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1-Azaspiro[4.4]nonane-2,6-dione and the separation and absolute configurations of its enantiomers

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Racemic 1-azaspiro[4.4]nonane-2,6-dione 1 was easily synthesized from Abstract: cyclopentanone in five steps and resolved via chiral acetals into enantiomers. Cephlotaxine (-)-3 and its enantiomer (+)-3 were obtained from (-)-1 and (+)-1, respectively, according to the literature. © 1997 Published by Elsevier Science Ltd. All rights reserved.

Key intermediate 2 for the total synthesis of (\pm) -cephalotaxine 3, as reported by Kuehne et al.¹ includes the 1-azaspiro[4.4]nonane-2,6-dione skeletone 1 (Scheme 1). Compound 2 has been prepared by the oxidative rearrangement of bicyclic ene-lactam 4 with lead tetraacetate.¹ This paper presents a facile synthesis of 1 and the resolution of its enantiomers via chiral acetals. The absolute configuration of each enantiomer was determined based on CD spectra and X-ray analysis and were converted to the optical active cephalotaxine (-)-3 and its enantiomer (+)-3 by the method of Kuehne without racemization.

The synthesis² of 1 and separation of its enantiomers were carried out as shown in Scheme 2. 2-(2-Methoxycarbonylethyl)cyclopentanone 5, readily available by the alkylation of the pyrrolidine enamine of cyclopentanone with methyl acrylate (68%),³ was used with isopropenyl acetate to obtain enol acetate 6 in 72% yield, which was nitrated to 7 with a mixture of trifluoroacetic anhydride and ammonium nitrate in 68% yield. The desired spirolactam 1 was obtained by reduction of 7 with zinc in acetic acid-ethanol (2:1) in good yield (80%).⁴ The acetalization of 1 with (R,R)-2,3-butanediol in the presence of catalytic p-toluenesulfonic acid gave a mixture of diastereomeric acetals (8 and 9) in quantitative yields. By HPLC on silica gel by elution with chloroform, 8 and 9 were easily separated in the first and second fractions, respectively. The hydrolysis of 8 and 9 with a mixture of AcOH-H₂O (5:95) gave (+)-1 and (-)-1 in quantitative yields, respectively. Each enantiomer was confirmed pure by HPLC using a chiral column (Figure 1).⁵



Scheme 1

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Scheme 2. Reagents and Conditions: a) isopropenyl acetate, p-TsOH, 110°C, 5 h; b) trifluoroacetic anhydride, ammonium nitrate, CH₂Cl₂, r.t., 30 min; c) Zn, AcOH-EtOH (2:1), reflux, 5 h; d) i) (R,R)-(-)-2,3-butanediol, p-TsOH, C₆H₆, reflux, 8 h, ii) column chromatography on silica gel by elution with CHCl₃; e) 5% AcOH, reflux, 2 h; f) NaH, 2-(3,4-methylenedioxyphenyl)ethanol p-toluenesulfonate, C₆H₆.



Figure 1.



Figure 2. The CD-spectra of (+)-1 (--) and (-)-1 (---) in MeOH.



Figure 3. ORTEP diagram of (+)-8.

The CD spectra of (+)-1 and (-)-1 showed opposite absorptions as shown in Figure 2. The absolute configurations of (+)-1 and (-)-1 were suggested by comparison of the CD spectra of (+)-1 and (-)-1 with those of the known (R)-(+)- and (S)-(-)-1,7-diazaspiro[4.4]nonane-2,6-diones.⁶ X-ray analysis⁷ of (+)-8 shown in Figure 3 confirmed this consideration.

Finally, the syntheses of optical active cephalotaxine (-)-3 and its enantiomer (+)-3 were successfully carried out from (-)-11⁸ and (+)-10⁸, respectively, by the method reported in the literature.^{1,9}

References

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2. All new compounds were fully characterized by spectrosopic data and microanalysis and/or molecular ion mass measurements. Representative data for some new compounds: 7: ¹H-NMR (300MHz, CDCl₃) δ : 1.98–2.84 (10H, m), 3.69 (3H, S), CIMS m/z: 216 (M⁺+1). 1: ¹H-NMR (300MHz, CDCl₃) δ : 1.85–2.54 (10H, m), 5.73 (1H, br). ¹³C-NMR (75MHz, CDCl₃) δ : 17.33, 29.58, 30.18, 34.70, 35.81, 66.99, 178.80, 217,17. MS m/z: 153 (M⁺). mp 133–134°C. (+)-1: [α]_D^{25.6} +68.2 (c=1.0, CHCl₃). (-)-1: [α]_D^{24.4} –68.0 (c=1.0, CHCl₃). (+)-8: ¹H-NMR (300MHz, CDCl₃) δ : 1.20 (3H, d, J=5.6Mz), 1.22 (3H, d, J=5.6Mz), 1.65–1.75 (3H, m), 1.76–1.82 (2H, m), 1.86–1.91 (2H, m), 2.35–2.40 (2H, m), 2.45–2.50 (1H, m), 3.56–3.63 (2H, m), 5.55 (1H, br). ¹³C-NMR (75MHz, CDCl₃) δ : 15.53, 15.98, 16.53, 29.26, 30.31, 32.24, 34.63, 67.37, 77.89, 79.29, 115.02, 177.51. [α]_D^{26.0} +50.2 (c=1.0, CHCl₃). MS m/z: 225 (M+). mp 121–122C°. (-)-9: ¹H-NMR (300MHz, CDCl₃) δ : 1.20 (3H, d, J=2.5MHz), 1.21 (3H, d, J=2.5MHz), 1.63–1.80 (4H, m), 1.67–1.94 (3H, m), 2.28–2.40 (2H, m), 2.47–2.52 (1H, m), 3.58–3.62 (2H, m), 5.54 (1H, br). ¹³C-NMR (75MHz, CDCl₃) δ : 1.20 (3H, m), 2.47–2.52 (1H, m), 3.58–3.62 (2H, m), 5.54 (1H, br). ¹³C-NMR (75MHz, CDCl₃) δ : 1.20 (3H, d, J=2.5MHz), 1.21 (3H, d, J=2.5MHz), 1.63–1.80 (4H, m), 1.67–1.94 (3H, m), 2.28–2.40 (2H, m), 2.47–2.52 (1H, m), 3.58–3.62 (2H, m), 5.54 (1H, br). ¹³C-NMR (75MHz, CDCl₃) δ : 16.19,

16.36, 16.83, 29.26, 30.66, 33.41, 35.29, 68.17, 78.78, 79.22, 115.46, 178.39. $[\alpha]_D^{25.6}$ -60.6 (c=1.0, CHCl₃). MS m/z: 225 (M⁺). mp 96–98C°.

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4. Catalytic hydrogenation (5% Pd–C, MeOH–AcOH 5:1, H₂ 2 atm, 5 h) of nitroester 7 afforded the desired amide 1 in low yield (35%). Various products were obtained in catalytic hydrogenations (5% Pd–C, H₂ 2.5 atm) using other solvents, such as hydroxyamine 12 in 21% yield in EtOH (24 h) and alcohol 13 in 22% and 46% yield in EtOH–AcOH (30:1) (9 h) and in EtOH–AcOH–H₂O (10:0.1:1) (12 h), respectively.



5. HPLC analysis was kindly carried out by Dr. Yasuo Dobashi, Tokyo University of Pharmacy and Life Science, using a chiral column (cited in Figure 1) prepared by himself. cf. Dobashi, Y.; Hara, S. J. Org. Chem. 1987, 52, 2490.

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7. Data were obtained using a Mac Scaience MXC18, radiation CuK α (γ =1541780Å) at 298K. Data collection of 2336 data, 2128 observed (F>3.00 σ (F)). All diagrams and calculation results were obtained using a Crystan 6.3.1 (Mac Science, Japan). Structural refinement was made to R=0.0537 and wR=0.0560. Crystal dimensions: 0.50×0.30×0.20 mm. Crystal system: orthorhombic; space group P2₁2₁2₁; Unit Cell dimensions *a*=11.2256(27)Å, *b*=24.7065(59)Å, *c*=9.0541(23)Å; Volume 2511.10(101)Å³; Z=8,density (calc): 1.351Mg m⁻³, Absorption Coefficient: none.: Authors thank Mr. Tadashi Hata, Sankyo Co. Ltd., for the X-ray analysis of (+)-8.

8. The direct alkylation of amides $\{(+)-1 \text{ and } (-)-1\}$ with alkyl halides in the presence of bases and phase transfer catalysts was not conducted successfully. For the alkylation of (+)-8 and (-)-9 to (+)-10 and (-)-10, respectively, tosylate (in Scheme 2) was better than any corresponding alkyl halide.

9. (-)-3, mp 118–120°C (lit.¹⁰ mp 122–124°C): $[\alpha]_D^{24.0}$ –184.6 (c=0.5,CHCl₃) {lit.¹⁰ $[\alpha]_D$ –188 (CHCl₃)} (+)-3, mp 118–120°C: $[\alpha]_D^{27.6}$ +188.9 (c=1.1, CHCl₃). Synthesis of (-)-cephalotaxine has recently been reported: Isono, N.; Mori, M. J. Org. Chem. **1995**, 60, 115.

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