## Journal Pre-proofs

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### **Graphical Abstract**

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# A novel and efficient method for the direct synthesis of pyrrolyl or indolyl substituted 9,10-dihydrophenanthren-9-ol analogues

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#### ARTICLE INFO

#### ABSTRACT

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Keywords: Intramolecular cycloaddtion Ring-opening aromatization Domino reaction 9,10-Dihydrophenanthren-9-ol analogues A novel domino intramolecular [3+2] cycloaddtion and ring-opening aromatization process has been successfully developed for the efficient direct synthesis of pyrrolyl or indolyl substituted 9,10-dihydrophenanthren-9-ol analogues. And 1-(phenanthren-9-yl)-1*H*-pyrroles can be easily obtained *via* dehydration of 10-(1*H*-pyrrol-1-yl)-9,10-dihydrophenanthren-9-ols. Furthermore, the MTT assay indicated that four compounds with indolyl substitutions showed obvious inhibitory activities against HepG2 cells, and compound *anti*-**4jb** displayed a lowest IC<sub>50</sub> value of 9.99  $\mu$ M.

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As an important class of polycyclic compounds, 9,10dihydrophenanthrene derivatives, existing in many synthetic and natural products,<sup>1</sup> have shown various biological activities, such

- 5 as antitumour,<sup>2</sup> antifungal,<sup>3</sup> antimicrobial,<sup>4</sup> antialgal,<sup>5</sup> antimalarial,<sup>6</sup> cytotoxic activities,<sup>7</sup> and so on. They are also useful building blocks in material science.<sup>8</sup> Therefore, the development of efficient methods for the synthesis of 9,10-dihydrophenanthrenes is of high interests. Hall and Turner<sup>9</sup>
- 10 reported the synthesis of 9,10-dihydrophenanthrene through the reaction of 2,2'-di-(bromomethtl)diphenyl with lithium phenyl 55 (eq 1). Suzuki<sup>10</sup> and Uemura<sup>11</sup> independently developed a SmI<sub>2</sub> 55 catalyzed intramolecular pinacol coupling strategy to obtain the *trans*-9,10-dihydrophenanthrene-9,10-diol from 2,2'-dicarbonyl-
- 15 1,1'-biphenyls.(eq 2). You and coworkers<sup>12</sup> modified the process into a one-pot cascade reaction. Sato and coworkers<sup>13</sup> reported a nickel-catalyzed [2+2+2] cycloaddition of arynes and alkenes (eq 3). Ray and coworkers<sup>14</sup> developed a palladium-assisted  $6\pi$ electrocyclic reaction to obtain 9,10-dihydrophenanthrenes (eq
- 20 4). Although the reported methods are somehow effective to the construction of a 9,10-dihydrophenanthrene skeleton, they suffer from some drawbacks, such as the starting materials were not commercially available, transition metal catalysts were unavoidable in many cases, and limited substrate scopes. To the
- 25 best of our knowledge, the direct synthesis of pyrrolyl or indolyl substituted 9,10-dihydrophenanthren-9-ols has not yet been 70 reported. Thus, the development of a novel and efficient approach to the direct construction of pyrrolyl or indolyl substituted 9,10-dihydrophenanthren-9-ol analogues is highly 20 desired.





In our previous work, we reported the synthesis of *N*-βhydroxyethyl pyrroles and indoles from the reaction of aldehyde 35 with *trans*-4-hydroxy-L-proline or indoline-2-carboxylic acid via 80 a domino [3+2] cycloaddition and ring-opening aromatization process.<sup>15</sup> However, the transformation was not applicable to the aromatic aldehydes bearing an electron-donating or a weak electron-withdrawing group due to the low reactivity between 40 aldehydes and azomethineylides in the intermolecular [3+2] cycloaddition. Fortunately, by replacing the two molecules of the aldehyde with a dialdehyde molecule participated in the intramolecular [3+2] cycloaddition, a novel methodology for the direct synthesis of pyrrolyl or indolyl substituted 9,10-

45 dihydrophenanthren-9-ol analogues with wide substituent

diversity was successfully accomplished and reported herein (eq 5).

At the outset, the assumption was tested by reaction of [1,1'biphenyl]-2,2'-dicarbaldehyde 1a with trans-4-hydroxy-L-proline 50 2a in N,N-dimethylformamide (DMF) at 150 °C. Gratifyingly, the expected product 10-(1H-pyrrol-1-yl)-9,10-dihydrophenanthren-9-ol 3aa was obtained in 85% yield (Table 1, entry 1). Inspired by this result, we started to optimize the reaction conditions to improve the reaction efficiency. Among the solvents screened, dimethyl sulfoxide (DMSO) was found to be the most efficient solvent with a desired product yield of 91% (Table 1, entry 2). Xylenes was inferior and produced the product in 40% yield (Table 1, entry 3). Other solvents, such as toluene, 1,2dichloroethane (DCE), acetonitrile, and ethanol, gave only trace 60 of products (Table 1, entries 3-7). Furthermore, no desired product was detected in the solvent of 1,4-dioxane, tetrahedronfuran (THF) and water (Table 1, entries 8-10). By increasing the amount of 2a (0.90 mmol), no considerable improvement was observed (Table 1, entry 11), whereas 65 decreasing the amount of 2a (0.60 mmol) lowered the yield obviously (Table 1, entry 12). In addition, by increasing or decreasing the reaction time did not show any further improvement in the product yield (Table 1, entries 13-14). Further investigation indicated that 140 °C was the optimal reaction temperature at which the product was obtained in a yield of 92% (Table 1, entries 15-17). Thus, the optimized condition was determined as follows: [1,1'-biphenyl]-2,2'-dicarbaldehyde 1a (0.5 mmol), trans-4-hydroxy-L-proline 2a

#### Table 1 Optimization of the reaction conditions<sup>a</sup>

		~				,
	CH 		HO	sol	vent	он
		+			<u>→ /</u>	$\succ$
		ĊHO	N <sup>×</sup> <sup>•</sup> CC	JOH T, I	time	
		1a	2a		\/	
75_		-				3aa
	Entry	2a	Solvent	T (°C)	Time (min)	Yieldb
_		(mmol)				(%)[dr] <sup>c</sup>
	1	0.75	DMF	150	30	85[87:13]
	2	0.75	DMSO	150	30	91[87:13]
	3	0.75	Xylenes	150	30	40[86:14]
	4	0.75	Toluene	150	30	trace
	5	0.75	DCE	150	30	trace
	6	0.75	MeCN	150	30	trace
	7	0.75	EtOH	150	30	trace
	8	0.75	1,4- Dioxane	150	30	0
	9	0.75	THF	150	30	0
	10	0.75	$H_2O$	150	30	0
	11	0.90	DMSO	150	30	90[87:13]
	12	0.60	DMSO	150	30	78[87:13]
	13	0.75	DMSO	150	20	88[87:13]
	14	0.75	DMSO	150	40	84[87:13]
	15	0.75	DMSO	160	30	87[87:13]
	16	0.75	DMSO	140	30	92[87:13]
	17	0.75	DMSO	130	30	72[87:13]

<sup>a</sup> Reaction conditions: [1,1'-biphenyl]-2,2'-dicarbaldehyde**1a**(0.5 mmol) and*trans*-4-hydroxy-L-proline in 1.5 mL solvent were stirred in a 10 mL-glass tube sealed with a cap in an oil bath at indicated temperature for 20-40 min.

<sup>b</sup> Isolated yields based on **1a**.

c Determined by <sup>1</sup>H-NMR.

(0.75 mmol), DMSO (1.5 mL) as solvent, and a 30-min reaction time at 140  $^{\circ}\mathrm{C}.$ 

With the optimized conditions in hand, the generality of the 85 reaction was evaluated by altering the diadldehydes and cyclic amino acids. At first, the symmetrical dialdehyde compounds were allowed to react with *trans*-4-hydroxy-L-proline **2a** or indoline-2-carboxylic acid **2b**. As shown in Table 2, the symmetrical [1,1'-biphenyl]-2,2'-dicarbaldehydes **1a-1g** afforded

dihydrophenanthren-9-ols **3aa-3ga** and 10-(1*H*-indol-1-yl)-9,10dihydrophenanthren-9-ols **4ab-4fb** in good to excellent yields of 79%-92% after the reaction. 2,2'-Oxydibenzaldehyde **1h** and 6,6'-

- 5 oxybis(3-methylbenzaldehyde) 1i generated pyrrol or indole substituted 10,11-dihydrodibenzo[b,f]oxepin-10-ols (3ha-3ia and 4hb) in moderate yields. Furthermore, 2,2'-thiodibenzaldehyde 1j
- **Table 2** Substrate scope of symmetrical dialdehydes <sup>a,b</sup>



<sup>a</sup> Reaction conditions: dialdehyde compound 1 (0.5 mmol), 2a or 2b (0.75 10 mmol) in 1.5 mL DMSO were stirred in a 10 mL-glass tube sealed with a cap in an oil bath at 140 °C for 30- or 60-min. <sup>b</sup> Isolated yields based on 1. Unless otherwise mentioned, the diastereomeric ratios were determined by <sup>1</sup>H NMR.
 <sup>c</sup> Diastereomeric ratios were calculated from the yields of the separated diastereomeric products.

- 15 was also applicable in this transformation and gave the corresponding 10,11-dihydrodibenzo[*b*,*f*]thiepin-10-ols **3ja** and **4jb** in 91% and 83% yields, respectively. It should be noted that both electron-withdrawing (-F and -CF<sub>3</sub>) and electron-donating groups (-Me, -OMe) on the phenyl ring werewell tolerated in this
- 20 transformation. Except for products **3ea**, **3ia**, **4eb** and **4hb**, the *anti* diastereomers were obtained as the major products in most cases. And 2,7-difluoro-10-(1*H*-pyrrol-1-yl)-9,10-

25 Next, the scope of asymmetrical dialdehydes was also investigated (Table 3). In order to verify the regioselectivity of the reaction, we began with the reaction of 3-fluoro-[1,1'-Biphenyl]-2,2'-dicarboxaldehyde 1k with trans-4-hydroxy-Lproline 2a. As anticipated, two regioisomers 3ka and 3ka' were 30 obtained by column chromatography with a ratio of 47:53. The structure of 1-fluoro-10-(1H-pyrrol-1-yl)-9,10dihvdrophenanthren-9-ol 3ka was proved by the presence of the doublet peak at  $\delta$  54.6 ppm in the <sup>13</sup>C NMR spectrum, while two doublet peaks at  $\delta$  72.0 and 64.1 ppm identified the structure of 35 3ka'. Similar results were also obtained for the reaction of dialdehyde 11 with trans-4-hydroxy-L-proline 2a, and dialdehyde 1k with indoline-2-carboxylic acid 2b. Interestingly, for the [1,1'biphenyl]-2,2'-dicarbaldehydes bearing only one substituent at the C4-position, the reactions with *trans*-4-hydroxy-L-proline 2a 40 could diastereoselectively afford the anti diastereomers (3ma and 3ma', 3na and 3na') in good yields of 86%-90%. The structures of 3ma and 3na were further confirmed by X-ray diffraction analysis (Fig. 1).<sup>17</sup> However, the reaction between dialdehyde 1n and indoline-2-carboxylic acid 2b lead to an 45 inseparable diastereomeric mixture of 4nb and 4nb' in a total vield of 88%.18

Table 3 Substrate scope of asymmetrical dialdehydes <sup>a,b</sup>



<sup>a</sup> Reaction conditions: dialdehyde compound 1 (0.5 mmol), 2a or 2b (0.75 mmol) in 1.5 mL DMSO were stirred in a 10 mL-glass tube sealed with a cap 50 in an oil bath at 140 °C for 30- or 60-min. <sup>b</sup> Isolated yields based on 1. The ratios of 3:3' and 4:4' were determined by <sup>1</sup>H NMR.



Scheme 1. Gram-scale synthesis of 3aa.

- 5 In order to prove the practicality of the method, we carried out a gram-scale experiment under optimized conditions (Scheme 1). The product 3aa was obtained in 90% yield with the reaction of 10 mmol of 1a and 15 mmol of 2a. Next, six 1-(phenanthren-9-yl)-1H-pyrroles (5aa-5da, 5fa and 5ma) were obained in 89-94%
- 10 yields through the dehydration of 10-(1H-pyrrol-1-yl)-9,10dihydrophenanthren-9-ols with  $P_2O_5$  in benzene (Scheme 2, (a)). We also attempted to synthesize 1-(phenanthren-9-yl)-1H-pyrrole 5aa from the reaction of 9-bromophenanthrene 6 and pyrrole 7. However, no desired product was detected under various 15 Ullmann reaction conditions (Scheme 2, (b))<sup>19</sup>.

Based on the literatures<sup>16b,20</sup> and our previous work,<sup>15</sup> a reaction of one aldehyde group from the dialdehyde compound with trans-4-hydroxy-L-proline 2a or indoline-2-carboxylic acid

20 2b produces an azomethineylide intermediate A or D, which undergoes intramolecular [3+2] cycloaddition with another aldehyde group and gave the bicyclic 1,3-oxazolidine B or E. 40 Dehydration of **B** forms intermediate **C**. The final product **3** or **4** was accomplished through the ring-opening aromatization of 25 corresponding intermediate C or E, respectively.



- Cs<sub>2</sub>CO<sub>3</sub> (2 equiv.), DMSO, 90 °C, 12-24 h.
- B. Cu(OAc)<sub>2</sub> (100 mol%), DBU (2 equiv.), DMSO 130-150 °C, MW, 20-30 min. C. Cu(II)(TMHD)<sub>2</sub> or Pd (OAc)<sub>2</sub> (20 mol%). KO<sup>t</sup>Bu (2 equiv.), DMF, 120 °C, 24 h.





30 Scheme 3. Proposed mechanism.

Table 4 IC<sub>50</sub> values of compounds on the viability of HepG2 cells

Compd.	IC <sub>50</sub> (µM)	Compd.	IC <sub>50</sub> (μM)
3aa	>50	4eb	>50
3ba	>50	anti-4eb	>50
3ca	>50	syn-4hb	>50 <sup>a</sup>
3da	>50	anti-4hb	>50
3ha	>50	anti-4jb	9.99
3ja	>50	5aa	>50
4ab	43.97	5ba	>50
anti-4ab	35.94	5ca	>50
4cb	>50	5da	>50
anti-4cb	22.45	5ma	>50
4db	>50	doxorubicin	5.75
anti-4db	>50		

plausible reaction mechanism was proposed in Scheme 3. The 35 dihydrophenanthren-9-ol analogues and 1-(phenanthren-9-yl)-With the synthesized pyrrolyl or indolyl substituted 9,10-1H-pyrroles in hand, we further evaluated their cytotoxicity on human hepatocellular carcinoma cells (HepG2) with the MTT assay. Among the tested compounds, four compounds showed obvious inhibitory activities, and the IC<sub>50</sub> values on the viability of HepG2 cells after 24 h incubation in the presence of the tested compounds were shown in Table 4. Interestingly, all the four active compounds contained the indoly-substitution, and the anti diastereomers showed lower IC50 values than those of their diastereomeric mixtures (anti-4ab, 35.94 µM vs 4ab, 43.97 µM; 45 anti-4cb, 22.45 µM vs 4cb, >50 µM). Compound anti-4jb exhibited the highest activity aginst HepG2 cells with an IC<sub>50</sub> value of 9.99  $\mu$ M, while the IC<sub>50</sub> value of the positive control doxorubicin was 5.75 µM. The results indicate that the indoly substituted 10,11-dihydrodibenzo[b,f]thiepin-10-ols might serve

50 as potential lead compounds for anticancer drug discovery.

In conclusion, a domino catalyst-free intramolecular [3+2] cycloaddition and ring-opening aromatization process between dialdehyde compounds and trans-4-hydroxy-L-proline or indoline-2-carboxylic acid has been successfully developed for 55 the efficient direct synthesis of pyrrolyl or indolyl substituted 9,10-dihydrophenanthren-9-ols, 10,11dihydrodibenzo[b,f]oxepin-10-ols and 10,11dihydrodibenzo[b,f]thiepin-10-ols. Both electron-withdrawing and electron-donating groups on the phenyl ring are well

- 60 tolerated in this transformation. And the reaction can be smoothly enlarged to gram-scale. Besides, 1-(phenanthren-9-yl)-1Hpyrroles can be easily obtained via the dehydration of 10-(1Hpyrrol-1-yl)-9,10-dihydrophenanthren-9-ols. Furthermore, the indolyl-substituted anti diastereomers showed potential better
- 65 activity against HepG2 cells than diastereomeric mixtures in the cytotoxic activity test, and the compound anti-4jb displayed the lowest IC<sub>50</sub> value of 9.99 µM. Further studies both on the application of the synthetic method and on the bioactivity of synthesized compounds are ongoing in our laboratory.

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# **Supplementary Material**

Supplementary data (experimental procedures, characterization data of products, and copies of <sup>1</sup>H and <sup>13</sup>C NMR spectra) 45 10 associated with this article can be found, in the online version, at http://....

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6	Tetrahedron
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#### **Declaration of interests**

 $\boxtimes$  The authors declare that they have no known competing financial interests or personal relationships that 10 could have appeared to influence the work reported in this paper.

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests:



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