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Palladium-Catalyzed Ring-Opening Coupling of Cyclobutenols with Aryl Halides

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Abstract A palladium(0)-catalyzed ring-opening cross-coupling reaction between *tert*-cyclobutenols and aryl halides produces γ -arylated β , γ -unsaturated ketones. In the case of aryl halides bearing functional groups at the *ortho* position, the resulting ring-opened ketones undergo intramolecular condensation to afford bicyclic aromatic compounds.

Key words palladium, cyclobutenols, arylation, ring opening, ketones, cross-coupling, condensation, homogeneous catalysis

The synthetic utility of transition-metal-catalyzed C-C bond-cleavage reactions has grown exponentially over the past two decades, enabling unique molecular transformations that are otherwise difficult to achieve.¹ Recently, transition-metal-catalyzed reactions of benzocyclobutenols have attracted considerable interest.² Such reactions encompass the formation of a transition-metal benzocyclobutenolate and subsequent β-carbon elimination involving selective $C(sp^2)-C(sp^3)$ bond cleavage, which leads to an aryl transition-metal species bearing an ortho-acylmethyl substituent. In addition, the palladium-catalyzed crosscoupling reactions of aryl halides and the rhodium- and iridium-catalyzed annulations of alkynes, α , β -unsaturated ketones, and diazoesters have also been developed. However, studies into comparable reactions involving nonbenzofused analogues, i.e., cyclobutenols, are rare. In this context, in 2013, we reported a rhodium(I)-catalyzed alkyne insertion into cyclobutenols to afford tetrasubstituted benzenes.^{3,4} Thus, we herein report the application of a previously developed palladium(0)-catalyzed ring-opening cross-coupling reaction using aryl halides to cyclobutenol substrates.⁵ We expect that this coupling reaction will proceed via ring opening by selective C-C bond-cleavage to furnish γ -arylated β , γ -unsaturated ketones.

To examine the initial cross-coupling reaction between a cyclobutenol and an aryl halide, 1-butyl-3-phenylcyclobut-2-enol (**1a**) and 4-bromotoluene (**2a**, 1.2 equiv to **1a**)



were reacted under the conditions employed for the coupling of benzocyclobutenols, i.e., $Pd_2(dba)_3^6$ (5 mol% Pd), DavePhos⁶ (10 mol%), and Ag_2CO_3 (1.1 equiv) in toluene at 65 °C (Table 1, entry 1). The expected cross-coupling reaction produced the ring-opened arylated ketone **3a** in 32% NMR yield. We expect that this reaction proceeded via the oxidative addition of **2a** to palladium(0) to give arylpalladium(II) bromide **A** (see Table 1), subsequent ligand exchange of **A** with **1a** to yield arylpalladium(II) cyclobutenolate **B**, ring opening by β -carbon elimination with selective C(sp²)–C(sp³) bond cleavage^{2,3,7} to afford alkenylpalladium(II) intermediate **C**, and finally, reductive elimination to yield the γ -arylated β , γ -unsaturated ketone **3a**.

We then focused on optimization of the reaction conditions to improve the yield of **3a**. Interestingly, the use of XPhos⁶ instead of DavePhos worked well in our case, affording 3a in 62% NMR yield (Table 1, entry 2). In addition, an isolated yield of 79% was achieved when 2a was used as the limiting reagent (Table 1, entry 3). In this case, **3a** was obtained as a 96:4 mixture of E and Z isomers in addition to a 9% yield of (E)-2-phenyl-2-octen-4-one $(\mathbf{4})$, which arose from the thermal ring opening of 2a.8 The reaction performed with 4-iodotoluene afforded the desired product 3a in 45% yield. Variation in the palladium/phosphine ratio from 1:2 to 1:1 led to a decrease in yield of 3a (Table 1, entry 4). Although silver(I) oxide was successful as an effective additive, a slightly lower yield was obtained compared to other conditions (Table 1, entry 5). We also confirmed that the use of silver salts was not essential, as the reaction with *t*-BuOK also yielded **3a**; however, a significant quantity of **4** was also formed, thereby demonstrating the superiority of the silver salts in this reaction to yield the desired products **3** (Table 1, entry 6).⁹ Furthermore, in the context of reaction temperature, the reaction performed at room temperature was slow and low yielding (7% yield), and the majority of starting material was recovered unchanged (Table 1, entry 7). In contrast, an increase in the reaction temperature to 110 °C resulted in significant competition from the thermal ring-opening reaction (Table 1, entry 8). With respect to the

reaction solvent, the use of 1,2-dichloroethane lowered the yield of **3a** to 24% (not shown in Table 1), while acetonitrile produced no reaction.

A range of cyclobutenols bearing different substituents were then subjected to the cross-coupling reaction with **2a** (Table 2).^{10,11} The reaction of 1-butylcyclobutenols **1b** and **1c**, bearing *para*-substituted phenyl groups at the 3-position, afforded enones **3b** and **3c** in 61% and 70% yields, respectively (Table 2, entries 1 and 2). In addition, the coupling of 3-hexylcyclobutenol **1d** with **2a** gave **3d** in a moderate yield (42%, Table 2, entry 3), in contrast to the previous rhodium(I)-catalyzed reaction with alkynes, which resulted in exclusive thermal ring opening of the cyclobutenol without formation of the coupling product. 1,3-Diarylcyclobutenols (**1e** and **1f**) and the thiophene-bearing **1g** were also suitable coupling partners (38–60% yields, Table 2, entries 4–6).

The scope of aryl halides for the coupling reaction was also examined (Table 3). The reaction of **1a** and bromobenzene (**2b**) afforded enone **3h** in 50% yield (Table 3, entry 1), and various substituted phenyl halides **2c-h** gave rise to the

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Table 2Ring-Opening Cross-Coupling Reaction of Cyclobutenols 1with 4-Bromotoluene (2a)^a



Entry	1 R ¹ , R ²	Time (h)	3 (%) ^b
1	1b 4-MeC ₆ H ₄ , Bu	11.5	3b 61
2	1c 4-MeOC ₆ H ₄ , Bu	10.5	3c 70
3	1d hexyl, Bu	16	3d 42
4	1e Ph, Ph	4	3e 60
5	1f Ph, 4-MeC ₆ H ₄	6	3f 59
6	1g Ph, 2-thienyl	11.5	3g 38

 a **1, 2a** (**1**/**2a** = 1.2:1), Pd₂(dba)₃ (5 mol% Pd), XPhos (10 mol%), and Ag₂CO₃ (1.1 equiv) were heated at 65 °C in toluene.

^b Isolated yield.

Table 1 Screening of the Reaction Conditions for the Ring-Opening Cross-Coupling of 1a with 2a^a



Entry	1a/2a	Ligand	Additive	Temp (°C)	3a (%) ^b	E/Z	4 (%) ^b
1	1:1.2	DavePhos	Ag ₂ CO ₃	65	32 ^c		
2	1:1.2	XPhos	Ag ₂ CO ₃	65	62°		
3	1.2:1	XPhos	Ag ₂ CO ₃	65	79 ^d	96:4	9
4	1.2:1	XPhos ^e	Ag ₂ CO ₃	65	71 ^c		
5	1.2:1	XPhos	Ag ₂ O	65	61 ^c		
6	1.2:1	XPhos	t-BuOK	65	30		45
7	1.2:1	XPhos	Ag ₂ CO ₃	r.t.	7 ^f		
8	1.2:1	XPhos	Ag ₂ CO ₃	110	44	62:38	39

^a 1a, 2a, $Pd_2(dba)_3$ (5 mol% Pd), ligand (10 mol%), and additive (1.1 equiv) were heated in toluene for 4–8 h.

^b Isolated yield unless otherwise noted.

^c Yield determined by ¹H NMR spectroscopy.

^d **3a** was isolated in 45% yield when the corresponding iodide was employed.

^e 5 mol% XPhos was employed.

^f 79% **1a** was recovered unreacted.

corresponding enones **3i–n** in 30–66% yields (Table 3, entries 2–7). In addition, 1-bromonaphthalene (**2i**) took part in the coupling reaction with **1a** (66% yield, Table 3, entry 8). Allyl bromide **2j** was also suitable as a coupling partner for the reaction with **1a**, providing dienone **3p** in 47% yield (Table 3, entry 9).

Ph—≺ 1a (1	OH Bu + R-Br .2 equiv) 2	Pd ₂ (dba) ₃ (2.5 mol%) XPhos (10 mol%) Ag ₂ CO ₃ (1.1 equiv) toluene, 65 °C	R Ph Bu 3
Entry	2 R	Time (h)	3 (%) ^b
1	2b Ph	14.5	3h 50
2	2c 4-MeOC ₆ H ₄ ^c	3	3i 59
3	2d 4-Me ₂ NC ₆ H ₄	6	3j 39
4	2e 4-AcC ₆ H ₄	6	3k 50
5	2f 3-MeC ₆ H ₄	19	3I 66
6	2g 2-MeC ₆ H ₄	28	3m 30
7	2h 3,5-Me ₂ C ₆ H	7	3n 49
8	2i 1-naphthyl	11.5	3o 66
9	2j allyl ^d	9	3p 47

^a Unless otherwise noted, **1a**, **2** (**1a**/**2** = 1.2:1), $Pd_2(dba)_3$ (5 mol% Pd), XPhos (10 mol%), and Ag_2CO_3 (1.1 equiv) were heated at 65 °C in toluene.

^b Isolated yield. ^c The corresponding iodide was employed.

^d **1 a/2i** = 1.1 2

^d **1a/2j** = 1:1.2.

Orellana previously reported that the reaction of benzocyclobutenols with aryl halides bearing electrophilic functional groups at the *ortho* position afforded phenanthrene products via condensation following the initial coupling reaction.^{2a} Thus, we also examined the coupling/condensation sequence using cyclobutenols (Table 4).^{12,13} Following the reaction of cyclobutenols **1a**, **1d**, and **1e** with 2formyliodobenzene (**2k**) under the optimized conditions, the reaction mixture was treated with *t*-BuONa (2 equiv). An intramolecular aldol condensation then occurred to afford 2-acylnaphthalenes (**5a–c**) in 45–50% yields (Table 4, entries 1–3). Similarly, 2-cyanoiodobenzene (**2l**) reacted with the cyclobutenols to deliver 2-acyl-1-aminonaphthalenes **5d–f** via an analogous aromatization process (entries 4–6).¹⁴

In the reaction of 1,3-diphenylcyclobutenol (**1e**) with 2iodoaniline (**2m**), the resulting enone spontaneously underwent intramolecular condensation to the expected seven-membered cyclic imine **6**, which was obtained in 43% yield. In contrast, surprisingly no trace of the corresponding

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Entry	1 R ¹ , R ²	2 R ³	5 R ⁴	Yield (%) ^b
1	1a Ph, Bu	2k CHO	5a H	46
2	1d hexyl, Bu	2k	5b H	45
3	1e Ph, Ph	2k	5c H	50
4	1a	2I CN	$5d \text{ NH}_2$	59
5	1d	21	5e NH ₂	33
6	1e	21	5f NH ₂	52

^a **1, 2** (**1**/**2** = 1.2:1), $Pd_2(dba)_3$ (5 mol% Pd), XPhos (10 mol%), and Ag_2CO_3 (1.1 equiv) were heated at 65 °C in toluene for 13–21 h, then *t*-BuOK (2 equiv) was added and the mixture heated further for 2–4 h. ^b Isolated yield.

seven-membered imine was obtained in the reaction of 1butylcyclobutenol **1a** with **2m**; instead, intramolecular oxidative C=N bond formation took place to yield the eightmembered cyclic imine **7** (Scheme 1).¹⁵



Scheme 1 Reaction of cyclobutenols 1a and 1e with 2-aminoiodobenzene (2m)

In conclusion, we have described a method for the palladium-catalyzed arylation of cyclobutenols using aryl halides that involves ring opening of the cyclobutenols to yield a range of γ -arylated β , γ -unsaturated ketones. Furthermore, we also demonstrated that the products obtained from the reaction with aryl halides bearing functional groups at the *ortho* position could be transformed into bicyclic aromatic compounds through an intramolecular condensation reaction.

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Supporting Information

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References and Notes

- (1) For recent reviews, see: (a) Nairoukh, Z.; Cormier, M.; Marek, I. Nat. Rev. Chem. 2017, 1, 0035. (b) Fumagalli, G.; Stanton, S.; Bower, J. F. Chem. Rev. 2017, 117, 9404. (c) Murakami, M.; Ishida, N. J. Am. Chem. Soc. 2016, 138, 13759. (d) Shaw, M. H.; Bower, J. F. Chem. Commun. 2016, 52, 10817. (e) Kondo, T. Eur. J. Org. Chem. 2016, 1232. (f) Souillart, L.; Cramer, N. Chem. Rev. 2015, 115, 9410. (g) Marek, I.; Masarwa, A.; Delaye, P.-O.; Leibeling, M. Angew. Chem. Int. Ed. 2015, 54, 414. (h) Liu, H.; Feng, M.; Jiang, X. Chem. Asian J. 2014, 9, 3360. (i) Chen, F.; Wang, T.; Jiao, N. Chem. Rev. 2014, 114, 8613. (j) Dermenci, A.; Coe, J. W.; Coe, J. W.; Dong, G. Org. Chem. Front. 2014, 1, 567. (k) Mack, D. J.; Njardarson, J. T. ACS Catal. 2013, 3, 272. (l) Dong, G. Synlett 2013, 24, 1. (m) Korotvicka, A.; Nečas, D.; Kotora, M. Curr. Org. Chem. 2012, 16, 1170. (n) Ruhland, K. Eur. J. Org. Chem. 2012, 2683. (o) Aïssa, C. Synthesis 2011, 3389. (p) Seiser, T.; Saget, T.; Tran, D. N.; Cramer, N. Angew. Chem. Int. Ed. 2011, 50, 7740. (q) Cramer, N.; Seiser, T. Synlett 2011, 449. (r) Murakami, M.; Matsuda, T. Chem. Commun. 2011, 47, 1100.
- (2) (a) Chtchemelinine, A.; Rosa, D.; Orellana, A. J. Org. Chem. 2011, 76, 9157. (b) Ishida, N.; Sawano, S.; Masuda, Y.; Murakami, M. J. Am. Chem. Soc. 2012, 134, 17502. (c) Ding, L.; Ishida, N.; Murakami, M.; Morokuma, K. J. Am. Chem. Soc. 2014, 136, 169. (d) Xia, Y.; Liu, Z.; Liu, Z.; Ge, R.; Ye, F.; Hossain, M.; Zhang, Y.; Wang, J. J. Am. Chem. Soc. 2014, 136, 3013. (e) Ishida, N.; Ishikawa, N.; Sawano, S.; Masuda, Y.; Murakami, M. Chem. Commun. 2015, 51, 1882. (f) Wang, Y.; Wang, Y.; Zhang, W.; Zhu, Y.; Wei, D.; Tang, M. Org. Biomol. Chem. 2015, 13, 6587. (g) Yu, J.; Yan, H.; Zhu, C. Angew. Chem. Int. Ed. 2015, 55, 1143. (h) Zhao, C.; Liu, L.-C.; Wang, J.; Jiang, C.; Zhang, Q.-W.; He, W. Org. Lett. 2016, 18, 328.
- (3) Matsuda, T.; Miura, N. Org. Biomol. Chem. 2013, 11, 3424.
- (4) For our recent studies on the rhodium(I)-catalyzed reactions of cyclobutanes and cyclobutenes, see: (a) Matsuda, T.; Yuihara, I. *Chem. Commun.* 2015, *51*, 7393. (b) Matsuda, T.; Matsumoto, T. *Org. Biomol. Chem.* 2016, *14*, 5023. (c) Matsuda, T.; Yuihara, I.; Kondo, K. *Org. Biomol. Chem.* 2016, *14*, 7024.
- (5) For palladium(0)-catalyzed ring-opening arylation reactions of cyclobutanols, see: (a) Nishimura, T.; Uemura, S. J. Am. Chem. Soc. 1999, 121, 11010. (b) Nishimura, T.; Matsumura, S.; Maeda, Y.; Uemura, S. Chem. Commun. 2001, 50. (c) Matsumura, S.; Maeda, Y.; Nishimura, T.; Uemura, S. J. Am. Chem. Soc. 2003, 125, 8862. (d) Ethirajan, M.; Oh, H.-S.; Cha, J. K. Org. Lett. 2007, 9, 2693. (e) Ziadi, A.; Martin, R. Org. Lett. 2012, 14, 1266.
- (6) Abbreviations: dba = dibenzylideneacetone; DavePhos = 2-dicyclohexylphosphino-2'-(*N*,*N*-dimethylamino)biphenyl; XPhos = 2-dicyclohexylphosphino-2',4',6'-triisopropylbiphenyl.

(7) The site selectivity might be attributed to stabilization by an interaction between palladium(II) and the C=C bond. See:
(a) Zhao, P.; Incarvito, C. D.; Hartwig, J. F. *J. Am. Chem. Soc.* 2006, 128, 3124. Pd–C(sp²) bond is stronger than Pd–C(sp³) bond by 8.7 kcal/mol, see: (b) Siegbahn, P. E. M. *J. Phys. Chem.* 1995, 99, 12723.

Cluster

- (8) Murakami, M.; Miyamoto, Y.; Ito, Y. J. Am. Chem. Soc. 2001, 123, 6441.
- (9) Silver(I) ion abstracts a halide ion to facilitate the generation of a cationic palladium(II) species, see: (a) Hirabayashi, K.; Mori, A.; Kawashima, J.; Suguro, M.; Nishihara, Y.; Hiyama, T. J. Org. Chem. 2000, 65, 5342. (b) Ladd, C. L.; Roman, D. S.; Charette, A. B. Org. Lett. 2013, 15, 1350.
- (10) General Procedure for Palladium(0)-Catalyzed Ring-Opening Cross-Coupling of *tert*-Cyclobutenols 1 with Organohalides 2 (GP-1)

A mixture of $Pd_2(dba)_3$ (2.3 mg, 2.5 µmol), XPhos (4.7 mg, 9.9 µmol), cyclobutenol **1** (0.120 mmol), organohalide **2** (0.100 mmol), and Ag_2CO_3 (30.3 mg, 0.11 mmol) in toluene (0.50 mL) was heated at 65 °C with stirring. The reaction mixture was filtered through a plug of Florisil[®] washing with hexane/AcOEt (3:1), and the filtrate was concentrated. The residue was purified by preparative TLC on silica gel to afford ring-opened enone **3**.

(11) Selected Data for Products (E)-1-(4-Methylphenyl)-2-phenyl-1-octen-4-one (3a)

The general procedure (GP-1) was followed using **1a** (24.3 mg, 0.120 mmol) and **2a** (17.1 mg, 0.100 mmol) for 6 h. Purification by preparative TLC on silica gel (hexane/AcOEt/toluene = 60:2:1, 4×) yielded **3a** (22.1 mg, 0.076 mmol, 76%) as a yellow oil. The *Z* isomer (1.0 mg, 0.003 mmol, 3%) and (*E*)-2-phenyl-2-octen-4-one (**4**, 2.1 mg, 0.010 mmol, 9%) were also isolated. ¹H NMR (500 MHz, CDCl₃): δ = 0.83 (t, *J* = 7.3 Hz, 3 H), 1.21 (sext, *J* = 7.5 Hz, 2 H), 1.44–1.51 (m, 2 H), 2.38 (t, *J* = 7.5 Hz, 2 H), 2.37 (s, 3 H), 3.78 (s, 2 H), 7.01 (s, 1 H), 7.15–7.22 (m, 4 H), 7.26–7.30 (m, 1 H), 7.33–7.38 (m, 2 H), 7.42–7.46 (m, 2 H). ¹³C NMR (126 MHz, CDCl₃): δ = 13.8, 21.2, 22.2, 25.8, 42.2, 45.6, 126.2, 127.4, 128.45, 128.53, 129.1, 131.3, 134.6, 134.8, 137.0, 142.3, 209.1. IR: v = 2954, 2931, 2870, 1712, 756, 702 cm⁻¹. ESI-HRMS: *m/z* calcd for C₂₁H₂₄NaO [M + Na]⁺: 315.1719; found: 315.1724.

(*E*)-4-(4-Methylphenyl)-1,3-diphenyl-3-buten-1-one (3e)

The general procedure (GP-1) was followed using **1e** (26.7 mg, 0.120 mmol) and **2a** (17.0 mg, 0.099 mmol) for 4 h. Purification by preparative TLC on silica gel (hexane/AcOEt = 60:1, 3×) yielded **3e** (18.6 mg, 0.060 mmol, 60%) as a colorless oil. ¹H NMR (500 MHz, CDCl₃): δ = 2.34 (s, 3 H), 4.40 (s, 2 H), 7.11–7.15 (m, 3 H), 7.17–7.21 (m, 2 H), 7.23–7.28 (m, 1 H), 7.30–7.35 (m, 2 H), 7.43–7.49 (m, 4 H), 7.55–7.59 (m, 1 H), 7.94–7.98 (m, 2 H). ¹³C NMR (126 MHz, CDCl₃): δ = 21.2, 41.5, 126.2, 127.3, 128.2, 128.38, 128.43, 128.6, 129.1, 131.3, 133.2, 134.7, 134.8, 136.7, 136.9, 142.2, 197.9. IR: v = 1682, 1327, 1211, 756, 694 cm⁻¹. ESI-HRMS: *m/z* calcd for C₂₃H₂₀NaO [M + Na]*: 335.1406; found: 335.1410.

(E)-7-Phenyl-7,10-undecadien-5-one (3p)

The general procedure (GP-1) was followed using **1a** (20.2 mg, 0.100 mmol) and **2j** (14.6 mg, 0.121 mmol) for 9 h. Purification by preparative TLC on silica gel (hexane/AcOEt = 30:1, 3×) yielded **3p** (11.4 mg, 0.047 mmol, 47%) as a yellow oil. ¹H NMR (500 MHz, CDCl₃): δ = 0.82 (t, *J* = 7.5 Hz, 3 H), 1.20 (sext, *J* = 7.4 Hz, 2 H), 1.47 (quint, *J* = 7.5 Hz, 2 H), 2.35 (t, *J* = 7.5 Hz, 2 H), 2.88–2.95 (m, 2 H), 3.56 (s, 2 H), 5.02 (dd, *J* = 10.5, 1.0 Hz, 1 H), 5.09 (dd, *J* = 17.3, 1.2 Hz, 1 H), 5.85 (ddt, *J* = 17.3, 10.2, 6.1 Hz, 1 H), 5.96 (t, *J* = 7.5 Hz, 1 H), 7.18–7.32 (m, 5 H). ¹³C NMR (126 MHz, CDCl₃): δ = 13.8, 22.2, 25.8, 33.2, 41.7, 45.0, 115.5, 126.0,

127.1, 128.4, 129.2, 134.2, 135.8, 142.2, 208.4. IR: v = 2954, 1720, 756 cm⁻¹. ESI-HRMS: *m/z* calcd for $C_{17}H_{22}NaO$ [M + Na]*: 265.1563; found: 265.1559.

(12) General Procedure for Palladium(0)-Catalyzed Ring-Opening Cross-Coupling of tert-Cyclobutenols 1 with ortho-Functionalized Iodobenzenes 2 (GP-2)

A mixture of $Pd_2(dba)_3$ (2.3 mg, 2.5 µmol), XPhos (4.7 mg, 9.9 µmol), cyclobutenol **1** (0.120 mmol), aryl iodide **2** (0.100 mmol), and Ag_2CO_3 (30.3 mg, 0.11 mmol), in toluene (0.50 mL) was heated at 65 °C with stirring. Then, *t*-BuOK (2 equiv) was added to the reaction mixture, and the mixture was further heated for 2–4 h. The reaction mixture was filtered through a plug of Florisil[®] washing with hexane/AcOEt (3:1), and the filtrate was concentrated. The residue was purified by preparative TLC on silica gel to afford naphthalene **4**.

(13) Selected Data for Products

2-Benzoyl-3-phenylnaphthalene (5c)

The general procedure (GP-2) was followed using **1e** (26.7 mg, 0.120 mmol) and **2k** (23.2 mg, 0.100 mmol) for 21 h. Purification by preparative TLC on silica gel (hexane/AcOEt = 20:1, 2×) yielded **5c** (15.3 mg, 0.050 mmol, 50%) as a colorless oil. ¹H NMR (500 MHz, CDCl₃): δ = 7.17–7.38 (m, 7 H), 7.42–7.47 (m, 1 H), 7.54–7.64 (m, 2 H), 7.70–7.73 (m, 2 H), 7.91–7.96 (m, 3 H), 8.04 (s, 1 H). ¹³C NMR (126 MHz, CDCl₃): δ = 126.8, 127.2, 127.8,

127.9, 128.1, 128.3, 128.4, 129.1, 129.2, 129.3, 130.0, 131.5, 132.8, 134.0, 137.4, 137.7, 138.5, 140.3, 198.3. IR: v = 1666, 1281, 756, 694 cm⁻¹. ESI-HRMS: *m/z* calcd for $C_{23}H_{16}NaO$ [M + Na]*: 331.1093; found: 331.1096.

1-Amino-2-benzoyl-3-phenylnaphthalene (5f)

The general procedure (GP-2) was followed using **1e** (26.7 mg, 0.120 mmol) and **2l** (22.0 mg, 0.096 mmol) for 14 h. Purification by preparative TLC on silica gel (hexane/AcOEt = 30:1, 2×) yielded **5f** (16.3 mg, 0.050 mmol, 52%) as a yellow oil. ¹H NMR (500 MHz, CDCl₃): δ = 5.66 (br s, 2 H), 7.02–7.14 (m, 5 H), 7.21–7.30 (m, 4 H), 7.39–7.43 (m, 2 H), 7.53–7.58 (m, 1 H), 7.59–7.64 (m, 1 H), 7.87 (d, *J* = 8.0 Hz, 1 H), 7.97 (d, *J* = 8.5 Hz, 1 H). ¹³C NMR (126 MHz, CDCl₃): δ = 116.5, 118.6, 121.6, 122.6, 125.6, 126.8, 127.6, 128.0, 128.1, 128.8, 129.1, 129.2, 131.7, 134.9, 139.9, 140.3, 141.9, 144.2, 200.4. IR: v = 1604, 1273, 756, 694 cm⁻¹. ESI-HRMS: *m/z* calcd for C₂₃H₁₈NO [M + H]⁺: 324.1383; found: 324.1380.

- (14) Without the addition of a base, the reaction with aldehyde 2k afforded naphthalene 5 in lower yields. In contrast, no condensation was observed in the reaction with nitrile 2l in the absence of base.
- (15) Further studies have been carried out to elucidate the details of this novel oxidative imination.