13-((2-Hydroxy-3-phenylpropyl)(methyl)amino)butyl dimethylcarbamate (7)

To a solution of $43(200 \mathrm{mg}, 1.15 \mathrm{mmol})$ in $\mathrm{MeCN}(5 \mathrm{~mL}), \mathrm{LiClO}_{4}(24 \mathrm{mg}, 0.23 \mathrm{mmol})$ and 2-benzyloxirane ( $185 \mathrm{mg}, 1.38 \mathrm{mmol}$ ) were added, and the reaction mixture was stirred at reflux overnight. The reaction mixture was diluted with EtOAc ( 40 mL ) and washed with water ( 30 mL ). The water phase was extracted with EtOAc $(2 \times 40 \mathrm{~mL})$.

The combined organic phases were dried $\left(\mathrm{MgSO}_{4}\right)$, filtered, and evaporated in vacuo. Purification of the crude product was obtained by DCVC (Heptane:EtOAc: $\mathrm{NH}_{3} / 100: 0: 0$ to $0: 99: 1$ ) yielding the product as a clear oil ( $218 \mathrm{mg}, 61 \%$ ). ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300\right.$ $\mathrm{MHz}) \delta$ 7.39-7.14 (m, 5H), 4.23-4.07 (m, 2H), 3.98-3.82 (m, 1H), 3.05-2.76 (m, 8H), $2.68(\mathrm{dd}, 1 \mathrm{H}, J=13.6,5.6 \mathrm{~Hz}), 2.56-2.12(\mathrm{~m}, 5 \mathrm{H}), 1.73-1.54(\mathrm{~m}, 1 \mathrm{H}), 1.36-1.24(\mathrm{~m}$, 1H), $1.03 \quad(\mathrm{~s}, \quad 3 \mathrm{H}) . \quad$ 3-((2-Hydroxy-3-phenylpropyl)-(methyl)amino)butyl dimethylcarbamate oxalate. The free amine of $7(218 \mathrm{mg}, 0.71 \mathrm{mmol})$ was converted into the oxalate salt as described under the general procedure affording white crystals (39 mg, 14\%). Mp: n.d. ${ }^{1} \mathrm{H}$ NMR ( $\mathrm{D}_{2} \mathrm{O}, 300 \mathrm{MHz}$ ) $\delta 7.32-7.09(\mathrm{~m}, 5 \mathrm{H}), 4.20-3.87(\mathrm{~m}$, $3 H), 3.63-3.38(\mathrm{~m}, 1 \mathrm{H}), 3.10-2.91(\mathrm{~m}, 2 \mathrm{H}), 2.82-2.66(\mathrm{~m}, 12 \mathrm{H}), 2.21-1.57(\mathrm{~m}, 2 \mathrm{H}), 1.25$ (dd, $3 \mathrm{H}, J=11.1,6.7 \mathrm{~Hz}) .{ }^{13} \mathrm{C}$ NMR ( $\mathrm{D}_{2} \mathrm{O}, 75 \mathrm{MHz}$ ) $\delta 165.64$ (2C); 157.93, 157.76 (diastereomers); 136.76, 136.69 (diastereomers); 129.52 (2C), 129.39, 128.82 (2C, diastereomers); 127.05; 67.27, 67.09, 66.35, 66.25 (diastereomers); 62.32, 61.88, 61.79, 61.02, 60.52 (diastereomers); 58.57, 57.79, 56.69 (diastereomers); 56.47, 56.09, 54.94 (diastereomers); 40.90, 40.74 (2C, diastereomers); 38.01, 37.28, 36.87 (diastereomers); 36.70, 36.13, 35.74 (diastereomers); 31.17, 31.06, 29.23, 27.95 (diastereomers); 14.27, 13.99, 12.18, 11.65 (diastereomers).

### 3.2.4. 3-[Phenylpropyl(methyl)amino]butyl dimethylcarbamate (8)

To a solution of $43(200 \mathrm{mg}, 1.15 \mathrm{mmol})$ and 3-phenylpropionaldehyde $(0.15 \mathrm{~mL}, 1.15$ mmol) dissolved in 1,2-dichloroethane ( 10 mL ) and $\mathrm{MeOH}(10 \mathrm{~mL})$ was added AcOH ( $0.07 \mathrm{~mL}, 1.15 \mathrm{mmol}$ ) and $\mathrm{NaBH}_{3} \mathrm{CN}(109 \mathrm{mg}, 1.75 \mathrm{mmol})$. The reaction mixture was stirred at room temperature for 72 hours. The reaction mixture was quenched by adding saturated aqueous $\mathrm{NaHCO}_{3}$ solution ( 20 mL ) and was then extracted with EtOAc ( $3 \times$ $30 \mathrm{~mL})$. The organic phases were combined, dried $\left(\mathrm{MgSO}_{4}\right)$, filtered, and evaporated in vacuo. The crude product was purified by DCVC (Toluene:DCM/100:0 to 95:5)
yielding a pale yellow oil. The product was purified further by extraction: The product was poured into water ( 30 mL ) and then acidified with aqueous HCl solution ( $30 \mathrm{~mL}, 2$ M), followed by extraction with $\operatorname{EtOAc}(30 \mathrm{~mL})$. The water phase was then basified with aqueous NaOH solution ( $30 \mathrm{~mL}, 2 \mathrm{M}$ ) and extracted with EtOAc $(2 \times 60 \mathrm{~mL})$. The combined organic phases were dried $\left(\mathrm{MgSO}_{4}\right)$, filtered, and evaporated in vacuo yielding the product as a pale yellow oil ( $127 \mathrm{mg}, 38 \%$ ). ${ }^{1} \mathrm{H} \mathrm{NMR}(\mathrm{CDCl} 3,300 \mathrm{MHz}) \delta$ 7.30-7.08 (m, 5H), 4.12-4.00 (m, 2H), $2.85(\mathrm{~s}, 6 \mathrm{H}), 2.70-2.53(\mathrm{~m}, 2 \mathrm{H}), 2.45-2.15(\mathrm{~m}$, $2 \mathrm{H}), 2.18(\mathrm{~s}, 3 \mathrm{H}), 1.98-1.61(\mathrm{~m}, 4 \mathrm{H}), 1.60-1.41(\mathrm{~m}, 1 \mathrm{H}), 0.99,0.97(2 \mathrm{~s}, 3 \mathrm{H}$, rotamers). 3-[Phenylpropyl(methyl)amino]butyl dimethylcarbamate oxalate. The free amine of 8 ( $104 \mathrm{mg}, 0.36 \mathrm{mmol}$ ) was converted into the oxalate salt as described under the general procedure affording white crystals ( $28 \mathrm{mg}, 27 \%$ ). Mp: 107.5-108.3 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}, 400 \mathrm{MHz}\right) \delta 7.44-7.34(\mathrm{~m}, 2 \mathrm{H}), 7.33-7.26(\mathrm{~m}, 3 \mathrm{H}), 4.24-4.13(\mathrm{~m}, 1 \mathrm{H}), 4.13-4.00$ $(\mathrm{m}, 1 \mathrm{H}), 3.62-3.50(\mathrm{~m}, 1 \mathrm{H}), 3.23-2.99(\mathrm{~m}, 2 \mathrm{H}), 2.86,2.84(2 \mathrm{~s}, 6 \mathrm{H}$, rotamers), 2.79-2.67 $(\mathrm{m}, 5 \mathrm{H}), 2.15-1.71(\mathrm{~m}, 4 \mathrm{H}), 1.34,1.28(2 \mathrm{~d}, 3 \mathrm{H}, J=6.8 \mathrm{~Hz}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(\mathrm{D}_{2} \mathrm{O}, 101 \mathrm{MHz}\right)$ $\delta 165.54$ (2C); 157.78, 157.74 (rotamers); 140.49, 140.42 (rotamers); 128.84 (2C); 128.47; 128.46; 126.58; 61.83 (2C); 58.14, 57.94 (rotamers); 53.29, 52.36 (rotamers); 35.70, 35.14 (rotamers); 31.74 (2C); 30.72; 28.56, 25.74 (rotamers); 13.79, 11.83 (rotamers).

### 3.2.5. 3-[Benzyl(methyl)amino]butyl methylcarbamate (9)

A solution of $40(626 \mathrm{mg}, 3.24 \mathrm{mmol})$ in dry toluene $(9 \mathrm{~mL})$ was added CDI ( 632 mg , 3.90 mmol ), stirred for 3 h at rt , and added dry THF ( 3.3 mL ). After further stirring for 2.5 h , methylamine ( $0.37 \mathrm{~mL}, 7.0 \mathrm{mmol}, 40 \%$ in $\mathrm{H}_{2} \mathrm{O}$ ) was added. The reaction mixture was stirred for 17 h , diluted with EtOAc ( 9 mL ), and washed with brine ( 8 mL ). The organic phase was dried $\left(\mathrm{MgSO}_{4}\right)$, filtered, and evaporated in vacuo. Purification by DCVC (DCM:MeOH: $\mathrm{NH}_{3} / 100: 0: 0$ to $300: 4.5: 0.5$ ) afforded the product as a yellow oil
$(572 \mathrm{mg}, 71 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}\right) \hat{\delta} 7.34-7.18(\mathrm{~m}, 5 \mathrm{H}), 4.48(\mathrm{bs}, 1 \mathrm{H}), 4.18(\mathrm{t}$, $2 \mathrm{H}, J=6 \mathrm{~Hz}), 3.59(\mathrm{~d}, 1 \mathrm{H}, J=12 \mathrm{~Hz}), 3.46(\mathrm{~d}, 1 \mathrm{H}, J=12 \mathrm{~Hz}), 2.89-2.82(\mathrm{~m}, 1 \mathrm{H}), 2.80$, $2.78(2 \mathrm{~s}, 3 \mathrm{H}$, rotamers), $2.14(\mathrm{~s}, 3 \mathrm{H}), 1.91-1.84(\mathrm{~m}, 1 \mathrm{H}), 1.65-1.56(\mathrm{~m}, 1 \mathrm{H}), 1.02(\mathrm{~d}$, $3 \mathrm{H}, J=9 \mathrm{~Hz}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right) \delta 157.8,140.6,129.1$ (2C), 128.5 (2C), 127.1, 63.4, 58.2, 54.5, 36.5, 33.6, 27.9, 13.4. 3-[Benzyl(methyl)amino]butyl methylcarbamate oxalate. The free amine of $9(470 \mathrm{mg}, 1.89 \mathrm{mmol})$ was converted into the oxalate salt as described under the general procedure affording white crystals (383 mg, 49\%) after recrystallization from acetone. Mp: 118.0-119.8 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}, 300 \mathrm{MHz}\right) \delta 7.43-7.40(\mathrm{~m}, 5 \mathrm{H}), 4.32(\mathrm{dd}, 1 \mathrm{H}, J=12,3 \mathrm{~Hz}), 4.21-4.13(\mathrm{~m}, 1 \mathrm{H})$, 4.09-3.98 (m, 2H), 3.52-3.38 (m, 1H), $2.68(\mathrm{~s}, 3 \mathrm{H}), 2.61,2.57(2 \mathrm{~s}, 3 \mathrm{H}$, rotamers), 2.452.38, 2.28-2.18, 2.09-1.99, 1.90-1.79 (4 m, 2H), 1.34, 1.32 ( $2 \mathrm{~d}, 3 \mathrm{H}, J=6 \mathrm{~Hz}$ ). ${ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}, 75 \mathrm{MHz}\right) \delta 165.8$ (2C); 159.0; 130.9 (2C); 130.4; 129.6 (2C); 61.4; 57.7, 57.5 (rotamers); 57.0, 56.5 (rotamers), 35.8, 35.1 (rotamers); 31.0, 29.1 (rotamers); 26.9; 13.9, 12.0 (rotamers).

### 3.2.6. 3-[Phenethyl(methyl)amino]butyl methylcarbamate (10)

Compound $\mathbf{1 0}$ was prepared as described for $\mathbf{9}$ from a mixture of $\mathbf{4 1}$ ( $2.96 \mathrm{~g}, 14.3$ $\mathrm{mmol})$, CDI ( $2.76 \mathrm{~g}, 17.0 \mathrm{mmol}$ ), and methylamine ( 1.65 mL , $29 \mathrm{mmol}, 40 \%$ in $\mathrm{H}_{2} \mathrm{O}$ ). Purification by DCVC (DCM:MeOH: $\mathrm{NH}_{3} / 100: 0: 0$ to 100:4.5:0.5) afforded the product as a yellow oil $(2.95 \mathrm{~g}, 80 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}\right) \delta 7.22-7.07(\mathrm{~m}, 5 \mathrm{H}), 4.48$ (bs, 1 H ), $3.99(\mathrm{t}, 2 \mathrm{H}, J=6 \mathrm{~Hz}), 2.78-2.43(\mathrm{~m}, 5 \mathrm{H}), 2.71,2.70(2 \mathrm{~s}, 3 \mathrm{H}$, rotamers), 2.18 $(\mathrm{s}, 3 \mathrm{H}), 1.72(\mathrm{dq}, 1 \mathrm{H}, J=12,6 \mathrm{~Hz}), 1.45(\mathrm{dq}, 1 \mathrm{H}, J=12,6 \mathrm{~Hz}), 0.88(\mathrm{~d}, 3 \mathrm{H}, J=6 \mathrm{~Hz})$. ${ }^{13} \mathrm{C}_{\mathrm{NMR}}\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right) \delta 157.7,141.1,129.1$ (2C), 128.7 (2C), 126.3, 63.4, 55.8, 55.4, $\quad 37.0, \quad 35.5, \quad 33.4, \quad 31.4, \quad 13.82 . \quad 3$ 3-[Phenethyl(methyl)amino]butyl methylcarbamate oxalate. The free amine of $\mathbf{1 0}(100 \mathrm{mg}, 0.38 \mathrm{mmol})$ was converted into the oxalate salt as described under the general procedure affording white crystals
(103 mg, $77 \%$ ). Mp: $141.1-142.3^{\circ} \mathrm{C} .{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{D}_{2} \mathrm{O}, 300 \mathrm{MHz}\right) \delta 7.36-7.25(\mathrm{~m}, 5 \mathrm{H})$, 4.13-3.95 (m, 2H), 3.62-3.22 (m, 3H), 3.08-2.89 (m, 2H), 2.79, $2.75(2 \mathrm{~s}, 3 \mathrm{H}$, rotamers), 2.62, $2.60(2 \mathrm{~s}, 3 \mathrm{H}$, rotamers), 2.10-1.95 (m, 1 H ), 1.89-1.74 (m, 1H), 1.29, 1.26 (2 d, $3 \mathrm{H}, J=6 \mathrm{~Hz}) .{ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}, 75 \mathrm{MHz}\right) \delta 165.9$ (2C); 159.0; 136.2; 129.4 (2C); 129.0 (2C); 127.8; 61.8, 61.5 (rotamers); 59.1, 58.5 (rotamers); 54.8, 53.4 (rotamers); 36.6, 35.5 (rotamers); 30.7, 30.5 (rotamers); 29.4; 26.9; 13.6, 12.0 (rotamers).

### 3.2.7. 3-((2-Hydroxy-2-phenethyl)(methyl)amino)butyl methylcarbamate (11)

Compound 11 was prepared as described for $\mathbf{7}$ from a mixture of $\mathbf{4 2}(944 \mathrm{mg}, 5.93$ mmol), $\mathrm{LiClO}_{4}$ ( $125 \mathrm{mg}, 1.17 \mathrm{mmol}$ ), and styrene oxide ( $850 \mathrm{mg}, 7.07 \mathrm{mmol}$ ). Purification by HPLC (MeOH: $\left.\mathrm{H}_{2} \mathrm{O}(\mathrm{pH} 10) / 40: 60\right)$ yielded the product as a clear oil ( 84 $\mathrm{mg}, 58 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}\right) \delta 7.40-7.22(\mathrm{~m}, 5 \mathrm{H}), 4.79(\mathrm{bs}, 1 \mathrm{H}), 4.70-4.64$ $(\mathrm{m}, 1 \mathrm{H}), 4.20-4.11(\mathrm{~m}, 2 \mathrm{H}), 2.96-2.74(\mathrm{~m}, 1 \mathrm{H}), 2.81,2.79(2 \mathrm{~d}, 3 \mathrm{H}, \mathrm{J}=6 \mathrm{~Hz}), 2.65,2.54$ ( $2 \mathrm{dd}, 1 \mathrm{H}, J=12,3 \mathrm{~Hz}$ ), 2.43, $2.39(2 \mathrm{dd}, 1 \mathrm{H}, J=12,3 \mathrm{~Hz}$ ), 2.34, $2.27(2 \mathrm{~s}, 3 \mathrm{H}$, rotamers), 1.93-1.75 (m, 1H), 1.68-1.58(m, 1H), $1.00(\mathrm{~d}, 3 \mathrm{H}, J=6 \mathrm{~Hz}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right) \delta 157.6 ; 142.8 ; 128.7$ (2C); 127.8; 126.2 (2C); 69.8, 69.4 (diastereomers); 64.7 (diastereomers); 63.2, 63.0 (diastereomers); 60.9 (diastereomers); 57.4, 55.6 (diastereomers); 37.9 (diastereomers); 34.11; 33.7, 33.5 (diastereomers); 27.8; 14.5, 13.7 (diastereomers). 3-((2-Hydroxy-2-phenethyl)(methyl)amino)butyl methylcarbamate oxalate. The free amine of $\mathbf{1 1}(65 \mathrm{mg}, 0.23 \mathrm{mmol})$ was converted into the oxalate salt as described under the general procedure affording white crystals ( $27 \mathrm{mg}, 22 \%$ ). Mp: 80.5-82.9 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $\mathrm{D}_{2} \mathrm{O}, 300 \mathrm{MHz}$ ) $\delta 4.98-4.94(\mathrm{~m}, 1 \mathrm{H}), 4.06-$ $3.93(\mathrm{~m}, 2 \mathrm{H}), 3.61-3.41(\mathrm{~m}, 1 \mathrm{H}), 3.35-3.11(\mathrm{~m}, 2 \mathrm{H}), 2.84,2.79,2.68(3 \mathrm{~s}, 3 \mathrm{H}$, diastereomers), 2.52, $2.50(2 \mathrm{~s}, 3 \mathrm{H}$, diastereomers), 2.13-1.95 (m, 1H), 1.84-1.61 (m, 1H), 1.26-1.16 (m, 3H). ${ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}, 75 \mathrm{MHz}\right) \delta 166.0$ (2C); 159.0; 139.5; 129.1 (2C); 126.3 (2C); 126.2; 68.4, 67.4 (diastereomers); 62.0, 61.8 (diastereomers); 61.3;
58.1; 38.1, 36.4 (diastereomers); 31.0, 29.5 (diastereomers); 28.9, 26.9 (diastereomers); 13.7, 12.4 (diastereomers).

### 3.2.8. 3-[Phenylpropyl(methyl)amino]butyl methylcarbamate (12)

Compound 12 was prepared as described for $\mathbf{8}$ from a mixture of $\mathbf{4 2}$ ( $200 \mathrm{mg}, 1.25$ mmol), 3-phenylpropionaldehyde ( $0.16 \mathrm{~mL}, 1.25 \mathrm{mmol}$ ), and $\mathrm{NaBH}_{3} \mathrm{CN}(109 \mathrm{mg}, 1.75$ $\mathrm{mmol})$ affording the product as a yellow oil $(104 \mathrm{mg}, 30 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 300\right.$ MHz) $\delta 7.40-7.10(\mathrm{~m}, 5 \mathrm{H}), 4.53(\mathrm{~s}, 1 \mathrm{H}), 4.14(\mathrm{~m}, 2 \mathrm{H}), 2.90-2.70(\mathrm{~m}, 4 \mathrm{H}), 2.69-2.51$ $(\mathrm{m}, 2 \mathrm{H}), 2.49-2.28(\mathrm{~m}, 2 \mathrm{H}), 2.18(\mathrm{~s}, 3 \mathrm{H}), 1.92-1.66(\mathrm{~m}, 3 \mathrm{H}) 1.64-1.43(\mathrm{~m}, 1 \mathrm{H}), 0.90$, 0.93 ( $2 \mathrm{~s}, 3 \mathrm{H}$, rotamers). 3-[Phenylpropyl-(methyl)amino]butyl methylcarbamate oxalate. The free amine of $\mathbf{1 2}(104 \mathrm{mg}, 0.37 \mathrm{mmol})$ was converted into the oxalate salt as described under the general procedure affording white crystals ( $52 \mathrm{mg}, 22 \%$ ). Mp: 121.5-123.3 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}, 400 \mathrm{MHz}\right) \delta 7.43-7.34(\mathrm{~m}, 2 \mathrm{H}), 7.33-7.24(\mathrm{~m}, 3 \mathrm{H})$, 4.25-4.14 (m, 1H), 4.12-3.99 (m, 1H), 3.61-3.48 (m, 1H), 3.22-2.97 (m, 2H), 2.80-2.62 $(\mathrm{m}, 8 \mathrm{H}), 2.15-1.95(\mathrm{~m}, 3 \mathrm{H}), 1.93-1.70(\mathrm{~m}, 1 \mathrm{H}), 1.33,1.27(2 \mathrm{~d}, 3 \mathrm{H}, J=6.5 \mathrm{~Hz}) .{ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}, 101 \mathrm{MHz}\right) \delta 165.53$ (2C); 140.50, 140.45 (rotamers); 128.84 (2C); 128.49; $128.47 ; 126.58 ; 118.92 ; 61.31 ; 58.12 ; 53.31 ; 35.11 ; 31.74 ; 30.61 ; 28.81,26.64$ (rotamers); 25.74, 25.72 (rotamers); 13.60, 11.81 (rotamers).

### 3.2.9. 3-(Dimethylamino)-6-phenylhexyl dimethylcarbamate (13)

A solution of $53(1.52 \mathrm{~g}, 6.87 \mathrm{mmol})$ in dry toluene $(25 \mathrm{~mL})$ was added CDI $(1.34 \mathrm{~g}$, $8.24 \mathrm{mmol})$. The reaction mixture was stirred for 2 h at rt . The reaction mixture was added THF ( 20 mL ) and stirred for additional one hour. Dimethylamine $(8.70 \mathrm{~mL}, 68.7$ $\mathrm{mmol}, 40 \%$ in water) was added the reaction mixture, and stirring continued overnight at rt. The reaction mixture was diluted with EtOAc ( 25 mL ). The water phase was washed with brine ( 25 mL ) and aqueous NaOH solution ( $1 \mathrm{M}, 2 \times 25 \mathrm{~mL}$ ). The
combined organic phases were dried $\left(\mathrm{MgSO}_{4}\right)$, filtered, and evaporated yielding the product as a yellow oil ( $1.54 \mathrm{~g}, 77 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta$ 7.29-7.27 (m, 1 H$)$, $7.25(\mathrm{~d}, 1 \mathrm{H}, J=1.8 \mathrm{~Hz}), 7.18-7.16(\mathrm{~m}, 3 \mathrm{H}), 4.10(\mathrm{~d}, 2 \mathrm{H}, J=13.7 \mathrm{~Hz}), 2.89(\mathrm{~s}, 6 \mathrm{H}), 2.61$ $(\mathrm{t}, 2 \mathrm{H}, J=7.6 \mathrm{~Hz}), 2.52(\mathrm{t}, 1 \mathrm{H}, J=5.8 \mathrm{~Hz}), 2.23(\mathrm{~s}, 6 \mathrm{H}), 1.82(\mathrm{dq}, 1 \mathrm{H}, J=13.6,6.8$ $\mathrm{Hz}), 1.70-1.64(\mathrm{~m}, 2 \mathrm{H}), 1.57(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=6.9 \mathrm{~Hz}), 1.35-1.26(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right.$, $101 \mathrm{MHz}) \delta 156.86,142.54,128.52$ (2C), 128.41 (2C), 125.84, 63.91, 61.00, 40.41 (4C), 36.20, 29.48, 29.10, 28.95. 3-(Dimethylamino)-6-phenylhexyl dimethylcarbamate oxalate. The free amine of $\mathbf{1 3}(380 \mathrm{mg}, 1.30 \mathrm{mmol})$ was converted into the oxalate salt as described under the general procedure affording white crystals (291 mg, 59\%). Mp: $97.9{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $\mathrm{D}_{2} \mathrm{O}, 400 \mathrm{MHz}$ ) $\delta 7.34-7.29(\mathrm{~m}, 2 \mathrm{H}), 7.24-7.20$ (m, 3H), 4.11 (ddd, 1H, $J=11.5,6.6,4.9 \mathrm{~Hz}), 4.02(\mathrm{ddd}, 1 \mathrm{H}, J=11.7,7.4,4.5 \mathrm{~Hz})$, 3.28-3.22 (m, 1H), 2.84 (s, 6H), 2.79, 2.77 (2 s, 6H, rotamers), 2.69-2.61 (m, 2H), 2.06 (ddt, $1 \mathrm{H}, J=15.1,7.6,4.9 \mathrm{~Hz}), 1.93-1.85(\mathrm{~m}, 1 \mathrm{H}), 1.73-1.61(\mathrm{~m}, 4 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}\right.$, $101 \mathrm{MHz}) \delta 165.63$ (2C), 157.67, 141.74, 128.69 (2C), 128.55 (2C), 126.23, 63.55 (2C), $62.28,39.54,38.77,34.36$ (2C), 27.72, 27.66, 26.70.

### 3.2.10. 3-(Dimethylamino)-7-phenylheptyl dimethylcarbamate (14)

Compound $\mathbf{1 4}$ was prepared as described for $\mathbf{1 3}$ from a mixture of $\mathbf{5 4}(800 \mathrm{~g}, 3.40$ $\mathrm{mmol})$, $\mathrm{CDI}(660 \mathrm{~g}, 4.08 \mathrm{mmol})$, and dimethylamine ( $4.3 \mathrm{~mL}, 34 \mathrm{mmol}, 33 \% \mathrm{in} \mathrm{EtOH}$ ) yielding the product as a yellow oil ( $860 \mathrm{mg}, 64 \%$ ). ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta$ 7.29-7.27 (m, 2H), 7.17 (dt, 3H, $J=1.4,0.7 \mathrm{~Hz}), 4.12(\mathrm{t}, 2 \mathrm{H}, J=6.9 \mathrm{~Hz}), 2.90(\mathrm{~s}, 6 \mathrm{H})$, $2.61(\mathrm{t}, 2 \mathrm{H}, J=7.8 \mathrm{~Hz}), 2.49-2.47(\mathrm{~m}, 1 \mathrm{H}), 2.24(\mathrm{~s}, 6 \mathrm{H}), 1.64-1.61(\mathrm{~m}, 2 \mathrm{H}), 1.59-1.52$ (m, 2H), 1.42-1.26 (m, 4H). 3-(Dimethylamino)-7-phenylheptyl dimethylcarbamate oxalate. The free amine of $\mathbf{1 4}(367 \mathrm{mg}, 1.20 \mathrm{mmol})$ was converted into the oxalate salt as described under the general procedure affording white crystals ( $346 \mathrm{mg}, 94 \%$ ). Mp : $90.4{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}, 400 \mathrm{MHz}\right) \delta 7.42-7.37(\mathrm{~m}, 2 \mathrm{H}), 7.33-7.27(\mathrm{~m}, 3 \mathrm{H}), 4.22(\mathrm{ddd}$,
$1 \mathrm{H}, J=11.5,6.5,5.1 \mathrm{~Hz}), 4.13(\mathrm{ddd}, 1 \mathrm{H}, J=11.7,7.4,4.5 \mathrm{~Hz}), 3.33(\mathrm{tt}, 1 \mathrm{H}, J=7.6$, $5.1 \mathrm{~Hz}), 2.90(\mathrm{~s}, 6 \mathrm{H}), 2.80,2.79(2 \mathrm{~s}, 6 \mathrm{H}$, rotamers), $2.70(\mathrm{t}, 2 \mathrm{H}, J=7.4 \mathrm{~Hz}), 2.15$ (ddt, $1 \mathrm{H}, J=15.0,7.6,5.0 \mathrm{~Hz}), 2.02-1.93(\mathrm{~m}, 1 \mathrm{H}), 1.82-1.68(\mathrm{~m}, 4 \mathrm{H}), 1.43(\mathrm{dt}, 2 \mathrm{H}, J=15.3$, $7.6 \mathrm{~Hz}) .{ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}, 101 \mathrm{MHz}\right) \delta 165.65(2 \mathrm{C}), 157.68,142.65,128.60(2 \mathrm{C}), 128.54$ (2C), 125.98, 63.57, 62.20, 39.42 (2C), 38.94 (2C), 34.51, 30.05, 28.18, 27.77, 24.58.

### 3.2.11. 3-(Dimethylamino)-8-phenyloctyl dimethylcarbamate (15)

Compound $\mathbf{1 5}$ was prepared as described for $\mathbf{1 3}$ from a mixture of $55(488 \mathrm{mg}, 1.96$ mmol ), CDI ( $381 \mathrm{mg}, 2.35 \mathrm{mmol}$ ), and dimethylamine ( $1.75 \mathrm{~mL}, 9.78 \mathrm{mmol}, 33 \% \mathrm{in}$ $\mathrm{EtOH})$ giving the product as a yellow oil ( $332 \mathrm{mg}, 53 \%$ ). ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 400 \mathrm{MHz}\right)$ $\delta 7.27-7.19(\mathrm{~m}, 2 \mathrm{H}), 7.18-7.10(\mathrm{~m}, 3 \mathrm{H}), 4.10(\mathrm{td}, 2 \mathrm{H}, J=6.7,1.9 \mathrm{~Hz}), 2.89(\mathrm{~s}, 6 \mathrm{H}), 2.61$ $(\mathrm{t}, 2 \mathrm{H}, J=7.8 \mathrm{~Hz}), 2.48(\mathrm{p}, 1 \mathrm{H}, J=6.3 \mathrm{~Hz}), 2.22(\mathrm{~s}, 6 \mathrm{H}), 1.84(\mathrm{dq}, 1 \mathrm{H}, J=14.1,6.3$ $\mathrm{Hz}), 1.67-1.51(\mathrm{~m}, 4 \mathrm{H}), 1.42-1.22(\mathrm{~m}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 101 \mathrm{MHz}\right) \delta 144.04$, $130.08,129.56$ (2C), 129.41 (2C), $126.80,65.23,62.32,40.85$ (2C), $37.00,36.32$ (2C) $32.78, \quad 30.74, \quad 30.53, \quad 30.45, \quad 28.13 . \quad 3$-(Dimethylamino)-8-phenyl-octyl dimethylcarbamate oxalate. The free amine of $\mathbf{1 5}(128 \mathrm{mg}, 0.40 \mathrm{mmol})$ was converted into the oxalate salt according to the general procedure affording a white solid ( 64 mg , $39 \%$ ). Mp: 84.7-86.0 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}, 400 \mathrm{MHz}\right) \delta 7.38-7.32(\mathrm{~m}, 2 \mathrm{H}), 7.30-7.21$ (m, $3 \mathrm{H}), 4.23-4.07(\mathrm{~m}, 2 \mathrm{H}), 3.27(\mathrm{~d}, 1 \mathrm{H}, J=6.8 \mathrm{~Hz}), 2.87,2.87(2 \mathrm{~s}, 6 \mathrm{H}$, rotamers), 2.80 $(\mathrm{dd}, 6 \mathrm{H}, J=6.3,1.5 \mathrm{~Hz}), 2.66-2.59(\mathrm{~m}, 2 \mathrm{H}), 2.17-2.06(\mathrm{~m}, 1 \mathrm{H}), 2.01-1.88(\mathrm{~m}, 1 \mathrm{H})$, 1.76-1.57 (m, 4H), 1.44-1.28 (m, 4H). ${ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}, 101 \mathrm{MHz}\right) \delta 165.64$ (2C), 152.37, 143.16, 128.59 (2C), 128.58 (2C), 125.88. 63.53, 62.31, 39.49, 38.87, 35.94, 35.56, 34.84, 34.82, 30.20, 28.28, 27.74, 24.87.
3.2.12. 3-(Dimethylamino)-6-hydroxy-6-phenylhexyl dimethylcarbamate (16)

To a solution of $61(0.92 \mathrm{~g}, 2.34 \mathrm{mmol})$ in $\mathrm{MeOH}(50 \mathrm{~mL})$ was added $p$-toluenesulfonic acid monohydrate $(0.89 \mathrm{~g}, 4.68 \mathrm{mmol})$. The reaction mixture was stirred for 2 h at rt . The reaction mixture was added saturated aqueous $\mathrm{NaHCO}_{3}$ solution $(50 \mathrm{~mL})$ and then reduced in vacuo. The remaining mixture was extracted with $\mathrm{DCM}(3 \times 50 \mathrm{~mL})$ and the combined organic phases were dried, filtered, and evaporated in vacuo. The crude product was purified by DCVC ( $\mathrm{DCM}: \mathrm{MeOH}: \mathrm{NH}_{3} / 90: 9: 1$ ) yielding the product as a yellow oil ( $90 \mathrm{mg}, 13 \%$ ). ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 7.38-7.19(\mathrm{~m}, 5 \mathrm{H}), 4.81$ (dd, $1 \mathrm{H}, J=6.0,3.6 \mathrm{~Hz}), 4.16-4.10(\mathrm{~m}, 1 \mathrm{H}), 4.05-3.99(\mathrm{~m}, 1 \mathrm{H}), 2.89(\mathrm{~s}, 6 \mathrm{H}), 2.56(\mathrm{tt}, 1 \mathrm{H}, J$ $=9.7,3.3 \mathrm{~Hz}), 2.33,2.25(2 \mathrm{~s}, 6 \mathrm{H}$, rotamers $)$, 2.17-2.09 $(\mathrm{m}, 1 \mathrm{H}), 1.97-1.87(\mathrm{~m}, 2 \mathrm{H})$,
 128.21 (2C), 126.59, 125.74 (2C), 72.17, 63.56, 61.51, 40.08 (4C), 37.21, 27.29, 26.73.

3-(Dimethylamino)-6-hydroxy-6-phenylhexyl dimethylcarbamate oxalate. The free amine of $\mathbf{1 6}(90 \mathrm{mg}, 0.29 \mathrm{mmol})$ was converted into the oxalate salt as described under the general procedure affording white crystals ( $67 \mathrm{mg}, 58 \%$ ). Mp: $119.3{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}, 400 \mathrm{MHz}\right) \delta 7.51-7.40(\mathrm{~m}, 5 \mathrm{H}), 4.81(\mathrm{~d}, 1 \mathrm{H}, J=6.3 \mathrm{~Hz}), 4.20(\mathrm{ddd}, 1 \mathrm{H}, J=11.5$, $6.8,4.9 \mathrm{~Hz}), 4.12(\mathrm{ddd}, 1 \mathrm{H}, J=11.7,7.3,4.6 \mathrm{~Hz}), 3.35(\mathrm{td}, 1 \mathrm{H}, J=6.2,2.1 \mathrm{~Hz}), 2.89(\mathrm{~s}$, $6 \mathrm{H}), 2.85,2.80\left(2 \mathrm{~s}, 6 \mathrm{H}\right.$, rotamers), 2.21-2.13(m, 1H), 2.04-1.68(m,5H). ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}, 101 \mathrm{MHz}\right) \delta 165.69(2 \mathrm{C}), 157.66,142.95,128.82(2 \mathrm{C}), 128.08,126.02(2 \mathrm{C})$, 73.01, 63.41, 62.23, 39.47 (2C), 38.87 (2C), 33.63, 27.71, 24.73.

### 3.2.13. 3-(Dimethylamino)-4-phenoxybutyl dimethylcarbamate (17)

56[31] ( $200 \mathrm{mg}, 0.98 \mathrm{mmol}$ ), CuI ( $6 \mathrm{mg}, 0.03 \mathrm{mmol}$ ), 3,4,7,8-tetramethyl-1,10phenanthroline ( $15 \mathrm{mg}, 0.07 \mathrm{mmol}$ ), iodobenzene ( $73 \mu \mathrm{~L}, 0.65 \mathrm{mmol}$ ), and $\mathrm{Cs}_{2} \mathrm{CO}_{3}$ (319 $\mathrm{mg}, 0.98 \mathrm{mmol})$ were mixed and added dry toluene $(0.50 \mathrm{~mL})$ under nitrogen. The reaction mixture was heated to $110{ }^{\circ} \mathrm{C}$ overnight, diluted with EtOAc ( 15 mL ), and filtered through a plug of silica washing additional times with EtOAc. The filtrate was
concentrated in vacuo yielding the product as a brown oil ( $136 \mathrm{mg}, 60 \%$ ). ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}\right) \delta 7.38-7.28(\mathrm{~m}, 2 \mathrm{H}), 7.05-6.88(\mathrm{~m}, 3 \mathrm{H}), 4.22(\mathrm{td}, 2 \mathrm{H}, J=6.6,1.9$ $\mathrm{Hz})$, 4.17-4.05 (m, 1H), 4.02-3.96 (m, 1H), 3.05-3.01 (m, 1H), 2.92, $2.89(2 \mathrm{~s}, 6 \mathrm{H}$, rotamers), $2.43(\mathrm{~s}, 6 \mathrm{H}), 1.99-1.87(\mathrm{~m}, 3 \mathrm{H}), 1.73-1.56(\mathrm{~m}, 3 \mathrm{H})$. 3-(Dimethylamino)-4phenoxybutyl dimethylcarbamate oxalate. The free amine of $\mathbf{1 7}(185 \mathrm{mg}, 0.66 \mathrm{mmol})$ was converted into the oxalate salt as described under the general procedure affording white crystals ( $66 \mathrm{mg}, 27 \%$ ). Mp: 91.3-92.5 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $\mathrm{D}_{2} \mathrm{O}, 300 \mathrm{MHz}$ ) $\delta 7.40-7.35$ $(\mathrm{m}, 2 \mathrm{H}), 7.09-7.00(\mathrm{~m}, 3 \mathrm{H}), 4.46(\mathrm{dd}, 1 \mathrm{H}, J=11.9,3.1 \mathrm{~Hz}), 4.31(\mathrm{dd}, 1 \mathrm{H}, J=11.9,6.6$ $\mathrm{Hz}), 4.23(\mathrm{dd}, 2 \mathrm{H}, J=9.8,5.7 \mathrm{~Hz}), 3.84-3.81(\mathrm{~m}, 1 \mathrm{H}), 2.93,2.88(2 \mathrm{~s}, 6 \mathrm{H}$, rotamers $)$, $2.77(\mathrm{~s}, 6 \mathrm{H}), 2.27(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}, 75 \mathrm{MHz}\right) \delta 241.85(2 \mathrm{C}), 215.94,186.69$, 144.11 (2C), 136.32, 128.79 (2C), 78.13, 77.16, 76.48, 55.46, 53.41, 50.18, 49.74, 38.60.

### 3.2.14. 3-(Dimethylamino)-4-(4-fluorophenoxy)butyl dimethylcarbamate (18)

Compound $\mathbf{1 8}$ was prepared as described for $\mathbf{1 7}$ from a mixture of $\mathbf{5 6}(200 \mathrm{mg}, 0.98$ mmol ), $\mathrm{CuI}(6 \mathrm{mg}, 0.03 \mathrm{mmol}), 3,4,7,8$-tetramethyl-1,10-phenanthroline ( $16 \mathrm{mg}, 0.07$ mmol ), 1-fluoro-4-iodobenzene ( $76 \mu \mathrm{~L}, 0.66 \mathrm{mmol}$ ), and $\mathrm{Cs}_{2} \mathrm{CO}_{3}(320 \mathrm{mg}, 0.98 \mathrm{mmol})$. Purification by reverse phase column chromatography (C18 silica) $\left(\mathrm{H}_{2} \mathrm{O}: \mathrm{AcOH}: \mathrm{MeCN} / 99.9: 0.1: 0\right.$ to $\left.94.9: 0.1: 5\right)$ followed by basic extraction of the product into EtOAc afforded the product as a yellow oil ( $46 \mathrm{mg}, 23 \%$ ). ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 300 \mathrm{MHz}\right) \delta 7.04-6.90(\mathrm{~m}, 3 \mathrm{H}), 4.19(\mathrm{t}, 2 \mathrm{H}, J=6.5 \mathrm{~Hz}), 4.07(\mathrm{qd}, 2 \mathrm{H}, J=$ 10.3, 5.0 Hz), 3.01-2.97 (m, 1H), $2.88(\mathrm{~s}, 5 \mathrm{H}), 2.37(\mathrm{~s}, 6 \mathrm{H}), 2.01-1.91(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{CD}_{3} \mathrm{OD}, 75 \mathrm{MHz}\right) \delta 160.06$, $158.11\left(\mathrm{C}-\mathrm{F}:{ }^{1} \mathrm{~J}=147 \mathrm{~Hz}\right) ; 156.92 ; 156.06,156.03$ (C-F: $\left.{ }^{4} J=2.1 \mathrm{~Hz}\right) ; 116.72,116.59\left(\mathrm{C}-\mathrm{F}:{ }^{2} J=8.5 \mathrm{~Hz}, 2 \mathrm{C}\right) ; 116.49,116.41\left(\mathrm{C}-\mathrm{F}:{ }^{3} J=5.8\right.$ Hz, 2C); 68.59; 64.57; 61.29; 41.61 (2C); 36.57; 36.13; 28.22. 3-(Dimethylamino)-4-(4-fluorophenoxy)butyl dimethylcarbamate oxalate. The free amine of $\mathbf{1 8}$ ( 46 mg ,
0.15 mmol ) was converted into the oxalate salt as described under the general procedure affording white crystals ( $43 \mathrm{mg}, 72 \%$ ). Mp: 97.6-98.3 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $\mathrm{D}_{2} \mathrm{O}, 300 \mathrm{MHz}$ ) $\delta$ 7.07-7.00 (m, 2H), 6.97-6.90 (m, 2H), 4.39 (dd, 1H, $J=11.8,3.1 \mathrm{~Hz}), 4.27-4.12(\mathrm{~m}$, $4 \mathrm{H}), 3.78(\mathrm{qd}, 1 \mathrm{H}, J=6.7,3.0 \mathrm{~Hz}), 2.87,2.82(2 \mathrm{~s}, 6 \mathrm{H}$, rotamers $), 2.68,2.66(2 \mathrm{~s}, 5 \mathrm{H}$, rotamers), 2.18-2.26 (m, 2H). ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}, 75 \mathrm{MHz}\right) \delta 165.59 ; 159.18$, $156.04(\mathrm{C}-\mathrm{F}:$ $\left.{ }^{1} J=245 \mathrm{~Hz}\right) ; 157.52 ; 153.29,153.26\left(\mathrm{C}-\mathrm{F}:{ }^{3} J=2.3 \mathrm{~Hz}, 2 \mathrm{C}\right) ; 116.19,115.80\left(\mathrm{C}-\mathrm{F}:{ }^{2} J=\right.$ $21 \mathrm{~Hz}) ; 115.88,115.69\left(\mathrm{C}-\mathrm{F}:{ }^{2} \mathrm{~J}=21 \mathrm{~Hz}\right) ; 64.48 ; 62.97 ; 62.24 ; 41.42 ; 39.16 ; 35.96$; 35.52; 24.36.
3.2.15. 3-(Dimethylamino)-4-(pyridin-2-yloxy)butyl dimethylcarbamate (19)

To a suspension of NaH ( $65 \mathrm{mg}, 1.62 \mathrm{mmol}, 60 \%$ in mineral oil) in dry DMF ( 6 mL ) was dropwisely added $56(300 \mathrm{mg}, 1.47 \mathrm{mmol})$ at $0^{\circ} \mathrm{C}$. After 1 h of stirring at $0^{\circ} \mathrm{C}, 2-$ fluoropyridine ( $0.25 \mathrm{~mL}, 2.94 \mathrm{mmol}$ ) was added. The reaction mixture was stirred overnight at rt , added water $(60 \mathrm{~mL})$, and extracted with $\mathrm{Et}_{2} \mathrm{O}(2 \times 60 \mathrm{~mL})$. The combined organic phases were dried $\left(\mathrm{MgSO}_{4}\right)$, filtered, and evaporated in vacuo. Purification by reverse phase column chromatography (C18 silica) $\left(\mathrm{H}_{2} \mathrm{O}:\right.$ TFA:MeCN/99.9:0.1:0 to 89.9:0.1:10) followed by basic extraction of the product into EtOAc afforded the product as a yellow oil ( $224 \mathrm{mg}, 54 \%$ ). ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 300\right.$ $\mathrm{MHz}) \delta 8.15(\mathrm{dd}, 1 \mathrm{H}, J=5.1,1.3 \mathrm{~Hz}), 7.75(\mathrm{ddd}, 1 \mathrm{H}, J=8.4,7.1,1.9 \mathrm{~Hz}), 7.04(\mathrm{ddd}$, $1 \mathrm{H}, J=7.1,5.2,1.0 \mathrm{~Hz}), 6.93(\mathrm{~d}, 1 \mathrm{H}, J=8.4 \mathrm{~Hz}), 4.80(\mathrm{dd}, 1 \mathrm{H}, J=13.1,3.0 \mathrm{~Hz}), 4.63$ (dd, $1 \mathrm{H}, J=13.1,6.6 \mathrm{~Hz}), 4.54-4.50(\mathrm{~m}, 1 \mathrm{H}), 4.31-4.24(\mathrm{~m}, 2 \mathrm{H}), 3.86(\mathrm{qd}, 1 \mathrm{H}, J=6.5$, 3.3 Hz ), $3.02(\mathrm{~s}, 6 \mathrm{H}), 2.86(\mathrm{~s}, 6 \mathrm{H}), 2.34-2.16(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 75 \mathrm{MHz}\right) \delta$ $163.25,157.57,147.31,140.95,119.18,112.10,63.82,63.12,62.83,41.19$ (2C), 36.68, 36.15, 26.30. 3-(Dimethylamino)-4-(pyridin-2-yloxy)butyl dimethylcarbamate oxalate. The free amine of $\mathbf{1 9}(224 \mathrm{mg}, 0.80 \mathrm{mmol})$ was converted into the oxalate salt as described under the general procedure affording white crystals (141 mg, 47\%). Mp:
$77.1^{\circ} \mathrm{C} .{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{D}_{2} \mathrm{O}, 300 \mathrm{MHz}\right) \delta 8.08(\mathrm{dd}, 1 \mathrm{H}, J=5.2,1.7 \mathrm{~Hz}), 7.77(\mathrm{ddd}, 1 \mathrm{H}, \mathrm{J}=$ 8.6, 7.1, 1.9 Hz ), 7.11-7.07 (m, 1H), $6.94(\mathrm{dd}, 1 \mathrm{H}, J=8.4,0.6 \mathrm{~Hz}), 4.66(\mathrm{dd}, 1 \mathrm{H}, J=$ $12.6,2.7 \mathrm{~Hz}), 4.50(\mathrm{dd}, 1 \mathrm{H}, J=12.6,6.1 \mathrm{~Hz}), 4.26-4.15(\mathrm{~m}, 2 \mathrm{H}), 3.87-3.81(\mathrm{~m}, 1 \mathrm{H})$, $2.90(\mathrm{~s}, 6 \mathrm{H}), 2.68,2.65\left(2 \mathrm{~s}, 5 \mathrm{H}\right.$, rotamers), 2.29-2.17(m, 2H). ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}, 75\right.$ $\mathrm{MHz}) \delta 166.09,161.69$ (2C), 157.48, 145.72, 141.37, 118.72, 111.00, 62.85, 62.71, 62.33, 36.02 (2C), 35.51 (2C), 24.62.

### 3.2.16. 4-(Benzyloxy)-3-(dimethylamino)butyl dimethylcarbamate (20)

Compound 20 was prepared as described for $\mathbf{1 9}$ from a mixture of $\mathrm{NaH}(44 \mathrm{mg}, 1.08$ $\mathrm{mmol}, 60 \%$ in mineral oil), 56 ( $200 \mathrm{mg}, 0.98 \mathrm{mmol}$ ), and benzyl bromide ( 0.23 mL , $1.96 \mathrm{mmol})$ yielding the product as a yellow oil $(94 \mathrm{mg}, 33 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300\right.$ $\mathrm{MHz}) \delta 7.38-7.27(\mathrm{~m}, 5 \mathrm{H}), 4.53(\mathrm{~s}, 2 \mathrm{H}), 4.14(\mathrm{t}, 2 \mathrm{H}, J=6.7 \mathrm{~Hz}), 3.62-3.47(\mathrm{~m}, 2 \mathrm{H})$, 2.92, 2.89 ( $2 \mathrm{~s}, 7 \mathrm{H}$, rotamers), 2.37 (s, 6H), 1.92-1.70 (m, 2H). 4-(Benzyloxy)-3(dimethylamino)butyl dimethylcarbamate oxalate. The free amine of 20 (94 mg, 0.32 mmol ) was converted into the oxalate salt as described under the general procedure affording white crystals ( $114 \mathrm{mg}, 93 \%$ ). Mp: 79.0-79.7 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $\mathrm{D}_{2} \mathrm{O}, 300 \mathrm{MHz}$ ) $\delta$ 7.40-7.34 (m, 5H), $4.58(\mathrm{~d}, 2 \mathrm{H}, J=0.7 \mathrm{~Hz}), 4.11-3.95(\mathrm{~m}, 2 \mathrm{H}), 3.77(\mathrm{dd}, 1 \mathrm{H}, J=12.0$, $3.7 \mathrm{~Hz}), 3.65(\mathrm{dd}, 1 \mathrm{H}, J=12.0,8.0 \mathrm{~Hz}), 3.53-3.41(\mathrm{~m}, 1 \mathrm{H}), 2.83-2.73(\mathrm{~m}, 12 \mathrm{H}), 2.18-$ $2.06(\mathrm{~m}, 1 \mathrm{H}), 1.92-1.85(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}, 75 \mathrm{MHz}\right) \delta 165.64,157.52(2 \mathrm{C})$, 136.70, 128.85 (2C), 128.69 (2C), 128.62, 73.14, 65.32, 63.14, 62.20, 41.22, 38.02, 36.08, 35.65, 24.02.

### 3.2.17. 3-(Dimethylamino)-4-(thiazol-2-yloxy)butyl dimethylcarbamate (21)

Compound 21 was prepared as described for $\mathbf{1 9}$ from a mixture of NaH ( $65 \mathrm{mg}, 1.62$ $\mathrm{mmol}, 60 \%$ in mineral oil), 56 ( $300 \mathrm{mg}, 1.47 \mathrm{mmol}$ ), and 2-bromothiazole ( 0.26 mL , 2.94 mmol ). Purification by reverse phase column chromatography (C18 silica)
$\left(\mathrm{H}_{2} \mathrm{O}: \mathrm{AcOH}: \mathrm{MeCN} / 99.9: 0.1: 0\right.$ to $\left.94.9: 0.1: 5\right)$ followed by basic extraction of the product into EtOAc afforded the product as a yellow oil ( $61 \mathrm{mg}, 14 \%$ ). ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 300 \mathrm{MHz}\right) \delta 7.17(\mathrm{~d}, 1 \mathrm{H}, J=3.8 \mathrm{~Hz}), 6.98(\mathrm{~d}, 1 \mathrm{H}, J=3.8 \mathrm{~Hz}), 4.88(\mathrm{dd}, 1 \mathrm{H}, J$ $=12.9,2.9 \mathrm{~Hz}), 4.77(\mathrm{dd}, 1 \mathrm{H}, J=13.0,6.5 \mathrm{~Hz}), 4.31-4.25(\mathrm{~m}, 2 \mathrm{H}), 3.97-3.86(\mathrm{~m}, 1 \mathrm{H})$, 3.00-2.84 (m, 6H), 2.30-2.19 (m, 2H). ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 75 \mathrm{MHz}\right) \delta$ 174.79, 157.80 , $137.69,114.11,68.36,63.84,62.93,41.40$ (2C), $36.87,36.35,26.36 .3-$ (Dimethylamino)-4-(thiazol-2-yloxy)butyl dimethylcarbamate oxalate. The free amine of $21(61 \mathrm{mg}, 0.21 \mathrm{mmol})$ was converted into the oxalate salt as described under the general procedure affording white crystals ( $53 \mathrm{mg}, 66 \%$ ). Mp: 122.4-122.7 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}, 300 \mathrm{MHz}\right) \delta 7.15(\mathrm{dd}, 1 \mathrm{H}, J=3.9,0.8 \mathrm{~Hz}), 6.97(\mathrm{dd}, 1 \mathrm{H}, J=3.9,0.9 \mathrm{~Hz})$, 4.81-4.76 (m, 1H), 4.67 (dd, $1 \mathrm{H}, J=12.6,6.1 \mathrm{~Hz}), 4.20$ (nonet, $2 \mathrm{H}, J=5.8 \mathrm{~Hz}$ ), 3.94$3.84(\mathrm{~m}, 1 \mathrm{H}), 2.94(\mathrm{~s}, 6 \mathrm{H}), 2.78(\mathrm{~s}, 6 \mathrm{H}), 2.30-2.20(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(\mathrm{D}_{2} \mathrm{O}, 75 \mathrm{MHz}\right) \delta$ 174.15 (2C), 165.72, 157.52, 136.11, 113.44, 67.29, 62.70, 62.25, 41.08, 39.81, 36.05, 35.57, 24.51.
3.2.18 3-(Dimethylamino)-5-phenylpentyl methylcarbamate (22).

Compound $\mathbf{2 2}$ was prepared as described for $\mathbf{1 3}$ from a mixture of $52(500 \mathrm{mg}, 2.41$ mmol ), CDI ( $469 \mathrm{mg}, 2.89 \mathrm{mmol}$ ), and methylamine ( $1.1 \mathrm{~mL}, 12.06 \mathrm{mmol}, 40 \% \mathrm{in}$ $\left.\mathrm{H}_{2} \mathrm{O}\right)$ affording the product as a yellow oil $(360 \mathrm{mg}, 57 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 400\right.$ $\mathrm{MHz}) \delta$ 7.28-7.23 (m, 2H), 7.21-7.12 (m, 3H), 4.14-4.04 (m, 2H), $2.69(\mathrm{~s}, 3 \mathrm{H}), 2.66-$ $2.60(\mathrm{~m}, 2 \mathrm{H}), 2.52(\mathrm{t}, 1 \mathrm{H}, J=6.3 \mathrm{~Hz}), 2.22(\mathrm{~s}, 6 \mathrm{H}), 1.92-1.79(\mathrm{~m}, 2 \mathrm{H}), 1.70-1.52(\mathrm{~m}$, $2 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 101 \mathrm{MHz}\right) \delta 143.70,129.58$ (2C), 129.52 (2C), 126.96, 64.19, 61.48, 40.74 (2C), 34.35, 30.27, 32.93, 27.55. 3-(Dimethylamino)-5-phenylpentyl methylcarbamate oxalate. The free amine of $\mathbf{2 2}(200 \mathrm{mg}, 0.76 \mathrm{mmol})$ was converted into the oxalate salt according to the general procedure affording a white solid ( 182 mg , $68 \%$ ). Mp: 136.0-136.7 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $\mathrm{D}_{2} \mathrm{O}, 400 \mathrm{MHz}$ ) $\delta 7.34-7.27(\mathrm{~m}, 2 \mathrm{H}), 7.25-7.18$
$(\mathrm{m}, 3 \mathrm{H}), 4.16-4.06(\mathrm{~m}, 1 \mathrm{H}), 4.02-3.92(\mathrm{~m}, 1 \mathrm{H}), 3.22-3.11(\mathrm{~m}, 1 \mathrm{H}), 2.70(\mathrm{~d}, 6 \mathrm{H}, J=11.3$ $\mathrm{Hz})$, $2.59(\mathrm{~s}, 3 \mathrm{H}), 2.10-1.94(\mathrm{~m}, 2 \mathrm{H}), 1.93-1.83(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}, 101 \mathrm{MHz}\right) \delta$. 165.48 (2C), 158.61, 140.04, 128.79 (2C), 128.45 (2C), 126.54, $62.40,61.14,39.49$, 38.70, 31.10, 27.83, 29.35, 26.53.

### 3.2.19. 3-(Dimethylamino)-6-phenylhexyl methylcarbamate (23)

Compound 23 was prepared as described for $\mathbf{1 3}$ from a mixture of $\mathbf{5 3}(1.48 \mathrm{~g}, 6.67$ $\mathrm{mmol}), \mathrm{CDI}(1.30 \mathrm{~g}, 8.00 \mathrm{mmol})$, and methylamine ( $5.80 \mathrm{~mL}, 68.7 \mathrm{mmol}, 40 \%$ in $\mathrm{H}_{2} \mathrm{O}$ ) affording the product as a yellow oil $(1.49 \mathrm{~g}, 80 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 7.29-$ $7.27(\mathrm{~m}, 1 \mathrm{H}), 7.25(\mathrm{~d}, 1 \mathrm{H}, J=1.8 \mathrm{~Hz}), 7.19-7.16(\mathrm{~m}, 3 \mathrm{H}), 4.60(\mathrm{~s}, 1 \mathrm{H}), 4.10(\mathrm{t}, 2 \mathrm{H}, J=$ $6.8 \mathrm{~Hz}), 2.79,2.78(2 \mathrm{~s}, 3 \mathrm{H}$, rotamers), $2.61(\mathrm{t}, 2 \mathrm{H}, J=7.6 \mathrm{~Hz}), 2.51(\mathrm{t}, 1 \mathrm{H}, J=6.0 \mathrm{~Hz})$, $2.23(\mathrm{~s}, 6 \mathrm{H}), 1.78(\mathrm{dq}, 1 \mathrm{H}, J=13.3,6.6 \mathrm{~Hz}), 1.69-1.64(\mathrm{~m}, 2 \mathrm{H}), 1.61-1.53(\mathrm{~m}, 2 \mathrm{H})$, 1.32-1.27 (m, 1H). ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right) \delta 157.39,142.52,128.54(2 \mathrm{C}), 128.42$ (2C), 125.84, 63.44, 61.02, 40.44 (3C), 36.17, 29.15, 28.90, 27.64.3-(Dimethylamino)-6-phenylhexyl methylcarbamate oxalate. The free amine of $\mathbf{2 3}$ ( $362 \mathrm{mg}, 1.30 \mathrm{mmol}$ ) was converted into the oxalate salt as described under the general procedure affording white crystals ( $291 \mathrm{mg}, 61 \%$ ). Mp: $91.8{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{D}_{2} \mathrm{O}, 400 \mathrm{MHz}\right) \delta 7.34-7.30(\mathrm{~m}$, $2 H), ~ 7.25-7.20(\mathrm{~m}, 3 \mathrm{H}), 4.14-4.07(\mathrm{~m}, 1 \mathrm{H}), 4.04-3.98(\mathrm{~m}, 1 \mathrm{H}), 3.26-3.23(\mathrm{~m}, 1 \mathrm{H}), 2.78$, $2.76(2 \mathrm{~s}, 6 \mathrm{H}$, rotamers), $2.65(\mathrm{t}, 2 \mathrm{H}, J=3.1 \mathrm{~Hz}), 2.63(\mathrm{~s}, 3 \mathrm{H}), 2.06-1.98(\mathrm{~m}, 1 \mathrm{H}), 1.92-$ $1.84(\mathrm{~m}, 1 \mathrm{H}), 1.68-1.58(\mathrm{~m}, 4 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}, 101 \mathrm{MHz}\right) \delta 165.68(2 \mathrm{C}), 158.70$, $141.78,128.69$ (2C), 128.57 (2C), 126.22, 63.57, 61.66, 39.34, 39.07, 34.37, 30.23, 27.68, 27.57, 26.71.
3.2.20. 3-(Dimethylamino)-7-phenylheptyl methylcarbamate (24)

Compound $\mathbf{2 4}$ was prepared as described for $\mathbf{1 3}$ from a mixture of $\mathbf{5 4}(970 \mathrm{mg}, 4.12$ mmol ), CDI ( $800 \mathrm{mg}, 4.95 \mathrm{mmol}$ ), and methylamine ( $4.0 \mathrm{~mL}, 41.2 \mathrm{mmol}, 40 \%$ in $\mathrm{H}_{2} \mathrm{O}$ )
yielding the product as a yellow oil $(890 \mathrm{mg}, 73 \%)$. H NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta$ 7.29-7.28 (m, 2H), 7.19-7.17 (m, 3H), 4.60 (s, 1H), 4.13-4.09 (m, 2H), 2.83-2.77 (m, $3 \mathrm{H}), 2.61(\mathrm{t}, 2 \mathrm{H}, J=7.8 \mathrm{~Hz}), 2.48-2.43(\mathrm{~m}, 1 \mathrm{H}), 2.24(\mathrm{~s}, 6 \mathrm{H}), 1.64-1.60(\mathrm{~m}, 2 \mathrm{H}), 1.56-$ $1.49(\mathrm{~m}, 2 \mathrm{H}), 1.40-1.33(\mathrm{~m}, 2 \mathrm{H}), 1.30-1.23(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right) \delta$ $157.33,142.64,128.51$ (2C), 128.43 (2C), 125.83, 63.17, 61.46, 40.42 (2C), 35.99, 31.73, 29.17, 27.64, 26.74, 25.67. 3-(Dimethylamino)-7-phenylheptyl methylcarbamate oxalate. The free amine of $\mathbf{2 4}(409 \mathrm{mg}, 1.40 \mathrm{mmol})$ was converted into the oxalate salt as described under the general procedure affording white crystals (224 mg, 55\%). Mp: 87.1 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $\mathrm{D}_{2} \mathrm{O}, 400 \mathrm{MHz}$ ) $\delta 7.42-7.38(\mathrm{~m}, 2 \mathrm{H}), 7.33-7.27$ $(\mathrm{m}, 3 \mathrm{H}), 4.22(\mathrm{dt}, 1 \mathrm{H}, J=11.4,5.7 \mathrm{~Hz}), 4.12(\mathrm{ddd}, 1 \mathrm{H}, J=11.4,7.4,4.5 \mathrm{~Hz}), 3.32(\mathrm{p}$, $1 \mathrm{H}, J=5.4 \mathrm{~Hz}), 2.83(\mathrm{~s}, 6 \mathrm{H}), 2.72-2.68(\mathrm{~m}, 6 \mathrm{H}), 2.14-2.07(\mathrm{~m}, 1 \mathrm{H}), 2.00-1.93(\mathrm{~m}, 1 \mathrm{H})$, 1.82-1.66 (m, 4H), $1.42(\mathrm{dt}, 2 \mathrm{H}, J=15.0,7.5 \mathrm{~Hz}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(\mathrm{D}_{2} \mathrm{O}, 101 \mathrm{MHz}\right) \delta 165.51$ (2C), 158.71, 142.69, 128.61 (2C), 128.56 (2C), 125.97, 63.65, 61.64, 39.33, 39.10, $34.51,30.09,28.06,27.75,26.63,24.54$.
3.2.21. 3-(Dimethylamino)-8-phenyloctyl methylcarbamate (25)

Compound $\mathbf{2 5}$ was prepared as described for $\mathbf{1 3}$ from a mixture of $\mathbf{5 5}(470 \mathrm{~g}, 1.89$ $\mathrm{mmol})$, CDI ( $370 \mathrm{~g}, 2.26 \mathrm{mmol}$ ), and methylamine ( 0.80 mL , 9.42 mmol , $40 \%$ in $\mathrm{H}_{2} \mathrm{O}$ ) furnishing the product as a yellow oil ( $425 \mathrm{mg}, 73 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 400 \mathrm{MHz}\right) \delta$ 7.27-7.20 (m, 2H), 7.18-7.10(m, 3H), 4.11-4.03(m, 2H), 2.68 (s, 3H), 2.61 (t, $2 \mathrm{H}, \mathrm{J}=$ $7.5 \mathrm{~Hz}), 2.21(\mathrm{~s}, 6 \mathrm{H}), 1.80(\mathrm{sxt}, 1 \mathrm{H}, J=6.5 \mathrm{~Hz}), 1.68-1.49(\mathrm{~m}, 4 \mathrm{H}), 1.26(\mathrm{bs}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{CD}_{3} \mathrm{OD}, 101 \mathrm{MHz}\right) \delta 129.56$ (2C), 129.41 (2C), 129.37, 126.79, 64.27, 62.22, 40.84 (2C), 37.00, 32.79, 30.79, 30.57, 30.53, 28.13, 27.54. 3-(Dimethylamino)-8phenyloctyl methylcarbamate oxalate. The free amine of $\mathbf{2 5}$ ( $212 \mathrm{mg}, 0.69 \mathrm{mmol}$ ) was converted into the oxalate salt according to the general procedure yielding a white solid ( $213 \mathrm{mg}, 78 \%$ ). Mp: 98.2-98.7 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $\mathrm{D}_{2} \mathrm{O}, 400 \mathrm{MHz}$ ) $\delta 7.34-7.27(\mathrm{~m}, 2 \mathrm{H})$,
7.25-7.17 (m, 3H), 4.19-3.99 (m, 2H), 3.20 (bs, 1 H$), 2.73(\mathrm{~s}, 6 \mathrm{H}), 2.64(\mathrm{~s}, 3 \mathrm{H}), 2.57(\mathrm{t}$, $2 \mathrm{H}, J=7.4 \mathrm{~Hz}), 2.09-1.95(\mathrm{~m}, 1 \mathrm{H}), 1.92-1.81(\mathrm{~m}, 1 \mathrm{H}), 1.70-1.60(\mathrm{~m}, 1 \mathrm{H}), 1.60-1.54$ $(\mathrm{m}, 3 \mathrm{H}), 1.39-1.24(\mathrm{~m}, 4 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}, 101 \mathrm{MHz}\right) \delta 165.64(2 \mathrm{C}), 158.71,143.17$, 128.59 (2C), 128.58 (2C), 125.88, 63.61, 61.71, 39.36, 39.07, 34.85, 30.24, 28.16, 27.80, 27.75, 26.66, 24.89.

### 3.2.22. 3-(Dimethylamino)-6-hydroxy-6-phenylhexyl methylcarbamate (26)

Compound 26 was prepared as described for $\mathbf{1 6}$ from a mixture of $\mathbf{6 0}(1.08 \mathrm{~g}, 2.85$ $\mathrm{mmol})$ and $p$-toluenesulfonic acid monohydrate $(1.08 \mathrm{~g}, 5.70 \mathrm{mmol}$ yielding the product as a yellow oil $(0.50 \mathrm{~g}, 60 \%) .{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta$ 7.39-7.19 $(\mathrm{m}, 10 \mathrm{H}$, diastereomers); $4.80(\mathrm{dd}, 1 \mathrm{H}, J=6.1,3.6 \mathrm{~Hz}), 4.66-4.58(\mathrm{~m}, 1 \mathrm{H})$ (diastereomers); 4.154.10, 4.05-4.00 ( $2 \mathrm{~m}, 2 \mathrm{H}$, diastereomers); 2.80, 2.79 ( $2 \mathrm{~s}, 6 \mathrm{H}$, diastereomers); 2.57-2.52 ( $\mathrm{m}, 2 \mathrm{H}$, diastereomers); 2.30, 2.23 ( $2 \mathrm{~s}, 6 \mathrm{H}$, diastereomers); 2.15-1.85 (m, 6 H , diastereomers); 1.73-1.63 ( $\mathrm{m}, 2 \mathrm{H}$, diastereomers); 1.59-1.51 (m, 1H, diastereomers); 1.45-1.29 (m, 3H, diastereomers). ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right) \delta$ 145.92, 145.64 (diastereomers); 128.08, 128.06 (2C, diastereomers); 126.69, 126.45 (diastereomers); 125.65, 125.57 (2C, diastereomers); 74.64, 72.06 (diastereomers); 62.90; 61.98, 61.16 (diastereomers); 40.10, 36.96 (diastereomers); 39.93, 39.50 (diastereomers); 30.26, 27.04 (diastereomers); 27.49; 26.58, 26.41 (diastereomers). 3-(Dimethylamino)-6-hydroxy-6-phenylhexyl methylcarbamate oxalate. The free amine of 26 ( 100 mg , 0.34 mmol ) was converted into the oxalate salt as described under the general procedure affording white crystals ( $86 \mathrm{mg}, 66 \%$ ). Mp: $90.3{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $\mathrm{D}_{2} \mathrm{O}, 400 \mathrm{MHz}$ ) $\delta 7.50-$ $7.34(\mathrm{~m}, 5 \mathrm{H}), 4.22-4.11(\mathrm{~m}, 1 \mathrm{H}), 4.10-4.00(\mathrm{~m}, 1 \mathrm{H}), 3.36-3.25(\mathrm{~m}, 1 \mathrm{H}), 2.78,2.75(2 \mathrm{~s}$, 6 H , rotamers), $2.68(\mathrm{~s}, 3 \mathrm{H}), 2.18-2.01(\mathrm{~m}, 1 \mathrm{H}), 2.01-1.80(\mathrm{~m}, 4 \mathrm{H}), 1.79-1.47(\mathrm{~m}, 2 \mathrm{H})$. ${ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}, 101 \mathrm{MHz}\right) \delta$ 158.71, 158.68 (diastereomers); 142.96, 142.88 (diastereomers); 128.83 (2C); 128.11, 128.08 (diastereomers); 126.12, 126.07 (2C,
diastereomers); 73.29, 73.05 (diastereomers); $63.69,63.45$ (diastereomers); 61.63, 61.62 (diastereomers); 39.35, 39.20 (diastereomers); 33.83, 33.61 (2C, diastereomers); 27.73, 27.64 (diastereomers); 26.75, 26.62 (diastereomers); 24.74, 24.56 (diastereomers).
3.2.23. N,N-Dimethyl-1-((1-methyl-1H-imidazol-2-yl)oxy)-7-phenylheptan-3-amine (27) To a stirred suspension of $\mathrm{NaH}(68 \mathrm{mg}, 1.71 \mathrm{mmol}, 60 \%$ in mineral oil) in dry THF ( 5 mL ) at $0^{\circ} \mathrm{C}$ was added a dry THF ( 6 mL ) solution of $\mathbf{5 4}(268 \mathrm{mg}, 1.14 \mathrm{mmol})$. After 30 $\min$ of stirring at $0^{\circ} \mathrm{C}, 2,4,5$-tribromo-1-methyl-1 H -imidazole[45] ( $399 \mathrm{mg}, 1.10 \mathrm{mmol}$ ) was added. The reaction mixture was refluxed overnight, then cooled, and poured into ice water ( $\sim 10 \mathrm{~mL}$ ). The resulting mixture was extracted with EtOAc $(2 \times 10 \mathrm{~mL})$. The combined organic phases were extracted with aq HCl solution $(1 \mathrm{M}, 2 \times 6 \mathrm{~mL})$. The combined water phases were basified to pH 14 with a concentrated aq NaOH solution and extracted with EtOAc $(3 \times 10 \mathrm{~mL})$. The combined organic phases were dried $\left(\mathrm{MgSO}_{4}\right)$, filtered, and evaporated in vacuo. Purification by DCVC (DCM:MeOH: $\mathrm{NH}_{3} / 100: 0: 0$ to $\left.100: 13: 1\right)$ gave 1-((4,5-dibromo-1-methyl-1 H -imidazol-2-yl)oxy- $N, N$-dimethyl-7-phenylheptan-3-amine as a yellow oil (155 mg, 29\%). ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 400 \mathrm{MHz}\right) \delta 7.26-7.20(\mathrm{~m}, 3 \mathrm{H}), 7.18-7.10(\mathrm{~m}, 3 \mathrm{H}), 4.41-4.29(\mathrm{~m}, 2 \mathrm{H})$, $3.36(\mathrm{~s}, 3 \mathrm{H}), 2.62(\mathrm{t}, 3 \mathrm{H}, J=7.5 \mathrm{~Hz}), 2.33(\mathrm{~s}, 6 \mathrm{H}), 2.00(\mathrm{dq}, 1 \mathrm{H}, J=14.4,7.8 \mathrm{~Hz}), 1.81$ $(\mathrm{dq}, 1 \mathrm{H}, J=7.8,6.8 \mathrm{~Hz}), 1.69-1.61(\mathrm{~m}, 4 \mathrm{H}), 1.39-1.37(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}\right.$, $101 \mathrm{MHz}) \delta 129.55$ (2C), 129.44 (2C), $126.85,110.88,69.83,62.27,40.79$ (2C), 36.88, 32.82, 31.03, 30.32, 27.78. A stirred solution of 1-((4,5-dibromo-1-methyl-1 H -imidazol-2-yl)oxy- $N, N$-dimethyl-7-phenylheptan-3-amine ( $178 \mathrm{mg}, 0.38 \mathrm{mmol}$ ) in dry THF ( 2 mL ) was cooled to $-78{ }^{\circ} \mathrm{C}$ and dropwisely added $n$ - $\operatorname{BuLi}(0.95 \mathrm{ml}, 2.27 \mathrm{mmol}$, 2.4 M in hexane). After 15 min of stirring at $-78^{\circ} \mathrm{C}$, the reaction mixture was added sat. aq $\mathrm{NaHCO}_{3}$ solution $(3.5 \mathrm{~mL})$ and then extracted with $\mathrm{DCM}(3 \times 7 \mathrm{~mL})$. The combined organic phases were dried $\left(\mathrm{MgSO}_{4}\right)$, filtered, and evaporated in vacuo. Purification by

DCVC (DCM:MeOH: $\mathrm{NH}_{3} / 100: 0: 0$ to $100: 2.8: 0.3$ ) gave 27 as a yellow oil ( 52 mg , $44 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 400 \mathrm{MHz}\right) \delta 7.26-7.20(\mathrm{~m}, 2 \mathrm{H}), 7.17-7.10(\mathrm{~m}, 3 \mathrm{H}), 6.64(\mathrm{~d}$, $1 \mathrm{H}, J=1.8 \mathrm{~Hz}), 6.52(\mathrm{~d}, 1 \mathrm{H}, J=1.8 \mathrm{~Hz}), 4.38-4.26(\mathrm{~m}, 2 \mathrm{H}), 3.37(\mathrm{~s}, 3 \mathrm{H}), 2.64-2.58(\mathrm{~m}$, $3 \mathrm{H}), 2.24(\mathrm{~s}, 6 \mathrm{H}), 1.95(\mathrm{~d}, 1 \mathrm{H}, J=6.3 \mathrm{~Hz}), 1.75(\mathrm{~d}, 1 \mathrm{H}, J=6.5 \mathrm{~Hz}), 1.63(\mathrm{~d}, 3 \mathrm{H}, J=7.5$ $\mathrm{Hz}), 1.42-1.33(\mathrm{~m}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 101 \mathrm{MHz}\right) \delta 129.55(2 \mathrm{C}), 129.43(2 \mathrm{C})$, 126.82, 122.83, 117.76, 69.45, 62.12, 40.82, 36.91, 32.91, $30.9130 .63,30.40,27.86$. The product was not converted into the oxalate salt but tested pharmacologically as the oil.
3.2.24. 1-(6-Bromopyridin-3-yl)-4-(3-phenylpropyl)-1,4-diazepane (28)

A solution of $58(598 \mathrm{mg}, 1.68 \mathrm{mmol})$ and TFA $(2.58 \mathrm{~mL}, 33.7 \mathrm{mmol})$ in DCM ( 20 mL ) was stirred at rt for 16 h . The reaction mixture was evaporated in vacuo. The residue was added aq NaOH solution $(4 \mathrm{M}, 20 \mathrm{~mL})$ and extracted with $\mathrm{DCM}(3 \times 100 \mathrm{~mL})$. The combined organic phases were dried $\left(\mathrm{MgSO}_{4}\right)$, filtered, and evaporated in vacuo giving 1-(6-bromopyridin-3-yl)-1,4-diazepane as a yellow oil (274 mg, 64\%) that crystallizes over time. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 7.84(\mathrm{~d}, 1 \mathrm{H}, J=3.3 \mathrm{~Hz}), 7.23(\mathrm{dd}, 1 \mathrm{H}, J=8.8$, $0.5 \mathrm{~Hz}), 6.86(\mathrm{dd}, 1 \mathrm{H}, J=8.9,3.4 \mathrm{~Hz}), 3.54(\mathrm{dt}, 4 \mathrm{H}, J=16.7,5.8 \mathrm{~Hz}), 3.08-3.00(\mathrm{~m}$, $2 \mathrm{H}), 2.89-2.80(\mathrm{~m}, 2 \mathrm{H}), 2.28(\mathrm{bs}, 1 \mathrm{H}), 1.97-1.85(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right)$ $\delta 143.90,133.91,127.59,126.76,121.08,51.71,47.95,47.91,47.89,28.92$. То а solution 1-(6-bromopyridin-3-yl)-1,4-diazepane (274 mg, 1.07 mmol$)$ and phenylpropanal ( $157 \mu \mathrm{~L}, 1.07 \mathrm{mmol}, 90 \%$ ) in dry DCE ( 10 mL ) was added freshly grinded $\mathrm{NaBH}(\mathrm{OAc})_{3}(340 \mathrm{mg}, 1.61 \mathrm{mmol})$. The reaction mixture was stirred overnight at rt. Sat. aq $\mathrm{NaHCO}_{3}$ solution ( 10 mL ) was added to the reaction mixture that was extracted with DCM ( $2 \times 5 \mathrm{~mL}$ ). The combined organic phases were washed with sat. aq $\mathrm{NaHCO}_{3}$ solution $(10 \mathrm{~mL})$, dried $\left(\mathrm{MgSO}_{4}\right)$, filtered, and evaporated in vacuo. Purification by DCVC (DCM:MeOH: $\mathrm{NH}_{3} / 100: 0: 0$ to $97.5: 2.25: 0.25$ ) afforded the
product as a yellow oil $(362 \mathrm{mg}, 90 \%)$. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 7.87(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=$ $3.3 \mathrm{~Hz}), 7.35-7.29(\mathrm{~m}, 3 \mathrm{H}), 7.26-7.18(\mathrm{~m}, 4 \mathrm{H}), 6.88(\mathrm{dd}, 1 \mathrm{H}, J=8.8,3.3 \mathrm{~Hz}), 3.55(\mathrm{t}$, $2 \mathrm{H}, J=5.0 \mathrm{~Hz}), 3.51(\mathrm{t}, 2 \mathrm{H}, J=6.2 \mathrm{~Hz}), 2.84-2.77(\mathrm{~m}, 2 \mathrm{H}), 2.70-2.61(\mathrm{~m}, 4 \mathrm{H}), 2.55(\mathrm{t}$, $2 \mathrm{H}, J=7.3 \mathrm{~Hz}), 2.03-1.95(\mathrm{~m}, 2 \mathrm{H}), 1.84(\mathrm{p}, 2 \mathrm{H}, J=7.5 \mathrm{~Hz}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101\right.$ $\mathrm{MHz}) \delta 144.39,142.02,133.91,128.35$ (2C), 128.29 (2C), 127.43, 126.68, 125.76, $121.05,57.05,54.62,54.38,48.90,47.94,33.43,29.08,27.24$. 1-(6-Bromopyridin-3-yl)-4-(3-phenylpropyl)-1,4-diazepane oxalate. The free amine of $\mathbf{2 8}$ ( $283 \mathrm{mg}, 0.71$ mmol) was converted into the oxalate salt as described under the general procedure with the exception that 2.0 eq of oxalic acid was added. Salt formation afforded white crystals ( $293 \mathrm{mg}, 81 \%$ ). Mp: $95.3{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $\mathrm{D}_{2} \mathrm{O}, 400 \mathrm{MHz}$ ) $\delta 7.81(\mathrm{~d}, 1 \mathrm{H}, J=3.3$ $\mathrm{Hz}), 7.47(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=9.0 \mathrm{~Hz}), 7.36-7.29(\mathrm{~m}, 2 \mathrm{H}), 7.28-7.17(\mathrm{~m}, 4 \mathrm{H}), 3.80-3.59(\mathrm{~m}, 3 \mathrm{H})$, 3.58-3.40 (m, 3H), 3.37-3.18 (m, 2H), 3.17-3.05 (m, 2H), $2.70(\mathrm{t}, 2 \mathrm{H}, J=7.2 \mathrm{~Hz}), 2.23-$ 2.11 (m, 2H), 2.10-1.99 (m, 2H). ${ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}, 101 \mathrm{MHz}\right) \delta 165.13$ (2C), 144.54, $141.60,140.28,132.31,132.27,128.78$ (2C), 128.45 (2C), 126.58, 124.32, 55.70, $54.01,53.58,46.36,43.48,31.72,25.26,23.62$.

### 3.2.25. 1-Cinnamyl-4-(pyridin-3-yl)-1,4-diazepane (29)

Compound 29 was prepared as described for 28 from a mixture of 1-(pyridin-3-yl)-1,4diazepane ( $100 \mathrm{mg}, 0.56 \mathrm{mmol}$ ), cinnamaldehyde ( $0.076 \mathrm{~mL}, 0.56 \mathrm{mmol}, 93 \%$ ), and $\mathrm{NaBH}(\mathrm{OAc})_{3}(178 \mathrm{mg}, 0.84 \mathrm{mmol})$. Purification by DCVC (DCM:MeOH: $\mathrm{NH}_{3} / 100: 0: 0$ to 100:2.3:0.3) afforded the product as a yellow oil ( $90 \mathrm{mg}, 55 \%$ ). ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right.$, $400 \mathrm{MHz}) \delta 8.14(\mathrm{~d}, 1 \mathrm{H}, J=3.0 \mathrm{~Hz}), 7.94(\mathrm{dd}, 1 \mathrm{H}, J=4.8,1.3 \mathrm{~Hz}), 7.41-7.36(\mathrm{~m}, 2 \mathrm{H})$, 7.35-7.30 (m, 2H), 7.27-7.22 (m, 1H), $7.11(\mathrm{ddd}, 1 \mathrm{H}, J=4.0 \mathrm{~Hz}), 6.95(\mathrm{ddd}, 1 \mathrm{H}, J=$ $8.6,3.2,1.3 \mathrm{~Hz}), 6.52(\mathrm{~d}, 1 \mathrm{H}, J=16.1 \mathrm{~Hz}), 6.30(\mathrm{dt}, 1 \mathrm{H}, J=9.0,6.5 \mathrm{~Hz}), 3.61(\mathrm{t}, 2 \mathrm{H}, J$ $=4.8 \mathrm{~Hz}), 3.54(\mathrm{t}, 2 \mathrm{H}, J=6.3 \mathrm{~Hz}), 3.33(\mathrm{~d}, 2 \mathrm{H}, J=6.3 \mathrm{~Hz}) 2.91-2.82(\mathrm{~m}, 2 \mathrm{H}), 2.75-$ $2.67(\mathrm{~m}, 2 \mathrm{H}), 2.09-2.00(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right) \delta 144.90,137.30$,
$136.79,134.41,133.02,128.58$ (2C), $127.58,126.33,123.54$ (2C), 120.59, 117.73, 60.72, 55.14, 54.66, 48.51, 47.67, 27.52. 1-Cinnamyl-4-(pyridin-3-yl)-1,4-diazepane oxalate. The free amine of $29(78 \mathrm{mg}, 0.27 \mathrm{mmol})$ was converted into the oxalate salt according to the general procedure with the exception that 2.0 eq of oxalic acid was added. Salt formation yielded a yellow solid ( $58 \mathrm{mg}, 56 \%$ ). Mp: 63.0-63.9 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}, 400 \mathrm{MHz}\right) \delta 8.15(\mathrm{~d}, 1 \mathrm{H}, J=2.5 \mathrm{~Hz}), 8.03(\mathrm{~d}, 1 \mathrm{H}, J=5.5 \mathrm{~Hz}), 7.89(\mathrm{dd}, 1 \mathrm{H}, J=$ $6.0,3.0 \mathrm{~Hz}), 7.83-7.77(\mathrm{~m}, 1 \mathrm{H}), 7.54(\mathrm{~d}, 2 \mathrm{H}, J=7.5 \mathrm{~Hz}), 7.47-7.37(\mathrm{~m}, 3 \mathrm{H}), 6.93(\mathrm{~d}$, $1 \mathrm{H}, J=15.8 \mathrm{~Hz}), 6.33(\mathrm{p}, 1 \mathrm{H}, J=7.5 \mathrm{~Hz}), 4.01(\mathrm{~d}, 2 \mathrm{H}, J=7.3 \mathrm{~Hz}), 3.97-3.89(\mathrm{~m}, 2 \mathrm{H})$, 3.85-3.70 (m, 2H), 3.70-3.50(m, 2H), 3.45-3.20 (m, 2H), 2.45-2.25 (m, 2H). ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}, 101 \mathrm{MHz}\right) \delta 165.13$ (2C), 147.02, 140.87, 135.16, 129.27, 129.02 (2C), 128.51, $127.60,127.16,126.97$ (2C), 124.90, 116.31, 59.44, 53.58, 53.46, 46.69, 43.39, 23.76.
3.2.26. 1-Benzyl-4-(pyridin-3-yl)-1,4-diazepane (30)
$\mathbf{5 8}$ was deprotected as described for $\mathbf{2 8}$ from a mixture of $\mathbf{5 8}$ ( $948 \mathrm{mg}, 3.42 \mathrm{mmol}$ ) and TFA ( $5.23 \mathrm{~mL}, 68.4 \mathrm{mmol}$ ) yielding 1-(pyridin-3-yl)-1,4-diazepane as a yellow oil (507 $\mathrm{mg}, 84 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 8.13(\mathrm{~d}, 1 \mathrm{H}, J=3.1 \mathrm{~Hz}), 7.92(\mathrm{dd}, 1 \mathrm{H}, J=$ $4.5,1.3 \mathrm{~Hz}), 7.10(\mathrm{dd}, 1 \mathrm{H}, J=8.5,4.6 \mathrm{~Hz}), 6.95(\mathrm{ddd}, 1 \mathrm{H}, J=8.6,3.1,1.3 \mathrm{~Hz}), 3.62-$ $3.54(\mathrm{~m}, 4 \mathrm{H}), 3.05(\mathrm{t}, 2 \mathrm{H}, J=5.3 \mathrm{~Hz}), 2.85(\mathrm{t}, 2 \mathrm{H}, J=5.8 \mathrm{~Hz}), 2.07(\mathrm{~s}, 1 \mathrm{H}), 1.93(\mathrm{p}$, $2 \mathrm{H}, J=6.0 \mathrm{~Hz}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right) \delta 144.40,137.49,134.54,123.84$, 117.91, $51.45,48.19,47.93,47.89,29.24$. Compound $\mathbf{3 0}$ was prepared as described for 28 from a mixture of 1-(pyridin-3-yl)-1,4-diazepane (100 mg, 0.56 mmol ), benzaldehyde ( $57 \mu \mathrm{~L}, 0.56 \mathrm{mmol}$ ), and $\mathrm{NaBH}(\mathrm{OAc})_{3}(178 \mathrm{mg}, 0.84 \mathrm{mmol})$ affording the product as a yellow oil $(104 \mathrm{mg}, 69 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 8.12(\mathrm{~d}, 1 \mathrm{H}, J$ $=3.0 \mathrm{~Hz}), 7.93(\mathrm{dd}, 1 \mathrm{H}, J=4.6,1.3 \mathrm{~Hz}), 7.34-7.28(\mathrm{~m}, 4 \mathrm{H}), 7.28-7.23(\mathrm{~m}, 2 \mathrm{H}), 7.10$ (ddd, $1 \mathrm{H}, J=8.5,4.6,0.6 \mathrm{~Hz}), 6.94(\mathrm{ddd}, 1 \mathrm{H}, J=8.6,3.1,1.3 \mathrm{~Hz}), 3.65(\mathrm{~s}, 2 \mathrm{H}), 3.54$ (dt, 4H, $J=13.3,5.6 \mathrm{~Hz}), 2.78(\mathrm{t}, 2 \mathrm{H}, J=5.0 \mathrm{~Hz}), 2.65(\mathrm{t}, 2 \mathrm{H}, J=5.6 \mathrm{~Hz}), 1.98(\mathrm{p}, 2 \mathrm{H}$,
$J=5.9 \mathrm{~Hz}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right) \hat{\delta} 145.07,138.94,137.23,134.47,128.97$ (2C), 128.44 (2C), 127.27, 123.72, 117.93, 62.48, 55.07, 54.75, 48.74, 47.99, 27.62. 1-Benzyl-4-(pyridin-3-yl)-1,4-diazepane oxalate. The free amine of $\mathbf{3 0}$ ( $104 \mathrm{mg}, 0.39$ mmol) was converted into the oxalate salt as described under the general procedure affording an orange solid (113 mg, 81\%). Mp: 124.0-124.8 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $\mathrm{D}_{2} \mathrm{O}, 400$ $\mathrm{MHz}) \delta 8.16(\mathrm{~d}, 1 \mathrm{H}, J=3.0 \mathrm{~Hz}), 8.06(\mathrm{~d}, 1 \mathrm{H}, J=5.3 \mathrm{~Hz}), 7.89(\mathrm{ddd}, 1 \mathrm{H}, J=9.1,3.0$, $1.1 \mathrm{~Hz}), 7.82(\mathrm{dd}, 1 \mathrm{H}, J=9.0,5.3 \mathrm{~Hz}), 7.60-7.53(\mathrm{~m}, 5 \mathrm{H}), 4.44(\mathrm{~s}, 2 \mathrm{H}), 3.92(\mathrm{t}, 2 \mathrm{H}, J=$ $4.6 \mathrm{~Hz}), 3.78(\mathrm{~s}, 4 \mathrm{H}), 3.62(\mathrm{t}, 2 \mathrm{H}, J=6.1 \mathrm{~Hz}), 3.59-3.44(\mathrm{~m}, 3 \mathrm{H}), 2.40-2.34(\mathrm{~m}, 2 \mathrm{H})$. ${ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}, 101 \mathrm{MHz}\right) \delta 170.15(2 \mathrm{C}), 147.06,131.23(2 \mathrm{C}), 130.32,129.34$ (2C), $128.79,128.58,127.42,127.07,124.30,61.07,53.93,53.71,46.61,42.97,23.45$.

### 3.2.27. 1-Phenethyl-4-(pyridin-3-yl)-1,4-diazepane (31)

Compound $\mathbf{3 1}$ was prepared as described for $\mathbf{2 8}$ from a mixture of 1-(pyridin-3-yl)-1,4diazepane ( $100 \mathrm{mg}, 0.56 \mathrm{mmol}$ ), 2-phenylacetaldehyde ( $65 \mu \mathrm{~L}, 0.56 \mathrm{mmol}$ ), and $\mathrm{NaBH}(\mathrm{OAc})_{3}(178 \mathrm{mg}, 0.84 \mathrm{mmol})$. Purification by DCVC (DCM:MeOH: $\mathrm{NH}_{3} / 100: 0: 0$ to 95:4.5:0.5) afforded the product as a brown oil ( $124 \mathrm{mg}, 79 \%$ ). ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400\right.$ $\mathrm{MHz}) \delta 8.20(\mathrm{~d}, 1 \mathrm{H}, J=3.0 \mathrm{~Hz}), 7.99(\mathrm{dd}, 1 \mathrm{H}, J=4.6,1.3 \mathrm{~Hz}), 7.37-7.32(\mathrm{~m}, 2 \mathrm{H})$, 7.30-7.23 (m, 3H), 7.16 (ddd, $1 \mathrm{H}, J=8.5,4.6,0.6 \mathrm{~Hz}), 7.01$ (ddd, $1 \mathrm{H}, J=8.6,3.1,1.3$ $\mathrm{Hz}), 3.63(\mathrm{t}, 2 \mathrm{H}, J=5.0 \mathrm{~Hz}), 3.57(\mathrm{t}, 2 \mathrm{H}, J=6.3 \mathrm{~Hz}), 2.94-2.91(\mathrm{~m}, 2 \mathrm{H}), 2.88-2.80(\mathrm{~m}$, $4 \mathrm{H}), 2.77-2.75(\mathrm{~m}, 2 \mathrm{H}), 2.05(\mathrm{p}, 2 \mathrm{H}, J=5.9 \mathrm{~Hz}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right) \delta$ $144.86,140.32,137.19,134.40,128.71$ (2C), 128.40 (2C), 126.04, 123.60, 117.75, 59.88, 54.95, 54.49, 48.76, 47.73, 34.09, 27.52. 1-Phenethyl-4-(pyridin-3-yl)-1,4diazepane oxalate. The free amine of $\mathbf{3 1}(124 \mathrm{mg}, 0.44 \mathrm{mmol})$ was converted into the oxalate salt as described under the general procedure affording an orange solid ( 98 mg , $60 \%$ ). Mp: 173.1-173.9 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{D}_{2} \mathrm{O}, 400 \mathrm{MHz}\right) \delta 8.18(\mathrm{~d}, 1 \mathrm{H}, J=3.0 \mathrm{~Hz}), 8.08-$ $8.07(\mathrm{~m}, 1 \mathrm{H}), 7.92(\mathrm{ddd}, 1 \mathrm{H}, J=9.1,3.0,1.0 \mathrm{~Hz}), 7.83(\mathrm{dd}, 1 \mathrm{H}, J=9.0,5.4 \mathrm{~Hz}), 7.46-$
$7.41(\mathrm{~m}, 2 \mathrm{H}), 7.40-7.35(\mathrm{~m}, 3 \mathrm{H}), 3.97(\mathrm{t}, 2 \mathrm{H}, J=4.7 \mathrm{~Hz}), 3.86-3.59(\mathrm{~m}, 4 \mathrm{H}), 3.58-3.32$ $(\mathrm{m}, 4 \mathrm{H})$, 3.18-3.12 (m, 2H), 2.43-2.33 (m, 2H). ${ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}, 101 \mathrm{MHz}\right): \delta 168.31$ (2C), 146.99, 136.16, 129.13 (2C), 128.81 (2C), 128.57, 127.54, 127.44, 127.15, $124.27,58.02,54.10,54.07,46.60,43.24,30.05,23.55$.
3.2.28. 1-(3-Phenylpropyl)-4-(pyridin-3-yl)-1,4-diazepane (32)

Compound $\mathbf{3 2}$ was prepared as described for $\mathbf{2 8}$ from a mixture of 1-(pyridin-3-yl)-1,4diazepane ( $100 \mathrm{mg}, 0.56 \mathrm{mmol}$ ), 3-phenylpropanal ( $82 \mu \mathrm{~L}, 0.56 \mathrm{mmol}, 90 \%$ ), and $\mathrm{NaBH}(\mathrm{OAc})_{3}(178 \mathrm{mg}, 0.84 \mathrm{mmol})$ yielding the product as an orange oil $(121 \mathrm{mg}$, $73 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 8.12(\mathrm{~d}, 1 \mathrm{H}, J=2.9 \mathrm{~Hz}), 7.92(\mathrm{dd}, 1 \mathrm{H}, J=4.6,1.3$ $\mathrm{Hz}), 7.29-7.25(\mathrm{~m}, 3 \mathrm{H}), 7.20-7.14(\mathrm{~m}, 3 \mathrm{H}), 7.09(\mathrm{ddd}, 1 \mathrm{H}, \mathrm{J}=8.5,4.6,0.7 \mathrm{~Hz}), 6.93$ (ddd, 1H, $J=8.6,3.1,1.3 \mathrm{~Hz}), 3.56(\mathrm{t}, 2 \mathrm{H}, J=4.9 \mathrm{~Hz}), 3.50(\mathrm{t}, 2 \mathrm{H}, J=6.3 \mathrm{~Hz}), 2.80-$ $2.77(\mathrm{~m}, 2 \mathrm{H}), 2.66-2.59(\mathrm{~m}, 4 \mathrm{H}), 2.52(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=7.4 \mathrm{~Hz}), 2.01-1.96(\mathrm{~m}, 2 \mathrm{H}), 1.81(\mathrm{p}$, $2 \mathrm{H}, J=7.5 \mathrm{~Hz}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right) \delta 144.99,142.16,137.31,134.48,128.53$ (2C), 128.47 (2C), 125.94, 123.73, 117.92, 57.23, 55.07, 54.54, 48.64, 47.86, 33.63, 29.12, 27.45. 1-(3-Phenylpropyl)-4-(pyridin-3-yl)-1,4-diazepane oxalate. The free amine of $\mathbf{3 2}$ ( $121 \mathrm{mg}, 0.41 \mathrm{mmol}$ ) was converted into the oxalate salt as described under the general procedure affording a yellow solid ( $55 \mathrm{mg}, 35 \%$ ). Mp: 59.3-60.3 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}, 400 \mathrm{MHz}\right) \delta 8.13(\mathrm{~d}, 1 \mathrm{H}, J=2.9 \mathrm{~Hz}), 8.07(\mathrm{~d}, 1 \mathrm{H}, J=5.2 \mathrm{~Hz}), 7.88(\mathrm{ddd}$, $1 \mathrm{H}, J=9.1,2.9,1.0 \mathrm{~Hz}), 7.82(\mathrm{dd}, 1 \mathrm{H}, J=9.0,5.3 \mathrm{~Hz}), 7.40-7.37(\mathrm{~m}, 2 \mathrm{H}), 7.32-7.28$ $(\mathrm{m}, 3 \mathrm{H}), 3.89(\mathrm{t}, 2 \mathrm{H}, J=4.7 \mathrm{~Hz}), 3.68-3.56(\mathrm{~m}, 4 \mathrm{H}), 3.50-3.25(\mathrm{~m}, 2 \mathrm{H}), 3.22-3.18(\mathrm{~m}$, $2 \mathrm{H})$, $2.75(\mathrm{t}, 2 \mathrm{H}, J=7.3 \mathrm{~Hz})$, 2.33-2.27 (m, 2H), 2.16-2.08 (m, 2H). ${ }^{13} \mathrm{C}$ NMR ( $\mathrm{D}_{2} \mathrm{O}$, $101 \mathrm{MHz}) \delta 168.27$ (2C), 146.92, 140.37, 128.79 (2C), 128.58 (2C), 128.50, 127.55, 127.16, 126.57, 124.21, 56.08, 54.04, 53.59, 46.51, 43.18, 31.71, 25.31, 23.47.

Compound $\mathbf{3 3}$ was prepared as described for $\mathbf{2 8}$ from a mixture of 1-(pyridin-3-yl)-1,4diazepane ( $281 \mathrm{mg}, 1.59 \mathrm{mmol}$ ), 2-(3-bromophenyl)acetaldehyde[46] (411 mg, 2.07 $\mathrm{mmol})$, and $\mathrm{NaBH}(\mathrm{OAc})_{3}(506 \mathrm{mg}, 2.39 \mathrm{mmol})$ giving the product as a yellow oil (179 $\mathrm{mg}, 31 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 8.12(\mathrm{~d}, 1 \mathrm{H}, J=2.9 \mathrm{~Hz}), 7.92(\mathrm{dd}, 1 \mathrm{H}, J=$ $4.6,1.3 \mathrm{~Hz}), 7.33-7.29(\mathrm{~m}, 2 \mathrm{H}), 7.14-7.07(\mathrm{~m}, 3 \mathrm{H}), 6.93(\mathrm{ddd}, 1 \mathrm{H}, \mathrm{J}=8.6,3.1,1.3 \mathrm{~Hz})$, $3.57(\mathrm{t}, 2 \mathrm{H}, J=4.9 \mathrm{~Hz}), 3.50(\mathrm{t}, 2 \mathrm{H}, J=6.3 \mathrm{~Hz}), 2.86(\mathrm{t}, 2 \mathrm{H}, J=4.9 \mathrm{~Hz}), 2.75(\mathrm{~s}, 4 \mathrm{H})$, $2.70(\mathrm{t}, 2 \mathrm{H}, J=5.5 \mathrm{~Hz}), 2.03-1.97(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right) \delta 144.90$, $142.58,137.47,134.50,131.89,130.09,129.37,127.51,123.76,122.55,117.97,59.56$, 55.11, 54.55, 48.68, 47.81, 33.69, 27.40. 1-(3-Bromophenethyl)-4-(pyridin-3-yl)-1,4diazepane oxalate. The free amine of $\mathbf{3 3}(180 \mathrm{mg}, 0.50 \mathrm{mmol})$ was converted into the oxalate salt as described under the general procedure with the exception that 2.0 eq of oxalic acid was added. Salt formation afforded a yellow solid (183 mg, 74\%). Mp: $130.5-131.4{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}, 400 \mathrm{MHz}\right) \delta 8.18(\mathrm{~d}, 1 \mathrm{H}, J=3.0 \mathrm{~Hz}), 8.08-8.07(\mathrm{~m}$, $1 \mathrm{H}), 7.92$ (ddd, 1H, $J=9.1,3.0,1.0 \mathrm{~Hz}), 7.84(\mathrm{dd}, 1 \mathrm{H}, J=9.0,5.4 \mathrm{~Hz}), 7.55-7.52(\mathrm{~m}$, $2 \mathrm{H}), 7.35-7.32(\mathrm{~m}, 2 \mathrm{H}), 3.97-3.94(\mathrm{~m}, 2 \mathrm{H}), 3.71-3.49(\mathrm{~m}, 8 \mathrm{H}), 3.16-3.12(\mathrm{~m}, 2 \mathrm{H}), 2.41-$ $2.35(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{D}_{2} \mathrm{O}, 101 \mathrm{MHz}\right) \delta 165.98(2 \mathrm{C}), 146.99,138.45,131.63$, $130.76,130.36,128.58,127.65,127.58,127.18,124.21,122.22,57.67,54.12$ (2C), 46.60, 43.22, 29.62, 23.55.

### 3.2.30. 1-Benzyl-4-(pyridin-3-yl)piperazine (34)

$\mathbf{6 2}[40]$ was deprotected as described for $\mathbf{2 8}$ from a mixture of $\mathbf{6 2}(0.77 \mathrm{~g}, 2.92 \mathrm{mmol})$ and TFA ( $4.5 \mathrm{~mL}, 58.4 \mathrm{mmol}$ ) yielding 1-(pyridin-3-yl)piperazine as a yellow oil (410 $\mathrm{mg}, 86 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 8.31(\mathrm{dd}, 1 \mathrm{H}, J=2.6,1.1 \mathrm{~Hz}), 8.10(\mathrm{dd}, 1 \mathrm{H}, J$ $=4.0,2.0 \mathrm{~Hz}), 7.19-7.13(\mathrm{~m}, 2 \mathrm{H}), 3.17(\mathrm{dd}, 4 \mathrm{H}, J=6.2,3.8 \mathrm{~Hz}), 3.05(\mathrm{dd}, 4 \mathrm{H}, J=6.2$, $3.9 \mathrm{~Hz}), 1.85(\mathrm{~s}, 1 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right) \delta 147.53,140.91,138.86,123.58$, 122.49, 49.79 (2C), 46.08 (2C). Compound 34 was prepared as described for 28 from a
mixture of 1-(pyridin-3-yl)piperazine ( $100 \mathrm{mg}, 0.61 \mathrm{mmol}$ ), benzaldehyde $(62 \mu \mathrm{~L}, 0.61$ mmol), and $\mathrm{NaBH}(\mathrm{OAc})_{3}(195 \mathrm{mg}, \quad 0.92 \mathrm{mmol})$. Purification by DCVC (DCM:MeOH: $\mathrm{NH}_{3} / 100: 0: 0$ to $92.5: 6.75: 0.75$ ) afforded the product as a brown oil ( 80 $\mathrm{mg}, 52 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 8.30(\mathrm{dd}, 1 \mathrm{H}, J=2.6,1.1 \mathrm{~Hz}), 8.09(\mathrm{dd}, 1 \mathrm{H}, J$ $=4.0,2.0 \mathrm{~Hz}), 7.36-7.27(\mathrm{~m}, 5 \mathrm{H}), 7.15(\mathrm{dt}, 2 \mathrm{H}, J=4.2,1.6 \mathrm{~Hz}), 3.57(\mathrm{~s}, 2 \mathrm{H}), 3.23(\mathrm{t}$, $4 \mathrm{H}, J=5.1 \mathrm{~Hz}), 2.62(\mathrm{t}, 4 \mathrm{H}, J=5.1 \mathrm{~Hz}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right) \delta 147.15$, 140.76, 138.72, 138.01, 129.30 (2C), 128.45 (2C), 127.35, 123.56, 122.36, 63.16, 52.95 (2C), 48.62 (2C). 1-Benzyl-4-(pyridin-3-yl)piperazine oxalate. The free amine of $\mathbf{3 4}$ ( $80 \mathrm{mg}, 0.32 \mathrm{mmol}$ ) was converted into the oxalate salt as described under the general procedure with the exception that 2.1 eq of oxalic acid was added. Salt formation afforded yellow crystals ( $99 \mathrm{mg}, 90 \%$ ). Mp: $148.3^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}, 400 \mathrm{MHz}\right) \delta 8.38$ $(\mathrm{d}, 1 \mathrm{H}, J=2.9 \mathrm{~Hz}), 8.23(\mathrm{dt}, 1 \mathrm{H}, J=5.5,0.8 \mathrm{~Hz}), 8.14(\mathrm{ddd}, 1 \mathrm{H}, J=9.0,2.9,1.0 \mathrm{~Hz})$, $7.91(\mathrm{dd}, 1 \mathrm{H}, J=9.0,5.6 \mathrm{~Hz}), 7.61-7.56(\mathrm{~m}, 5 \mathrm{H}), 4.48(\mathrm{~s}, 2 \mathrm{H}), 4.11(\mathrm{~d}, 2 \mathrm{H}, J=9.0 \mathrm{~Hz})$, $3.71(\mathrm{~d}, 2 \mathrm{H}, J=8.7 \mathrm{~Hz}), 3.38(\mathrm{~d}, 4 \mathrm{H}, J=9.1 \mathrm{~Hz}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(\mathrm{D}_{2} \mathrm{O}, 101 \mathrm{MHz}\right) \delta 165.07$ (2C), 147.60, 131.37, 131.31, 131.23 (2C), 130.42, 129.36 (2C), 128.00, 127.70, $127.34,60.52,50.46$ (2C), 44.04 (2C).

### 3.2.31. 1-Phenethyl-4-(pyridin-3-yl)piperazine (35)

Compound 35 was prepared as described for 28 from a mixture of 1-(pyridin-3yl)piperazine ( $100 \mathrm{mg}, 0.61 \mathrm{mmol}$ ), 2-phenylacetaldehyde ( $71 \mu \mathrm{~L}, 0.61 \mathrm{mmol}$ ), and $\mathrm{NaBH}(\mathrm{OAc})_{3}$ (195 mg, 0.92 mmol ). Purification by DCVC (DCM:MeOH: $\mathrm{NH}_{3} / 100: 0: 0$ to $92.5: 6.75: 0.75)$ afforded the product as a brown oil $(90 \mathrm{mg}, 55 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right.$, $400 \mathrm{MHz}) \delta 8.21(\mathrm{bs}, 2 \mathrm{H}), 7.32-7.28(\mathrm{~m}, 2 \mathrm{H}), 7.24-7.18(\mathrm{~m}, 5 \mathrm{H}), 3.26(\mathrm{t}, 4 \mathrm{H}, J=5.1$ $\mathrm{Hz}), 2.85(\mathrm{dd}, 2 \mathrm{H}, J=9.9,6.3 \mathrm{~Hz}), 2.71-2.65(\mathrm{~m}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right) \delta$ $147.09,140.79,140.25,138.68,128.83$ (2C), 128.57 (2C), 126.28, 123.60, 122.44, 60.53, 53.07 (2C), 48.60 (2C), 33.73. 1-Phenethyl-4-(pyridin-3-yl)piperazine oxalate.

The free amine of $\mathbf{3 5}(40 \mathrm{mg}, 0.15 \mathrm{mmol})$ was converted into the oxalate salt as described under the general procedure with the exception that 2.1 eq of oxalic acid was added. Salt formation afforded yellow crystals ( $31 \mathrm{mg}, 58 \%$ ). Mp: $150.7^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}, 400 \mathrm{MHz}\right) \delta 8.40(\mathrm{~d}, 1 \mathrm{H}, J=2.9 \mathrm{~Hz}), 8.24(\mathrm{~d}, 1 \mathrm{H}, J=5.5 \mathrm{~Hz}), 8.15(\mathrm{ddd}, 1 \mathrm{H}, J=$ $9.0,2.9,1.0 \mathrm{~Hz}), 7.92(\mathrm{dd}, 1 \mathrm{H}, J=9.0,5.5 \mathrm{~Hz}), 7.49-7.38(\mathrm{~m}, 5 \mathrm{H}), 4.12(\mathrm{~d}, 2 \mathrm{H}, J=12.0$ $\mathrm{Hz}), 3.84(\mathrm{~d}, 2 \mathrm{H}, J=10.4 \mathrm{~Hz}), 3.59-3.55(\mathrm{~m}, 2 \mathrm{H}), 3.41(\mathrm{p}, 4 \mathrm{H}, J=12.6 \mathrm{~Hz}), 3.19(\mathrm{dd}$, $2 \mathrm{H}, J=9.4,6.8 \mathrm{~Hz}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}, 101 \mathrm{MHz}\right) \delta$ 147.60, 136.01, 131.36, 131.32, 129.13 (2C), 128.79 (2C), $127.69,127.46,127.35,57.51,51.01$ (2C), 44.05 (2C), 29.62.

### 3.2.32. 1-(3-Phenylpropyl)-4-(pyridin-3-yl)piperazine (36)

Compound 36 was prepared as described for 28 from a mixture of 1-(pyridin-3yl)piperazine ( $200 \mathrm{mg}, 1.23 \mathrm{mmol}$ ), 3-phenylpropanal ( $163 \mu \mathrm{~L}, 1.23 \mathrm{mmol}$ ), and $\mathrm{NaBH}(\mathrm{OAc})_{3}$ ( $392 \mathrm{mg}, 1.85 \mathrm{mmol}$ ). Purification by DCVC (DCM:MeOH: $\mathrm{NH}_{3} / 100: 0: 0$ to 95:4.5:0.5) afforded the product as a brown oil ( $200 \mathrm{mg}, 58 \%$ ). ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400\right.$ $\mathrm{MHz}) \delta 8.31(\mathrm{dd}, 1 \mathrm{H}, J=2.7,0.9 \mathrm{~Hz}), 8.09(\mathrm{dd}, 1 \mathrm{H}, J=4.1,1.8 \mathrm{~Hz}), 7.31-7.26(\mathrm{~m}$, $2 \mathrm{H}), 7.21-7.15(\mathrm{~m}, 5 \mathrm{H}), 3.23(\mathrm{t}, 4 \mathrm{H}, J=5.1 \mathrm{~Hz}), 2.67(\mathrm{t}, 2 \mathrm{H}, J=7.7 \mathrm{~Hz}), 2.60(\mathrm{t}, 4 \mathrm{H}, J$ $=5.1 \mathrm{~Hz}), 2.45-2.41(\mathrm{~m}, 2 \mathrm{H}), 1.86(\mathrm{dt}, 2 \mathrm{H}, J=15.2,7.6 \mathrm{~Hz}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101\right.$ $\mathrm{MHz}) \delta 147.11,142.19,140.79,138.71,128.55$ (2C), 128.47 (2C), 125.95, 123.55, 122.33, 58.03, 53.08 (2C), 48.62 (2C), 33.78, 28.69. 1-(3-Phenylpropyl)-4-(pyridin-3yl)piperazine oxalate. The free amine of $\mathbf{3 6}(200 \mathrm{mg}, 0.71 \mathrm{mmol})$ was converted into the oxalate salt as described under the general procedure with the exception that 2.1 eq of oxalic acid was added. Salt formation afforded yellow crystals ( $242 \mathrm{mg}, 92 \%$ ). Mp: $154.2{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}, 400 \mathrm{MHz}\right) \delta 8.40-8.38(\mathrm{~m}, 1 \mathrm{H}), 8.25-8.22(\mathrm{~m}, 1 \mathrm{H}), 8.15-8.12$ $(\mathrm{m}, 1 \mathrm{H}), 7.93-7.89(\mathrm{~m}, 1 \mathrm{H}), 7.46-7.32(\mathrm{~m}, 5 \mathrm{H}), 4.11-4.06(\mathrm{~m}, 2 \mathrm{H}), 3.78-3.72(\mathrm{~m}, 2 \mathrm{H})$, 3.44-3.36(m, 2H), 3.32-3.22(m, 4H), 2.81-2.76(m, 2H), 2.20-2.11 (m, 2H). ${ }^{13}$ C NMR
$\left(\mathrm{D}_{2} \mathrm{O}, 101 \mathrm{MHz}\right) \delta 165.52(2 \mathrm{C}), 147.61,140.42,131.34$ (2C), 128.83 (2C), 128.51 (2C), $127.70,127.33,126.60,56.23,50.87$ (2C), 44.10 (2C), 31.68, 24.94.

### 3.2.33. 3-[Benzyl(methyl)amino]butyl dimethylcarbamate (37)

Compound $\mathbf{3 7}$ was prepared as described for 9 from a mixture of $\mathbf{4 0}(6.47 \mathrm{~g}, 33.5$ mmol ), CDI ( $7.05 \mathrm{~g}, 43.5 \mathrm{mmol}$ ), and dimethylamine ( $15.1 \mathrm{~mL}, 77.0 \mathrm{mmol}, 33 \% \mathrm{in}$ ethanol (EtOH)). Purification by FCC (EtOAc:heptane/0:100 to 100:0) afforded the product as a yellow oil $(7.48 \mathrm{~g}, 85 \%) .{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 7.35-7.15(\mathrm{~m}$, $5 \mathrm{H}), 4.23-4.10(\mathrm{~m}, 2 \mathrm{H}), 3.58(\mathrm{~d}, 1 \mathrm{H}, J=13.3 \mathrm{~Hz}), 3.47(\mathrm{~d}, 1 \mathrm{H}, J=13.6 \mathrm{~Hz}), 2.95-2.70$ $(\mathrm{m}, 7 \mathrm{H}), 2.14(\mathrm{~s}, 3 \mathrm{H}), 1.95-1.82(\mathrm{~m}, 1 \mathrm{H}), 1.65-1.55(\mathrm{~m}, 1 \mathrm{H}), 1.02,1.00(2 \mathrm{~s}, 3 \mathrm{H}$, rotamers). ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right) \delta 156.74,140.01,128.59$ (2C), 128.10 (2C), 126.65, 63.36, 57.85, 53.97, 36.28, 36.02, 35.72, 33.14, 13.04.

### 3.2.34. Methyl 3-[benzyl(methyl)amino]but-2-enoate (38)

To a stirred suspension of methyl acetoacetate ( $5.00 \mathrm{~g}, 43.1 \mathrm{mmol}$ ) in toluene $(100 \mathrm{~mL})$ containing AcOH ( 2.15 mL ) was added methyl benzylamine ( $10.4 \mathrm{~g}, 85.8 \mathrm{mmol}$ ), and the mixture was stirred at room temperature (rt) for 5 days. The reaction mixture was washed with saturated (sat.) aqueous (aq) $\mathrm{NaHCO}_{3}$ solution $(2 \times 100 \mathrm{~mL})$, and the organic phase dried $\left(\mathrm{MgSO}_{4}\right)$, filtered, and evaporated in vacuo. Purification by Kugelrohr distillation $\left(0.4 \mathrm{mmHg}, 200{ }^{\circ} \mathrm{C}\right)$ afforded the product as a yellow oil $(9.34 \mathrm{~g}$, $>99 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}\right) \delta 7.35-7.23(\mathrm{~m}, 3 \mathrm{H}), 7.09(\mathrm{~d}, 2 \mathrm{H}, J=9 \mathrm{~Hz}), 4.71$ $(\mathrm{s}, 1 \mathrm{H}), 4.49(\mathrm{~s}, 2 \mathrm{H}), 3.62(\mathrm{~s}, 3 \mathrm{H}), 2.90(\mathrm{~s}, 3 \mathrm{H}), 2.53(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 75\right.$ MHz) $\delta 170.0,161.8,137.7,129.2(2 \mathrm{C}), 127.8,126.6$ (2C), 84.9, 55.3, 50.5, 38.9, 15.6.

Compound $\mathbf{3 9}$ was prepared as described for $\mathbf{3 8}$ from a mixture of methyl acetoacetate $(5.00 \mathrm{~g}, 43.1 \mathrm{mmol}), \mathrm{AcOH}(2.15 \mathrm{~mL})$, and methyl phenethylamine ( $11.6 \mathrm{~g}, 85.8 \mathrm{mmol}$ ) yielding the product as a yellow oil $(9.57 \mathrm{~g}, 93 \%) .{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}\right) \delta 7.31-$ $7.13(\mathrm{~m}, 5 \mathrm{H}), 4.60(\mathrm{~s}, 1 \mathrm{H}), 3.62(\mathrm{~s}, 3 \mathrm{H}), 3.48(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=8 \mathrm{~Hz}), 2.83-2.78(\mathrm{~m}, 5 \mathrm{H}), 2.40$ (s, 3H). ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right) \delta 170.0,161.3,138.9,129.1$ (2C), 129.1 (2C), 127.0, 84.3, 54.1, 50.5, 38.9, 34.8, 15.5.

### 3.2.36. 3-[Benzyl(methyl)amino]butan-1-ol (40)

$38(4.25 \mathrm{~g}, 19.4 \mathrm{mmol})$ dissolved in dry $i-\operatorname{PrOH}(19 \mathrm{~mL})$ and dry THF ( 48 mL ) was treated with sodium $(3.84 \mathrm{~g}, 166 \mathrm{mmol})$ at $0{ }^{\circ} \mathrm{C}$. The reaction mixture was allowed to warm to rt over 17 h . The reaction mixture was poured into sat. aq $\mathrm{NH}_{4} \mathrm{Cl}$ solution (50 $\mathrm{mL})$, extracted with DCM ( $3 \times 50 \mathrm{~mL}$ ), and the organic phase evaporated in vacuo. The residue was dissolved in aq HCl solution $(4 \mathrm{M}, 20 \mathrm{~mL})$ and washed with $\mathrm{Et}_{2} \mathrm{O}(2 \times 20$ $\mathrm{mL})$. The water phase was neutralized with solid $\mathrm{NaHCO}_{3}$ and extracted with DCM ( $3 \times$ $20 \mathrm{~mL})$. The combined organic phases were dried $\left(\mathrm{MgSO}_{4}\right)$, filtered, and evaporated in vacuo. Purification by DCVC (DCM:MeOH: $\mathrm{NH}_{3} / 100: 0: 0$ to 100:4.5:0.5) afforded the product as a brown oil ( $880 \mathrm{mg}, 24 \%$ ). ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}\right) \delta 7.33-7.25(\mathrm{~m}$, $5 \mathrm{H}), 5.99(\mathrm{bs}, 1 \mathrm{H}), 3.81-3.77(\mathrm{~m}, 2 \mathrm{H}), 3.70(\mathrm{~d}, 1 \mathrm{H}, J=12 \mathrm{~Hz}), 3.52(\mathrm{~d}, 1 \mathrm{H}, J=12 \mathrm{~Hz})$, $3.11-3.00(\mathrm{~m}, 1 \mathrm{H}), 2.19(\mathrm{~s}, 3 \mathrm{H}), 2.01-2.87(\mathrm{~m}, 1 \mathrm{H}), 1.34(\mathrm{dq}, 1 \mathrm{H}, J=15,3 \mathrm{~Hz}), 1.02(\mathrm{~d}$, $3 \mathrm{H}, J=6 \mathrm{~Hz}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right) \delta 138.9,129.4$ (2C), 128.9 (2C), 127.6, 64.5, 59.7, 59.2, 35.7, 34.6, 12.4.

### 3.2.37. 3-[Phenethyl(methyl)amino]butan-1-ol (41)

Compound 41 was prepared as described for $\mathbf{4 0}$ from a mixture of $\mathbf{3 9}(5.00 \mathrm{~g}, 21.4$ mmol) and sodium ( $4.43 \mathrm{~g}, 193 \mathrm{mmol})$. Purification by DCVC (DCM:MeOH: $\mathrm{NH}_{3} / 100: 0: 0$ to $100: 9: 1$ ) yielded the product as a yellow oil $(2.97 \mathrm{~g}$,
$67 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}\right) \delta 7.31-7.10(\mathrm{~m}, 5 \mathrm{H}), 5.84(\mathrm{bs}, 1 \mathrm{H}), 3.84-3.70(\mathrm{~m}$, $2 H), 3.05-2.96(\mathrm{~m}, 1 \mathrm{H}), 2.82-2.71(\mathrm{~m}, 2 \mathrm{H}), 2.62-2.55(\mathrm{~m}, 2 \mathrm{H}), 2.29(\mathrm{~s}, 3 \mathrm{H}), 1.92-1.78$ $(\mathrm{m}, 1 \mathrm{H}), 1.30(\mathrm{dq}, 1 \mathrm{H}, J=12,3 \mathrm{~Hz}), 0.93(\mathrm{~d}, 3 \mathrm{H}, J=6 \mathrm{~Hz}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right)$ $\delta 140.2,129.1$ (2C), 129.1 (2C), 126.9, 65.5, 62.7, 60.6, 36.4, 35.4, 34.7, 13.0.
3.2.38. 3-(Methylamino)butyl methylcarbamate acetate (42)
$9(645 \mathrm{mg}, 2.59 \mathrm{mmol}), \mathrm{AcOH}(6 \mathrm{~mL})$, and $\mathrm{Pd} / \mathrm{C}(32 \mathrm{mg}, 5 \% \mathrm{w} / \mathrm{w})$ were added to a flask that was evacuated and flushed with $\mathrm{H}_{2}$ and stirred at rt for 52 h under a positive pressure of $\mathrm{H}_{2}$. The reaction mixture was diluted with EtOAc and filtered through a plug of Celite. The filtrate was evaporated in vacuo to give the product as a brown oil (645 mg , quantitative). ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}, 300 \mathrm{MHz}\right) \delta 4.20-4.05(\mathrm{~m}, 2 \mathrm{H}), 3.36-3.29(\mathrm{~m}, 1 \mathrm{H})$, $2.64(\mathrm{~s}, 6 \mathrm{H}), 2.11-1.98(\mathrm{~m}, 1 \mathrm{H}), 1.90(\mathrm{~s}, 3 \mathrm{H}), 1.90-1.82(\mathrm{~m}, 1 \mathrm{H}), 1.29(\mathrm{~d}, 3 \mathrm{H}, J=9 \mathrm{~Hz})$. ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{D}_{2} \mathrm{O}, 75 \mathrm{MHz}\right) \delta 178.1,157.4,61.3,53.0,32.3,30.2,27.8,23.8,16.1 .42$ ( $3.40 \mathrm{~g}, 15.6 \mathrm{mmol}$ ) was dissolved in aq NaOH solution $(2 \mathrm{M}, 70 \mathrm{~mL}$ ) and extracted with EtOAc ( $3 \times 70 \mathrm{~mL}$ ). The combined organic phases were dried $\left(\mathrm{MgSO}_{4}\right)$, filtered, and evaporated in vacuo affording the free amine of $\mathbf{4 2}(2.03 \mathrm{~g}, 82 \%)$.

### 3.2.39. 3-(Methylamino)butyl dimethylcarbamate (43)

Compound $\mathbf{4 3}$ was prepared as described for $\mathbf{4 2}$ from a mixture of $\mathbf{3 7}(2.53 \mathrm{~g}, 9.59$ $\mathrm{mmol})$ and $\mathrm{Pd} / \mathrm{C}(127 \mathrm{mg}, 5 \% \mathrm{w} / \mathrm{w})$. Basic extraction afforded the free amine of 43 as a transparent oil ( $170 \mathrm{mg}, 69 \%$ ). ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 4.17-4.03(\mathrm{~m}, 2 \mathrm{H}), 2.87$ (s, 6H), $2.65(\mathrm{sxt}, 1 \mathrm{H}, J=6.3 \mathrm{~Hz}), 2.38(\mathrm{~s}, 3 \mathrm{H}), 1.92(\mathrm{bs}, 1 \mathrm{H}), 1.86-1.73(\mathrm{~m}, 1 \mathrm{H}), 1.66-$ $1.54(\mathrm{~m}, 1 \mathrm{H}), 1.07,1.05\left(2 \mathrm{~s}, 3 \mathrm{H}\right.$, rotamers). ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right) \delta 156.58$, 62.80, 60.27, 52.20, 35.78, 33.49 (2C), 19.73.

To $\mathbf{4 4}$ [38] ( $2.08 \mathrm{~g}, 10.2 \mathrm{mmol}$ ) was added dimethylamine $(7.28 \mathrm{~mL}, 40.8 \mathrm{mmol}, 33 \%$ in $\mathrm{EtOH})$, and the reaction mixture was stirred for 4 days at rt . The solvent was evaporated in vacuo giving the product as a yellow oil ( $2.30 \mathrm{~g}, 91 \%$ ). ${ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CD}_{3} \mathrm{OD}, 400 \mathrm{MHz}\right)$ $\delta 7.28-7.22(\mathrm{~m}, 2 \mathrm{H}), 7.21-7.11(\mathrm{~m}, 3 \mathrm{H}), 4.11(\mathrm{q}, 2 \mathrm{H}, J=6.8 \mathrm{~Hz}), 2.98(\mathrm{p}, 1 \mathrm{H}, J=6.4$ Hz) $2.64(\mathrm{t}, 2 \mathrm{H}, J=7.9 \mathrm{~Hz}), 2.57(\mathrm{dd}, 1 \mathrm{H}, J=14.8,6.0 \mathrm{~Hz}), 2.29(\mathrm{dd}, 1 \mathrm{H}, J=14.8,7.3$ $\mathrm{Hz}), 2.23(\mathrm{~s}, 6 \mathrm{H}), 1.90-1.79(\mathrm{~m}, 1 \mathrm{H}), 1.69-1.58(\mathrm{~m}, 1 \mathrm{H}), 1.23(\mathrm{t}, 4 \mathrm{H}, J=7.0 \mathrm{~Hz}) .{ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{CD}_{3} \mathrm{OD}, 101 \mathrm{MHz}\right) \delta 174.84,143.47,129.58$ (2C), 129.52 (2C), 127.01, 61.89, 61.77, 40.68 (2C), $35.32,34.21,34.00,14.61$.

### 3.2.41. Ethyl 3-(dimethylamino)-6-phenylhexanoate (49)

Compound 49 was prepared as described for 48 from a mixture of dimethylamine (11.7 $\mathrm{mL}, 65.4 \mathrm{mmol}, 33 \%$ in EtOH$)$ and $45[37,38](3.57 \mathrm{~g}, 16.4 \mathrm{mmol})$ yielding the product as a yellow oil $(4.05 \mathrm{~g}, 94 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 7.28-7.26(\mathrm{~m}, 1 \mathrm{H}), 7.25(\mathrm{t}$, $1 \mathrm{H}, J=2.2 \mathrm{~Hz}), 7.18-7.15(\mathrm{~m}, 3 \mathrm{H}), 4.12(\mathrm{q}, 2 \mathrm{H}, J=7.1 \mathrm{~Hz}), 3.06-2.99(\mathrm{~m}, 1 \mathrm{H}), 2.62(\mathrm{t}$, $2 \mathrm{H}, J=7.7 \mathrm{~Hz}), 2.51(\mathrm{dd}, 1 \mathrm{H}, J=14.4,5.7 \mathrm{~Hz}), 2.23(\mathrm{~s}, 6 \mathrm{H}), 2.16(\mathrm{dd}, 1 \mathrm{H}, J=14.2,7.7$ $\mathrm{Hz}), 1.74-1.62(\mathrm{~m}, 2 \mathrm{H}), 1.60-1.52(\mathrm{~m}, 1 \mathrm{H}), 1.40-1.31(\mathrm{~m}, 1 \mathrm{H}), 1.23(\mathrm{t}, 3 \mathrm{H}, J=7.1 \mathrm{~Hz})$. ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right) \delta 173.32,142.46,128.54$ (2C), 128.40 (2C), 125.83, $61.13,60.53,40.39$ (2C), $35.96,34.33,30.83,28.51,14.34$.

### 3.2.42. Ethyl 3-(dimethylamino)-7-phenylheptanoate (50)

Compound $\mathbf{5 0}$ was prepared as described for $\mathbf{4 8}$ from a mixture of dimethylamine (8.0 $\mathrm{mL}, 51 \mathrm{mmol}, 33 \%$ in EtOH$)$ and $\mathbf{4 6}[37,38](2.98 \mathrm{~g}, 12.8 \mathrm{mmol})$ furnishing the product as a yellow oil $(3.52 \mathrm{~g}, 99 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 7.29-7.26(\mathrm{~m}, 2 \mathrm{H}), 7.19-$ $7.16(\mathrm{~m}, 3 \mathrm{H}), 4.12(\mathrm{q}, 2 \mathrm{H}, J=7.1 \mathrm{~Hz}), 2.98(\mathrm{dt}, 1 \mathrm{H}, J=13.6,7.1 \mathrm{~Hz}), 2.61(\mathrm{t}, 2 \mathrm{H}, J=$ $7.8 \mathrm{~Hz}), 2.50(\mathrm{dd}, 1 \mathrm{H}, J=14.5,6.0 \mathrm{~Hz}), 2.23(\mathrm{~s}, 6 \mathrm{H}), 2.15(\mathrm{dd}, 1 \mathrm{H}, J=14.5,7.5 \mathrm{~Hz})$, $1.66-1.59(\mathrm{~m}, 3 \mathrm{H}), 1.42-1.30(\mathrm{~m}, 3 \mathrm{H}), 1.25(\mathrm{t}, 3 \mathrm{H}, J=7.1 \mathrm{~Hz}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101\right.$
$\mathrm{MHz}) \delta 173.25,142.77,128.52(2 \mathrm{C}), 128.39(2 \mathrm{C}), 125.77,61.18,60.51,40.40(2 \mathrm{C})$, 36.01, 34.44, 31.67, 31.03, 26.53, 14.36.

### 3.2.43. Ethyl-3-(dimethylamino)-8-phenyloctanoate (51)

Compound $\mathbf{5 1}$ was prepared as described for $\mathbf{4 8}$ from a mixture of dimethylamine (4.85 $\mathrm{mL}, 27.2 \mathrm{mmol}, 33 \% \mathrm{in} \mathrm{EtOH}$ ) and $47[37,38](1.12 \mathrm{~g}, 4.50 \mathrm{mmol})$ furnishing the product as a yellow oil $(1.18 \mathrm{~g}, 90 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 400 \mathrm{MHz}\right) \delta 7.27-7.20(\mathrm{~m}$, $2 \mathrm{H}), 7.18-7.10(\mathrm{~m}, 3 \mathrm{H}), 4.11(\mathrm{q}, 2 \mathrm{H}, J=7.3 \mathrm{~Hz}), 3.55-3.35(\mathrm{~m}, 1 \mathrm{H}), 3.00-2.75(\mathrm{~m}, 1 \mathrm{H})$, $2.65-2.50(\mathrm{~m}, 7 \mathrm{H}), 2.50-2.40(\mathrm{~m}, 1 \mathrm{H}), 1.90-1.77(\mathrm{~m}, 1 \mathrm{H}), 1.63(\mathrm{p}, 2 \mathrm{H}, J=7.4 \mathrm{~Hz}), 1.57-$ $1.47(\mathrm{~m}, 1 \mathrm{H}), 1.45-1.32(\mathrm{~m}, 5 \mathrm{H}), 1.28(\mathrm{t}, 3 \mathrm{H}, J=7.3 \mathrm{~Hz}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 101\right.$ $\mathrm{MHz}) \delta 144.24,129.76$ (2C), 129.62 (2C), 127.01, 62.74, 61.98, 40.98, 40.94, 37.17, $35.81,32.95,27.94,30.57,32.22,14.84$.

### 3.2.44. 3-(Dimethylamino)-5-phenylpentan-1-ol (52)

A mixture of $48(2.30 \mathrm{~g}, 9.22 \mathrm{mmol})$ in dry $\mathrm{Et}_{2} \mathrm{O}(12 \mathrm{~mL})$ was slowly added to a slurry mixture of $\mathrm{LiAlH}_{4}(0.70 \mathrm{~g}, 18.5 \mathrm{mmol})$ in dry $\mathrm{Et}_{2} \mathrm{O}(12 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$. The reaction mixture was allowed to reach rt and stirring was continued overnight. $\mathrm{H}_{2} \mathrm{O}(2 \mathrm{~mL})$, aq NaOH solution $(5 \mathrm{M}, 1 \mathrm{~mL})$, and $\mathrm{H}_{2} \mathrm{O}(4 \mathrm{~mL})$ were slowly added at $0^{\circ} \mathrm{C}$. The precipitate was isolated and washed thoroughly with $\mathrm{Et}_{2} \mathrm{O} . \mathrm{H}_{2} \mathrm{O}$ was added to the filtrate and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 60 \mathrm{~mL})$. The combined organic phases were dried $\left(\mathrm{MgSO}_{4}\right)$, filtered, and evaporated in vacuo yielding the product as a yellow oil $(1.51 \mathrm{~g}, 79 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 400 \mathrm{MHz}\right) \delta 7.28-7.22(\mathrm{~m}, 2 \mathrm{H}), 7.21-7.12(\mathrm{~m}, 3 \mathrm{H}), 3.68-3.62(\mathrm{~m}, 2 \mathrm{H})$, 2.72-2.63 (m, 1H), 2.63-2.54 (m, 2H), 2.23 (s, 6H) 1.91-1.72 (m, 2H), 1.61-1.48 (m, $2 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 101 \mathrm{MHz}\right) \delta 143.49,129.33(4 \mathrm{C}), 126.78,63.17,62.28,40.38$ (2C), 34.26, 32.60, 31.93.
3.2.45. 3-(Dimethylamino)-6-phenylhexan-1-ol (53)

Compound $\mathbf{5 3}$ was prepared as described for $\mathbf{5 2}$ from a mixture of $\mathbf{4 9}(4.05 \mathrm{~g}, 15.4$ $\mathrm{mmol})$ and $\mathrm{LiAlH}_{4}(0.79 \mathrm{~g}, 20.8 \mathrm{mmol})$ yielding the product as a yellow oil $(3.03 \mathrm{~g}$, $89 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 7.28(\mathrm{~d}, 2 \mathrm{H}, J=7.5 \mathrm{~Hz})$, 7.18-7.16 (m, 3H), 3.81$3.78(\mathrm{~m}, 2 \mathrm{H}), 2.70-2.57(\mathrm{~m}, 3 \mathrm{H}), 2.28(\mathrm{~s}, 6 \mathrm{H}), 1.77-1.57(\mathrm{~m}, 4 \mathrm{H}), 1.55-1.46(\mathrm{~m}, 2 \mathrm{H})$. ${ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right) \delta 142.18,128.51$ (2C), 128.44 (2C), 126.01, 66.49, 64.18, 40.05 (2C), 36.19, 31.01, 29.45, 26.19.

### 3.2.46. 3-(Dimethylamino)-7-phenylheptan-1-ol (54)

Compound $\mathbf{5 4}$ was prepared as described for $\mathbf{5 2}$ from a mixture of $\mathbf{5 0}(2.88 \mathrm{~g}, 10.4$ $\mathrm{mmol})$ and $\mathrm{LiAlH}_{4}(0.79 \mathrm{~g}, 20.8 \mathrm{mmol})$ affording the product as a yellow oil $(2.13 \mathrm{~g}$, $87 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 7.30-7.27(\mathrm{~m}, 2 \mathrm{H}), 7.20-7.16(\mathrm{~m}, 3 \mathrm{H}), 6.30(\mathrm{~s}$, $1 \mathrm{H}), 3.79(\mathrm{dd}, 2 \mathrm{H}, J=7.3,3.2 \mathrm{~Hz}), 2.64-2.58(\mathrm{~m}, 3 \mathrm{H}), 2.27(\mathrm{~s}, 6 \mathrm{H}), 1.71-1.56(\mathrm{~m}, 4 \mathrm{H})$, $1.47(\mathrm{q}, 1 \mathrm{H}, J=2.9 \mathrm{~Hz}), 1.44-1.42(\mathrm{~m}, 1 \mathrm{H}), 1.26-1.07(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101\right.$ MHz) $\delta 142.54,128.50$ (2C), 128.44 (2C), 125.87, 66.65, 64.46, 40.03 (2C), 35.99, 31.75, 30.91, 27.24, 26.25.

### 3.2.47. 3-(Dimethylamino)-8-phenyloctan-1-ol (55)

Compound $\mathbf{5 5}$ was prepared as described for $\mathbf{5 2}$ from a mixture of $\mathbf{5 1}(1.37 \mathrm{~g}, 4.70$ $\mathrm{mmol})$ and $\mathrm{LiAlH}_{4}(0.36 \mathrm{~g}, 9.40 \mathrm{mmol})$ furnishing the product as a yellow oil $(0.975 \mathrm{~g}$, $83 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 400 \mathrm{MHz}\right) \delta 7.26-7.21(\mathrm{~m}, 2 \mathrm{H}), 7.18-7.11(\mathrm{~m}, 3 \mathrm{H}), 3.65(\mathrm{ddd}$, $2 \mathrm{H}, J=7.3,5.5,1.8 \mathrm{~Hz}), 2.61(\mathrm{t}, 2 \mathrm{H}, J=7.5 \mathrm{~Hz}), 2.57-2.49(\mathrm{~m}, 1 \mathrm{H}), 2.23(\mathrm{~s}, 6 \mathrm{H}), 1.74$ (sxt, 1H, $J=7.5 \mathrm{~Hz}), 1.64(\mathrm{t}, 2 \mathrm{H}, J=7.3 \mathrm{~Hz}), 1.49(\mathrm{dq}, 2 \mathrm{H}, J=5.5 \mathrm{~Hz}), 1.39-1.33(\mathrm{~m}$, $4 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 101 \mathrm{MHz}\right) \delta 129.55(2 \mathrm{C}), 129.41(2 \mathrm{C}), 126.80,64.12,62.60$, 40.65 (2C), 36.99, 33.01, 32.79, 29.87, 28.25, 30.60.

A solution of $\mathbf{5 8 [ 4 0 ]}(416 \mathrm{mg}, 1.50 \mathrm{mmol})$ in dry $\mathrm{MeCN}(15 \mathrm{~mL})$ was added NBS (267 $\mathrm{mg}, 1.50 \mathrm{mmol}$ ) and stirred for 1 h at rt . The reaction mixture was evaporated in vacuo. Purification by DCVC (DCM:MeOH: $\mathrm{NH}_{3} / 100: 0: 0$ to $97.5: 2.25: 0.25$ ) afforded the product as a yellow solid ( 598 mg , quantitative). ${ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 7.86(\mathrm{~d}$, $1 \mathrm{H}, J=3.0 \mathrm{~Hz}), 7.25(\mathrm{~s}, 1 \mathrm{H}), 6.92-6.86(\mathrm{~m}, 1 \mathrm{H}), 3.48-3.62(\mathrm{~m}, 6 \mathrm{H}), 3.35(\mathrm{t}, 1 \mathrm{H}, J=5.6$ $\mathrm{Hz}), 3.25(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}=6.0 \mathrm{~Hz}), 1.99-1.90(\mathrm{~m}, 2 \mathrm{H}), 1.42(\mathrm{~s}, 5 \mathrm{H}), 1.38(\mathrm{~s}, 4 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right) \delta 177.38 ; 155.16,154.75$ (rotamers); 142.91, 142.80 (rotamers); 133.86, 133.62 (rotamers); 127.87, 127.78 (rotamers); 121.45; 79.89; 49.94, 49.83 (rotamers); 48.62, 48.01 (rotamers); 46.15, 45.90 (rotamers); 45.84, 45.68 (rotamers); 29.56 (3C); 28.31, 28.26 (rotamers).
3.2.49. 3-(Dimethylamino)-6-phenyl-6-((tetrahydro-2H-pyran-2-yl)oxy)hexyl methylcarbamate (60) A mixture of $57[39](6.12 \mathrm{~g}, 26.1 \mathrm{mmol})$ and 3,4-dihydro-2Hpyran ( $4.78 \mathrm{~mL}, 52.2 \mathrm{mmol}$ ) was added a few drops of concentrated aq HCl solution. The reaction mixture was stirred for 2 days at rt . The reaction mixture was added $\mathrm{Et}_{2} \mathrm{O}$ ( 100 mL ) and extracted with sat. aq $\mathrm{NaHCO}_{3}$ solution (2 x 50 mL ) and $\mathrm{H}_{2} \mathrm{O}(50 \mathrm{~mL})$. The combined organic phases were dried $\left(\mathrm{MgSO}_{4}\right)$, filtered, and evaporated in vacuo. The crude product was purified by DCVC (heptane:EtOAc/2:1) giving ( $E$ )-ethyl 6-phenyl-6-((tetrahydro-2H-pyran-2-yl)oxy)hex-2-enoate as a yellow oil (6.85 g, quantitative). (Ethyl 3-(dimethylamino)-6-phenyl-6-((tetrahydro-2H-pyran-2yl)oxy)hexanoate was prepared as described for 48 from a mixture of ( $E$ )-ethyl 6-phenyl-6-((tetrahydro-2H-pyran-2-yl)oxy)hex-2-enoate (6.82 g, 21.4 mmol$)$ and dimethylamine ( $15.3 \mathrm{~mL}, 85.7 \mathrm{mmol}, 33 \% \mathrm{in} \mathrm{EtOH}$ ) yielding the product as a yellow oil (7.68 g, 99\%). 3-(Dimethylamino)-6-phenyl-6-((tetrahydro-2H-pyran-2-yl)oxy)hexan-1-ol was prepared as described for 52 from a mixture of (ethyl 3-
(dimethylamino)-6-phenyl-6-((tetrahydro-2H-pyran-2-yl)oxy)hexanoate (7.68 g, 21.1 $\mathrm{mmol})$ and $\mathrm{LiAlH}_{4}(1.60 \mathrm{~g}, 42.3 \mathrm{mmol})$ affording the product as a yellow oil $(6.48 \mathrm{~g}$, $95 \%$ ). Compound 60 was prepared as described for 13 from a mixture of 3-(dimethylamino)-6-phenyl-6-((tetrahydro-2H-pyran-2-yl)oxy)hexan-1-ol (1.00 g, 3.11 $\mathrm{mmol})$, CDI ( $0.61 \mathrm{~g}, 3.73 \mathrm{mmol}$ ), and methylamine ( $2.62 \mathrm{~mL}, 31.1 \mathrm{mmol}, 40 \%$ in $\mathrm{H}_{2} \mathrm{O}$ ) yielding the product as a yellow oil ( $1.08 \mathrm{~g}, 92 \%) .{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 7.38$ $7.15(\mathrm{~m}, 11 \mathrm{H}$, diastereomers); 4.86-4.81 (m, 1H), $4.66(\mathrm{ddd}, 1 \mathrm{H}, J=7.7,5.7,2.0 \mathrm{~Hz})$ (diastereomers); $4.56(\mathrm{t}, 1 \mathrm{H}, J=6.4 \mathrm{~Hz}), 4.40(\mathrm{t}, 1 \mathrm{H}, J=3.6 \mathrm{~Hz})$ (diastereomers); 4.134.01 (m, 4H), 3.94 (dt, 1H, $J=11.2,5.7 \mathrm{~Hz}$ ) (diastereomers); 3.60-3.45 (m, 2H), 3.32$3.23(\mathrm{~m}, 1 \mathrm{H})$ (diastereomers); 2.96-2.83 (m, 12H, diastereomers); 2.52-2.39 (m, 2H, diastereomers); 2.19, 2.18, 2.17, 2.17 (4 s, 3 H , diastereomers); 1.95-1.11 (m, 27H, diastereomers). ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right) \delta 142.39,142.43$ (diastereomers); 128.26, 128.04, 127.40, 127.38, 126.95, 126.94, 126.93, 126.45, 126.44 (diastereomers); 97.87, 97.83, $95.29,95.27$ (diastereomers); 78.84, 78.80 (diastereomers); 63.85, 63.82 (diastereomers); 62.92, 62.40, 61.88 (diastereomers); 60.84, 60.80, 60.78 (diastereomers); 40.28, 40.23, 40.21, 40.19 (diastereomers); 35.64, 35.58 (diastereomers); 34.12, 34.07 (diastereomers); 30.76, 30.71 (diastereomers); 28.91, 28.79, 28.69, 28.60 (diastereomers); 26.21, 25.88, 25.55, 25.42, 25.30 (diastereomers); 19.55, 19.19 (diastereomers). LC-MS $\mathrm{m} / \mathrm{z}=379.1[\mathrm{M}+\mathrm{H}]^{+}$.

### 3.2.50. 3-(Dimethylamino)-6-phenyl-6-((tetrahydro-2H-pyran-2-yl)oxy)hexyl dimethylcarbamate (61)

Compound $\mathbf{6 1}$ was prepared as described for $\mathbf{2 2}$ from a mixture of 3-(dimethylamino)-6-phenyl-6-((tetrahydro-2H-pyran-2-yl)oxy)hexan-1-ol (1.00 g, 3.11 mmol ) (described above), CDI ( $0.61 \mathrm{~g}, 3.73 \mathrm{mmol}$ ), and dimethylamine ( $3.94 \mathrm{~mL}, 31.1 \mathrm{mmol}, 40 \% \mathrm{in}$ $\left.\mathrm{H}_{2} \mathrm{O}\right)$ yielding the product as a yellow oil ( $\left.0.92 \mathrm{~g}, 75 \%\right) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta$
7.40-7.21 (m, 13H, diastereomers); 7.20-7.09 (m, 2H, diastereomers); 4.88-4.80 (m, $1 \mathrm{H})$, 4.72-4.52 (m, 4H), 4.42-4.35 (m, 1H) (diastereomers); 4.18-4.00 (m, 5H), 3.99$3.84(\mathrm{~m}, 1 \mathrm{H})$ (diastereomers); 3.61-3.45 (m, 3H), 3.33-3.23 (m, 1H) (diastereomers); 2.81-2.73 (m, 6H, diastereomers); 2.52-2.38 (m, 2H, diastereomers); 2.19, $2.18(2 \mathrm{~s}, 3 \mathrm{H}$, diastereomers); 2.16 (s, 6 H , diastereomers); 1.96-1.09 (m, 31H, diastereomers). ${ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right) \delta 129.02,128.28,128.21,128.06,127.42,126.99,126.96$, 126.94, 126.52, 126.48, 125.28 (diastereomers); 97.80, 95.43, 95.35, 94.66 (diastereomers); 78.73, 77.23 (diastereomers); 63.37, 62.93, 62.52, 62.44 (diastereomers); 61.93, 61.58 (diastereomers); 60.86, 60.81, 60.68 (diastereomers); 40.34, 40.28, 40.23 (diastereomers); 35.64; 34.07; 30.81, 30.78, 30.70 (diastereomers); 28.97, 28.84, 28.78, 28.68 (diastereomers); 26.04, 25.56 (diastereomers); 25.44,.25.30, 25.00 (diastereomers); 19.81, 19.76, 19.62, 19.58, 19.20, 19.12 (diastereomers). LC-MS $\mathrm{m} / \mathrm{z}=393.0[\mathrm{M}+\mathrm{H}]^{+}$.

### 3.3 Pharmacology

### 3.3.1. Materials

Culture media, serum, antibiotics and buffers for cell culture were obtained from Invitrogen (Paisley, UK). (S)-Nicotine was obtained from Sigma (St. Louis, MO) and $\left[{ }^{3} \mathrm{H}\right]$ epibatidine from PerkinElmer (Boston, MA). The FLIPR ${ }^{\text {TM }}$ Membrane Potential Blue (FMP) dye was purchased from Molecular Devices (Crawley, UK). The HEK293 cell lines stably expressing the rat $\alpha 3 \beta 4$ and rat $\alpha 4 \beta 4$ were generous gifts from Drs. K. Kellar and Y. Xiao (Georgetown University School of Medicine, Washington DC), and the stable mouse $\alpha 4 \beta 2$-HEK293T cell line and the rat $\alpha 4 \beta 2$-HEK293 cell line were generously provided by Dr. J.A. Stitzel (University of Colorado, Boulder, CO) and Dr. J.H. Steinbach (Washington University School of Medicine, St. Louis, MO), respectively [47-49].

### 3.3.2. Cell culture

The tsA201 and HEK293 cells lines were maintained at $37{ }^{\circ} \mathrm{C}$ in a humidified $5 \% \mathrm{CO}_{2}$ incubator in culture medium [Dulbecco's Modified Eagle Medium supplemented with penicillin ( $100 \mathrm{U} / \mathrm{ml}$ ), streptomycin ( $100 \mathrm{mg} / \mathrm{ml}$ ) and $10 \%$ fetal bovine serum]. The culture medium used for the stable HEK293 cell lines expressing rat $\alpha 4 \beta 2, \alpha 4 \beta 4$ and $\alpha 3 \beta 4 \mathrm{nAChRs}$ were supplemented with $1 \mathrm{mg} / \mathrm{ml} \mathrm{G-418}$, for the $\alpha 4 \beta 2$-HEK293T cells were supplemented with $0.1 \mathrm{mg} / \mathrm{ml}$ zeocin and $0.5 \mathrm{mg} / \mathrm{ml}$ hygromycin $B$.

### 3.3.3. [3]H]Epibatidine Binding

The binding experiments with stable HEK293 cell lines expressing rat $\alpha 4 \beta 2$, $\alpha 3 \beta 4$ and $\alpha 4 \beta 4$ nAChRs were performed essentially as previously described [41]. Briefly, cells were harvested at $80-90 \%$ confluency and scraped into assay buffer $[140 \mathrm{mM} \mathrm{NaCl} / 1.5$ $\mathrm{mM} \mathrm{KCl} / 2 \mathrm{mM} \mathrm{CaCl} 2 / 1 \mathrm{mM} \mathrm{Mg} 2 \mathrm{SO}_{4} / 25 \mathrm{mM}$ HEPES ( pH 7.4 )], homogenized using a polytron for 10 sec and centrifuged for 20 min at $50.000 \times g$. Cell pellets were resuspended in fresh assay buffer, homogenized and centrifuged at $50.000 \times g$ for another 20 min . Then the cell pellet were resuspended in the assay buffer, and the cell membranes were incubated with $10 \mathrm{pM}\left[{ }^{3} \mathrm{H}\right]$ epibatidine in the presence of various concentrations of compounds in a total assay volume of 2 mL . Nonspecific binding was determined in reactions with $100 \mathrm{mM}(S)$-nicotine. The reactions were incubated for 4 h at room temperature while shaking. Whatman GF/C filters were presoaked for 1 h in a $0.2 \%$ polyethyleneimine solution, and binding was terminated by filtration through these filters using a 48-well cell harvester and washing with $3 \times 4 \mathrm{ml}$ ice-cold isotonic NaCl solution. Following this, the filters were dried, 3 ml Opti-Fluor ${ }^{\mathrm{TM}}$ (Packard) was added, and the amount of bound radioactivity was determined in a scintillation counter.

The fraction of specifically bound radioligand was always $<10 \%$ of the total amount of radioligand. The binding experiments were performed in duplicate at least three times for each compound.

### 3.3.4. The FLIPR Membrane Potential Blue assay

The functional characterization of the compounds at the $\alpha 3 \beta 4$-HEK293 and $\alpha 4 \beta 2$ HEK293T cell lines in the assay was performed essentially as previously described.[50] The cells were split into poly-D-lysine-coated black 96-well plates with clear bottom (BD Biosciences, Bedford, MA). On the day of the experiment the medium was aspirated, and the cells were washed with $100 \mu \mathrm{l}$ Krebs buffer ( $140 \mathrm{mM} \mathrm{NaCl} / 4.7 \mathrm{mM}$ $\mathrm{KCl} / 2.5 \mathrm{mM} \mathrm{CaCl} 2 / 1.2 \mathrm{mM} \mathrm{MgCl} 2 / 11 \mathrm{mM}$ HEPES/ 10 mM D-glucose, pH 7.4$)$. Then $100 \mu \mathrm{l}$ Krebs buffer supplemented with FMP dye ( $0.5 \mathrm{mg} / \mathrm{ml}$ ) was added to the wells (in the antagonist experiments, various concentrations of the antagonists were dissolved in the buffer). After an incubation at $37^{\circ} \mathrm{C}$ in humidified $5 \% \mathrm{CO}_{2}$ for 30 min , the plate was then assayed in a FlexStation ${ }^{3}$ Benchtop Multi-Mode Microplate Reader (Molecular Devices, Sunnyvale, CA) measuring emission at 565 nm [in fluorescence units (FU)] caused by excitation at 525 nm 20 sec before and 70 sec after addition of ligand solution. In the antagonist experiments, $\mathrm{EC}_{70}-\mathrm{EC}_{80} \mathrm{ACh}$ was used as the final agonist concentration. Experiments were performed in duplicate at least three times for each of the receptors.

### 3.3.5. Data Analysis

Data from the binding experiments were fitted to the equation \% Bound $=100 \%$ Bound $/\left(1+([\mathrm{L}] / \mathrm{IC} 50)^{\mathrm{n}}\right)$. Since the concentrations of $\left[{ }^{3} \mathrm{H}\right]$ epibatidine and $\left[{ }^{3} \mathrm{H}\right]$ GR65630 used in the binding experiments were lower than the $\mathrm{K}_{\mathrm{D}}$ values determined for the radioligands the respective receptors, it was deduced from the Cheng-Prusoff equation
[51] that the $K_{i}$ values for the compounds were similar to the obtained $\mathrm{IC}_{50}$ values. The concentration-response curves for agonists and concentration-inhibition curves for antagonists in the FLIPR Membrane Potential Blue assay were constructed based on the differences in the fluorescence units ( $\triangle \mathrm{FU}$ ) between the maximal fluorescence levels recorded before and after addition of eight different concentrations of the various ligands.

### 3.3.6. Two-electrode voltage clamp

Whole-cell currents were measured using a two-electrode voltage clamp with a Geneclamp 500B amplifier together with a Powerlab/200 (AD Instruments, Sydney, Australia) and Chart version 3.5 for PC as previously described.[52] In brief, lobes from ovaries of
female adult Xenopus laevis were removed and defolliculated to obtain isolated oocytes. Oocytes were injected with a total of ;25 ng of cRNA encoding rat $\alpha 4$ and $\beta 2$, in the ratios $1: 4$ and $10: 1$, respectively, and were then incubated $15-18^{\circ} \mathrm{C}$. Recording microelectrodes were filled with 3 M KCl and had resistance between 0.2 and $1 \mathrm{M} \Omega$. Two to 5 days post-injection, oocytes held at -60 mV were used for recording. To prevent the activation of $\mathrm{Ca}^{2+}$ activated chloride currents, oocytes were superfused with calcium-free buffer ( $115 \mathrm{mM} \mathrm{NaCl}, 2.5 \mathrm{mM} \mathrm{KCl}, 1.8 \mathrm{mM} \mathrm{BaCl} 2,10 \mathrm{mM}$ HEPES, pH 7.4) while recording until a stable base current was reached. Concentration-response curves for $\mathbf{3 0}$ were constructed by measuring the peak current responses elicited from a range of agonist concentrations ( $10 \mathrm{nM}-10 \mathrm{mM}$ ). Responses were normalized to the maximum ACh currents (I/Imax) to compare data from different oocytes. Oocytes were washed for 10 min between agonist applications the $\alpha 4 \beta 2$ to ensure receptors were not desensitized from the previous agonist application.

### 3.3.7. SPR biosensor analyses

SPR biosensor interaction experiments were performed on a Biacore T200 apparatus (GE Healthcare, General Electric Co.) as described previously.[27] Briefly, Ac-AChBP and $L s$-AChBP in Acetate 5.0 solution ( $10 \mathrm{mM} \mathrm{NaAc} ,\mathrm{pH} \mathrm{5.0} ,\mathrm{Biacore} \mathrm{AB}, \mathrm{Sweden)}$ were immobilized on separate flow channels on a CM5 sensor chip (Series S, GE Healthcare Bio-Sciences AB, Sweden) using the Amine Coupling Kit (Biacore AB, Sweden) to give $\sim 10000$ response units (RUs). This step was followed by injection of ethanolamine $\mathrm{HCl}-\mathrm{NaOH}$ ( 1 M solution, pH 8.5 , Biacore AB , Sweden) to give a final immobilization of $\sim 3000 \mathrm{RU}$ for Ac -AChBP and $\sim 5000 \mathrm{RU}$ for $L s$-AChBP. On a separate flow channel, a reference surface was prepared using the same conditions and its signals were subtracted from the other channels during analysis. The signals obtained by blank injections were used to correct analyte curves, and solvent correction curves were recorded to assess solvent effects. Freshly prepared running buffer (phosphate buffered saline (PBS (1X) solution; 5\% DMSO; 0.005\% Surfactant P20 (GE Healthcare, Bio-Sciences AB) in deionized, filtered and degassed ultrapure water) was used in all experiments and solutions. Compounds were diluted using a 4 -fold dilution scheme to six concentrations. Compounds were injected for 60 s and allowed to dissociate for 300 s where after regeneration ( $10 \%$ DMSO in running buffer) and wash (50\% DMSO in running buffer) steps were run. Sensorgrams were analyzed with Biacore T200 Evaluation Software v1.0 (GE Healtcare Bio-Sciences AB).

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## ASSOCIATED CONTENT

Supporting Information. Results from elemental analysis of all target compounds, ${ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}$ NMR of representative compounds and co-crystal structures of $L s$-AChBP with $\mathbf{4}$ and $\mathbf{6}$.

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#### Abstract

Abbreviations Ac, Aplysia californica; ACh, acetylcholine; AChBP, acetylcholine binding protein; CNS, central nervous system; CDI, 1,1'-carbonyldiimidazole; Ct, Capitella teleta; DMABC, 3(dimethylamino)butyl dimethylcarbamate; Ls, Lymnaea stagnalis; mAChRs, muscarinic acetylcholine receptors; nAChRs, nicotinic acetylcholine receptors; PNS, peripheral nervous system; SARs, structure-activity relationships; SPR, surface plasmon resonance; THP, tetrahydropyranyl; DCVC, dry column vacuum chromatography; FLIPR, fluorescent imaging plate reader; FMP, FLIPR membrane potential; FCC, flash chromatography; TEVC, two-electrode-voltage-clamp.


Figure 1. Chemical structures of ACh (1), Lobeline (2), 3-6 and general structures of the novel analogs of 4 (7-27) and 6 (28-36). (1 column)

Figure 2. AChBP co-crystal structures with (S)-nicotine and elongated ligands $\mathbf{2}$ and $\mathbf{3}$ displayed with a calculated molecular interaction field (upper row) and surfaces (lower row) to illustrate size and shape of the cavities. A) and D) $L s$-AChBP co-crystallised with ( $S$ )-nicotine (green carbons) [24]; B) and E) Ac-AChBP co-crystallised with 2 (orange carbons) [17]; C) and F) $A c-A C h B P$ cocrystallised with $\mathbf{3}$ (deep purple carbons) [27]. Residues constituting the aromatic box are shown as sticks. Molecular interaction fields (C3 probe) is shown as a grey mesh contoured at $-1 \mathrm{kcal} / \mathrm{mol}$. Surface representation is shown in grey. Residues are named in accordance with Dougherty et al [21] and the amino acid numbering for $L s$-AChBP (1UW6), $A c$-AChBP (2BYS), $A c$-AChBP (2Y58) and $\alpha 4 \beta 2$ (according to Harpsøe et al [15]) corresponds to TyrA1: 185/188/186/223, TyrA2: 192/195/193/230, TrpB: 143/147/145/182, TyrC: 89/93/91/126 and Trp/TyrD: 53/55/53/82 respectively. (2 columns)

Figure 3. Overlay of co-crystal structures of agonists 4 and 6 on co-crystal structures with elongated ligands $\mathbf{2}$ and 3. Overlay of X-ray structures with A) $\mathbf{4}$ (grey carbons, PDB: 3ZDG) and $\mathbf{2}$ (orange carbons, PDB: 2BYS). The overlay indicates that introduction of substituents on the amine of $\mathbf{4}$ could lead to analogs that would address the same cavity as 2. B) Overlay of $\mathbf{4}$ and $\mathbf{3}$ (deep purple carbons, PDB: 2Y58). The overlay indicates that introduction of substituents in the C-3position of $\mathbf{4}$ could lead to analogs that would protrude along the interfacial cleft. C) Overlay of $\mathbf{6}$ (cyan carbons, PDB: 3U8J) and 2; D) 6 and 3. The overlays in C and D indicate that introduction of substituents on the N-4 of the 1,4-diazepane ring of $\mathbf{6}$ potentially could address both cavities. LsAChBP structures are shown in slate/wheat and $A c$ - AChBP in cyan/olive for the principal and complementary subunit, respectively. ( 1.5 column)

Figure 4. Presentation of suggested binding mode of $(R)$ - $\mathbf{1 5}$ (A) and $\mathbf{3 2}$ (B) docked into a homology model of the $\alpha 4 \beta 2$ nAChR.. A) ( $R$ )-15 (light magenta carbons) docked into the model; B) $\mathbf{3 2}$ (green carbons) docked into the model. Residues surrounding the ligand binding site from the principal side (density carbons) and complementary side (orange carbons) are shown. Molecular interaction (C3 probe) is shown at an isocontour level of $-1 \mathrm{kcal} / \mathrm{mol}$. Residue naming is according to Dougherty et al.[21] and the amino acid numbering for $\alpha 4 \beta 2$ (according to Harpsøe et al.[15]) corresponds to TyrA1: 223, TyrA2: 230, TrpB: 182, TyrC126 and Trp/TyrD: 82. (1.5 column)

Figure 5. Agonist-evoked responses for compound 30 at $\alpha 4 \beta 2$ nAChRs expressed in Xenopus laevis oocyte measured by two-electrode voltage-clamp electrophysiology. To yield uniform populations of receptors in $(\alpha 4)_{2}(\beta 2)_{3}(\bullet)$ and $(\alpha 4)_{3}(\beta 2)_{2}(■)$ stichiometries, oocytes were injected with $\alpha 4$ and $\beta 2$ cRNA in a 1:4 and 10:1, respectively. Peak current amplitudes were recorded,
background subtracted and normalized to the amplitude evoked by $300 \mu \mathrm{M}$ at $(\alpha 4)_{2}(\beta 2)_{3}$ and 1 mM ACh at $(\alpha 4)_{3}(\beta 2)_{2}$ (ACh_control) on the same oocyte. Calculated percentage activation were depicted as a function of the concentration $\pm$ S.E. and fitted to a monophasic Hill equation. Data points are from $n=3-6$ oocytes. Regression results are presented in Table 4. (1 column)

Scheme 1. Synthesis of $N$-arylalkyl analogs of 4 (7-12). (a) methyl acetoacetate, AcOH , toluene, rt; (b) sodium, $i$-PrOH, THF, $0{ }^{\circ} \mathrm{C}$ then rt; (c) (1) CDI, toluene, rt, (2) THF, rt, (3) 9 and 10: $\mathrm{CH}_{3} \mathrm{NH}_{2}$ ( $40 \%$ in $\mathrm{H}_{2} \mathrm{O}$ ), 37: $\left(\mathrm{CH}_{3}\right)_{2} \mathrm{NH}(33 \%$ in EtOH$)$, rt ; (d) $\mathrm{Pd} / \mathrm{C}(5 \%(\mathrm{w} / \mathrm{w})), \mathrm{H}_{2}, \mathrm{AcOH}$, rt; (e) 7: 2benzyloxirane, 11: styrene oxide, $\mathrm{LiClO}_{4}$, MeCN , reflux; (f) 3-phenylpropionaldehyde, $\mathrm{NaBH}_{3} \mathrm{CN}$, DCE, MeOH , rt.

Scheme 2. Synthesis of C-3-arylalkyl analogs of $\mathbf{4}\left(\mathbf{1 3 - 1 5}, \mathbf{2 2 - 2 5}\right.$, 27) (a) $\mathrm{NH}\left(\mathrm{CH}_{3}\right)_{2}$ (33\% in EtOH ); (b) $\mathrm{LiAlH}_{4}, \mathrm{Et}_{2} \mathrm{O}, 0^{\circ} \mathrm{C}$ then rt; (c) (1) CDI, toluene, rt, (2) THF, rt, (3) 22-25: $\mathrm{NH}_{2} \mathrm{CH}_{3}$ ( $40 \%$ in $\mathrm{H}_{2} \mathrm{O}$ ), 13-15: $\mathrm{NH}\left(\mathrm{CH}_{3}\right)_{2}$ ( $33 \%$ in EtOH ), rt; (d) (1) NaH ( $60 \%$ in mineral oil), $2,4,5-$ tribromo-1-methyl- $1 H$-imidazole, THF, reflux, (2) $n$ - BuLi , THF, $-78{ }^{\circ} \mathrm{C}$, (3) $\mathrm{H}_{2} \mathrm{O},-78{ }^{\circ} \mathrm{C}$. ( 1 column)

Scheme 3. Synthesis of C-3-arylalkyl analogs of 4 (16-21, 26) (a) 17: iodobenzene, 18: 1-fluoro-4-iodo-benzene, CuI, 3,4,7,8-tetramethyl-1,10-phenanthroline, $\mathrm{Cs}_{2} \mathrm{CO}_{3}$, toluene, reflux; (b) (1) NaH (60\% in mineral oil), DMF, $0^{\circ} \mathrm{C}$, (2) 19: 2-fluoropyridine, 20: benzylbromide, 21: 2-bromothiazole, rt ; (c) $\mathrm{NH}\left(\mathrm{CH}_{3}\right)_{2}(33 \%$ in EtOH$)$; (d) 3,4-dihydro-2H-pyran, concentrated HCl , rt; (e) $\mathrm{LiAlH}_{4}$, $\mathrm{Et}_{2} \mathrm{O}, 0{ }^{\circ} \mathrm{C}$ then rt; (f) (1) CDI, toluene, rt, (2) THF, rt, (3) 22-25: $\mathrm{NH}_{2} \mathrm{CH}_{3}\left(40 \%\right.$ in $\left.\mathrm{H}_{2} \mathrm{O}\right), \mathbf{1 3 - 1 5}$ : $\mathrm{NH}\left(\mathrm{CH}_{3}\right)_{2}(33 \%$ in EtOH$)$, rt ; (g) $p$-toluenesulfonic acid monohydrate, MeOH , rt. (1 column)

Scheme 4. Synthesis of $N$-arylalkyl 3-pyridyl diazacycle analogs 28-36. (a) NBS, MeCN, rt; (b) TFA, DCM, rt; (c) 28, 32, and 36: hydrocinnamaldehyde, 30 and 34: benzaldehyde, 31 and 35: phenylacetaldehyde, 33: 2-(3-bromophenyl)acetaldehyde, 29: cinnamaldehyde, $\mathrm{NaBH}(\mathrm{OAc})_{3}, \mathrm{DCE}$, rt. (1 column)

Table 1. Binding characteristics of $\mathbf{7 - 2 7}$ and reference compounds $\mathbf{1 , 2}, \mathbf{4}$ and $\mathbf{5}$ to rat $\alpha 4 \beta 2, \alpha 4 \beta 4$, and $\alpha 3 \beta 4$ nAChRs stably expressed in HEK293 cells in a $\left[{ }^{3} \mathrm{H}\right]$-epibatidine binding assay. ${ }^{\text {a }}$ (2 columns)



${ }^{\text {a }}$ Binding affinities are given as $K_{i}$ in $\mu \mathrm{M}$ with $\mathrm{p} K_{i} \pm$ SEM values in brackets based on three experiments for each compound. ${ }^{\mathrm{b}} K_{i}$ values were determined in a $\left[{ }^{3} \mathrm{H}\right]$ epibatidine binding assay with HEK293 cells stably expressing rat $\alpha 4 \beta 2, \alpha 4 \beta 4$, and $\alpha 3 \beta 4$ as previously described.[32] ${ }^{\text {c }}$ Binding data from Jensen et al.[41] ${ }^{\text {d Binding data from Hansen et }}$ al.[31] ${ }^{\mathrm{e}}$ Binding data from Rohde et al.[36]

Table 2. Binding characteristics of 28-36 and reference compound $\mathbf{6}$ to rat $\alpha 4 \beta 2, \alpha 4 \beta 4$, and $\alpha 3 \beta 4$ nAChRs stably expressed in HEK293 cells in a $\left[{ }^{3} \mathrm{H}\right]$-epibatidine binding assay. ${ }^{\text {a }}$ (1 column)

|  <br> 28 |   <br> 33 <br> 34 |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | Competition | nding, ${ }^{\text {b }} K_{i}$ ( $\mu \mathrm{M}$ ) |  | Selectivity ratio |
| Compd | $\mathrm{R}_{1}$ | $\mathrm{R}_{2}$ | $\alpha 4 \beta 2$ | $\alpha 4 \beta 4$ | $\alpha 3 \beta 4$ | $\alpha 3 \beta 4 / \alpha 4 \beta 2$ |
| $6^{\text {e }}$ | - ( | - | $\begin{aligned} & 0.00072 \\ & {[9.14 \pm 0.09]} \end{aligned}$ | n.d. | n.d. |  |
| 28 | phenylpropyl | Br | $\begin{aligned} & 0.82 \\ & {[6.08 \pm 0.07]} \end{aligned}$ | $\begin{aligned} & 1.5 \\ & {[5.82 \pm 0.05]} \end{aligned}$ | $\begin{aligned} & 2.2 \\ & {[5.66 \pm 0.02]} \end{aligned}$ | 3 |
| 29 | cinnamyl | H | $\begin{aligned} & 0.099 \\ & {[7.00 \pm 0.12]} \end{aligned}$ | $\begin{aligned} & 1.6 \\ & {[5.79 \pm 0.13]} \end{aligned}$ | $\begin{aligned} & 2.9 \\ & {[5.53 \pm 0.07]} \end{aligned}$ | 29 |
| 30 | benzyl | H | $\begin{aligned} & 0.104 \\ & {[6.98 \pm 0.06]} \end{aligned}$ | $\begin{aligned} & 3.4 \\ & {[5.47 \pm 0.03]} \end{aligned}$ | $\begin{aligned} & 2.6 \\ & {[5.59 \pm 0.03]} \end{aligned}$ | 25 |
| 31 | phenethyl | H | $\begin{aligned} & 0.11 \\ & {[6.98 \pm 0.03]} \end{aligned}$ | $\begin{aligned} & 2.4 \\ & {[5.63 \pm 0.05]} \end{aligned}$ | $\begin{aligned} & 3.4 \\ & {[5.47 \pm 0.03]} \end{aligned}$ | 31 |
| 32 | phenylpropyl | H | $\begin{aligned} & 0.082 \\ & {[7.09 \pm 0.02]} \end{aligned}$ | $\begin{aligned} & 3.8 \\ & {[5.42 \pm 0.07]} \end{aligned}$ | $\begin{aligned} & 3.8 \\ & {[5.42 \pm 0.07]} \end{aligned}$ | 46 |
| 33 | (3-bromo-phenyl)ethyl | H | $\begin{aligned} & 0.13 \\ & {[6.88 \pm 0.06]} \end{aligned}$ | $\begin{aligned} & 1.9 \\ & {[5.72 \pm 0.05]} \end{aligned}$ | $\begin{aligned} & 0.78 \\ & {[6.11 \pm 0.05]} \end{aligned}$ | 6 |


| 34 | benzyl | - | 9.6 | $\sim 100$ | $>300$ | $>31$ |
| :--- | :--- | :--- | :--- | :--- | :--- | :---: |
|  |  |  | $[5.02 \pm 0.02]$ | $[\sim 4.0]$ | $[<3.5]$ |  |
| $\mathbf{3 5}$ | phenethyl | - | 0.56 | 8.9 | 8.2 | 15 |
|  |  |  | $[6.25 \pm 0.02]$ | $[5.05 \pm 0.05]$ | $[5.08 \pm 0.04]$ |  |
| $\mathbf{3 6}$ | phenylpropyl | - | 0.49 | 3.5 | $\sim 20$ | $\sim 41$ |
|  |  |  | $[6.30 \pm 0.09]$ | $[5.46 \pm 0.06]$ | $[\sim 4.7]$ |  |

${ }^{\text {a }}$ Binding affinities are given as $K_{i}$ in $\mu \mathrm{M}$ with $\mathrm{p} K_{i} \pm$ SEM values in brackets based on three experiments for each compound. ${ }^{\mathrm{b}} K_{i}$ values were determined in a $\left[{ }^{3} \mathrm{H}\right]$ epibatidine binding assay with HEK293 cells stably expressing rat $\alpha 4 \beta 2, \alpha 4 \beta 4$, and $\alpha 3 \beta 4$ as previously described.[32] ${ }^{\text {c }}$ Binding data from Jensen et al.[41] ${ }^{\text {d }}$ Binding data from Hansen et al.[31] ${ }^{\mathrm{e}}$ Binding data from Rohde et al.[36]

Table 3. Functional properties of target compounds 13-15, 23-25, 27, 29, 35, and 36 at mouse $\alpha 4 \beta 2$-HEK293T and rat $\alpha 3 \beta 4$-HEK293 HEK293 cell lines in the FLIPR Membrane Potential Blue assay. (1 column)

| $\alpha 4 \beta 2^{\text {a }}$ |  |  | $\alpha 3 \beta 4^{\text {a }}$ |  |
| :---: | :---: | :---: | :---: | :---: |
| Compound | $\mathrm{EC}_{50}(\mu \mathrm{M})$ | $\mathrm{IC}_{50} \quad(\mu \mathrm{M})$ | $\mathrm{EC}_{50} \quad(\mu \mathrm{M})$ | $\mathrm{IC}_{50} \quad(\mu \mathrm{M})$ |
|  | $\left[\mathrm{pEC}_{50} \pm\right.$ S.E.M.] | [pIC ${ }_{50} \pm$ S.E.M.] | $\left[\mathrm{pEC}_{50} \pm\right.$ S.E.M. $]$ | $\left[\mathrm{pIC} \mathrm{S}_{50} \pm\right.$ S.E.M. $]$ |
| 13 | Agonist ${ }^{\text {b }}$ |  | Agonist ${ }^{\text {c }}$ | - |
| 14 | Agonist ${ }^{\text {b }}$ |  | Agonist ${ }^{\text {c }}$ | - |
| 15 | Agonist ${ }^{\text {d }}$ | - | Agonist ${ }^{\text {e }}$ | - |
| 23 | Agonist ${ }^{\text {b }}$ | - | Agonist ${ }^{\text {c }}$ | - |
| 24 | Agonist ${ }^{\text {b }}$ | - | Agonist ${ }^{\text {c }}$ | - |
| 27 | Agonist ${ }^{\text {d }}$ | - | Agonist ${ }^{\text {e }}$ | - |
| 29 |  | $1.1[5.96 \pm 0.09]$ | - | Weak antagonist ${ }^{\text {f }}$ |
| 35 | Agonist ${ }^{\text {g }}$ | - | Agonist ${ }^{\text {g }}$ | - |
| 36 | Agonist ${ }^{\text {e }}$ | - | Agonist ${ }^{\text {c }}$ | - |

${ }^{\mathrm{a}} \mathrm{EC}_{50}$ and $\mathrm{IC}_{50}$ values are given in $\mu \mathrm{M}$ with $\mathrm{pEC}_{50} \pm \mathrm{SEM}$ and $\mathrm{pIC}_{50} \pm$ SEM values in brackets. The data are the means of three individual experiments performed in duplicate. ${ }^{\text {b,c, } \mathrm{de}, \mathrm{f}}$ Complete concentration-response or concentration-
inhibitions curves could not be obtained for the compounds in the tested concentration ranges. ${ }^{\mathrm{b}}$ The compound elicited a significant agonist response at $100 \mu \mathrm{M}$ and higher concentrations. ${ }^{\mathrm{c}}$ The compound elicited a significant agonist response at $300 \mu \mathrm{M} .{ }^{\mathrm{d}}$ The compound elicited a significant agonist response at concentrations of $30 \mu \mathrm{M}$ and higher. ${ }^{\text {e }}$ The compound elicited a significant agonist response at concentrations of $100 \mu \mathrm{M}$ and higher. ${ }^{\mathrm{f}}$ The compound exhibited antagonist activity at concentrations of $100 \mu \mathrm{M}$ and higher.

Table 4. Functional activity of 30 in two-electrode voltage clamp at $(\alpha 4)_{2}(\beta 2)_{3}$ and $(\alpha 4)_{3}(\beta 2)_{2}$ nAChRs expressed in Xenopus oocytes. ${ }^{\text {a }}$ ( 1 column)

|  | Activity $(\alpha 4)_{2}(\beta 2)_{3}$ |  | Activity $(\alpha 4)_{3}(\beta 2)_{2}$ |  |
| :--- | :--- | :--- | :--- | :--- |
| Compound | $\mathrm{EC}_{50}(\mu \mathrm{M})$ | $\mathrm{R}_{\text {max }}$ | $\mathrm{EC}_{50}(\mu \mathrm{M})$ | $\mathrm{R}_{\text {max }}$ |
|  | $\left[\mathrm{pEC}_{50} \pm \mathrm{SEM}\right]$ |  |  |  |
| $\mathbf{3 0}$ | $1.8[5.75 \pm 0.07]$ | $16 \pm 0.8$ | $7.9[5.1 \pm 0.20]$ | $6 \pm 0.7$ |

${ }^{\text {a }}$ The $(\alpha 4)_{2}(\beta 2)_{3}$ receptors were expressed by a 1:4 ratio of $\alpha: \beta$. The $(\alpha 4)_{3}(\beta 2)_{2}$ receptors were expressed by a $10: 1$ ratio of $\alpha: \beta$.

Table 5. Dissociation constants of $\mathbf{3 0}$ and $\mathbf{3 2}$ and reference compounds ( $S$ )-nicotine and 2. ${ }^{\text {a }}$ (1 column)

|  | $K_{D}(\mu \mathrm{M})^{\mathrm{a}}$ | $K_{i}(\mu \mathrm{M})^{\mathrm{b}}$ |  |
| :--- | :--- | :--- | :--- |
| Compound | $L s$-AChBP | $A c$-AChBP | $\alpha 4 \beta 2$ |
| $(S)$-Nicotine | $0.32[6.5 \pm 0.2]$ | $3.16[5.5 \pm 0.1]$ | $0.0076[8.12 \pm 0.05]^{\mathrm{c}}$ |
| $\mathbf{2}$ | $0.63[6.2 \pm 0.1]$ | $0.013[7.9 \pm 0.1]$ | $0.0050[8.3 \pm 0.1]^{\mathrm{d}}$ |
| $\mathbf{3 0}$ | $0.16[6.8 \pm 0.1]$ | $>1.0[<6]^{\mathrm{e}}$ | $0.104[6.98 \pm 0.06]$ |

$$
0.020[7.7 \pm 0.1] \quad>1.0[<6]^{\mathrm{e}}
$$

$$
0.082[7.09 \pm 0.02]
$$

${ }^{a}$ SPR binding affinity values for $L s$ - and $A c$-AChBP represented as $K_{\mathrm{D}}(\mu \mathrm{M})$ with $\mathrm{p} K_{\mathrm{D}} \pm$ SEM values shown in brackets, determined by steady-state affinity analysis as previously described.[27] ${ }^{\mathrm{b}}$ Receptor binding affinity values for $\alpha 4 \beta 2$ represented as $K_{i}(\mu \mathrm{M})$ with $\mathrm{p} K_{i} \pm$ SEM values shown in brackets, determined in a $\left.{ }^{3} \mathrm{H}\right]$ epibatidine binding assay with HEK293 cells stably expressing rat $\alpha 4 \beta 2$ as previously described.[32] ${ }^{\mathrm{c}}$ Data from Ussing et al.[35] ${ }^{\mathrm{d}}$ Data from Edink et al.[27] ${ }^{\mathrm{e}}$ Binding isotherm did not reach plateau within the tested range.

## Figure 1






4,(DMABC) $\mathrm{R}=\mathrm{CH}_{3}$ 5, $R=H$


1-(Pyridin-3-yl)-1,4-diazepane (6)


13-26


28-33



27


34-36

Figure 2


Figure 3


Figure 4


Figure 5


## Scheme 1




9: $n=1, R=H$
40: $n=1$
10: $n=2, R=H$


41: $n=2$


42: $R=H$
7: $\mathrm{n}=1, \mathrm{R}=\mathrm{CH}_{3}$
11: $n=0, R=H$
12: $n=3, R=H$


8: $n=3, R=\mathrm{CH}_{3}$

Scheme 2


Scheme 3


60: $R_{3}=H$
61: $R_{3}=\mathrm{CH}_{3}$
26: $\mathrm{R}_{3}=\mathrm{H}$

Scheme 4


## Highlights:

- Elongated $\alpha 4 \beta 2$ nAChR ligands have retained partial agonist profile
- SAR indicate classical binding mode for the core agonist part of the molecule
- Substituents bind along the subunit interface
- Ligands showing preference for $L s-\mathrm{AChBP}$ over $A c-\mathrm{AChBP}$


## Supporting Information

## Exploration of the Molecular Architecture of the Orthosteric

## Binding Site in the $\alpha 4 \beta 2$ Nicotinic Acetylcholine Receptor with

 Analogs of 3-(Dimethylamino)butyl Dimethylcarbamate(DMABC) and 1-(Pyridin-3-yl)-1,4-diazepane

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Figure S1: Co-crystal structures of $L s$-AChBP with $\mathbf{4}^{1}$ and $\mathbf{6}^{2}$
Elemental analysis data
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Figure S1. Co-crystal structures of $L s$ - AChBP with $\mathbf{4}^{1}$ and $\mathbf{6}^{2}$. ${ }^{\text {. }}$

${ }^{\text {a }}$ Residues at the principal and complementary side are colored blue and orange, respectively.
Dashed lines indicate short contact distances $(<3.5 \AA$ ) representing ionic or electrostatic interactions between ligands and the protein or a water molecule (red sphere). A) 4 (wheat carbons) in complex with $L s$ - AChBP . A polarised hydrogen on one of the methyl groups on the protonated nitrogen forms a direct interaction to Tyr89. B) $\mathbf{6}$ (pale green) in complex with $L s$-AChBP. $\mathbf{6}$ forms a direct hydrogen bond to Tyr89.

## Elemental analysis data

| Compound |  | Elemental analysis calculated <br> (found) |  | Deviation |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | C | H | N |  |
| 7 | $\mathrm{C}_{17} \mathrm{H}_{27} \mathrm{~N}_{2} \mathrm{O}_{3} \cdot \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4}$ | 57.42 | 7.35 | 7.05 | $<0.4$ |
|  |  | $(57.14)$ | $(7.66)$ | $(6.97)$ |  |


| 8 | $\mathrm{C}_{17} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}_{2} \cdot \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4}$ | $\begin{gathered} 59.67 \\ (59.55) \end{gathered}$ | $\begin{gathered} 7.91 \\ (7.56) \end{gathered}$ | $\begin{aligned} & 7.32 \\ & (7.32) \end{aligned}$ | $<0.4$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 9 | $\mathrm{C}_{14} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{2} \cdot \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4}$ | $\begin{gathered} 56.46 \\ (56.49) \end{gathered}$ | $\begin{gathered} 7.11 \\ (6.98) \end{gathered}$ | $\begin{gathered} 8.23 \\ (8.13) \end{gathered}$ | $<0.4$ |
| 10 | $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{2} \cdot \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4}$ | $\begin{gathered} \hline 57.61 \\ (57.28) \end{gathered}$ | $\begin{gathered} 7.39 \\ (7.26) \end{gathered}$ | $\begin{gathered} 7.90 \\ (8.02) \end{gathered}$ | $<0.4$ |
| 11 | $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{3} \cdot 1.1 \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4}$ | $\begin{gathered} 54.45 \\ (54.26) \end{gathered}$ | $\begin{gathered} 6.96 \\ (6.92) \end{gathered}$ | $\begin{aligned} & 7.38 \\ & (7.08) \end{aligned}$ | $<0.4$ |
| 12 | $\mathrm{C}_{16} \mathrm{H}_{26} \mathrm{~N}_{2} \mathrm{O}_{2} \cdot 1.1 \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4}$ | $\begin{gathered} 57.92 \\ (57.62) \end{gathered}$ | $\begin{gathered} 7.53 \\ (6.98) \end{gathered}$ | $\begin{aligned} & 7.42 \\ & (7.94) \end{aligned}$ | 0.55 |
| 13 | $\mathrm{C}_{17} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}_{2} \cdot \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4}$ | 59.67 <br> (59.68) | $\begin{gathered} 7.91 \\ (7.72) \end{gathered}$ | $\begin{gathered} 7.32 \\ (7.37) \end{gathered}$ | $<0.4$ |
| 14 | $\mathrm{C}_{18} \mathrm{H}_{30} \mathrm{~N}_{2} \mathrm{O}_{2} \cdot \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4}$ | $\begin{gathered} 60.59 \\ (60.53) \end{gathered}$ | $\begin{gathered} 8.14 \\ (7.92) \end{gathered}$ | $\begin{gathered} 7.07 \\ (7.06) \end{gathered}$ | $<0.4$ |
| 15 | $\mathrm{C}_{19} \mathrm{H}_{32} \mathrm{~N}_{2} \mathrm{O}_{2} \cdot \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4} \cdot 0 \cdot 1 \mathrm{H}_{2} \mathrm{O}$ | $\begin{gathered} 61.17 \\ (60.92) \end{gathered}$ | $\begin{gathered} 8.36 \\ (8.11) \end{gathered}$ | $\begin{aligned} & 6.79 \\ & (6.88) \end{aligned}$ | $<0.4$ |
| 16 | $\mathrm{C}_{17} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}_{3} \cdot \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4}$ | $\begin{gathered} 57.27 \\ (57.15) \end{gathered}$ | $\begin{gathered} 7.59 \\ (7.38) \end{gathered}$ | $\begin{gathered} 7.03 \\ (7.03) \end{gathered}$ | $<0.4$ |
| 17 | $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{3} \cdot \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4}$ | $\begin{gathered} 55.13 \\ (54.86) \end{gathered}$ | $\begin{aligned} & \hline 7.08 \\ & (6.78) \end{aligned}$ | $\begin{gathered} 7.56 \\ (7.28) \end{gathered}$ | $<0.4$ |
| 18 | $\mathrm{C}_{15} \mathrm{H}_{23} \mathrm{FN}_{2} \mathrm{O}_{3} \cdot \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4}$ | 52.57 | 6.49 | 7.21 | $<0.4$ |


|  |  | (52.40) | (6.59) | (7.11) |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 19 | $\mathrm{C}_{14} \mathrm{H}_{23} \mathrm{~N}_{3} \mathrm{O}_{3} \cdot 1.1 \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4}$ | $\begin{aligned} & \hline 51.15 \\ & (51.08) \end{aligned}$ | $\begin{aligned} & 6.68 \\ & (6.79) \end{aligned}$ | $\begin{aligned} & 11.05 \\ & (11.05) \end{aligned}$ | <0.4 |
| 20 | $\mathrm{C}_{16} \mathrm{H}_{26} \mathrm{~N}_{2} \mathrm{O}_{3} \cdot \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4}$ | $\begin{aligned} & 56.24 \\ & (56.30) \end{aligned}$ | $7.34$ <br> (7.37) | $\begin{gathered} 7.29 \\ (7.36) \end{gathered}$ | $<0.4$ |
| 21 | $\mathrm{C}_{12} \mathrm{H}_{21} \mathrm{~N}_{3} \mathrm{O}_{3} \mathrm{~S} \cdot 1.2 \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4} \cdot 0.1 \mathrm{H}_{2} \mathrm{O}$ | $\begin{gathered} 43.54 \\ (43.60) \end{gathered}$ | $\begin{aligned} & 5.99 \\ & (6.08) \end{aligned}$ | $\begin{aligned} & 10.58 \\ & (10.53) \end{aligned}$ | $<0.4$ |
| 22 | $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{2} \cdot \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4} \cdot 0.1 \mathrm{C}_{3} \mathrm{H}_{6} \mathrm{O}$ | $\begin{aligned} & 57.69 \\ & (57.91) \end{aligned}$ | $7.44$ (7.24) | 7.78 <br> (7.75) | $<0.4$ |
| 23 | $\mathrm{C}_{16} \mathrm{H}_{26} \mathrm{~N}_{2} \mathrm{O}_{2} \cdot \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4}$ | $\begin{aligned} & 58.68 \\ & (58.89) \end{aligned}$ | $7.66$ <br> (7.52) | $\begin{aligned} & 7.60 \\ & (7.62) \end{aligned}$ | $<0.4$ |
| 24 | $\mathrm{C}_{17} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}_{2} \cdot \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4} \cdot 0.2 \mathrm{H}_{2} \mathrm{O}$ | $\begin{gathered} 59.11 \\ (58.77) \end{gathered}$ | $\begin{gathered} 7.94 \\ (7.59) \end{gathered}$ | $\begin{aligned} & 7.26 \\ & (7.60) \end{aligned}$ | $<0.4$ |
| 25 | $\mathrm{C}_{18} \mathrm{H}_{30} \mathrm{~N}_{2} \mathrm{O}_{2} \cdot \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4}$ | $\begin{gathered} 60.59 \\ (60.54) \end{gathered}$ | $\begin{aligned} & 8.14 \\ & (8.04) \end{aligned}$ | $\begin{gathered} 7.07 \\ (7.00) \end{gathered}$ | $<0.4$ |
| 26 | $\mathrm{C}_{16} \mathrm{H}_{26} \mathrm{~N}_{2} \mathrm{O}_{3} \cdot \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4}$ | $\begin{gathered} 56.24 \\ (55.87) \end{gathered}$ | $7.34$ <br> (7.10) | $\begin{gathered} 7.29 \\ (7.25) \end{gathered}$ | $<0.4$ |
| 28 | $\mathrm{C}_{19} \mathrm{H}_{24} \mathrm{BrN}_{3} \cdot 1 \cdot 4 \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4} \cdot 0.2 \mathrm{C}_{3} \mathrm{H}_{6} \mathrm{O}$ | 52.55 <br> (52.77) | $\begin{gathered} 5.51 \\ (5.29) \end{gathered}$ | $\begin{aligned} & 8.21 \\ & (7.97) \end{aligned}$ | $<0.4$ |
| 29 | $\mathrm{C}_{19} \mathrm{H}_{23} \mathrm{~N}_{3} \cdot 2.5 \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4} \cdot 0.4 \mathrm{C}_{3} \mathrm{H}_{6} \mathrm{O}$ | $\begin{gathered} 55.87 \\ (55.89) \end{gathered}$ | $\begin{gathered} 5.66 \\ (5.79) \end{gathered}$ | $\begin{aligned} & 7.76 \\ & (7.85) \end{aligned}$ | $<0.4$ |


| $\mathbf{3 0}$ | $\mathrm{C}_{17} \mathrm{H}_{21} \mathrm{~N}_{3} \cdot 1.6 \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4} \cdot 0.4 \mathrm{C}_{4} \mathrm{H}_{10} \mathrm{O}$ | 59.36 | 6.44 | 9.53 | $<0.4$ |
| :---: | :--- | :---: | :---: | :---: | :---: |
| $\mathbf{3 1}$ | $\mathrm{C}_{18} \mathrm{H}_{23} \mathrm{~N}_{3} \cdot 1.5 \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4}$ | $(59.66)$ | $(6.39)$ | $(9.75)$ |  |
| $\mathbf{3 2}$ | $\mathrm{C}_{19} \mathrm{H}_{25} \mathrm{~N}_{3} \cdot 1.8 \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4} \cdot 0.1 \mathrm{C}_{3} \mathrm{H}_{6} \mathrm{O}$ | $(60.57)$ | $(6.18)$ | $(10.01)$ |  |
| $\mathbf{3 3}$ | $\mathrm{C}_{18} \mathrm{H}_{22} \mathrm{BrN}_{3} \cdot 2.0 \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4}$ | $(59.39)$ | $(6.51)$ | $(9.16)$ |  |
| $\mathbf{3 4}$ | $\mathrm{C}_{16} \mathrm{H}_{19} \mathrm{~N}_{3} \cdot 2.6 \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4}$ | 48.90 | 4.85 | 7.78 | $<0.4$ |
| $\mathbf{3 5}$ | $\mathrm{C}_{17} \mathrm{H}_{21} \mathrm{~N}_{3} \cdot 2.7 \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4} \cdot 0.1 \mathrm{C}_{3} \mathrm{H}_{6} \mathrm{O}$ | $(48.67)$ | $(4.69)$ | $(7.56)$ |  |
| $\mathbf{3 6}$ | $\mathrm{C}_{18} \mathrm{H}_{23} \mathrm{~N}_{3} \cdot 2.2 \mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{4}$ | 52.24 | 5.00 | 8.62 | $<0.4$ |
|  | $(52.87)$ | $(5.19)$ | $(8.21)$ |  |  |

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