

Synthesis and Quantum Chemical Studies of New 4-aminoquinazoline Derivatives as Aurora A/B Kinase Inhibitors

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Nine novel 4-aminoquinazoline derivatives were designed and synthesized. Biochemical and cellular analyses demonstrated that most of the derivatives exhibited a strong activity to inhibit Aurora A and B kinases and to suppress the proliferation of a panel of human tumor cell lines (U937, K562, A549, LoVo, and HT29). Quantum chemical studies were also carried out to determine the structural features of these compounds engaged in the inhibition of Aurora kinases.

Key words: 4-aminoquinazolines, anti-proliferative activity, Aurora A and B, quantum chemistry

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Mammalian Aurora kinases constitute a small family of three closely related Ser/Thr kinases, named as Aurora A, B, and C. These kinases contain a highly conserved catalytic site and are involved in the progression of mitosis or meiosis, although the function of Aurora C is less understood (1,2). Aurora A localizes to the spindle poles and

has a crucial role in the bipolar spindle formation (2). Aurora B is a chromosomal passenger protein and localizes at the centromeres during prometaphase and associates with midzone microtubules and midbodies during anaphase and telophase (1,3). Aurora C is specifically expressed in the testis and may play a role in spermatogenesis (4). The initial link between Aurora kinases and cancer was based on the observation that Aurora A and B were frequently overexpressed in colon cancer (5). Further studies found that Aurora A and B were overexpressed in a variety of solid tumors including prostate (6,7), pancreatic (8), breast (9), and thyroid cancers (10). Overexpression of Aurora A or B was mainly attributed to gene amplification and correlated with histological malignancy and poor prognosis (11). Recently, several small molecules with selective anti-Aurora activities have been developed and entered into preclinical or clinical trials for cancer treatment (12–14).

ZM447439 (AstraZeneca, Wilmington, DE), which has a 4-aminoquinazoline skeleton, was the first reported small inhibitor of Aurora kinases. *In vitro*, ZM447439 inhibits both Aurora A and B with IC₅₀ values of 110 and 130 nM, respectively; impairs certain mitosis events, including chromosome alignment, chromosome segregation, and cytokinesis; and induces ultimately cell deaths (15,16). The report herein described our effort to improve the efficacy of inhibiting Aurora kinases by introducing an active imatinib group into several 4-aminoquinazoline analogs. This was achieved by intramolecular cyclization of the amide group on the benzene ring. We found that several new compounds generated by this approach have acquired improved potency in inhibiting both Aurora A and B kinases (Figure 1).

Methods and Materials

Chemistry

The strategy for synthesis of 4-aminoquinazoline derivatives is outlined in Scheme 1 (17). Interaction of cyclocondensation of 2-amino-4-fluorobenzoic acid **1** with formamidine acetate in the solvent of 2-methoxyethanol produced a high yield of compound **2**. Compound **2** was then treated with propylene glycol in the presence of sodium hydride to form compound **3**, which was further used to interact with thionyl chloride to generate chlorinated compound **4**. Using

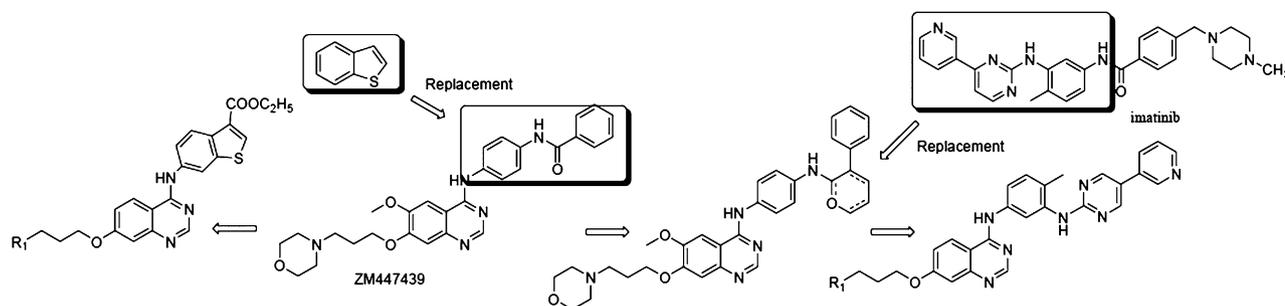
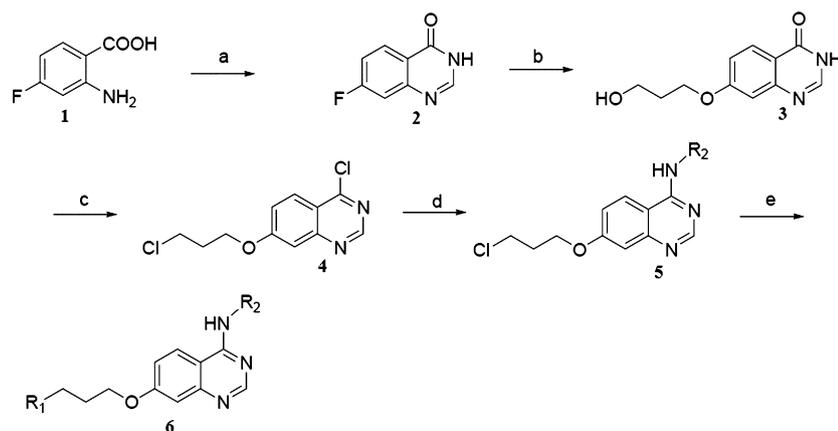


Figure 1: Design of the compounds.



Scheme 1: Synthesis route of the Target Compounds. Reagents and conditions: a) Formamidine Acetate, 2-methoxyethanol; b) propylene glycol and NaH; c) SOCl₂/DMF; d) aniline derivatives, isopropanol; e) secondary amine/DMF.

isopropanol as solvent, compound **4** reacted with aniline derivatives to yield 4-aminoquinazolinone **5**, which furnished the final compound **6** upon treatment with the secondary amine in *N,N*-dimethylformamide.

Cell proliferation assay

The effect of the synthesized compounds on cell proliferation was assessed by MTT (Sigma-Aldrich Shanghai Trading Co Ltd Shanghai, China). The cell lines that were assessed include the following: U937 human acute monocytic leukemia cells, K562 human chronic myeloid leukemia cells, A540 human non-small cell lung cancer cells, and LoVo and HT29 human colon cancer cells. 5×10^3 cells were seeded in each well in a 96-well culture plate. The cell cultures were added with compounds to be tested at various concentrations in triplicates. Seventy-two hours later, the cell growth medium was removed, and each well was added with 5 mg/mL of MTT followed by incubation at 37 °C. After 4-h incubation, 50 μ L of 20% SDS was added to each well, which was further incubated for overnight. Absorbance was measured with a plate reader at 570 nm with correction at 630 nm. The results were expressed as the percentage of absorbance of treated wells versus that of vehicle control. IC₅₀, the drug concentration causing 50% growth inhibition, was calculated via sigmoid curve fitting using GRAPHPAD PRISM 5.0 (GraphPad Software, Inc., La Jolla, CA, USA).

Aurora kinase assays

The kinase activity of Aurora A and B was measured using HTRF-KinEASE-STK discovery kit (Cisbio, Bedford, MA, USA). Briefly, Aurora A (65 nM) or Aurora B (316 nM) (Millipore, Billerica, MA, USA) was incubated with 1 μ M STK substrate 2-biotin (Cisbio) and 15 μ M ATP (Sigma-Aldrich) in the presence of tested compounds for 1 h at room temperature. The reaction was terminated by adding 200 mM EDTA. Phosphorylated substrates were incubated with STK antibody labeled with Eu³⁺-cryptate and streptavidin-XL665. The Eu³⁺-cryptate-conjugated antibody was excited at 340 nm, and cryptate and fluorescence resonance energy transfer emissions were detected at 615 and 665 nm, respectively. GRAPHPAD PRISM was used to generate IC₅₀ values.

Quantum chemical calculation

The chemical structures of new compounds were drawn using HYPERCHEM 7.0 and optimized with its module of molecular mechanics (MM+ force field). The calculated structures at low energy were subjected to conformational analysis using HyperChem's Conformational Search module, and only those that have the most stable conformation were used in the next level of quantum chemical calculations. These calculations were performed using density functional theory (DFT) with the aid of GAUSSIAN03 software (Gaussian, Inc., Wallingford, CT, USA). B3LYP functional

and the 6-31G (d) basis set were used in the DFT calculations^b (18–21). The fundamental vibrations were also calculated using the same method to determine whether they were true minima. The highest occupied molecular orbital (HOMO) and lowest unoccupied molecular orbital (LUMO) values, which were calculated as molecular orbital coefficients in artificial unit (a.u.), were converted into energy units in eV. The plots of HOMO and LUMO were generated using GAUSSVIEW 3.0 (Gaussian, Inc.).

Experimental Protocols

All reagents were purchased from commercial sources and used without further purification. Melting points were measured in open capillaries without correction. IR spectra were determined as KBr pellets on a Thermo Nicolet 6700 spectrophotometer. ¹H-NMR and ¹³C-NMR spectra were recorded in DMSO-*d*₆ on a Bruker Avance 300 or 500 spectrometer; chemical shifts (δ) were reported in parts per million (ppm) relative to tetramethylsilane and used as an internal standard. Mass spectra (MS) were obtained from Agilent 1100LC/MS Spectrometry Services. Elementary analyses were performed on Elementar Vario EL III instrument. The purity of all compounds was routinely verified by TLC with silica gel GF-254 glass plates and viewed under UV light at 254 nm.

Compounds **6a–6i** were characterized as follows:

4-methyl-N¹-(7-(3-morpholinopropoxy)quinazolin-4-yl)-N³-(5-(pyridin-3-yl)pyrimidin-2-yl)benzene-1,3-diamine(6a)

Yield 52.3%, mp 143 °C. IR (KBr, per cm): 3243.73, 2958.31, 2858.03, 1618.01, 1577.51, 1529.30, 1452.16, 1419.37, 1332.59, 1299.81, 1224.60, 1118.53, 1025.96, 794.54, 649.90; ¹H-NMR (DMSO-*d*₆, 300 MHz) δ (ppm): 1.90–1.97 (m, 2H, -CH₂-CH₂-CH₂-O-), 2.24 (s, 3H, -CH₃), 2.40 (m, 4H, 2 × morpholine-CH₂), 2.50 (m, 4H, 2 × morpholine-CH₂), 3.59 (s, 2H, -CH₂-CH₂-CH₂-O-), 4.20 (t, *J* = 6.06 Hz, 2H, -CH₂-CH₂-CH₂-O-), 7.14 (s, 1H, Ar-H), 7.25 (t, *J* = 8.31 Hz, 2H, Ar-H), 7.41–7.49 (m, 2H, Ar-H), 7.59 (d, *J* = 8.31 Hz, 1H, Ar-H), 8.13 (s, 1H, Ar-H), 8.43 (s, 2H, Ar-H), 8.52 (t, *J* = 2.91 Hz, 2H, Ar-H), 8.67 (d, *J* = 3.51 Hz, 2H, Ar-H), 8.98 (s, 1H, Ar-H), 9.28 (s, 1H, -NH-), 9.66 (s, 1H, -NH-); ¹³C-NMR (DMSO-*d*₆, 75 MHz) δ (ppm): 17.61, 25.52, 53.20, 54.64, 66.01, 66.16, 107.41, 107.51, 109.30, 117.50, 118.71, 119.07, 123.65, 124.68, 127.54, 129.81, 132.12, 134.19, 137.28, 137.63, 148.11, 151.30, 151.94, 155.01, 157.37, 159.39, 161.16, 161.51, 161.88; ESI-MS *m/z*: 549.33 [M + H]⁺.

4-methyl-N¹-(7-(3-(piperidin-1-yl)propoxy)quinazolin-4-yl)-N³-(5-(pyridin-3-yl)pyrimidin-2-yl)benzene-1,3-diamine(6b)

Yield 52.1%, mp 170 °C. IR (KBr, per cm): 3423.08, 2944.82, 1618.01, 1577.51, 1523.51, 1452.16, 1413.59,

1334.52, 1297.88, 1224.60, 1132.03, 1002.82, 804.18, 651.83; ¹H-NMR (DMSO-*d*₆, 300 MHz) δ (ppm): 1.50 (m, 2H, -CH₂-CH₂-CH₂-O-), 1.71 (s, 4H, 2 × piperidine-CH₂), 2.24 (s, 5H, -CH₃ and piperidine-CH₂), 2.95 (s, 4H, 2 × piperidine-CH₂), 3.33 (s, 2H, -CH₂-CH₂-CH₂-O-), 4.21 (s, 2H, -CH₂-CH₂-CH₂-O-), 7.22 (d, 3H, Ar-H), 7.42–7.56 (d, 3H, Ar-H), 8.13 (s, 1H, Ar-H), 8.43–8.51 (d, 4H, Ar-H), 8.66 (s, 1H, Ar-H), 8.98 (s, 1H, Ar-H), 9.28 (s, 1H, -NH-), 9.70 (s, 1H, -NH-); ¹³C-NMR (DMSO-*d*₆, 75 MHz) δ (ppm): 17.61, 22.16, 23.18, 23.91, 52.52, 53.69, 65.70, 107.42, 107.67, 109.43, 117.35, 118.74, 119.10, 123.66, 124.78, 127.58, 129.81, 132.12, 134.20, 137.26, 137.62, 148.11, 151.31, 151.91, 155.05, 157.39, 159.40, 161.16, 161.51, 161.62; ESI-MS *m/z*: 547.3 [M + H]⁺.

N¹-(7-(3-(diethylamino)propoxy)quinazolin-4-yl)-4-methyl-N³-(5-(pyridin-3-yl)pyrimidin-2-yl)benzene-1,3-diamine(6c)

Yield 46.8%, mp 157 °C. IR (KBr, per cm): 3384.51, 2979.53, 2651.68, 1618.01, 1597.44, 1525.44, 1452.16, 1415.52, 1332.59, 1295.95, 1224.60, 1130.10, 1004.75, 987.39, 800.33, 707.76, 653.76; ¹H-NMR (DMSO-*d*₆, 300 MHz) δ (ppm): 1.89 (s, 2H, -CH₂-CH₂-CH₂-O-), 1.97–2.04 (d, 5H, diethylamine-CH₂-CH₃), 2.24 (s, 3H, -CH₃), 2.78–2.87 (t, 5H, diethylamine-CH₂-CH₃), 3.98–4.05 (m, 2H, -CH₂-CH₂-CH₂-O-), 4.21 (s, 2H, -CH₂-CH₂-CH₂-O-), 7.15–7.24 (t, *J* = 7.05 Hz, 3H, Ar-H), 7.40–7.46 (t, *J* = 4.95 Hz, 2H, Ar-H), 7.59 (d, *J* = 7.74 Hz, 1H, Ar-H), 8.14 (s, 1H, Ar-H), 8.43 (s, 2H, Ar-H), 8.49–8.57 (dd, *J*₁ = 4.80 Hz, *J*₂ = 9.18 Hz, 2H, Ar-H), 8.66 (s, 1H, Ar-H), 8.99 (s, 1H, Ar-H), 9.28 (s, 1H, -NH-), 9.71 (s, 1H, -NH-); ¹³C-NMR (DMSO-*d*₆, 75 MHz) δ (ppm): 7.78, 9.97, 14.04, 17.64, 20.71, 24.51, 46.26, 48.04, 58.56, 59.70, 65.71, 107.43, 107.58, 109.43, 117.53, 118.78, 119.14, 123.68, 124.84, 127.59, 129.82, 132.15, 134.22, 137.21, 137.64, 148.14, 151.33, 151.96, 155.07, 157.42, 159.42, 161.18, 161.53, 161.73; ESI-MS *m/z*: 535.34 [M + H]⁺.

4-methyl-N³-(5-(pyridin-3-yl)pyrimidin-2-yl)-N¹-(7-(3-(pyrrolidin-1-yl)propoxy)quinazolin-4-yl)benzene-1,3-diamine(6d)

Yield 31.5%, mp 125 °C. IR (KBr, per cm): 3423.08, 2960.24, 2805.96, 1618.01, 1577.51, 1529.30, 1454.09, 1419.37, 1334.52, 1297.88, 1224.60, 1130.10, 798.40, 705.83, 651.83; ¹H-NMR (DMSO-*d*₆, 500 MHz) δ (ppm): 1.68 (s, 4H, 2 × pyrrolidine-CH₂), 1.90–1.98 (m, 2H, -CH₂-CH₂-CH₂-O-), 2.25 (s, 3H, -CH₃), 2.46–2.50 (m, 4H, 2 × pyrrolidine-CH₂), 2.59 (t, *J* = 7.15 Hz, 2H, -CH₂-CH₂-CH₂-O-), 4.17 (t, *J* = 6.35 Hz, 2H, -CH₂-CH₂-CH₂-O-), 7.15 (d, *J* = 2.45 Hz, 1H, Ar-H), 7.20–7.24 (m, 2H, Ar-H), 7.41 (d, *J* = 5.15 Hz, 1H, Ar-H), 7.45–7.47 (dd, *J*₁ = 4.80 Hz, *J*₂ = 4.75 Hz, 1H, Ar-H), 7.55–7.57 (dd, *J*₁ = 2.05 Hz, *J*₂ = 2.00 Hz, 1H, Ar-H), 8.13 (s, 1H, Ar-H), 8.41–8.44 (m, 2H, Ar-H), 8.46–8.51 (dd, *J*₁ = 9.25 Hz, *J*₂ = 5.10 Hz, 1H, Ar-H), 8.65–8.67 (dd, *J*₁ = 1.55 Hz, *J*₂ = 1.50 Hz, 1H, Ar-H), 8.96 (s, 1H, Ar-H), 9.28 (s, 1H,

(-NH-), 9.59 (s, 1H, -NH-); ^{13}C -NMR (DMSO- d_6 , 125 MHz) δ (ppm): 17.63, 23.06, 27.91, 52.08, 53.55, 66.25, 107.43, 107.52, 109.29, 117.54, 118.68, 119.04, 123.67, 124.58, 127.55, 129.85, 132.14, 134.20, 137.28, 137.67, 148.13, 151.32, 151.96, 155.04, 157.36, 159.41, 161.18, 161.53, 161.93; ESI-MS m/z : 533.32 [M + H] $^+$.

Ethyl 6-(7-(3-morpholinopropoxy)quinazolin-4-ylamino)benzo[b]thiophene-3-carboxylate(6e)

Yield 61.3%, mp 145 °C. IR (KBr, per cm): 3234.09, 2942.89, 1664.29, 1567.87, 1554.37, 1403.95, 1357.66, 1274.74, 1201.45, 1114.67, 958.46, 821.54; ^1H -NMR (DMSO- d_6 , 500 MHz) δ (ppm): 1.36 (t, $J = 7.15\text{ Hz}$, 3H, -CH $_2$ -CH $_3$), 1.93–1.97 (dd, $J_1 = 6.80\text{ Hz}$, $J_2 = 6.65\text{ Hz}$, 2H, -CH $_2$ -CH $_3$), 2.38 (s, 4H, 2 \times morpholine-CH $_2$), 2.44–2.50 (dd, $J_1 = 7.10\text{ Hz}$, $J_2 = 12.85\text{ Hz}$, 2H, -CH $_2$ -CH $_2$ -CH $_2$ -O-), 3.58 (t, $J = 4.40\text{ Hz}$, 4H, 2 \times morpholine-CH $_2$), 4.19 (t, $J = 6.30\text{ Hz}$, 2H, -CH $_2$ -CH $_2$ -CH $_2$ -O-), 4.34–4.38 (m, 2H, -CH $_2$ -CH $_2$ -CH $_2$ -O-), 7.18 (d, $J = 2.25\text{ Hz}$, 1H, Ar-H), 7.24–7.26 (dd, $J_1 = 2.00\text{ Hz}$, $J_2 = 2.00\text{ Hz}$, 1H, Ar-H), 7.94–7.96 (dd, $J_1 = 1.60\text{ Hz}$, $J_2 = 1.50\text{ Hz}$, 1H, Ar-H), 8.04 (d, $J = 8.80\text{ Hz}$, 1H, Ar-H), 8.20 (s, 1H, Ar-H), 8.47–8.53 (m, 3H, Ar-H), 9.83 (s, 1H, -NH-); ^{13}C -NMR (DMSO- d_6 , 125 MHz) δ (ppm): 14.12, 25.68, 53.33, 54.73, 61.39, 66.18, 66.26, 107.54, 109.29, 117.81, 118.25, 122.69, 123.52, 124.57, 130.74, 133.66, 136.41, 136.97, 138.80, 152.05, 154.52, 157.43, 161.97, 162.05; ESI-MS m/z : 493.22 [M + H] $^+$.

Ethyl 6-(7-(3-(4-methylpiperazin-1-yl)propoxy)quinazolin-4-ylamino)benzo[b]thiophene-3-carboxylate(6f)

Yield 43.7%, mp 162 °C. IR (KBr, per cm): 3274.59, 2937.10, 1712.51, 1619.94, 1577.51, 1529.30, 1450.23, 1413.59, 1292.09, 1230.38, 1149.38, 1072.24, 1014.39, 782.97, 754.04; ^1H -NMR (DMSO- d_6 , 500 MHz) δ (ppm): 1.34 (t, $J = 8.30\text{ Hz}$, 3H, -CH $_2$ -CH $_3$), 1.90–1.93 (t, $J = 6.20$, 2H, -CH $_2$ -CH $_3$), 2.18 (s, 4H, 2 \times morpholine-CH $_2$), 2.40–2.49 (m, 9H, 2 \times morpholine-CH $_2$, -CH $_2$ -CH $_2$ -CH $_2$ -O- and -CH $_3$), 4.17 (d, $J = 5.60\text{ Hz}$, 2H, -CH $_2$ -CH $_2$ -CH $_2$ -O-), 4.33–4.37 (m, 2H, -CH $_2$ -CH $_2$ -CH $_2$ -O-), 7.15 (s, 1H, Ar-H), 7.24 (d, $J = 8.85\text{ Hz}$, 1H, Ar-H), 7.97 (d, $J = 8.55\text{ Hz}$, 1H, Ar-H), 8.03 (d, $J = 8.70\text{ Hz}$, 1H, Ar-H), 8.19 (s, 1H, Ar-H), 8.50–8.53 (m, 3H, Ar-H), 9.86 (s, 1H, -NH-); ^{13}C -NMR (DMSO- d_6 , 125 MHz) δ (ppm): 14.10, 25.95, 45.31, 52.34, 54.15, 54.45, 61.36, 66.25, 107.50, 109.30, 117.74, 118.20, 122.63, 123.49, 124.63, 130.72, 133.62, 136.38, 137.00, 138.77, 152.03, 154.90, 157.42, 161.95, 162.02; ESI-MS m/z : 506.26 [M + H] $^+$.

Ethyl 6-(7-(3-(piperidin-1-yl)propoxy)quinazolin-4-ylamino)benzo[b]thiophene-3-carboxylate(6g)

Yield 57.9%, mp 170–171 °C. IR (KBr, per cm): 3390.58, 2933.73, 1709.27, 1619.05, 1570.93, 1526.82, 1454.64, 1410.53, 1228.07, 1127.82, 1069.67, 1053.63, 756.89;

^1H -NMR (DMSO- d_6 , 300 MHz) δ (ppm): 1.32–1.39 (m, 5H, -CH $_2$ -CH $_3$), 1.50 (d, 4H, 2 \times piperidine-CH $_2$), 1.93 (t, $J = 6.96\text{ Hz}$, 2H, -CH $_2$ -CH $_2$ -CH $_2$ -O-), 2.34–2.44 (m, 6H, 3 \times piperidine-CH $_2$), 4.18 (t, $J = 6.33\text{ Hz}$, 2H, -CH $_2$ -CH $_2$ -CH $_2$ -O-), 4.32–4.39 (m, 2H, -CH $_2$ -CH $_2$ -CH $_2$ -O-), 7.17 (d, $J = 2.37\text{ Hz}$, 1H, Ar-H), 7.24–7.27 (dd, $J_1 = 2.37\text{ Hz}$, $J_2 = 2.37\text{ Hz}$, 1H, Ar-H), 7.93–7.97 (d, $J_1 = 1.98\text{ Hz}$, $J_2 = 1.86\text{ Hz}$, 1H, Ar-H), 8.03 (d, $J = 8.82\text{ Hz}$, 1H, Ar-H), 8.21 (s, 1H, Ar-H), 8.47–8.53 (m, 3H, Ar-H), 9.85 (s, 1H, -NH-); ^{13}C -NMR (DMSO- d_6 , 125 MHz) δ (ppm): 14.11, 24.10, 25.57, 26.10, 54.07, 55.00, 61.38, 66.42, 107.53, 109.26, 117.82, 118.26, 122.69, 123.54, 124.56, 130.74, 133.65, 136.41, 138.79, 152.05, 154.91, 157.42, 161.96, 162.08; ESI-MS m/z : 491.4 [M + H] $^+$.

Ethyl 6-(7-(3-(pyrrolidin-1-yl)propoxy)quinazolin-4-ylamino)benzo[b]thiophene-3-carboxylate(6h)

Yield 36.9%, mp 170 °C. IR (KBr, per cm): 3426.94, 2964.10, 2875.39, 2788.61, 1618.01, 1575.58, 1519.66, 1456.02, 1432.58, 1417.45, 1338.38, 1230.38, 1130.10, 81.32, 717.40, 676.90; ^1H -NMR (DMSO- d_6 , 300 MHz) δ (ppm): 1.85–2.00 (m, 5H, -CH $_2$ -CH $_3$), 2.49–2.54 (m, 4H, 2 \times pyrrolidine-CH $_2$), 2.65 (t, $J = 6.93\text{ Hz}$, 2H, -CH $_2$ -CH $_2$ -CH $_2$ -O-), 3.51–3.55 (t, 4H, 2 \times pyrrolidine-CH $_2$), 3.85 (d, $J = 6.15\text{ Hz}$, 2H, -CH $_2$ -CH $_2$ -CH $_2$ -O-), 4.20 (t, $J = 6.27\text{ Hz}$, 2H, -CH $_2$ -CH $_2$ -CH $_2$ -O-), 7.17 (d, $J = 2.43\text{ Hz}$, 1H, Ar-H), 7.23–7.27 (dd, $J_1 = 2.43\text{ Hz}$, $J_2 = 2.43\text{ Hz}$, 1H, Ar-H), 7.81–7.84 (dd, $J_1 = 2.07\text{ Hz}$, $J_2 = 1.92\text{ Hz}$, 1H, Ar-H), 7.96–8.02 (dd, $J_1 = 5.10\text{ Hz}$, $J_2 = 11.91\text{ Hz}$, 2H, Ar-H), 8.49–8.62 (m, 3H, Ar-H), 10.02 (s, 1H, -NH-); ^{13}C -NMR (DMSO- d_6 , 125 MHz) δ (ppm): 22.96, 23.51, 26.15, 27.19, 47.25, 48.47, 51.86, 53.45, 66.01, 107.61, 109.37, 113.64, 117.69, 117.87, 118.37, 122.12, 123.11, 124.65, 125.14, 134.80, 136.54, 139.64, 140.58, 152.01, 154.97, 155.60, 157.55, 160.79, 161.92; ESI-MS m/z : 477.2 [M + H] $^+$.

Ethyl 6-(7-(3-(diethylamino)propoxy)quinazolin-4-ylamino)benzo[b]thiophene-3-carboxylate(6i)

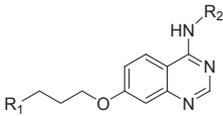
Yield 35.7%, mp 160 °C. IR (KBr, per cm): 3455.87, 3025.81, 2971.81, 2933.24, 2802.11, 1706.72, 1618.01, 1577.51, 1526.27, 1450.23, 1409.73, 1326.81, 1172.53, 1149.38, 1130.10, 1070.32, 1020.18, 912.18, 755.97, 674.97; ^1H -NMR (DMSO- d_6 , 300 MHz) δ (ppm): 0.89–0.97 (m, 6H, 2 \times diethylamine-CH $_3$), 1.28–1.37 (m, 3H, -CH $_2$ -CH $_3$), 1.85–1.91 (dd, $J_1 = 6.54\text{ Hz}$, $J_2 = 5.12\text{ Hz}$, 2H, -CH $_2$ -CH $_2$ -CH $_2$ -O-), 2.42–2.57 (m, 6H, 2 \times diethylamine-CH $_2$ and -CH $_2$ -CH $_3$), 4.17 (t, $J = 6.12\text{ Hz}$, 2H, -CH $_2$ -CH $_2$ -CH $_2$ -O-), 4.32–4.39 (t, 2H, -CH $_2$ -CH $_2$ -CH $_2$ -O-), 7.16 (s, 1H, Ar-H), 7.26 (d, $J = 9.03\text{ Hz}$, 1H, Ar-H), 7.97 (d, $J = 8.91\text{ Hz}$, 1H, Ar-H), 8.05 (d, $J = 8.76\text{ Hz}$, 1H, Ar-H), 8.20 (s, 1H, Ar-H), 8.47–8.53 (m, 3H, Ar-H), 9.86 (s, 1H, -NH-); ^{13}C -NMR (DMSO- d_6 , 125 MHz) δ (ppm): 11.83, 14.09, 26.45, 46.42, 48.57, 61.35, 66.23, 107.46, 109.27, 117.76, 118.20, 122.63, 123.48, 124.56, 130.71, 133.63, 136.38, 137.00, 138.77, 139.36, 152.05, 154.87, 157.41, 161.94, 162.10; ESI-MS m/z : 479.4 [M + H] $^+$.

Results and Discussion

Analysis of the inhibitory activity of new compounds against Aurora A/B kinases

We have synthesized nine compounds derived from 4-aminoquinazoline. To explore the mechanism of action of these new compounds, they were analyzed for the inhibition of Aurora A/B kinases. As shown in Table 1, most of the derivatives exhibit a higher selectivity for Aurora A than Aurora B. For example, compound **6d** has a lower IC₅₀ value (77 nM) for Aurora A than that for Aurora B (144 nM). Introduction of the active group of imatinib at the amine of C-4 of the quinazoline ring in compounds **6a–6d** improved the inhibitory activity as compared to aniline analogs

Table 1: Aurora A/B inhibitory activity of 4-aminoquinazoline derivatives



Compound	R ₁	R ₂	IC ₅₀ (nM) ^a	
			Aurora A	Aurora B
6a			179	191
6b			323	489
6c			98	202
6d			77	145
6e			NT	NT
6f			111	136
6g			822	NT
6h			62	146
6i			788	893
ZM447439			138	156

NT, not tested.

^aThe mean of at least two experiments.

6e–6i. To facilitate structure–activity relationship (SAR) analysis, we maintained the aniline group at R₂ while varying the side chain substituent at R₁. We found that variations of the secondary amine substituent at R₁ also resulted in dramatic changes in the inhibition of Aurora kinases. Compounds **6d** and **6h** with a pyrrole side chain showed the strongest inhibitory activity. Compounds **6b** and **6g** with a piperidine were significantly less potent than that of **6d** and **6h**. Compound **6a** with a morpholine at R₁ had also a modest activity to inhibit Aurora kinases. However, compound **6e**, which has a similar side chain, did not show a detectable activity against Aurora A/B. Replacement of the pyrrole group in compound **6d** with diethylamine group did not change the anti-Aurora activity (**6c**). On the other hand, the similar replacement in **6h** led to a significant reduction in the potency (**6i**). Compound **6f**, which has an *N*-methyl piperazine at R₁, exhibited a similar activity as ZM447439. Thus, changes in R₁ and R₂ side chains of 4-aminoquinazoline can improve the potency against Aurora kinases.

Analysis of antiproliferation activity

All the new compounds were evaluated for the potential to inhibit the growth of human tumor cells. For this purpose, we chose five cell lines representing different tumor types, including human acute monocytic leukemia cell line (U937), human chronic myeloid leukemia cell line (K562), human non-small cell lung cancer (A549), and human colon cancer (LoVo and HT29). The result of the study was summarized in Table 2. Most of the compounds had evident antiproliferation effect on solid tumor cell lines (A549, LoVo, and HT29) but poor effect on leukemia cells (U937 and K562). In particular, compounds **6c** and **6d** have IC₅₀ values ranging from 0.99 to 4.35 μM toward solid tumor cells, which are similar to or better than that of ZM447439 (IC₅₀ = 1.87–3.36 μM). In addition, compounds **6c** and **6d** also showed increased potency against leukemia cell lines as compared to ZM447439. We proposed that the higher cytotoxicity of these compounds could be attributed to the active group of imatinib substituted at the R₂ side chain. Indeed, compound **6b**, which has the same active group, also showed a similar activity against solid tumor cell lines with an IC₅₀ value of 2.01–2.85 μM. Also compound **6g**, which has a R₁ side chain (piperidine) same as **6b** but has an aniline group at R₂, exhibited a significant decrease in the cytotoxic activity (IC₅₀ > 100 μM for the most tumor cells) and poor potency against Aurora A/B kinases. On the other hand, compounds **6d** and **6h**, which have a pyrrolidine ring at the R₁ side chain but with different R₂ groups, inhibited the growth of solid tumor cells with IC₅₀ values from 1.98 to 5.44 μM. Thus, the side chain at R₁ also has a significant impact on the antiproliferation activity. Another interesting observation is that compound **6f**, which has an *N*-methyl piperazine at R₁ and a similar anti-Aurora kinase activity as ZM447439, was more potent than ZM447439 in inhibiting tumor cell growths. This difference may be

Compound	IC ₅₀ (μM) ^a				
	U937	K562	A549	LoVo	HT29
6a	>100	>100	>100	6.09	9.90
6b	11.30	>100	2.85	2.20	2.01
6c	7.20	4.00	0.99	1.65	4.35
6d	4.90	2.00	2.85	1.98	1.99
6e	>100	>100	6.23	5.88	2.16
6f	4.90	2.00	2.26	1.21	1.89
6g	>100	>100	>100	6.88	>100
6h	10.60	>100	2.53	5.44	4.26
6i	>100	>100	8.85	7.35	>100
ZM447439	>100	>100	3.36	1.87	2.23

^aCellular proliferation was determined by MTT assay.

Table 2: *In vitro* cytotoxic activity of target compounds against cancer cells

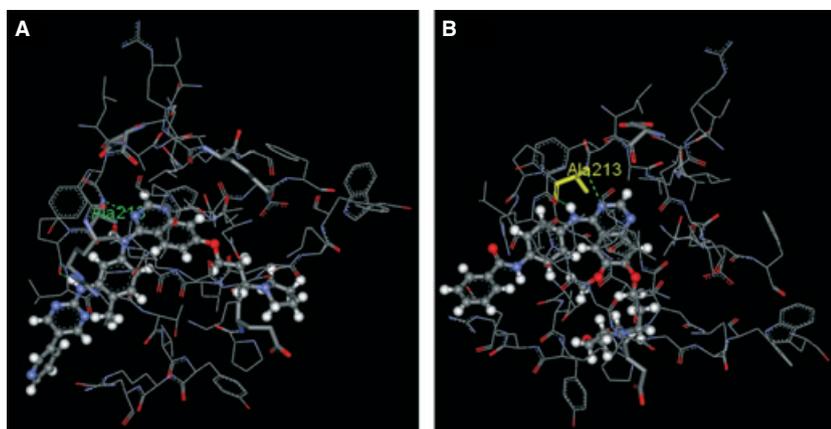


Figure 2: (A) Binding mode of compounds **6d** with Aurora A; (B) Binding interaction of ZM-447439 with Aurora A.

due to different permeability of these two compounds into cells.

Molecular docking analysis

The coordinate for the Aurora A structure was obtained from the RCSB Protein Data Bank (PDB ID: 2NP8). The protein

structure was presented using SYBYL 7.3 software (Tripos International, St. Louis, MO, USA) and subjected to molecular modeling, which indicates that compound **6d** would dock strongly into the ATP-binding site of Aurora A (Figure 2A). It also reveals that the compound participates in a hydrogen bond interaction with the amine group of Ala213 within the protein hinge region (Figure 2A). Interestingly, the

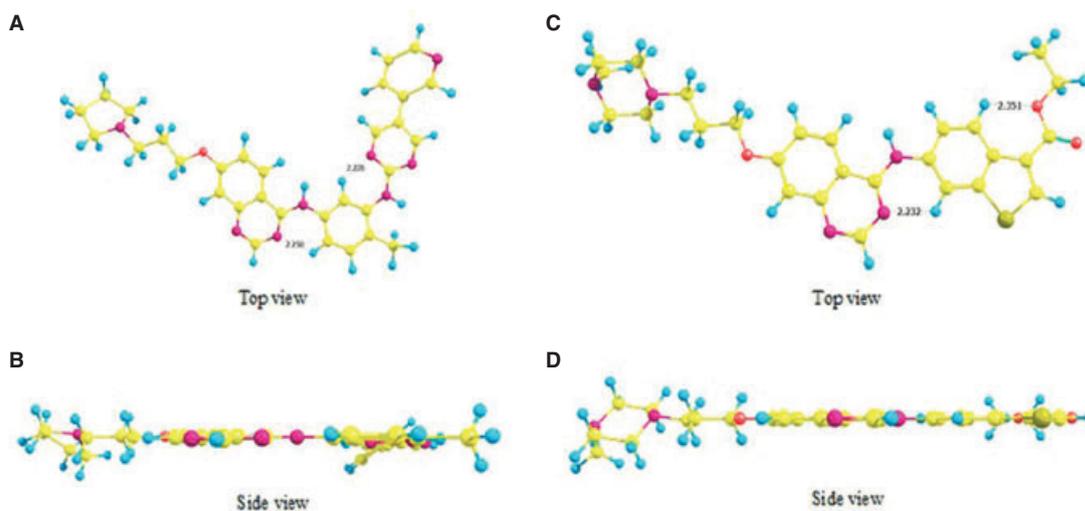


Figure 3: The optimized structures of the most active compounds **6d** (A, B) and **6f** (C, D).

amine between benzene and pyrimidine ring also makes a hydrogen bond to the hinge region of the kinase, which is apparently absent in the binding mode of ZM 447439 with Aurora A (Figure 2B). We proposed that this extra hydrogen bond may contribute to the apparent high activity of compound **6d** to inhibit Aurora kinases and cell growth.

Quantum chemical studies

To gain further insight into the structural basis for these compounds to interact with their targets, conformational analysis was performed using MM+ force field in HYPERCHEM 7.0^a. The analysis demonstrated that the lowest energy-minimized structures of compounds **6a–d** take a coplanar configuration of the benzopyrimidine, benzene, and pyrimidine moieties. Such configuration is apparently stabilized by intramolecular hydrogen bond between the CH group in the benzene ring and the adjacent nitrogen atom in the pyrimidine ring (Figure 3A, B). The compounds **6a–c** have a structure similar to **6d**. As for compounds **6e–i**, their benzopyrimidine and benzothiophene moieties are also configured into a structural plane (Figure 3C, D).

It has been reported that molecular electronic effects play a critical role in controlling the pharmacological activities of

drugs (22). In SAR analysis, quantum chemical parameters have yielded promising estimation for the correlation of the electronic effects with the biological activity (23–25). These parameters include the HOMO and the LUMO (26), which reflect the frontier molecular orbitals that dominate the reactivity of a molecule. Thus, we calculated the orbital energies of both HOMO and LUMO and their gaps for all the new compounds, the result of which is presented in Table 3. Significantly, we observed that compounds **6d** and **6f** have the lowest energy gap (ΔE) of 4.0115 and 3.9426 eV, respectively, and both exhibit the highest cytotoxic activity.

We also plotted the HOMO and LUMO of these compounds to analyze the atomic contributions of these orbitals to the sites for electronic transfer between individual compound and its biological target. As the representatives, the plots of the HOMO and LUMO of the most active compounds (**6d** and **6f**) are presented in Figure 4. The results illustrate that the HOMO molecular orbital of **6d** is mainly located in the benzopyrimidine, benzene, and pyrimidine moieties, indicative of the existence of possible reactive sites that may be involved in electrophilic attacks. On the other hand, the LUMO of **6d** is primarily concentrated on pyridine and pyrimidine rings in which the negatively charged polar residues of the receptor may be

Table 3: Energies of both HOMO and LUMO and their gaps (in eV) calculated for all compounds

Compound	E_{HOMO} (eV)	E_{LUMO} (eV)	ΔE (eV)
6a	-5.36205	-1.34126	4.020786
6b	-5.33647	-1.32493	4.011534
6c	-5.33783	-1.32520	4.012623
6d	-5.33538	-1.32384	4.011534
6e	-5.56205	-1.54507	4.016976
6f	-5.46627	-1.52358	3.942689
6g	-5.53484	-1.51024	4.024596
6h	-5.53321	-1.50834	4.024868
6i	-5.53647	-1.51052	4.025956

HOMO, highest occupied molecular orbital; LUMO, lowest unoccupied molecular orbital.

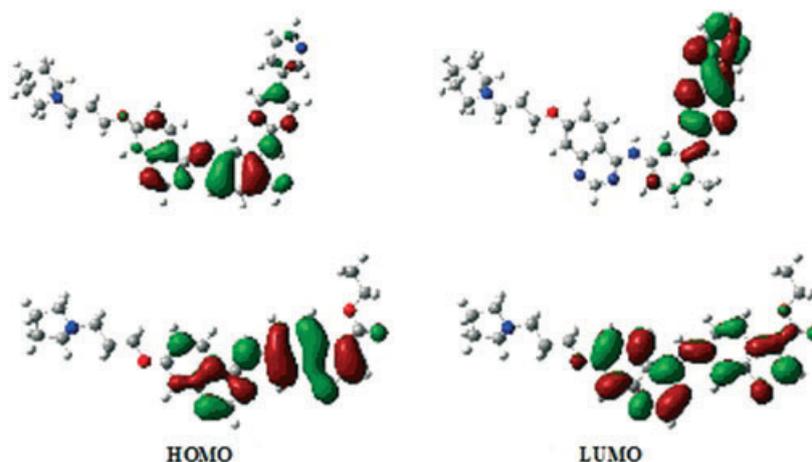


Figure 4: Plots of the highest occupied molecular orbital (HOMO) and lowest unoccupied molecular orbital (LUMO) of the most active compounds **6d** (upper) and **6f** (lower).

favorable. When compared with **6d**, the HOMO and LUMO of **6f** are quite similar and delocalized in benzopyrimidine and benzothiophene moieties.

Conclusion

We have designed a series of 4-aminoquinazoline derivatives and analyzed their anti-Aurora kinase activities. Most of the compounds showed a significant antiproliferation activity against three solid tumor cell lines, but no or poor activity against two leukemia cell lines. Introduction of the active group of imatinib at the R2 side chain or a pyrrolidine ring at the R1 side chain impacts significantly the activity for both anti-Aurora kinases and antigrowth of tumor cells. Molecular modeling suggested that the strong inhibitory activity of compound **6d** may be attributed to an extra hydrogen bond between the amine at the pyrimidine ring and the hinge region of targeted kinases. Quantum chemical calculation of electronic parameters including HOMO and LUMO indicates an association of molecular orbitals at low energy with antiproliferation activities of these compounds. These findings may be useful in improving the efficacy of Aurora A/B kinase inhibitors by incorporating some structural groups that are able to modify the electronic property of the existing anti-Aurora compounds.

Acknowledgments

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Notes

^a(2002) HyperChem (Molecular Modeling System) Hypercube, Inc., Gainesville, FL, USA.

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