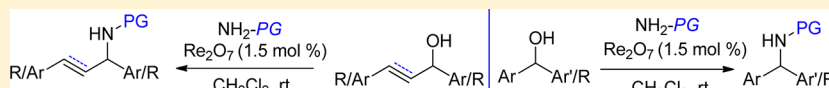


Direct Substitution of Hydroxy Group of π -Activated Alcohols with Electron-Deficient Amines Using Re_2O_7 Catalyst

Braja Gopal Das, Rajender Nallagonda, and Prasanta Ghorai*

Department of Chemistry, Indian Institute of Science Education and Research Bhopal, Bhopal-462023, India

S Supporting Information



ABSTRACT: The first example of simple Re_2O_7 -catalyzed direct dehydrative coupling between allylic alcohols with electron-deficient amines has been achieved under mild and open flask conditions. The protocol has also been successfully applied to benzylic and propargylic alcohols. The mechanistic proof for the $\text{S}_{\text{N}}1$ -type process has also been provided.

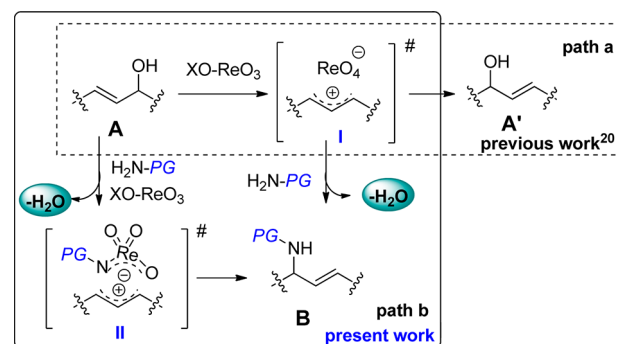
INTRODUCTION

The direct catalytic amination of *underivatized* π -activated alcohols, producing water as the only byproduct, is a very powerful strategy for the synthesis of various π -activated amines. In recent years, therefore, there have been continuous efforts to develop such reaction protocols using novel catalytic systems with improved efficiency.¹ The use of alcohol is limited by its leaving group ability despite the atom-economical and ecological advantage of this strategy.² For example, for the synthesis of allylic amines via π -allyl Pd complexes, which have been widely used,³ very few reports exist wherein an allylic alcohol is indeed used as a substrate^{4,5} compared to activated derivatives of allyl alcohols (esters, carbonates, carbamates, phosphates, halides, etc.)⁶ or allylic alcohols with activators.⁷ However, such palladium-catalyzed reactions are limited to electron-rich amines as nucleophiles; electron-deficient nitrogen nucleophiles did not undergo any reaction due to their low nucleophilicity. However, the higher nucleophilicity of the monoalkylated product in comparison to that of the starting amine results in further alkylation at the nitrogen center, which becomes a limiting factor in such a synthetic strategy.

On the other hand, the protocols with electron-deficient nitrogen nucleophiles have been developed using various Lewis acids^{8–17} or Brønsted acids.^{18,19} However, besides the advantages, most of these catalysts are *not broadly applicable to all classes* (namely, allyl, benzyl, and propargyl) of π -activated alcohols. Additionally, in many cases, the reaction using such acids is *restricted to aromatic amines only*, or *disubstituted amines are used* to avoid overalkylation. Therefore, a milder catalyst with high oxophilicity as well as applicability to all classes of π -activated alcohols toward simple monoprotected amines is highly desirable.

Recently, the metal-oxo complexes display high catalytic activity for allyl alcohol transposition, and oxo-rhenium(VII) catalysts have been shown to be superior in terms of reactivity and chemoselectivity, with no competitive oxidation (path a, Scheme 1).²⁰ Unfortunately, these oxo-rhenium catalysts have never been utilized for allylic amination reactions.²¹ In continuation of our research directed toward the development

Scheme 1. Oxo-Re Catalysis^a



^aPath a: allylic transposition of allyl alcohol. Path b: working hypothesis of allylic amination of allyl alcohol.

of atom-economic reactions catalyzed by oxo-rhenium catalysts²² because of their mild, moisture/air-tolerant nature,²³ we, *for the first time*, initiated a study of oxo-rhenium-catalyzed direct substitution of allylic alcohols toward various electron-deficient amines. Hypothetically, there could be a possibility of the reaction proceeding via an intramolecular aza-Cope rearrangement of in situ formed aza-/oxa-rhenate intermediate (II), which might enhance the rate of the reaction (path b, Scheme 1).^{24,25}

RESULTS AND DISCUSSION

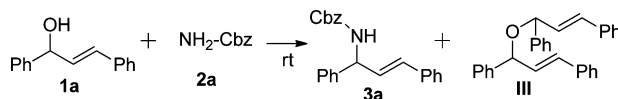
Table 1 summarizes the results of the catalyst screening of the allylic amination reaction of the diphenyl allylic alcohol (1a) with Cbz-NH₂ (2a) as the model substrate. As shown in entry 1, using MTO as catalyst, the desired product was given in low yield and ether (III) was obtained as the major product.

Among different oxo-rhenium(V) catalysts, (entries 2–4, Table 1), $\text{ReCl}_3\text{O}(\text{SMe}_2)(\text{OPPh}_3)$ showed moderate reactivity (60% conversion only). Interestingly, the best catalytic activity

Received: April 6, 2012

Published: June 11, 2012

Table 1. Optimization of the Reaction Conditions



	catalyst (1.5 mol %)	solvent	min	3a [%] ^a
1	MeReO ₃ (MTO)	CH ₂ Cl ₂	15	18
2	ReCl ₃ O(PPh ₃) ₂	CH ₂ Cl ₂	15	0
3	ReIO ₃ (PPh ₃) ₂	CH ₂ Cl ₂	15	6
4	ReCl ₃ O(SMe ₂)(OPPh ₃)	CH ₂ Cl ₂	15	61
5	Re ₂ O ₇	CH ₂ Cl ₂	10	100 (99) ^b

^aConversion was determined by ¹H NMR spectroscopy of the reaction mixture using *p*-anisaldehyde as internal standard. ^bIn parentheses, the yield of the isolated product after column chromatography.

was shown by using 1.5 mol % Re₂O₇ at room temperature in CH₂Cl₂; an almost quantitative yield of the corresponding allylamine (3a) was obtained with excellent chemoselectivity within 10 min. However, the reaction took place in a less-effective manner, using solvents, such as THF (74%), Et₂O (48%), toluene (36%), and MeOH (0%).

To broaden the scope of amines used, we carried out the reaction of the same alcohol with several protected amines (summarized in Table 2) under the optimized conditions as mentioned above. As shown in entries 1–7, a variety of primary amines protected with sensitive carbamates, such as Cbz-, Boc-, and EtOCO-, primary amides (Ac-, Bz-), and tosylate (Ts-, pNosyl-) participate in the reaction to provide excellent yields of the corresponding amines (3a–3g, respectively). As shown in entries 8–10, cyclic amides also reacted smoothly (3h–3j, respectively). Nevertheless, the aromatic amine (2k) also provided the allylamine (3k). However, benzoylhydrazine (2l) remained unreactive under similar reaction conditions (entry 12).

The scope of the reaction with respect to the alcohol substrate is summarized in Table 3. For example, nonbenzylic acyclic allylic alcohols (1b and 1c, entries 1 and 2) were

successful in reaction with protected amine (2a). Interestingly, the reaction of 1b afforded the linear amine (4a), whereas 1c provided a more substituted allyl amine (4b), in a highly regioselective manner in both the cases (Table 3, entries 1 and 2). The regioselective formation of 4b probably resulted from the most stabilized allyl-cation intermediates (e.g., intermediate like I or II in Scheme 1), which indirectly supports the S_N1-like reaction pathway. Nonbenzylic cyclic alcohols (entries 3–6) also worked efficiently under these conditions.

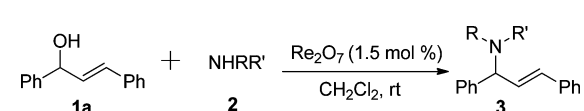
Next, we applied the present protocol for propargylic alcohols (Table 3, 1g–1j) with the above-mentioned amines (entries 7–12). Although the reaction time was longer than the allylation reaction, notably, only propargylic products were observed with high regioselectivity, without any interference from the allenic product.²⁶

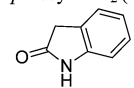
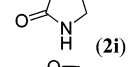
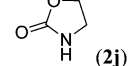
Finally, N-benzoylation of various amides was also examined from benzyl alcohols in the presence of the Re₂O₇ catalyst, as summarized in Table 4. For example, Cbz-NH₂ (2a) and Ts-NH₂ (2f) with diphenyl alcohol (5a) proceeded smoothly to afford the corresponding desired products in high yield (entries 1 and 2, respectively). The cyclic amide (2h) also worked well, except it required a longer reaction time (entry 3). Other diaryl alcohols (5b–5e) underwent conversion to provide the desired products (entries 4–7).

To gain further insight into the mechanism of this Re₂O₇ catalysis, we performed the following reactions. An optically active alcohol 1c-crl with 50% ee was treated with 2a under the specified reaction conditions, and the racemic product 4b was obtained (see Scheme 2). This result also suggested that the substitution probably proceeded through an S_N1-like mechanism.

CONCLUSIONS

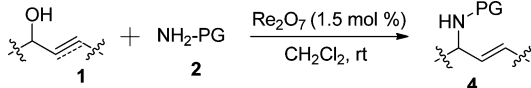
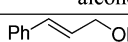
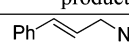
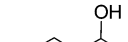
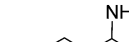
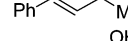
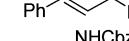
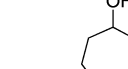
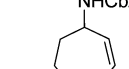
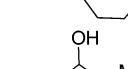
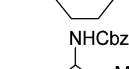
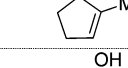
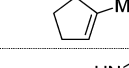
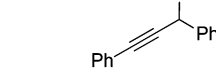
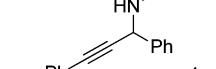
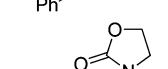
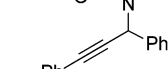
In conclusion, the ligand- and additive-free simple oxo-rhenium catalyst (1.5 mol % Re₂O₇) has been utilized, for the first time, for the direct substitution of allylic alcohols under mild and open flask conditions. Also, other π -activated alcohols, namely, benzylic and propargylic alcohols, worked smoothly under

Table 2. Allylic Amination Reaction^{a,b}


z	NHRR', 2	min	3	[%] ^a
1	Cbz-NH ₂ (2a)	10	3a	99
2	Boc-NH ₂ (2b)	15	3b	99
3	EtOCO-NH ₂ (2c)	60	3c	99
4	Ac-NH ₂ (2d)	4h ^b	3d	95
5	Bz-NH ₂ (2e)	60	3e	97
6	Ts-NH ₂ (2f)	10	3f	95
7	pNosyl-NH ₂ (2g)	60	3g	93
8	 (2h)	10	3h	99
9	 (2i)	20h	3i	99
10	 (2j)	30	3j	99
11	<i>p</i> -NO ₂ -C ₆ H ₄ NH ₂ (2k)	6h ^b	3k	98
12	PhCO-NHNH ₂ (2l)	12	-	0

^aIsolated yield after column chromatography. ^bThe reaction was carried out at 45 °C.

Table 3. Amination of Allylic and Propargylic Alcohols^{a,b,c}

				
	alcohol, 1	2 , [h]	product, 4	
			[%] ^a	
1		2a , 4 ^b		61
2		2a , 3		99 ^c
3		2a , 6		76
4		2f , 4		89
5		2a , 6		61
6		2a , 6		72
7		2a , 24		68
8		2j , 24		83
9		2f , 12 ^b		77
10		2a , 15		74
11		2a , 6		91
12		2a , 12		62

^aIsolated yield after column chromatography. ^bThe reaction was carried out at 45 °C. ^cOnly regioisomer observed.

these conditions. Excellent chemoselective formation of only the monoalkylated product was observed. Finally, the proof for the S_N1-type process has also been developed.

EXPERIMENTAL SECTION

General Experimental Methods. All reagents and solvents were used as supplied commercially. Commercial Re₂O₇, ranging in color from yellow to brown-black, was stored in a desiccator over CaCl₂. Reactions were conducted in an open atmosphere. Analytical thin-layer chromatography (TLC) was performed on silica gel 60 F254 plates. Visualization was accomplished with UV light (254 nm) and exposure to either phosphomolybdic acid (PMA)-CeSO₄/10% H₂SO₄, anisaldehyde, or KMnO₄ solution, followed by heating. Melting points are uncorrected. ¹H NMR spectra were acquired on either a 400 or a

500 MHz MNR spectrometer, and chemical shifts are reported relative to the residual solvent peak. ¹³C NMR spectra were acquired on either a 100 or a 125 MHz NMR spectrometer, and chemical shifts are reported in parts per million relative to the residual solvent peak. Unless noted, NMR spectra were acquired in CDCl₃; individual peaks are reported as multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet), integration, and coupling constants in hertz (Hz). All IR spectra were obtained as neat films, and selected absorbances are reported in cm⁻¹. Low-resolution and high-resolution mass spectrometry data were acquired using a Q-TOF analyzer in CH₃CN/H₂O as solvent (unless otherwise noted).

Standard Procedure of Amination Reaction from π -Activated Alcohol and Amines. To a stirred solution of alcohol (0.5 mmol) and amine (0.6 mmol, 1.2 equiv) in CH₂Cl₂ (2 mL) was added Re₂O₇ (0.0075 mmol, 1.5 mol %), and the solution was stirred

143.2, 139.6, 137.7, 136.0, 132.1, 129.4, 128.7, 128.4, 128.2, 127.9, 127.8, 127.3, 127.0, 126.5, 59.7, 21.4; HRMS (ESI, m/z) [$M + Na$]⁺ calculated for C₂₂H₂₁NNaO₅S, 386.1191; found, 386.1199.

(*E*)-*N*-(1,3-Diphenylallyl)-4-nitrobenzenesulfonamide (**3g**): 0.175 g, 93% yield; R_f = 0.24 (30:70 = EtOAc/*n*-hexane); colorless solid; mp 156 °C; IR (neat) 3279, 3028, 2914, 1606, 1528, 1347, 1311, 1160, 1090, 1027, 971, 853, 736, 697 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) 8.12 (d, J = 8.8 Hz, 2H), 7.87 (d, J = 8.80 Hz, 2H), 7.28–7.19 (m, 10H), 6.45 (d, J = 15.92 Hz, 1H), 6.14 (dd, J = 6.90 and 9.04 Hz, 1H), 5.39 (s, 1H), 5.27 (d, J = 6.67 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) 149.6, 146.6, 138.6, 135.4, 133.0, 128.9, 128.6, 128.4, 128.3, 127.3, 127.1, 126.4, 123.9, 60.2; HRMS (ESI, m/z) [$M - H$]⁺ calculated for C₂₁H₁₇N₂O₄S, 393.0908; found, 393.0918.

(*E*)-1-(1,3-Diphenylallyl)indolin-2-one (**3h**): 0.161 g, 99% yield; R_f = 0.15 (20:80 = EtOAc/*n*-hexane); light brown liquid; IR (neat) 3055, 3028, 1712, 1612, 1450, 1338, 1199, 968 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) 7.54–7.22 (12H), 7.19–6.93 (2H), 6.90–6.73 (2H), 6.66 (d, J = 7.6 Hz, 1H), 6.52 (d, J = 6.4 Hz, 1H), 3.69 (s, 2H); ¹³C NMR (100 MHz, CDCl₃) 174.9, 143.0, 137.8, 136.2, 135.0, 128.7, 128.6, 128.1, 127.6, 127.4, 127.1, 126.7, 124.6, 124.5, 124.3, 122.1, 111.2, 56.0, 35.8; HRMS (ESI, m/z) [$M + Na$]⁺ calculated for C₂₃H₂₀NO, 326.1545; found, 326.1544.

(*E*)-1-(1,3-Diphenylallyl)pyrrolidin-2-one (**3i**): 0.137 g, 99% yield; R_f = 0.15 (10:90 = EtOAc/*n*-hexane); colorless solid; mp 63–65 °C; IR (neat) 3028, 1681, 1492, 1415, 1265, 972 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) 7.55–7.20 (10H), 6.65 (d, J = 16 Hz, 1H), 6.48 (dd, J = 6.4, 15.6 Hz, 1H), 6.12 (d, J = 6.8 Hz, 1H), 3.52–3.40 (1H), 3.23–3.09 (1H), 2.70–2.40 (2H), 2.20–1.90 (2H); ¹³C NMR (100 MHz, CDCl₃) 174.7, 138.7, 136.4, 133.4, 128.7, 128.6, 127.9, 127.8, 127.7, 126.5, 125.4, 56.2, 43.4, 31.2, 18.1; HRMS (ESI, m/z) [$M + H$]⁺ calculated for C₁₉H₂₀NO, 278.1545; found, 278.1547.

(*E*)-3-(1,3-Diphenylallyl)oxazolidin-2-one (**3j**):^{12b} 0.138 g, 99% yield; R_f = 0.23 (20:85 = EtOAc/*n*-hexane); colorless liquid; IR (neat) 3029, 1732, 1494, 1417, 1250, 974 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) 7.50–7.26 (10H), 6.71 (d, J = 16 Hz, 1H), 6.49 (dd, J = 6.4, 16 Hz, 1H), 5.84 (d, J = 6.4 Hz, 1H), 4.45–4.25 (2H), 3.68–3.57 (1H), 3.37–3.25 (1H); ¹³C NMR (100 MHz, CDCl₃) 158.2, 138.0, 136.2, 133.8, 128.9, 128.7, 128.17, 128.17, 127.8, 126.6, 124.8, 62.1, 58.4, 41.0; HRMS (ESI, m/z) [$M + Na$]⁺ calculated for C₁₈H₁₇NaNO₂, 302.1157; found, 302.1149.

(*E*)-*N*-(1,3-Diphenylallyl)-4-nitroaniline (**3k**):^{18c} 0.140 g, 98% yield; R_f = 0.3 (15:85 = EtOAc/*n*-hexane); yellow solid; mp 135–137 °C; IR (neat) 3390, 3375, 1600, 1500, 1275, 1111, 968 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) 8.15–8.03 (2H), 7.55–7.15 (10H), 6.70–6.55 (3H), 6.41 (dd, J = 6, 16 Hz, 1H), 5.23 (d, J = 6 Hz, 1H), 4.93 (b s, 1H); ¹³C NMR (100 MHz, CDCl₃) 152.1, 140.3, 138.5, 136.0, 132.3, 129.2, 128.7, 128.6, 128.2, 128.1, 127.1, 126.6, 126.2, 112.1, 60.0; HRMS (ESI, m/z) [$M + Na$]⁺ calculated for C₂₁H₁₈NaN₂O₂, 353.1266; found, 353.1278.

(*E*)-Benzyl 4-Phenylbut-3-enoate (**4a**): 0.080 g, 61% yield; R_f = 0.25 (10:90 = EtOAc/*n*-hexane); colorless oil; IR (neat) 3414, 3332, 3028, 2924, 1725, 1516, 1246, 964 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) 7.55–7.20 (10H), 6.54 (d, J = 15.6 Hz, 1H), 6.35–6.05 (1H), 5.17 (s, 2H), 4.92 (b s, 1H), 4.02 (2H); ¹³C NMR (100 MHz, CDCl₃) 156.2, 136.5, 131.8, 129.2, 128.6, 128.5, 128.3, 128.1, 127.7, 127.0, 126.4, 126.0, 125.7, 66.8, 43.2; HRMS (ESI, m/z) [$M + H$]⁺ calculated for C₁₇H₁₈NO₂, 268.1332; found, 268.1320.

(*E*)-Benzyl (4-Phenylbut-3-en-2-yl)carbamate (**4b**): 0.139 g, 99% yield; R_f = 0.2 (10:90 = EtOAc/*n*-hexane); colorless solid; mp 66–68 °C; IR (neat) 3321, 3028, 1697, 1527, 1450, 1234, 1049, 964 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) 7.60–7.15 (10H), 6.54 (d, J = 16 Hz, 1H), 6.18 (dd, J = 6, 16 Hz, 1H), 5.15 (s, 2H), 4.83 (b s, 1H), 4.51 (b s, 1H), 1.36 (d, J = 6.8 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) 155.6, 136.6, 136.5, 131.1, 129.6, 128.6, 128.5, 128.1, 127.6, 126.4, 66.7, 48.4, 21.1; HRMS (ESI, m/z) [$M + 2H$]⁺ calculated for C₁₈H₂₀NO₂, 282.1494; found, 282.1493.

Benzyl Cyclohex-2-en-1-ylcarbamate (**4c**):¹⁴ 0.088 g, 76% yield; R_f = 0.26 (10:90 = EtOAc/*n*-hexane); colorless solid; mp 83–85 °C; IR (neat) 3313, 2916, 181, 1539, 1315, 1257, 1037 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) 7.50–7.30 (5H), 5.92–5.80 (1H), 5.68–5.55 (1H),

5.12 (1H), 4.78 (1H), 4.25 (1H), 2.10–1.85 (3H), 1.75–1.50 (3H); ¹³C NMR (100 MHz, CDCl₃) 155.6, 136.6, 130.8, 128.5, 128.4, 128.1, 128.0, 127.7, 66.5, 46.3, 29.7, 24.7, 19.5; HRMS (ESI, m/z) [$M + Na$]⁺ calculated for C₁₄H₁₈NO₂, 232.1338; found, 232.1359.

N-(Cyclohex-2-en-1-yl)-4-methylbenzenesulfonamide (**4d**):¹⁴ 0.112 g, 89% yield; R_f = 0.23 (5:95 = EtOAc/*n*-hexane); colorless solid; mp 98–100 °C; IR (neat) 3282, 2927, 1597, 1427, 1327, 1157, 1072 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) 7.79 (d, J = 7.6 Hz, 2H), 7.31 (d, J = 7.6 Hz, 2H), 5.76 (d, J = 10 Hz, 1H), 5.36 (d, J = 9.8 Hz, 1H), 4.66 (d, J = 8.4 Hz, 1H), 3.82 (s, 1H), 2.44 (s, 3H), 2.10–1.45 (6H); ¹³C NMR (100 MHz, CDCl₃) 143.2, 138.3, 131.5, 129.6, 127.06, 127.00, 48.9, 30.2, 24.4, 21.5, 19.3; HRMS (ESI, m/z) [$M + Na$]⁺ calculated for C₁₃H₁₇NNaO₂S, 274.0878; found, 274.0881.

Benzyl Cyclohept-2-en-1-ylcarbamate (**4e**): 0.075 g, 61% yield; R_f = 0.29 (5:95 = EtOAc/*n*-hexane); colorless solid; mp 83–85 °C; IR (neat) 3422, 3314, 2916, 1685, 1541, 1400, 1254, 1038 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) 7.46–7.30 (5H), 5.95–5.70 (1H), 5.65–5.50 (1H), 5.12 (s, 2H), 4.90 (s, 1H), 4.28 (s, 1H), 2.30–1.80 (4H), 1.78–1.32 (4H); ¹³C NMR (100 MHz, CDCl₃) 155.5, 136.6, 134.9, 132.1, 128.5, 128.2, 128.1, 66.6, 52.2, 34.2, 28.4, 27.5, 26.6; HRMS (ESI, m/z) [$M + H$]⁺ calculated for C₁₅H₂₀NO₂, 246.1494; found, 246.1486.

Benzyl (2-Methylcyclopent-2-en-1-yl)carbamate (**4f**): 0.083 g, 72% yield; R_f = 0.29 (10:90 = EtOAc/*n*-hexane); colorless solid; mp 50–52 °C; IR (neat) 3325, 2935, 1693, 1519, 1234, 1060, 1026 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) 7.44–7.28 (5H), 5.60–5.00 (3H), 4.85–4.40 (2H), 2.50–2.05 (3H), 1.85–1.50 (4H); ¹³C NMR (100 MHz, CDCl₃) 156.1, 144.9, 139.7, 136.7, 136.6, 128.52, 128.50, 128.1, 128.0, 125.0, 66.5, 59.8, 35.2, 32.3, 29.8, 16.6, 13.7; HRMS (ESI, m/z) [$M + Na$]⁺ calculated for C₁₄H₁₇NNaO₂, 254.1157; found, 254.1165.

Benzyl (1,3-Diphenylprop-2-yn-1-yl)carbamate (**4g**):^{16a} 0.115 g, 68% yield; R_f = 0.34 (10:90 = EtOAc/*n*-hexane); colorless solid; mp 88–90 °C; IR (neat) 3305, 3032, 1689, 1531, 1253, 1041 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) 7.75–7.25 (15H), 5.99 (d, J = 8 Hz, 1H), 5.59 (d, J = 7.2 Hz, 1H), 5.29–5.09 (2H); ¹³C NMR (100 MHz, CDCl₃) 155.4, 139.0, 136.2, 131.8, 128.7, 128.6, 128.5, 128.3, 128.2, 127.0, 122.4, 86.9, 85.1, 67.2, 47.4; HRMS (ESI, m/z) [$M + Na$]⁺ calculated for C₂₃H₁₉NNaO₂, 364.1308; found, 364.1302.

3-(1,3-Diphenylprop-2-yn-1-yl)oxazolidin-2-one (**4h**): 0.115 g, 83% yield; R_f = 0.2 (10:90 = EtOAc/*n*-hexane); light brown liquid; IR (neat) 3491, 3059, 2916, 1955, 1894, 1743, 1489, 1415, 1246, 1068, 1029, 759 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) 7.67–7.60 (2H), 7.58–7.51 (2H), 7.47–7.30 (6H), 4.48–4.37 (1H), 4.35–4.24 (1H), 3.87–3.75 (1H), 3.32–3.20 (1H); ¹³C NMR (100 MHz, CDCl₃) 157.8, 136.0, 131.9, 128.5, 127.4, 122.0, 87.4, 83.4, 62.2, 49.8, 40.4; HRMS (ESI, m/z) [$M + H$]⁺ calculated for C₁₈H₁₆NO₂, 278.1181; found, 278.1186.

N-(1,3-Diphenylprop-2-yn-1-yl)-4-methylbenzenesulfonamide (**4i**):^{12b} 0.138 g, 77% yield; R_f = 0.2 (10:90 = EtOAc/*n*-hexane); colorless solid; mp 182–184 °C; IR (neat) 3267, 1593, 1489, 1427, 1323, 1153, 1049, 813 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) 7.84 (d, J = 8 Hz, 2H), 7.58 (d, J = 8 Hz, 2H), 7.45–7.20 (8H), 7.14 (d, J = 7.6 Hz, 2H), 5.58 (d, J = 9.2 Hz, 1H), 4.96 (d, J = 9.2 Hz, 1H), 2.34 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) 143.5, 137.4, 131.5, 129.5, 128.7, 128.6, 128.5, 128.1, 127.5, 127.3, 121.9, 86.7, 85.4, 49.8, 21.4; HRMS (ESI, m/z) [$M + Na$]⁺ calculated for C₂₂H₁₉NaNO₂S, 384.1034; found, 384.1078.

Benzyl (1-(Naphthalen-2-yl)-3-phenylprop-2-yn-1-yl)carbamate (**4j**):^{12b} 0.145 g, 74% yield; R_f = 0.32 (10:90 = EtOAc/*n*-hexane); colorless solid; mp 160–165 °C; IR (neat) 3309, 2918, 1684, 1653, 1506, 1322, 1226, 1045, 972, 814, 757 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) 8.06 (s, 1H), 7.95–7.82 (3H), 7.67 (d, J = 8.4 Hz, 1H), 7.58–7.47 (4H), 7.45–7.32 (7H), 6.14 (d, J = 8.0 Hz, 1H), 5.45 (d, J = 7.6 Hz, 1H), 5.29–5.11 (2H); ¹³C NMR (100 MHz, CDCl₃) 155.4, 136.1, 133.2, 133.1, 131.8, 128.7, 128.6, 128.5, 128.3, 128.2, 128.1, 127.6, 126.4, 126.3, 125.8, 124.8, 122.4, 86.9, 85.4, 67.2, 47.6; HRMS (ESI, m/z) [$M + H$]⁺ calculated for C₂₇H₂₂NO₂, 392.1651; found, 392.1640.

Benzyl (1-Phenyl-3-(triisopropylsilyl)prop-2-yn-1-yl)carbamate (**4k**): 0.189 g, 91% yield; R_f = 0.31 (10:90 = EtOAc/*n*-hexane); colorless oil; IR (neat) 3318, 2943, 2866, 1715, 1699, 1496, 1456,

1307, 1232, 1041, 917, 883, 750, 696 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) 7.65–7.45 (2H), 7.40–7.27 (8H), 5.75 (d, $J = 8$ Hz, 1H), 5.32–5.00 (3H), 1.30–0.80 (21H); ^{13}C NMR (100 MHz, CDCl_3) 155.3, 139.1, 136.2, 128.6, 128.5, 128.2, 128.1, 128.0, 127.0, 104.9, 86.5, 67.1, 47.6, 18.6, 11.1; HRMS (ESI, m/z) $[\text{M} + \text{Na}]^+$ calculated for $\text{C}_{26}\text{H}_{35}\text{NaNO}_2\text{Si}$, 444.2335; found, 444.2331.

Benzyl (1-Phenylpent-2-yn-1-yl)carbamate (4l): 0.101 g, 62% yield; $R_f = 0.26$ (05:95 = EtOAc/*n*-hexane); colorless oil; IR (neat) 3324, 2958, 2931, 1715, 1496, 1455, 1309, 1231, 1115, 1027, 915, 752, 697 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) 7.65–7.15 (10H), 5.70 (s, 1H), 5.38–5.00 (3H), 2.27 (t, $J = 6.8$ Hz, 2H), 1.70–1.10 (4H), 0.94 (t, $J = 6.8$ Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3) 155.4, 139.7, 136.2, 128.6, 128.5, 128.1, 127.9, 126.9, 85.9, 77.8, 67.0, 47.1, 30.6, 29.7, 21.9, 18.4, 13.5; HRMS (ESI, m/z) $[\text{M} + \text{Na}]^+$ calculated for $\text{C}_{21}\text{H}_{24}\text{NO}_2$, 322.1807; found, 322.1811.

Benzyl Benzhydrylcarbamate (6a):^{16a} 0.157 g, 99% yield; $R_f = 0.29$ (10:90 = EtOAc/*n*-hexane); colorless solid; mp 113–115 $^{\circ}\text{C}$; IR (neat) 3321, 3028, 1693, 1535, 1492, 1238, 1141, 1029 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) 7.50–7.20 (15H), 6.02 (s, 1H), 5.46 (s, 1H), 5.15 (s, 1H); ^{13}C NMR (100 MHz, CDCl_3) 155.6, 141.6, 136.3, 128.7, 128.5, 128.2, 127.5, 127.2, 67.0, 58.9; HRMS (ESI, m/z) $[\text{M} + \text{H}]^+$ calculated for $\text{C}_{21}\text{H}_{20}\text{NO}_2$, 318.1489; found, 318.1488.

N-Benzhydryl-4-methylbenzenesulfonamide (6b):^{18c} 0.103 g, 97% yield; $R_f = 0.22$ (10:90 = EtOAc/*n*-hexane); colorless solid; mp 155–157 $^{\circ}\text{C}$; IR (neat) 3248, 3028, 1597, 1492, 1450, 1315, 1161, 1091, 1056, 937 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) 7.63–7.55 (2H), 7.35–7.05 (12H), 5.60 (d, $J = 7.2$ Hz, 1H), 5.32 (d, $J = 7.2$ Hz, 1H), 2.39 (s, 1H); ^{13}C NMR (100 MHz, CDCl_3) 143.1, 140.5, 137.3, 129.3, 128.5, 127.5, 127.3, 127.2, 61.3, 21.4; HRMS (ESI, m/z) $[\text{M} + \text{Na}]^+$ calculated for $\text{C}_{20}\text{H}_{19}\text{NNaO}_2\text{S}$, 360.1034; found, 360.1049.

1-Benzhydrylindolin-2-one (6c): 0.130 g, 88% yield; $R_f = 0.25$ (10:90 = EtOAc/*n*-hexane); colorless solid; mp 154–156 $^{\circ}\text{C}$; IR (neat) 3059, 2916, 1712, 1612, 1481, 1330, 1234, 1199, 1168 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) 7.50–6.90 (14H), 6.52–6.34 (1H), 3.68 and 3.60 (2H); ^{13}C NMR (100 MHz, CDCl_3) 175.2, 143.8, 137.7, 129.3, 128.6, 128.5, 128.3, 127.7, 127.2, 124.5, 124.3, 122.0, 111.9, 58.1, 35.6; HRMS (ESI, m/z) $[\text{M} + \text{Na}]^+$ calculated for $\text{C}_{21}\text{H}_{18}\text{NO}$, 300.1388; found, 300.1391.

N-(4-Bromophenyl)(phenyl)methyl-4-methylbenzene sulfonamide (6d): 0.185 g, 89% yield; $R_f = 0.25$ (10:90 = EtOAc/*n*-hexane); colorless solid; mp 121–123 $^{\circ}\text{C}$; IR (neat) 3271, 3059, 2920, 1597, 1489, 1450, 1327, 1161, 1087 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) 7.56 (d, $J = 8$ Hz, 2H), 7.33 (d, $J = 8$ Hz, 2H), 7.22 (s, 3H), 7.15 (d, $J = 8$ Hz, 2H), 7.13–6.98 (4H), 5.53 and 5.43 (two d, $J = 7.2$ Hz, 1H), 2.41 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3) 143.4, 140.0, 139.5, 137.1, 131.5, 129.4, 129.1, 128.7, 127.8, 127.2, 127.1, 121.5, 60.8, 21.5; HRMS (ESI, m/z) $[\text{M} + \text{Na}]^+$ calculated for $\text{C}_{20}\text{H}_{18}\text{BrNNaO}_2\text{S}$, 438.0139; found, 438.0109.

N-((4-Methoxyphenyl)(phenyl)methyl)-4-methylbenzenesulfonamide (6e): 0.169 g, 92% yield; $R_f = 0.26$ (10:90 = EtOAc/*n*-hexane); colorless solid; mp 139–141 $^{\circ}\text{C}$; IR (neat) 3275, 2927, 1608, 1512, 1454, 1323, 1253, 1157, 1029 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) 7.65–7.55 (2H), 7.27–7.09 (7H), 7.06–6.97 (2H), 6.80–6.70 (2H), 5.54 (d, $J = 7.2$ Hz, 1H), 5.22 (d, $J = 7.2$ Hz, 1H), 3.77 (s, 3H), 2.39 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3) 158.9, 143.1, 140.7, 137.4, 132.7, 129.3, 128.4, 127.4, 127.3, 127.2, 113.8, 60.8, 55.2, 21.4; HRMS (ESI, m/z) $[\text{M} + \text{Na}]^+$ calculated for $\text{C}_{21}\text{H}_{21}\text{NNaO}_3\text{S}$, 390.1140; found, 390.1175.

4-Methyl-N-(phenyl(thiophen-2-yl)methyl)benzene Sulfonamide (6f): 0.147 g, 86% yield; $R_f = 0.24$ (10:90 = EtOAc/*n*-hexane); colorless solid; mp 121–123 $^{\circ}\text{C}$; IR (neat) 3259, 2920, 1597, 1492, 1438, 1319, 1161, 1053, 933 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) 7.63–7.53 (2H), 7.30–7.05 (8H), 6.93–6.87 (1H), 6.80 (dd, $J = 1.2$, 5.2 Hz, 1H), 5.65 (d, $J = 7.6$ Hz, 1H), 5.37 (d, $J = 7.6$ Hz, 1H), 2.39 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3) 143.2, 141.8, 140.2, 137.4, 129.3, 128.5, 127.6, 127.2, 127.1, 126.6, 126.4, 122.8, 57.5, 21.5; HRMS (ESI, m/z) $[\text{M} + \text{Na}]^+$ calculated for $\text{C}_{18}\text{H}_{17}\text{NNaO}_2\text{S}_2$, 366.0598; found, 366.0587.

N-(9H-Fluoren-9-yl)-4-methylbenzenesulfonamide (6g): 0.152 g, 91% yield; $R_f = 0.24$ (10:90 = EtOAc/*n*-hexane); colorless solid; mp

174–176 $^{\circ}\text{C}$; IR (neat) 3305, 3028, 2920, 1597, 1435, 1327, 1157, 1064, 740 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) 8.20–6.90 (14H), 5.40 (d, $J = 9.6$ Hz, 1H), 4.82 (d, $J = 9.6$ Hz, 1H), 2.52 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3) 143.7, 143.3, 140.0, 138.4, 129.9, 129.8, 128.9, 127.8, 127.3, 125.2, 119.9, 58.3, 21.6; HRMS (ESI, m/z) $[\text{M} + \text{Na}]^+$ calculated for $\text{C}_{20}\text{H}_{17}\text{NNaO}_2\text{S}$, 358.0878; found, 358.0859.

■ ASSOCIATED CONTENT

Supporting Information

Copies of ^1H and ^{13}C NMR spectra for compounds **3a–3k**, **4a–4l**, and **6a–6g**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

■ AUTHOR INFORMATION

Corresponding Author

*E-mail: pghorai@iiserb.ac.in.

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

We would like to thank the Department of Science & Technology (Sanction No. SR/FT/CS-71/2010) and IISER Bhopal for financial support. Our special thanks go to Dr. Deepak Chopra for useful discussions. B.G.D. and R.N. would like to thank CSIR, India, for a fellowship.

■ REFERENCES

- (1) (a) Trost, B. M.; Crawley, M. L. *Chem. Rev.* **2003**, *103*, 2921. (b) Johannsen, M.; Jørgensen, K. A. *Chem. Rev.* **1998**, *98*, 1689.
- (2) Trost, B. M. *Angew. Chem., Int. Ed.* **1995**, *34*, 259.
- (3) For reviews, see: Tsuji, J. *Transition Metal Reagents and Catalysts*; Wiley: New York, 2000.
- (4) Using Pd catalysts: (a) Ghosh, R.; Sarkar, A. *J. Org. Chem.* **2011**, *76*, 8508. (b) Hikawa, H.; Yokoyama, Y. *Org. Lett.* **2011**, *13*, 6512. (c) Hikawa, H.; Yokoyama, Y. *J. Org. Chem.* **2011**, *76*, 8433. (d) Hikawa, H.; Yokoyama, Y. *Org. Biomol. Chem.* **2011**, *9*, 4044. (e) Tao, Y.; Wang, B.; Wang, B.; Qu, L.; Qu, J. *Org. Lett.* **2010**, *12*, 2726. (f) Usui, I.; Schmidt, S.; Keller, M.; Breit, B. *Org. Lett.* **2008**, *10*, 1207. (g) Piechaczyk, O.; Thoumazet, C.; Jean, Y.; Le Floch, P. *J. Am. Chem. Soc.* **2006**, *128*, 14306. (h) Kinoshita, H.; Shinokubo, H.; Oshima, K. *Org. Lett.* **2004**, *6*, 4085. (i) Ozawa, F.; Okamoto, H.; Kawagishi, S.; Yamamoto, S.; Minami, T.; Yoshifuji, M. *J. Am. Chem. Soc.* **2002**, *124*, 10968.
- (5) Using Pt catalysts: (a) Das, K.; Shibuya, R.; Nakahara, Y.; Germain, N.; Ohshima, T.; Mashima, K. *Angew. Chem., Int. Ed.* **2012**, *51*, 150. (b) Ohshima, T.; Miyamoto, Y.; Ipposhi, J.; Nakahara, Y.; Utsunomiya, M.; Mashima, K. *J. Am. Chem. Soc.* **2009**, *131*, 14317.
- (6) References cited therein: Yang, S.-C.; Hsu, Y.-C.; Gan, K.-H. *Tetrahedron* **2006**, *62*, 3949.
- (7) (a) Tao, Y.; Zhou, Y.; Qu, J.; Hidai, M. *Tetrahedron Lett.* **2010**, *51*, 1982. (b) Kimura, M.; Futamata, M.; Shibata, K.; Tamaru, Y. *Chem. Commun.* **2003**, 234.
- (8) Using Ru catalyst: Nishibayashi, Y.; Milton, M. D.; Inada, Y.; Yoshikawa, M.; Wakiji, I.; Hidai, M.; Uemura, S. *Chem.—Eur. J.* **2005**, *11*, 1433.
- (9) Using rare earth metal salts: Noji, M.; Ohno, T.; Fujii, K.; Futaba, N.; Tajima, H.; Ishii, K. *J. Org. Chem.* **2003**, *68*, 9340.
- (10) Using gold catalysts: (a) Ohshima, T.; Nakahara, Y.; Ipposhi, J.; Miyamoto, Y.; Mashima, K. *Chem. Commun.* **2011**, *47*, 8322. (b) Mukherjee, P.; Widenhoefer, R. A. *Org. Lett.* **2010**, *12*, 1184. (c) Lu, Y.; Fu, X.; Chen, H.; Du, X.; Jia, X.; Liu, Y. *Adv. Synth. Catal.* **2009**, *351*, 129. (d) Guo, S.; Song, F.; Liu, Y. *Synlett* **2007**, 964. (e) Terrasson, V.; Marque, S.; Georgy, M.; Campagne, J.-M.; Prima, D. *Adv. Synth. Catal.* **2006**, *348*, 2063.

- (11) Using In catalysts: (a) Vicennati, P.; Cozzi, P. G. *Eur. J. Org. Chem.* **2007**, 2248. (b) Liu, Y.-L.; Liu, L.; Wang, D.; Chen, Y.-J. *Tetrahedron* **2009**, 65, 3473.
- (12) Using Bi catalysts: (a) Zhan, Z.-P.; Yang, W.-Z.; Yang, R.-F.; Yu, J.-L.; Li, J.-P.; Liu, H.-J. *Chem. Commun.* **2006**, 3352. (b) Qin, H.; Yamagiwa, N.; Matsunaga, S.; Shibasaki, M. *Angew. Chem., Int. Ed.* **2007**, 46, 409.
- (13) Using Fe catalysts: (a) Zhan, Z.-P.; Yu, J.-L.; Liu, H.-J.; Cui, Y.-Y.; Yang, R.-F.; Yang, W.-Z.; Li, J.-P. *J. Org. Chem.* **2006**, 71, 8298. (b) Jana, U.; Maiti, S.; Biswas, S. *Tetrahedron Lett.* **2008**, 49, 858. (c) Guérinot, A.; Serra-Muns, A.; Gnam, C.; Bensoussan, C.; Reymond, S.; Cossy, J. *Org. Lett.* **2010**, 12, 1808.
- (14) Using calcium catalysts: Haubenreisser, S.; Niggemann, M. *Adv. Synth. Catal.* **2011**, 353, 469.
- (15) Using Ag catalysts: Sreedhar, B.; Reddy, P. S.; Reddy, M. A.; Neelima, B.; Arundhati, R. *Tetrahedron Lett.* **2007**, 48, 8174.
- (16) Using Mo catalysts: (a) Reddy, C. R.; Madhavi, P. P.; Reddy, A. S. *Tetrahedron Lett.* **2007**, 48, 7169. (b) Yang, H. W.; Fang, L.; Zhang, M.; Zhu, C. J. *Eur. J. Org. Chem.* **2009**, 666. (c) Zhang, M.; Yang, H.; Cheng, Y.; Zhu, Y.; Zhu, C. *Tetrahedron Lett.* **2010**, 51, 1176.
- (17) Using Hg catalysts: Yamamoto, H.; Ho, E.; Namba, K.; Imagawa, H.; Nishizawa, M. *Chem.—Eur. J.* **2010**, 16, 11271.
- (18) Using Bronsted acids: (a) Motokura, K.; Nakagiri, N.; Mori, K.; Mizugaki, T.; Ebitani, K.; Jitsukawa, K.; Kaneda, K. *Org. Lett.* **2006**, 8, 4617. (b) Motokura, K.; Nakagiri, N.; Mizugaki, T.; Ebitani, K.; Kaneda, K. *J. Org. Chem.* **2007**, 72, 6006. (c) Wang, G. W.; Shen, Y. B.; Wu, X. L. *Eur. J. Org. Chem.* **2008**, 4367. (d) Reddy, C. R.; Jithender, E. *Tetrahedron Lett.* **2009**, 50, 5633.
- (19) Amination reaction of alcohols through a redox process; see: (a) Cui, X.; Zhang, Y.; Shi, F.; Deng, Y. *Chem.—Eur. J.* **2011**, 17, 1021. (b) Guillena, G.; Ramon, D. J.; Yus, M. *Chem. Rev.* **2010**, 110, 1611. (c) Dobreiner, G. E.; Crabtree, R. H. *Chem. Rev.* **2010**, 110, 681. (d) Shi, F.; Tse, M. K.; Zhou, S. L.; Pohl, M. M.; Radnik, J.; Hubner, S.; Jahnisch, K.; Bruckner, A.; Beller, M. *J. Am. Chem. Soc.* **2009**, 131, 1775. (e) Cui, X.; Shi, F.; Zhang, Y.; Deng, Y. *Tetrahedron Lett.* **2010**, 51, 2048. (f) Shimizu, K.; Nishimura, M.; Satsuma, A. *ChemCatChem* **2009**, 1, 497.
- (20) For allylic transposition using oxo-rhenium catalysts: (a) Herrmann, A. T.; Saito, T.; Stivala, C. E.; Tom, J.; Zakarian, A. *J. Am. Chem. Soc.* **2010**, 132, 5962. (b) Hansen, E. C.; Lee, D. J. *Am. Chem. Soc.* **2006**, 128, 8142. (c) Morrill, C.; Grubbs, R. H. *J. Am. Chem. Soc.* **2005**, 127, 2842. (d) Sherry, B. D.; Radosevich, A. T.; Toste, F. D. *J. Am. Chem. Soc.* **2003**, 125, 6076. (e) Bellemin-Laponnaz, S.; Giesie, H.; Le Ny, J.-P.; Osborn, J. A. *Angew. Chem., Int. Ed. Engl.* **1997**, 36, 976.
- (21) Amination reaction of propargyl alcohol using oxo-rhenium(V) catalyst: Ohri, R. V.; Radosevich, A. T.; Hrovat, K. J.; Musich, C.; Huang, D.; Holman, T. R.; Toste, F. D. *Org. Lett.* **2005**, 7, 2501.
- (22) Pramanik, S.; Ghorai, P. *Chem. Commun.* **2012**, 48, 1820.
- (23) For recent application of oxo-rhenium catalysts: (a) Volchkov, I.; Park, S.; Lee, D. *Org. Lett.* **2011**, 13, 3530. (b) Yun, S. Y.; Hansen, E. C.; Volchkov, I.; Cho, E. J.; Lo, W. Y.; Lee, D. *Angew. Chem., Int. Ed.* **2010**, 49, 4261. (c) Bellemin-Laponnaz, S. *ChemCatChem* **2009**, 1, 357. (d) Ghorai, P.; Dussault, P. H. *Org. Lett.* **2009**, 11, 213. (e) Tadpetch, K.; Rychnovsky, S. D. *Org. Lett.* **2008**, 10, 4839. (f) Nolin, K. A.; Krumper, J. R.; Pluth, M. D.; Bergman, R. G.; Toste, F. D. *J. Am. Chem. Soc.* **2007**, 129, 14684.
- (24) Aza-phospha-oxa-Cope rearrangements: Lee, E. E.; Batey, R. A. *J. Am. Chem. Soc.* **2005**, 127, 14887.
- (25) Imidozirconium complex mediated transpositions: (a) Lalic, G.; Krinsky, J. L.; Bergman, R. G. *J. Am. Chem. Soc.* **2008**, 130, 4459. (b) Lalic, G.; Blum, S. A.; Bergman, R. G. *J. Am. Chem. Soc.* **2005**, 127, 16790.
- (26) Stefanoni, M.; Luparia, M.; Porta, A.; Zanoni, G.; Vidari, G. *Chem.—Eur. J.* **2009**, 15, 3940.