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Synthesis of 3D-rich heterocycles: Hexahydropyrazolo[1,5-a]pyridin-2(1H)ones and octahydro-2H-2a,2a1-diazacyclopenta[cd]inden-2-ones.

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Synthesis of 3D-rich heterocycles: Hexahydropyrazolo[1,5-*a*]pyridin-2(1*H*)ones and octahydro-2*H*-2a,2a¹-diazacyclopenta[*cd*]inden-2-ones.

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Abstract — Two cyclic azomethine imines, 7-methyl- and 7-phenyl-2-oxo- Δ^7 hexahydropyrazolo[1,5-*a*]pyridin-8-ium-1-ide were prepared in seven steps from the respective commercially available δ -keto acids. The addition of Grignard reagents followed by N-alkylation at position 1 afforded the 1,7,7-trisubstituted hexahydropyrazolo[1,5-*a*]pyridin-2(1*H*)-ones, whereas 1,3-dipolar cycloadditions of these dipoles to typical acetylenic and olefinic dipolarophiles gave 4a-substituted 2a,2a¹-diazacyclopenta[*cd*]indene derivatives, as the first representatives of a novel heterocyclic system. Regio- and stereoselectivity as well as the mechanism of these [3+2]-cycloadditions were evaluated using computational and experimental

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methods. The data obtained were in agreement with the polar concerted cycloaddition mechanism *via* the energetically favorable *syn/endo*-transition states.

Keywords: pyrazolo[1,5-*a*]pyridine; 2a,2a¹-diazacyclopenta[*cd*]indene; azomethine imines; [3+2]-cycloadditions; Grignard reagents; saturated heterocycles.

1. Introduction

Heterocyclic systems are common building blocks for the synthesis of various biologically important and naturally occurring compounds. Consequently, heterocycles are commonly used building blocks for applications in medicinal chemistry, catalysis, and material science.¹ In this context, pyrazolo[1,5-*a*]pyridine (1)² belongs to a group of well-explored systems with over 100,000 hits and over 2,500 references according to a SciFinder®³ substructure search. Derivatives of **1** exhibit different biological activities, such as antiviral,⁴ inhibition of reverse transcriptase,⁵ dopamine D3 and D4 antagonist,⁶ dopamine D3 agonist,⁷ diuretic adenosine A1 antagonist,⁸ and intercalating activity.⁹ A phosphodiesterase inhibitor, Ibudilast (**2**), is an approved anti-inflammatory drug.¹⁰ In contrast to thousands of known derivatives of pyrazolo[1,5-*a*]pyridine (**1**), only ~120 fully saturated derivatives of **3** are known to date,³ whereas the tricyclic analogues **4** (2a,2a¹-diazacyclopenta[*cd*]indenes) are unknown to the best of our knowledge. Note that two related examples can be found in the literature. The first example is a theoretical report on **4** as a part of a heterofullerene system,¹¹ while in the second example **4** was a part of a cage compound (Figure 1).¹²



Figure 1. Pyrazolo[1,5-*a*]pyridine (1), Ibudilast (2), less explored saturated analogues 3, the unknown saturated tricyclic system 4, and the target structures 5 and 6.

In the context of our ongoing work on the synthesis of 3-pyrazolidinones and pyrazole analogues of histamine,¹³ we recently reported two syntheses of tetrahydropyrazolo[1,5-c]pyrimidine-2,7(1*H*,3*H*)-diones as the first representatives of a novel saturated heterocyclic system.^{14,15} Subsequently, a library of related tetrahydropyrazolo[1,5-c]pyrimidine-3-carboxamides as novel conformationally constrained pyrazole analogues of histamine was also synthesized.¹⁶ In continuation, we focused on 1,7,7-trisubstituted hexahydropyrazolo[1,5-a]pyridin-2(1*H*)-ones **5** and their tricyclic analogues (3,4,4a-trisubstituted octahydro-2*H*-2a,2a¹-diazacyclopenta[cd]inden-2-one) **6** (Figure 1). A literature search revealed that scaffolds **5** and **6** were unknown, which prompted us to focus our attention on their synthesis since the availability of this type of template would enable the preparation of compound libraries, suitable for screening for various activities or applications. The results of this study are reported herein.

2. Results and Discussion

Initially, we attempted to access the title compounds *via* 7-substituted 2-oxo-2,3,3a,4,5,6-hexahydropyrazolo[1,5-*a*]pyridin-8-ium-1-ides **12** as key-intermediates available by microwave-assisted cyclization of pent-4-en-1-yl N-Boc-hydrazones **11**¹⁷ following the procedure described recently by Beauchemin and co-workers.¹⁸ First, hex-5-en-2-one (**9a**) was prepared by Cu(I)-catalyzed treatment of acetyl chloride (**7a**) with pent-4-en-1-ylmagnesium bromide (**8a**).¹⁹ The crude ketone **9a** was, without purification, transformed further with Boc-carbazate (**10**) into the corresponding hydrazone **11a**, which was isolated in 56% yield over two steps. Subsequent cyclization of hydrazone **11a** was performed in trifluoromethylbenzene under microwave irradiation at 150 °C to afford the desired azomethine imine **12a** in 60% yield.¹⁸ Finally, stereoselective reduction of dipole **12a** with excess PhMgBr at 0–20 °C followed by workup using column chromatography furnished the (3a*S**,7*S**)-isomer **13'a** in 31% yield (Scheme 1).



Scheme 1. Four-step synthesis of compound 13'a.



Reaction conditions: (i) pent-4-en-1-ylmagnesium bromide (8a), THF, CuI (4 mol%), r.t. (Ref. 19); (ii) BocNHNH₂ (10), MeOH, AcOH, r.t. (Ref. 18); (iii) μ -waves, 300 W, C₆H₅CF₃, 150 °C, 3 h (Ref. 18); and (iv) excess PhMgBr (8b), THF, 0 \rightarrow 20 °C, followed by column chromatography.

The successful preparation of **13'a** confirmed the viability and simplicity of the original synthetic approach. However, the microwave-assisted cyclization of **11a** into **12a** was the bottleneck of this synthetic sequence because in our hands the reaction was reproducible only on a ~0.3 mmol scale, i.e. on a similar scale as that reported previously (0.2 mmol).¹⁸ In addition, the incomplete conversion and the formation of by-products required a tedious chromatographic workup to obtain pure **12a**. Thus, despite its simplicity, the original synthetic approach was not suitable to provide sufficient amounts of the key-intermediates **12** for further transformations. Consequently, a seven-step synthesis of **12** was developed based on a synthetic method applied previously for the preparation of related pyrazolidinones.¹⁵ The synthesis commenced with an almost quantitative one-pot transformation of commercially available γ -acetyl- (**14a**) and γ -benzoylbutyric acid (**14b**) into the δ -keto acid ketals **16a**²⁰ and **16b**;²¹ these steps were composed

of ketalization and esterification with ethylene glycol and trimethyl orthoformate (TMOF) and were followed by hydrolysis of the intermediate ketal-esters 15a,b.^{21,22} Masamune-Claisen condensation of the acids 16 afforded the corresponding β -keto esters 17a,b in quantitative yields. Then, reduction of ketones 17, followed by O-mesylation of alcohols 18, and cyclization of O-mesylates 19 with hydrazine hydrate furnished the pyrazolidinones 20a and 20b in good yields over three steps. Finally, acidolytic removal of the ketal protecting group and concomitant cyclization furnished the desired key-intermediates $12a^{18}$ and $12b^{23}$ in 80% and 60% yield, respectively (Scheme 2).





Reaction conditions: (i) ethylene glycol, TMOF, H₂SO₄ (cat.), r.t. (Ref. 22); (ii) 2 M aq. NaOH, H₂O–MeOH, r.t. (Ref. 22); (iii) CDI, THF, r.t., then MeO₂CCH₂CO₂K, MgCl₂, r.t.; (iv) NaBH₄,

2 3	MaOH 0.8C; (c) MaCl meriding 0.8C; (ci) N.H. H.O. MaOH 50.8C; and (ciii) EtOH TEA (ast)
4	MeOH, 0 °C; (V) MsCl, pyridine, 0 °C; (VI) N_2H_4 · H_2O , MeOH, 50 °C; and (VII) EtOH, 1FA (cat.),
6	reflux.
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Next, the addition of Grignard reagents to dipoles **12a** and **12b** was studied. First, we attempted to add excess PhMgBr (**8b**) to the dipole **12a** at a lower temperature; however, at -78 °C, no reaction occurred after several hours. When the reaction was performed at -20 °C for 1 h followed by treatment at room temperature for 12 h, pure ($3aS^*,7S^*$)-isomer **13'a** was isolated in 54% yield. The other epimer could not be detected in the reaction mixture. As expected, the epimer **13a** was exclusively obtained in 69% yield upon treatment of the 7-phenyl analogue **12b** with excess MeMgBr (**8c**) under the same reaction conditions. However, the addition of MeMgBr (**8c**) to **12a** gave compound **13b** in 31% yield. N-Alkylation of **13a**, **13'a**, and **13b** with methyl iodide or benzyl bromide in DMF in the presence of K₂CO₃ furnished the title compounds **5a**, **5'a**, **5'b**, and **5c** in good yields. The stereoselectivity of the addition reaction is explainable by the preferential attack of the Grignard reagent **8** to the less hindered face of the dipole **12** to give the major isomer with *syn*-oriented R' and H–3a (Scheme 3).

Scheme 3. Synthesis of title bicyclic compounds 5, 5', 13, and 13' and the proposed stereochemistry of the addition to the C=N bond.



Reaction conditions: (i) excess PhMgBr (**8b**) or MeMgBr (**8c**), THF, $-20 \text{ °C} \rightarrow \text{r.t.}$; (ii) MeI or BnBr, K₂CO₃, DMF, r.t.

The 1,3-dipolar characteristic of azomethine imines **12a** and **12b** was tested in [3+2]cycloadditions to acetylenic- (**21a–d**) and olefinic dipolarophiles **22a–c**. Most cycloadditions were highly regio- and stereoselective and gave the corresponding cycloadducts **23–26** as single isomers upon workup using flash chromatography. Cycloadditions of **12a,b** to dimethyl

acetylenedicarboxylate (DMAD) (21a) and terminal ynones 21b-d proceeded at room temperature to give the major $(4aS^*, 7aS^*)$ -isomers 23a,b and 24a-f in 42-73% yields. Surprisingly, methyl propiolate (21e) did not react at room temperature, and heating at 80 °C was required to obtain cycloadducts 24g and 24h. In the reaction of 12b with 21e, the minor isomer 24'h was also isolated. The CuI-catalyzed reactions of 12a,b with 21e occurred at room temperature to give inseparable 85:15 mixtures of cycloadducts 24gh and methyl (E)-3-[ethyl(isopropyl)amino]acrylate (29).²⁴ The reactions of dipoles 12 with olefinic dipolarophiles **22a–c** required heating at 80 °C to achieve satisfactory conversion into the products. Treatment of 12a with methyl acrylate (22a) produced a mixture of products; upon chromatographic separation, the *endo*-cycloadduct **25a** and the regioisomeric *exo*-adduct **25'a** were isolated in 11% and 16% yield, respectively. The reactions of the 7-phenyl analogue 12b with methyl (22a) and tert-butyl acrylate (22b) were highly regio- and stereoselective and afforded the major endoisomers 25b and 25c as single products. Cycloadditions of 12a and 12b to N-phenylmaleimide (22c) followed by chromatographic separation furnished the major *exo*-isomers 26a and 26b in 13% and 40% yield, respectively. To evaluate the further diversification of the core scaffold, the acidolytic deprotection of the carboxy function of cycloadduct 25c gave the carboxylic acid 27 in 48% yield. Amidation of 27 using bis(pentafluorophenyl) carbonate (BPC) as the activating reagent furnished carboxamides **28a-c** in 70–79% yields (Scheme 4, Table 1).



Reaction conditions: (i) DMAD (**21a**), toluene or CH₂Cl₂, r.t.; (ii) ynone **21b–d** or methyl propiolate (**21e**), CH₂Cl₂, r.t. or 80 °C (pressure vessel); (iii) methyl propiolate (**21e**), CuI (20 mol%), DIPEA (20 mol%), CH₂Cl₂, r.t.; (iv) methyl acrylate (**22a**) or *t*-butyl acrylate (**22b**), toluene or CH₂Cl₂ (pressure vessel), 80 °C; (v) *N*-phenylmaleimide (**22c**), toluene, 80 °C followed by column chromatography; (vi) CH₂Cl₂–TFA (2:1), r.t.; and (vii) BPC, Et₃N, DMF, r.t., 1 h followed by RR'NH, Et₃N, r.t., 24 h.

^{*a*}) In the CuI-catalyzed reactions (see iii), by-product **29** was also formed.²⁴

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Table 1. Experimental	data or	n tricyclic	compounds	23-28.
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23a Me 94:6 60	
23b Ph 100:0 56	
24a Me Me - 93:7 73	
24b Ph Me - 91:9 69	
24c Me Ph - 90:10 67	
24d Ph Ph - 100:0 71	
24e Me CH ₂ NHBoc - 93:7 42	
24f Ph CH ₂ NHBoc - 89:11 50	
24g Me OMe - 93:7 (100:0 ^b) 56 (47 ^b)	
24h Ph OMe - $85:15(100:0^b) 60(59^b)$	
24'h ^c Ph OMe - $100:0^d$ 11	
25a Me Me - $100:0^d$ 16	
25'a Me Me - ^{<i>d,e</i>} 11	
25b Ph Me - ^e 44	
25c Ph <i>t</i> -Bu - $100:0^{e,f}$ 77	
26a Me 78:22 13	
26b Ph 90:10 40	
27 Ph 48	
28a - H Bn - 79	
28b - H (CH ₂) ₃ OH - 76	
28c - piperidin-1-yl - 70	

^{*a*}) Determined using ¹H NMR. ^{*b*}) CuI-catalyzed reaction. ^{*c*}) Minor epimer. ^{*d*}) Upon separation by

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column chromatography. ^{*e*}) The isomer ratio could not be determined due to overlapped signals in the ¹H NMR spectrum of the crude product. ^{*f*}) Upon purification using flash column chromatography.

The regioselectivity of the cycloadditions to terminal ynones **21b**–**d** and alkyl acrylates **22** was in agreement with the regioselectivity of closely related thermal^{13,25,26} and Cu-catalyzed reactions.^{13,26,27} The preferential formation of the regioisomers **24** and **25** is in line with the electrostatically controlled approach of the polarized dipolarophile **21** or **22** to the mesomeric structure **12** via the proposed transition states **TS1** and **TS2** (Scheme 5). Facial selectivity of cycloadditions to **12a,b** is explainable by the preferential attack of the dipolarophile **21** or **22** from the less hindered face of the dipole **12** via the proposed transition states **TS1**–**TS3**. Accordingly, the *endo*-attack of the acrylate **22** via **TS2** should lead to the major diastereomer **25**, whereas the *exo*-approach of maleimide **22c** via **TS3** should give the major *exo*-isomers **26** (Scheme 5).





3. Structure determination.

The structures of novel compounds 5a,c, 5'a,b, 13a,b, 13'a, 17a,b, 18a,b, 19a,b, 20a,b, 23a,b, 24a–h, 24'h, 25a–c, 25'a, 27, and 28a–c were determined using spectroscopic methods (IR, ¹H and ¹³C NMR, COSY, HSQC, HMBC and NOESY spectroscopy, and MS). The structure and purity of compounds 12b, 5'a, and 5'b were additionally determined via elemental analyses for

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C, H, and N. Crude intermediates **16a,b**, **17a,b**, **18a,b**, **19a,b**, and **20a,b** were used in the following transformation without any purification.

The relative configurations of bicyclic (5, 5', 13, and 13') and tricyclic compounds 23–26 were determined by ¹H NMR and NOESY spectroscopy.²⁸ The structures of structurally representative compounds 5'b, 13'a, 23b, 24b, 26b, and 28a were unambiguously determined using X-ray diffraction.²⁸ The crystal structure of cycloadduct 24b is depicted in Figure 2.



Figure 2. ORTEP drawing of the **24b** molecule showing the atom labelling. The displacement ellipsoids are drawn with a 30% probability level and the hydrogen atoms are shown as small spheres of arbitrary radii.

4. Computational determination of the mechanism and selectivity of [3+2]-cycloadditions

In contrast to highly regioselective catalyzed reactions,²⁷ thermal cycloadditions of azomethine imines to terminal acetylenes usually furnish mixtures of regioisomers.^{25,26} Intrigued

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by the high regioselectivity of thermal cycloadditions of dipoles **12** (cf. Scheme 4), we attempted to find a plausible mechanistic explanation²⁹ using computational methods. Dipoles **12a,b**, 3-butyn-2-one (**21b**), methyl propiolate (**21e**), and methyl acrylate (**22a**) were chosen as model reactants.

All computations were performed using the Gaussian 09 program suite.³⁰ Geometry optimization of all stationary points was performed using DFT methods at the B3LYP/6-311+G(d,p) level of theory.³¹ First, the ideal gas approximation under the standard conditions was assumed and then the polarizable continuum model (PCM) for solvation by toluene was used for the computations. The DFT study started with an evaluation of the energetic and structural aspects of possible regio- and stereoisomeric transition states. The *syn/anti*-approach refers to facial selectivity with respect to the angular proton H–3a, while for the acetylenic dipolarophiles **21b** and **21e**, the *endo*-orientation refers to the orientation of the C=O function in the transition state.²⁸

The calculated activation and distortion/interaction parameters³² as well as asynchronicity parameter $(\Delta d_{\text{TS/P}})^{32\text{g,h}}$ are reported in Table 2. In all reactions, the *syn*-transition states were found to be energetically favourable. The differences of the Gibbs energy of activation values between the *syn* and the *anti* forms in the range of 2.7–4.0 kcal mol⁻¹ demonstrate that the reaction channel prefers the *syn*-approach to the dipole **12**. The typical asynchronicity measure value, $\Delta d_{\text{TS/P}} \sim 0.3$, suggests that reactions are concerted, although asynchronous. The computed free energy of activation values in toluene as the reaction medium are significantly smaller ($\Delta \Delta G^{\ddagger}$ ≈ 8 kcal mol⁻¹) for the ynone-derived cycloadducts **24** compared to the acrylate-derived cycloadducts **25**. In the reactions with acrylate **22a**, the 7-phenyl dipole **12b** has a lower energy and more asynchronous transition state than its 7-methyl analogue **12a**. Transition states leading to the minor regioisomers **25'** are higher in energy than those for the major isomers **25**. Finally, the energy difference between the 4a-methyl regioisomers **25a** and **25'a** ($\Delta \Delta G^{\ddagger} = 2.8$ kcal/mol) is

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smaller than that for the 4a-phenyl analogs **25b** and **25'b** ($\Delta\Delta G = 6.1^{\ddagger}$ kcal/mol). The calculated parameters given in Table 2 are in agreement with the experimental results in terms of reactivity and selectivity (cf. Scheme 4, Table 1).²⁸

Table 2. B3LYP/6-311+G(d,p) calculated activation energies, distortion energies (ΔE_d^{\ddagger}), interaction energies (ΔE_i^{\ddagger}), and asynchronicity degrees for transition states.

TS^{\ddagger}		B3LYP/6-331+G(d,p)							
		$\Delta G^{\ddagger a}$	$\Delta G^{\ddagger b}$	$\Delta E^{\ddagger c}$		$\Delta E_d^{\ddagger d}$		$\Delta E_i^{\ddagger d}$	$\Delta d_{\mathrm{TS/P}}^{e}$
					12	21/22	Total	_	
1	24a-anti/exo	28.9	32.0	16.2	10.6	14.0	24.6	-8.4	0.30
2	24a-anti/endo	26.6	31.8	16.0	11.0	13.9	24.9	-8.8	0.28
3	24a-syn/exo	29.3	28.8	13.6	8.2	14.2	22.4	-8.8	0.40
4	24a-syn/endo	26.5	27.9	13.7	9.2	13.8	23.0	-9.3	0.33
5	24g-anti/exo	29.4	31.5	16.2	9.2	18.1	27.3	-11.2	0.53
6	24g-anti/endo	28.8	31.0	12.3	10.7	14.3	25.0	-12.7	0.53
7	24g-syn/exo	26.2	27.4	12.3	11.4	20.4	31.8	-19.5	0.57
8	24g-syn/endo	26.1	27.4	13.5	9.2	14.2	23.5	-10.0	0.70
9	25a-anti/exo	36.8	40.0	23.3	17.1	14.1	31.2	-7.9	0.18
10	25a-anti/endo	36.2	39.3	22.3	16.6	14.7	31.2	-8.9	0.27
11	25a-syn/exo	33.6	36.6	19.7	15.5	14.2	29.8	-10.1	0.19
12	25a-syn/endo	32.5	35.3	18.6	14.7	14.6	29.3	-10.7	0.24
13	25'a-anti/exo	38,3	_	24,6	_	_	_	_	0.30
14	25'a-anti/endo	38,8	_	25,1	_	_	_	_	0.37

15	25'a-syn/exo	35,3	—	21,5	-	-	-	-	0.36
16	25'a-syn/endo	35,3	—	21,6	-	_	-	-	0.37
17	25b-anti/exo	36.0	37.2	22.4	13.5	15.4	28.9	-6.5	0.30
18	25b-anti/endo	34.1	35.1	20.3	10.4	17.0	27.4	-7.2	0.38
19	25b-syn/exo	34.3	35.2	20.1	13.1	15.7	28.8	-8.7	0.30
20	25b-syn/endo	31.2	32.4	17.6	12.1	15.3	27.4	-9.8	0.33
21	25'b-anti/exo	40,0	_	26,3	-	-	-	-	0.20
22	25'b-anti/endo	39,4	_	25,4	_	_	_	_	0.24
23	25'b-syn/exo	38,0	_	24,0	_	_	_	_	0.22
24	25'b-syn/endo	37,3	-	23,4	-	-	-	-	0.24

^{*a*} ΔG[‡]=G_{TS}-G_{dipole}-G_{dipolarophile} at 298 K in gas. ^{*b*} ΔG[‡] at 298 K in toluene as a solvent. ^{*c*} Zero-point energy corrected values (EZPE) of B3LYP/6-311+G(d,p). ^{*d*} ΔE_d[‡]_{dipole}, ΔE_d[‡]_{dipolarophile}, and ΔE_d[‡]_{total} are the distortion energies of the dipole, dipolarophile, and total distortion energy. ΔE_i[‡] indicates the interaction energy between distorted fragments. ^{*e*} Δd_{TS/P} = |(C-C)_{TS}/(C-C)_P-(C-N)_{TS}/(C-N)_P|.

Transition states leading to regioisomers **25a** and **25'a** are shown in Figure 3. In the transition state for the major isomer, TS**25a**-*syn/endo*, the C–N bond is shorter than the C–C bond, while these values are inversed in TS**25'a**-*syn/endo*. This result suggests that the C–C bond formation is more advanced in TS**25'a**,**b**-*syn/endo*, while the formation of the C–N bond is more advanced in TS**25'a**-*syn/endo*.



TS25a-syn/endo



Figure 3. Most favourable transition states for cycloaddition, $12a + 22a \rightarrow 25a + 25'a$, in the gas phase, calculated at the 6-311+G(d,p) level.

The electrophilicity ω and nucleophilicity *N* values³³ for the dipoles **12a**,**b** and dipolarophiles **21b**, **21e**, and **22a** are displayed in Table 3. All dipolarophiles, **21b**, **21e**, and **22a**, have high electrophilicity indices, 1.93, 1.97, and 2.20 eV, respectively, and are classified as strong electrophiles on the electrophilicity scale.³⁴ However, the dipoles **12a** and **12b** present moderate to strong respective electrophilicity indices of 1.39 and 2.13 eV, respectively, while both are classified as a strong nucleophiles due to their high nucleophilicity index, *N* > 3 eV.

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Entry	Compound	η	μ	ω (eV)	N(eV)
1	7-Me-dipole 12a	4.87	-3.67	1.39	3.04
2	7-Ph-dipole 12b	3.92	-4.09	2.13	3.10
3	3-Butyn-2-one (21b)	5.71	-5.01	2.20	1.27
4	Methyl propiolate (21e)	6.47	-5.05	1.97	0.85
5	Methyl acrylate (22a)	6.33	-4.94	1.93	1.03

Table 3. Electrophilicity ω and nucleophilicity N of dipoles **12a**,**b** and dipolarophiles.

The frontier molecular orbital (FMO) analysis for the cycloadditions studied show that the main interactions occur between the HOMO_{dipole} of dipoles **12a**,**b** and the LUMO_{dipolarophile} of the electron-poor dipolarophiles **21b**,**e** and **22a** due to the very different energy gaps, $\Delta E' - \Delta E > 1.5$ eV (Figure 4). In terms of favourable FMO interactions,²⁹ similar HOMO orbital coefficients at N(1) and C(7) in **12a** and larger coefficients at N(1) in the phenyl analogue **12b**²⁸ indicate a greater regioselectivity for the phenyl analogue **12b**, which was also observed experimentally.



Figure 4. FMO diagram of HOMO-LUMO orbitals calculated by NBO6/6-311+G(d,p) using PCM: toluene.

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The high asynchronicity of the cycloaddition of dipole **12a** to 3-butyn-2-one (**21b**) that was determined theoretically (cf. Table 2, entry 8) suggested the possibility of a stepwise mechanism.²⁹ To check this possibility experimentally, the kinetics of this cycloaddition was investigated. The reaction progress was followed using ¹H NMR in CDCl₃, CD₃CN, and DMSO- d_6 by monitoring the disappearance of the dipole **12a**. The finding of no significant solvent effect on the reaction kinetics was clearly in agreement with the concerted 1,3-dipolar reaction mechanism (Figure 5).



Figure 5. The kinetics of the reaction $12a + 21b \rightarrow 24a$ in CDCl₃, CD₃CN, and DMSO-*d*₆.

Next, the reaction kinetics was measured in CD₃CN at different temperatures (302, 312, 322, and 332 K) as a pseudo-first order reaction with respect to butynone **21b**.²⁸ Acetonitrile was selected as the solvent of choice due to the appropriate solubility of all reactants within the temperature range needed to construct the Eyring plot (Figure 6).³⁵ The corresponding experimental activation parameters were determined as $\Delta H^{\ddagger} = 13.8 \pm 0.1$ kcal/mol, $\Delta S^{\ddagger} = -27.2 \pm 0.2$ cal/mol K, and $\Delta G^{\ddagger} = 21.9 \pm 0.1$ kcal/mol. The experimental results are in fairly good

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agreement with the computed values (cf. Table 2, entry 8); however, the strong negative entropy value suggests a highly ordered rate-determining transition state, as expected for a polar concerted cycloaddition.²⁹



Figure 6. Eyring plot for the cycloaddition between dipole 12a and butynone 21b.

5. Conclusion

A seven-step synthesis of C,N,N-cyclic azomethine imines, 7-substituted 2-oxo- Δ^7 -hexahydropyrazolo[1,5-*a*]pyridin-8-ium-1-ides **12**, from δ -acyl butyric acids **14** was developed as an alternative to the previously described three-step process starting with acid chlorides **7** and pent-4-en-1-ylmagnesium bromide (**8**).¹⁸ Though requiring a longer synthesis time, the present method allows large-scale preparation of cyclic dipoles **12**, while the shorter and more elegant three-step synthesis¹⁸ has a scale limitation (< 0.5 mmol). The stereoselective addition of

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Grignard reagents to cyclic azomethine imines 12 gave 7,7-disubstituted hexahydropyrazolo[1,5alpyridin-2(1H)-ones 13 or 13', which were further N-alkylated into the title 1,7,7-trisubstituted compounds 5 and 5'. [3+2]-Cycloadditions of 12 were highly stereoselective particularly in reactions with acetylenes 21 (one regioisomer, d.r. \geq 89:11), whereas with olefins 22, the stereoselectivity was lower (d.r. \geq 78:22). Interestingly, thermal cycloadditions to terminal acetylenes **21b**–e were as regioselective as CuI-catalyzed reactions with methyl propiolate (**21e**). Moreover, the non-catalyzed reactions were even cleaner because they did not lead to the byproduct enaminone 29, which was difficult to separate. Acidolytic deprotection of the carboxy function gave the carboxylic acid 27, which was amidated into carboxamides 28a-c. Both reactions, the addition and [3+2]-cycloaddition exhibit the same stereocontrol leading to the major isomers in which R-C(7) and H-C(3a) from the parent dipole become anti-oriented. Regio- and stereoselectivity as well as mechanism of these [3+2]-cycloadditions were evaluated by computational and experimental methods supporting a polar concerted cycloaddition mechanism with the most favorable energetically syn/endo-transition states ($\Delta\Delta G \sim 3$ kcal/mol). To the best of our knowledge, the title compounds, 23–28, are the first known representatives of $2a_{a}2a^{1}$ -diazacyclopenta[*cd*]indene, which is an unexplored saturated heterocyclic system. In summary, we developed a viable synthetic protocol for the preparation of cyclic azomethine imines 12 as useful intermediates in the synthesis of 3D-rich saturated heterocycles, which may serve as a starting point in the search for novel lead compounds in medicinal chemistry, chemical biology, and material science.

6. Experimental

6.1. General methods. Melting points were determined on a Kofler micro hot stage and on an automated melting point system. The NMR spectra were recorded in CDCl₃ and DMSO-d₆ using TMS as the internal standard on a 300 MHz or 500 MHz instrument at 300 and 500 MHz for ¹H and at 75.5 and 126 MHz for ¹³C nucleus, respectively. Mass spectra were recorded on time-of-flight (TOF) mass spectrometer equipped with a double orthogonal electrospray source at athmospheric pressure ionization (ESI) coupled to an HPLC instrument. IR spectra were recorded on a FTIR ATR spectrophotometer. Microanalyses were performed by combustion analysis on a CHN analyzer. Microwave-assisted reactions were performed in a single mode microwave instrument in pressure reaction vessels. Reaction times refer to hold times at the temperature indicated not the total irradiation times. The temperature was measured using IR temperature sensor of the instrument. Column chromatography and flash column chromatography were performed on silica gel (particle size $35-70 \mu m$). Acetyl chloride (7a), Grignard reagents 8a-c, *tert*-butyl carbazate (10), γ -acetyl- (14a) and γ -benzoyl butyric acid (14b), 1,1'carbonyldiimidazole, bis(pentafluorophenyl) carbonate, potassium monomethyl malonate, benzylamine, 3-amino-1-propanol, piperidine, and dipolarophiles 21a-e and 22a-c are commercially available. Hex-5-en-2-one $(9a)^{19}$ and *tert*-butyl 2-(hept-6-en-2-vlidene)hydrazine-1-carboxylate $(11a)^{18}$ were prepared according to the literature procedures.

6.2. Synthesis of 7-methyl-2-oxo-2,3,3a,4,5,6-hexahydropyrazolo[1,5-*a*]pyridin-8-ium-1ide (12a) by microwave-assisted cyclization of hydrazone 11a. Compound 12a was prepared following slightly modified literature procedure.¹⁸ A 5 mL Pyrex reaction vessel was charged with hydrazone 9a (57 mg, 0.25 mmol) and trifluoromethylbenzene (2 mL) and the mixture was heated under microwave irradiation (P = 300 W) at 150 °C for 3 h. Volatile components were evaporated in vacuo, the residue was dissolved in a mixture of MeOH and CH₂Cl₂ (1:5, 10 mL),

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silica gel (500 mg), and the suspension was carefully evaporated in vacuo. The so formed silica gel with adsorbed reaction product(s) was poured into a stabilized chromatographic column (silica gel, 1.5×5 cm, EtOAc). First, the non-reacted hydrazine **11a** was eluted with EtOAc, followed by elution of the product **12a** with CH₂Cl₂-MeOH (10:1). Fractions containing the product were combined and evaporated in vacuo to give **12a**. Yield: 22 g (60%) of a beige solid; mp 125–126 °C (decomp.). ¹H NMR (500 MHz, CDCl₃): δ 1.67–1.84 (2H, m); 2.00–2.07 (1H, m); 2.30 (3H, br t, *J* = 0.8 Hz); 2.33–2.39 (1H, m); 2.57 (1H, dd, *J* = 15.6, 10.4 Hz); 2.55–2.65 (1H, m); 2.72 (1H, dd, *J* = 20.5, 6.8 Hz); 2.81 (1H, dd, *J* = 15.6, 8.4 Hz); 4.17 (1H, br q, *J* = 10.2 Hz). ¹³C NMR (126 MHz, DMSO-d₆): δ 18.3, 20.8, 27.3, 30.1, 37.5, 64.5, 148.8, 180.4. *m/z* (ESI) = 153 (MH⁺). *m/z* (HRMS) Found: 153.1021 (MH⁺). C₈H₁₃N₂O requires: *m/z* = 153.1022. v_{max} (ATR) 3382, 2936, 1674, 1584, 1373, 1337, 1082, 765, 667 cm⁻¹. Physical and spectral data of compound **12a** were in agreement with the literature data.¹⁸

6.3. A seven-step synthesis of 7-substituted-2-oxo-2,3,3a,4,5,6-hexahydropyrazolo[1,5*a*]pyridin-8-ium-1-ides 12a and 12b from γ-acyl butyric acids 14a and 14b.

6.3.1. Synthesis of 4-(2-substituted-1,3-dioxolan-2-yl)butanoic acids 16a and 16b. Compounds 16a and 16b were prepared following literature procedure for the synthesis of related compounds.²² A mixture of carboxylic acid 14 (5 mmol), anh. CH₂Cl₂ (10 mL), ethylene glycol (1.4 mL, 25 mmol), TMOF (1.6 mL, 15 mmol), and H₂SO₄ (96%, 25 μ L) was stirred at r.t. for 6 h. Then, NaHCO₃ (250 mg) was added and the mixture was stirred at r.t. for 10 min. Volatile components were evaporated in vacuo (2 mbar, 40 °C), MeOH (7 mL) and 2 M aq. NaOH (5 mL) were added, and the mixture was stirred at r.t. for 12 h. The mixture was concentrated to a half of the initial volume by evaporation in vacuo (2 mbar, 40 °C) and aqueous residue was acidified with citric acid to pH ~2. The product was extracted with EtOAc (3×20 mL), the combined organic phases were dried over anh. sodium sulfate, filtered, the filtrate was evaporated in vacuo to give **16**.

6.3.1.1. 4-(2-Methyl-1,3-dioxolan-2-yl)butanoic acid (16a). Prepared from 14a (596 µL, 5 mmol). Yield: 800 mg (91%) of pale yellow oil. ¹H NMR (500 MHz, CDCl₃): δ 1.33 (3H, s); 1.63–1.79 (4H, m); 2.39 (2H, t, *J* =7.1 Hz); 3.88–4.02 (4H, m); 10.42 (1H, br s). ¹³C NMR (126 MHz, CDCl₃): δ 19.2, 23.8, 33.8, 38.2, 64.7, 109.7, 178.7. Physical and spectral data of 16a were in agreement with the literature data.²⁰

6.3.1.2. 4-(2-Phenyl-1,3-dioxolan-2-yl)butanoic acid (16b). Prepared from 14b (960 mg, 5 mmol). Yield: 1.180 g (98%) of white solid; mp 68–71 °C. ¹H NMR (500 MHz, CDCl₃): δ 1.71 (1H, tt, J = 10.6, 6.3 Hz); 1.89–1.99 (2H, m); 2.35 (2H, t, J = 7.6 Hz); 3.73–3.81 (2H, m); 3.96–4.07 (2H, m); 7.27–7.36 (3H, m), 7.43–7.46 (2H, m). ¹³C NMR (126 MHz, CDCl₃): δ 18.9, 33.6, 39.5, 64.5, 110.1, 125.6, 127.9, 128.1, 142.3, 178.2. Physical and spectral data of 16b were in agreement with the literature data.²¹

6.4. Synthesis of methyl 3-oxo-6-(2-substituted-1,3-dioxolan-2-yl)hexanoates 17a and 17b. Under argon 1,1'-carbonyldiimidazole (815 mg, 5.2 mmol) was added to a solution of carboxylic acid 16 (5 mmol) in anh. THF (15 mL) and the mixture was stirred at r.t. for 1 h. Then, a solid well homogenized mixture of anh. MgCl₂ (395 mg, 4.8 mmol), and potassium monomethyl malonate (1.130 g, 7.5 mmol) was added, and the suspension was stirred at r.t. for 12 h. Volatile components were evaporated in vacuo, the residue was taken up in EtOAc (20 mL), and the suspension was washed with 1 M NaHSO₄ (3×20 mL) and brine (3×10 mL). The organic phase was dried over anh. sodium sulfate, filtered, and the filtrate was evaporated in vacuo to give 17.

6.4.1. Methyl 6-(2-methyl-1,3-dioxolan-2-yl)-3-oxohexanoate (17a). Prepared from 16a (800 mg, 5 mmol). Yield: 977 mg (98%) of brownish oil. ¹H NMR (500 MHz, CDCl₃): δ 1.33

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(3H, s); 1.64–1.77 (4H, m); 2.60 (2H, t, J = 7.1); 3.47 (2H, s); 3.76 (3H, s); 3.90–4.00 (4H, m). ¹³C NMR (126 MHz, CDCl₃): δ 18.0, 23.7, 38.0, 42.8, 49.0, 52.4, 64.6, 64.6, 109.8, 167.7, 202.5. m/z (ESI) = 231 (MH⁺). m/z (HRMS) Found: 231.1224 (MH⁺). C₁₁H₁₉O₅ requires: m/z =231.1227. v_{max} (ATR) 2954, 2883, 2078, 1737, 1713, 1055 cm⁻¹.

6.4.2. *Methyl* 6-(2-phenyl-1,3-dioxolan-2-yl)-3-oxohexanoate (17b). Prepared from 16b (800 mg, 5 mmol). Yield: 1.100 g (99%) of yellow oil. ¹H NMR (500 MHz, CDCl₃): δ 1.64–1.70 (2H, m); 1.88–1.91 (2H, m); 2.54 (2H, t, J = 7.4 Hz); 3.41 (2H, s); 3.71 (3H, s); 3.73–3.78 (2H, m); 3.98–4.03 (2H, m); 7.30–7.27 (1H, m); 7.31–7.35 (2H, m); 7.41–7.45 (2H, m). ¹³C NMR (126 MHz, CDCl₃): δ 17.7, 39.3, 42.8, 49.0, 52.3, 64.5, 64.5, 110.1, 125.7, 127.9, 128.2, 142.3, 167.7, 202.5. *m/z* (ESI) = 293 (MH⁺). HRMS (ESI): MH⁺, found 293.1384. C₁₆H₂₁O₅ requires 293.1384. v_{max} (ATR) 2953, 2889, 1966, 1154, 1075, 1039, 949 cm⁻¹.

6.5. Synthesis of methyl 3-hydroxy-6-(2- substituted-1,3-dioxolan-2-yl)hexanoates 18a and 18b. Finely ground NaBH₄ (188 mg, 5 mmol) was added slowly in several portions to a cold (0 °C, ice-bath) solution of β -keto ester 17 (5 mmol) in MeOH (10 mL) and the mixture was stirred at 0 °C for 1.5 h. Then, brine (5 mL) was added, ice-bath was removed, the mixture was stirred at r.t. for 5 min., and the product was extracted with CH₂Cl₂ (3×20 mL). The combined organic phases were dried over anh. sodium sulfate, filtered, and the filtrate was evaporated in vacuo to give 18.

6.5.1. Methyl 3-hydroxy-6-(2-methyl-1,3-dioxolan-2-yl)hexanoate (18a). Prepared from 17a (800 mg, 3.5 mmol) and NaBH₄ (132 mg, 3.5 mmol) in MeOH (8 mL). Yield: 698 mg (85%) of yellow oil. ¹H NMR (500 MHz, CDCl₃): δ 1.34 (3H, s); 1.43–1.53 (2H, m); 1.53–1.62 (2H, m); 1.66–1.72 (2H, m); 2.45 (1H, dd, J = 16.5, 9.1 Hz); 2.54 (1H, dd, J = 16.4, 3.1 Hz); 2.94 (1H, d, J = 3.9 Hz); 3.72 (3H, s); 3.92–4.00 (4H, m); 4.04 (1H, ddd, J = 11.3, 7.7, 3.7 Hz). ¹³C

NMR (126 MHz, CDCl₃): δ 20.0, 23.8, 36.5, 38.9, 41.1, 51.8, 64.7, 64.7, 67.9, 110.0, 173.5. m/z(ESI) = 171 (MH - H₂O - C₂H₄O⁺). m/z (HRMS) Found: 171.1006 (MH - H₂O - C₂H₄O⁺). C₉H₁₅O₃ requires: m/z = 171.1021. v_{max} (ATR) 3420, 2951, 1733, 1652, 1118, 1043 cm⁻¹.

6.5.2. *Methyl* 3-hydroxy-6-(2-phenyl-1,3-dioxolan-2-yl)hexanoate (18b). Prepared from 17b (800 mg, 2.5 mmol) and NaBH₄ (94 mg, 3.5 mmol) in MeOH (5 mL). Yield: 600 mg (85%) of pale yellow oil. ¹H NMR (500 MHz, CDCl₃): δ 1.35–1.46 (2H, m); 1.46–1.59 (2H, m); 1.87– 1.97 (2H, m); 2.38 (1H, dd, J = 16.4, 9.1 Hz); 2.47 (1H, dd, J = 16.4, 3.2 Hz); 2.87 (1H, br s); 3.69 (3H, s); 3.74–3.78 (2H, m); 3.94–3.99 (1H, m); 3.99–4.03 (2H, m); 7.26–7.31 (1H, m); 7.31–7.36 (2H, m); 7.46–7.41 (2H, m), ¹³C (126 MHz, CDCl₃) δ 19.6, 36.4, 40.2, 41.1, 51.8, 64.5, 64.5, 67.9, 110.3, 125.7, 127.8, 128.1, 142.5, 173.4. *m/z* (ESI) = 233 (MH - H₂O - C₂H₄O⁺). *m/z* (HRMS) Found: 233.1176 (MH - H₂O - C₂H₄O⁺). C₁₄H₁₇O₃ requires: *m/z* = 233.1178. v_{max} (ATR) 3467, 3050, 1731, 1171, 1102, 1072, 1026 cm⁻¹.

6.6. Synthesis of methyl 3-(methylsulfonyl)oxy-6-(2-substituted-1,3-dioxolan-2yl)hexanoates 19a and 19b. Mesyl chloride (450 μ L, 5.8 mmol) was added to a cold (0 °C, icebath) solution of the ester 18 (5 mmol) in anh. pyridine (5 mL) and the mixture was stirred at 0 °C for 3 h. The reaction mixture was diluted with toluene (30 mL) and washed with 1 M aq. NaHSO₄ until pH of aqueous phase was around 2. The organic phase was washed again with brine (2×10 mL), dried over anh. sodium sulfate, filtered, and the filtrate was evaporated in vacuo to give 19.

6.6.1. Methyl 6-(2-methyl-1,3-dioxolan-2-yl)-3-((methylsulfonyl)oxy)hexanoate (**19a**). Prepared from **18a** (600 mg, 2.5 mmol) and mesyl chloride (225 μ L, 2.9 mmol) in anh. pyridine (2.5 mL). Yield: 620 mg (80%) of orange oil. ¹H NMR (500 MHz, CDCl₃): δ 1.33 (3H, s); 1.48–1.59 (2H, m); 1.65–1.76 (2H, m); 1.76–1.91 (2H, m); 2.68 (1H, dd, *J* = 16.4, 4.8 Hz); 2.81 (1H,

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dd, J = 16.4, 7.9 Hz); 3.06 (3H, s); 3.74 (3H, s); 3.91–4.02 (4H, m); 5.04–5.09 (1H, m). ¹³C NMR (126 MHz, CDCl₃): δ 19.3, 23.9, 34.9, 38.4, 38.5, 39.2, 52.1, 64.7, 64.7, 79.0, 109.7, 170.4. m/z(ESI) = 311 (MH⁺). m/z (HRMS) Found: 311.1154 (MH⁺). C₁₂H₂₃O₇S requires: m/z = 311.1159. v_{max} (ATR) 3021, 2954, 1734, 1710, 1334, 1167, 968, 902 cm⁻¹.

6.6.2. *Methyl* 3-(*methylsulfonyl*)*oxy*-6-(2-*phenyl*-1,3-*dioxolan*-2-*yl*)*hexanoate* (19b). Prepared from 18b (1.4 g, 5 mmol). Yield: 1.55 g (95%) of pale orange oil. ¹H NMR (500 MHz, CDCl₃): δ 1.37–1.53 (2H, m); 1.69–1.85 (2H, m); 1.85–2.00 (2H, m); 2.61 (1H, dd, J = 16.4, 4.9); 2.75 (1H, dd, J = 16.4, 7.9 Hz); 2.97 (3H, s); 3.69 (3H, s); 3.73–3.81 (2H, m); 3.93–4.08 (2H, m); 4.98 (1H, dtd, J = 7.9, 6.2, 4.8 Hz); 7.28–7.31 (1H, m); 7.31–7.36 (2H, m); 7.40–7.46 (2H, m). ¹³C NMR (126 MHz, CDCl₃): δ 19.0, 34.7, 38.3, 39.2, 39.7, 52.0, 64.5, 64.5, 79.1, 110.0, 125.7, 127.9, 128.2, 142.3, 170.4. *m/z* (ESI) = 373 (MH⁺). *m/z* (HRMS) Found: 373.1312 (MH⁺). C₁₇H₂₅O₇S requires: *m/z* = 373.1316. v_{max} (ATR) 2951, 2915, 2884, 1736, 1441, 1355, 1337, 1156, 1111, 910, 887, 702 cm⁻¹.

6.7. Synthesis of 5-(3-(2-substituted-1,3-dioxolan-2-yl)propyl)pyrazolidin-3-one 20a and 20b. Hydrazine monohydrate (1.3 mL, 26 mmol) was added to a solution of ester 19 (5 mmol) in MeOH (20 mL) and the mixture was stirred at 50 °C for 3 days. Volatile components were evaporated in vacuo and the crude product was purified by column chromatography (silica gel, EtOAc–MeOH, 10:1). Fractions containing the product were combined and evaporated in vacuo to give 20.

4.7.1. 5-(3-(2-Methyl-1,3-dioxolan-2-yl)propyl)pyrazolidin-3-one (20a). Prepared from 19a (295 mg, 1 mmol) and hydrazine monohydrate (260 μ L, 5 mmol) in MeOH (5 mL). Yield: 160 mg (74%) of yellow oil. ¹H NMR (500 MHz, CDCl₃): δ 1.82 (1H, dtd, J = 13.6, 11.6, 5.8 Hz); 1.92–2.02 (1H, m); 2.02–2.10 (1H, m); 2.40 (1H, ddt, J = 13.4, 5.8, 3.7 Hz); 2.55 (1H, dd, J =

15.9, 9.3 Hz); 2.80 (1H, dd, J = 15.9, 8.6 Hz); 2.94 (1H, dddd, J = 18.9, 7.4, 4.5, 1.3 Hz); 3.07 (1H, dtd, J = 18.9, 7.8, 1.9 Hz); 4.30–4.40 (1H, m); 7.41–7.45 (3H, m); 8.02–8.07 (2H, m). ¹³C NMR (126 MHz, CDCl₃): δ 18.3, 27.4, 28.6, 36.7, 65.9, 128.2, 129.4, 130.9, 132.5, 144.4, 181.8. m/z (ESI) = 215 (MH⁺). m/z (HRMS) Found: 215.1389 (MH⁺). $C_{10}H_{19}N_2O_3$ requires: m/z = 215.1390. v_{max} (ATR) 3217, 2943, 2877, 1674, 1376, 1219, 1060, 948, 863 cm⁻¹.

6.7.2. 5-(3-(2-Phenyl-1,3-dioxolan-2-yl)propyl)pyrazolidin-3-one (20b). Prepared from 19b (7.44 g, 20 mmol) and hydrazine monohydrate (2.5 mL, 50 mmol) in MeOH (100 mL). Yield: 5.40 g (97 %) of pale yellow oil. ¹H NMR (500 MHz, CDCl₃): δ 1.24–1.36 (1H, m); 1.39–1.54 (2H, m); 1.56–1.65 (2H, m); 1.85–1.93 (1H, m); 1.92 (1H, br t, J = 7.6 Hz); 2.16 (1H, dd, J = 16.3, 8.5 Hz); 2.49 (1H, br dd, J = 16.4, 7.0 Hz); 3.61 (1H, br quintet, J = 7.1 Hz); 3.73–3.78 (2H, m); 3.97–4.02 (2H, m); 6.77 (1H, br s); 7.29 (1H, br t, J = 7.2 Hz); 7.34 (2H, t, J = 7.3 Hz); 7.43 (2H, br d, J = 7.2 Hz). ¹H NMR (500 MHz, DMSO-d₆): δ 1.21–1.47 (4H, m); 1.80 (2H, t, J = 7.7 Hz); 1.86 (1H, dd, J = 15.7, 7.6 Hz); 2.26 (1H, dd, J = 16.0, 7.0); 3.25–3.32 (1H, m); 3.71–3.60 (2H, m); 3.91–3.98 (2H, m); 5.04 (1H, s); 7.24–7.43 (5H, m); 8.88 (1H, s). ¹³C NMR (126 MHz, DMSO-d₆): δ 20.3, 32.9, 38.1, 57.6, 62.8, 64.1, 64.1, 109.6, 125.4, 127.7, 128.0, 142.4, 175.8, m/z (ESI) = 277 (MH⁺). m/z (HRMS) Found: 277.1547 (MH⁺). C₁₅H₂₁N₂O₃ requires: m/z = 277.1547. v_{max} (ATR) 3177, 2947, 2887, 1681, 1171, 1047, 1023, 939, 914, 733, 702 cm⁻¹.

6.8. Synthesis of 7-substituted-2-oxo-2,3,3a,4,5,6-hexahydropyrazolo[1,5-*a*]pyridin-8ium-1-ides 12a and 12b. TFA (3 drops) was added to a solution of pyrazolidinone 20 (1 mmol) in anh. EtOH (5 mL) and the mixture was stirred under reflux for 6 h. Volatile components were evaporated in vacuo and the crude product was purified by column chromatography (silica gel, first EtOAc–MeOH, 5:1, then CH₂Cl₂–MeOH, 9:1). Fractions containing the product were combined and evaporated in vacuo to give 12.

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6.8.1. 7-methyl-2-oxo-2,3,3a,4,5,6-hexahydropyrazolo[1,5-a]pyridin-8-ium-1-ide (12a).
Prepared from 20a (214 mg, 1 mmol). Yield: 122 mg (80%) of pale yellow solid; mp 106–110
°C. Physical and spectral data for compound 12a are given in section 6.2. These data are in agreement with the literature data.¹⁸

6.8.2. 2-oxo-7-phenyl-2,3,3a,4,5,6-hexahydropyrazolo[1,5-a]pyridin-8-ium-1-ide (12b). Prepared from 20b (230 mg, 1 mmol). Yield: 130 mg (60%) of orange solid; mp 110 °C (decomp.). ¹H NMR (500 MHz, CDCl₃): δ 1.84 (1H, dtd, J = 13.7, 11.7, 5.7 Hz); 1.91–2.03 (1H, m); 2.04–2.12 (1H, m); 2.42 (1H, ddt, J = 13.4, 5.5, 3.6 Hz); 2.57 (1H, dd, J = 15.9, 9.3 Hz); 2.84 (1H, dd, J = 15.9, 8.6 Hz); 2.95 (1H, dddd, J = 19.2, 7.6, 4.6, 1.4 Hz); 3.08 (1H, dtd, J = 19.2, 7.8, 2.0 Hz); 4.36 (1H, dddd, J = 9.3, 7.6, 3.6, 1.8 Hz); 7.41–7.46 (3H, m); 8.02–8.06 (2H, m). ¹³C NMR (126 MHz, CDCl₃): δ 18.3, 27.4, 28.6, 36.7, 65.9, 128.2, 129.4, 130.9, 132.5, 144.4, 181.8. m/z (ESI) = 215 (MH⁺). m/z (HRMS) Found: 215.0079 (MH⁺). C₁₃H₁₅N₂O requires: m/z =215.1179. (Found: C 71.23, H 6.60, N 12.83. C₁₃H₁₄N₂O·¹/₄H₂O requires: C 71.37, H 6.68, N 12.81.); v_{max} (ATR) 2930, 1648, 1572, 1559, 1324, 1299, 901, 753, 691, 668, 635 cm⁻¹. IR data are in agreement with the literature data.²³

6.9. Synthesis of 7,7-disubstituted ($3aS^*,7R^*$)-hexahydropyrazolo[1,5-*a*]pyridin-2(1*H*)ones 13'a, 13a, and 13b. Under argon, azomethine imine 12 (5 mmol) was dissolved in anh. THF (25 mL) and the solution was cooled to -20 °C (ice–salt bath). Then, Grignard reagent 8 (1 M, 25 mL, 25 mmol) was added dropwise, and the mixture was stirred at -20 °C for 1 h. The dry ice– salt bath was removed, and the reaction mixture stirred at r.t. for 12 h. Excess Grignard reagent was quenched by addition of saturated aq. NH₄Cl (20 mL) and the product was extracted with EtOAc (3×20 mL). The combined organic phases were washed with brine (30 mL), dried over anh. sodium sulfate, filtered, and the filtrate was evaporated in vacuo. The crude product was purified by column chromatography (silica gel, EtOAc-hexanes). Fractions containing the product were combined and evaporated in vacuo to give **13'a** or **13a** or **13b**.

6.9.1. (3*a*S*,7S*)-7-*methyl*-7-*phenylhexahydropyrazolo*[1,5-*a*]*pyridin*-2(1*H*)-one (13'*a*). Prepared from 12*a* (154 mg, 1 mmol) and PhMgBr (5 mL, 5 mmol) in anh. THF (5 mL), column chromatography (silica gel, EtOAc–hexanes, 1:1). Yield: 124 mg (54%) of white crystals; mp 176–178 °C. ¹H NMR (500 MHz, CDCl₃): δ 1.45 (3H, s), 1.51–1.74 (4H, m), 1.83–1.94 (1H, m), 2.19 (2H, dd, *J* = 16.0, 5.9 Hz), 2.58 (1H, dd, *J* = 16.0, 7.4 Hz), 3.41 (1H, br dq, *J* = 10.6, 6.0 Hz), 7.16–7.28 (1H, m), 7.29–7.39 (2H, m), 7.59 (2H, dd, *J* = 8.2, 1.4 Hz), 8.76 (1H, br s). ¹³C NMR (126 MHz, CDCl₃): δ 19.6, 26.9, 31.2, 31.6, 39.5, 55.5, 61.0, 126.6, 127.0, 128.4, 144.1, 174.9. *m/z* (ESI) = 231 (MH⁺). *m/z* (HRMS) Found: 231.1490 (MH⁺). C₁₄H₁₉N₂O requires: *m/z* = 231.1492. v_{max} (ATR) 3153, 3055, 2953, 2941, 2925, 2866, 1681, 1598, 763, 726, 700 cm⁻¹.

6.9.2. $(3aS^*, 7R^*)$ -7-methyl-7-phenylhexahydropyrazolo[1,5-a]pyridin-2(1H)-one (13a). Prepared from **12b** (214 mg, 1 mmol) and MeMgBr (5 mL, 5 mmol) in anh. THF (5 mL), column chromatography (silica gel, EtOAc–hexanes, 1:1). Yield: 160 mg (69%) of orange solid; mp 112–113 °C. ¹H NMR (500 MHz, CDCl₃): δ 1.51 (3H, s); 1.58 –1.76 (4H, m); 1.79–1.93 (2H, m); 2.28 (1H, dd, J = 15.5, 11.6 Hz); 2.47 (1H, dd, J = 15.5, 6.1 Hz); 3.41 (1H, br s); 5.97 (1H, br s); 7.23–7.28 (1H, m); 7.35 (2H, t, J = 7.8 Hz); 7.57 (2H, d, J = 7.5 Hz). ¹³C NMR (126 MHz, CDCl₃): δ 12.7, 21.0, 28.7, 39.1, 56.5, 58.1, 60.7, 125.8, 127.5, 128.9, 146.5, 174.8. m/z (ESI) = 231 (MH⁺). m/z (HRMS) Found: 231.1492 (MH⁺). C₁₄H₁₉N₂O requires: m/z = 231.1492. v_{max} (ATR) 3136, 2936, 2920, 2849, 1680, 1382, 1350, 1237, 1220, 1094, 1069, 757, 729, 695, 670 cm⁻¹.

6.9.3. (RS)-7,7-dimethylhexahydropyrazolo[1,5-a]pyridin-2(1H)-one (13b). Prepared from
12a (608 mg, 4 mmol) and MeMgBr (1 M in Bu₂O, 15 mL, 16 mmol) in anh. THF (20 mL), flash

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column chromatography (EtOAc). Yield: 211 mg (31%) of brownish solid; mp 143–144 °C. ¹H NMR (500 MHz, CDCl₃): δ 1.06 (3H, s); 1.17 (3H, s); 1.46–1.64 (5H, m); 1.79 (1H, brd, J = 11.3 Hz); 2.23 (1H, dd, J = 15.6, 12.6 Hz); 2.39 (1H, dd, J = 15.6, 6.5 Hz); 3.13 (1H, br q, J = 10.8 Hz); 7.62 (1H, s). ¹³C NMR (126 MHz, CDCl₃): δ 20.4, 29.0, 30.0, 37.1, 38.9, 55.6, 57.3, 77.4, 175.2. m/z (ESI) = 169 (MH⁺). m/z (HRMS) Found: 169.1336 (MH⁺). C₉H₁₆N₂O requires: m/z = 169.1335. v_{max} (ATR) 2958, 2842, 1688 (C=O), 1236, 1094, 824, 764, 718, 665 cm⁻¹.

6.10. Synthesis of 7,7-disubstituted 1-alkylhexahydropyrazolo[1,5-*a*]pyridin-2(1*H*)-ones 5'a, 5'b, 5a, and 5c. Under argon, K_2CO_3 (688 mg, 5 mmol) and MeI or BnBr (15 mmol) were added to a solution of pyrazolidinone 13a or 13'a (5 mmol) in anh. DMF (25 mL) and the mixture was stirred at r.t. for 3 days. Volatile components were evaporated in vacuo and the residue was taken up with EtOAc (30 mL). The organic phase was washed with H_2O (2×20 mL) and brine H_2O (2×20 mL), dried over anh. sodium sulfate, filtered, and the filtrate was evaporated in vacuo. The residue was purified by column chromatography (silica gel, EtOAc–hexanes). Fractions containing the product were combined and evaporated in vacuo to give 5a, 5c, or 5'a,b.

6.10.1. (3*a*S*,7S*)-1,7-*dimethyl*-7-*phenylhexahydropyrazolo*[1,5-*a*]*pyridin*-2(1*H*)-*one* (**5**'*a*). Prepared from **13**'*a* (196 mg, 0.85 mmol), K₂CO₃ (117 mg, 0.85 mmol), and MeI (157 µL, 2.55 mmol) in anh. DMF (3 mL), column chromatography (EtOAc–hexanes, 1:1). Yield: 140 mg (68%) of white crystals; mp 131–134 °C. ¹H NMR (500 MHz, CDCl₃): δ 1.52–1.61 (1H, m); 1.57 (3H, s); 1.77–1.64 (3H, m); 2.03–1.84 (2H, m); 2.16 (1H, dd, *J* = 16.2, 7.4 Hz); 2.43 (3H, s); 2.74 (1H, br dd, *J* = 16.2, 14.3 Hz); 4.10 (1H, dddd, *J* = 14.1, 7.9, 6.6, 1.9 Hz); 7.22 (1H, t, *J* = 7.3 Hz); 7.32 (2H, dd, *J* = 12.9, 5.5 Hz); 7.61 (2H, d, *J* = 7.6 Hz). ¹³C NMR (126 MHz, CDCl₃): δ 15.8, 20.2, 26.4, 32.5, 36.7, 41.8, 57.7, 63.0, 126.1, 126.6, 128.0, 149.8, 173.6. *m/z* (ESI) = 245 (MH⁺). *m/z* (HRMS) Found: 245.1646 (MH⁺). C₁₅H₂₁N₂O requires: *m/z* = 245.1648. (Found: C

73.88, H 8.35, N 11.16. C₁₅H₂₀N₂O requires: C 73.74, H 8.25, N 11.47.); ν_{max} (ATR) 2938, 2917, 2861, 1672 (C=O), 1441, 1409, 1180, 950, 691 cm⁻¹.

6.10.2. (3*a*S*,7S*)-1-benzyl-7-methyl-7-phenylhexahydropyrazolo[1,5-*a*]pyridin-2(1H)-one (5'b). Prepared from **13'a** (196 mg, 0.85 mmol), K₂CO₃ (117 mg, 0.85 mmol), and BnBr (305 µL, 2.55 mmol) in anh. DMF (3 mL), column chromatography (EtOAc–hexanes, 1:1). Yield: 150 mg (55%) of white crystals; mp 118–120 °C. ¹H NMR (500 MHz, CDCl₃): δ 1.49–1.56 (1H, m); 1.65 (3H, s); 1.65–1.72 (3H, m); 1.88–1.95 (2H, m); 2.23 (1H, dd, *J* = 16.3, 7.3 Hz); 2.84 (1H, br dd, *J* = 16.3, 14.3 Hz); 3.05 (1H, d, *J* = 15.4); 4.01 (1H, ddt, *J* = 15.0, 7.7, 4.0 Hz); 4.89 (1H, d, *J* = 15.4); 6.76–6.83 (2H, m); 7.21–7.15 (3H, m); 7.21–7.27 (3H, m) 7.44 (2H, t, *J* = 16.1 Hz) ¹³C NMR (126 MHz, CDCl₃): δ 15.5, 20.3, 26.4, 36.6, 42.2, 46.8, 58.6, 63.1, 126.6, 127.0, 127.7, 127.7, 128.0, 128.1, 137.1, 149.6, 174.8. *m/z* (ESI) = 321 (MH⁺). *m/z* (HRMS) Found: 321.1960 (MH⁺). C₂₁H₂₅N₂O requires: *m/z* = 321.1961. (Found: C 78.44, H 7.73, N 8.60. C₂₁H₂₄N₂O requires: C 78.71, H 7.55, N 8.74.); v_{max} (ATR) 2932, 2895, 1669 (C=O), 1601, 1494, 1432, 1229, 748, 696, 615 cm⁻¹.

6.10.3. (3*a*S*,7*R**)-1,7-*dimethyl*-7-*phenylhexahydropyrazolo*[1,5-*a*]*pyridin*-2(1*H*)-*one* (5*a*). Prepared from **13a** (176 mg, 0.33 mmol), K₂CO₃ (45 mg, 0.33 mmol), and MeI (68 µL, 0.99 mmol) in anh. DMF (1.5 mL), column chromatography (EtOAc–hexanes, 1:1). Yield: 50 mg (66%) of orange oil. ¹H NMR (500 MHz, CDCl₃): δ 1.56 (3H, s); 1.57–1.65 (3H, m); 1.65–1.72 (1H, m); 1.84–1.97 (2H, m); 2.22 (1H, t, *J* = 14.5 Hz); 2.44 (1H, dd, *J* = 15.2, 6.2 Hz); 2.72 (3H, s); 3.22 (1H, dddd, *J* = 13.5, 10.7, 6.2, 2.3 Hz); 7.21–7.26 (1H, m); 7.29–7.37 (2H, m); 7.68 (2H, d, *J* = 8.5 Hz).¹³C NMR (126 MHz, CDCl₃): δ 13.0, 21.5, 29.3, 33.1, 37.7, 43.9, 57.0, 61.6, 126.1, 127.1, 128.1, 149.2, 172.1. *m/z* (ESI) = 245 (MH⁺). *m/z* (HRMS) Found: 245.1645 (MH⁺). C₁₅H₂₁N₂O requires: *m/z* = 245.1648. v_{max} (ATR) 2944, 2909, 1688, 1373, 1221, 1117, 1035, 1029, 772, 747, 701 cm^{-1} .

6.10.4. (*RS*)-1-benzyl-7,7-dimethylhexahydropyrazolo[1,5-a]pyridin-2(1H)-one (5c). Prepared from **13b** (90 mg, 0.53 mmol), K₂CO₃ (138 mg, 1 mmol), and BnBr (190 µL, 1.59 mmol) in anh. DMF (2 mL), column chromatography (EtOAc–hexanes, 1:2). Yield: 58 mg (42%) of orange oil. ¹H NMR (500 MHz, DMSO-d₆): δ 1.15 (3H, s); 1.16 (3H, s); 1.42–1.49 (2H, m); 1.50–1.59 (2H, m); 1.67–1.81 (2H, m); 2.12 (1H, dd, *J* = 15.9, 6.8 Hz); 2.63 (1H, t, *J* = 15.2 Hz); 3.52–3.62 (1H, m); 4.33 (1H, d, *J* = 15.4 Hz); 5.12 (1H, d, *J* = 15.4 Hz); 7.23–7.34 (5H, m). ¹³C NMR (126 MHz, DMSO-d₆): δ 16.1, 26.9, 36.7, 36.7, 38.9, 50.8, 50.8, 59.3, 127.4, 128.2, 128.5, 137.4, 176.7. *m/z* (ESI) = 259 (M⁺). *m/z* (HRMS) Found: 259.1801 (MH⁺). C₁₆H₂₂N₂O requires: *m/z* = 259.1805. v_{max} (ATR) 2935, 1686 (C=O), 1455, 1385, 1250, 1084, 774, 700 cm⁻¹.

6.11. Thermal [3+2]-cycloadditions of azomethine imines 12a and 12b. Synthesis of cycloadducts 23–26.

A mixture of azomethine imine **12** (0.25 mmol), anh. CH_2Cl_2 or toluene (1 mL), and dipolarophile **21** or **22** (0.3 mmol, 1.2 equiv.) was stirred at r.t. or at 80 °C for 24–96 h.

Workup A. The precipitate was collected by filtration to give **23b**.

Workup B. Volatile components were evaporated in vacuo and the residue was purified by column chromatography (silica gel, EtOAc–hexanes or CH₂Cl₂–MeOH). Fractions containing the product were combined and evaporated in vacuo to give cycloadduct **23–26**.

6.11.1. Dimethyl (4aS*, 7aS*)-4a-methyl-2-oxo-1,4a,5,6,7,7a-hexahydro-2H-2a,2a1diazacyclopenta[cd]indene-3,4-dicarboxylate (23a). Prepared from 12a (38 mg, 0.25 mmol) and DMAD (21a) (36 μL, 0.30 mmol) in anh. CH₂Cl₂ (1 mL), r.t., 72 h, Workup B, column chromatography (CH₂Cl₂–MeOH, 50:1). Yield: 44 mg (60%) of orange oil. ¹H NMR (500 MHz, CDCl₃): δ 1.40 (3H, s); 1.66–1.74 (1H, m); 1.76–1.91 (3H, m); 1.94 (1H, dt, J = 14.1, 3.4 Hz); 2.06 (1H, td, J = 13.8, 3.8 Hz); 2.46 (1H, dd, J = 15.2, 4.9 Hz); 2.55 (1H, dd, J = 15.2, 13.0 Hz); 3.50 (1H, ddt, J = 13.0, 11.7, 4.7 Hz); 3.72 (3H, s); 3.94 (3H, s). ¹³C NMR (126 MHz, CDCl₃); δ 16.4, 23.2, 25.5, 28.6, 40.7, 51.8, 53.3, 60.3, 66.7, 115.9, 139.8, 161.3, 163.5, 176.5. m/z (ESI) = 295 (MH⁺). m/z (HRMS) Found: 295.1288 (MH⁺). C₁₄H₁₉N₂O₅ requires: m/z = 295.1288. v_{max} (ATR) 2952, 1731, 1706, 1607, 1435, 1368, 1150, 842 cm⁻¹.

6.11.2. Dimethyl (4aR,7aS)-2-oxo-4a-phenyl-1,4a,5,6,7,7a-hexahydro-2H-2a,2a1-diazacyclopenta[cd]indene-3,4-dicarboxylate (23b). Prepared from 12b (43 mg, 0.2 mmol) and DMAD (21a) (30 µL, 0.24 mmol) in anh. toluene (2 mL), r.t., 24 h, Workup A. Yield: 42 mg (59%) of brownish oil. ¹H NMR (500 MHz, CDCl₃): δ 1.46–1.56 (1H, m); 1.65–1.76 (2H, m); 1.82–1.92 (1H, m); 2.40 (1H, td, J = 14.4, 3.5 Hz); 2.58 (2H, dd, J = 15.2, 4.7 Hz); 2.62 (1H, dt, J= 14.5, 3.6 Hz); 2.70 (1H, dd, J = 15.2, 13.0 Hz); 3.71 (1H, dddd, J = 13.0, 9.8, 4.9, 3.6 Hz); 3.71 (3H, s); 3.88 (3H, s); 7.24–7.29 (1H,m); 7.35 (2H, br t, J = 7.7 Hz); 7.73 (2H, br d, J = 7.1 Hz). ¹³C NMR (126 MHz, CDCl₃): δ 16.4, 23.2, 29.3, 40.6, 51.9, 53.2, 59.9, 72.0, 115.9, 126.2, 127.4, 128.4, 138.6, 143.9, 161.1, 164.0, 176.2. m/z (ESI) = 357 (MH⁺). m/z (HRMS) Found: 357.1446 (MH⁺). C₁₉H₂₀N₂O₅ requires: m/z = 357.1445. v_{max} (ATR) 1770, 1742, 1670, 1609, 1437, 1303, 1223, 1146, 754, 704 cm⁻¹.

6.11.3. $(4aS^*, 7aS^*)$ -3-Acetyl-4a-methyl-1,4a,5,6,7,7a-hexahydro-2H-2a,2a1-diazacyclopenta-[cd]inden-2-one (24a). Prepared from 12a (38 mg, 0.25 mmol) and 3-butyn-2-one (21b) (23.5 μ L, 0.3 mmol) in anh. DCM (1 mL), r.t., 24 h, Workup B, column chromatography (CH₂Cl₂– MeOH, 100:1). Yield: 40 mg (73%) of beige solid; mp 83–87 °C. ¹H NMR (500 MHz, CDCl₃): δ 1.36 (3H, s); 1.64–1.98 (5H, m); 2.05–2.08 (1H, m); 2.27 (3H, s); 2.44 (1H, dd, J = 15.1, 4.7 Hz); 2.57 (1H, dd, J = 15.1, 13.2 Hz); 3.40 (1H, ddt, J = 13.0, 11.9, 4.5 Hz); 7.30 (1H, s); minor isomer 24'a 2.82 (1H, dd, J = 17.1, 7.5 Hz). ¹³C NMR (126 MHz, CDCl₃): δ 16.8, 23.1, 25.8,

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27.5, 28.2, 40.9, 60.9, 66.6, 128.2, 136.2, 177.2, 193.5. m/z (ESI) = 221 (MH⁺). m/z (HRMS) Found: 221.1284 (MH⁺). $C_{12}H_{17}N_2O_2$ requires: m/z = 221.1285. v_{max} (ATR) 3068, 2965, 1728, 1648, 1575, 1233, 1186, 660, 612 cm⁻¹.

6.11.4. (4aR*,7aS*)-3-Acetyl-4a-phenyl-1,4a,5,6,7,7a-hexahydro-2H-2a,2a1-diazacyclopenta[cd]inden-2-one (**24b**). Prepared from **12b** (53 mg, 0.25 mmol) and 3-butyn-2-one (**21b**) (23.5 μL, 0.3 mmol) in anh. DCM (1 mL), r.t., 96 h, Workup B, column chromatography (EtOAc-hexanes, 1:3). Yield: 49 mg (69%) of pale yellow crystals; mp 196–200 °C. ¹H NMR (500 MHz, CDCl₃): δ 1.44–1.54 (1H, m); 1.62–1.73 (2H, m); 1.78–1.86 (1H, m); 2.23 (3H, s); 2.29 (1H, td, *J* = 14.5, 3.7 Hz); 2.56 (1H, dd, *J* = 15.1, 4.6 Hz); 2.69 (1H, dt, *J* = 14.4, 3.5 Hz); 2.72 (1H, dd, *J* = 15.1, 13.1 Hz); 3.59 (1H, ddt, *J* = 13.1, 11.6, 4.8 Hz); 7.22–7.25 (1H, m); 7.28 (1H, s); 7.31–7.34 (2H, m); 7.79–7.81 (2H, m); minor isomer **24'b** 3.26 (1H, dd, *J* = 16.5, 7.5 Hz). ¹³C NMR (126 MHz, CDCl₃): δ 16.7, 23.1, 27.9, 29.4, 40.8, 60.3, 71.8, 126.6, 127.1, 127.6, 128.3, 136.2, 145.2, 176.9, 194.1. *m/z* (ESI) = 283 (MH⁺). *m/z* (HRMS) Found: 283.1441 (MH⁺). $C_{17}H_{19}N_2O_2$ requires: *m/z* = 283.1441. v_{max} (ATR) 3078, 2959, 1742, 1648, 1569, 1296, 1222, 1102, 760, 706 cm⁻¹.

6.11.5. (4aS*,7aS*)-4-Benzoyl-4a-methyl-1,4a,5,6,7,7a-hexahydro-2H-2a,2a1-

diazacyclopenta[cd]inden-2-one (24c). Prepared from **12a** (38 mg, 0.25 mmol) and 1phenylprop-2-yn-1-one (**21c**) (39 mg, 0.3 mmol) in anh. DCM (1 mL), r.t., 42 h, Workup B, column chromatography (EtOAc–hexanes, 1:4). Yield: 47 mg (67%) of brownish solid; mp 157– 161 °C. ¹H NMR (500 MHz, CDCl₃): δ 1.50 (3H, s); 1.79–1.87 (2H, m); 1.90–1.99 (1H, m); 2.12–2.20 (2H, m); 2.46 (1H, dd, J = 15.2, 4.7 Hz); 2.60 (1H, dd, J = 15.1, 13.2 Hz); 3.48 (1H, ddt, J = 13.2, 12.1, 4.5 Hz); 7.17 (1H, s); 7.41–7.45 (2H, m); 7.50–7.54 (1H, m) 7.61–7.64 (2H, m); *minor isomer* **24'c** 3.26 (1H, dd, J = 16.6, 7.5 Hz). ¹³C NMR (126 MHz, CDCl₃): δ 16.9, 23.2, 26.1, 28.5, 40.9, 61.1, 67.6, 126.6, 128.1, 128.6, 131.8, 138.1, 140.2, 177.0, 191.9. m/z(ESI) = 283 (MH⁺). m/z (HRMS) Found: 283.1445 (MH⁺). $C_{17}H_{19}N_2O_2$ requires: m/z = 283.1441. v_{max} (ATR) 3077, 2957, 1745, 1619, 1571, 1295, 1237, 1174, 732 cm⁻¹.

6.11.6. (4aR*,7aS*)-4-Benzoyl-4a-phenyl-1,4a,5,6,7,7a-hexahydro-2H-2a,2a1-

diazacyclopenta[cd]inden-2-one (24d). Prepared from 12b (53 mg, 0.25 mmol) and 1phenylprop-2-yn-1-one (21c) (39 mg, 0.3 mmol) in anh. DCM (1 mL), r.t., 40 h, Workup B, column chromatography (EtOAc-hexanes, 1:4). Yield: 61 mg (71%) of orange solid; mp 171– 174 °C. ¹H NMR (500 MHz, CDCl₃): δ 1.52–1.63 (1H, m); 1.68–1.79 (2H, m); 1.83–1.91 (1H, m); 2.55 (1H, dd, J = 15.1, 4.6 Hz); 2.60 (1H, dd, J = 14.4, 3.7 Hz); 2.73 (1H, dt, J = 14.4, 3.5 Hz); 2.74 (1H, dd, J = 15.1, 13.0 Hz); 3.65 (1H, ddt, J = 13.0, 11.7, 4.8 Hz); 7.09 (1H, s); 7.23– 7.27 (1H, m); 7.32–7.39 (2H, m); 7.46–7.50 (1H, m); 7.51–7.54 (2H, m); 7.79–7.82 (2H, m). ¹³C NMR (126 MHz, CDCl₃): δ 16.8, 23.3, 30.1, 40.8, 60.6, 73.3, 126.2, 126.4, 127.3, 128.3, 128.51, 128.52, 132.0, 137.6, 140.1, 145.5, 176.6, 193.1. m/z (ESI) = 345 (MH⁺). m/z (HRMS) Found: 345.1599 (MH⁺). C₂₂H₂₁N₂O₂ requires: m/z = 345.1598. v_{max} (ATR) 3059, 2922, 1741, 1612, 1564, 1554, 1294, 1231, 721 cm⁻¹.

6.11.7. tert-butyl (4aS*,7aS*)-(2-(4a-methyl-2-oxo-1,4a,5,6,7,7a-hexahydro-2H-2a,2a1diazacyclopenta[cd]inden-3-yl)-2-oxoethyl)carbamate (24e). Prepared from 12a (38 mg, 0.25 mmol) and tert-butyl (2-oxobut-3-yn-1-yl)carbamate (21d) (46 mg, 0.3 mmol) in anh. DCM (1 mL), r.t., 48 h, Workup B, column chromatography (CH₂Cl₂–MeOH, 100:1). Yield: 35 mg (42%) of yellow resin. ¹H NMR (500 MHz, CDCl₃): δ 1.36 (3H, s); 1.45 (9H, s); 1.65–2.08 (6H, m); 2.46 (1H, dd, J = 15.1, 4.7 Hz); 2.57 (1H, dd, J = 15.2, 13.1 Hz); 3.38 (1H, tq, J = 12.6, 4.5 Hz); 4.15–4.25 (2H, m); 5.29 (1H, br s); 7.40 (1H, s); minor isomer 24'e 2.81 (1H, dd, J = 17.2, 7.4 Hz). ¹³C NMR (126 MHz, CDCl₃): δ 16.8, 23.1, 25.8, 28.3, 28.5, 40.8, 47.4, 60.8, 66.8, 79.9,

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124.9, 135.9, 155.8, 176.7, 190.5. m/z (ESI) = 336 (MH⁺). m/z (HRMS) Found: 336.1923 (MH⁺). C₁₇H₂₆N₃O₄ requires: m/z = 336.1918. v_{max} (ATR) 3367, 2968, 1751, 1709, 1653, 1577, 1163, 937, 861, 730 cm⁻¹.

6.11.8. tert-butyl (4aR*, 7aS*)-(2-oxo-2-(2-oxo-4a-phenyl-1,4a,5,6,7,7a-hexahydro-2H-2a,2a1-diazacyclopenta[cd]inden-3-yl)ethyl)carbamate (24f). Prepared from 12b (54 mg, 0.25 mmol) and tert-butyl (2-oxobut-3-yn-1-yl)carbamate (21d) (46 mg, 0.3 mmol) in anh. DCM (1 mL), r.t., 48 h, Workup B, column chromatography (EtOAc-hexanes, 1:3). Yield: 50 mg (50%) of brownish resin. ¹H NMR (500 MHz, CDCl₃): δ 1.43 (9H, s); 1.46–1.59 (1H, m); 1.62–1.73 (2H, m); 1.75–1.90 (1H, m); 2.30 (1H, td, J = 14.3, 3.6 Hz); 2.57 (1H, dd, 15.1, 4.8 Hz); 2.70 (1H, dt, J = 14.3, 3.5 Hz); 2.72 (1H, dd, J = 15.2, 12.6 Hz); 3.58 (1H, ddt, J = 12.7, 11.3, 5.0 Hz); 3.98 (1H, dd, J = 18.3, 4.3 Hz); 4.30 (1H, dd, J = 18.2, 5.7 Hz); 5.27 (1H, br s); 7.21–7.26 (1H, m); 7.29–7.35 (2H, m); 7.38 (1H, s); 7.73–7.77 (2H, m); minor isomer 24'f 3.21 (1H, dd, J = 17.2, 7.6 Hz). ¹³C NMR (126 MHz, CDCl₃): δ 16.7, 23.2, 28.4, 29.3, 40.8, 47.5, 60.2, 72.1 79.9, 124.5, 126.5, 127.3, 128.4, 135.8, 144.8, 155.7, 176.4, 191.1. m/z (ESI) = 398 (MH⁺). m/z (HRMS) Found: 398.2073 (MH⁺). C₂₂H₂₈N₃O₄ requires: $m/z = 398.2074_{\perp} v_{max}$ (ATR) 3406, 2938, 1759, 1699, 1676, 1585, 1574, 1522, 1153, 702 cm⁻¹.

6.11.9. Methyl (4aS*,7aS*)-4a-methyl-2-oxo-1,4a,5,6,7,7a-hexahydro-2H-2a,2a1-diazacyclopenta[cd]-indene-3-carboxylate (24g). Prepared from 12a (38 mg, 0.25 mmol) and methyl propiolate (21e) (27 μL, 0.3 mmol) in anh. DCM (1 mL), 80 °C (pressure vessel), 24 h, Workup B, column chromatography (CH₂Cl₂–MeOH, 100:1). Yield: 33 mg (56%) of colourless oil. ¹H NMR (500 MHz, CDCl₃): δ 1.39 (3H, s); 1.64–2.02 (6H, m); 2.42 (1H, dd, J = 15.1, 4.7 Hz); 2.56 (1H, dd, J = 15.0, 13.2 Hz); 3.41 (1H, ddt, J = 13.1, 12.1, 4.5 Hz); 3.73 (3H, s); 7.31 (1H, s); minor isomer 24'g 3.16 (1H, dd, J = 16.5, 7.5 Hz). ¹³C NMR (126 MHz, CDCl₃) δ 16.8, 23.2,

26.1, 28.4, 40.9, 51.5, 60.8, 65.9, 118.7, 135.5, 164.6, 176.9. m/z (ESI) = 237 (MH⁺). m/z (HRMS) Found: 237.1234 (MH⁺). $C_{12}H_{17}N_2O_3$ requires: m/z = 237.1234. v_{max} (ATR) 2952, 1750, 1698, 1595, 1224, 1169, 1076, 765 cm⁻¹.

6.11.10. Methyl $(4aR^*, 7aS^*)$ -2-oxo-4a-phenyl-1,4a,5,6,7,7a-hexahydro-2H-2a,2a1-diazacyclopenta[cd]-indene-3-carboxylate (24h) and its (4aS^*, 7aS^*)-epimer 24'h. Prepared from 12b (53 mg, 0.25 mmol) and methyl propiolate (21e) (27 µL, 0.3 mmol) in anh. DCM (1 mL), 80 °C (pressure vessel), 24 h, Workup B, column chromatography (EtOAc–hexanes, 1:4).

6.11.10.1. Methyl (4aR*,7aS*)-2-oxo-4a-phenyl-1,4a,5,6,7,7a-hexahydro-2H-2a,2a1diaza-cyclopenta[cd]-indene-3-carboxylate (24h). Yield: 45 mg (60%) of brownish solid; mp 140–144 °C. ¹H NMR (500 MHz, CDCl₃): δ 1.47–1.57 (1H, m); 1.65–1.74 (2H, m); 1.79–1.87 (1H, m); 2.31 (1H, dt. *J* = 14.4, 3.6 Hz); 2.53 (1H, dd, *J* = 15.0, 4.6 Hz); 2.62 (1H, dt, *J* = 14.3, 3.5 Hz); 2.71 (1H, dd, *J* = 15.0, 13.1 Hz); 3.60 (1H, ddt, *J* = 13.2, 11.7, 4.8 Hz); 3.72 (3H, s); 7.23–7.27 (1H, m); 7.29 (1H, s); 7.32–7.36 (2H, m); 7.80–7.82 (2H, m). ¹³C NMR (126 MHz, CDCl₃): δ 16.7, 23.2, 29.3, 40.9, 51.5, 60.3, 71.1, 118.2, 126.4, 127.2, 128.4, 135.2, 145.1, 165.1, 176.6. *m/z* (ESI) = 299 (MH⁺). *m/z* (HRMS) Found: 299.1393 (MH⁺). C₁₇H₁₉N₂O₃ requires: *m/z* = 299.1390. v_{max} (ATR) 2932, 1739, 1691, 1604, 1590, 1107, 748, 705 cm⁻¹.

6.11.10.2. Methyl (4aS*,7aS*)-2-oxo-4a-phenyl-1,4a,5,6,7,7a-hexahydro-2H-2a,2a1-diazacyclopenta[cd]-indene-3-carboxylate (24'h). Yield: 8 mg (11%) of brownish resin. ¹H NMR (500 MHz, CDCl₃): δ 1.39–1.49 (1H, m); 1.51–1.62 (3H, m); 1.75–1.86 (2H, m); 2.46 (1H, d, J = 16.5 Hz); 2.81–2.88 (1H, m), 3.21 (1H, dd, J = 16.5, 7.5 Hz); 3.58 (3H, s); 7.28 (1H, br t, J = 7.3 Hz); 7.37 (2H, br t, J = 7.7 Hz); 7.41 (1H, s); 7.78 (2H, br d, J = 7.5 Hz). ¹³C NMR (126 MHz, CDCl₃): δ 19.3, 27.4, 31.6, 42.9, 51.5, 53.4, 68.8, 125.0, 126.5, 127.5, 128.1, 128.3, 140.8, 163.2, 164.0. m/z (ESI) = 299 (MH⁺). m/z (HRMS) Found: 299.1392 (MH⁺). $C_{17}H_{19}N_2O_3$ requires: m/z =

299.1390. v_{max} (ATR) 2932, 1690, 1577, 1410, 1310, 1194, 1090, 756, 696 cm⁻¹.

6.11.11. A mixture of methyl ($4S^*$, $4aS^*$, $7aS^*$)-4a-methyl-2-oxo-octahydro-2H-2a, 2aldiazacyclopenta-[cd]indene-4-carboxylate (25a) and its isomers 25'a. Prepared from 12a (152 mg, 1 mmol) and methyl acrylate (22a) (450μ l, 5 mmol) in anh. CH₂Cl₂ (5 mL), 80 °C (pressure vessel), 72 h, Workup B, flash column chromatography (EtOAc). Yield: 167 mg (70%) of brownish oil. The isomeric products 25a and 25'a were separated by column chromatography (EtOAc–hexanes, 1:3 to elute the non-polar by-products, then EtOAc–hexanes, 1:1 to elute 25'a, finally EtOAc to elute 25a). Fractions containing the products were combined and evaporated in vacuo to give 25a and 25'a.

6.11.11.1 Methyl (3*R**,4*a*S*,7*a*S*)-4*a*-methyl-2-oxooctahydro-2*H*-2*a*,2*a*1-diazacyclopenta-[*cd*]*indene-3-carboxylate* (**25'a**). Yield: 38 mg (16%) of yellowish oil. ¹H NMR (500 MHz, CDCl₃): δ 1.12 (3H, s); 1.63–1.76 (6H, m); 2.20 (1H, dd, *J* = 12.9, 9.6 Hz); 2.31 (1H, dd, *J* = 14.3, 4.8 Hz); 2.55 (1H, d, *J* = 12.9 Hz); 2.60 (1H, t, *J* = 14.2 Hz); 3.07–3.16 (1H, m); 3.77 (3H, s); 4.80 (1H, d, *J* = 9.5 Hz). ¹³C NMR (126 MHz, CDCl₃): δ 17.3, 24.2, 26.1, 32.4, 40.1, 41.9, 53.1, 59.3, 61.2, 63.2, 171.8, 176.4. *m/z* (ESI) = 239 (MH⁺). *m/z* (HRMS) Found: 239.1394 (MH⁺). C₁₂H₁₉N₂O₃ requires: *m/z* = 239.1390. v_{max} (ATR) 2949, 1728, 1703, 1436, 1349, 1197, 1112, 1016, 638 cm⁻¹.

6.11.11.2 Methyl (4S*,4aS*,7aS*)-4a-methyl-2-oxo-octahydro-2H-2a,2a1-diazacyclopenta-[cd]indene-4-carboxylate (25a). Yield: 26 mg (11%) of yellowish oil. ¹H NMR (500 MHz, CDCl₃): δ 1.34 (3H, s); 1.57–1.66 (2H, m); 1.68–1.92 (4H, m); 2.31 (1H, br dd, J = 13.9, 5.7 Hz); 2.46 (1H, t, J = 13.9 Hz); 3.00 (1H, d, J = 6.7 Hz); 3.40 (1H, br dd, J = 12.1, 6.7 Hz); 3.58 (1H, br s); 3.72 (3H, s); 4.42 (1H, br d, J = 12.1 Hz). ¹³C NMR (126 MHz, CDCl₃): δ 17.5, 24.1, 28.4, 29.1, 39.9, 46.6, 52.1, 55.1, 60.9, 64.5, 173.4, 173.6. m/z (ESI) = 239 (MH⁺). m/z (HRMS)

Found: 239.1392 (MH⁺). $C_{12}H_{19}N_2O_3$ requires: m/z = 239.1390. v_{max} (ATR) 2951, 1728, 1697, 1371, 1194, 1176, 1162, 1111, 635 cm⁻¹.

6.11.12. Methyl (4S*,4aS*,7aS*)-2-oxo-4a-phenyloctahydro-2H-2a,2a1-diazacyclopenta[cd]indene-4-carboxylate (25b). Prepared from 12b (50 mg, 0.25 mmol) and methyl acrylate (22a) (25 μl, 0.4 mmol) in anh. CH₂Cl₂ (5 mL), 80 °C, 12 h, Workup B, column chromatography (EtOAc). Yield: 33 mg (44%) of brownish oil. ¹H NMR (500 MHz, CDCl₃): δ 1.34–1.45 (1H, m); 1.55–1.80 (3H, m); 2.02–2.08 (2H, m); 2.42 (1H, dd, J =14.1, 5.3 Hz); 2.62 (1H, td, J = 14.0, 1.1 Hz); 2.90 (1H, ddd, J = 12.0, 6.4, 1.1 Hz); 3.37 (1H, d, J = 6.4 Hz); 3.75–3.86 (1H, m); 3.80 (3H, s); 4.32 (1H, d, J = 11.9 Hz); 7.27 (1H, d, J = 7.3 Hz); 7.37 (2H, br t, J = 7.7 Hz); 7.73 (2H, br d, J = 7.2 Hz). ¹³C NMR (126 MHz, CDCl₃): δ 17.1, 24.4, 30.9, 40.2, 45.6, 52.2, 56.9, 60.1, 70.4, 126.7, 127.4, 128.4, 146.0, 173.0, 173.4. m/z (ESI) = 301 (MH⁺). m/z (HRMS) Found: 301.1543 (MH⁺). C₁₇H₂₁N₂O₃ requires: m/z = 301.1547. v_{max} (ATR) 2949, 1730, 1701, 1491, 1361, 1197, 1177, 1058, 706 cm⁻¹.

6.11.13. tert-Butyl (4S*,4aS*,7aS*)-2-oxo-4a-phenyloctahydro-2H-2a,2a1-diazacyclopenta-[cd]indene-4-carboxylate (25c). Prepared from 12a (214 mg, 1 mmol) and tert-butyl acrylate (22b) (1.5 mL, 10 mmol) in anh. toluene (80 mL), 80 °C, 24 h, Workup B, column chromatography (EtOAc). Yield: 265 mg (77%) of yellow oil. ¹H NMR (500 MHz, CDCl₃): δ 1.19–1.25 (1H, m); 1.48 (9H, s); 1.49–1.61 (2H, m), 1.69–1.77 (1H, m); 1.95 (1H, dt, *J* = 13.9, 3.4 Hz); 2.16 (1H, td, *J* = 14.0, 3.2 Hz); 2.29 (1H, dd, *J* = 14.2, 5.3 Hz); 2.53 (1H, m); 2.66 (1H, dd, *J* = 12.0, 6.5 Hz); 3.20 (1H, d, *J* = 6.3 Hz); 3.71 (1H, ddt, *J* = 13.8, 11.7, 5.1 Hz); 4.08 (1H, d, *J* = 12.0 Hz); 7.25–7.30 (1H, m); 7.39 (2H, br t, *J* = 7.7 Hz); 7.60 (2H, dt, *J* = 8.3, 1.7 Hz). ¹³C NMR (126 MHz, CDCl₃): δ 16.6, 23.9, 27.6, 30.0, 39.7, 45.7, 57.5, 59.3, 69.7, 81.4, 126.2, 127.0, 128.1, 146.2, 171.3, 172.3. *m/z* (ESI) = 343 (MH⁺). *m/z* (HRMS) Found: 343.2007 (MH⁺).

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 $C_{20}H_{26}N_2O_3$ requires: m/z = 343.2016. v_{max} (ATR) 1650, 1573, 1557, 1444, 1300, 1087, 774, 692 cm⁻¹.

6.11.14. $(2aS^*, 5aS^*, 5bS^*, 8aS^*)$ -5*a*-methyl-7-phenyloctahydro-2*a*1, 7,8*b*-triazadicyclopenta-[*a*,*cd*]*indene*-1,6,8(7*H*)-trione (**26a**) Prepared from **12a** (78 mg, 0.5 mmol) and *N*phenylmaleimide (**22c**) (104 mg, 0.6 mmol) in toluene (2 mL), 80 °C, 18 h, Workup B, column chromatography (CH₂Cl₂–MeOH, 50:1). Yield: 21 mg (13%) of white solid; mp 169–172 °C. ¹H NMR (500 MHz, CDCl₃): δ *major isomer* **26a** 1.30 (3H, s); 1.72 (2H, m); 1.81–1.97 (3H, m); 2.09 (1H, m); 2.41 (1H, dd, *J* = 14.5, 4.5 Hz); 2.66 (1H, t, *J* = 14.0 Hz); 3.14 (1H, m); 3.56 (1H, d, *J* = 9.0 Hz); 5.17 (1H, d, *J* = 9.0 Hz); 7.28 (2H, m); 7.41 (1H, m); 7.48 (2H, m); *minor isomer* **26'a** 5.00 (1H, d, *J* = 7.1 Hz). ¹³C NMR (126 MHz, CDCl₃): δ 18.1, 22.8, 24.8, 33.0, 40.4, 53.8, 63.8, 64.8, 65.3, 126.1, 128.9, 129.3, 131.2, 172.4, 172.6, 181.3. *m/z* (ESI) = 326 (MH⁺). *m/z* (HRMS) Found: 326.1503 (MH⁺). C₁₈H₂₀N₃O₃ requires: *m/z* = 326.1499. v_{max} (ATR) 1714, 1499, 1391, 1279, 1181, 1114, 733, 691, 660 cm⁻¹.

6.11.15. (2aS*,5aS*,5bS*,8aS*)-5a,7-diphenyloctahydro-2a1,7,8b-triazadicyclopenta[a,cd]indene-1,6,8(7H)-trione (**26b**) Prepared from **12b** (108 mg, 0.5 mmol) and *N*-phenylmaleimide (**22c**) (104 mg, 0.6 mmol) in toluene (2 mL), 80 °C, 18 h, Workup B, column chromatography (CH₂Cl₂–MeOH, first 50:1, then 25:1). Yield: 77 mg (40%) of white solid; mp 195–198 °C. ¹H NMR (500 MHz, CDCl₃): δ major isomer **26b** 1.44 (1H, m); 1.70 (2H, m); 1.86 (1H, m); 2.36 (1H, td, *J* = 14.0, 3.5 Hz); 2.50 (1H, dd, *J* = 14.5, 4.5 Hz); 2.63 (1H, dt, *J* = 14.0, 4.0 Hz); 2.76 (1H, t, *J* = 14.0 Hz); 3.51 (1H, m); 3.79 (1H, d, *J* = 8.5 Hz); 5.22 (1H, d, *J* = 8.5 Hz), 6.35 (2H, m), 7.19–7.40 (7H, m); 7.88 (1H, br s); minor isomer **26'b** 3.89 (1H, d, *J* = 6.1 Hz). ¹³C NMR (126 MHz, CDCl₃) δ 17.7, 24.5, 33.8, 40.1, 56.9, 62.6, 63.9, 71.4, 126.2, 128.2, 128.4 (br), 128.6, 128.7, 130.7, 139.6, 171.67, 171.69, 179.6. *m/z* (ESI) = 388 (MH⁺). *m/z* (HRMS) Found:

388.1655 (MH⁺). C₂₃H₂₂N₃O₃ requires: m/z = 388.1656. v_{max} (ATR) 1716, 1498, 1391, 1283, 1196, 755, 705, 689 cm⁻¹.

6.12. CuI-catalyzed [3+2]-cycloadditions of azomethine imines 12a and 12b to methyl propiolate. Synthesis of cycloadducts 24g and 24h.

A mixture of azomethine imine **12** (0.25 mmol), anh. CH_2Cl_2 (1 mL), methyl propiolate (**21e**) (27 μ L, 0.3 mmol), CuI (10 mg, 0.05 mmol), and DIPEA (9 μ L, 0.05 mmol) was stirred at r.t. for 72 h. Volatile components were evaporated in vacuo and the residue was purified by column chromatography (silica gel, EtOAc–hexanes, 1:4). Fractions containing the product were combined and evaporated in vacuo to give cycloadducts **24g** and **24h**.

6.12.1. Methyl $(4aS^*, 7aS^*)$ -4a-methyl-2-oxo-1,4a,5,6,7,7a-hexahydro-2H-2a,2a1-diazacyclopenta[cd]-indene-3-carboxylate (24g) and methyl (E)-3-(ethyl(isopropyl)amino)acrylate (29). Prepared from 12a (38 mg, 0.25 mmol) and methyl propiolate (21e). Yield: 28 mg (47%) of brownish oil, 24g:29 = 84:16. Characterization data for compound 24g are given in Section 6.11.9.

6.12.2. Methyl (4aR*,7aS*)-2-oxo-4a-phenyl-1,4a,5,6,7,7a-hexahydro-2H-2a,2a1-diazacyclopenta[cd]-indene-3-carboxylate (24h) and methyl (E)-3-(ethyl(isopropyl)amino)acrylate (29). Prepared from 12b (53 mg, 0.25 mmol) and methyl propiolate (27 μ L, 0.3 mmol). Yield: 30 mg (40%) of brownish solid, 24g:29 = 84:16. Characterization data for compound 24h are given in Section 6.11.10.

6.12.3. Methyl (E)-3-(ethyl(isopropyl)amino)acrylate (29). A mixture of anh. CH₂Cl₂ (1 mL), methyl propiolate (21e) (14 μ L, 0.16 mmol), CuI (30 mg, 0.16 mmol), and DIPEA (24 μ L, 0.14 mmol) was stirred at r.t. for 72 h. Volatile components were evaporated in vacuo and the residue was purified by flash column chromatography (silica gel, EtOAc). Fractions containing

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the product were combined and evaporated in vacuo to give **29**. Yield: 14 mg (57%) of a brownish oil. ¹H NMR (500 MHz, CDCl₃): δ 1.15 (3H, t, J = 7.2 Hz), 1.21 (6H, d, J = 6.7 Hz), 3.13 (2H, q, J = 7.2 Hz), 3.47–3.56 (1H, m), 3.66 (s, 3H), 4.58 (1H, d, J = 13.0 Hz), 7.51 (1H d, J = 13.0 Hz). ¹³C NMR (126 MHz, CDCl₃): δ 12.7, 21.8, 40.9, 50.5, 56.5, 83.0, 149.5, 170.5. These data were in agreement with the literature data.²⁴

6.13. Synthesis of $(4S^*, 4aS^*, 7aS^*)$ -2-oxo-4a-phenyloctahydro-2*H*-2a, 2a1-diazacyclopenta[*cd*]indene-4-carboxylic acid (27). A mixture of ester 25c (0.5 mmol) CH₂Cl₂ (4 mL) and CF₃CO₂H (3 mL) was stirred at r.t. for 24 h. Volatile component were evaporated in vacuo and the residue was triturated with Et₂O (10 mL). The precipitate was collected by filtration and washed with Et₂O (2×3 mL) to give carboxylic acid 27. Yield: 69 mg (48%) of white solid; mp 175–176 °C. ¹H NMR (500 MHz, DMSO-d₆): δ 1.20 (1H, dtdd, J = 16.1, 12.4, 8.4, 4.8 Hz), 1.48–1.61 (2H, m), 1.66–1.77 (1H, m), 1.96 (1H, dt, J = 13.9, 3.4 Hz), 2.20 (1H, td, J = 14.2, 3.3 Hz), 2.26 (1H, dd, J = 14.2, 5.2 Hz), 2.52 (1H, overlapped by the signal for DMSO), 2.66 (1H, dd, J = 11.8 Hz), 7.23–7.31 (1H, m), 7.38 (2H, t, J = 7.7 Hz), 7.61–7.67 (2H, m), 12.99 (1H, s). ¹³C NMR (126 MHz, DMSO-d₆): δ 16.5, 23.8, 30.4, 39.8, 45.8, 56.3, 59.2, 69.1, 126.2, 126.8, 128.0, 146.2, 172.5, 173.9. *m/z* (ESI) = 287 (MH⁺). *m/z* (HRMS) Found: 287.1389 (MH⁺). C₁₆H₁₈N₂O₃ requires: *m/z* = 287.1390. v_{max} (ATR) 1732, 1723, 1674, 1413, 1205, 757, 702 cm⁻¹.

6.14. Synthesis of $(4S^*, 4aS^*, 7aS^*)$ -2-oxo-4a-phenyloctahydro-2*H*-2a,2a1-diazacyclopenta[*cd*]indene-4-carboxamides (28a–c). Et₃N (35 µL, 0.25 mmol) was added to a suspension of carboxylic acid 27 (71 mg, 0.25 mmol) in anh. DMF (2 mL) and the mixture was stirred at r.t. for 10 min. Then, bis(pentafluorophenyl) carbonate (99 mg, 0.25 mmol) was added and the mixture was stirred at r.t. for 1 h. Amine (0.25 mmol) and Et₃N (35 µL, 0.25 mmol) were added and stirring at r.t. was continued for 24 h. Volatile components were evaporated in vacuo (50 °C,

2 mbar) and the residue was purified by flash column chromatography (silica gel, CH_2Cl_2 – MeOH, 50:1). Fractions containing the product were combined and evaporated in vacuo to give **28**.

6.14.1. (4S*,4aS*,7aS*)-N-benzyl-2-oxo-4a-phenyloctahydro-2H-2a,2a1-diazacyclopenta-

[*cd*]*indene-4-carboxamide* (**28a**). Prepared from **27** (71 mg, 0.25 mmol) and benzylamine (28 μ L, 0.25 mmol). Yield: 74 mg (79%) of yellow solid, mp 202–203 °C. ¹H NMR (500 MHz, CDCl₃): δ 1.35 (1H, tddd, *J* = 13.9, 10.6, 7.5, 3.3 Hz); 1.56–1.66 (2H, m, overlapped by the signal for H₂O); 1.71 (1H, dddd, *J* = 12.6, 10.5, 7.5, 5.0 Hz); 1.95 (1H, dt, *J* = 13.4, 3.3 Hz); 2.26 (1H, td, *J* = 13.8, 3.4 Hz); 2.40 (1H, dd, *J* = 14.0, 5.3 Hz); 2.60 (1H, t, *J* = 13.9 Hz); 2.91 (1H, br dd, *J* = 11.8, 6.6 Hz); 3.02 (1H, d, *J* = 6.5 Hz), 4.07–4.16 (1H, m); 4.27 (1H, d, *J* = 11.8 Hz); 4.45 (1H, dd, *J* = 14.5, 5.9 Hz); 6.01 (1H,br t, *J* = 5.6 Hz); 7.22–7.28 (1H, m); 7.30–7.40 (7H, m); 7.63 (2H, br d, *J* = 7.1 Hz). ¹³C NMR (126 MHz, CDCl₃): δ 17.1, 24.2, 30.0, 40.0, 44.0, 45.7, 58.5, 59.4, 70.2, 126.5, 127.2, 127.9, 128.2, 128.3, 128.9, 137.7, 146.3, 171.5, 173.0. *m/z* (ESI) = 376 (MH⁺). *m/z* (HRMS) Found: 376.2017 (MH⁺). C₂₃H₂₅N₃O₂ requires: *m/z* = 376.2020. ν_{max} (ATR) 3354, 1687, 1639, 1523, 1382, 1240, 758, 699 cm⁻¹.

6.14.2. $(4S^*, 4aS^*, 7aS^*)$ -N-(3-hydroxypropyl)-2-oxo-4a-phenyloctahydro-2H-2a,2a1diazacyclopenta-[cd]indene-4-carboxamide (**28b**). Prepared from **27** (71 mg, 0.25 mmol) and 3hydroxypropylamine (19 µL, 0.25 mmol). Yield: 65 mg (76%) of pinkish resin. ¹H NMR (500 MHz, CDCl₃): δ 1.35 (1H, ddtt, J = 14.2, 10.6, 7.2, 3.7 Hz), 1.55–1.66 (2H, m), 1.68–1.84 (3H, m), 1.98 (1H, dt, J = 13.5, 3.4 Hz), 2.22 (1H, td, J = 13.8, 3.1 Hz), 2.35–2.48 (1H, m), 2.37 (1H, dd, J = 14.1, 5.3 Hz), 2.60 (2H, t, J = 13.9 Hz), 2.85 (1H, dd, J = 11.5, 6.5 Hz), 3.22 (2H, d, J =6.4 Hz), 3.37–3.49 (2H, m), 3.69 (1H, t, J = 5.8 Hz), 4.11–4.20 (1H, m), 4.17 (1H, d, J = 11.7Hz), 7.23 (1H, br t, J = 7.3 Hz), 7.33 (2H, br t, J = 7.7 Hz), 7.73 (2H, br d, J = 7.5 Hz). ¹³C NMR (126 MHz, CDCl₃): δ 17.1, 24.3, 30.3, 32.0, 37.3, 40.3, 45.8, 58.2, 59.6, 60.3, 70.0, 126.8, 127.2, 128.3, 146.4, 172.9, 173.0. m/z (ESI) = 344 (MH⁺). m/z (HRMS) Found: 344.1966 (MH⁺). C₁₉H₂₆N₃O₃ requires: m/z = 344.1969. v_{max} (ATR) 1680, 1581, 1395, 1290, 1078, 705 cm⁻¹.

6.14.3. (4S*,4aS*,7aS*)-4a-phenyl-4-(piperidine-1-carbonyl)octahydro-2H-2a,2a1-

diazacyclopenta[cd]inden-2-one (28c). Prepared from **27** (71 mg, 0.25 mmol) and piperidine (25 μ L, 0.25 mmol). Yield: 62 mg (70%) of pale orange resin. ¹H NMR (500 MHz, CDCl₃): δ 1.37 (1H, tddd, J = 13.8, 10.5, 7.1, 3.2 Hz), 1.56–1.65 (4H, m), 1.67–1.77 (5H, m), 2.03 (1H, dt, J = 13.3, 3.3 Hz), 2.18 (1H, td, J = 13.8, 3.3 Hz), 2.45 (1H, dd, J = 14.2, 5.3 Hz), 2.60 (1H, br t, J = 13.9 Hz), 2.91 (1H, ddd, J = 11.5, 6.4, 1.2 Hz), 3.46 (1H, d, J = 6.3 Hz), 3.58–3.75 (4H, m), 4.14 (1H, ddt, J = 13.7, 12.0, 5.2 Hz), 4.29 (1H, d, J = 11.5 Hz), 7.29 (1H, br t, J = 7.1 Hz), 7.39 (2H, br t, J = 7.7 Hz), 7.64 (2H, br d, J = 7.2 Hz). ¹³C NMR (126 MHz, CDCl₃): δ 17.4, 24.3, 24.6, 25.8, 26.8, 29.8, 40.1, 43.3, 46.9, 47.4, 53.2, 59.3, 70.2, 126.2, 127.4, 128.6, 146.7, 170.7, 173.5. *m/z* (ESI) = 354 (MH⁺). *m/z* (HRMS) Found: 354.2173 (MH⁺). C₂₁H₂₈N₃O₂ requires: *m/z* = 354.2176. v_{max} (ATR) 2939, 1668, 1580, 1446, 1381, 1288, 702 cm⁻¹.

Supporting Information contains copies of the NMR spectra, data on structure determination by NMR, X-ray diffraction data, computational details, additional Tables and Figures.

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(1) (a) Joule, J. A.; Mills, K. In *Heterocyclic Chemistry*, 5th ed., Wiley-Blackwell, 2010. (b)
 Patrick, G. L. In *An Introduction to Medicinal Chemistry*, 4th ed., Oxford University Press:
 Oxford, UK, 2009. (c) Pernerstorfer, J. Molecular Design and Combinatorial Compound
 Libraries. In *Handbook of Combinatorial Chemistry*. *Drugs, Catalysts, Materials; Vol. 2*;
 Nicolaou, K. C., Hanko, R., Hartwig, W., Eds.; Wiley-VCH Verlag GmbH: Weinheim, Germany,
 2002; pp 725–742. (d) Dolle, R. E. Solid-phase Synthesis of Heterocyclic Systems (Heterocycles
 Containing One Heteroatom). In *Handbook of Combinatorial Chemistry*. *Drugs, Catalysts, Materials; Vol. 2*;
 Nicolaou, K. C., Hanko, R., Hartwig, R., Hartwig, W., Eds.; Wiley-VCH Verlag GmbH:

(2) (a) Couty, F. Evano, G. Pyrazolo[1,5–a]pyridine in Bicyclic 5–6 Systems with One Bridgehead (Ring Junction) Nitrogen Atom: One Extra Heteroatom 1:0. In *Comprehensive Heterocyclic Chemistry III*, Katritzky, A. R.; Ramsden, C. A.; Scriven, E. F. V.; Taylor, R. J. K., Eds.; Vol. 11; Cossy, J., Ed.; Elsevier: Oxford, UK, 2008, pp. 409–424. (b) Howard, A. S. Pyrazolo[1,5–a]pyridine in Bicyclic 5–6 Systems with One Bridgehead (Ring Junction) Nitrogen Atom: One Extra Heteroatom 1:0. In *Comprehensive Heterocyclic Chemistry III*, Katritzky, A. R.; Rees, C. W.; Scriven, E. F. V., Eds.; Vol. 8; Jones, G., Ed.; Elsevier: Oxford, UK, 1996, pp. 249–258.

(3) SciFinder® Scholar Substructure search performed on May 18th 2016.

(4) Allen, S. H.; Johns, B. A.; Gudmundsson, K. S.; Freeman, G. A.; Boyd, F. L.; Secxton,C. H.; Selleseth, D. W.; Creech, K. L.; Moniri, K. R. *Bioorg. Med. Chem.* 2006, *14*, 944–954.

(5) Timári, G.; Soós, T.; Hajós, G.; Messner, A.; Nacsa, J.; Molnár, J. Bioorg. Med. Chem.
 Lett. 1996, 6, 2831–2836.

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54
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60

(6) (a) Bettinetti, L.; Schlotter, K.; Hübner, H.; Gmeiner, P. J. Med. Chem. 2002, 45, 4594-

4597. (b) Lôber, S.; Hübner, H.; Utz, W.; Gmeiner, P. J. Med. Chem. 2001, 44, 2691-2694.

(7) Lôber, S.; Hübner, H.; Gmeiner, P. Bioorg. Med. Chem. Lett. 2002, 12, 2377–2380.

(8) Akahane, A.; Katayama, H.; Mitsunaga, T.; Kita, Y.; Kusunoki, T.; Terai, T.; Yoshida,

K.; Shiokawa, Y. Bioorg. Med. Chem. Lett. 1996, 6, 2059-2062.

(9) Csàny, D.; Hajós, G.; Riedl, Z.; Timári, G.; Bajor, Z.; Cochard, F.; Sapi, J.; Laronze, J.-Y
 Bioorg. Med. Chem. Lett. 2000, 10, 1767–1769.

(10) Gibson, L. C. D.; Hastings, S. F.; McPhee, I.; Clayton, R. A.; Darroch, C. E.; MacKenzie, A.; Mackenzie, F. L.; Nagasawa, M.; Stevens, P. A.; Mackenzie, S. J. *Eur. J. Pharm.*2006, *538*, 39–42.

(11) Xu, X.; Xing, Y.; Shang, Z.; Wang, G.; Cai, Z.; Pan, Y.; Zhao, X. Chem. Phys. 2003, 287, 317–333.

(12) (a) Hoffman, P.; Hünig, S.; Walz, L.; Peters, K.; Schnering, H.-G. *Tetrahedron* 1995, *51*, 13197–13216. (b) Beck, K.; Hünig, S.; Reinold, P. *Tetrahedron* 1988, *44*, 3296–3308.

(13) For a recent review see: Grošelj, U.; Svete, J. Arkivoc 2015, part vi, 175–205; and references cited therein.

(14) Grošelj, U.; Podlogar, A.; Novak, A.; Dahmann, G.; Golobič, A.; Stanovnik, B.; Svete, J. *Synthesis* **2013**, *45*, 639–650.

(15) Mirnik, J.; Grošelj, U.; Novak, A.; Dahmann, G.; Golobič, A.; Kasunič, M.; Stanovnik,B.; Svete, J. *Synthesis* 2013, *45*, 3404–3412.

(16) Lombar, K.; Grošelj, U.; Dahmann, G.; Stanovnik, B.; Svete, J. Synthesis 2015, 47, 497–506.

(17) (a) Beauchemin, A. M.; Clavette, C.; Gan, W.; Markiewicz, T.; Toderian, A. B.
WO2013067646; *Chem. Abstr.* 2013, *158*, 728238. (b) Berger, R.; Duff, K.; Leighton, J. L. J.

49

ACS Paragon Plus Environment

Am. Chem. Soc. 2004, 126, 5686-5687.

(18) Gan, W.; Moon, P. J.; Clavette, C.; Neves, N. D.; Markiewicz, T.; Toderian, A. B.;Beauchemin, A. M. Org. Lett. 2013, 15, 1890–1893.

(19) (a) Gribkov, D. V.; Hultzsch, K. C.; Hampel, F. J. Am. Chem. Soc. 2006, 128, 3748-

3759. (b) Baggelaar, M. P.; Huang, Y.; Feringa, B. L.; Dekker, F. J.; Minnaard, A. J. *Bioorg. Med. Chem.* 2013, 21, 5271–5274.

(20) Duggan, M. E.; Hartman, G. D. WO 9818460; Chem. Abstr. 1998, 128, 321933.

(21) Marcin, L. R.; Thompson, L. A., III; Boy, K. M.; Guernon, J. M.; Higgins, M. A.; Shi, J.;
Wu, Y.-J.; Zhang, Y.; Macor, J. E. WO 2010083141; *Chem. Abstr.* 2010, *153*, 204359.

(22) Bodi, J.; Janos, E., Szöke, K.; Vukics, K.; Gáti, T.; Temesvári, K., Kiss-Bartos, D. WO 2007/072088; Chem. Abstr. 2007, 147, 95527.

(23) Bongers, A.; Moon, P. J.; Beauchemin, A. M. Angew. Chem. Int. Ed. 2015, 54, 15516–15519.

(24) Compound 29 has been prepared before in the ZnBr₂-catalyzed treatment of DIPEA with
21e: Lee, K. Y; Lee, C. G.; Na, J. E.; Kim, J. N. *Tetrahedron Lett.* 2005, *46*, 69–74. Formation of
29 by the same mechanism was confirmed by another experiment, where the enaminone 29 was formed, exclusively, upon treatment of 21e with DIPEA in the presence of CuI.

(25) Grashey, R. Azomethine Imines. In *1,3-Dipolar Cycloaddition Chemistry; Vol. 1*; Padwa, A., Ed.; John Wiley & Sons, Inc.: New York, USA, 1984; pp 733–814.

(26) For recent review on cycloadditions of azomethine imines see: Nájera, C.; Sansano, J.M.; Yus, M. Org. Biomol. Chem. 2015, 13, 8596–8636.

(27) For an illustration see: (a) Pezdirc, L.; Stanovnik, B.; Svete, J. Aust. J. Chem. 2009, 62, 1661–1666. (b)
Keller, M.; Sido, A. S. S.; Pale, P.; Sommer, J. Chem. Eur. J. 2009, 15, 2810–2817. (c) Chassaing, S.; Alix, A.;
Bonongari, T.; Sido, K. S. S.; Keller, M.; Kuhn, P.; Louis, B.; Sommer, J.; Pale, P. Synthesis 2010, 1557–1567. (d)

The Journal of Organic Chemistry

Mizuno, N.; Kamata, K.; Nakagawa, Y.; Oishi, T.; Yamaguchi, K. *Catalysis Today*, 2010, *157*, 359–363. (e) Oishi,
T.; Yoshimura, K.; Yamaguchi, K.; Mizuno, N. *Chem. Lett*, 2010, *39*, 1086–1087. (f) Yoshimura, K.; Oshi, T.;
Yamaguchi, K.; Mizuno, N. *Chem. Eur. J.* 2011, *17*, 3827–3831. (g) Shao, C.; Zhang, Q.; Cheng, G.; Wang, X.; Hu,
Y. *Eur. J. Org. Chem.* 2013, 6443–6448. (h) Pušavec, E.; Mirnik, J.; Šenica, L.; Grošelj, U.; Stanovnik, B.; Svete, J. *Z. Naturforsch.* 2014, *69b*, 615–626. (i) Pušavec Kirar, E.; Grošelj, U.; Mirri, G.; Požgan, F.; Strle, G.; Štefane, B.;
Jovanovski, V.; Svete, J. *J. Org. Chem.* 2016, *81*, 5988–5997.

(28) For details see Supporting Information.

(29) (a) Huisgen, R. 1,3-Dipolar Cycloadditions – Introduction, Survey, Mechanism. In *1,3-Dipolar Cycloaddition Chemistry; Vol. 1*; Padwa, A., Ed.; John Wiley & Sons, Inc.: New York, USA, 1984; pp 1–176. (b) Houk, K. N.; Yamaguchi, K. Theory of 1,3-Dipolar Cycloadditions. In *1,3-Dipolar Cycloaddition Chemistry; Vol. 2*; Padwa, A., Ed.; John Wiley & Sons, Inc.: New York, USA, 1984; pp 407–447.

(30) Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Mennucci, B.; Petersson, G. A.; Nakatsuji, H.; Caricato, M.; Li, X.; Hratchian, H. P.; Izmaylov, A. F.; Bloino, J.; Zheng, G.; Sonnenberg, J. L.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.; Montgomery, J. A., Jr; Peralta, J. E.; Ogliaro, F.; Bearpark, M.; Heyd, J. J.; Brothers, E.; Kudin, K. N.; Staroverov, V. N.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell, A.; Burant, J. C.; Iyengar, S. S.; Tomasi, J.; Cossi, M.; Rega, N.; Millam, J. M.; Klene, M.; Knox, J. E.; Cross, J. B.; Bakken, V.; Adamo, C.; Jaramillo, J.; Gomperts, R.; Stratmann, R. E.; Yazyev, O.; Austin, A. J.; Cammi, R.; Pomelli, C.; Ochterski, J. W.; Martin, R. L.; Morokuma, K.; Zakrzewski, V. G.; Voth, G. A.; Salvador, P.; Dannenberg, J. J.; Dapprich, S.; Daniels, A. D.; Farkas, O.; Foresman, J. B.; Ortiz, J. V.; Cioslowski, J.; Fox, D. J. Gaussian 09, Gaussian, Inc., Wallingford CT, 2009.

(31) (a) Becke, A. D. J. Chem. Phys., 1993, 98, 5648–5652. (b) Lee, C.; Yang, W.; Parr, R. G.
Phys. Rev. B 1988, 37, 785–789. (c) Hehre, W. J.; Radom, L.; Schleyer, P. V. R.; Pople, J. A. In
Ab Initio Molecular Orbital Theory, Wiley: New York, NY, 1986, pp. 1–576.

(32) (a) Lopez, S. A.; Munk, M. E.; Houk, K. N. J. Org. Chem. 2013, 78, 1576–1582. (b) Ess,
D. H.; Houk, K. N. J. Am. Chem. Soc., 2008, 130, 10187–10198. (c) Ess, D. H.; Houk, K. N. J. Am. Chem. Soc., 2007, 129, 10646–10647. (d) Ziegler, T.; Rauk, A. Theor. Chim. Acta, 1977, 46,
1–10. (e) Ziegler, T.; Rauk, A. Inorg. Chem., 1979, 18, 1755–1759. (f) Bickelhaupt, F. M.; Ziegler, T.; Schleyer, P. V. R. Organometallics, 1995, 14, 2288–2296. (g) Contini, A.; Leone, S.; Menichetti, S.; Viglianisi, C.; Trimarico, P. J. Org. Chem. 2006, 71, 5507–5514. (h) Legnani, L.; Lunghi, C.; Marinone Albini F.; Nativi, C.; Richichi, B.; Toma, L. Eur. J. Org. Chem. 2007, 3547–3554.

(33) (a) Domingo, L. R.; Pérez, P. Org. Biomol. Chem. 2011, 9, 7168–7175. (b) Domingo, L.
R.; Sáez, J. A. Org. Biomol. Chem. 2009, 7, 3576–3583. (c) Domingo, L. R.; Chamorro, E.;
Pérez, P. J. Org. Chem. 2008, 73, 4615–4624.

(34) (a) Parr, R. G.; Von Szentpaly, L.; Liu, S. J. Am. Chem. Soc., 1999, 121, 1922–1924. (b)
Domingo, L. R.; Aurell, M. J.; Pérez, P.; Contreras, R. Tetrahedron 2002, 58, 4417–4423.
(35) Eyring, H. J. Chem. Phys. 1935, 3, 107–115.

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