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# Intramolecular aglycon delivery for $(1 \rightarrow 2)$ - $\beta$ -mannosylation: towards the synthesis of phospholipomannan of Candida albicans



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Candida albicans, the most widespread human pathogen responsible for cutaneous and systemic fungal infections, synthesize a unique sequence of  $(1 \rightarrow 2)$ - $\beta$ -oligomannans on its cell surface that act as adhesin, induce production of pro-inflammatory cytokines and antibodies. One of the key  $(1 \rightarrow 2)$ - $\beta$ -mannan expressed by *C*. albicans is the membrane-anchored glucosphingolipid (GSL), named phospholipomannan<sup>1</sup> (PLM, Fig. 1) where oligomeric  $(1 \rightarrow 2)$ - $\beta$ -mannan domain is linked through a unique  $\alpha$ -1,2-mannosyl anomeric phosphodiester moiety to a mannose-inositolphosphoceramide (MIPC) glycolipid anchor, which is embedded in the cell wall of the C. albicans. The PLM of Candida albicans is highly immunogenic due to its  $(1 \rightarrow 2)$ - $\beta$ -mannan structural motif, which is absent in human host. When C. albicans infects and interacts with the macrophages (the first-line-of-defence cells of the human host), a large amount of soluble PLM fragments (largely  $(1 \rightarrow 2)$ - $\beta$ -mannans) are rapidly shed by the pathogen resulting in severe pro-inflammatory response<sup>2</sup> (TNF- $\alpha$  production and release) from host cells.<sup>3</sup> Several biochemical<sup>4</sup> and immunological studies<sup>5</sup> have shown that PLM motif provides a scaffold for anchoring of various virulence factors to the cell surface of the pathogen, a mechanism quite similar to that used by the mammalian cells for anchoring various cell-surface proteins and glycans through the glycosylphosphatidylinositol<sup>6</sup> (GPI) anchor. A comparison of the structures of PLM of C. albicans (pathogen) and GPI anchor of

## ABSTRACT

A high yielding method for 1,2-*cis*-β-D-mannosylation by intra-molecular aglycon delivery (IAD) through p-methoxy benzyl ether/acetal exchange and phenylsulfoxide donor is reported, along with its application in iterative assembly of antigenic  $(1 \rightarrow 2)$ - $\beta$ -pentamannoside domain of phospholipomannan (PLM) of fungal pathogen Candida albicans.

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human cells (host) reveals remarkable biomimetic features (Fig. 1). These include: (a)  $(1 \rightarrow 2)$ - $\beta$ -mannan in PLM in place of  $(1 \rightarrow 2)$ - $\alpha$ -mannan motif in GPI, (b)  $(1 \rightarrow 2)$ - $\alpha$ -mannose linked to *myo*-inositol in place of  $(1 \rightarrow 6)$ - $\alpha$  glucosamine-inositol motif in GPI; and c) presence of phytoceramide in PLM and glycerolipid in GPI. Perhaps, these remarkable bio-mimetic features represent mechanisms of molecular and evolutionary adaptations between the species. In continuation to our long-term interest in chemistry



Figure 1. Phospholipomannan of C. albicans and GPI anchor of H. sapiens.

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and biology of GPI anchors,<sup>7</sup> we initiated the synthesis of PLM of *C. albicans*, which present substantial opportunities.

Arguably, one of the most challenging problems in carbohydrate chemistry is the construction of thermodynamically unstable  $(1 \rightarrow 2)$ - $\beta$ -mannoside bond, ubiquitous in several pathogenic microorganisms. First synthesis of  $\beta$ -oligomannosides was reported by Hindsgaul,<sup>8a</sup> and Stork<sup>8b,c</sup> followed by Ogawa and Ito<sup>9</sup> using intramolecular aglycon delivery (IAD) and temporary tethering concept. Later on, a number of other leading groups designed  $(1 \rightarrow 2)$ - $\beta$ -glycosidations via intermolecular glycosylation using orthogonal functional groups.<sup>10</sup>

Here, we report the first application of intramolecular glycosylation approach by linkage of the accepting atom to the donor via a bifunctional group (intramolecular aglycon delivery) for the synthesis of  $(1 \rightarrow 2)$ - $\beta$ -mannosides. Previously, Crich<sup>11</sup> and Seeberger<sup>12</sup> reported the synthesis of  $(1 \rightarrow 2)$ - $\beta$ ,  $(1 \rightarrow 3)$ - $\beta$  and  $(1 \rightarrow 4)$ - $\beta$ -mannosides using 4,6-O-benzylidene protected mannosyl-1-O-sulfoxides or thiomannosides respectively through intermolecular glycosylation. In our approach, we used mannosyl-1-O-sulfoxide donor first time in PMB mediated IAD approach which because of good leaving nature gave high yield of  $(1 \rightarrow 2)$ - $\beta$ -mannosides even without the 4,6-benzylidene group.

The intramolecular aglycon delivery method (IAD) has been successfully used for the synthesis of  $(1 \rightarrow 3)$ - $\beta$  and  $(1 \rightarrow 4)$ - $\beta$ mannosides, but the most challenging  $(1 \rightarrow 2)$ - $\beta$ -mannosides have not been successful by IAD method so far. Keeping in view the challenge, we designed a new strategy for the synthesis of  $(1 \rightarrow 2)$ - $\beta$ -mannosides via IAD approach where we envisioned that a *p*-methoxybenzyl group at C-2 position of mannosyl residue will participate in both ether/acetal exchange as well as in intra-molecular aglycon delivery for glycosylation. To test the hypothesis, we first synthesized and coupled 2-O-p-methoxybenzyl-3,4,6-tri-Obenzylmannosyl phenyl sulfoxide 6 as donor with suitably protected  $\alpha$ -phenyl thiopyranosides **4** as acceptor in the presence of DDQ (1.4 equiv), under anhydrous conditions which afforded the mixed acetals 7 quite smoothly.<sup>9,13,14</sup> Further the mixed acetal was used for glycosylation by performing reaction with triflic anhydride (Tf<sub>2</sub>O, 0.95 equiv) in the presence of 2.6-di-tert-butyl-4-methylpyridine (DTBMP, 3.0 equiv) at -78 °C which gave exclusively the desired  $(1 \rightarrow 2)$ - $\beta$ -mannopyranoside **8** with 84% yield as revealed by spectroscopic studies (Scheme 1). The starting materials 4 and 6 were used in this glycosylation, which in turn were synthesized from corresponding orthoester donor of mannose 2 as shown in Scheme 2.<sup>15,16</sup>

The stereochemistry of compound **8** was determined by <sup>1</sup>H NMR and further confirmed by 2D NMR spectral data analysis. The  $\alpha$ - and  $\beta$ -configurations of **8** were assigned by the appearance of anomeric protons as doublets at  $\delta$  5.60, 4.58 with coupling constant values of  $J_{1,2}$  1.2 and 6.4 Hz, respectively. The HSQC and HMBC correlation of anomeric proton at  $\delta$  5.60 (d, J = 1.2 Hz) with <sup>13</sup>C ( $\delta$  85.2) indicated the  $\alpha$ -orientation whereas the other anomeric proton at  $\delta$  4.58 (d, J = 6.4 Hz) with <sup>13</sup>C ( $\delta$  96.6) confirmed the  $\beta$ -orientation. The relative configuration of **8** was authenticated by NOESY correlation as being compatible with computer modelling in which the close contact of atoms in space calculated were consistent with NOESY correlation. In NOESY correlation H-1 exhibited



**Scheme 1.** Formation of  $(1 \rightarrow 2) \beta$ -mannosylation by IAD.



**Scheme 2.** Synthesis of mannoside fragments **4** and **6**. Reagents and conditions: (i) BF<sub>3</sub>Et<sub>2</sub>O, PhSH; (ii) NaOMe, MeOH; 90% over 2 steps; (iii) PMBCl, NaH, DMF; (iv) *m*-CPBA, DCM, 85%.

a correlation with H-1' indicating that the two protons H-1 & H-1' were situated in a same face and assigned as  $\alpha$ -proton with the  $\beta$ -linkage (Fig. 2 of ESI).

Now we extended our IAD method for the synthesis of final product **1**, the  $(1 \rightarrow 2)$ - $\beta$ -pentamannoside of PLM. The intermediates were prepared as shown in Schemes 2–5. For this, we started with mannose and converted it into an orthoester donor 2 (Scheme 2) using reported method.<sup>16</sup> The orthoester was successfully converted to an acceptor 10<sup>17</sup> as well as mannosyl TCA donor **12** (Scheme 3).<sup>18</sup>

Having the new IAD method and the key intermediates in hand, the  $(1 \rightarrow 2)$ - $\beta$ -mannopyranoside **8** was now successfully converted into the corresponding sulfoxide donor **14** (Scheme 4).<sup>19</sup> The synthesis of the second key trisaccharide intermediate **16** was achieved by the coupling of mannopyranosyl donor **12** with the acceptor **10** (Scheme 5), which on further deacetylation by using sodium methoxide provided the alcohol **15**.<sup>20</sup> The  $\alpha$ -stereochemistry of the resultant glycoside **15** was established by NMR analysis.

To construct the required trisaccharide **16**, the disaccharide **15** was coupled with donor **6** using our intra-molecular aglycon deliv-



**Scheme 3.** Synthesis of mannoside fragments 10 and 11. Reagents and conditions: (i) BF<sub>3</sub>Et<sub>2</sub>O, AllOH; (ii) NaOMe, MeOH; 90% over 2 steps; (iii) *p*-TSA, DCE; (iv) Tri chloro acetonitrile, K<sub>2</sub>CO<sub>3</sub>, DCM, 90% over 2 steps.



**Scheme 4.** Synthesis of  $\beta$ -dimannoside donor 14. Reagents and conditions: (i) PMBCI, NaH, DMF; (ii) *m*-CPBA, DCM, 96%, over 2 steps.



Scheme 5. Synthesis of  $\beta$ -trimannoside acceptor 16. Reagents and conditions: (i) TMSOTf, DCM; (ii) NaOMe, MeOH; (iii) 6, DDQ, 1.5 equiv; (iv) DTBMP, Tf<sub>2</sub>O, DCM, -78 °C to rt.



Scheme 6. Synthesis of β-pentamannoside by 2+3 coupling. Reagents and conditions: (i) DDQ, (1.5 equiv); (ii) DTBMP, Tf<sub>2</sub>O, DCM, -78 °C to rt.



Scheme 7. Iterative synthesis of β-pentamannoside 1. Reagents and conditions: (i) 6, DDQ, (1.5 equiv); (ii) DTBMP, Tf<sub>2</sub>O, DCM, -78 °C to rt; (iii) BnBr, NaH, DMF; (iv) 20, Pd/C 10%, THF, MeOH, H<sub>2</sub>, 94%.

ery (IAD) condition and the desired product obtained in 86% yield. The stereochemistry of **16** having  $\alpha$ - and  $\beta$ -anomeric linkages was determined by HSOC NMR. Bundle et al. also reported the synthesis of compound **16** but by a different synthetic strategy.<sup>21</sup>

To further confirm the stereochemistry at the anomeric position of trisaccharide 16, disaccharide 15 was then converted to the trisaccharides **17** and **18** by  $1,2-\alpha$ -glycosylation with **12** (Scheme 5b, of ESI) and then compared with trisaccharide 16 (HSQC given in ESI).

Finally, the key synthetic step to construct  $(1 \rightarrow 2)$ - $\beta$ -pentamannoside via 2+3 glycosylation with the mannosyl donor 14 and the acceptor trimannoside 16 was tried, which gave the desired pentamannoside 19 in only low 18% yield; not satisfactory for the total synthesis of PLM (Scheme 6).

Therefore, we modified our approach to construct the  $(1 \rightarrow 2)$ - $\beta$ -pentamannoside **19** by an iterative synthetic strategy, wherein the trimannoside 16 was first converted into tetra-mannoside 18b and subsequently into pentamannoside 19 using the donor 6 as shown in Scheme 7. The  $(1 \rightarrow 2)$ - $\beta$ -pentamannoside **19** was further converted into perbenzylated pentamannoside 20. The stereochemistry of  $(1 \rightarrow 2)$ - $\beta$ -pentamannoside **20** was confirmed by NMR and HSQC experiment. Finally, controlled global hydrogenolysis with Pd/C afforded the target  $(1 \rightarrow 2)$ - $\beta$ -pentamanan **1** in 94% isolated yield.

In summary, we have designed a high yielding method for  $(1 \rightarrow 2)$ - $\beta$ -mannosylation employing IAD via PMB ether/acetal intermediate and also achieved the first synthesis of  $(1 \rightarrow 2)$ - $\beta$ - pentamannoside domain of phospholipomannan (PLM) of Candida albicans.

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## Supplementary data

Supplementary data (spectral data for all compounds 1-20) associated with this article can be found, in the online version, at http://dx.doi.org/10.1016/j.tetlet.2014.03.099.

### **References and notes**

- 1. Trinel, P. A.; Maes, E.; Zanetta, J. P.; Delplace, F.; Coddeville, B.; Jouault, T.; Strecker, G.; Poulain, D. J. Biol. Chem. 2002, 277, 37260.
- 2 Jouault, T.; Bernigaud, A.; Lepage, G.; Trinel, P.; Poulain, D. Immunology 1994, 83, 268
- 3 Jouault, T.; Fradin, C.; Trinel, P. A.; Bernigaud, A.; Poulain, D. J. Infect. Dis. 1998, 178, 792.
- 4 Fradin, F.; Slomianny, M. C.; Mille, C.; Masset, A.; Robert, R.; Sendid, B.; Ernst, J. F.; Michalski, J. C.; Poulain, D. Infect. Immun. 2008, 76, 4509.
- (a) Murciano, C.; Moyes, D. L.; Runglall, M.; Islam, A.; Mille, C.; Fradin, C.; Poulain, D.; Gow, N. A. R.; Naglik, J. R. Infect. Immun. 2011, 79, 4902; (b) Johnson, M. A.; Bundle, D. R. Chem. Soc. Rev. 2013, 42, 4327.
- (a) Ferguson, M. A. J.; Homans, S. W.; Dwek, R. A.; Rademacher, T. W. Science 1988, 239, 753; (b) Homans, S. W.; Ferguson, M. A. J.; Dwek, R. A.; Rademacher, T. W.; Anand, R.; Williams, A. F. *Nature* **1988**, 333, 269; (c) McConville, M. J.; Feguson, M. A. J. *Biochem. J.* **1993**, 294, 305; (d) Ruhela, D.; Banerjee, P Vishwakarma, R. A. Curr. Sci. 2012, 102, 194; (e) Tsai, Y. H.; Liu, X.; Seeberger, P. H. Angew. Chem., Int. Ed. 2012, 51, 11438. and references cited therein.
- (a) Ali, A.; Gowda, D. C.; Vishwakarma, R. A. Chem. Commun. 2005, 519; (b) Vishwakarma, R. A.; Menon, A. K. Chem. Commun. 2005, 453; (c) Vishwakarma, R. A.; Vehring, S.; Mehta, A.; Sinha, A.; Pomorski, T.; Herrmann, A.; Menon, A. K. Org. Biomol. Chem. 2005, 3, 1275; (d) Ali, A.; Vishwakarma, R. A. Tetrahedron 2010, 66, 4357; (e) Ruhela, D.; Vishwakarma, R. A. J. Org. Chem. 2003, 68, 4446; (f) Chawla, M.; Vishwakarma, R. A. J. Lipid Res. 2003, 44, 594; (g) Saikam, V.; Raghupathy, R.; Yadav, M.; Gannedi, V.; Singh, P. P.; Qazi, N. A.; Sawant, S. D.; Vishwakarma, R. A. Tetrahedron Lett. 2011, 52, 4277.
- (a) Barresi, F.; Hindsgaul, O. J. Am. Chem. Soc. 1991, 113, 9376; (b) Stork, G.; Kim, 8. G. J. Am. Chem. Soc. 1992, 114, 1087; (c) Stork, G.; La Clair, J. L. J. Am. Chem. Soc. **1996**, 118, 247.
- (a) Dan, A.; Ito, Y.; Ogawa, T. J. Org. Chem. 1995, 60, 4680; (b) Ishiwata, A.; Lee, Y. J.; Ito, Y. Org. Biomol. Chem. 2010, 8, 3596; (c) Ishiwata, A.; Sakurai, A.; Nishimiya, Y.; Tsuda, S.; Ito, Y. J. Am. Chem. Soc. 2011, 133, 19524.
- (a) Codée, J. D. C.; Hossain, L. H.; Seeberger, P. H. Org. Lett. **2005**, 7, 3251; (b) Baek, J. Y.; Choi, T. J.; Jeon, H. B.; Kim, K. S. Angew. Chem., Int. Ed. 2006, 45, 7436; (c) Crich, D.; Sun, S. J. Org. Chem. 1997, 62, 1198; (d) El Ashry, E. S. H.; Rashed, N.; Ibrahim, E. S. I. Curr. Org. Synth. 2005, 2, 175.
- (a) Crich, D.; Li, W.; Li, H. J. Am. Chem. Soc. 2004, 126, 15081; (b) Crich, D.; Li, H.; Yao, Q.; Wink, D. J.; Sommer, R. D.; Rheingold, A. L. J. Am. Chem. Soc. 2001, 123, 5826: (c) Crich, D. Acc. Chem. Res. 2010, 43, 1144: (d) Huang, M.: Garrett, G. E.: Birlirakis, N.; Bohe, L.; Pratt, D. A.; Crich, D. Nat. Chem. 2012, 4, 663.
- Jeroen, D. C.; Krock, C. L.; Castagner, B.; Seeberger, P. H. Chem. Eur. J. 2008, 14, 12. 3987
- 13. Ito, Y.; Ogawa, T. J. Am. Chem. Soc. 1997, 119, 5562.
- Ito, Y.; Ohnishi, Y.; Ogawa; Nakahara, Y. Synlett 1998, 1102. 14.
- Franks, N. E.; Montgomery, R. Carbohydr. Res. 1968, 6, 286. 15.
- 16.
- Zhang, Y. M.; Mallet, J. M.; Sinay, P. *Carbohydr. Res.* **1992**, 236, 73. Liu, X.; Stocker, B. L.; Seeberger, P. H. *J. Am. Chem. Soc.* **2006**, 128, 3638. 17
- Mayor, G. T.; Schmidt, R. R. Eur. J. Org. Chem. 1998, 1153. 18.
- Ito, Y.; Ogawa, T. Angew. Chem., Int. Ed. 1994, 33, 1765. 19.
- Grathwohl, M.: Schmidt, R. R. Synthesis 2001, 15, 2263. 20.
- 21. Wu, X. Y.; Bundle, D. R. J. Org. Chem. 2005, 70, 7381.