Synthesis of 1,4-Benzoxazepine Derivatives *via* a Novel Domino Aziridine Ring-Opening and Isocyanide-Insertion Reaction

Fei Ji,^a Mei-fang Lv,^a Wen-bin Yi,^a and Chun Cai^{a,*}

^a Chemical Engineering College, Nanjing University of Science and Technology, Nanjing 210094, People's Republic of China Fax: (+86)-25-8431-5030; e-mail: c.cai@mail.njust.edu.cn

Received: July 24, 2013; Published online: November 14, 2013

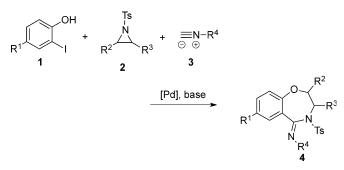
Supporting information for this article is available on the WWW under http://dx.doi.org/10.1002/adsc.201300650.

Abstract: A novel and efficient domino process has been developed for the synthesis of 1,4-benzoxazepine derivatives from a range of readily accessible *N*-tosylaziridines, 2-iodophenols and isocyanides. This process involves the aziridine ring-opening reaction with 2-iodophenol, followed by a palladiumcatalyzed isocyanide-insertion reaction. This regioselective and high-yielding transformation could be extended to its application for the synthesis of natural products and biologically interesting heterocycles.

Keywords: aziridine ring-opening; 1,4-benzoxazepines; domino process; isocyanide-insertion; regioselectivity

The importance of N-heterocyclic scaffolds in nature products, pharmaceuticals and synthetic materials is well known,^[1] thus significant efforts have been devoted to the discovery of methodologies for the preparation of N-heterocyclic compounds in organic chemistry. Among them, 1,4-benzoxazepine derivatives are a very important class of seven-membered rings and such structural units could widely exist in numerous medicinal heterocyclic molecules with promising biological and pharmaceutical activities.^[2] These biological characteristics have stimulated organic researchers to explore synthetic methods for 1,4-benzoxazepines and their structural analogues.^[3] However, the major drawback of the reported methods is their multistep nature,^[3a] which prevents them from generating a large number of structural derivatives efficiently. Over the past decades, domino reactions as one of the most powerful tools for generating complex molecules have attracted more and more attention.^[4] This strategy leads to the rapid formation of two or more new bonds under the same conditions, which avoids the separation of the intermediates and minimizes the amount of chemical waste, energy, and time. Thus, the development of a novel domino reaction for the synthesis of 1,4-benzoxazepine derivatives is highly desired.

The aziridines as strained three-membered ring heterocycles are considered to be important building blocks for the synthesis of various N-heterocyclic compounds.^[5] They could undergo ring-opening reactions with various nucleophilic reagents to provide a direct access to a range of substituted amines,^[6] which are important precursors and have been explored in the synthesis of biological heterocycles.^[7] Recently, the Alper group reported a new domino aziridine ring-opening/carboxamidation reaction of Ntosylaziridine and 2-halophenols for the synthesis of 1,4-benzoxazepine derivatives.^[7a] However, the use of toxic carbon monoxide and the harsh reaction conditions limited the further use and exploration of these reactions.^[8] Isocyanides, a kind of unsaturated molecules similar to carbon monoxide, have emerged as versatile building blocks in the construction of medicinally important molecules and natural products.^[9] Thus, transition metal-catalyzed reactions involving isocyanides have been studied intensively.^[10] This methodology has several advantages such as simple handling, mild conditions and wide substrate scope (variable group at nitrogen) over carbon monoxide.^[8] As a result, the transition metal-catalyzed isocyanideinsertion reactions could offer more opportunities for the synthesis of N-heterocyclic compounds. In this context, we designed a palladium-catalyzed domino reaction involving 2-iodophenols 1, N-tosylaziridines 2, and isocyanides 3, which would allow an efficient access to 1,4-benzoxazepines 4 by the aziridine ringopening and isocyanide-insertion events (Scheme 1). To the best of our knowledge, this novel domino strategy based on isocyanides and aziridines has not been reported up to now.



Scheme 1. Approaches towards the 1,4-benzoxazepine structure.

To test the hypothesis, we commenced the investigation with 2-iodophenol 1a (0.55 mmol), N-tosylaziridine of cyclohexene 2a (0.50 mmol) and tert-butyl isocyanide 3a (0.75 mmol) as model substrates catalyzed by $Pd(PPh_3)_2Cl_2$ (5 mol%) and Cs_2CO_3 (2 equiv.) in toluene under reflux for 24 h. The desired 1,4-benzoxazepine 4a was obtained in 65% yield (Table 1, entry 1). To our delight, the isolated yield of 4a was enhanced further to 78% when the amount of base was increased (Table 1, entry 2). However, further raising the amount of base to 4 equiv. could not improve the yield significantly (Table 1, entry 3). When K₂CO₃ and DIPEA were employed as bases instead of Cs₂CO₃, the desired product 4a was isolated in 62% and 23% yields, respectively (Table 1, entries 4 and 5). Next, several Pd catalysts were screened: it was found that $Pd(PPh_3)_2Cl_2$ was the best choice (Table 1, entries 2, and 6-9). Moreover, no obvious improvement in the yield was observed when the solvent was switched to DMF, MeCN, THF and 1,4-dioxane (Table 1, entries 2, and 10–13).

The structure of the novel 1,4-benzoxazepine 4a was fully characterized by mass spectrometry, ¹H NMR, ¹³C NMR spectroscopy and elemental analysis. The *trans* stereochemistry of the product 4a was confirmed by single-crystal X-ray analysis (Figure 1). According to this single-crystal structure analysis of 4a, the effect of the steric hindrance may lead to the Z configuration of the imino bond (see the Supporting Information).

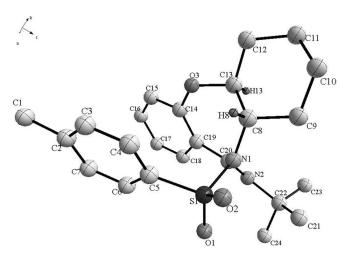


Figure 1. Single-crystal X-ray analysis of 4a.

OH +	N-Ts	+ <u>}</u> N≡ ⊕ ⊙	[Pd], base solvent, <i>T</i> [°C]		
1a	2a	3a		4a >	
Catalyst	Base (equiv.)		Solvent	<i>T</i> [°C]	Yield [9
$Pd(PPh_3)_2Cl_2$ Cs_2CO_3 (2.0)		toluene	110	65	
$Pd(PPh_3)_2Cl_2$	Cs_2CO_3 (3.0)		toluene	110	78
$Pd(PPh_3)_2Cl_2$		$_{2}CO_{3}(4.0)$	toluene	110	75
Pd(PPh ₂) ₂ Cl ₂	DIPEA(30)		toluene	110	23

Table 1. Optimization of the domino aziridine ring-opening/isocyanide-insertion reaction.^[a]

Entry	Catalyst	Base (equiv.)	Solvent	$T [^{\circ}C]$	Yield [%] ^[b]
1	Pd(PPh ₃) ₂ Cl ₂	Cs_2CO_3 (2.0)	toluene	110	65
2	$Pd(PPh_3)_2Cl_2$	Cs_2CO_3 (3.0)	toluene	110	78
3	$Pd(PPh_3)_2Cl_2$	Cs_2CO_3 (4.0)	toluene	110	75
4	$Pd(PPh_3)_2Cl_2$	DIPEA (3.0)	toluene	110	23
5	$Pd(PPh_3)_2Cl_2$	$K_2CO_3(3.0)$	toluene	110	62
6	$Pd(OAc)_2$	$Cs_2CO_3(3.0)$	toluene	110	45
7	$Pd(dppf)Cl_2$	$Cs_2CO_3(3.0)$	toluene	110	66
8	$Pd(amphos)_2Cl_2$	$Cs_2CO_3(3.0)$	toluene	110	75
9	PdCl ₂ /Xantphos (1:2)	Cs_2CO_3 (3.0)	toluene	110	32
10	$Pd(PPh_3)_2Cl_2$	$Cs_2CO_3(3.0)$	DMF	110	72
11	$Pd(PPh_3)_2Cl_2$	$Cs_2CO_3(3.0)$	THF	66	67
12	$Pd(PPh_3)_2Cl_2$	$Cs_2CO_3(3.0)$	MeCN	81	75
13	$Pd(PPh_3)_2Cl_2$	$Cs_2CO_3(3.0)$	1,4-dioxane	100	65

[a] Reaction conditions: 2-iodophenol 1a (0.55 mmol), N-tosylaziridine of cyclohexene 2a (0.50 mmol), tert-butyl isocyanide 3a (0.75 mmol), 5 mol% Pd catalyst, base, solvent (4 mL), 24 h.

[b] Isolated yield based on 2a.

```
3402
                asc.wiley-vch.de
```

© 2013 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim

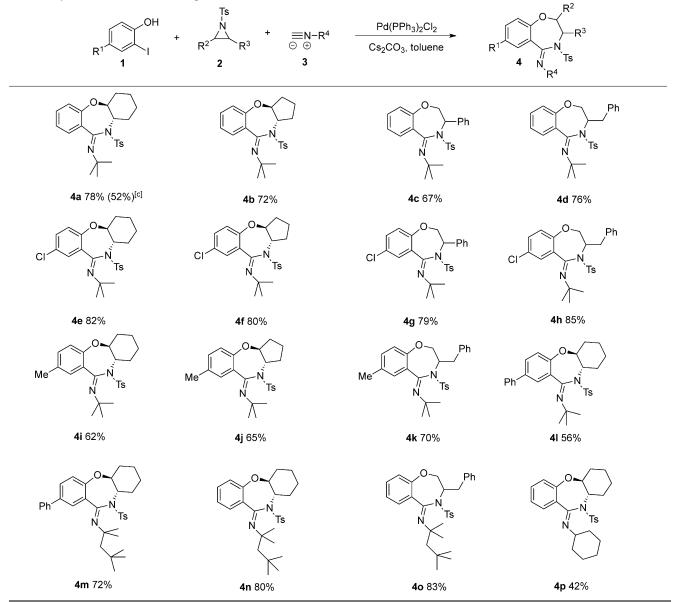


Table 2. Synthesis of 1,4-benzoxazepine derivatives.^[a,b]

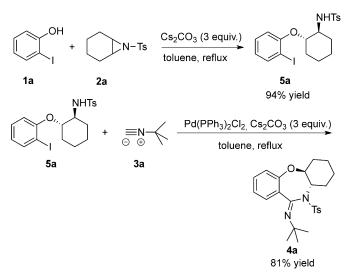
[a] Reaction conditions: 2-iodophenols 1 (0.55 mmol), N-tosylaziridines 2 (0.50 mmol), isocyanides 3 (0.75 mmol), Pd-(PPh₃)₂Cl₂ (5 mol%), Cs₂CO₃ (3 equiv.), toluene (4 mL), reflux, 24 h.

^[b] Isolated yield based on **2**.

^[c] *Reation conditions:* 2-bromophenol (0.55 mmol), *N*-tosylaziridine of cyclohexene (0.50 mmol), *tert*-butyl isocyanide (0.75 mmol), Pd(PPh₃)₂Cl₂ (5 mol%), Cs₂CO₃ (3 equiv.), toluene (4 mL), reflux, 24 h.

With these optimized reaction conditions in hand, we turned our attention to the scope of the domino aziridine ring-opening/isocyanide-insertion reaction. Different *N*-tosylaziridines, 2-iodophenols and isocyanides were examined and the results are summarized in Table 2. Various *N*-tosylaziridines, including cyclic and acyclic ones, reacted with *tert*-butyl isocyanide and 2-iodophenol, leading to the desired 1,4-benzox-azepines in high yields (Table 2, **4a-4d**). As for acyclic *N*-tosylaziridines, the reaction was completely regioselective and provided only one regioisomer (Table 2,

4c and **4d**). The reactions employing various substituted 2-iodophenols bearing a 4-chloro, 4-methyl or 4-phenyl group also worked well to furnish the corresponding 1,4-benzoxazepines (Table 2, **4e**, **4i** and **4l**). Unfortunately, no reaction occurred when 2-iodo-4-nitrophenol was employed in this transformation. We reasoned that the introduction of the electron-with-drawing nitro group decreased the nucleophilic activity of 2-iodo-4-nitrophenol. Next, different isocyanides were also applied to probe the scope of the reaction. Generally, when *tert*-butyl isocyanide and 1,1,3,3-tet-

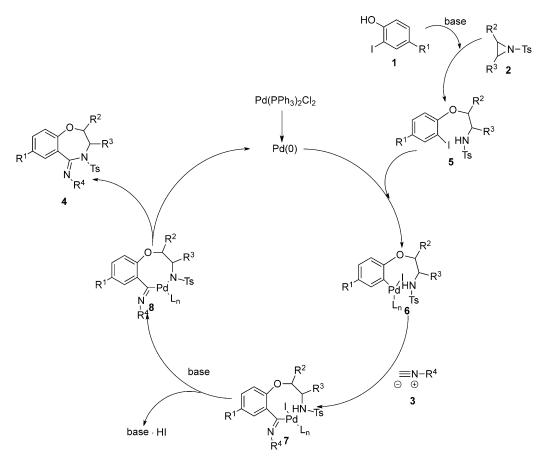


Scheme 2. Investigation of the reaction mechanism.

ramethylbutyl isocyanide were used, the reaction gave the corresponding products in high yields (Table 2, 4a and 4n). However, when cyclohexyl isocyanide was applied in this process under the optimized conditions, the desired product was only obtained in moderate yield (Table 2, 4p). Finally, the treatment of 2bromophenol, *N*-tosylaziridine of cyclohexene and *tert*-butyl isocyanide afforded the corresponding 1,4benzoxazepine **4a** in only 52% yield. The lower rate of oxidative addition of the Ar-Br moiety to the *in situ* formed palladium(0) species may lead to the difference in behaviour compared to that of 2-iodophenol.

To gain insights of the reaction mechanism, several control experiments were conducted as shown in Scheme 2. When 2-iodophenol **1a** and *N*-tosylaziridine **2a** were treated in toluene using Cs_2CO_3 (3 equiv.) as a base, the ring-opening product **5a** was obtained in up to 94% yield. Next, amide **5a** was reacted with *tert*-butyl isocyanide **3a** under the optimized conditions, leading to the desired product **4a** in 81% yield. The experimental results revealed that this domino reaction may proceed through the base-catalyzed ring-opening of *N*-tosylaziridines with 2-iodophenols, and then the palladium-catalyzed isocyanide-insertion reaction.

On the basis of these preliminary results and the control experiments conducted, a possible catalytic cycle involved in the present process was outlined as shown in Scheme 3.^[7a,8] Initially, the base-catalyzed ring-opening reaction of *N*-tosylaziridine **2** with 2-io-dophenol **1** generates the amide **5**. The oxidative ad-



Scheme 3. Possible mechanism of the domino aziridine ring-opening/isocyanide-insertion reaction.

3404 asc.wiley-vch.de

© 2013 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim

dition of **5** to the *in situ* generated palladium(0) species leads to the palladium complex **6**. This could then undergo the isocyanide-insertion process to give the intermediate **7**. Next, with the aid of the base, hydroiodic acid is extruded out of **7** to generate the eightmembered product **8**. Finally, reductive elimination affords **4**, regenerating the palladium (0) catalyst.

In conclusion, we have demonstrated a novel and efficient procedure for the synthesis of a new class of substituted 1,4-benzoxazepine derivatives *via* a palladium-catalyzed three-component domino reaction of *N*-tosylaziridines, 2-iodophenols and isocyanides. This process involved the aziridine ring-opening reaction with 2-iodophenol, followed by palladium-catalyzed isocyanide-insertion reaction. This domino reaction was successful with a range of *N*-tosylaziridines, 2-io-dophenols and isocyanides, and could afford the desired products in moderate to high yields. More importantly, this methodology may be instrumental to the rapid construction of important 1,4-benzoxazepine derivatives with promising biological and pharmaceutical activities.

Experimental Section

General Procedure for 1,4-Benzoxazepine Derivatives

A mixture of 2-iodophenol (0.55 mmol), *N*-tosylaziridine (0.50 mmol), isocyanide (0.75 mmol), Pd(PPh₃)₂Cl₂ (18 mg) and Cs₂CO₃ (489 mg) was stirred in toluene (4 mL) under reflux for 24 h. Upon completion of the reaction, the mixture was concentrated under vacuum and the resulting residue was purified by column chromatography.

Characterization Data of a Representative 1,4-Benzoxazepine (4a)

White solid; 144–146 °C; ¹H NMR (500 MHz, CDCl₃): δ = 1.14–1.15 (m, 1H), 1.32–1.47 (m, 2H), 1.59 (s, 9H), 1.50– 1.60 (m, 1H), 1.77–1.84 (m, 2H), 2.06–2.08 (m, 1H), 2.31 (s, 3H), 2.44–2.47 (m, 1H), 3.67–3.72 (m, 1H), 3.92–3.98 (m, 1H), 6.23 (d, *J* = 8.0 Hz, 1H), 6.91–6.98 (m, 3H), 7.07–7.12 (m, 3H), 7.63–7.65 (m, 1H); ¹³C NMR (125 MHz, CDCl₃): δ =20.47, 22.79, 23.51, 29.85, 32.03, 32.85, 57.72, 64.65, 79.38, 117.35, 120.62, 124.75, 126.80, 127.65, 129.26, 132.12, 134.13, 137.72, 141.74, 156.32; MS (ESI): *m*/*z* = 427 [M+H]; anal. calcd. for C₂₄H₃₀N₂O₃S: C 67.58, H 7.09, N 6.57; found: C 67.67, H 7.1, N 6.55.

Crystallographic data for 1,4-benzoxazepine **4a** have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication no. CCDC 937174. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre *via* www.ccdc.cam.ac.uk/data_request/cif.

Acknowledgements

We are thankful for financial support by NUST research funding (2011ZDJH07). We also acknowledge the Analysis and Testing Center of Nanjing University of Science and Technology for the NMR analyses reported in this work.

References

- [1] a) D. P. Walsh, Y.-T. Chang, *Chem. Rev.* 2006, 106, 2476; b) M. D. Burke, S. L. Schreiber, *Angew. Chem.* 2004, 116, 48; *Angew. Chem. Int. Ed.* 2004, 43, 46; c) P. Arya, D. T. H. Chou, M.-G. Baek, *Angew. Chem.* 2001, 113, 351; *Angew. Chem. Int. Ed.* 2001, 40, 339.
- [2] a) T. MiKi, M. Kori, H. Mabuchi, R.-T. Tozawa, T. Nishimoto, Y. Sugiyama, K. Teshima, H. Yukimasa, J. Med. Chem. 2002, 45, 4571; b) R. A. Smits, H. D. Lim, B. Stegink, R. A. Bakker, I. J. P. de Esch, R. Leurs, J. Med. Chem. 2006, 49, 4512; c) Y. Liao, B. J. Venhuis, N. Rodenhuis, W. Timmerman, H. Wikstrom, E. Meier, G. D. Bartoszyk, H. Bottcher, C. A. Seyfried, S. Sundell, J. Med. Chem. 1999, 42, 2235; d) L. Banfi, A. B. G. Guanti, P. Lecinska, R. Riva, Org. Biomol. Chem. 2006, 4, 4236; e) V. J. Merluzzi, K. D. Hargrave, M. Labadia, K. Grozinger, M. Skoog, J. C. Wu, C. K. Shih, K. Eckner, S. Hattox, J. Adams, A. S. Rosenthal, R. Faanes, R. J. Eckner, R. A. Koup, J. L. Sullivan, Science 1990, 250, 1411.
- [3] a) D. Tsvelikhovsky, S. L. Buchwald, J. Am. Chem. Soc. 2011, 133, 14228; b) L. K. Ottesen, F. Ek, R. Olsson, Org. Lett. 2006, 8, 1771; c) X. Xu, S. Guo, Q. Dang, J. Chen, X. Bai, J. Comb. Chem. 2007, 9, 773; d) F. Shi, X. Xu, L. Zheng, Q. Dang, X. Bai, J. Comb. Chem. 2008, 10, 158; e) R. Fu, X. Xu, Q. Dang, X. Bai, J. Org. Chem. 2005, 70, 10810; f) J. Yang, X. Che, Q. Dang, Z. Wei, S. Gao, X. Bai, Org. Lett. 2005, 7, 1541; g) K. Kamei, N. Maeda, K. Nomura, M. Shibata, R. Katsuragi-Ogino, M. Koyama, M. Nakajima, T. Inoue, T. Ohno, T. Tatsuoka, Bioorg. Med. Chem. 2006, 14, 1978; h) K. Kamei, N. Maeda, R. Ogino, M. Koyama, M. Nakajima, T. Tatsuoka, T. Ohno, T. Inoue, Bioorg. Med. Chem. Lett. 2011, 11, 595; i) M. M. Cavalluzzi, A. Catalano, C. Bruno, A. Lovece, A. Carocci, F. Corbo, C. Franchini, G. Lentini, V. Tortorella, Tetrahedron: Asymmetry 2007, 18, 2409.
- [4] a) L. F. Tietze, *Chem. Rev.* 1996, 96, 115; b) P. J. Parsons, C. S. Penkett, A. J. Shell, *Chem. Rev.* 1996, 96, 195; c) S.-I. Ikeda, *Acc. Chem. Res.* 2000, 33, 511; d) M. M. Hussain, P. G. Walsh, *Acc. Chem. Res.* 2008, 41, 883.
- [5] a) G. S. Singh, M. D'hooghe, N. De Kimpe, *Chem. Rev.* 2007, 107, 2080; b) C. A. Olsen, H. Franzyk, J. W. Jaroszewski, *Eur. J. Org. Chem.* 2007, 1717; c) I. D. G. Watson, L. Yu, A. K. Yudin, *Acc. Chem. Res.* 2006, 39, 194; d) X. E. Hu, *Tetrahedron* 2004, 60, 2701; e) J. B. Sweeney, *Chem. Soc. Rev.* 2002, 31, 247; f) S. Minakata, *Acc. Chem. Res.* 2009, 42, 1172.
- [6] a) J. Wu, X. Sun, Y. Li, *Eur. J. Org. Chem.* 2005, 4271;
 b) D. Sureshkumar, S. M. Koutha, S. Chandrasekaran, *J. Am. Chem. Soc.* 2005, 127, 12760;
 c) X.-L. Hou, R.-H. Fan, L.-X. Dai, *J. Org. Chem.* 2002, 67, 5295;
 d) H.

© 2013 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim

Han, I. Base, E. J. Yoo, J. Lee, Y. Do, S. Chang, Org. Lett. 2004, 6, 4109.

- [7] a) G. Chouhan, H. Alper, Org. Lett. 2010, 12, 192; b) F. Zeng, H. Alper, Org. Lett. 2010, 12, 5567; c) R. K. Rao, A. B. Naidu, G. Sekar, Org. Lett. 2009, 11, 1923; d) R. K. Rao, I. Karthikeyan, G. Sekar, Tetrahedron 2012, 68, 9090; e) R. K. Rao, G. Sekar, Tetrahedron: Asymmetry 2011, 22, 948; f) S. Bhadra, L. Adak, S. Samanta, A. K. M. Maidul Islam, M. Mukherjee, B. C. Ranu, J. Org. Chem. 2010, 75, 8533.
- [8] a) T. Vlaar, B. U. W. Maes, E. Ruijter, R. V. A. Orru, Angew. Chem. Int. Ed. 2013, 52, 7084; b) S. Lang, Chem. Soc. Rev. 2013, 42, 4867; c) G. Qiu, Q. Ding, J. Wu, Chem. Soc. Rev. 2013, 42, 5257.
- [9] a) A. V. Lygin, A. de Meijere, Angew. Chem. 2010, 122, 9280; Angew. Chem. Int. Ed. 2010, 49, 9094; b) T. Yue, M.-X. Wang, D.-X. Wang, G. Masson, J. Zhu, J. Org. Chem. 2009, 74, 8396; c) H. Mihara, Y. Xu, N. E. Shepherd, S. Matsunaga, M. Shibasaki, J. Am. Chem. Soc. 2009, 131, 8384; d) R. Scheffelaar, M. Paravidino, D. Muilwijk, M. Lutz, A. L. Spek, F. J. J. de Kanter, R. V. A. Orru, E. Ruijter, Org. Lett. 2009, 11, 125; e) T. Pirali, G. C. Tron, G. Masson, J. Zhu, Org. Lett. 2007, 9, 5275; f) S.-X. Wang, M.-X. Wang, D.-X. Wang, J. Zhu, Org. Lett. 2007, 9, 3615; g) T. Pirali, G. C. Tron, J. Zhu, Org. Lett. 2006, 8, 4145; h) P. Janvier, M. Bois-Choussy, H. Bienayme, J. Zhu, Angew. Chem. 2003, 115, 835; Angew. Chem. Int. Ed. 2003, 42, 811.
- [10] a) Y. Wang, Q. Zhu, Adv. Synth. Catal. 2012, 354, 1902; b) V. Tyagi, S. Khan, A. Giri, H. M. Gauniyal, B. Sridhar, P. M. S. Chauhan, Org. Lett. 2012, 14, 3126; c) G. Qiu, Y. He, J. Wu, Chem. Commun. 2012, 48, 3836; d) T. Vlaar, E. Ruijter, A. Znabet, E. Janssen, F. J. J. de Kanter, B. U. W. Maes, R. V. A. Orru, Org. Lett. 2011, 13, 6496; e) M. Tobisu, S. Imoto, S. Ito, N. Chatani, J. Org. Chem. 2010, 75, 4835; f) G. Qiu, G. Liu, S. Pu, J. Wu, Chem. Commun. 2012, 48, 2903; g) T. Nanjo, C. Tsukano, Y. Takemoto, Org. Lett. 2012, 14, 4270; h) B. Liu, Y. Li, H. Jiang, M. Yin, H. Huang, Adv. Synth. Catal. 2012, 354, 2288; i) Z. Hu, D. Liang, J. Zhao, J. Huang, Q. Zhu, Chem. Commun. 2012, 48, 7371; j) C. Zhu, W. Xie, J. R. Falck, Chem. Eur. J. 2011, 17, 12591; k) G. Van Baelen, S. Kuijer, L. Rýček, S. Sergeyev, E. Janssen, F. J. J. de Kanter, B. U. W. Maes, E. Ruijter, R. V. A. Orru, Chem. Eur. J. 2011, 17, 15039; 1) Q. Cai, F. Zhou, T. Xu, L. Fu, K. Ding, Org. Lett. 2011, 13, 340; m) K. Komeyama, D. Sasayama, T. Kawabata, K. Takehira, K. Takaki, Chem. Commun. 2005, 5, 634; n) C. G. Saluste, R. J. Whitby, M. Furber, Angew. Chem. 2000, 112, 4326; Angew. Chem. Int. Ed. 2000, 39, 4156; o) P. J. Boissarie, Z. E. Hamilton, S. Lang, J. A. Murphy, C. J. Suckling, Org. Lett. 2011, 13, 6256; p) J. Peng, L. Liu, Z. Hu, J. Huang, Q. Zhu, Chem. Commun. 2012, 48, 3772; q) T. Vlaar, P. Mampuys, M. Helliwell, B. U. W. Maes, R. V. A. Orru, E. Ruijter, J. Org. Chem. 2013, 78, 6735; r) F. Ji, M.-F. Lv, W.-B. Yi, C. Cai, Synthesis 2013, 45, 1965.