Synthesis and Antitumor Activity of Amino Derivatives of Pyridine-2carboxaldehyde Thiosemicarbazone

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Various substituted pyridine-2-carboxaldehyde thiosemicarbazones (12 compounds) have been synthesized and evaluated for antineoplastic activity in mice bearing the L1210 leukemia. Oxidation of 3-nitro-2-picoline, 5-nitro-2-picoline, 3-nitro-2,4-lutidine, and 5-nitro-2,4-lutidine with selenium dioxide was employed to generate the corresponding pyridine-2-carboxaldehydes, which were then converted to cyclic ethylene acetals and subsequently reduced to amino and hydroxyamino derivatives by catalytic hydrogenation. Condensation of nitro aldehydes and acetals with thiosemicarbazide afforded the respective thiosemicarbazones. Acetylation of the amino acetals and alkylsulfonation of the 5-amino acetal, followed by condensation with thiosemicarbazide was employed to yield amide thiosemicarbazones. The most active compounds synthesized were 3-aminopyridine-2-carboxaldehyde thiosemicarbazone and 3-amino-4-methylpyridine-2-carboxaldehyde thiosemicarbazone which produced against the L1210 leukemia, % T/C values of 246 and 255, and 40% 60-day long-term survivors at two daily doses of 40 mg/kg and 10 mg/kg. respectively, for six consecutive days.

The reductive conversion of ribonucleotides to deoxyribonucleotides appears to be a rate-controlling step in the biosynthesis of DNA. Thus, the level of the enzyme activity that catalyzes this reaction, ribonucleoside diphosphate reductase, is closely correlated with the rate of cellular replication.¹ Furthermore, since deoxyribonucleoside triphosphates are present in extremely low levels in mammalian cells, Corey and Chiba² have hypothesized that DNA synthesis would be more affected by an inhibitor of ribonucleoside diphosphate reductase than by an inhibitor of DNA polymerase. α -(N)-Heterocyclic carboxaldehyde thiosemicarbazones (HCTs) represent some of the most potent known inhibitors of ribonucleoside diphosphate reductase, being 80-5000 times more effective, depending upon the HCT, than hydroxyurea, a clinically useful agent that acts, at least in part, by interfering with the enzymatic reduction of ribonucleotides. One of the members of the HCT series, 5-hydroxypyridine-2-carboxaldehyde thiosemicarbazone (5-HP), has received a phase I trial in patients with cancer.^{3,4} These studies have shown that the broad tumor inhibitory activity of 5-HP in animal systems did not occur in patients with cancer, although it is important to stress that this agent did not undergo phase II evaluation. The inactivity of 5-HP as an antineoplastic agent in humans appeared to be the result of (a) its relatively low inhibitory potency for the target

enzyme ribonucleoside diphosphate reductase,⁵ (b) its short biological half-life in humans, which resulted from the rapid formation and elimination of an inactive glucuronide conjugate,³ and (c) its relatively low therapeutic index.³ 5-Aminoisoquinoline-1-carboxaldehyde thiosemicarbazone (AIQ-1), which was about 100 times more potent than 5-HP as an inhibitor of ribonucleoside diphosphate reductase, was considered as a second generation drug of this class for trial against human cancer. However, since the 5-acetyl derivative of AIQ-1 was completely devoid of carcinostatic activity,⁶ the 4-methyl derivative of AIQ-1 (MAIQ-1) was designed and synthesized to minimize the formation of the inactive 5-acetyl derivative. The presence of a 4-methyl group in MAIQ-1 created steric hindrance, minimizing enzymatic acetylation of the 5-amino group.^{6,7} Unfortunately, MAIQ-1 did not appear to have sufficient water solubility to permit adequate formulation for use in humans.

French and Blanz⁸ reported the synthesis and antitumor activity of 5-(acetylamino)pyridine-2-carboxaldehyde thiosemicarbazone (5-AAP, 26), which showed both strong inhibitory activity against ribonucleoside diphosphate reductase in vitro and significant antitumor activity in vivo. By analogy to the relationship between AIQ-1 and its 5-acetyl derivative, we hypothesized that 5-aminopyridine-2-carboxaldehyde thiosemicarbazone (5-AP, 18), might well be more effective than 5-AAP. Furthermore, since 5-AP was a relatively small molecule capable of

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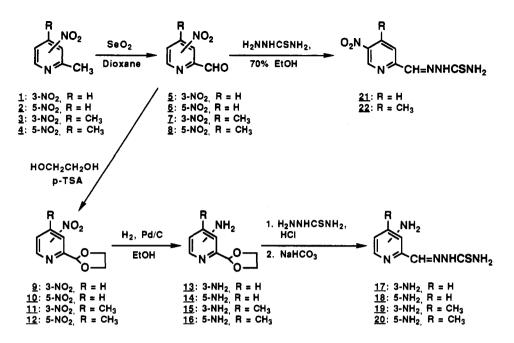
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Scheme I



forming an acidic salt of the amino group, we speculated further that it would be sufficiently water soluble to permit formulation for clinical usage. In this paper, we report the syntheses of 5-AP, the related agent 3-aminopyridine-2-carboxaldehyde thiosemicarbazone (3-AP, 17), and their 4-methyl derivatives (3-AMP and 5-AMP, 19 and 20, respectively), in which a methyl group has been placed at the 4-position of the pyridine ring in an effort to minimize enzymatic acetylation. For comparison, we have also synthesized 5-AAP (26) by a different method from that of French and Blanz⁸ and two of its analogues, the 5-(acetylamino)-4-methyl derivative (28) and 3-(acetylamino)-4-methylpyridine-2-carboxaldehyde thiosemicarbazone (27). In addition, two nitro (21 and 22) and two alkylsulfonamide (31 and 32) derivatives were synthesized to evaluate further the structural features of this class of compounds that are required for antineoplastic activity.

Chemistry

Various 3-amino, 5-amino-, and 5-nitro-substituted pyridine-2-carboxaldehyde thiosemicarbazones (17-22) were synthesized by methodology shown in Scheme I. Oxidation^{9,10} of 3-nitro-, 5-nitro-, 4-methyl-3-nitro-, and 4-methyl-5-nitro-2-picolines (1-4) with selenium dioxide in refluxing dioxane yielded the corresponding pyridine-2-carboxaldehydes (5-8). To reduce the nitro groups in compounds 5-8 to amino functions, the aldehydes were protected by conversion to the cyclic ethylene acetals (9-12), which were then reduced by catalytic hydrogenation using Pd/C as a catalyst to give the corresponding amino acetals (13-16).⁶ Compounds 13-16 were then reacted with thiosemicarbazide in ethanol containing 10% concentrated hydrochloric acid to form the desired thiosemicarbazone hydrochlorides; the free bases (17-20) were liberated by treatment with aqueous sodium bicarbonate solution. Condensation of 5-nitropyridine-2-carboxaldehyde and 4-methyl-5-nitropyridine-2-carboxaldehyde (6 and 8, respectively), with thiosemicarbazide in aqueous ethanol, yielded the corresponding 5-nitro-substituted thiosemicarbazones, 21 and 22.

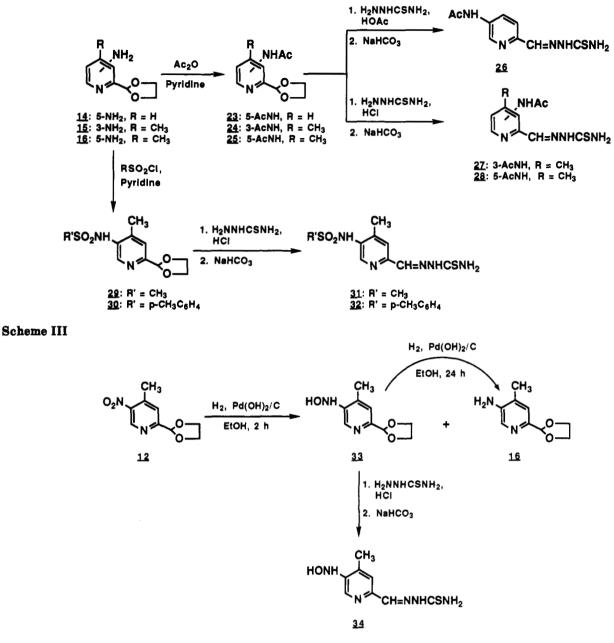
The acetamide and alkylsulfonamide derivatives of 3-amino- and 5-aminopyridine-2-carboxaldehyde thiosemicarbazone (26-28 and 31, 32, respectively) were prepared as described in Scheme II. Acetylation of compounds 14-16 with acetic anhydride in anhydrous pyridine gave the acetamide derivatives 23-25, which were then condensed with thiosemicarbazide to produce 5-(acetylamino)pyridine-2-carboxaldehyde thiosemicarbazone (26) and 3- and 5-(acetylamino)-4-methylpyridine-2-carboxaldehyde thiosemicarbazones (27 and 28, respectively). During the process of acidic hydrolysis of the ethylene acetal groups, some hydrolysis of the acetamide functions occurred even though reaction conditions were carefully controlled. The pure desired compounds were obtained, however, by recrystallization from ethanol or by silica gel chromatography. Treatment of 16 with methanesulfonyl chloride or p-toluenesulfonyl chloride in anhydrous pyridine afforded the corresponding 5-(methylsulfonyl)amino and 5-(p-tolylsulfonyl)amino derivatives, 29 and 30, respectively, which were then treated with thiosemicarbazide in the presence of concentrated hydrochloric acid to afford the corresponding 5-(methylsulfonyl)amino and 5-(p-tolylsulfonyl)amino derivatives of 4-methylpyridine-2-carboxaldehyde thiosemicarbazones, 31 and 32.

5-(Hydroxyamino)-4-methylpyridine-2-carboxaldehyde thiosemicarbazone (34) was synthesized by the procedure described in Scheme III. Hydrogenation of compound 12 in ethanol using $Pd(OH)_2/C$ as a catalyst under 50 psi of hydrogen yielded the 5-hydroxyamino derivative 33 contaminated with about 10-15% of the corresponding 5-amino derivative 16. Compound 33 was easily purified by recrystallization from ethanol. The structure of 33 was assigned by NMR, mass spectroscopy, and elemental analysis. During the reduction process, the rate of absorption of hydrogen decreased considerably after the formation of the 5-hydroxyamino derivative 33 and the reaction was terminated at this stage. However, when the reaction was allowed to proceed until the absorption

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Scheme II



of hydrogen was complete (about 24 h), the 5-amino derivative 16 was obtained in nearly quantitative yield. Condensation of 33 with thiosemicarbazide in the presence of concentrated hydrochloric acid, followed by treatment with sodium bicarbonate, afforded the desired 5-(hydroxyamino)-4-methylpyridine-2-carboxaldehyde thiosemicarbazone (34).

Biological Evaluation

The tumor-inhibitory properties of the substituted pyridine-2-carboxaldehyde thiosemicarbazones were determined by measuring their effects on the survival time of CD_2F_1 mice bearing the L1210 leukemia. Compounds were administered in suspension by intraperitoneal (ip) injection to groups of 5–10 tumor-bearing mice by the methodology previously described;¹¹ the results of these tests are shown in Table I. The 5-nitro derivatives, 21 and 22, the acetyl derivatives, 26-28, and the alkylsulfonamide derivatives, 31 and 32, were inactive and the results of these tests are not included. A wide range of dosage levels of each compound of from 10 to 60 mg/kg per day \times 6 were tested; however, only the prolonged life span produced by the maximum effective daily dose of each compound is listed. For comparison, the effects of 5-HP, the agent of this series that has undergone clinical trial, was also included as a positive control. The 3-amino derivatives, 17 (3-AP) and 19 (3-AMP), were comparable in their antitumor efficacy against the L1210 leukemia and were the most effective of all the agents synthesized in prolonging the survival time of mice bearing this neoplasm. The 5-amino derivatives, 18 (5-AP) and 20 (5-AMP) and the 5-hydroxyamino derivative, 34, were comparable to 5-HP as antineoplastic agents. The % T/C value of 5-HP against the L1210 leukemia obtained in these experiments was similar to our previous results with 5-HP;7 however, a significant difference existed between our value for this agent (% T/C = 133) and that (% T/C

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Table I. Effects of Pyridine-2-carboxaldehyde Thiosemicarbazone Derivatives on the Survival Time of Mice Bearing the L1210 Leukemia

compd	R1	R²	R ³	optimum daily dosage, ^a mg/kg	av∆ wt, ^b %	av survival, days	T/C × 100°				
5-HP	Н	Н	ОН	40	+2.0	10.4	133				
17	NH_2	Н	H	40	-5.9	14.6	187				
18	Н	н	NH_2	20	-2.8	11.0	140				
19	NH_2	Me	н	60	+2.0	14.9	190				
20	н	Me	NH_2	20	-7.0	10.8	138				
34	Н	Me	HONH	10	-2.7	10.6	136				

^a Drugs were administered in suspension by intraperitoneal injection, beginning 24 h after tumor implantation, once daily for 6 consecutive days, with 5–10 mice per group. ^b Average change in body weight from onset to termination of the rapy. $^{\rm c}$ T/C \times 100 represents the ratio of the survival time of treated to control animals \times 100. The average survival time of untreated L1210 tumor-bearing control animals was 7.8 days.

= 268) reported by French and Blanz.⁸ This may be due to differences between the L1210 leukemia cell lines employed and/or to differences in the schedule of drug administration. Although 5-HP was administered daily by intraperitoneal injection starting 24 h after tumor inoculation in both studies, our experiments employed six daily treatments, while Blanz and French¹² used daily treatments that were continued until 50% of the animals had died.

The four most active compounds, i.e., the 3-amino derivatives, 3-AP (17) and 3-AMP (19) and the 5-amino derivatives, 5-AP (18) and 5-AMP (20), were further screened against L1210 leukemia bearing CD_2F_1 female mice with a schedule of drug administration of twice a day for six consecutive days. The results of this investigation are summarized in Table II. 3-AMP showed the least toxicity in this series and the group that received two daily doses of 40 mg/kg of 3-AMP gave a % T/C value of 255 and 40% 60-day long-term survivors. Conversely, the groups receiving two daily doses of 30 mg/kg of 3-AP and two daily doses of 30 mg/kg of 5-AP died after an average of 7.3 days (% T/C = 96) and 6.9 days (% T/C = 91), respectively. The dramatic loss in body weight post drug injection (an average decrease of 12.8% and 16.0% from the original body weight, respectively) suggests that their deaths were attributable to drug toxicity rather than to the leukemia. However, 3-AP and 5-AP showed much better antitumor activity at lower dosages and the groups that received two daily doses of 10 mg/kg of 3-AP and two daily doses of 10 mg/kg of 5-AP gave % T/C values of 246 and 232, and produced 40% 60-day long-term survivors, and 10% 60-day long-term survivors, respectively. 5-AMP was less toxic than 3-AP and 5-AP and its antitumor activity was at least equal to that of 5-AP. Comparison of the results in Table I and Table II shows that the activity of compounds 3- and 5-AP and 3- and 5-AMP were schedule dependent with much better therapeutic effects obtained by twice daily than by once daily treatment. 3-AP

prepared by a reported procedure⁹ with minor modification. A mixture of 3-nitropyridine¹⁰ (1; 1.3 g, 9.4 mmol) and selenium dioxide (1.1 g, 9.4 mmol) in 1,4-dioxane (30 mL), containing 0.8 mL of water, was refluxed under an atmosphere of nitrogen for 20 h. The reaction mixture was cooled and filtered to remove the precipitated black selenium. The filtrate was evaporated in vacuo to dryness, and the residue was chromatographed on a silica gel column (CH₂Cl₂, R_f 0.82) to afford 0.32 g (23%) of white crystals: mp 61-62 °C (lit.9 mp 63 °C); 1H NMR (90 MHz, CDCl₃) δ 7.65 (dd, 1 H, 5-H, $J_{4,5}$ = 6.0 Hz, $J_{5,6}$ = 4.5 Hz), 8.32 (d, 1 H, 4-H, $J_{4,5} = 6.0$ Hz), 9.05 (d, 1 H, 6-H, $J_{5,6} = 4.5$ Hz), 10.31 (s, 1 H, 2-CHO).

5-Nitropyridine-2-carboxaldehyde (6). This compound was prepared from the nitro derivative 2 by the procedure employed for the synthesis of 5, except that anhydrous dioxane was used as the solvent and the reaction time was 4 h: yield 2.0 g (42%);

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Table II. Effects of 3-AP, 5-AP, 3-AMP, and 5-AMP Administered Twice Daily on the Survival Time of Mice Bearing
the L1210 Leukemia

compd	daily dosage,ª mg/kg	av ∆ wt,° %	av survival, days ^c	T/C × 100 ^d	long-term survivors ^e
3-AP (17)	10×2	-6.4	18.7	246	4/10
	15×2	-3.3	19.8	262	0/5
	20×2	-12.4	14.6	192	1/10
	30×2	-12.8	7.3	96	0/5
5-AP (18)	10×2	-3.8	17.6	232	1/10
	15×2	-7.7	16.8	221	0/5
	20×2	-12.3	14.0	185	1/10
	30×2	-16.0	6.9	91	0/5
3-AMP (19)	10×2	+1.4	14.8	195	2/10
	20×2	-3.8	16.9	222	0/10
	30×2	-2.4	17.6	232	1/10
	40×2	-8.1	19.4	255	2/5
5-AMP (20)	10×2	+1.4	18.6	245	1/10
()	15×2	-7.1	21.0	276	0/10
	20×2	-12.1	17.9	236	0/10
	30×2	-12.4	14.4	190	0/5

^a Drugs were administered in suspension by intraperitoneal injection, beginning 24 h after tumor implantation, twice daily for 6 consecutive days, with 5-10 mice per group. ^b Average change in body weight from onset to termination of therapy. ^c Average survival time includes only those mice that died prior to day 60. $^{\circ}$ T/C \times 100 represents the ratio of the survival time of treated to control animals \times 100. The average survival time of the untreated tumor-bearing control animals was 7.6 days. e Long-term survivors are the number of mice that survived for >60 days relative to the total number of treated mice.

the L1210 leukemia to any other agent in this series of α -(N)-heterocyclic carboxaldehyde thiosemicarbazones reported to date^{8,13} and, therefore, further evaluation of these two promising agents as potential anticancer drugs is merited.

Experimental Section

321 - 356

Melting points were determined with a Thomas-Hoover Unimelt apparatus and are uncorrected. ¹H NMR spectra were recorded on a Varian EM-390 90 MHz NMR spectrometer or a Bruker WM-500 500 MHz spectrometer with Me₄Si as the internal reference. The mass spectra (at 70 eV) were provided by the Yale University Chemical Instrumentation Center. TLC was performed on EM precoated silica gel sheets containing a fluorescent indicator. Elemental analyses were carried out by the Baron Consulting Co., Orange, CT. Where analyses are indicated only by symbols of the elements, the analytical results for those elements were within $\pm 0.4\%$ of the theoretical value.

and 3-AMP appear to be superior in their activities against

³⁻Nitropyridine-2-carboxaldehyde (5). This compound was

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mp 66–67 °C (lit.⁹ mp 66.5–67.5 °C); TLC R_f 0.85 (EtOAc); ¹H NMR (90 MHz, CDCl₃) δ 8.00 (d, 1 H, 3-H, $J_{3,4}$ = 4.5 Hz), 8.30 (d, 1 H, 4-H, $J_{3,4}$ = 4.5 Hz), 9.25 (s, 1 H, 6-H), 10.45 (s, 1 H, 2-CHO).

2-(1,3-Dioxolan-2-yl)-3-nitropyridine (9). A mixture of aldehyde 5 (1.5 g, 9.7 mmol), ethylene glycol (7.0 g, 62 mmol), and p-toluenesulfonic acid monohydrate (60 mg, 0.24 mmol) in toluene (150 mL) was refluxed until the starting material was no longer observed by TLC (CH₂Cl₂/EtOAc, 4:1, v/v). The reaction mixture was cooled and washed with aqueous sodium bicarbonate solution, water, and brine. The organic layer was dried over MgSO₄. The filtrate was evaporated in vacuo to dryness, and the residue was chromatographed on a silica gel (120 g) column (CH₂Cl₂/EtOAc, 4:1, v/v; R_f 0.50). The product was obtained as almost colorless needles (1.65 g, 87%): mp 65-67 °C; ¹H NMR (90 MHz, CDCl₃) δ 4.15 (s, 4 H, CH₂CH₂), 6.52 (s, 1 H, 2-CH), 7.52 (m, 1 H, 5-H), 8.15 (dd, 1 H, 4-H, $J_{4,5} = 7$ Hz, $J_{4,6} = 1$ Hz), 8.85 (dd, 1 H, 6-H, $J_{5,6} = 4$ Hz, $J_{4,6} = 1$ Hz). Anal. (C₈H₈N₂O₄) C, H, N.

2-(1,3-Dioxolan-2-yl)-5-nitropyridine (10). This compound was prepared from 6 by the procedure employed for the synthesis of 9: yield 2.0 g (84%); mp 104–105 °C; TLC R_f 0.73 (CH₂Cl₂/ EtOAc, 1:1, v/v); ¹H NMR (90 MHz, CDCl₃) δ 4.10 (s, 4 H, CH₂-CH₂), 5.92 (s, 1 H, 2-CH), 7.72 (d, 1 H, 3-H, $J_{3,4}$ = 8 Hz), 8.50 (dd, 1 H, 4-H, $J_{3,4}$ = 8 Hz, $J_{4,6}$ = 2 Hz), 9.42 (d, 1 H, 6-H, $J_{4,6}$ = 2 Hz). Anal. (C₈H₈N₂O₄) C, H, N.

3-Amino-2-(1,3-dioxolan-2-yl)pyridine (13). A solution of 9 (1.0 g, 5.1 mmol) in ethanol (100 mL) was hydrogenated overnight in a Parr apparatus at 50 psi of hydrogen in the presence of 10% Pd/C (0.1 g). The reaction mixture was filtered through a Celite pad and the catalyst was washed with ethanol. The combined filtrate and washings were evaporated in vacuo to dryness and coevaporated with benzene. The resulting solid was recrystallized from benzene to afford 0.81 g (96%) of product as white crystals: mp 73-74 °C; TLC R_f 0.4 (EtOAc); ¹H NMR (90 MHz, CDCl₃) δ 4.10 (m, 4 H, CH₂CH₂), 4.15 (br s, 2 H, 3-NH₂, D₂O exchangeable), 5.80 (s, 1 H, 2-CH), 6.90 (m, 2 H, 4-H and 6-H), 7.95 (dd, 1 H, 5-H). Anal. (C₈H₁₀N₂O₂·0.1H₂O) C, H, N.

5-Amino-2-(1,3-dioxolan-2-yl)pyridine (14). This compound was prepared from 10 by the procedure employed for the synthesis of 13: yield 2.6 g (93%); mp 81-82 °C; TLC R_f 0.18 (EtOAc); ¹H NMR (90 MHz, CDCl₃) δ 3.85 (br s, 2 H, 5-NH₂, D₂O exchangeable), 4.05 (m, 4 H, CH₂CH₂), 5.72 (s, 1 H, 2-CH), 6.95 (dd, 1 H, 4-H, $J_{3,4}$ = 8 Hz, $J_{4,6}$ = 2 Hz), 7.30 (d, 1 H, 3-H, $J_{3,4}$ = 8 Hz), 8.08 (d, 1 H, 6-H, $J_{4,6}$ = 2 Hz). Anal. (C₈H₁₀N₂O₂) C, H, N.

3-Aminopyridine-2-carboxaldehyde Thiosemicarbazone (17). To a solution of 13 (0.80 g, 4.8 mmol) in 10 mL of ethanol, 8 mL of water, and 2 mL of concentrated hydrochloric acid was added 0.48 g (5.3 mmol) of thiosemicarbazide. The mixture was stirred at room temperature overnight, refluxed for 1 h, cooled, and filtered. The crude yellow hydrochloride salt was dissolved in 50 mL of hot water and filtered. To the hot filtrate was added 10 mL of 5% sodium bicarbonate solution. The mixture was stirred at room temperature for 1 h, filtered, and washed with water, followed by ethanol: yield 0.72 g (77%); mp 240-241 °C dec; MS m/e 194 (M⁺); ¹H NMR (90 MHz, DMSO- d_6) δ 6.48 (br s, 2 H, 3-NH₂, D₂O exchangeable), 7.12 (m, 2 H, 4-H and 6-H), 7.83 (dd, 1 H, 5-H), 8.10 (br s, 2 H, CSNH₂, D₂O exchangeable), 8.10 (s, 1 H, 2-CH), 10.95 (s, 1 H, NNH, D₂O exchangeable). Anal. (C₇H₉N₅S) C, H, N.

5-Aminopyridine-2-carboxaldehyde Thiosemicarbazone (18). This compound was prepared from 14 by the procedure employed for the synthesis of 17: yield 1.9 g (82%); mp 205-207 °C; MS m/e 194 (M⁺); ¹H NMR (DMSO- d_6 , 500 MHz) δ 5.60 (br s, 2 H, 3-NH₂, D₂O exchangeable), 6.95 (dd, 1 H, 4-H, $J_{3,4}$ = 8 Hz, $J_{4,6}$ = 1.5 Hz), 7.65 (s, 1 H, 2-CH), 7.75 (d, 1 H, 6-H, $J_{4,6}$ = 1.5 Hz), 7.90 (d, 1 H, 3-H, $J_{3,4}$ = 8 Hz), 7.85 and 8.10 (two br s, 2 H, CSNH₂, D₂O exchangeable), 11.05 (s, 1 H, NNH, D₂O exchangeable). Anal. (C₇H₉N₅S-0.4H₂O) C, H, N.

3-Amino-4-methylpyridine-2-carboxaldehyde Thiosemicarbazone (19). This compound was prepared from 15^{14} by the procedure employed for the synthesis of 17: yield 0.5 g (76%); mp 227–228 °C; MS m/e 208 (M⁺); ¹H NMR (500 MHz, DMSOd₆) δ 2.25 (s, 3 H, 4-CH₃), 6.18 (s, 2 H, 3-NH₂, D₂O exchangeable), 7.01 (d, 1 H, 5-H, $J_{5,6} = 6$ Hz), 7.78 (d, 1 H, 6-H, $J_{5,6} = 6$ Hz), 7.90 (s, 2 H, CSNH₂, D₂O exchangeable), 8.32 (s, 1 H, 2-CH), 11.31 (s, 1 H, NNH, D₂O exchangeable). Anal. (C₈H₁₁N₅S·HCl·H₂O) C, H, N.

5-Amino-4-methylpyridine-2-carboxaldehyde Thiosemicarbazone (20). This compound was prepared from 16^{14} by the procedure employed for the synthesis of 17: yield 0.64 g (78%); mp 235-236 °C; MS m/e 208 (M⁺); ¹H NMR (500 MHz, DMSO d_6) δ 2.10 (s, 3 H, 4-CH₃), 5.48 (s, 2 H, 5-NH₂, D₂O exchangeable), 7.80 (s, 2 H, CSNH₂, D₂O exchangeable), 7.80 (s, 1 H, 3-H), 7.95 (s, 1 H, 6-H), 8.00 (s, 1 H, 2-CH), 11.50 (s, 1 H, NNH, D₂O exchangeable). Anal. (C₈H₁₁N₅S·HCl·H₂O) C, H, N.

5-Nitropyridine-2-carboxaldehyde Thiosemicarbazone (21). A mixture of 6 (0.50 g, 3.3 mmol) and thiosemicarbazide (0.36 g, 4 mmol) in 20 mL of 70% aqueous ethanol solution was refluxed for 2 h, cooled, and filtered. The yellow precipitate that formed was washed with water and recrystallized from ethanol to give 0.54 g (73%) of product: mp 215-217 °C; ¹H NMR (90 MHz, DMSO- d_6) δ 8.25 (d, 1 H, 3-H), 8.35 and 8.55 (two br s, 2 H, CSNH₂, D₂O exchangeable), 8.90 (dd, 1 H, 4-H), 9.75 (d, 1 H, 6-H), 10.15 (s, 1 H, 2-CH), 11.95 (s, 1 H, NNH, D₂O exchangeable). Anal. (C₇H₇N₅O₂S·HCl) C, H, N.

4-Methyl-5-nitropyridine-2-carboxaldehyde Thiosemicarbazone (22). This compound was prepared from 8 by the procedure employed for the synthesis of 21: yield 0.38 g (88%); mp 220-222 °C; ¹H NMR (90 MHz, DMSO- d_6) δ 2.55 (s, 3 H, 4-CH₃), 8.12 (s, 1 H, 3-H), 8.30 and 8.50 (two br s, 2 H, CSNH₂, D₂O exchangeable), 9.15 (s, 1 H, 6-H), 9.45 (s, 1 H, 2-CH), 11.85 (s, 1 H, NNH, D₂O exchangeable). Anal. (C₈H₉N₅O₂S-HCl) C, H, N.

5-(Acetylamino)-2-(1,3-dioxolan-2-yl)pyridine (23). To a stirred solution of 14 (2.0 g, 12 mmol) in 15 mL of anhydrous pyridine in an ice bath was added dropwise 2 mL of acetic anhydride at 0-5 °C. The reaction mixture was stirred overnight and evaporated in vacuo to dryness. The residue was co-evaporated with ethanol (10 mL) and recrystallized from ethanol to yield 2.1 g (82%) of product: mp 145-147 °C; ¹H NMR (90 MHz, CDCl₃) δ 2.05 (s, 3 H, CH₃), 4.10 (m, 4 H, CH₂CH₂), 5.82 (s, 1 H, 2-CH), 7.10 (dd, 1 H, 4-H, J_{3,4} = 8 Hz, J_{4,6} = 2 Hz), 7.45 (d, 1 H, 3-H, J_{3,4} = 8 Hz), 8.40 (br s, 1 H, NH, D₂O exchangeable), 8.68 (d, 1 H, 6-H, J_{4,6} = 2 Hz). Anal. (C₁₀H₁₂N₂O₃) C, H, N.

3-(Acetylamino)-2-(1,3-dioxolan-2-yl)-4-methylpyridine (24). A mixture of 15 (500 mg, 2.78 mmol), 5 mL of acetic anhydride and 15 mL of anhydrous pyridine was refluxed overnight and evaporated in vacuo to dryness. The residue was dissolved in CH₂CH₂ (30 mL), washed with 10% sodium bicarbonate, brine, and water, and then dried (anhydrous MgSO₄). The solvent was removed, and the residue was purified on a silica gel column (CH₂Cl₂/CH₃OH, 10:1, v/v; R_1 0.67) to produce 440 mg (72%) of product: mp 77-79 °C; ¹H NMR (90 MHz, CDCl₃) δ 2.10 (s, 3 H, COCH₃), 2.17 (s, 3 H, 4-CH₃), 2.20 (br s, 1 H, NH, D₂O exchangeable), 3.95 (m, 4 H, CH₂CH₂), 5.70 (s, 1 H, 2-CH), 7.15 (d, 1 H, 5-H, J_{5,6} = 6 Hz), 8.42 (d, 1 H, 6-H, J_{5,6} = 6 Hz). Anal. (C₁₁H₁₄N₂O₃) C, H, N.

5-(Acetylamino)-2-(1,3-dioxolan-2-yl)-4-methylpyridine (25). This compound was prepared from 16 by the procedure employed for the synthesis of 24: yield 0.5 g (82%); mp 98–99 °C; TLC R_{1} 0.65 (CH₂Cl₂/EtOH, 10:1, v/v); ¹H NMR (90 MHz, CDCl₃) δ 2.08 (s, 3 H, COCH₃), 2.15 (s, 3 H, 4-CH₃), 4.05 (m, 4 H, CH₂CH₂), 5.72 (s, 1 H, 2-CH), 7.30 (s, 1 H, 3-H), 8.05 (br s, 1 H, NH, D₂O exchangeable), 8.58 (s, 1 H, 6-H). Anal. (C₁₁H₁₄N₂O₃) C, H, N.

5-(Acetylamino)pyridine-2-carboxaldehyde Thiosemicarbazone (26). A mixture of 23 (0.6 g, 3.6 mmol), thiosemicarbazide (0.4 g, 4.4 mmol), 1 mL of glacial acetic acid, and 10 mL of ethanol was heated with stirring at 50 °C for 6 h, cooled, and filtered. The acetic acid salt was dissolved in hot water and filtered into 15 mL of 5% sodium bicarbonate solution, and the mixture was stirred at room temperature for 1 h. The yellow precipitate that formed was filtered, washed with water, and recrystallized from ethanol twice to give 0.46 g (54%) of product: mp 215-217 °C; ¹H NMR (90 MHz, DMSO- d_6) δ 2.05 (s, 3 H, COCH₃), 8.00 (m, 3 H, 2-CH, 3-H and 4-H), 8.05 and 8.15 (two br s, 2 H, CSNH₂, D₂O exchangeable), 8.80 (d, 1 H, 6-H, $J_{4,6} =$

⁽¹⁴⁾ Wang, Y. Q.; Liu, M. C.; Lin, T. S.; Sartorelli, A. C. Unpublished results.

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1.5 Hz), 10.30 (s, 1 H, 5-NH, D₂O exchangeable), 11.30 (s, 1 H, NNH, D₂O exchangeable). Anal. (C₉H₁₁N₅OS) C, H, N.

3-(Acetylamino)-4-methylpyridine-2-carboxaldehyde Thiosemicarbazone (27). A mixture of 24 (0.41 g, 1.9 mmol), thiosemicarbazone (0.2 g, 2.2 mmol), 1 mL of concentrated hydrochloric acid and 10 mL of ethanol was stirred at room temperature overnight. The yellow precipitate (hydrochloride salt) that formed was filtered and washed with water, followed by ethanol. The hydrochloride salt was dissolved in hot water and stirred with 10 mL of 5% sodium bicarbonate solution for 1 h, filtered, and washed with water. The crude product was chromatographed on a silica gel column (CH₂Cl₂/CH₃OH, 4:1, $v/v; R_f 0.52$) to give 0.21 g (45%) of product: mp 225-227 °C; ¹H NMR (90 MHz, DMSO-d₆) δ 2.05 (s, 3 H, COCH₃), 2.18 (s, 3 H, 4-CH₃), 7.30 (d, 1 H, 5-H, $J_{5.6}$ = 6 Hz), 8.14 (d, 1 H, 2-CH), 7.90 and 8.30 (two br s, 2 H, CSNH₂, D₂O exchangeable), 8.35 (d, 1 H, 6-H, $J_{5.6} = 6$ Hz), 9.71 (s, 1 H, 3-NH, D₂O exchangeable), 11.53 (s, 1 H, NNH, D₂O exchangeable). Anal. (C₁₀H₁₃N₅OS) C, H, N.

5-(Acetylamino)-4-methylpyridine-2-carboxaldehyde Thiosemicarbazone (28). This compound was prepared from 25 by the procedure employed for the synthesis of 27: yield 0.42 g (76%); mp 229–231 °C; TLC, R_f 0.52 (CH₂Cl₂/CH₃OH, 4:1, v/v); ¹H NMR (90 MHz, DMSO-d₆) δ 2.10 (s, 3 H, COCH₃), 2.25 (s, 3 H, 4-CH₃), 7.90 (s, 1 H, 3-H), 8.05 (s, 1 H, 2-CH), 8.05 and 8.35 (two br s, 2 H, CSNH₂, D₂O exchangeable), 8.60 (s, 1 H, 6-H), 9.62 (s, 1 H, 5-NH, D₂O exchangeable), 11.60 (s, 1 H, NNH, D₂O exchangeable). Anal. (C₁₀H₁₃N₅OS) C, H, N.

2-(1,3-Dioxolan-2-yl)-5-[(methylsulfonyl)amino]-4-methylpyridine (29). To a stirred solution of 16 (0.8 g, 4.4 mmol) in 10 mL of anhydrous pyridine in an ice bath was added dropwise 0.6 g (5.3 mmol) of methanesulfonyl chloride at 0-5 °C. The mixture was stirred at room temperature overnight and evaporated in vacuo to dryness. The residue was coevaporated with toluene (10 mL) and then partitioned between CH_2CH_2 (30 mL) and water (10 mL). The organic layer was washed with 10%sodium bicarbonate, brine, and water, dried with anhydrous MgSO₄, and filtered. The filtrate was concentrated to a small volume and purified on a silica gel column (EtOAc, R_f 0.40) to give 0.68 g of product: mp 128-130 °C; 1H NMR (90 MHz, CDCl₃) δ 2.40 (s, 3 H, 4-CH₃), 3.03 (s, 3 H, CH₃SO), 4.12 (m, 4 H, CH₂-CH₂), 5.65 (s, 1 H, 2-CH), 7.40 (s, 1 H, 3-H), 8.10 (br s, 1 H, NH, D_2O exchangeable), 8.52 (s, 1 H, 6-H). Anal. ($C_{10}H_{14}N_2O_4S$) C, H, N.

2-(1,3-Dioxolan-2-yl)-4-methyl-5-[(p-tolylsulfonyl)amino]pyridine (30). This compound was prepared from 16 with p-toluenesulfonyl chloride by a procedure similar to that employed for the synthesis of **29**: yield 0.75 g (77%); mp 155–156 °C; ¹H NMR (90 MHz, CDCl₃) δ 2.05 (s, 3 H, ArCH₃), 2.32 (s, 3 H, 4-CH₃), 4.10 (m, 4 H, CH₂CH₂), 5.70 (s, 1 H, 2-CH), 6.50 (br s, 1 H, NH, D₂O exchangeable), 7.20–7.40 (m, 5 H, ArH and 3-H), 8.20 (s, 1 H, 6-H). Anal. (C₁₆H₁₈N₂O₄S) C, H, N.

5-[(Methylsulfonyl)amino]-4-methylpyridine-2-carboxaldehyde Thiosemicarbazone (31). A mixture of 29 (0.93 g, 3.6 mmol), thiosemicarbazide (0.37 g, 4.0 mmol), and 10 mL of 5% concentrated hydrochloric acid solution was heated with stirring at 60 °C for 4 h and cooled. The yellow precipitate (hydrochloride salt) that formed was filtered and washed with a small amount of water. The hydrochloride salt was then stirred in 10 mL of 1 N NaOH solution for 30 min and filtered. The filtrate was neutralized with dilute acetic acid, filtered, and washed with water followed by ethanol to give 0.65 g (63%) of product: mp 210–212 °C; ¹H NMR (90 MHz, DMSO-d₆) δ 2.35 (s, 3 H, 4-CH₃), 3.05 (s, 3 H, 4-CH₃SO), 8.05 (s, 1 H, 3-H), 8.22 (s, 1 H, 6-H), 8.30 (s, 1 H, 2-CH), 8.15 and 8.35 (two br s, 2 H, CSNH₂, D₂O exchangeable), 9.45 (br s, 1 H, SO₂NH, D₂O exchangeable), 11.45 (s, 1 H, NNH, D₂O exchangeable). Anal. (C₉H₁₃N₅O₂S₂·0.75H₂O) C, H, N.

4-Methyl-5-[(p-tolylsulfonyl)amino]pyridine-2-carboxaldehyde Thiosemicarbazone (32). This compound was prepared from 30 with p-toluenesulfonyl chloride by the procedure employed for the synthesis of 31: yield 0.43 g (80%); mp 234-236 °C; ¹H NMR (90 MHz, DMSO- d_6) δ 2.12 (s, 3 H, ArCH₃), 2.40 (s, 3 H, 4-CH₃), 7.30-7.40 (m, 4 H, ArH), 8.05 (s, 1 H, 3-H), 8.20 (s, 1 H, 6-H), 8.30 (s, 1 H, 2-CH), 8.35 and 8.55 (two br s, 2 H, CSNH₂, D₂O exchangeable), 9.40 (br s, 1 H, SO₂NH, D₂O exchangeable), 11.55 (s, 1 H, NNH, D₂O exchangeable). Anal. (C₁₅H₁₇N₅O₂S₂) C, H, N.

2-(1,3-Dioxolan-2-yl)-5-(hydroxyamino)-4-methylpyridine (33). A solution of 12 (1.8 g, 8.6 mmol) in ethanol (100 mL) was hydrogenated for 2 h in a Parr apparatus at 50 psi of hydrogen in the presence of 20% Pd(OH)₂/C (0.2 g). The reaction mixture was filtered through a Celite pad, and the catalyst was washed with ethanol. The combined filtrate and washings were evaporated in vacuo to dryness. The residue was recrystallized from ethanol twice to give 1.0 g (60%) of product: mp 180–181 °C as white crystals; MS m/e 167 (M⁺ + 1); ¹H NMR (500 MHz, DMSO-d₆) δ 2.08 (s, 3 H, 4-CH₃), 3.90–4.05 (m, 4 H, CH₂CH₂), 5.54 (s, 1 H, 2-CH), 7.13 (s, 1 H, 3-H), 8.18 (s, 1 H, 6-H), 8.29 (s, 1 H, NH, D₂O exchangeable), 8.45 (s, 1 H, OH, D₂O exchangeable). Anal. (C₉H₁₂N₂O₃) C, H, N.

5-(Hydroxyamino)-4-methylpyridine-2-carboxaldehyde Thiosemicarbazone (34). This compound was prepared from 33 by the procedure employed for the synthesis of 17: yield 0.45 g (77%); mp 197–198 °C; MS m/e 224 (M⁺); ¹H NMR (90 MHz, DMSO- d_6) δ 2.35 (s, 3 H, 4-CH₃), 8.12 (m, 3 H, 3-H, 6-H, and 2-CH), 8.55 and 8.75 (two br s, 2 H, CSNH₂, D₂O exchangeable), 9.15 (br s, 2 H, HONH, D₂O exchangeable), 12.10 (s, 1 H, NNH, D₂O exchangeable). Anal. (C₈H₁₁N₅OS) C, H, N.

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