A Novel Procedure for the Synthesis of Benzo[*b*][1,8]naphthyridine-3-carboxylate Derivatives from Morita–Baylis–Hillman Adduct Acetates

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Abstract: A novel procedure has been designed for the preparation of benzo[*b*][1,8]naphthyridine-3-carboxylate derivatives from the reaction of Morita–Baylis–Hillman adduct acetates with primary amines, ammonium acetate or benzenesulfonamides. The approach, which involves readily available starting materials and mild reaction conditions, gives excellent yields after a convenient workup procedure.

Key words: benzo[*b*][1,8]naphthyridine-3-carboxylate, Morita– Baylis–Hillman adduct acetates, primary amines, ammonium acetate, benzenesulfonamides

Quinoline and its annelated derivatives¹ are well known to both synthetic and biological chemists. They are important structural units present in many biologically important molecules that show a broad range of biological activities.^{2,3}

To our knowledge, only one preparative method has been reported in the literature (Scheme 1).³ This approach uses methyl 3-(2-chloroquinolin-3-yl)acrylate as substrate in combination with methylamine in acetonitrile and is followed by reaction with the Vilsmeier reagent. The method finally afforded methyl benzo[*b*][1,8]naphthyridine-3carboxylate (1) in only 42% yield. However, this approach has several drawbacks, including harsh reaction conditions, low reaction efficiency, and requires the use of hazardous and toxic reagents. In addition, the preparation of 1,2-dihydrobenzo[*b*][1,8]naphthyridines has not so far been documented. Thus, the development of simple and efficient methods for selectively constructing such heterocycles employing 'green' reaction conditions remains highly desirable.

The Morita–Baylis–Hillman (MBH) reaction is a well known coupling reaction of aldehydes and activated alkenes, which takes place in the presence of a tertiary base and affords a highly functionalized product. The versatility of the reaction has led to an exponential increase in the synthetic utility of this approach over the last decade,⁴ and increasing numbers of research groups have initiated work on different facets of this reaction.^{5,6}

To access structurally complex and diverse molecules through simple starting substrates under mild reaction

conditions has been one of the underlying principles of chemical research.7 During our ongoing studies toward the exploitation of MBH adduct acetates in heterocyclic chemistry, we have reported that several oxygen- and nitrogen-fused heterocycles, including polyhydrochromenes, polyhydroquinolines and 1,2,4-triazole derivatives, could be readily synthesized from MBH adducts under solvent-free conditions with good to excellent yields.⁸ Moreover, very recently, the direct conversion of acetanilides into 2-chloro-3-formyl quinolines 2 in good isolated yields, has been successfully achieved.⁹ It occurred to us that it would be useful to directly convert MBH adduct acetates such as 3 (derived from 2-chloro-3formyl quinolines) into the desired title compounds by treatment under environmentally benign and clean conditions.

Herein, we wish to report a simple methodology for constructing the target compounds under environmentally friendly conditions.





At the onset of the research, we examined the MBH reaction between 2-chloro-3-formyl quinolines 2 and activated alkenes at room temperature, in the presence of DABCO as a base, under solvent-free conditions. The MBH adducts were then transformed into the acetates 3via acetylation with acetyl chloride (Scheme 2). The results, illustrated in Table 1, show that the substituent group played a minimal role in governing the reactivity of the substrates.

Generally, the MBH reaction proceeded well and afforded the desired products **3a–f** in moderate overall yields (Table 1).

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Scheme 2

 Table 1
 Synthesis of MBH Adduct Acetates 3 under Solvent-Free Conditions^a

Entry	\mathbb{R}^1	\mathbb{R}^2	Product	Yield (%) ^b
1	Н	CO ₂ Me	3a	72
2	Н	CO ₂ Et	3b	71
3	Н	$CO_2(n-Bu)$	3c	72
4	Н	CN	3d	70
5	Me	CO ₂ Me	3e	73
6	Cl	CO ₂ Me	3f	72

^a Reaction conditions: (1) **2** (1 equiv), alkene (2 equiv), DABCO (1.2 equiv), r.t., 2 d; (2) AcCl (9.6 equiv), Et_3N (9.6 equiv), CH_2Cl_2 , r.t., 30 min.

^b Isolated total yield based on **2**.

In light of the successful formation of the MBH adduct acetates **3**, we intended to test the one-pot construction of methyl benzo[*b*][1,8]naphthyridine-3-carboxylate (**1**), from the MBH adduct acetate **3a** and tosylamide (Scheme 3).¹⁰ The reaction conditions were optimized and the results are summarized in Table 2. When an equimolar amount of substrate **3a** was mixed with tosylamide and 1.1 equivalents of potassium carbonate in *N*,*N*-dimethyl-formamide at 120 °C for five hours, the target compound **1** was successfully synthesized in 80% isolated yield.



Scheme 3

 Table 2
 Synthesis of Methyl Benzo[b][1,8]naphthyridine-3-carboxylate (1) under Different Reaction Conditions^a

Entry	Solvent	Base (1.1 equiv)	Temp (°C)	Time (h)	Yield (%) ^b
1	DMF	K ₂ CO ₃	120	5	80
2	DMF	Et ₃ N	120	5	45
3	95% EtOH	K ₂ CO ₃	reflux	30	18
4	95% EtOH	Et ₃ N	reflux	30	20

^a Reaction conditions: **3a** (1 equiv), TsNH₂ (1 equiv), K₂CO₃ (1.1 equiv), DMF, 120 °C, 5 h.

^b Isolated yield based on **3a**.

Interestingly, when an excess of tosylamide (2.5 equiv) was used under the same conditions, the unexpected product **5a** was exclusively obtained in 85% yield (Scheme 4).

We presumed that the formation of product **5a** might proceed through a two-step successive reaction: S_N2' substitution cyclization of the tosylamide with the MBH adduct acetate **3a**, followed by a 1,4-addition process. Enlightened by the research and in order to confirm our hypothesis, the reaction was carried out as shown in Scheme 5. Cyclization of compound **3a** with tosylamide (1.0 equiv) yielded the intermediate compound **4p**, which was then treated with *p*-chlorobenzenesulfonamide (1.5 equiv) and stirred for four hours. The progress of the reaction was monitored by TLC. Gratifyingly, the desired product **5b** was successfully isolated in good yield (83%).

Encouraged by these results, we further investigated the formation of 1,2-dihydrobenzo[b][1,8]naphthyridine-3-carboxylates **4** from MBH adduct acetates **3** and either primary amines or ammonium acetate.

Preliminary experiments were carried out using compound **3a** and aniline (Scheme 6) as typical substrates. The reaction was performed in acetone at room temperature for 10 hours without any catalyst or additive. Unfortunately, the desired product **4a** was obtained in only 45% yield (Table 3, entry 1). In order to optimize the reaction conditions, various factors including bases, solvents, reaction temperature and time were investigated and the results are summarized in Table 3. In the presence of either



Scheme 5

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Scheme 6

triethylamine or potassium carbonate, the required reaction time could be significantly reduced and higher yields were observed (Table 3, entries 3 and 5). When the reaction was carried out at elevated temperature, compound **4a** was formed in good yield within a shorter reaction time, however, the formation of some unknown byproducts were also observed (Table 3, entry 7). After screening a variety of reaction media, 95% ethanol was determined to be the best solvent for generation of the desired product (Table 3, entries 4 and 5). It is worthwhile mentioning that the workup procedure was very convenient; the product could be isolated by simple filtration without further purification when this reaction was performed at room temperature using 95% ethanol as a solvent and triethylamine as a base.

Table 3 Synthesis of Methyl 1-Phenyl-1,2-dihydroben-zo[b][1,8]naphthyridine-3-carboxylate (**4a**) under Different ReactionConditions^a

Entry	Solvent	Base	Temp (°C)	Time (h)	Yield (%) ^b
1	acetone	-	r.t.	10	45
2	acetone	-	reflux	5	55
3	acetone	Et ₃ N	r.t.	5	75 (81) ^c
4	95% EtOH	Et ₃ N	r.t.	4	88 (85) ^c
5	95% EtOH	K ₂ CO ₃	r.t.	4	87 (85) ^c
6	EtOH	Et ₃ N	r.t.	4	81
7	DMF	K ₂ CO ₃	90	0.2	84
8	DMF	K ₂ CO ₃	r.t.	4	79
9	CHCl ₃	Et ₃ N	r.t.	4	64
10	CH_2Cl_2	Et ₃ N	r.t.	4	57
11	DMSO	Et ₃ N	r.t.	4	76

^a Reaction conditions: 3a (1 equiv), amine (1 equiv), base (1.1 equiv).
 ^b Isolated yield based on 3a.

^c Isolated yield when the reaction was carried out at reflux for 1 h.

With the optimal conditions established, a series of primary amines, ammonium acetate and tosylamide were reacted with MBH adduct acetates in order to assess the scope of this method (Table 4).

As shown in Table 4, the reaction was compatible with a variety of primary amines. It can be seen that the primary amines with electron-withdrawing groups required longer

 Table 4
 Synthesis of 1,2-Dihydrobenzo[b][1,8]naphthyridine-3carboxylates 4 from MBH Adduct Acetates 3^a

Entry	\mathbb{R}^1	R ²	R ³	Time (h)	Product	Yield (%) ^b
1	Н	CO ₂ Me	Ph	4	4 a	88
2	Н	CO ₂ Et	Ph	4	4b	89
3	Н	$CO_2(n-Bu)$	Ph	4	4c	89
4	Н	CN	Ph	4	4d	76
5	Me	CO ₂ Me	Ph	4	4e	88
6	Cl	CO ₂ Me	Ph	4	4f	87
7	Н	CO ₂ Me	4-MeC ₆ H ₄	3	4g	87
8	Н	CO ₂ Me	4-MeOC ₆ H ₄	3	4h	90
9	Н	CO ₂ Me	4-ClC ₆ H ₄	6	4i	84
10	Н	CO ₂ Me	$3,4-F_2C_6H_3$	5	4j	85
11	Н	CO ₂ Me	benzyl	2	4k	90
12	Н	CO ₂ Me	cyclohexyl	3	41	89
13	Н	CO ₂ Me	<i>n</i> -Bu	3	4m	88
14	Н	CO ₂ Me	<i>t</i> -Bu	3	4n	87
15	Н	CO ₂ Me	Н	5	40	86
16	Н	CO ₂ Me	Ts	10	4p	20 (89)°

^a Reaction conditions: **3** (1 equiv), R³NH₂ (1.1 equiv), Et₃N (1.5 equiv), r.t., 95% EtOH (5 mL).

^b Isolated yield based on MBH adduct acetates 3.

 $^{\rm c}$ Isolated yield when carried out with K_2CO_3 in DMF at 90 °C for 15 min.

reaction times and gave lower yields than those of primary amines with electron-donating groups. The presence of electron-withdrawing or electron-donating groups in substrate **3** did not appear to exert much influence on either the rate or efficiency of the reaction (Table 4, entries 5 and 6). On the other hand, the presence of a nitrile in substrate **3** gave a lower yield than those of the esters (Table 4, entry 4).

According to the above results, a possible mechanism for the formation of compounds 1, 4 and 5 from MBH adduct acetates 3 can be illustrated as shown in Scheme 7.¹⁰

In summary, we have developed a novel and efficient strategy for the synthesis of various target compounds from MBH adducts and either primary amines, ammonium acetate or benzenesulfonamides under environmentally friendly and clean conditions. The merits of the present process are the simple experimental procedure, high selectivity and high to excellent yields.

Melting points were determined using a Büchi B-540 capillary melting point apparatus and are uncorrected. IR spectra were recorded on a Nicolet Avatar-370 instrument. ¹H and ¹³C NMR spectra were recorded on Varian (400 MHz) instruments using TMS as



Scheme 7

an internal standard. Mass spectra were measured with a Finnigan Trace DSQ instrument. High-resolution mass spectral (HRMS) analyses were measured on an APEX (Bruker) mass III spectrometer using ESI (electrospray ionization) techniques. All spectroscopic data for the products were identical to data from authentic samples. Starting materials and solvents were purchased from common commercial sources and were used without additional purification. Silica-gel for chromatography was purchased from Qingdao Haiyang Chemical Co., Ltd. (200–300 mesh). Petroleum ether (PE), where used, had a boiling range 60–90 °C.

Synthesis of MBH Adduct Acetates (3); General Procedure

A mixture of 2-chloro-3-formyl quinoline (**2**; 10 mmol) and either acrylate or acrylonitrile (20 mmol) and DABCO (1.2 mmol) was kept at r.t. for 2 d under solvent-free conditions. The progress of the reaction was monitored by TLC. After completion of the reaction, the mixture was extracted with EtOAc (2×20 mL) and the combined organic layer was dried over anhydrous Na₂SO₄ and concentrated in vacuo. The desired product was obtained by flash column chromatography on silica gel (PE–EtOAc, 4:1) to give a white solid **2'**.

AcCl (9.6 mmol) at 0 °C was slowly added to a solution of **2'** (8 mmol) and Et_3N (9.6 mmol) in CH_2Cl_2 and stirred at r.t. for 30 min. After completion of the reaction, as indicated by TLC, H_2O (30 mL) was added and the reaction mixture was extracted with EtOAc (2 × 20 mL). The organic phase was collected, dried over anhydrous Na_2SO_4 , and evaporated under reduced pressure to give a crude product, which was purified by flash chromatography on silica gel (PE–EtOAc, 5:1) to give a white solid **3**.

Methyl 2-[Acetoxy(2-chloroquinolin-3-yl)methyl]acrylate (3a) White solid; mp 81.3–83.2 °C; $R_f = 0.50$ (hexanes–EtOAc, 4:1).

IR (KBr): 3078, 2950, 1746, 1708, 1632, 1567, 1402, 1299, 1230, 1048, 750 $\rm cm^{-1}.$

¹H NMR (400 MHz, CDCl₃): δ = 2.18 (s, 3 H, CH₃), 3.75 (s, 3 H, OCH₃), 5.82 (s, 1 H, CH₂), 6.59 (s, 1 H, CH₂), 7.12 (s, 1 H, CH), 7.58 (t, *J* = 8.0 Hz, 1 H, ArH), 7.75 (t, *J* = 8.0 Hz, 1 H, ArH), 7.83 (d, *J* = 8.0 Hz, 1 H, ArH), 8.03 (d, *J* = 8.0 Hz, 1 H, ArH), 8.14 (s, 1 H, ArH).

¹³C NMR (100 MHz, CDCl₃): δ = 20.7, 52.2, 69.9, 126.7, 127.3, 127.6, 128.2, 128.4, 129.7, 130.9, 137.3, 137.5, 147.3, 149.7, 164.9, 168.9.

MS (ESI): m/z (%) = 320 (100) [M⁺ + 1].

HRMS: m/z [M⁺] calcd for C₁₆H₁₄ClNO₄: 319.0611; found: 319.0617.

Ethyl 2-[Acetoxy(2-chloroquinolin-3-yl)methyl]acrylate (3b)

Viscous oil; $R_f = 0.50$ (hexanes–EtOAc, 4:1).

IR (neat): 2982, 1750, 1720, 1566, 1490, 1370, 1224, 1051, 1024 $\rm cm^{-1}.$

¹H NMR (400 MHz, CDCl₃): $\delta = 1.24$ (t, J = 8.0 Hz, 3 H, CH₂CH₃), 2.17 (s, 3 H, CH₃), 4.19 (q, J = 8.0 Hz, 2 H, CH₂CH₃), 5.79 (s, 1 H, CH₂), 6.58 (s, 1 H, CH₂), 7.13 (s, 1 H, CH), 7.58 (t, J = 8.0 Hz, 1 H, ArH), 7.76 (t, J = 8.0 Hz, 1 H, ArH), 7.83 (d, J = 8.0 Hz, 1 H, ArH), 8.03 (d, J = 8.0 Hz, 1 H, ArH), 8.14 (s, 1 H, ArH).

¹³C NMR (100 MHz, CDCl₃): $\delta = 13.9$, 20.8, 61.2, 70.0, 126.8, 127.3, 127.7, 128.1, 128.3, 129.9, 130.9, 137.4, 137.8, 147.3, 149.7, 164.5, 169.0.

MS (ESI): m/z (%) = 334 (100) [M⁺ + 1].

HRMS: m/z [M⁺] calcd for $C_{17}H_{16}CINO_4$: 333.0768; found: 333.0762.

Butyl 2-[Acetoxy(2-chloroquinolin-3-yl)methyl]acrylate (3c) Viscous oil; $R_f = 0.50$ (hexanes-EtOAc, 4:1).

IR (neat): 2960, 1751, 1720, 1566, 1490, 1370, 1223, 1051, 1024 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): $\delta = 0.84$ [t, J = 8.0 Hz, 3 H, (CH₂)₃CH₃], 1.23–1.28 (m, 2 H, CH₂), 1.53–1.59 (m, 2 H, CH₂), 2.17 (s, 3 H, CH₃), 4.13 (t, J = 8.0 Hz, 2 H, CH₂), 5.80 (s, 1 H, CH₂), 6.59 (s, 1 H, CH₂), 7.13 (s, 1 H, CH), 7.58 (t, J = 8.0 Hz, 1 H, ArH), 7.75 (t, J = 8.0 Hz, 1 H, ArH), 7.83 (d, J = 8.0 Hz, 1 H, ArH), 8.03 (d, J = 8.0 Hz, 1 H, ArH), 8.14 (s, 1 H, ArH).

 ^{13}C NMR (100 MHz, CDCl₃): δ = 13.5, 18.9, 20.7, 30.0, 65.1, 70.0, 126.7, 127.3, 127.6, 128.1, 128.3, 129.8, 130.9, 137.4, 137.8, 147.3, 149.7, 164.6, 168.9.

MS (ESI): m/z (%) = 362 (100) [M⁺ + 1].

HRMS: m/z [M⁺] calcd for C₁₉H₂₀ClNO₄: 361.1081; found: 361.1083.

1-(2-Chloroquinolin-3-yl)-2-cyanoallyl Acetate (3d)

White solid; mp 88.4–89.6 °C; $R_f = 0.55$ (hexanes–EtOAc, 4:1).

IR (KBr): 2230, 1749, 1617, 1399, 1208, 1035 cm⁻¹.

 cm^{-1} .

¹H NMR (400 MHz, CDCl₃): δ = 2.26 (s, 3 H, CH₃), 6.22 (s, 1 H, CH₂), 6.26 (s, 1 H, CH₂), 6.80 (s, 1 H, CH), 7.62 (t, *J* = 8.0 Hz, 1 H, ArH), 7.79 (t, *J* = 8.0 Hz, 1 H, ArH), 7.91 (d, *J* = 8.0 Hz, 1 H, ArH), 8.04 (d, *J* = 8.0 Hz, 1 H, ArH), 8.40 (s, 1 H, ArH).

¹³C NMR (100 MHz, CDCl₃): δ = 20.7, 71.1, 115.5, 120.8, 126.8, 127.7, 127.9, 128.3, 131.4, 132.0, 134.6, 137.1, 147.5, 148.0, 168.7.

MS (ESI): m/z (%) = 287 (100) [M⁺ + 1].

HRMS: m/z [M⁺] calcd for C₁₅H₁₁ClN₂O₂: 286.0509; found: 286.0518.

Methyl 2-[Acetoxy(2-chloro-6-methylquinolin-3-yl)methyl]acrylate (3e)

White solid; mp 115.7–116.6 °C; *R_f* = 0.50 (hexanes–EtOAc, 4:1). IR (KBr): 2952, 1739, 1710, 1630, 1496, 1401, 1236, 1051, 821

¹H NMR (400 MHz, CDCl₃): δ = 2.17 (s, 3 H, CH₃), 2.53 (s, 3 H, CH₃), 3.74 (s, 3 H, OCH₃), 5.80 (s, 1 H, CH₂), 6.58 (s, 1 H, CH₂), 7.09 (s, 1 H, CH), 7.58 (d, *J* = 8.0 Hz, 1 H, ArH), 7.60 (s, 1 H, ArH), 7.92 (d, *J* = 8.0 Hz, 1 H, ArH), 8.04 (s, 1 H, ArH).

¹³C NMR (100 MHz, CDCl₃): δ = 20.8, 21.5, 52.2, 70.0, 126.6, 126.8, 127.9, 128.4, 129.6, 133.2, 136.8, 137.4, 137.6, 145.9, 148.8, 165.0, 169.0.

MS (ESI): m/z (%) = 334 (100) [M⁺ + 1].

HRMS: m/z [M⁺] calcd for C₁₇H₁₆ClNO₄: 333.0768; found: 333.0771.

Methyl 2-[Acetoxy(2,6-dichloroquinolin-3-yl)methyl]acrylate (3f)

White solid; mp 135.8–136.6 °C; $R_f = 0.50$ (hexanes–EtOAc, 4:1). IR (KBr): 3120, 1739, 1711, 1630, 1439, 1401, 1236, 1183, 1051 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): δ = 2.18 (s, 3 H, CH₃), 3.75 (s, 3 H, OCH₃), 5.82 (s, 1 H, CH₂), 6.59 (s, 1 H, CH₂), 7.08 (s, 1 H, CH), 7.69 (d, *J* = 8.0 Hz, 1 H, ArH), 7.83 (s, 1 H, ArH), 7.97 (d, *J* = 8.0 Hz, 1 H, ArH), 8.06 (s, 1 H, ArH).

¹³C NMR (100 MHz, CDCl₃): δ = 21.0, 52.5, 70.1, 126.7, 127.8, 128.9, 130.1, 131.2, 132.1, 133.5, 136.6, 137.5, 145.9, 150.3, 165.1, 169.1.

MS (ESI): m/z (%) = 354 (100) [M⁺ + 1].

HRMS: m/z [M⁺] calcd for C₁₆H₁₃Cl₂NO₄: 353.0222; found: 353.0227.

Preparation of Methyl Benzo[b][1,8]naphthyridine-3-carboxylate (1); Typical Procedure

A stirred solution of MBH adduct acetate **3a** (319 mg, 1 mmol), TsNH₂ (172 mg, 1 mmol), and K₂CO₃ (152 mg, 1.1 mmol) in DMF (5 mL) was mixed at 120 °C for 5 h. After completion of the reaction, as indicated by TLC, H₂O (10 mL) was added and the reaction mixture was extracted with EtOAc (3×10 mL). The organic phase was collected, dried over anhydrous Na₂SO₄, and evaporated under reduced pressure to give the crude product **1**, which was purified by flash chromatography on silica gel (PE–EtOAc, 4:1) to give **1** as a pale solid.

Yield: 190 mg (80%); yellowish solid; mp 256.8–258.1 °C; $R_f = 0.40$ (hexanes–EtOAc, 2:1).

IR (KBr): 3132, 1705, 1618, 1400, 1338, 1256, 1159, 1090 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): δ = 4.07 (s, 3 H, OCH₃), 7.68 (t, *J* = 8.0 Hz, 1 H, ArH), 7.94 (t, *J* = 8.0 Hz, 1 H, ArH), 8.11 (d, *J* = 8.0 Hz, 1 H, ArH), 8.44 (d, *J* = 8.0 Hz, 1 H, ArH), 8.99 (s, 1 H, ArH), 9.11 (s, 1 H, ArH), 9.80 (s, 1 H, ArH).

¹³C NMR (100 MHz, CDCl₃): δ = 52.7, 119.5, 123.0, 127.3 (2 × CH), 128.2 (2 × CH), 130.3, 132.7, 140.0, 140.9, 151.9, 155.2, 165.2.

MS (ESI): m/z (%) = 239 (100) [M⁺ + 1].

HRMS: m/z [M⁺] calcd for $C_{14}H_{10}N_2O_2$: 238.0742; found: 238.0739.

Preparation of 1,2-Dihydrobenzo[*b*][1,8]naphthyridine-3-carboxylate (4); General Procedure

To a solution of MBH adduct acetate **3** (1 mmol) in 95% EtOH (5 mL), was added either ammonium acetate (1.1 mmol) or primary amine (1.1 mmol) and Et₃N (1.5 mmol). The mixture was stirred vigorously at r.t. for 4 h, and the progress of the reaction was monitored by TLC. After completion of the reaction, the reaction was cooled to 0 °C and the desired product **4** was obtained by simple filtration without further purification.

Methyl 1-Phenyl-1,2-dihydrobenzo[*b*][1,8]naphthyridine-3-carboxylate (4a)

Yellow solid; mp 228.0–230.5 °C; *R_f* = 0.60 (hexanes–EtOAc, 4:1). IR (KBr): 3145, 1718, 1652, 1615, 1494, 1441, 1384, 1232, 1183,

 $\begin{array}{c} \text{IR} (\text{RD1}), 5145, 1718, 1052, 1015, 1494, 1441, 1564, 1252, 1185, \\ 753 \text{ cm}^{-1}. \end{array}$

¹H NMR (400 MHz, CDCl₃): δ = 3.83 (s, 3 H, OCH₃), 4.91 (s, 2 H, CH₂), 7.17 (t, *J* = 8.0 Hz, 1 H, ArH), 7.24–7.27 (m, 1 H, ArH), 7.41–7.49 (m, 6 H, ArH), 7.53 (d, *J* = 8.0 Hz, 2 H, ArH), 7.66 (s, 1 H, ArH).

¹³C NMR (100 MHz, CDCl₃): δ = 50.9, 52.0, 118.0, 123.2, 124.4, 125.4 (2 × CH), 125.5, 127.0, 127.3, 127.6, 128.9 (2 × CH), 130.4, 133.6, 136.3, 144.1, 148.6, 153.5, 165.1.

MS (ESI): m/z (%) = 317 (100) [M⁺ + 1].

HRMS: m/z [M⁺] calcd for $C_{20}H_{16}N_2O_2$: 316.1212; found: 316.1218.

Ethyl 1-Phenyl-1,2-dihydrobenzo[*b*][1,8]naphthyridine-3-carboxylate (4b)

Yellow solid; mp 134.3–135.9 °C; $R_f = 0.60$ (hexanes–EtOAc, 4:1).

IR (KBr): 3228, 1715, 1617, 1400, 1229, 1176 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): $\delta = 1.34$ (t, J = 8.0 Hz, 3 H, CH₂CH₃), 4.28 (q, J = 8.0 Hz, 2 H, CH₂CH₃), 4.89 (s, 2 H, CH₂), 7.16 (t, J = 8.0 Hz, 1 H, ArH), 7.25–7.27 (m, 1 H, ArH), 7.39–7.47 (m, 6 H, ArH), 7.52 (d, J = 8.0 Hz, 2 H, ArH), 7.65 (s, 1 H, ArH).

¹³C NMR (100 MHz, CDCl₃): δ = 14.2, 50.9, 61.0, 118.0, 123.1, 124.4, 125.4 (2 × CH), 125.8 (2 × CH), 126.9, 127.5, 128.8 (2 × C), 130.3, 133.3, 136.2, 144.1, 148.5, 153.5, 164.6.

MS (ESI): m/z (%) = 331 (100) [M⁺ + 1].

HRMS: m/z [M⁺] calcd for $C_{21}H_{18}N_2O_2$: 330.1368; found: 330.1364.

Butyl 1-Phenyl-1,2-dihydrobenzo[*b*][1,8]naphthyridine-3-carboxylate (4c)

Yellow solid; mp 118.9–120.8 °C; $R_f = 0.60$ (hexanes–EtOAc, 4:1).

IR (KBr): 3132, 1714, 1651, 1614, 1496, 1399, 1239 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): $\delta = 0.97$ [t, J = 8.0 Hz, 3 H, (CH₂)₃CH₃], 1.39–1.48 (m, 2 H, CH₂), 1.66–1.73 (m, 2 H, CH₂), 4.23 (t, J = 8.0 Hz, 2 H, CH₂), 4.90 (s, 2 H, CH₂), 7.16 (t, J = 8.0 Hz, 1 H, ArH), 7.23–7.27 (m, 1 H, ArH), 7.39–7.46 (m, 6 H, ArH), 7.50 (s, 1 H, ArH), 7.52 (d, J = 8.0 Hz, 1 H, ArH), 7.66 (s, 1 H, ArH).

 ^{13}C NMR (100 MHz, CDCl₃): δ = 13.9, 19.4, 30.9, 51.1, 65.1, 118.3, 123.4, 124.6, 125.6 (3 \times CH), 126.1, 127.2, 127.8, 129.1 (2 \times CH), 130.5, 133.5, 136.4, 144.3, 148.8, 153.7, 164.9.

MS (ESI): m/z (%) = 359 (100) [M⁺ + 1].

HRMS: m/z [M⁺] calcd for C₂₃H₂₂N₂O₂: 358.1681; found: 358.1679.

1-Phenyl-1,2-dihydrobenzo[*b*][1,8]naphthyridine-3-carbonitrile (4d)

Yellow solid; mp 179.8–182.1 °C; $R_f = 0.60$ (hexanes–EtOAc, 4:1). IR (KBr): 3143, 2201, 1641, 1618, 1493, 1400, 753 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): δ = 4.81 (s, 2 H, CH₂), 7.20–7.29 (m, 3 H, ArH), 7.37–7.47 (m, 6 H, ArH), 7.55 (d, *J* = 8.0 Hz, 1 H, ArH), 7.64 (s, 1 H, ArH).

¹³C NMR (100 MHz, CDCl₃): δ = 29.6, 51.6, 106.7, 116.8, 123.7, 124.2, 125.2 (2 × CH), 125.9, 127.2, 127.8, 129.0 (2 × CH), 131.0, 136.3, 138.6, 143.2, 148.6, 152.5.

MS (ESI): m/z (%) = 284 (100) [M⁺ + 1].

HRMS: *m*/*z* [M⁺] calcd for C₁₉H₁₃N₃: 283.1109; found: 283.1114.

Methyl 7-Methyl-1-phenyl-1,2-dihydrobenzo[b][1,8]naphthyridine-3-carboxylate (4e)

Yellow solid; mp 224.4–225.2 °C; $R_f = 0.60$ (hexanes–EtOAc, 4:1). IR (KBr): 3147, 1715, 1650, 1494, 1400, 1236, 1186, 823 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): δ = 2.40 (s, 3 H, CH₃), 3.83 (s, 3 H, OCH₃), 4.89 (s, 2 H, CH₂), 7.25–7.32 (m, 3 H, ArH), 7.44–7.52 (m, 6 H, ArH), 7.61 (s, 1 H, ArH).

¹³C NMR (100 MHz, CDCl₃): δ = 21.1, 44.8, 52.1, 118.0, 124.2, 125.2 (2 × CH), 125.4, 125.5, 126.4, 126.9, 129.2 (2 × CH), 132.6, 133.0, 133.6, 136.1, 142.3, 143.8, 152.8, 165.0.

MS (ESI): m/z (%) = 331 (100) [M⁺ + 1].

HRMS: m/z [M⁺] calcd for $C_{21}H_{18}N_2O_2$: 330.1368; found: 330.1365.

Methyl 7-Chloro-1-phenyl-1,2-dihydrobenzo[*b*][1,8]naphthyridine-3-carboxylate (4f)

Yellow solid; mp 241.4–245.0 °C; $R_f = 0.60$ (hexanes–EtOAc, 4:1).

IR (KBr): 3131, 1719, 1652, 1616, 1497, 1399, 1077 cm⁻¹.

¹H NMR (400 MHz, DMSO- d_6): $\delta = 3.77$ (s, 3 H, OCH₃), 4.84 (s, 2 H, CH₂), 7.23 (d, J = 8.0 Hz, 1 H, ArH), 7.26–7.29 (m, 1 H, ArH), 7.39 (d, J = 2.4 Hz, 1 H, ArH), 7.41–7.46 (m, 4 H, ArH), 7.57 (s, 1 H, ArH), 7.72 (d, J = 2.0 Hz, 1 H, ArH), 8.00 (s, 1 H, ArH).

¹³C NMR (100 MHz, CDCl₃): δ = 51.2, 52.2, 118.7, 124.9, 125.5 (3 × CH), 125.9, 126.2 (2 × CH), 128.3, 129.1 (2 × CH), 130.9, 133.1, 135.2, 143.7, 146.8, 153.6, 164.9.

MS (ESI): m/z (%) = 351 (100) [M⁺ + 1].

HRMS: m/z [M⁺] calcd for C₂₀H₁₅ClN₂O₂: 350.0822; found: 350.0824.

Methyl 1-*p*-Tolyl-1,2-dihydrobenzo[*b*][1,8]naphthyridine-3-carboxylate (4g)

Yellow solid; mp 210.0–211.7 °C; $R_f = 0.60$ (hexanes–EtOAc, 4:1).

IR (KBr): 3136, 1717, 1652, 1610, 1489, 1442, 1399, 1232, 1178 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): δ = 2.39 (s, 3 H, CH₃), 3.81 (s, 3 H, OCH₃), 4.87 (s, 2 H, CH₂), 7.14 (t, *J* = 8.0 Hz, 1 H, ArH), 7.23 (d, *J* = 8.0 Hz, 2 H, ArH), 7.32 (d, *J* = 8.0 Hz, 2 H, ArH), 7.38–7.51 (m, 4 H, ArH), 7.61 (s, 1 H, ArH).

¹³C NMR (100 MHz, CDCl₃): δ = 21.1, 51.2, 52.0, 117.9, 123.0, 124.3, 125.3 (2 × CH), 125.4, 126.9, 127.6, 129.5 (2 × CH), 130.3, 133.7, 135.1, 136.2, 141.5, 148.7, 153.6, 165.1.

MS (ESI): m/z (%) = 331 (100) [M⁺ + 1].

HRMS: m/z [M⁺] calcd for C₂₁H₁₈N₂O₂: 330.1368; found: 330.1369.

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Methyl 1-(4-Methoxyphenyl)-1,2-dihydrobenzo[*b*][1,8]naph-thyridine-3-carboxylate (4h)

Yellow solid; mp 208.5–209.9 °C; $R_f = 0.60$ (hexanes–EtOAc, 4:1). IR (KBr): 3139, 3007, 1715, 1651, 1613, 1509, 1443, 1381, 1239, 1180 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): δ = 3.81 (s, 3 H, OCH₃), 3.85 (s, 3 H, OCH₃), 4.86 (s, 2 H, CH₂), 6.96–6.99 (m, 2 H, ArH), 7.13 (t, *J* = 8.0 Hz, 1 H, ArH), 7.32–7.44 (m, 4 H, ArH), 7.49 (d, *J* = 8.0 Hz, 2 H, ArH), 7.60 (s, 1 H, ArH).

¹³C NMR (100 MHz, CDCl₃): δ = 51.6, 52.0, 55.4, 114.3 (2 × CH), 117.7, 122.9, 124.2, 125.2, 126.9, 127.0 (2 × CH), 127.6, 130.3, 133.8, 136.2, 137.0, 148.8, 153.8, 157.3, 165.1.

MS (ESI): m/z (%) = 347 (100) [M⁺ + 1].

HRMS: m/z [M⁺] calcd for C₂₁H₁₈N₂O₃: 346.1317; found: 346.1310.

Methyl 1-(4-Chlorophenyl)-1,2-dihydrobenzo[b][1,8]naph-thyridine-3-carboxylate (4i)

Yellow solid; mp 177.0–178.8 °C; $R_f = 0.60$ (hexanes–EtOAc, 4:1).

IR (KBr): 3133, 1718, 1706, 1651, 1615, 1494, 1399, 1239 cm⁻¹.

¹H NMR (400 MHz, DMSO- d_6): δ = 3.76 (s, 3 H, OCH₃), 4.80 (s, 2 H, CH₂), 7.21 (t, *J* = 8.0 Hz, 1 H, ArH), 7.28 (d, *J* = 8.0 Hz, 1 H, ArH), 7.43–7.50 (m, 5 H, ArH), 7.59 (s, 1 H, ArH), 7.64 (d, *J* = 8.0 Hz, 1 H, ArH), 8.05 (s, 1 H, ArH).

¹³C NMR (100 MHz, DMSO- d_6): δ = 50.4, 52.0, 117.6, 123.2, 124.2, 125.3, 126.0, 127.4 (2 × CH), 127.9, 128.6 (2 × CH), 129.2, 130.4, 132.7, 136.5, 142.7, 147.7, 152.8, 164.3.

MS (ESI): m/z (%) = 351 (100) [M⁺ + 1].

HRMS: m/z [M⁺] calcd for C₂₀H₁₅ClN₂O₂: 350.0822; found: 350.0816.

Methyl 1-(3,4-Difluorophenyl)-1,2-dihydrobenzo[*b*][1,8]naph-thyridine-3-carboxylate (4j)

Yellow solid; mp 186.5–186.9 °C; $R_f = 0.60$ (hexanes–EtOAc, 4:1). IR (KBr): 3128, 1707, 1650, 1613, 1516, 1480, 1398, 1218, 1174 cm⁻¹.

¹H NMR (400 MHz, DMSO- d_6): δ = 3.77 (s, 3 H, OCH₃), 4.80 (s, 2 H, CH₂), 7.22 (t, *J* = 8.0 Hz, 1 H, ArH), 7.29–7.33 (m, 2 H, ArH), 7.44–7.53 (m, 2 H, ArH), 7.59–7.66 (m, 3 H, ArH), 8.06 (s, 1 H, ArH).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 50.6, 51.9, 115.4, 115.6, 117.1, 117.3, 117.5, 122.7, 123.2, 124.2, 125.2, 126.0, 127.9, 130.5, 132.7, 136.5, 140.6, 147.6, 152.8, 164.3.

MS (ESI): m/z (%) = 353 (100) [M⁺ + 1].

HRMS: m/z [M⁺] calcd for $C_{20}H_{14}F_2N_2O_2$: 352.1023; found: 352.1026.

Methyl 1-Benzyl-1,2-dihydrobenzo[*b*][1,8]naphthyridine-3carboxylate (4k)

Yellow solid; mp 177.0–179.0 °C; $R_f = 0.60$ (hexanes–EtOAc, 4:1).

IR (KBr): 3126, 3023, 1703, 1650, 1614, 1393, 1244, 1217, 1080 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): δ = 3.77 (s, 3 H, OCH₃), 4.46 (s, 2 H, CH₂), 4.97 (s, 2 H, CH₂), 7.13 (t, *J* = 8.0 Hz, 1 H, ArH), 7.24–7.33 (m, 3 H, ArH), 7.39–7.47 (m, 6 H, ArH), 7.50 (d, *J* = 8.0 Hz, 1 H, ArH).

¹³C NMR (100 MHz, CDCl₃): δ = 48.5, 50.4, 52.0, 116.8, 122.4, 123.8, 124.7, 126.3, 127.2, 127.7, 128.4 (2 × CH), 128.5 (2 × CH), 130.4, 133.8, 136.1, 137.4, 149.4, 153.7, 165.2.

MS (ESI): m/z (%) = 331 (100) [M⁺ + 1].

HRMS: m/z [M⁺] calcd for C₂₁H₁₈N₂O₂: 330.1368; found: 330.1373.

Methyl 1-Cyclohexyl-1,2-dihydrobenzo[*b*][1,8]naphthyridine-3-carboxylate (4l)

Yellow solid; mp 182.1–184.3 °C; $R_f = 0.60$ (hexanes–EtOAc, 4:1). IR (KBr): 3132, 2929, 2852, 1719, 1657, 1616, 1492, 1400, 1231

cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 1.50–1.62 (m, 5 H, CH₂), 1.71– 1.88 (m, 5 H, CH₂), 3.82 (s, 3 H, OCH₃), 4.48 (s, 2 H, CH₂), 4.94– 4.95 (m, 1 H, CH), 7.08 (t, *J* = 8.0 Hz, 1 H, ArH), 7.34 (s, 1 H, ArH), 7.40–7.44 (m, 3 H, ArH), 7.52 (d, *J* = 8.0 Hz, 1 H, ArH).

¹³C NMR (100 MHz, CDCl₃): δ = 25.9 (3 × CH₂), 28.6 (2 × CH₂), 42.8, 51.9, 52.6, 117.6, 122.0, 123.4, 124.9, 126.2, 127.5, 130.1, 133.8, 135.6, 149.5, 153.6, 165.5.

MS (ESI): m/z (%) = 323 (100) [M⁺ + 1].

HRMS: $\textit{m/z}~[M^+]$ calcd for $C_{20}H_{22}N_2O_2{:}$ 322.1681; found: 322.1680.

Methyl 1-Butyl-1,2-dihydrobenzo[b][1,8]naphthayridine-3-carboxylate (4m)

Yellow solid; mp 112.9–114.5 °C; $R_f = 0.60$ (hexanes–EtOAc, 4:1).

IR (KBr): 3137, 2957, 2926, 1719, 1651, 1614, 1510, 1400, 1235, 1175 $\rm cm^{-1}.$

¹H NMR (400 MHz, CDCl₃): $\delta = 0.99$ [t, J = 8.0 Hz, 3 H, (CH₂)₃CH₃], 1.39–1.45 (m, 2 H, CH₂), 1.63–1.70 (m, 2 H, CH₂), 3.66 (t, J = 8.0 Hz, 2 H, CH₂), 3.82 (s, 3 H, OCH₃), 4.56 (s, 2 H, CH₂), 7.08 (t, J = 8.0 Hz, 1 H, ArH), 7.36 (s, 1 H, ArH), 7.40–7.44 (m, 3 H, ArH), 7.51 (d, J = 8.0 Hz, 1 H, ArH).

 ^{13}C NMR (100 MHz, CDCl₃): δ = 14.0, 20.2, 27.9, 47.2, 49.0, 52.0, 117.1, 122.0, 123.5, 124.6, 126.2, 127.7, 130.3, 134.1, 135.7, 149.7, 153.8, 165.4.

MS (ESI): m/z (%) = 297 (100) [M⁺ + 1].

HRMS: m/z [M⁺] calcd for $C_{18}H_{20}N_2O_2$: 296.1525; found: 296.1519.

Methyl 1-*tert*-Butyl-1,2-dihydrobenzo[*b*][1,8]naphthyridine-3-carboxylate (4n)

Yellow solid; mp 127.4–130.5 °C; $R_f = 0.60$ (hexanes–EtOAc, 4:1). IR (KBr): 3136, 2956, 1693, 1615, 1436, 1397, 1249, 720 cm⁻¹.

¹H NMR (400 MHz, DMSO- d_6): $\delta = 1.61$ (s, 9 H, *t*-Bu), 3.76 (s, 3 H, OCH₃), 4.45 (s, 2 H, CH₂), 7.16–7.19 (m, 1 H, ArH), 7.42 (s, 1 H, ArH), 7.47 (d, J = 3.2 Hz, 2 H, ArH), 7.57 (d, J = 8.0 Hz, 1 H, ArH), 7.81 (s, 1 H, ArH).

¹³C NMR (100 MHz, DMSO- d_6): $\delta = 27.5$ (3 × CH₃), 43.9, 51.8, 56.9, 119.5, 122.5, 122.9, 125.9, 126.5, 127.6, 130.0, 133.2, 135.1, 147.4, 154.4, 164.7.

MS (ESI): m/z (%) = 297 (100) [M⁺ + 1].

HRMS: $\textit{m/z}~[M^+]$ calcd for $C_{18}H_{20}N_2O_2{:}$ 296.1525; found: 296.1520.

Methyl 1,2-Dihydrobenzo[*b*][1,8]naphthyridine-3-carboxylate (40)

Yellow solid; mp 186.2–188.1 °C; $R_f = 0.60$ (hexanes–EtOAc, 4:1). IR (KBr): 3142, 1709, 1651, 1615, 1437, 1398, 1242, 754 cm⁻¹.

¹H NMR (400 MHz, DMSO- d_6): $\delta = 3.75$ (s, 3 H, OCH₃), 4.46 (s, 2 H, CH₂), 7.08 (t, J = 8.0 Hz, 1 H, ArH), 7.16 (s, 1 H, NH), 7.32 (d, J = 8.0 Hz, 1 H, ArH), 7.39–7.44 (m, 2 H, ArH), 7.52 (d, J = 8.0 Hz, 1 H, ArH), 7.79 (s, 1 H, ArH).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 42.7, 51.9, 116.1, 121.9, 123.6, 124.5, 125.2, 128.1, 130.4, 132.9, 135.8, 148.9, 154.7, 164.7. MS (ESI): *m*/*z* (%) = 241 (100) [M⁺ + 1].

HRMS: m/z [M⁺] calcd for C₁₄H₁₂N₂O₂: 240.0899; found: 240.0897.

Methyl 1-Tosyl-1,2-dihydrobenzo[*b*][1,8]naphthyridine-3-carboxylate (4p)

White solid; mp 208.3–209.0 °C; $R_f = 0.45$ (hexanes–EtOAc, 4:1).

IR (KBr): 3133, 1705, 1618, 1399, 1338, 1255, 1159, 1089 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): δ = 2.40 (s, 3 H, CH₃), 3.88 (s, 3 H, OCH₃), 5.03 (s, 2 H, CH₂), 7.30–7.38 (m, 3 H, ArH), 7.48 (s, 1 H, ArH), 7.58–7.63 (m, 2 H, ArH), 7.76–7.81 (m, 2 H, ArH), 8.18 (d, *J* = 8.0 Hz, 2 H, ArH).

¹³C NMR (100 MHz, CDCl₃): δ = 21.6, 45.5, 52.3, 118.7, 125.5, 125.6, 126.7, 127.7, 127.8, 128.9 (2 × CH), 129.1 (2 × CH), 130.8, 132.8, 136.4, 137.1, 144.1, 146.8, 148.5, 164.6.

MS (ESI): m/z (%) = 395 (100) [M⁺ + 1].

HRMS: m/z [M⁺] calcd for $C_{21}H_{18}N_2O_4S$: 394.0987; found: 394.0986.

Methyl 4-(4-Methylphenylsulfonamido)benzo[b][1,8]naphthyridine-3-carboxylate (5a)

A stirred solution of **3a** (319 mg, 1 mmol), TsNH₂ (430 mg, 2.5 mmol), and K_2CO_3 (152 mg, 1.1 mmol) in DMF (5 mL) was heated at 120 °C for 5 h. After completion of the reaction, as indicated by TLC, H₂O (10 mL) was added and the reaction mixture was extracted with EtOAc (3 × 10 mL). The organic phase was collected, dried over anhydrous Na₂SO₄, and evaporated under reduced pressure to give the crude product **5a**, which was purified by flash chromatography on silica gel (PE–EtOAc, 2:1).

Yield: 346 mg (85%); yellow solid; mp 271.5–273.4 °C; $R_f = 0.30$ (hexanes–EtOAc, 2:1).

IR (KBr): 3414, 3133, 1726, 1619, 1508, 1401, 1261, 1081 cm⁻¹.

¹H NMR (400 MHz, DMSO- d_6): δ = 2.43 (s, 3 H, CH₃), 3.93 (s, 3 H, OCH₃), 7.43–7.46 (m, 3 H, ArH), 7.78 (d, *J* = 8.0 Hz, 1 H, ArH), 7.88–7.92 (m, 3 H, ArH), 8.53 (d, *J* = 8.0 Hz, 1 H, ArH), 9.31 (s, 1 H, ArH), 9.75 (s, 1 H, ArH), 13.49 (s, 1 H, NH).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 20.9, 52.4, 111.1, 118.2, 119.4, 119.5, 123.8, 125.6 (2 × CH), 127.8, 129.4 (2 × CH), 135.2, 139.9, 141.1, 141.4, 142.0, 151.2, 155.5, 160.2, 164.4.

MS (ESI): m/z (%) = 406 (100) [M⁺ – 1).

HRMS: m/z [M⁺] calcd for C₂₁H₁₇N₃O₄S: 407.0940; found: 407.0937.

Methyl 4-(4-Chlorophenylsulfonamido)benzo[b][1,8]naphthyridine-3-carboxylate (5b)

A stirred solution of **3a** (319 mg, 1 mmol), TsNH₂ (172 mg, 1 mmol), and K₂CO₃ (152 mg, 1.1 mmol) in DMF (5 mL) was heated at 90 °C for 15 min. *p*-Chlorobenzenesulfonamide (287 mg, 1.5 mmol) was added and the reaction was stirred at 120 °C for 4 h. After completion of the reaction, as indicated by TLC, H₂O (10 mL) was added and the reaction mixture was extracted with EtOAc (3×10 mL). The organic phase was collected, dried over anhydrous Na₂SO₄, and evaporated under reduced pressure to give the crude product **5b**, which was purified by flash chromatography on silica gel (PE–EtOAc, 2: 1).

Yield: 354 mg (83%); yellow solid; mp 272.0–273.6 °C; $R_f = 0.30$ (hexanes–EtOAc, 2:1).

IR (KBr): 3427, 3170, 1724, 1619, 1508, 1400, 1259, 1080 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): δ = 4.01 (s, 3 H, OCH₃), 7.40 (t, *J* = 8.0 Hz, 1 H, ArH), 7.52–7.57 (m, 3 H, ArH), 7.77 (t, *J* = 8.0 Hz, 1 H, ArH), 8.05–8.09 (m, 2 H, ArH), 8.68 (d, *J* = 8.0 Hz, 1 H, ArH), 9.27 (s, 1 H, ArH), 9.84 (s, 1 H, ArH).

¹³C NMR (100 MHz, CDCl₃): δ = 52.6, 112.2, 117.4, 119.9, 120.5, 124.0, 127.7 (2 × CH), 128.9 (2 × CH), 129.1, 135.2, 137.9, 139.7, 141.8, 142.6, 151.0, 155.8, 161.2, 165.0.

MS (ESI): m/z (%) = 426 (100) [M⁺ – 1].

HRMS: m/z [M⁺] calcd for $C_{20}H_{14}ClN_3O_4S$: 427.0394; found: 427.0388.

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