

Allylic Amination

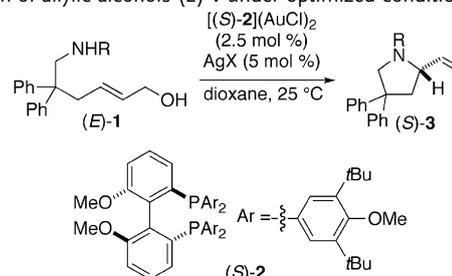
Gold(I)-Catalyzed Enantioselective Intramolecular Dehydrative Amination of Allylic Alcohols with Carbamates**

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The transition-metal-catalyzed enantioselective amination of allylic esters and carbonates represents one of the most well-established routes to chiral, nonracemic allylic amines.^[1] With the potential to condense synthetic sequences and reduce waste streams, the dehydrative amination of allylic alcohols as a route to enantiomerically enriched allylic amines has gained considerable interest. However, while the stereospecific amination of chiral secondary allylic alcohols has been demonstrated,^[2–4] the enantioselective amination of allylic alcohols remains problematic.^[5] Carreira et al. have reported the Ir^I-catalyzed enantioselective amination of 1-cyclohexyl-prop-2-enol with sulfamic acid in 70% *ee*.^[6] Hartwig et al. have reported the Ir^I/BPh₃-catalyzed enantioselective intermolecular amination of primary allylic alcohols with aromatic amines with up to 94% *ee*, but this method was restricted to cinnamyl alcohols in the absence of a stoichiometric Lewis acid promoter.^[7] The groups of Yamamoto^[8] and Kitamura^[9] have independently reported the enantioselective intramolecular amination of allylic alcohols catalyzed by Hg^{II} and Ru^{II} complexes, respectively. However, these methods were restricted to sulfonamide nucleophiles and high enantioselectivity was realized only for the formation of arene-fused nitrogen heterocycles. Herein we report a gold-catalyzed protocol for the intramolecular enantioselective amination of allylic alcohols with carbamates to form five- and six-membered aliphatic nitrogen heterocycles with up to 95% *ee*.

We recently reported the intramolecular dehydrative amination of allylic alcohols with alkylamines catalyzed by an achiral gold(I) phosphine complex.^[4,10] Encouraged by the high efficiency and stereospecificity of this transformation and guided by both our previous work in the area of gold(I)-catalyzed enantioselective allene hydroamination^[11,12] and Bandini's recent demonstration of gold(I)-catalyzed enantioselective arylation^[13] and alkoxylation^[14] of allylic alcohols,^[15] we targeted axially chiral bis(gold) complexes as catalysts for the intramolecular enantioselective amination of the ϵ -benzylamino allylic alcohol (*E*)-**1a** (Table 1). Unfortunately, optimization within this framework^[16] proved largely unsuccessful: treatment of (*E*)-**1a** with a catalytic 1:2 mixture of [(*S*)-**2**](AuCl)₂ and AgSbF₆ in dioxane at 25 °C for 5 h led to

Table 1: Effect of the nucleophile on the gold(I)-catalyzed intramolecular amination of allylic alcohols (*E*)-**1** under optimized conditions.^[16,18]



Entry	1 + 3 , R ^[a]	X	Time [h]	Yield [%] ^[b]	<i>ee</i> [%] ^[c]
1	a , Bn	SbF ₆	5	100 ^[d]	29
2	b , Cbz	ClO ₄	48	99	79
3	c , Boc	ClO ₄	48	97	80
4 ^[e]	d , Troc	ClO ₄	48	62	84
5	e , CO ₂ Me	ClO ₄	48	97	75
6	f , Ts	ClO ₄	48	98	76
7	g , Fmoc	ClO ₄	48	95	91

[a] Bn = benzyl, Cbz = benzyloxycarbonyl, Boc = *tert*-butyloxycarbonyl, Troc = 2,2,2-trichloroethoxycarbonyl, Ts = 4-toluenesulfonyl, Fmoc = fluorenylmethyloxycarbonyl. [b] Yield of isolated product. [c] Determined by HPLC analysis on chiral support. [d] Conversion. [e] Reaction run at 40 °C.

quantitative conversion to 2-vinylpyrrolidine **3a**, but with only 29% *ee* (Table 1, entry 1).^[17] We then focused our attention on the manipulation of the nitrogen nucleophile as a means to amplify stereinduction (Table 1). These experiments proved fruitful and gold(I)-catalyzed cyclization of Fmoc-protected ϵ -amino allylic alcohol (*E*)-**1g** employing an optimized catalyst system comprised of [(*S*)-**2**](AuCl)₂ (2.5 mol %) and AgClO₄ (5 mol %) in dioxane at room temperature for 48 h led to the isolation of (*S*)-**3g** in 95% yield with 91% *ee* (Table 1, entry 7).^[16,18]

The scope of this gold(I)-catalyzed enantioselective intramolecular amination was evaluated as a function of alkene configuration, substitution, and ring size (Table 2). The enantioselectivity of the amination was sensitive to the alkene configuration: (*Z*)-**1g** was converted into **3g** in 99% yield with $\leq 5\%$ *ee* (Table 2, entry 1). Although ϵ -amino allylic alcohols that possessed *gem*-dialkyl substitution at the homoallylic position cyclized with higher enantioselectivity than did an unsubstituted ϵ -amino allylic alcohol (Table 2, entries 2–4), homoallylic *gem*-disubstitution was not required for high enantioselectivity (Table 2, entries 5 and 6). For example, gold(I)-catalyzed cyclization of **4**, which possessed a single phenyl group at the homoallylic position, led to isolation of pyrrolidine **5** in 87% yield as a 1:1 mixture of

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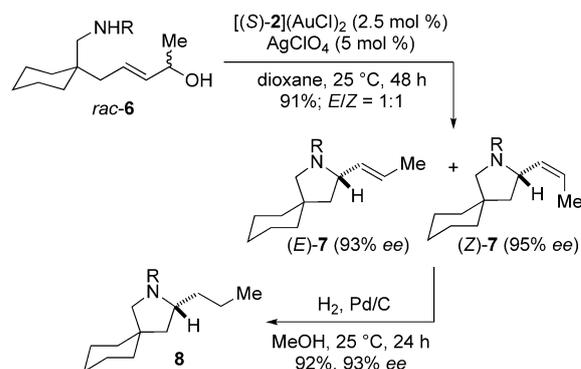
Table 2: Substrate scope of the intramolecular amination of allylic alcohols (0.6 M) catalyzed by a 1:2 mixture of [(*S*)-2](AuCl)₂ (2.5 mol %) and AgClO₄ (5 mol %) in dioxane at 25 °C for 48 h.^[18]

Entry	Substrate	Heterocycle	Yield [%] ^[a]	ee [%] ^[b]
1			99	≤ 5
2			94	90
3			95	94
4			89	62
5			87 ^[d]	90/92
6			98 ^[d]	85/91
7			87 ^[d]	88/90
8			69	77
9 ^[c]			86	94
10 ^[c]			99	92
11 ^[c]			99	91

[a] Yield of isolated product. [b] Determined by HPLC analysis on chiral support. [c] Compounds **4** and **5**: R = Ph (entry 5). [d] Diastereomeric ratio ≈ 1:1. [e] Reaction run at 50 °C.

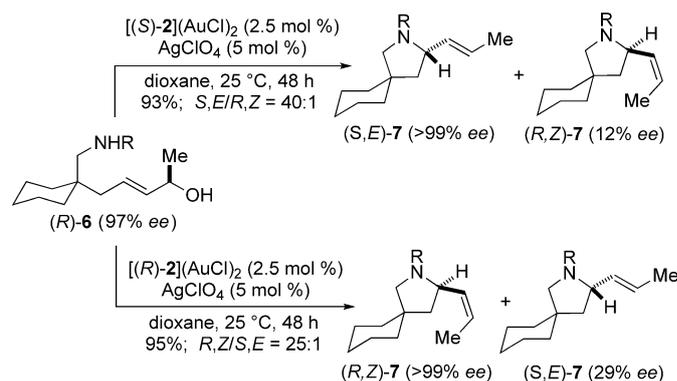
cis and *trans* diastereomers, both of which were formed with ≥ 90% *ee*, indicative of overriding catalyst control of stereoselection. The gold-catalyzed enantioselective amination also tolerated *gem*-dialkyl substitution at the hydroxy-bound carbon atom (Table 2, entry 7) and was applicable to the synthesis of six-membered nitrogen heterocycles (Table 2, entries 8–11), proving particularly effective for the synthesis of differently protected 2-vinylpiperazines (Table 2, entries 9–11).

The effect of a chiral secondary allylic alcohol moiety on the efficiency and stereoselectivity of this gold-catalyzed allylic amination was evaluated employing *ε*-amino allylic alcohol **6**. In one experiment (Scheme 1), cyclization of *rac*-**6** catalyzed by [(*S*)-2](AuCl)₂/AgClO₄ led to isolation of a 1:1 mixture of (*E*)-**7** and (*Z*)-**7** in 91% combined yield. Hydrogenation of this mixture formed 2-propylpyrrolidine **8** in 92% yield with 93% *ee*, which established that (*E*)-**7** and (*Z*)-**7**



Scheme 1. Cyclization of *rac*-**6** catalyzed by [(*S*)-2](AuCl)₂/AgClO₄. R = Fmoc.

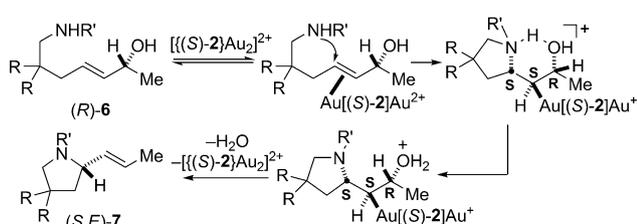
possessed the same absolute configuration (*S* by analogy),^[18] and HPLC analysis of the conversion of *rac*-**6** to **7** revealed that both enantiomers of **6** reacted at similar rates (*k_S/k_R* = 1.06). In two additional experiments (Scheme 2), cyclization of enantiomerically enriched (*R*)-**6** catalyzed by [(*S*)-2](AuCl)₂/AgClO₄ led to isolation of a 40:1 mixture of (*S,E*)-**7** and (*R,Z*)-**7** in 93% combined yield while cyclization of (*R*)-**6** catalyzed by [(*R*)-2](AuCl)₂/AgClO₄ led to isolation of a 25:1



Scheme 2. Cyclization of enantiomerically enriched (*R*)-**6** (97% *ee*) catalyzed by [(*S*)-2](AuCl)₂/AgClO₄ (top pathway) and [(*R*)-2](AuCl)₂/AgClO₄ (bottom pathway). R = Fmoc.

mixture of (*R,Z*)-**7** and (*S,E*)-**7** in 95% combined yield. Together, these results established that asymmetric induction is determined solely by the catalyst configuration (*S* → *S*; *R* → *R*) and that *E/Z* selectivity is determined by the stereochemical relationship between the incipient N-bound stereocenter and the extant O-bound stereocenter (*S/R* → *E*; *R/R* → *Z*), consistent with the net *syn* displacement of the hydroxy group by the attacking carbamate nucleophile.

The net *syn* displacement of the hydroxy group by the nitrogen nucleophile, which was also documented for the amination of allylic alcohols catalyzed by achiral mono(gold) complexes,^[3,4] is consistent with a mechanism involving π -complexation of gold to the C=C bond followed by *anti*-addition of the nucleophile and *anti*-elimination of the hydroxy group, perhaps facilitated by an intramolecular N–H–O hydrogen bond (Scheme 3).^[19] Alternatively, *syn*-sub-



Scheme 3. Proposed mechanism of the gold-catalyzed allylic amination of (*R*)-**6**. RR = (CH₂)₅, R' = Fmoc.

stitution is also consistent with a mechanism involving σ -activation of the hydroxy group followed by concerted S_N2' displacement^[20] and Toste et al. have recently demonstrated that bis(gold) phosphine complexes are sufficiently Lewis acidic to acidify the hydroxy proton of an alcohol.^[21] However, the failure of either triflic acid or BF₃·OEt₂ (10 mol %, 25 °C, 48 h) to catalyze the cyclization of (*E*)-**1g** argues against a σ -activation pathway for this allylic amination.

In summary, we have developed a gold(I)-catalyzed protocol for the intramolecular enantioselective amination of allylic alcohols with carbamates to form five- and six-membered nitrogen heterocycles with up to 95% *ee*. Cyclization of chiral ε -amino allylic alcohols that possessed a stereogenic homoallylic or hydroxy-bound carbon atom occurred with overriding catalyst control of asymmetric induction. Stereochemical analysis of the cyclization of (*R*)-**6**, which possessed a secondary allylic alcohol moiety, established the net *syn*-displacement of the hydroxy group by the carbamate nucleophile.

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- [17] Using a 1:1 mixture of [(*S*)-**2**](AuCl)₂ and AgSbF₆ led to significant decrease in both rate and enantioselectivity.^[16]
- [18] The absolute configuration of (*S*)-**3b** was determined by comparison to the HPLC trace of (*R*)-**3b**.^[11a] The *S* configuration of the heterocycles depicted in Table 2, entries 4 and 10 was established by chemical correlation. The absolute configurations of all other heterocycles were assigned by analogy.
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