ORIGINAL RESEARCH



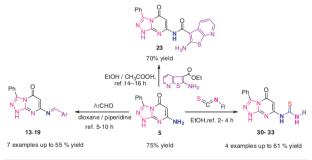
Synthesis of novel 1, 2, 4-triazolopyrimidines and their evaluation as antimicrobial agents

Ameen A. Abu-Hashem^{1,2} · Hoda A. R. Hussein¹ · Khadeja M. Abu-zied¹

Received: 27 April 2016 / Accepted: 13 October 2016 © Springer Science+Business Media New York 2016

Abstract 7-aminopyrimidin-5(1H)-one (5) was utilized as a starting material for the preparation of new, Schiff bases (13–19), amides 23, 24, imide 25, as well as thiourea derivatives 30–33 incorporating the triazolopyrimidines nucleus in their molecules. Structural elucidations for the new compounds were based upon compatible microanalytical and spectroscopic measurement. Biological evaluation indicated that compounds 23, 31, 33, 32, and 30 possess promising antimicrobial activities.

Graphical Abstract



Antimicrobial activity of compounds 23, 31, 33, 32 and 30 have found higher than commonly used antibiotics like [Penicillin G or Streptomycin] and [Itraconazole or Clotrimazole]

Electronic supplementary material The online version of this article (doi:10.1007/s00044-016-1733-5) contains supplementary material, which is available to authorized users.

Ameen A. Abu-Hashem aminaliabuhashem@yahoo.com

² Chemistry Department, Faculty of Science, Jazan University, Jezan 2097, Saudi Arabia **Keywords** Triazolopyrimidines · Benzohydrazide · Isothiocyocyanate · Antimicrobial activity

Introduction

Pyrimidines and purines are the main bases constituents of nucleic acids. In addition of other rings to the pyrimidines core is expected to result in novel biological potentialities (Aly et al. 2012; Mohamed et al. 2010; Abdel-Latif et al. 2007). Consequently, the aza analogs of purines, mainly the triazolopyrimidines, also are important (Mohamed et al. 2010). The study of compounds incorporating the triazolopyrimidine developed due to their varied effects in diverse domains. Triazolopyrimidines (TPs), a subtype of purine analogs, have been the subject of chemical and biological studies due to their interesting pharmacology including antibacterial (Broom et al. 1976; Grivsky et al. 1980; Nargund et al. 1991), antifungal (Chen et al. 2008), antimicrobial (Donkor et al. 1995), antihypertensive (Degraw et al. 1974), antileishmanial (Ellingboe and Princeton 1996; Pandey et al. 2004) anticonvulsant (Devanov et al. 1991), antitumor (Navarro et al. 1998), cytotoxicity (Magan et al. 2004), therapeutic potentiality (Magan et al. 2005), potent and selective ATP site directed inhibition of the EGFreceptor protein tyrosine kinase (Traxler et al. 1996) and cardiovascular (Sato et al. 1980) activities. In addition, triazolopyrimidines are versatile ligands and their derived coordination compounds can be considered as model systems for metal-ligand interactions observed in biological systems (Fischer 2008; Salas et al. 1999). Trapidil, the wellknown TP derivative behaves as a phosphodiesterase inhibitor and as a platelet-derived growth factor antagonist

¹ Photochemistry Department (Heterocyclic Unit), National Research Center, Dokki, Giza 12622, Egypt

(Fischer 2008). Triazolopyrimidine antibiotic was isolated from nature Essramycin and was proved to be potent antibiotic agents (El-Gendy et al. 2008). Bearing these facts in mind, coupled with our growing interest in the realm of biologically active heterocycles (Abu-Hashem et al. 2011; Abu-Hashem and Aly 2012; Hussein et al. 2010; Abu-Hashem et al. 2011; Khidre et al. 2011), we report now on the synthesis and antimicrobial evaluation of novel triazolepyrimidines. Hybird presence of the trizole and pyrimidine moieties in the same compound is hoped to boost the bioactivity.

Results and discussion

Chemistry

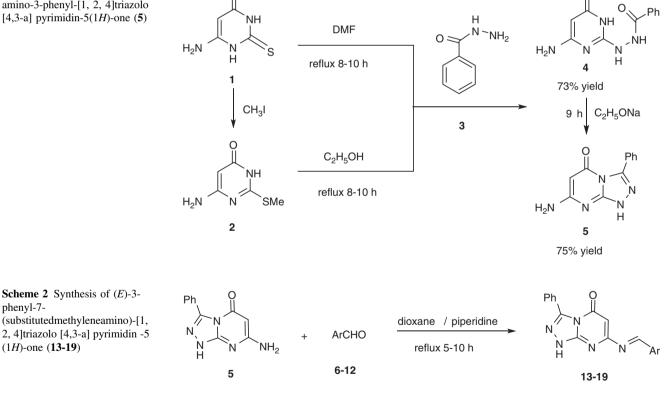
In this investigation, the starting 7-Aminopyrimidin-5(1H)one key derivative (5) was prepared via heating under reflux of 6-aminothiouracil (1) with benzohydrazide (3) in

Scheme 1 Synthesis of 7amino-3-phenyl-[1, 2, 4]triazolo [4,3-a] pyrimidin-5(1H)-one (5)

dimethyl formamide, followed by stirring under reflux of the formed pyrimdine (4) in sodium ethoxide. In another route compound (4) was prepared by alkylation of 6aminothiouracil (1) with methyl-iodide in ethanolic potassium hydroxide solution afforded 6-amino-2-(methylthio) pyrimidin-4(3H)-one (2) (Guiney et al. 2003). The last compound (2) reacted with benzohydrazide (3) in absolute ethanol to give the corresponding (4) (Scheme 1).

Reaction of 7-amino-3-phenyl-[1,2,4]triazolo [4,3-a] pyrimidin-5(1H)-one (5) with some aromatic aldehydes namely: 4-fluorobenzaldehyde (6), 4-chlorobenzaldehyde (7), 4-methoxybenzaldehyde (8), thiophene-2-carbaldehyde (9), isonicotinaldehyde (10) in dioxane containing catalytic amount of piperidine or with 5-chloro-3-methyl-1-phenyl-1H-pyrazole-4-carbaldehyde (11) or 2-azido-6-methylquinoline-3-carbaldehyde (12) in ethanol containing catalytic amount of acetic acid afforded the corresponding Schiff bases (13-19), (Scheme 2).

Furthermore, reaction of compound 5 with ethyl 3aminothieno [2,3-b] pyridine-2-carboxylate (20), ethyl

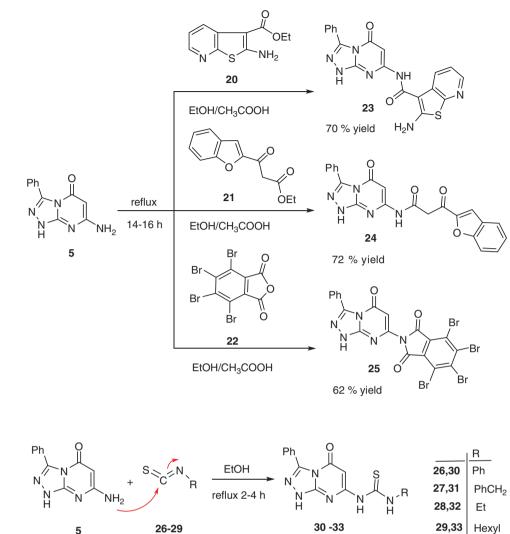


6,13; Ar=4-F-C₆H₄; 7,14; Ar=4-Cl-C₆H₄; 8,15; Ar=4-OCH₃-C₆H₄; 9,16; Ar=2-thienyl

(substitutedmethyleneamino)-[1, 2, 4]triazolo [4,3-a] pyrimidin -5 (1H)-one (13-19)

phenvl-7-

Scheme 3 Synthesis of amides 23, 24, and imide 25



Scheme 4 Synthesis of thiourea

derivatives 30-33

aureus. While compounds (23–25) and (30–33) were the most active against *Bacillus subtilis*; showing moderate influence on *Pseudomonas aeruginosa*. Compounds (30–33) were active against *Escherichia coli* and also exhibited marked antifungal properties.

Structure-activity relationships (SAR)

All compounds were tested in vitro of antimicrobial activity, comparing the results obtained for the antimicrobial properties of the compounds reported in this study with their structures, the following SAR are postulated (Fig. 1): (i) It seems that most of the Schiff's bases, amids, imide and thiourea derivatives are more potent than the starting TP **5** which may be due to the presence of the N=CH-R, NHCOR and NHCSNHR moieties, respectively. (ii) The antimicrobial activities of Schiff's bases **13–15** follow the order **13** < **14** < **15**, which may be attributed to the electronic effect of the *p*-substituents, i.e, OCH₃>Cl>F by +M.

4-(2,3-dihydro benzofuran-2-yl)-2,4-dioxobutanoate (**21**) or 4,5,6,7-tetrabromoiso-benzofuran-1,3-dione (**22**) in a mixture of absolute ethanol and few drops of glacial acetic acid as catalyst afforded the corresponding amides (**23**, **24**) and imide (**25**), respectively (Scheme 3).

Moreover, compound 5 was treated with isothiocyocyanate derivatives namely: phenyl- isothiocyanate (26), benzylisothiocyanate (27), ethylisothiocyanate (28) and cyclohexyl- isothiocyanate (29), respectively, under refluxing in ethanol to afford the corresponding thiourea derivatives (30–33), (Scheme 4). The structures of new compounds confirmed by elemental analyses and spectral data are presented in experimental Section.

Biological activity

In vitro Antimicrobial screening (Cruickshank et al. 1975) of the new product (Tables 1 and 2) indicated that compounds 23 and (30–33) were the most potent against *S*.

Compounds	Gram positive bacteria		Gram negative bacteria	
	Staphylococcus aureus (RCMB 000106) ^d	Bacillus subtilis (RCMB 000107)	Pseudomonas aeruginosa (RCMB 000102)	Escherichia coli (RCMB 000103)
1	10.30 ± 0.52	8.85 ± 0.45	NA	NA
2	11.10 ± 0.32	9.10 ± 0.60	NA	NA
3	NA	NA	NA	NA
4	12.90 ± 0.10	10.40 ± 0.20	NA	7.10 ± 0.52
5	14.40 ± 0.28	12.80 ± 0.50	NA	7.80 ± 0.74
13	15.50 ± 0.62	14.50 ± 0.56	NA	8.20 ± 0.60
14	17.60 ± 0.81	15.10 ± 0.34	NA	9.50 ± 0.12
15	19.40 ± 0.27	16.60 ± 0.25	8.55 ± 0.28	10.22 ± 0.08
16	22.34 ± 0.10	20.70 ± 0.33	11.20 ± 0.42	13.60 ± 0.85
17	20.80 ± 0.05	18.80 ± 0.48	10.30 ± 0.82	12.45 ± 0.07
18	23.50 ± 0.50	22.40 ± 0.75	12.90 ± 0.35	14.20 ± 0.15
19	24.70 ± 0.35	24.10 ± 0.52	13.80 ± 0.55	15.30 ± 0.70
23	30.20 ± 0.52	33.80 ± 0.66	22.40 ± 0.25	24.50 ± 0.09
24	25.80 ± 0.15	26.90 ± 0.50	15.30 ± 0.54	18.40 ± 0.35
25	25.40 ± 0.28	25.50 ± 0.32	14.60 ± 0.70	16.70 ± 0.80
30	26.60 ± 0.08	28.60 ± 0.80	16.70 ± 0.40	19.10 ± 0.20
31	29.10 ± 0.35	32.50 ± 0.40	20.10 ± 0.65	23.60 ± 0.10
32	27.30 ± 0.24	29.40 ± 0.70	18.50 ± 0.50	20.80 ± 0.40
33	28.50 ± 0.40	30.10 ± 0.30	19.20 ± 0.80	21.50 ± 0.50
Penicillin G	30.50 ± 0.62	34.25 ± 0.85	29.45 ± 0.20	35.40 ± 0.55
Streptomycin	28 ± 0.30	30 ± 0.50	26 ± 0.30	27 ± 0.46

^a Mean zone of inhibition in mm±standard deviation beyond well diameter (6 mm) produced on a range of environmental and clinically pathogenic microorganisms using (10 mg/mL) concentration of tested samples. The concentration used for the standard antibiotic was (30 μg/mL)

 b the test was done using the diffusion agar technique. Well diameter: 6.0 mm (100 μ L was tested)

^c Data are expressed in the form of mean \pm SD

^d RCMB Regional center for mycology and biotechnology culture collection

^e NA No activity

(iii) Compounds 23 and 30–33 have good antimicrobial activities, which may be attributed to presence of thiophene and thiourea moieties respectively. (iv) The antimicrobial activities of thiourea 30–33 follow the order 31 > 33 > 32 > 30, which may be attributed to the withdrawing effect of phenyl moiety (vi) Compound 16 has good antimicrobial activities than the other Schiff 's bases, which may be attributed to presence of thiophene (Fig. 1).

Conclusions

Table 1 Antibacterial activity

of compounds a, b, c

7-amino-3-phenyl-[1, 2, 4] triazolo [4, 3-a] pyrimidin-5 (1*H*)-one (5) have potential to act as antimicrobial agent by modification of their structure. Introduction of C=N, CONH, CSNH group to these position vanished the antimicrobial activity as well as affecting the solubility. Further

studies on the evaluation of relationship between bioactivities and structure are suggested.

Experimental

Materials and methods

All melting points are in degree centigrade (uncorrected) and were determined on Gallenkamp electric melting point apparatus. Thin layer chromatography (TLC) analysis was carried out on silica gel 60 F254 precoated aluminum sheets. The Infrared (IR) spectra were recorded (KBr) on a Perkin–Elmer 1430 spectrometer (λ , cm-1) in National Research Center, Egypt. ¹H-NMR,\¹³C-NMR Spectra were measured on JEOL-ECA 500 and JEOL JNM-LA-400 FT NMR Spectrometers at 500, 125 MHz, respectively, using tetramethylsilane as an internal reference and DMSO-d6 as solvent at the Microanalytical Center in National Research

Table 2 Antifungal activity of
compounds ^{a, b, c}

Compounds	Fungi				
	Aspergillus fumigatus (RCMB 002003) ^d	Geotrichum candidum (RCMB 052006)	Candida albicans (RCMB 005002)	Syncephalastrum racemosum (RCMB 005003)	
1	NA	NA	NA	NA	
2	NA	7.70 ± 0.60	NA	NA	
3	NA	NA	NA	NA	
4	8.10 ± 0.05	8.50 ± 0.80	7.10 ± 0.40	NA	
5	9.50 ± 0.60	9.80 ± 0.50	8.60 ± 0.50	NA	
13	10.10 ± 0.75	10.50 ± 0.85	9.80 ± 0.44	NA	
14	11.30 ± 0.20	11.10 ± 0.10	10.20 ± 0.30	NA	
15	12.80 ± 0.45	12.20 ± 0.25	11.90 ± 0.06	NA	
16	15.20 ± 0.50	14.80 ± 0.60	13.70 ± 0.05	NA	
17	14.60 ± 0.08	13.40 ± 0.09	12.30 ± 0.15	NA	
18	16.82 ± 0.20	15.50 ± 0.40	14.80 ± 0.33	8.20 ± 0.60	
19	17.40 ± 0.61	16.30 ± 0.81	15.60 ± 0.40	9.30 ± 0.40	
23	25.10 ± 0.45	24.50 ± 0.60	23.20 ± 0.80	18.50 ± 0.10	
24	19.70 ± 0.10	18.60 ± 0.30	18.20 ± 0.70	11.10 ± 0.50	
25	18.20 ± 0.40	17.20 ± 0.55	16.10 ± 0.82	10.40 ± 0.70	
30	20.10 ± 0.09	19.70 ± 0.10	19.10 ± 0.40	12.50 ± 0.60	
31	24.20 ± 0.55	23.10 ± 0.85	22.40 ± 0.60	16.40 ± 0.30	
32	21.50 ± 0.30	20.40 ± 0.20	20.20 ± 0.50	14.60 ± 0.80	
33	22.80 ± 0.60	21.90 ± 0.50	21.50 ± 0.30	15.80 ± 0.20	
Itraconazole	30 ± 0.08	29 ± 0.40	28 ± 0.05	24 ± 0.10	
Clotrimazole	28 ± 0.20	25 ± 0.50	20 ± 0.30	22 ± 0.08	

^a Mean zone of inhibition in mm \pm standard deviation beyond well diameter (6 mm) produced on a range of environmental and clinically pathogenic microorganisms using (10 mg/mL) concentration of tested samples. The concentration used for the standard antibiotic was (30 g/mL)

 b the test was done using the diffusion agar technique. Well diameter: 6.0 mm (100 μL was tested)

^c Data are expressed in the form of mean \pm SD

^d RCMB Regional center for mycology and biotechnology culture collection

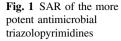
^e NA No activity

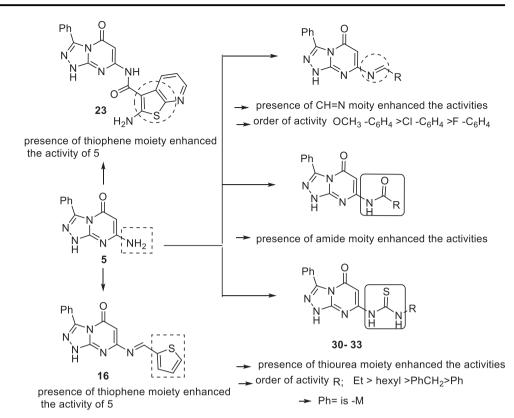
Center, Egypt. The mass spectra (EI) were recorded on GCMS-QP 1000 EX (Shimadzu) at National Research Center, Egypt. Elemental analyses (C, H and N) were carried out at the Microanalytical Center in National Research Center, Egypt. The elemental analyses were found to agree favorably with the calculated values. Biological activities were carried in Regional Center for Mycology and Biotechnology, Al-Azhar University, Nasr City, Cairo, Egypt.

Synthesis of N'-(6-amino-1, 2, 3, 4-tetrahydro-4oxopyrimidin-2-yl) benzo-hydrazide (4)

To a solution of compound 1, (1.43 g, 10 mmol) or 2 (1.57 g, 10 mmol) in DMF (30 ml) and ethanol (25 mL) respectively, benzohydrazide 3 (1.36 g, 10 mmol) was added. The reaction mixture was heated under reflux for 8–10 h, and then allowed to cool, poured into ice/water and

the formed solid product was collected by filtration and crystallized from ethanol/benzene to give compound 4; Yellow crystals, yield 73 % and m.p. >325 °C; IR (KBr): $(\nu/cm^{-1}) = 3360$ (br, NH, NH₂), 3035 (CH–Ar), 1690, 1680 (2CO), 1622 (C=N); 1555 (C=C); ¹H NMR (DMSO-d₆): δ (ppm) = 4.66 (s, 1H, C₅-H), 6. 35 (br, NH₂, D₂O exchangeable) 7.48-7.90 (m, 5H, ArH): 11.51, 11.61 (br, 2NH, D₂O exchangeable), 12.10 (br, NH, D₂O exchangeable); 13 C–NMR (DMSO-d6): δ ppm) = 82.5 (C₅, pyrimidine), 127.5, 128.7, 132.1, 132.4 (6C, Ar-C), 158.6 (C₆, pyrimidine), 161.2 (C₄, pyrimidine), 164.5 (C₂ pyrimidine), 165.1(CONH); MS (EI, 70 eV): m/z (%) = 247 (M⁺+2, 2.1), 245 (M⁺, 6.3), 242 (100), 241 (18.5), 229 (12.6), 196 (34.4), 184 (25.3), 156 (24.2), 125 (17.9), 110 (15.7), 97 (7.4), 68 (15.7), 67 (45.3), 59(11.6). Chemical Formula: C₁₁H₁₁N₅O₂ (245.24); Calcd. C, 53.87; H, 4.52; N, 28.56 %. Found: C, 53.80; H, 4.52; N, 28.56 %.





Synthesis of 7-amino-3-phenyl-[1, 2, 4]triazolo[4,3-a] pyrimidin-5 (1H)-one (5)

A solution of compound 4 (2.45 g, 10 mmol) in ethanol (50 ml) containing sodium ethoxide (prepared by dissolving sodium metal (0.23 g, 10 mmol in ethanol) was heated under reflux for 9 h, the reaction mixture was cooled and the deposited precipitate was filtered off, washed with ethanol. The formed salt was dissolve in water (50 ml) and acidified with 10 % HCl, the formed precipitate was filtered, dried and crystallized from methanol to afford 5; Yellowish crystals, 75 % and m.p. >325 °C; IR (KBr): (ν / cm^{-1}) = 3312, 3352 (br, 2NH, NH₂), 3038 (CH aryl), 1675 (CO), 1620 (C=N), 1558(C=C); ¹H NMR (DMSO-d₆): δ (ppm) = 4.66 (s,1H,C₆-H), 6.34 (br, 2H, NH₂, D₂O exchangeable), 7.49–7.91 (m, 5H, phenyl) and 11.65 (br, 1H, NH, D₂O exchangeable); 13 C–NMR (DMSO-d₆): δ (ppm) = 83.1 (C₆, pyrimidine), 128.1, 128.6, 128.8, 131.2 (6C, Ar-C), 150.8 (C₃, triazole ring); 158.5 (C₇, pyrimidine), 160.4 (C_{8a}, pyrimidine); 162.5 (C₅, pyrimidine); MS (EI, 70 eV): m/z (%) = 229 (M⁺+2, 11.6) 227 (M⁺, 20.2), 203 (27.9), 125 (100), 103 (18.5), 91 (51.1), 77 (37.0), 63 (74.4); Chemical Formula: C₁₁H₉N₅O (227.22); Calcd.: C, 58.14; H, 3.99; N, 30.82 %; Found: C, 58.10; H, 3.95; N, 30.75 %.

General procedure for synthesis of 7-(substitutedarylideneamino)-3-phenyl-[1, 2, 4] triazolo-[4,3-a] pyrimidin-5 (1H)-one (**13–17**)

A mixture of compound (5) (2.27 g, 10 mmol) and 4-fluorobenzaldehyde (6) (1.24 g, 10 mmol), 4-chlorobenzaldehyde (7) (1.40 g, 10 mmol), 4-methoxybenzaldehyde (8) (1.36 g, 10 mmol), isonicotinaldehyde (9) (1.07 g, 10 mmol) or thiophene-2-carbaldehyde (10) (1.2 mL, 10 mmol) in dioxane (40 ml) containing a catalyst amount of piperidine (0.5 mL), was stirred and heated under reflux for 5–10 h (TLC control).The reaction mixture was cooled, the formed precipitate filtered off, dried and recrystallized from the appropriate solvent to afford (13–17).

7-(4-Fluorobenzylideneamino)-3-phenyl-[1, 2, 4]triazolo [4,3-a]pyrimidin-5(1H)-one (**13**)

Yellow crystals, yield 65 %, m.p. 240–242 °C, reaction time 5 h and crystallized from EtOH / benzene; IR (KBr): (ν / cm⁻¹) = 3310 (br, NH), 3030 (CH aryl), 1670 (CO), 1615 (C=N), 1550 (C=C);¹H NMR (DMSO-*d*₆): δ (ppm) = 4.66 (s, 1H, C₆-H), 7.57–7.65 (m, 5H, phenyl), 7.82–7.84 (dd, 2H, *J* = 8.40 Hz, ArH), 7.93–7.95 (dd, 2H, *J* = 8.44 Hz, Ar–H), 8.23 (s, 1H, CH=N), 11.64 (br, NH, D₂O

exchangeable); ¹³C-NMR (DMSO- d_6): δ (ppm) = 112.1 (C₆, pyrimidine), 128.3, 128.7, 128.9, 129.4, 130.5, 131.4, 132.8, 146.2 (12C, Ar–C), 151.3 (C₃, triazole), 158.1 (C₇, pyrimidine), 162.2 (C_{8a}, pyrimidine), 163.5 (CH=N) 165.8 (C₅, pyrimidine); Chemical Formula: C₁₈H₁₂FN₅O (333.32); Calcd.: C, 64.86; H, 3.63; N, 21.01 %; Found: C, 64.80; H, 3.60; N, 21.08 %.

7-(4-Chlorobenzylideneamino)-3-phenyl-[1, 2, 4]triazolo [4,3-a]pyrimidin-5(1H)-one (**14**)

Brown powder, yield 70 %, m.p. 290–292 °C, reaction time 7 h and crystallized from MeOH; IR (KBr): (ν/cm^{-1}) = 3312 (br, NH), 3025 (CH aryl), 1672 (CO), 1618 (C=N), 1552 (C=C); ¹H NMR (DMSO-*d₆*): δ (ppm) = 4.67 (s, 1H, C₆-H), 7.56–7.64 (m, 5H, phenyl), 7.83–7.85 (dd, 2H, *J* = 8.30 Hz, ArH), 7.94–7.96 (dd, 2H, *J* = 8.33 Hz, ArH), 8.23 (s, 1H, CH=N) and 11.65 (br, NH, D₂O exchangeable); ¹³C-NMR (DMSO-*d₆*): δ (ppm) = 112.1 (C₆, pyrimidine), 128.2, 128.6, 128.8, 129.1, 130.1, 130.8, 132.1, 136.8 (12C, Ar–C), 151.6 (C₃, triazole), 157.8 (C₇, pyrimidine), 161.1(C_{8a}, pyrimidine), 162.7 (CH=N), 163.1 (C₅, pyrimidine); Chemical Formula: C₁₈H₁₂ClN₅O (349.77); Calcd.: C, 61.81; H, 3.46; N, 20.02 %; Found: C, 61.85; H, 3.40; N, 20.10 %.

7-(4-Methoxybenzylideneamino)-3-phenyl-[1,2,4]triazolo [4,3-a]pyrimidin-5(1H)-one(15)

White crystals, yield 64 %, m. P. 260-262 °C, reaction time 6 h and crystallized from EtOH / benzene; IR (KBr): (ν / cm^{-1}) = 3315 (br, NH), 3032 (CH aryl), 1671 (CO), 1619 (C=N), 1551(C=C); ¹H NMR (DMSO- d_6): δ (ppm) = 3.85 (s, 3H, OCH₃), 4.67 (s, 1H, C₆-H), 7.55–7.63 (m, 5H, phenyl), 7.81–7.83 (dd, 2H, J = 8.36 Hz, Ar–H), 7.92–7.94 (dd, 2H, J = 8.39 Hz, Ar-H), 8.24 (s, 1H, CH=N), 11.66 (br, NH, D_2O exchangeable); ¹³C-NMR (DMSO-d₆): δ (ppm) = 55.9 (OCH₃), 112.2 (C₆, pyrimidine), 124.5, 126.5, 128.4, 128.8, 129.5, 130.8, 131.6, 149.1 (12C, Ar-C), 151.5 (C₃, triazole), 158.4 (C₇, pyrimidine), 162.8 (C_{8a}, pyrimidine), 163.7 (CH=N). 165.1 (C₅, pyrimidine); MS (EI, 70 eV): m/z (%) = 345(M⁺, 44.4), 270 (22.1), 230 (33.2), 195(100), 139 (66.4), 59(88.8); Chemical Formula: C₁₉H₁₅N₅O₂ (345.35); Calcd.; C, 66.08; H, 4.38; N, 20.28 Found: C, 66.10; H, 4.38; N, 20.32 %.

3-Phenyl-7-(thiophen-2-ylmethyleneamino)-[1,2,4]triazolo [4,3-a]pyrimidin-5(1H)-one (**16**)

Yellowish crystals, yield 85 %; m. P. 226–228 °C, reaction time 10 h and crystallized from MeOH; IR (KBr): (ν /cm⁻¹) = 3320 (br, NH), 3030 (CH aryl), 1680 (CO), 1618 (C=N),1555 (C=C); ¹H NMR (DMSO-*d*₆): δ (ppm) = 4.67 (C₆-H, pyrimidine), 7.25 (t, 1H, J = 2.20 Hz, thiophene ring), 7.34–7.42 (dd, 2H, J = 2.24 Hz, thiophene ring), 7.52–7.88 (m, 5H, phenyl), 8.24 (s, 1H, CH=N) and 11.68 (br, NH, D₂O exchangeable); ¹³C-NMR (DMSO-d₆): δ (ppm) = 112.5 (C₆, pyrimidine), 127.4, 128.2, 128.6, 128.7, 128.8, 130.1, 130.3, 144.5 (10 C, aromatic carbon atoms), 150.8 (C₃, triazole), 158.8 (C₇, pyrimidine), 162.5 (C_{8a}, pyrimidine),163.6 (CH=N) and 165.2 (C₅, pyrimidine); Chemical formula: C₁₆H₁₁N₅OS (321.36)); Calcd.: C, 59.80; H, 3.45; N, 21.79 %; Found: C, 59.75; H, 3.40; N, 21.70 %.

3-Phenyl-7-(pyridin-4-ylmethyleneamino)-[1, 2, 4]triazolo [4,3-a]pyrimidin-5(1H)-one (17)

White powder in 75 % yield, m.p. $302-304 \,^{\circ}$ C, reaction time 8 h and crystallized from MeOH; IR (KBr): (ν/cm^{-1}) = 3317 (br, NH), 3028 (CH aryl), 1677 (CO), 1615 (C=N) and 1550 (C=C); ¹H NMR (DMSO-*d*₆): δ (ppm) = 4.68 (C₆-H, pyrimidine), 7.04–7.12 (dd, 2*H*, *J* = 8.10 Hz, pyridine ring), 7.25–7.33 (dd, 2*H*, *J* = 8.14 Hz, pyridine ring), 7.50–7.85 (m, 5H, phenyl), 8.25 (s, 1H, CH=N); 11.65 (br, NH, D₂O exchangeable); ¹³C-NMR (DMSO-*d*₆): δ (ppm) = 111.9 (C₆, pyrimidine), 114.5, 116.8, 118.9, 124.4, 127.2, 131.5, 137.7 (11C, Ar–C), 149.6 (C₃, triazole), 158.7 (C₇, pyrimidine), 159.1(C_{8a}, pyrimidine), 159.3 (CH=N) 159.9 (C₅, pyrimidine); Chemical formula: C₁₇H₁₂N₆O (316.32); Calcd.: C, 64.55; H, 3.82; N, 26.57 %; Found: C, 64.50; H, 3.78; N, 26.50 %.

General procedure for synthesis of 7-((5-chloro-3-methyl-1phenyl-1H-pyrazol-4-yl)methyleneamino)-3-phenyl-[1,2,4] triazolo [4,3-a]pyrimidin-5(1H)-one (18) and 7-((2-Azido-6-methylquinolin-3-yl)methyleneamino)-3-phenyl[1,2,4] triazolo[4,3-a] pyrimidin-5(1H)-one (19)

To a suspension of compound **5** (2.27 g, 10 mmol) in glacial acetic acid (50 mL) or in mixture of absolute ethanol (35 mL) containing glacial acetic acid (1 mL), 5-chloro-3-methyl-1-phenyl-1*H*-pyrazole-4-carbaldehyde (**16**) (2.20 g, 10 mmol) or ethyl 2-azido-6-methylquino-line-3-carbaldehyde (**17**) (2.12 g, 10 mmol) was added, respectively. The reaction mixture was stirred and heated under reflux for 10 and 15 h, respectively. The reaction mixture was cooled, the formed precipitate filtered off, dried and recrystallized from the appropriate solvent to afford **18** and **19**.

7-((5-Chloro-3-methyl-1-phenyl-1H-pyrazol-4-yl) methyleneamino)-3-phenyl-[1, 2, 4] triazolo [4, 3-a] pyrimidin-5(1H)-one (**18**)

Brown crystals, yield 60 %, m. P. 340–342 °C, reaction time 6 h and crystallized from EtOH; IR (KBr): $(\nu/cm^{-1}) = 3322$

(br, NH), 3035(CH aryl), 2995(CH aliph.), 1685 (CO), 1615 (C=N), 1552 (C=C); ¹H NMR (DMSO- d_6): δ (ppm) = 2.35 (s, 3H, CH₃), 4.66 (s, 1H, C₆-H, pyrimidine), 7.21–7.92 (m,10H, Ar–H), 8.23 (s, 1H, CH=N), 11.69 (br, NH, D₂O exchangeable); ¹³C-NMR (DMSO- d_6): δ (ppm) = 15.2 (CH₃), 112.4 (C₆, pyrimidine), 116.8, 121.2, 128.3, 128.5, 128.9, 129.8, 130.3, 143.6 (12C, Ar–C), 144.5, 146.1, 149 (3C, pyrazole ring), 151.1 (C₃, triazole), 158.4 (C₇, pyrimidine), 162.5 (C_{8a}, pyrimidine), 163.7(CH=N), 165.9 (C₅, pyrimidine); MS (EI, 70 eV): m/z (%) = 430 (M⁺, 35.7), 384 (100), 342 (21.4), 125 (35.7), 60 (42.8); Chemical Formula: C₂₂H₁₆ClN₇O (429.87)); Calcd.: C, 61.47; H, 3.75; N, 22.81 %; Found: C, 61.55; H, 3.70; N, 22.76 %.

7-((2-Azido-6-methylquinolin-3-yl)methyleneamino)-3phenyl[1,2,4]triazolo[4,3-a] pyrimidin-5(1H)-one (19)

White crystals, 55 % yield, m.p. >350 °C (melted) and crystallized from benzene; IR (KBr): $(\nu/cm^{-1}) = 3320$ (br, NH), 3030(CH aryl), 2992 (CH aliph.), 1682 (CO), 1617 (C=N), 1551 (C=C); ¹H NMR (DMSO- d_6): δ (ppm) = 2.35 (s, 3H, CH₃), 4.68 (C₆-H, pyrimidine), 7.45-7.52 (dd,2H, J = 8.55 Hz, quinoline ring), 7.60 (s,1H, quinoline ring), 7.64-7.94 (m,5H, phenyl), 8.25 (s, 1H, CH=N); 11.64 (br, NH, D₂O exchangeable); ¹³C-NMR (DMSO- d_6): δ (ppm) = 38.5 (1C, CH₃), 81.9 (C₆, pyrimidine), 121.1, 124.8, 126.2, 128.5, 128.6, 128.7, 129.2, 130.2, 131.5, 135.6, 136.5, 146.5, 147.1 (15C, Ar-C), 151.5 (C₃, triazole), 161.1 (C7, pyrimidine), 162.4 (C8a, pyrimidine), 163.5 (CH=N) 169.1 (C₅, pyrimidine); MS (EI, 70 eV): *m/z* (%) $= 423 (M^{+}+2, 98.0), 421 (M^{+}, 100), 379 (20.0), 353 (27.3),$ 325 (15.5), 307(19.1), 274 (11.8), 213 (20.1), 121 (19.0), 77 (22.7); Chemical Formula: C₂₂H₁₅N₉O (421.42); Calcd.: C, 62.70; H, 3.59; N, 29.91 %; Found: C, 62.65; H, 3.50; N, 29.85 %.

General procedure for synthesis of 2-amino-N-(1,5dihydro-5-oxo-3-phenyl-[1,2,4] triazolo [4,3-a]pyrimidin-7-yl)benzo[b]thiophene-3-carboxamide (23) 3-(Benzofuran-2-yl)-3-oxo-N-(5-Oxo-3-phenyl-1,5-dihydro-[1,2,4]triazolo [4,3-a]pyrimidin-7-yl)propanamide (24) and 4,5, 6,7tetrabromo-2-(1,5-dihydro-5-oxo-3-phenyl-[1,2,4]triazolo [4,3-a]pyrimidin-7-yl)isoindoli ne-1,3-dione(25)

A mixture of **5** (2.27 g, 10 mmol) and ethyl 3-aminothieno [2, 3-b] pyridine-2-carboxylate (**20**), (2.22 g, 10 mmol), ethyl 4-(2,3-dihydrobenzofuran-2-yl)-2,4-dioxobutanoate (**21**) (2.62 g,10 mmol) or 4,5,6,7-tetrabromoisobenzofuran-1,3-dione (**22**), (4.63 g,10 mmol) in mixture of absolute ethanol (40 mL) containing glacial acetic acid (1 mL) was refluxed for 14–16 h. The reaction mixture was allowed to

cool to room temperature, filtered off, washed with ethanol to give **23–25**.

2-Amino-N-(1,5-dihydro-5-oxo-3-phenyl-[1,2,4]triazolo [4,3-a]pyrimidin-7-yl)benzo[b] thiophene-3-carboxamide (23)

Yellowish crystals, yield 70 %, m.p. >350 °C, reaction time 6 h and crystallized from dioxane; IR (KBr): $(\nu/cm^{-1}) =$ 3380 (br, 2NH, NH₂), 3040 (CH aryl), 1695, 1684 (2CO), 1625 (C=N); 1556 (C=C); ¹H NMR (DMSO- d_6): δ (ppm) = 4.68 (s, 1H, C₆-H, pyrimidine), 6. 35 (br, NH₂, D₂O exchangeable), 7.12 (t,1H, J = 8.20 Hz, pyridine ring), 7.15 (d, 1H, J = 8.22 Hz, pyridine ring), 7.18 (d, 1H, J = 8.23 Hz,pyridine ring), 7.48-7.91 (m,5H, phenyl), 11.51 (br, NH, D₂O exchangeable); 11.61 (br, NH, D₂O exchangeable); ¹³C-NMR (DMSO- d_6): δ (ppm) = 90.4 (C₆, pyrimidine), 121.4, 128.2, 128.4, 128.8, 129.1, 129.6, 130.1, 131.4, 137.5, 145.5, 152.4 (14C, Arc), 151.5 (C₃, triazole), 157.1 (C₇, pyrimidine), 161.3 (C_{8a}, pyrimidine), 164.5 (C₅, pyrimidine), 165.8 (CONH); MS (EI, 70 eV): m/z (%) = 405 (M⁺+2, 29.3), 403 (M⁺, 33.5), 366 (26.8), 342 (19.5), 302 (21.9), 279 (9.7), 220(36.6), 153(9.7), 125 (100), 115 (9.7), 63 (4.8); Chemical formula: $C_{19}H_{13}N_7O_2S$ (403.42)); Calcd.: C, 56.50; H, 3.20; N, 24.26 %; Found: C, 56.55; H, 3.16; N, 24.20 %.

3-(Benzofuran-2-yl)-3-oxo-N-(5-oxo-3-phenyl-1,5-dihydro-[1,2,4]triazolo[4,3-a]pyrimidin-7-yl)propanamide (24)

Yellow crystals, yield 72 %, m.P.162-164 °C, reaction time 10 h and crystallized from MeOH; IR (KBr): $(\nu/cm^{-1}) =$ 3395 (br, 2NH), 3045 (CH aryl), 2994 (CH aliph.), 1740, 1730, 1694, 1685 (4CO), 1624 (C=N),1552 (C=C); ¹H NMR (DMSO- d_6): δ (ppm) = 4.15 (s, 2H, CH₂), 4.67 (s, 1H, C₅-H, pyrimidine), 5.15 (s, 1H, benzofuran), 7.01–7.25 (m,4H, benzofuran), 7.50-7.90 (m, 5H, phenyl), 11.52 (br, NH, D₂O exchangeable); 11.62 (br, NH, D₂O exchangeable); ¹³C-NMR (DMSO- d_6): δ (ppm) = 62.8 (CH₂), 90.4 (C₆, pyrimidine), 114.2, 121.1, 125.2, 127.3, 128.4, 128.6, 128.8,129.9,130.3, 151.2, 152.1 (13C, Ar-C), 154.2 (C₃, triazole), 158.5 (C₇, pyrimidine), 160.5 (C_{8a}, pyrimidine), 163.6 (C₅, pyrimidine), 165.8 (CONH) 190.1 (COCO), 198.8 (CH₂CO); MS (EI, 70 eV): m/z (%) = 413 (M⁺, 45.5), 414 (12.7), 415 (9.1), Chemical formula: C₂₂H₁₅N₅O₄ (413.39); Calcd.: C, 63.92; H, 3.66; N, 16.94 %; Found: C, 63.85; H, 3.75; N, 16.82 %.

4,5,6,7-Tetrabromo-2-(1,5-dihydro-5-oxo-3-phenyl-[1,2,4] triazolo[4,3-a]pyrimidin-7-yl) isoindoline-1,3-dione (**25**)

Yellowish crystals in 62 % yield, m.p. 232–234 °C, reaction time 14 h and crystallized from DMF; IR (KBr):

 $(\nu/cm^{-1}) = 3332$ (br, NH), 3042 (CH aryl), 1720, 1702, 1682 (3CO), 1622 (C=N), 1554(C=C); ¹H NMR (DMSO*d*₆): δ (ppm) = 4.68 (s, 1*H*, C₅-H, pyrimidine), 7.48–7.90 (m, 5H, phenyl), 11.66 (br, NH, D₂O exchangeable); ¹³C-NMR (DMSO-*d*₆): δ (ppm) = 90.2 (C₆, pyrimidine), 125.1, 128.2, 128.5, 128.9, 129.8, 130.2, 139.4 (12C, Ar–C) 150.5 (C₃, triazole), 153.4 (C₇, pyrimidine), 156.8 (C_{8a}, pyrimidine), 162.9 (C₅, pyrimidine), 166.2 (2CON); Chemical formula: C₁₉H₇Br₄N₅O₃ (672.91)); Calcd.: C, 33.91; H, 1.05; N, 10.41 %; Found: C, 33.85; H, 1.10; N, 10.50 %.

General procedure for synthesis of 1-(5-oxo-3-phenyl-1,5-dihydro-[1,2,4] triazolo [4,3-a] pyrimidin-7-yl)-3-phenyl thiourea (**30**), 1-benzyl-3-(5-oxo-3-phenyl-1,5-dihydro-[1,2,4] tria zolo[4,3-a]pyrimidin-7-yl)thiourea (**31**), 1-ethyl-3-(5-oxo-3-phenyl-1,5-dihydro-[1,2,4] tria zolo[4,3-a] pyrimidin-7-yl)thiourea (**32**), 1-cyclohexyl-3-(5-oxo-3-phenyl-1, 5-dihydro-[1, 2,4] triazolo[4,3-a]pyrimidin-7-yl)thiourea (**33**); A mixture of **5** (2.27 g, 10 mmol) and the appropriate phenylisothiocyanate (**26**), (1.35 g, 10 mmol), benzylisothiocyanate (**27**), (1.49 g, 10 mmol), ethylisothoicyanate (**28**), (0.87 g, 10 mmol) or cyclohexylisothiocyanate (**29**) (1.41 g, 10 mmol) in absolute ethanol (30 ml) was refluxed for 2–4 h (TLC) and then left to cool. The precipitate which formed was filtered off, dried and crystallized from the appropriate solvent to give (**30–33**).

1-(5-Oxo-3-phenyl-1,5-dihydro-[1,2,4] triazolo [4,3-a] pyrimidin-7-yl)-3-phenyl thiourea (**30**)

White crystals, yield 66 %; m.p. 206–208 °C, reaction time 2 h and crystallized from MeOH; IR (KBr): $(\nu/cm^{-1}) =$ 3335, 3255, 3220 (3NH), 3040 (CH aryl), 1680 (CO), 1355 (C=S), 1621 (C=N), 1552 (C=C); ¹H NMR (DMSO-*d₆*): δ (ppm) = 4.67 (s, 1H, C₅-H, pyrimidine), 7.01–7.90 (m,10H, Ar–H), 11.51 (br, NH, D₂O exchangeable), 11.62 (br, NH, D₂O exchangeable), 12.10 (br, NH, D₂O exchangeable); ¹³C-NMR (DMSO-*d₆*): δ (ppm) = 92.5 (C₆, pyrimidine), 126.4, 128.1, 128.2, 128.5, 128.9, 129.2, 130.5, 138.8 (12 C, Ar–C), 150.2 (C₃, triazole), 153.2 (C₇, pyrimidine), 157.4 (C_{8a}, pyrimidine), 163.5 (C₅, pyrimidine), 176.5 (C=S); Its MS: [M⁺], m/z 362 (90 %); Chemical formula: C₁₈H₁₄N₆OS (362.41); Calcd.: C, 59.65; H, 3.89; N, 23.19 %; Found: C, 59.60; H, 3.80; N, 23.14 %.

1-Benzyl-3-(5-oxo-3-phenyl-1, 5-dihydro-[1, 2, 4]triazolo [4,3-a]pyrimidin-7-yl)thiourea (**31**)

Yellowish crystals, yield 60 %, m.p. 174–176 °C, reaction time 4 h and crystallized from acetone; IR (KBr): (ν /cm⁻¹) = 3334, 3254, 3221 (3NH), 3035 (CH aryl), 2985 (CH aliph), 1682 (CO), 1352 (C=S), 1622 (C=N), 1551 (C=C); ¹H NMR (DMSO- d_6): δ (ppm) = 4.66 (s, 1*H*, C₅-H, pyrimidine), 4.82 (d, 2*H*, CH₂-Ph), 7.02–7.95 (m, 10*H*, ArH), 11.50 (br, NH, D₂O exchangeable), 11.60 (br, NH, D₂O exchangeable), 12.12 (br, NH, D₂O exchangeable); ¹³C-NMR (DMSO-*d*₆): δ (ppm) = 60.1 (1C, CH₂), 95.1 (C₆, pyrimidine), 126.6,126.9,128.2, 128.4, 128.6,128.8, 130.2,138.1 (12C, Ar–C), 151.5 (C₃, triazole), 154.4 (C₇, pyrimidine), 158.2 (C_{8a}, pyrimidine), 164.1 (C₅, pyrimidine), 180.8 (C=S); MS (EI, 70 eV): *m*/*z* (%) = 378 (M⁺+2, 1.3) 366 (M⁺, 10.5), 253 (3.9), 239 (3.4), 112 (13.6), 57 (100); Chemical formula: C₁₉H₁₆N₆OS (376.43); Calcd.: C, 60.62; H, 4.28; N, 22.33 %; Found: C, 60.55; H, 4.20; N, 22.27 %.

1-Ethyl-3-(5-oxo-3-phenyl-1, 5-dihydro-[1, 2, 4]triazolo [4,3-a] pyrimidin-7-yl)thiourea (**32**)

White crystals in 70 % yield, m.p. 190-192 °C, reaction time 2 h and crystallized from dioxane: IR (KBr): (ν/cm^{-1}) = 3330, 3250, 3220 (3NH), 3040 (CH aryl), 2975 (CH aliph), 1684 (CO),1355 (C=S), 1618 (C=N), 1550 (C=C); ¹H NMR (DMSO- d_6): δ (ppm) = 1.32 (t, 3H, CH₃), 4.45 (q, 2H, CH₂), 4.67 (s, 1H, C₅-H, pyrimidine), 7.52-7.92 (m, 5H, phenyl), 11.52 (br, NH, D₂O exchangeable), 11.61 (br, NH, D₂O exchangeable), 12.11 (br, NH, D₂O exchangeable); ¹³C-NMR (DMSO- d_6): δ (ppm) = 18.5 (CH₃), 55.5 (CH₂), 91.2(C₆, pyrimidine), 128.3, 128.5, 128.9, 130.4 (6C, Ar-C), 150.2 (C₃, triazole), 158.5 (C₇, pyrimidine), 159.5 (C_{8a}, pyrimidine), 165.5 (C₅, pyrimidine), 185.5 (C=S); MS (EI, 70 eV): m/z (%) = 314 (M⁺, 30.1), 239 (45.7), 165 (40.0), 125 (85.1), 55(100); Chemical formula: C₁₄H₁₄N₆OS (314.37); Calcd.: C, 53.49; H, 4.49; N, 26.73 %; Found: C, 53.40; H, 4.40; N, 26.65 %.

1-Cyclohexyl-3-(5-oxo-3-phenyl-1,5-dihydro-[1,2,4] triazolo[4,3-a]pyrimidin-7-yl)thiourea (**33**)

Yellowish crystals, yield 61 %, m.p. 217-219 °C, reaction time 4 h and crystallized from methanol; IR (KBr): (ν / cm^{-1}) = 3340, 3258, 3230 (3NH), 3035 (CH aryl), 2970 (CH aliph.), 1682 (CO), 1358(C=S), 1615 (C=N), 1554 (C=C);¹H NMR (DMSO-*d*₆): δ (ppm) = 1.21–1.75 (m, 10H, cyclohexyl), 2.65 (m, 1H, cyclohexyl), 4.68 (s, 1H, C₅-H, pyrimidine),7.50–7.90 (m, 5H, phenyl), 11.53 (br, NH, D₂O exchangeable), 11.64 (br, NH, D₂O exchangeable), 12.13 (br, NH, D₂O exchangeable); ¹³C-NMR (DMSO-d6, 8, ppm); 25.5, 25.8, 32.8, 55.1(6C, cyclohexyl), 94.1(C₆, pyrimidine), 128.2, 128.6, 128.8, 130.2 (6C, Ar-C), 149.8 (C₃, triazole), 155.2 (C₇, pyrimidine), 161.5 (C_{8a}, pyrimidine), 166.2 (C₅, pyrimidine), 184.2 (C=S); Chemical formula: C₁₈H₂₀N₆OS (368.46); Calcd.: C, 58.68; H, 5.47; N, 22.81 %; Found: C, 58.60; H, 5.41; N, 22.75 %.

Microbiology

Evaluation of in vitro antibacterial and antifungal activity

The tested compounds were evaluated by the agar diffusion technique (Cruickshank et al. 1975) using a 2 mg mL^{-1} solution in DMSO. The test organisms were four bacterial strains: Staphylococcus aureus and Bacillus subtilis (as Gram positive bacteria) and Pseudomonas aeruginosa, Escherichia coli and Salmonella typhi (as Gram negative bacteria). And two fungi: Aspergillus fumigatus, Geotrichum candidum, Candida albicans, Syncephalastrum racemosum. A control using DMSO without the test compound was included for each organism. Penicillin G, and Streptomycin were purchased form the Egyptian market and used in a concentration 10 mg/mL as reference drugs for antibacterial activity, whereas, Itraconazole, Clotrimazole were used as reference drugs for antifungal activity. The bacteria and fungi were tested on nutrient agar and potato dextrose agar media, respectively. Three plates were used for each compound as replicates. The plates were incubated for 24 h and seven days for bacteria and fungi, respectively. After the incubation period, the diameter of inhibition

zone was measured as an indicator for the activity of the compounds.

Acknowledgments We deeply appreciate the Regional Center for Mycology and Biotechnology, Al-Azhar University, Nasr City, Cairo, Egypt, for carrying out the biological activity tests. The presented work was supported by the Department of Photochemistry (Heterocyclic unit); Chemical Industries Research Division, National Research Centre in Cairo, Egypt.

Compliance with ethical standards

Conflict of interest The authors declare that they have no competing interests.

References

- Abdel-Latif NA, Sabry NM, Mohamed AM, Abdulla MM (2007) Synthesis, analgesic and anti-parkinsonism profiles of some pyridine pyrazoline and thinopyrimidine derivatives. Monatsh Chem 138:715–724
- Abu-Hashem AA, Youssef MM, Hussein HAR (2011) Synthesis, antioxidant, antituomer activities of some new thiazolopyrimidines, pyrrolothiazolopyrimidines and triazolopyrrolothiazolopyrimidines derivatives. J Chinese Chem Soc 58:41–48
- Abu-Hashem AA, Aly AS (2012) Synthesis of new pyrazole, triazole, and thiazolidine-pyrimido[4, 5-b] quinoline derivatives with potential antitumor activity. Arch Pharm Res 35:437–445
- Abu-Hashem AA, Abu-Zied KM, El-Shehry MF (2011) Synthetic utility of bifunctional thiophene derivatives and antimicrobial evaluation of the newly synthesized agents. Monatshefte fur Chemie-Chemical Monthly 142:539–545

- Aly AA, Ramadan M, Mohamed AM, Eshak EA (2012) Thieno[2,3-d] pyrimidines in the synthesis of new fused heterocyclic compounds of prospective antitumor and antioxidant agents. J Heterocycl Chem 49:1009–1018
- Broom D, Shim JL, Anderson GL (1976) Synthetic studies leading to various oxopyrido[2,3-d]pyrimidines. J Org Chem 41: 1095–1099
- Chen Q, Zhu XL, Jiang LL, Liu ZM, Yang GF (2008) Synthesis, antifungal activity and Co MFA analysis of novel 1, 2, 4-triazolo [1, 5-a] pyrimidine derivatives. Eur J Med Chem 43:595–603
- Cruickshank R, Duguid JP, Marion BP, Swain RHA (1975) In: Medicinal microbiology. 12th edn. Churchill Livingstone, London II, pp 196–202
- Degraw JI, Kisliuk RL, Gaumont Y, Baugh CM (1974) Antimicrobial activity of 8-deazafolic acid. J Med Chem 17:470–471
- Deyanov B, Niyazov RK, Nazmetdivov FY, Syropyatov BY, Kolla VE, Konshin ME (1991) Synthesis and biological activity of amides and nitriles of 2- arylamino- 5-carboxy (carbethoxy)-6methylnicotinic acids and 1- aryl-6-carbethoxy-7-methyl-4-oxo-1, 4-dihydropyrido [2, 3-d]pyrimidines. J Pharm Chem 25:248–250
- Donkor CL, Klein L, Liang N, Zhu E, Bradley AM, Clark AM (1995) Synthesis and antimicrobial activity of 6, 7- annulated pyrido [2, 3-d] pyrimidines. J Pharm Sci 84:661–664
- El-Gendy MMA, Shaaban M, Shaaban KA, El-Bondkly AM, Laatsch H (2008) Essramycin: a first triazolopyrimidine antibiotic isolated from nature. J Antibiotics 61:149–157
- Ellingboe JW, Princeton NJ (1996) Substituted pyridopyrimidines and antihypertensives. Chemical Abstracts 124:176134q
- Fischer G (2008) Recent progress in 1, 2, 4- Triazolo[1, 5-a]pyrimidine chemistry. Adv Heterocycl Chem 95:143–219
- Grivsky EM, Lee S, Sigel CW, Duch DS, Nichol CA (1980) Synthesis and antitumor activity of 2, 4-diamino-6-(2, 5-dimethoxybenzyl)-5-methylpyrido [2, 3-d] pyrimidine. J Med Chem 23:327–329
- Guiney D, Gibson CL, Suckling CJ (2003) Syntheses of highly functionalized 6- substituted pteridines. Org Biomol Chem 1:664–675
- Hussein HAR, Abu-Zied KHM, Abu-Hashem AA (2010) Novel synthesis of fused pyrazolopyrimidines and C-nucleosides of thienopyrimidone with expected antimicrobial activity. Egypt J Chem 53:527–540
- Khidre RE, Abu-Hashem AA, El-Shazly M (2011) Synthesis and antimicrobial activity of some 1- substituted amino- 4, 6-dimethyl-2oxo-pyridine-3-carbonitrile derivatives. Eur J Med Chem 46:5057–5064
- Magan R, Marin C, Salas JM, Perez MB, Rosales M (2004) Cytotoxicity of three new triazolo-pyrimidine derivatives against the plant trypanosomatid: Phytomonas sp. isolated from Euphorbia characias. J Mem Inst Oswaldo Cruz 99:651–656
- Magan R, Marin C, Rosales MJ, Salas JM, Sánchez-Moreno M (2005) Therapeutic potential of new Pt (II) and Ru (III) triazolepyrimidine complexes against Leishmania donovani. Pharmacology 73:41–48
- Mohamed AM, Alsharari MA, Ali AA (2010) Rapid and facile synthesis of spiro [indole-3, 3' [1, 2, 4] triazol]-2(1*H*)-ones. J Chem Res 34(3):200–202
- Nargund LVG, Reddy YSR, Jose R (1991) Synthesis and antibacterial activity of pyrido[1, 2-a] pyrimidin-4 (1*H*) ones. Ind Drugs 29:45–46
- Navarro JAR, Salas JM, Romero MA, Vilaplana R, Faure R (1998) Cis-[PtCl₂(4,7-H-5-methyl-7-oxo[1,2,4]triazolo[1,5-a]pyrimidine)2]: A sterically restrictive new cisplatin analogue. Reaction kinetics with model nucleobases, DNA interaction studies, antitumor activity, and structure-activity relationships. J Med Chem 41:332–338

- Pandey S, Suryawanshi SN, Gupta S, Srivastava VML (2004) Synthesis and ant leishmanial profile of some novel terpenyl pyramid. Eur J Med Chem 39:969–973
- Salas JM, Romero MA, Sánchez MP, Quirós MA (1999) Metal complexes of 1, 2, 4- triazolo[1, 5-a] pyrimidine derivatives. Coord Chem Rev 193:1119–1142
- Sato Y, Shimoji Y, Fujita H, Nishino H, Mizuno H, Kobayashi S, Kumakura S (1980) Synthesis and coronary vasodilating and

antihypertensive activities of 1, 2, 4-triazolo [1, 5-a] pyrimidines fused to heterocyclic systems. J Med Chem 23:927–937

Traxler PM, Furet P, Mett H, Buchdunger E, Meyer T, Lydon NJ (1996) 4-(Phenylamino) pyrrolopyrimidines: Potent and selective, ATP site directed inhibitors of the EGF-receptor protein tyrosine kinase. J Med Chem 39:2285–2292