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# Divergent Syntheses of Silicon-Containing Alkanes by Iron-Catalysed Intermolecular 1,2-Difunctionalizations of Styrenes and Conjugated Alkenes with Silanes and Nucleophiles

Yuan Yang, Ren-Jie Song\*, Xuan-Hui Ouyang, Cheng-Yong Wang, Jin-Heng Li\*, and Shenglian Luo\*

Dedicated to Professor Jiannan Xiang on the occasion of his 60th birthday

**Abstract:** The first iron-catalysed 1,2-difunctionalizations of styrenes and conjugated alkenes with silanes and *N*- or *C*-nucleophiles using an oxidative radical strategy is described. Employing FeCl<sub>2</sub> and di-*tert*-butyl peroxide (DTBP) allows divergent alkene 1,2-difunctionalizations, including 1,2-aminosilylation, 1,2-arylsilylation and 1,2-alkylsilylation, which rely on a wide range of nucleophiles, namely, amines, amides, indoles, pyrrole and 1,3-dicarbonyls, thus providing a powerful platform for producing diverse silicon-containing alkanes.

Organosilicon chemistry has attracted increased attention from scientific researchers because organosilicon compounds have significant chemical, physical and bioactive properties, which grant them wide applications in organic chemistry, materials science, agrochemistry and medicinal chemistry.<sup>[1,2]</sup> Specifically, the recently expanding studies of both silicon analogues of known drugs and entirely new silicon-containing compounds have become an emerging leading research field in the pharmaceutical industry. For example, 1-amino-2-silylalkanes exhibit fascinating biological activities that exist in many potential pharmaceuticals and drug lead compounds,<sup>[2]</sup> such as Silavenlafaxine (1),<sup>[2a-b]</sup> compound 2<sup>[2c]</sup> and TMS-alanine (3)<sup>[2d-e]</sup> (Figure 1). Traditionally, synthesis of such 1-amino-2silylalkanes and other silicon-containing amino compounds relies on the introduction of amino or silicon functional groups into the corresponding silicon-based or amino-based frameworks;[1-3] however, these methods suffer from harsh reaction conditions with strong bases, narrow substrate scope, multiple steps and/or use of expensive silicon reagents. Thus, new general and divergent strategies for producing siliconcontaining compounds in a straightforward manner, especially 1amino-2-silylalkanes, are desirable and urgently needed.

Recently, transformations of alkenes with silanes have emerged as a powerful tool to assemble silicon-containing molecules in syntheses,<sup>[4-7]</sup> with the vast majority of examples concerning the hydrosilylation or dehydrogenative silylation of alkenes through alkyl-metal or alkyl radical intermediates.<sup>[4,5]</sup>

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Figure 1. Examples of medicinal interest molecules comprising silicon.

Alternatively, 1,2-difunctionalization of alkenes, such as 1,2-bissilvlation,[6a-c] 1,2-arylsilvlation[6d-f] and 1,2-hydroxysilvlation,[6g] is particularly attractive in that it allows the simultaneous introduction of a silicon group and the other functional group beyond the hydrogen atom across the C=C bond to prepare complex functional silicon-containing molecules. However, examples of such alkene difunctionalizations are quite rare and challenging probably due to complications from dominant competitive reactions including hydrosilylation, dehydrogenative silvlation, silane redistribution, and silane dehydrocoupling.[4-6] To the best of our knowledge, there is no report of intermolecular 1,2-aminosilylation of alkenes for the synthesis of 1-amino-2-silylalkanes. Herein, we report a new iron-catalysed intermolecular 1,2-aminosilylation of styrenes and conjugated alkenes with silanes and amines or amides for the preparation of diverse 1-amino-2-silylalkanes using a DTBP oxidant (Scheme 1),<sup>[7,8]</sup> wherein a silicon group and an amino group are simultaneously incorporated across the C=C bond. Interestingly, the reaction is expanded to 1,2-carbosilylation of alkenes with silanes and a wide range of carbon nucleophiles, such as indoles, pyrrole and 1,3-dicarbonyls.



Scheme 1. Three-component intermolecular 1,2-difunctionalization of alkenes.

As shown in Table 1, the three-component reaction of 4methoxystyrene (**1a**) with HSi(TMS)<sub>3</sub> (**2a**) and morpholine (**3a**) was chosen for the optimization of the 1,2-aminosilylation conditions. Treatment of alkene **1a** with silane **2a**, amine **3a**, 10 mol% FeCl<sub>2</sub> and 3 equiv of DTBP in PhCF<sub>3</sub> at 120 °C for 20 h afforded the desired product **4aaa** in 67% yield (entry 1).

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Inspired by these results, the amount of  $\text{FeCl}_2$  was subsequently examined. The results revealed 10 mol%  $\text{FeCl}_2$  as the best option (entries 1-4). Notably, without  $\text{FeCl}_2$  the reaction could still occur, albeit with the yield diminished to 7% (entry 2). Other alternative Fe catalysts, such as  $\text{Fe}(\text{OTf})_2$  and  $\text{FeCl}_3$ , were tested (entries 5 and 6), but none were more efficient than  $\text{FeCl}_2$ . However, no desired reaction was observed in the absence of DTBP (entry 7), and both lower and higher amounts of DTBP had negative effects (Table S1 in Supporting Information). Gratifyingly, a reaction scale up to 1 mmol of alkene **1a** was successful at accessing **4aaa** in useful yield (entry 8).

#### **Table 1.** Screening of optimal reaction conditions<sup>[a]</sup>

MeO 1a	+ H-SI(TMS) <sub>3</sub> + (N 2a) + H 3a + H + H 3a + H + H + H + H + H + H + H + H + H + H	MeO 4aaa
Entry	Variation from the Standard Conditions	Isolated Yield [%]
1	none	67
2	without FeCl <sub>2</sub>	7
3	FeCl <sub>2</sub> (5 mol %)	47
4	FeCl <sub>2</sub> (20 mol %)	68
5	Fe(OTf) <sub>2</sub> instead of FeCl <sub>2</sub>	30
6	FeCl <sub>3</sub> instead of FeCl <sub>2</sub>	48
7	without DTBP	0
8 <sup>[b]</sup>	none	63

[a] Reaction conditions: **1a** (0.2 mmol), **2a** (2 equiv), **3a** (2 equiv), FeCl<sub>2</sub> (10 mol%), DTBP (3 equiv), PhCF<sub>3</sub> (2 mL), argon, 120 °C, and 20 h. Some byproducts, including the hydrosilylation product **4aa'** and dehydrogenative silylation product **4aa''**, were determined by GC-MS analysis. [b] **1a** (1 mmol) and 24 h.

Having optimized the reaction conditions, we set out to investigate the substrate scope of this alkene 1,2-aminosilylation protocol (Table 2). Initially, a wide range of terminal alkenes, including arylalkenes 1b-n, alkylalkenes 1o-p, 1,3-diene 1q and 1-en-3-yne 1r, were examined with silane 2a and amine 3a (3baa-raa). An array of substituents, namely, Me, Cl, Br, CN, CF<sub>3</sub> and MeO, on the aryl ring at the terminal alkene were well tolerated, and their electronic properties and positions affected the reactivity (3baa-naa). While alkene 1c bearing an electrondonating 4-MeC<sub>6</sub>H<sub>4</sub> group delivered 4caa in 67% yield, alkenes 1d-g and 1i having a weak (CI or Br) or a strong (CN or CF<sub>3</sub>) electron-withdrawing group afforded 4daa-gaa and 4iaa with diminishing yields. Importantly, chloride and bromide groups were tolerated, thus providing the opportunity for subsequent modification of the halogenated position (4daa-eaa). Arylalkenes 1h and 1j with a MeO group on the meta and ortho position were highly reactive (4haa and 4jaa). Using alkenes containing disubstituted phenyl (1k-I), naphthalen-2-yl (1m) and 3methylthiophen-2-yl (1n) groups were successfully converted into 4kaa-naa in 39%-72% yields. Unfortunately, aliphatic alkenes 10 and 1p both had no reactivity (40aa-paa). Noted that conjugated alkenes 1g-r were suitable substrates (4gaa-raa). Gratifyingly, the optimal conditions were applicable to a wide array of nitrogen nucleophiles, including secondary amines (3bd), primary amines (3e-1j) and amides or an imide (3k-n), which

underwent the aminosilylation reaction with alkene **1a** and  $HSi(TMS)_3$  **2a** to afford **4aab-aan** in 40%-70% yields. Importantly, a number of silanes, such as *tert*-butyldimethylsilane (**2b**), triethylsilane (**2c**), tri*iso*propylsilane (**2d**), triphenylsilane (**2e**), methyldiphenylsilane (**2f**) and dimethyl(phenyl)silane (**2g**), were competent reaction partners for producing **4aba-aea** and **4afh-agh**, albeit giving decreased yields in comparison with those of HSi(TMS)\_3 **2a**.



 Table 2. Variation of the Alkenes (1), Silanes (2) and N-Nucleophiles (3). [a]

 Reaction conditions: see Table 1. [b] >90% of alkene 1 was recovered.

Next, the scope with respect to carbon nucleophiles was explored under the optimal conditions (Table 3). Indole 3o as a suitable C-nucleophile smoothly underwent the 1,2-arylsilylation reaction with various alkenes 1a-g, 1h, 1j-k, 1q-r and 1s providing 5aao-gao, 5hao, 5jao-kao, 5qao-rao and 5sao in 44%-67% yields.<sup>[9]</sup> 1-Methyl-1-phenylethene (1t) also exhibited reactivity and delivered 5tao in moderate yield. For internal 1,2-dihydronaphthalene (1u) and prop-1-en-1alkenes. ylbenzene (1v), both were viable for the 1,2-arylsilylation reaction (5uao-vao). Subsequently, a series of indoles 3p-v bearing different substitution patterns were tested by reacting with alkene 1a and HSi(TMS)3 2a (5aap-aav). 1-Methyl-1Hindole (3p) was transformed to 5aap with 57% yield. Several substituents, such as 5-Me, 5-MeO, 5-CN, 7-CO2Me, 4-CHO and 2-Me, on the indole core were perfectly tolerated (5aaq-aav). Other tertiary silane 2b-c, secondary silane 2h and primary silane 2i were also suitable for preparing 5abo-aco and 5aho-

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aio. Further exploitation of *C*-nucleophiles revealed that pyrrole **3w** and 1,3-dicarbonyls **3x-y** were viable *C*-nucleophiles. Using pyrrole **3w** could furnish useful fully-substituted pyrrole **5aaw** in 62% yield. Pleasingly, both pentane-2,4-dione (**3x**) and ethyl 3-oxo-3-phenylpropanoate (**3y**) could succeed to furnish **5aax** and **5aay** in moderate yields.



 Table 3. Variation of the Alkenes (1), Silanes (2) and C-Nucleophiles (3). [a]

 Reaction conditions: see Table 1. [b] Diphenylsilane (2h) was used.

On the basis of the above results, a possible mechanism for the alkene difunctionalization reaction was proposed (Scheme 2 and Scheme S1 in Supporting Information).<sup>[4-8]</sup> Initially, DTBP is split by the active Fe<sup>II</sup> species under heating takes place to afford the *tert*-butoxyl radical and the Fe<sup>III</sup>(*t*BuO) species.<sup>[6,7]</sup> Hydrogen-abstraction from HSi(TMS)<sub>3</sub> 2a by the *t*-butoxyl radical gives the silicon-centred radical **A**,<sup>[6,7]</sup> which sequentially adds across the C=C bond of alkene 1a to grant the alkyl radical intermediate **B**. Oxidation of the-intermediate **B** by the Fe<sup>III</sup>(*t*BuO) species delivers the alkyl cation intermediate **C**. Finally, reaction of the intermediate **C** with *N*- or *C*-nucleophile 3 provides the desired product 4 or **5**.



Scheme 2. Possible Reaction Mechanism.

In summary, we have developed the first iron-catalysed intermolecular 1,2-difunctionalization of styrenes and conjugated alkenes with silanes and *N*- or *C*-nucleophiles. The reaction is initiated by a key silicon-centred radical from oxidative cleavage of the Si-H bond followed by addition across the C=C bond in addition to N-H oxidative functionalization cascades, thus building the 1-amino-2-silylalkane cores. Moreover, the reaction can be expanded to three-component 1,2-carbosilylation of alkenes via Si-H/C-H oxidative functionalization. Notably, the reaction exhibits high selectivity, excellent tolerance of functional groups and broad scope of substrates, including a wide range of alkenes, silanes, and *N*- and *C*-nucleophiles. Further studies on mechanism and applications are currently underway.

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#### **Conflict of interest**

The authors declare no conflict of interest.

Keywords: iron • difunctionalization • alkenes • silanes • nucleophiles

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## Entry for the Table of Contents

## COMMUNICATION



Employing FeCl<sub>2</sub> and DTBP divergent alkene 1,2-difunctionalization reactions, including 1,2-aminosilylation, 1,2-arylsilylation and 1,2-alkylsilylation, are achieved by using different nucleophiles, thereby providing straightforward and practical access to 1-amino-2-silylalkanes and other functionalized silicon-containing alkanes with broad substrate scope and high selectivity.

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