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Two-step syntheses of 2,4,6-triisopropylbenzenesulfonyl aziridines

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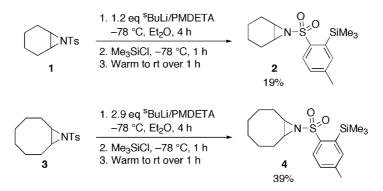
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Abstract—In a continuing study on the α -lithiation of *N*-tosyl aziridines, it is reported that *ortho*-lithiation of the *N*-tosyl group is occurring under typical α -lithiation conditions (*s*-BuLi/PMDETA). Thus, a simple, two-step synthesis of *N*-2,4,6-triisopropylbenzenesulfonyl aziridines was optimised. The route involves epoxide ring opening using a sulfonamide, mesylation and base-mediated ring closure. The scope and limitations of this route were assessed and five new *N*-2,4,6-triisopropylbenzenesulfonyl aziridines were prepared in good yields. Since this method was unsuitable for the preparation of cyclooctene aziridine, an alternative two-step method was developed. This method involved aminobromination of cyclooctene using NBS/2,4,6-triisopropylbenzenesulfonamide.

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Recent work in our group has focused on the alkyllithium-mediated α -lithiation of *N*-tosyl aziridines such as **1** and **3**.^{1–3} Suspecting that lithiation *ortho* to the sulfonamide of the *N*-tosyl group^{4–6} may be occurring under the reaction conditions, we decided to trap any intermediate organolithiums with Me₃SiCl. Thus, lithiation of *N*-tosyl aziridine **1** using *s*-BuLi in the presence of pentamethyldiethylenetriamine (PMDETA) at $-78 \,^{\circ}$ C for 4 h and subsequent trapping with Me₃SiCl gave substituted aziridine **2**⁷ in 19% yield together with recovered starting aziridine **1** (25% yield). Under comparable conditions, aziridine **3** gave **4**⁸ in 39% yield (with 35% recovered starting material). In contrast to cyclooctene oxide,⁹ no trapping at the α -position of the aziridine was observed. Clearly, the *N*-tosyl-aziridino protecting group is not an innocent bystander in alkyllithium reactions and this is probably why an excess of alkyllithium is typically employed for lithiation of *N*-tosyl aziridines.^{1–3}

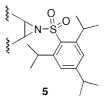
With a view to developing aziridine α -lithiation conditions that avoid excess alkyllithium as the base, we considered using *N*-2,4,6-triisopropylbenzenesulfonyl aziridines **5**. The 2,4,6-triisopropylbenzenesulfonyl group is of course well known from its use in the Shapiro reaction.^{4a,10} To our surprise, *N*-sulfonyl aziridines **5** have never previously been prepared and so a convenient route to these compounds needed to be devised.



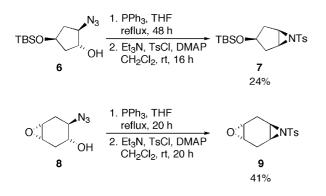
Keywords: Aziridines; Epoxides; Sulfonamides.

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Despite extensive studies on aziridination methodology,¹¹ there are in fact few synthetic methods for the synthesis of 'non-tosyl' *N*-sulfonyl aziridines. The most obvious approach is sulfonylation of the parent NH aziridine (normally prepared by Staudinger reduction of the corresponding azido alcohol,¹² itself synthesised from the epoxide and sodium azide). We have utilised such an approach for proof of stereochemistry in some of our previous studies (e.g. $6 \rightarrow 7$ and $8 \rightarrow 9$)^{13,14} but these reactions have always proved low yielding in our hands (24% yield of 7 and 41% yield of 9 are representative).

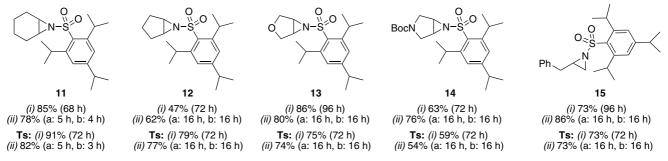


Alternative and popular approaches for the direct synthesis of aziridines from alkenes involve use of iodinanes PhI = NSO₂R^{15,16} (which can also be prepared in situ¹⁷) and *N*-chloramine sulfonamide salts RSO₂NCl⁻-Na⁺,^{18,19} both of which we have used in other studies.^{1,13,20} With iodinanes, Andersson et al. have reported the preparation of a few *N*-sulfonyl aziridines¹⁶ and the synthesis of *N*-SES-protected aziridines, pioneered by Dauban and Dodd, is well documented.^{11a,17,21} For *N*-chloramine sulfonamide salts, commercially available Chloramine-T is almost always used¹⁸ although Sharpless and co-workers have also had success with ^{*t*}BuSO₂NCl⁻Na⁺.¹⁹ Thus, these commonly employed methods each have their own limitations.

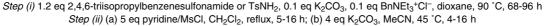
Due to the lack of widespread syntheses of different *N*-sulfonyl aziridines, we were attracted to a report by Albanese et al. on the three-step synthesis of *N*-tosyl aziridines starting from epoxides.^{22,23} The epoxides were ring opened with *p*-toluenesulfonamide and the resulting amino alcohols were tosylated and cyclised using K_2CO_3 . In one example, another sulfonamide (*p*-NO₂C₆H₄SO₂NH₂) was used. By slightly modifying Albanese's original methods,^{22a} we have now successfully synthesised a range of *N*-2,4,6-triisopropylbenzenesulfonyl aziridines **5** using a simple two-step protocol. In addition, we also developed a two-step synthesis of *N*-2,4,6-triisopropylbenzenesulfonyl cyclooctene aziridine using an aminobromination approach.²⁴ Herein, we describe our results.

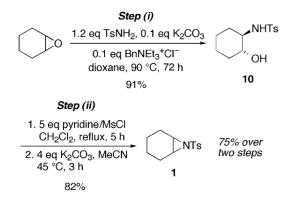
To start with, we optimised conditions for the synthesis of *N*-tosyl aziridine **1**, culminating in a two-step approach that delivered **1** in 75% overall yield. Thus, cyclohexene oxide was ring-opened using *p*-toluenesul-fonamide in refluxing dioxane in the presence of K_2CO_3 and $BnNEt_3^+Cl^-$ to give amino alcohol **10** in 91% yield after chromatography. An extended reaction time (72 h) was necessary for a high conversion to **10**. In an improvement to Albanese's route, we mesylated **10** and the crude mesylate was then directly cyclised to *N*-tosyl aziridine **1**. In this way, **1** was produced in 82% yield after chromatography. Our approach to **1** benefits from use of less equivalents of sulfonamide, a shorter mesylation reaction time and the need for only two chromatographic purifications; in addition, our overall yield of **1** is higher.

When *p*-toluenesulfonamide was replaced with *N*-2,4,6triisopropylbenzenesulfonamide, some of the reaction times for step (i) were increased to 96 h to optimise the yields. Using this two-step approach, aziridines **11–15** were prepared in satisfactory yields over the two steps (29-69%).²⁵ Despite *N*-2,4,6-triisopropylbenzenesulfonamide being more sterically hindered, the results obtained compared well with *p*-toluenesulfonamide for each of **11–15**. Notably, the synthesis of the *N*-tosyl analogue of dihydrofuran aziridine **13** using our protocol (56% yield over two steps) is more efficient (in terms of yield and overall reaction time) than a recently

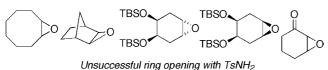




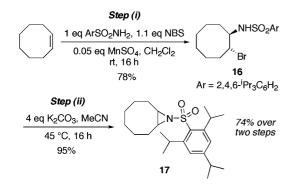




reported approach.²⁶ We briefly explored the scope of the approach and have identified some limitations. For example, ring opening of the five epoxides shown below using *p*-toluenesulfonamide were unsuccessful.



For our planned lithiation studies, the preparation of N-2,4,6-triisopropylbenzenesulfonyl cyclooctene aziridine was of particular importance. Therefore, since ring opening of cyclooctene oxide was unsuccessful we investigated another route to this aziridine based on aminobromination, as reported by Sudalai and co-workers.²⁴ Thus, reaction of cyclooctene with N-2,4,6-triisopropylbenzenesulfonamide in the presence of NBS and manganese sulfate gave amino bromide 16 in 78% yield after chromatography. In contrast to Sudalai's observation, we found that the yield of 16 was only moderately reduced (to 69%) if the manganese sulfate catalyst was omitted. Cyclisation of amino bromide 16 to aziridine 17 was readily accomplished in near-quantitative yield using K₂CO₃. Hence, this approach also represents a simple two-step synthesis of N-sulfonyl aziridines.



In summary, two straightforward routes to previously unknown N-2,4,6-triisopropylbenzenesulfonyl aziridines 11–15 and 17 have been developed. These approaches have utility for the preparation of 'non-tosyl' N-sulfonyl aziridines. The usefulness of such aziridines in alkyllithium-mediated reactions will be reported in due course.

Acknowledgements

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- 7. Data for **2**: colourless oil; $R_{\rm f}$ (1:1 petrol–Et₂O) 0.7; IR (CH₂Cl₂): 2957, 2856, 1320, 1158 cm⁻¹; ¹H NMR (400 MHz, CDCl₃): 7.85 (d, J = 1.0, 1H), 7.56 (d, J = 8.0, 1H), 7.29 (dd, J = 8.0, 1.0, 1H), 2.99–2.45 (m, 2H), 2.42 (s, 3H), 1.78–1.74 (m, 4H), 1.43–1.33 (m, 2H), 1.25–1.17 (m, 2H), 0.45 (s, 9H); ¹³C NMR (100.6 MHz, CDCl₃): 142.3, 141.1, 140.5, 137.3, 129.6, 129.1, 39.6, 22.7, 21.6, 19.5, 1.1; MS (CI, NH₃) m/z 2324 [100%, (M+H)⁺], 308 (70); HRMS (CI, NH₃) m/z calcd for C₁₆H₂₅NO₂SiS (M+H)⁺ 324.1452, found 324.1454.
- 8. Data for 4: colourless oil; $R_{\rm f}$ (1:1 petrol–Et₂O) 0.7; IR (CH₂Cl₂): 2928, 2856, 1320, 1246, 1161 cm⁻¹; ¹H NMR (400 MHz, CDCl₃): 7.87 (d, J = 8.0, 1H), 7.55 (br s, 1H), 7.28 (dd, J = 8.0, 1.0, 1H), 2.82–2.73 (m, 2H), 2.42 (s, 3H), 2.03–1.98 (m, 2H), 1.58–1.25 (m, 10H), 0.42 (s, 9H); ¹³C NMR (100.6 MHz, CDCl₃): 142.5, 141.2, 140.4, 137.2, 129.6, 129.0, 43.7, 26.4, 26.1, 25.2, 21.6, 1.1; MS (CI, NH₃) *m*/*z* asc2 [100%, (M+H)⁺], 336 (40); HRMS (CI, NH₃) *m*/*z* calcd for C₁₈H₂₉NO₂SiS (M+H)⁺ 352.1770, found 352.1767.
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cooling, CH₂Cl₂ (5 mL) was added and the solids were removed by filtration through Celite. The filtrate was dried (MgSO₄) and evaporated under reduced pressure to give the crude product. Purification by flash chromatography on silica with petrol-EtOAc (4:1) as eluent gave hydroxy sulfonamide (737 mg, 86%) as a white solid, mp 141-143 °C; R_f (1:1 petrol-Et₂O) 0.2; IR (CH₂Cl₂): 3496, 3250, 1314, 1150 cm⁻¹; ¹H NMR (400 MHz, CDCl₃): 7.17 (s, 2H), 4.83 (d, J = 8.0, 1H), 4.41 (td, J = 5.0, 3.0, 1H), 4.11 (sept, J = 7.0, 2H), 4.04 (dd, J = 10.0, 5.0, 1H), 3.96 (dd, J = 10.0, 5.0, 1H), 3.75–3.66 (m, 1H), 3.65 (dd, J = 10.0,3.0, 1H), 3.52 (dd, J = 10.0, 3.0, 1H), 2.91 (sept, J = 7.0, 1H), 2.35 (br s, 1H), 1.29–1.25 (m, 18H); ¹³C NMR (67.9 MHz, CDCl₃): 153.3, 150.3, 132.3, 124.0, 73.6, 71.3, 61.1, 34.1, 29.7, 24.84, 24.76, 23.5; MS (CI, NH₃) m/z 387 $[25\% (M+NH_4)^+]$, 370 $[100 (M+H)^+]$; HRMS (CI, NH₃) m/z calcd for $C_{19}H_{31}NO_4S$ (M+H)⁺ 370.2058, found 370.2052. A solution of hydroxy sulfonamide (500 mg, 1.36 mmol) in CH₂Cl₂ (7.5 mL) was added dropwise to a stirred solution of pyridine (0.47 mL, 6.78 mmol) and mesyl chloride (0.45 mL, 6.78 mmol) in CH₂Cl₂ (7.5 mL) at 0 °C under N₂. After 20 min, the solution was heated at reflux for 16 h. After cooling, the solution was washed with brine (15 mL), dried (MgSO₄) and evaporated under reduced pressure to give the crude mesylate. Then, a stirred mixture of the crude mesylate and K₂CO₃ (750 mg, 5.44 mmol) in MeCN (15 mL) was heated at 45 °C under N2 for 16 h. After cooling, CH2Cl2 (15 mL) was added and the organic solution was washed with water (20 mL) and brine (20 mL), dried (MgSO₄) and evaporated under reduced pressure. Purification by flash chromatography on silica with petrol-EtOAc (9:1) as eluent gave aziridine **13** (380 mg, 80%) as a white solid, mp 177–180 °C; $R_{\rm f}$ (1:1 petrol-Et₂O) 0.6; IR(CH₂Cl₂): 3089, 2955, 1313, 1152 cm⁻¹; ¹H NMR (400 MHz, CDCl₃): 7.18 (s, 2H), 4.33 (sept, J = 7.0, 2H), 3.99 (d, J = 10.0, 1H), 3.71 (d, J = 10.0, 1H), 3.68 (s, 2H), 2.91 (sept, J = 7.0, 1H), 1.26 (d, J = 7.0, 18H; ¹³C NMR (100.6 MHz, CDCl₃): 153.5, 150.9, 130.8, 123.8, 68.0, 44.6, 34.2, 29.6, 24.8, 23.5; MS (CI, NH₃) m/z 352 [100, (M+H)⁺]; HRMS (CI, NH₃) m/zcalcd for $C_{19}H_{29}NO_3S(M+H)^+$ 352.1952, found 352.1946.

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