



Synthesis of quinazoline based chiral ligands and application in the enantioselective addition of phenylacetylene to aldehydes

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ABSTRACT

Five novel 4-phenylquinazolinols were synthesised in three steps. Their application as ligands in the titanium tetraisopropoxide promoted catalytic enantioselective addition of phenylacetylene to a variety of aldehydes gave propargylic alcohols. Under the optimised reaction conditions, the best enantioselectivity was obtained using L-lactic acid derived 4-phenylquinazolinol and apart from the cyclohexylcarbaldehyde derivative, 16 propargylic alcohols were then synthesised in moderate to excellent enantiomeric excess from 53% to 97%.

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1. Introduction

Catalytic asymmetric C–C bond formation is an intense research area in the field of modern organic chemistry.¹ Among the variety of nucleophilic reactions, the asymmetric alkynylzinc addition to aldehydes has been a particularly significant method for the synthesis of propargylic alcohols, which are very useful precursors for various organic compounds involving natural products, pharmaceuticals and macromolecules.² In 1994, Ishizaki and Hoshino published the first asymmetric alkynylzinc reagents addition to aldehydes in the presence of chiral pyridyl alcohol ligand **1**³ (Fig. 1). After this seminal report, many chiral ligands such as Salen,⁴ pyridyl,⁵ amino alcohol,^{6–8} BINOL^{9–11} and sulfonamide alcohols^{10,12–14} and its derivatives^{15,16} have been emerged to obtain high enantioselectivity in C–C bond forming reactions. In 2000, Carreira and co-workers discovered a highly enantioselective alkynylzinc addition to aldehydes.¹⁷ This result was based on a stoichiometric approach using Zn(OTf)₂ and N-methylephedrine. A year later they came up with a catalytic version of this approach.¹⁸ Another efficient method was reported by Pu et al. using BINOL derivatives and Et₂Zn with Ti(O*i*Pr)₄.¹⁹ In the last decades, many types of catalytic asymmetric reactions have been carried out with various

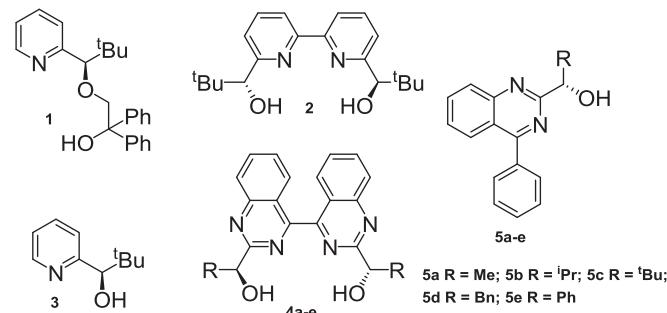


Fig. 1. Structure of some chiral heterocyclic ligands containing nitrogen (**1–4**) and evaluated ligands (**5a–e**).

heterocyclic ligands containing nitrogen such as pyridyl alcohols **1–3**,²⁰ quinazolinone^{21,22} and quinazoline alcohols **4a–e**²³ (Fig. 1).

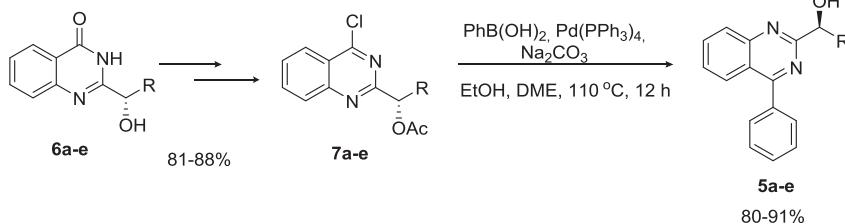
In our previous report,²³ we synthesised a series of (1*S*,1'*S*)-4,4'-biquinazoline alcohols **4a–e** from commercially available chiral α -hydroxy acids and α -amino acids in a few steps. Their application as ligands in enantioselective phenylacetylene addition to aldehydes furnished good yields with moderate enantioselectivities. In this respect, we describe here the synthesis of a series of quinazoline alcohols **5a–e** containing a phenyl group at their 4-position. The addition of phenylacetylene to aromatic aldehydes was then

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studied with the aim of determining the catalytic effects of these ligands.

2. Results and discussion

The quinazolinones **6a–e** were prepared from α -hydroxy acids and α -amino acids in four steps in high yields and without a need for chromatography (**Scheme 1**).²³ Enantiomerically pure quinazolinones **6a–e** were treated with Ac₂O in the presence of pyridine as a base to protect the alcohols followed by exposing them to POCl₃ and *N,N*-diethyl aniline in order to obtain chloroquinazolines **7a–e** (**Scheme 1**).

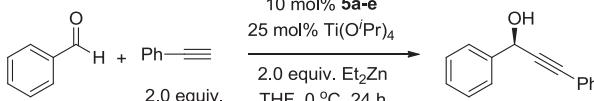


Scheme 1.

With chloroquinazolines **7a–e** in hand, the synthesis of chiral 4-phenylquinazoline alcohols was carried out via Suzuki coupling reactions with phenylboronic acid in the presence of 2 mol % of Pd(PPh₃)₄ at reflux. Except for **5d** (72% ee), the other chiral 4-phenylquinazolines **5a–c** and **5e** were obtained in enantiopure form (99% ee) and high yields (80–91%).

First of all, we carried out the alkynylation of benzaldehyde in the presence of 10 mol % of ligand **5a** both with and without $\text{Ti(O}^{\prime}\text{Pr)}_4$ in dry THF at 0 °C under nitrogen atmosphere for 24 h. As can be seen in Table 1, in the absence of $\text{Ti(O}^{\prime}\text{Pr)}_4$ the reaction yield was moderate but the enantioselectivity too low (Table 1, entry 1). Pleasingly, the loading of 25% $\text{Ti(O}^{\prime}\text{Pr)}_4$ into the reaction conditions increased both enantiomeric excess and yield enormously (Table 1, entry 2). However a rise in ees and yields was not observed substantially when 10 and 40 mol % $\text{Ti(O}^{\prime}\text{Pr)}_4$ was used (Table 1, entries

Table 1
Asymmetric alkynylation of benzaldehyde catalysed by **5a–e**



Entry	Ligand (mol %)	ee ^a (%)	Yield ^b (%)	Config. ^f
1	5a (10) ^c	14	64	R
2	5a (10)	78	98	R
3	5a (10) ^d	70	90	R
4	5a (10) ^e	77	91	R
5	5b (10)	45	70	R
6	5c (10)	4	75	R
7	5d (10)	54	93	R
8	5e (10)	60	80	R
9	5a (15)	78	90	R
10	5a (5)	74	87	R

^a Determined by chiral HPLC (Chiralcel OD-H).

^b Determined by isolated yields.

^c Without Ti(OⁱPr)₄.

^d Ti(O*i*Pr)₄: 10 mol %.

^e Ti(O*i*Pr)₄: 40 mol %.
^f The absolute configurations of adducts were assigned in accordance with literature.²⁴

3 and 4). To identify the appropriate ligand, the synthesised quinazolines **5a–e** were then examined in the condition mentioned above. Although the better differentiated *tert*-leucine derived chiral centre containing ligand **5c** gave the least enantioselectivity (**Table 1**, entry 6), interestingly, the least bulky chiral centre bearing one **5a** was still the best ligand choice (**Table 1**, entry 2). Additionally, 1,3-diphenylprop-2-yn-1-ol was accomplished between 45 and 60% enantioselectivities (**Table 1**, entries 5, 7 and 8) with the other ligands (**5b**, **5d** and **5e**). In order to seek a more efficient catalytic system, we studied the effect of several reaction parameters such as the choice of solvent, the reaction temperature and the amount of the chiral ligand used. As a result of increasing the amount of ligand

from 10 to 15 mol %, the ees remained the same. However, decreasing the amount of the ligand to 5 mol %, diminished slightly the reaction ees and yields.

After these observations we turned our attention to solvent screening in order to improve ees. No matter which solvent was used, we could not get any better enantioselectivity apart from in THF. Decreasing the reaction temperature from 0 °C to –10 °C led to an increase in ee, but it reduced the reaction yield (**Table 2**, entry 9). Extending the reaction time from 24 to 40 h allowed us to increase the reaction yield from 65 to 89% (**Table 2**, entry 10).

Table 2
Reaction optimisation for asymmetric alkynylation of benzaldehyde

Entry	Solvent	Temp (°C)	ee ^a (%)	Yield ^b (%)	Config. ^c
1	Hexane	0	17	17	R
2	THF	0	78	98	R
3	Diethyl ether	0	51	75	R
4	Dichloromethane	0	27	96	R
5	Chloroform	0	7	38	R
6	Toluene	0	54	71	R
7	1,4-Dioxane	0	35	70	R
8	THF	25	70	95	R
9	THF	-10	84	65	R
10	THF	-10^d	84	89	R
11	THF	-20	80	50	R

^a Determined by chiral HPLC (Chiralcel OD-H).

^b Isolated yields.

^c The absolute configurations of adducts were assigned in accordance with literature.²⁴

^d The reaction was conducted for 40 h.

The addition of phenylacetylene to a series of aldehydes was carried out under the optimised reaction conditions (THF as a solvent, 10% **5a**, 25% $Ti(O^iPr)_4$ at $-10\text{ }^\circ C$ for 40 h) (**Table 2**, entry 10). As shown in **Table 3**, 17 aldehydes were then converted into their propargylic alcohols in good to excellent yields (67–98%). While propargylic alcohols derived from aromatic aldehydes (**Table 3**, entries 1–13) gave highly enantioselective results up to 97%, aliphatic ones afforded moderate enantioselectivities (from 53 to 76% ee). Interestingly, among the aldehydes tested in the reaction, the

Table 3Enantioselective addition of phenylacetylene to aldehydes using ligand **5a**

Entry	Aldehyde	Product	ee ^a (%)	Yield ^b (%)	Config. ^c	10 mol% 5a	
						2.0 equiv.	25 mol% Ti(O <i>i</i> Pr) ₄
1	Benzaldehyde	8a	84	89	R ²⁴		
2	2-Methoxybenzaldehyde	8b	88	75	R ²⁵		
3	3-Methoxybenzaldehyde	8c	75	98	R ²⁶		
4	4-Methoxybenzaldehyde	8d	86	73	R ²⁴		
5	2-Chlorobenzaldehyde	8e	78	67	R ²⁷		
6	3-Chlorobenzaldehyde	8f	83	96	R ²⁴		
7	4-Chlorobenzaldehyde	8g	87	95	R ²⁴		
8	3-Bromobenzaldehyde	8h	80	78	R ²⁸		
9	4-Bromobenzaldehyde	8i	97	76	R ²⁴		
10	3-Methylbenzaldehyde	8j	91	98	R ²⁴		
11	4-Methylbenzaldehyde	8k	89	77	R ²⁴		
12	1-Naphthaldehyde	8l	91	94	R ²⁵		
13	Furfural	8m	88	85	R ²⁹		
14	Crotonaldehyde	8n	76	74	NA		
15	Propionaldehyde	8o	53	68	R ³⁰		
16	Isobutyraldehyde	8p	67	70	R ³¹		
17	Cyclohexanecarbaldehyde	8r	13	82	S ³¹		

^a Determined by chiral HPLC (Chiralcel OD-H).^b Isolated yields.^c The absolute configurations of adducts were assigned in accordance with literature.

cyclohexanecarbaldehyde corresponding adduct accomplished neither a good enantioselectivity nor the same sense of stereoselectivity.

3. Conclusions

In summary, we successfully prepared novel chiral quinazolinols (**5a–e**) from commercially available starting materials in a few steps. Among these ligands, quinazolinol **5a** exhibited high catalytic activity and asymmetric induction in the alkynylation of aldehydes. It is of note that the best enantioselectivity (75%) of propargylic alcohols obtained in our earlier report²³ is the worst one in this present work for aromatic aldehydes. Application of these ligands in other catalytic enantioselective reactions is currently under investigation in our laboratory.

4. Experimental section

4.1. General procedure 1 for the preparation of (*S*)-1-(4-phenylquinazolin-2-yl)ethanol **5a**

To a solution of (*S*)-1-(4-chloroquinazolin-2-yl)ethyl acetate **7a** (5.00 g, 19.90 mmol) in DME (15 mL) and ethanol (15 mL) were added phenylboronic acid (2.55 g, 20.94 mmol), Pd(PPh₃)₄ (454 mg, 0.39 mmol) and Na₂CO₃ (22 mL, 2 M in water). The resulting mixture was then refluxed for 12 h, cooled to room temperature and diluted with dichloromethane (150 mL). The mixture was washed with water (3×75 mL). The organic layer was dried over Na₂SO₄, and evaporated under reduced pressure. The crude product was purified by column chromatography (silica gel, hexane/EtOAc, 7:1), and then crystallised on addition of ethanol to give (*S*)-1-(4-phenylquinazolin-2-yl)ethanol **5a** as a colourless solid (4.48 g, 90%). Mp: 85–87 °C; [α]_D +7.5 (c 3.2, CH₂Cl₂); ee: 99%; retention time 6.9 min, Chiralcel OD-H, 90:10 n-hexane/*i*-PrOH, flow rate of 1 mL/min, 254 nm; IR (KBr, cm⁻¹) 3232, 3048, 2978, 2909, 1613, 1486, 1132, 1107; ¹H NMR (400 MHz, CDCl₃, ppm) δ 8.13 (d, J=8.8 Hz, 1H), 8.09 (d, J=8.4 Hz, 1H), 7.93–7.57 (m, 7H), 5.17–5.16 (m, 1H), 4.6 (s, 1H), 1.71 (d, J=6.7 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃, ppm) δ 168.9, 167.3, 150.7, 137.0, 134.0, 130.2, 130.0, 128.6, 128.4,

127.3, 127.2, 121.8, 69.8, 23.4; HRMS calculated for C₁₆H₁₄N₂O: 251.1186. Found: 251.1187.

4.2. (*S*)-2-Methyl-1-(4-phenylquinazolin-2-yl)propan-1-ol **5b**

General procedure 1 was followed using (*S*)-1-(4-chloroquinazolin-2-yl)-2-methylpropyl acetate **7b** (4.50 g, 16.10 mmol), phenylboronic acid (2.06 g, 16.95 mmol), Pd(PPh₃)₄ (371 mg, 0.32 mmol) and Na₂CO₃ (16 mL, 2 M in water) in a solution in mixture of DME (15 mL) and ethanol (15 mL). The crude product was purified by column chromatography (hexane-EtOAc, 7:1), and then crystallised on addition of ethanol to give (*S*)-2-methyl-1-(4-phenylquinazolin-2-yl)propan-1-ol **5b** as a colourless solid (4.08 g, 91%). Mp: 69–70 °C; [α]_D −5.29 (c 1.7, CH₂Cl₂); ee: 99%; retention time 5.1 min, Chiralcel OD-H, 90:10 n-hexane-*i*-PrOH, flow rate of 1 mL/min, 254 nm; IR (KBr, cm⁻¹) 3455, 3061, 2959, 2928, 1613, 1487, 1143, 1019; ¹H NMR (400 MHz, CDCl₃, ppm) δ 8.12 (d, J=7.5 Hz, 1H), 8.08 (d, J=8.4 Hz, 1H), 7.94–7.58 (m, 7H), 4.92 (dd, J=5.6, 3.4 Hz, 1H), 4.46 (d, J=5.6 Hz, 1H), 2.54–2.46 (m, 1H), 1.17 (d, J=7.0 Hz), 0.77 (d, J=6.8 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃, ppm) δ 168.6, 166.1, 150.5, 137.1, 133.9, 130.1, 128.6, 128.4, 127.4, 127.2, 121.7, 34.0, 19.9, 15.4, 0.02; HRMS calculated for C₁₆H₁₄N₂O: 279.1499. Found: 279.1503.

4.3. (*S*)-2,2-Dimethyl-1-(4-phenylquinazolin-2-yl)propan-1-ol **5c**

General procedure 1 was followed using (*S*)-1-(4-chloroquinazolin-2-yl)-2,2-dimethylpropyl acetate **7c** (2.30 g, 7.85 mmol), phenylboronic acid (1.01 g, 8.24 mmol), Pd(PPh₃)₄ (180 mg, 0.157 mmol) and Na₂CO₃ (17.3 mL, 2 M in water) in a solution in mixture of DME (15 mL) and ethanol (15 mL). The crude product was purified by column chromatography (hexane/EtOAc, 7:1), and then crystallised on addition of ethanol to give (*S*)-2,2-dimethyl-1-(4-phenylquinazolin-2-yl)propan-1-ol **5c** as a colourless solid (1.84 g, 80%). Mp: 101–103 °C; [α]_D +7.08 (c 2.4, CH₂Cl₂); ee: 99%; retention time 5.0 min, Chiralcel OD-H, 90:10 n-hexane/*i*-PrOH, flow rate of 1 mL/min, 254 nm; IR (KBr, cm⁻¹) 3459, 3062, 2957, 2866, 1614, 1565, 1546, 1487, 1391, 1068; ¹H NMR (400 MHz, CDCl₃, ppm) δ 8.15 (d, J=8.4 Hz, 1H), 8.10 (d, J=8.5 Hz, 1H), 7.92–7.58 (m, 7H), 4.73 (d, J=7.6 Hz, 1H), 4.47 (d, J=7.6 Hz, 1H), 1.06 (s, 9H); ¹³C NMR (100 MHz, CDCl₃, ppm) δ 167.4, 165.3, 150.5, 137.1, 133.8, 130.1, 128.6, 127.3, 127.1, 121.8, 80.9, 30.9, 26.3; HRMS calculated for C₁₆H₁₄N₂O: 293.1655. Found: 293.1653.

4.4. (*S*)-Phenyl(4-phenylquinazolin-2-yl)methanol **5d**

General procedure 1 was followed using (*S*)-(4-chloroquinazolin-2-yl)(phenyl)methyl acetate **7d** (5.40 g, 17.26 mmol), phenylboronic acid (2.20 g, 18.12 mmol), Pd(PPh₃)₄ (396 mg, 0.34 mmol) and Na₂CO₃ (7.8 mL, 2 M in water) in a solution in mixture of DME (6 mL) and ethanol (6 mL). The crude product was purified by column chromatography (hexane/EtOAc, 10:1), and then crystallised on addition of ethanol to give (*S*)-2,2-dimethyl-1-(4-phenylquinazolin-2-yl)propan-1-ol **5d** as a colourless solid (4.58 g, 85%). Mp: 102–106 °C; [α]_D +54.33 (c 2.0, CH₂Cl₂); ee: 72%; retention time 8.9 min (major), 11.7 min (minor), Chiralcel OD-H, 90:10 n-hexane/*i*-PrOH, flow rate of 1 mL/min, 254 nm; IR (KBr, cm⁻¹) 3423, 3058, 1613, 1565, 1547, 1389, 1058; ¹H NMR (400 MHz, CDCl₃, ppm) δ 8.19–7.31 (m, 14H), 6.10 (d, J=6.3 Hz, 1H), 5.29 (d, J=6.3 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃, ppm) δ 168.9, 165.2, 150.7, 142.5, 136.9, 134.1, 130.2, 128.6, 128.5, 128.2, 127.6, 127.5, 127.2, 126.9, 121.9, 75.5; HRMS calculated for C₁₆H₁₄N₂O: 313.1342. Found: 313.1335.

4.5. (S)-2-Phenyl-1-(4-phenylquinazolin-2-yl)ethanol 5e

General procedure 1 was followed using (S)-1-(4-chloroquinazolin-2-yl)-2-phenylethyl acetate **7e** (5.00 g, 15.30 mmol), phenylboronic acid (1.96 g, 16.06 mmol), Pd(PPh₃)₄ (352 mg, 0.31 mmol) and Na₂CO₃ (15 mL, 2 M in water) in a solution in mixture of DME (20 mL) and ethanol (20 mL). The crude product was purified by column chromatography (hexane/EtOAc, 6:1), and then crystallised on addition of ethanol to give (S)-2-phenyl-1-(4-phenylquinazolin-2-yl)ethanol **5e** as a colourless solid (4.24 g, 85%). Mp: 66–68 °C; [α]_D –70.74 (c 2.7, CH₂Cl₂); ee: 99%; retention time 6.2 min, Chiralcel OD-H, 90:10 n-hexane/i-PrOH, flow rate of 1 mL/min, 254 nm; IR (KBr, cm^{–1}) 3430, 3058, 1613, 1565, 1547, 1389, 1058; ¹H NMR (400 MHz, CDCl₃, ppm) δ 8.12 (d, J=8.7 Hz, 1H), 8.06 (d, J=8.4 Hz, 1H), 7.92–7.23 (m, 12H), 5.36–5.29 (m, 1H), 4.48 (d, J=5.8 Hz, 1H), 3.51 (dd, J=13.8, 3.95 Hz, 1H), 3.17 (dd, J=13.8, 7.64 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃, ppm) δ 168.7, 165.7, 150.7, 138.2, 137.0, 134.0, 130.2, 130.08, 129.7, 128.6, 128.5, 128.1, 127.4, 127.2, 126.2, 121.83, 74.24, 43.59; HRMS calculated for C₁₆H₁₄N₂O: 327.1499. Found: 327.1485.

4.6. Typical procedure for catalytic asymmetric addition of phenylacetylene to aldehydes

Under nitrogen atmosphere, phenylacetylene (1.96 mmol) and diethylzinc (1.96 mmol, 1 M in hexane) were dissolved in freshly distilled THF (3 mL) and refluxed for 1 h. The resulting solution was brought to ambient temperature and following that a solution of chiral ligand **5a** (0.098 mmol) in THF (3 mL) was added. After stirring for 30 min, Ti(O*i*Pr)₄ (0.25 mmol) was added to the solution and stirred for a further 1 h. Finally, the mixture was cooled to –10 °C and aldehyde (0.98 mmol in 2 mL THF) was added. After stirring for 40 h at –10 °C under a nitrogen atmosphere, the reaction was then quenched with saturated ammonium chloride (10 mL). The aqueous phase was extracted with ethyl acetate (3×20 mL), washed with brine (3×20 mL), dried with anhydrous Na₂SO₄, filtered and concentrated under reduced pressure. The crude product was purified by column chromatography (silica gel, EtOAc/hexane 9:1) to give the desired product. The enantiomeric excess was determined by chiral HPLC using Chiralcel OD-H column.

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Supplementary data

Supplementary data associated with this article can be found in the online version, at <http://dx.doi.org/10.1016/j.tet.2014.08.067>.

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