

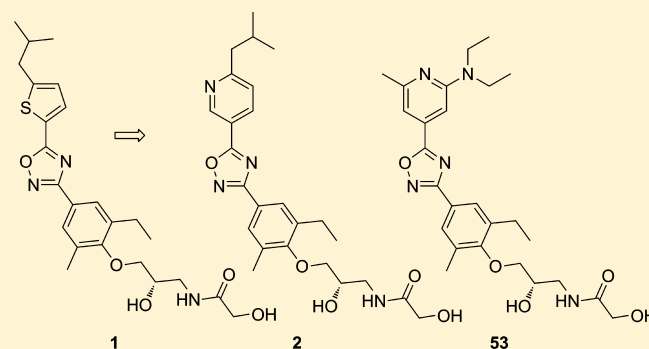
Novel S1P₁ Receptor Agonists – Part 3: From Thiophenes to Pyridines

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Supporting Information

ABSTRACT: In preceding communications we summarized our medicinal chemistry efforts leading to the identification of potent, selective, and orally active S1P₁ agonists such as the thiophene derivative **1**. As a continuation of these efforts, we replaced the thiophene in **1** by a 2-, 3-, or 4-pyridine and obtained less lipophilic, potent, and selective S1P₁ agonists (e.g., **2**) efficiently reducing blood lymphocyte count in the rat. Structural features influencing the compounds' receptor affinity profile and pharmacokinetics are discussed. In addition, the ability to penetrate brain tissue has been studied for several compounds. As a typical example for these pyridine based S1P₁ agonists, compound **53** showed EC₅₀ values of 0.6 and 352 nM for the S1P₁ and S1P₃ receptor, respectively, displayed favorable PK properties, and penetrated well into brain tissue. In the rat, compound **53** maximally reduced the blood lymphocyte count for at least 24 h after oral dosing of 3 mg/kg.



■ INTRODUCTION

Synthetic sphingosine 1-phosphate 1 (S1P₁) receptor agonists hold great promise for treating inflammatory disorders and autoimmune diseases,^{1–3} and the number of clinical trials investigating their safety and efficacy is growing rapidly. While fingolimod^{4–7} has approval to treat relapsing multiple sclerosis, siponimod,^{8–10} ponesimod,^{11,12} and ONO-4641 (structure undisclosed)^{13,14} successfully completed phase II and RPC-1063 (structure undisclosed)¹⁵ and MT-1303 (structure undisclosed)¹⁶ recently entered phase II clinical trials for this debilitating disease. The last two compounds are also evaluated in patients suffering from ulcerative colitis¹⁷ and inflammatory bowel disease,¹⁸ respectively. Results of an open-label phase I study in MS patients treated with CS-0777 have been published recently.^{19,20} In addition, siponimod is currently studied in patients suffering from secondary progressive multiple sclerosis²¹ as well as in patients with polymyositis and dermatomyositis.²² Ponesimod, on the other hand, successfully completed a phase II trial for chronic plaque psoriasis.^{23,24} Furthermore, fingolimod's ability to reduce graft rejection has been studied in renal transplant patients,²⁵ and KRP-203²⁶ was investigated in patients with subacute cutaneous lupus erythematosus²⁷ and refractory ulcerative colitis.²⁸ APD334 (structure undisclosed)^{29,30} recently entered phase I clinical trials. Additional molecules emerged from discovery programs at Almirall,^{31,32} Glaxo,^{33–35} Bristol-Myers Squibb,³⁶ e.g.³⁷ and

other companies,^{38–41} but it is not known whether they entered clinical development yet.

The lysophospholipid sphingosine 1-phosphate (S1P) is the natural ligand of five G-protein-coupled receptors named S1P₁–S1P₅.⁴² S1P–S1P receptor signaling is involved in many cellular responses including survival, proliferation, differentiation, adhesion, migration, and chemotaxis^{43–45} and leads to angiogenesis, endothelial barrier enhancement, airway and blood vessel constriction, alveolar epithelial barrier disruption, heart rate modulation, neurite extension, and bone homeostasis.^{46–49} The physiological role of S1P receptor signaling has been studied in several organs and tissues such as the nervous system,^{50–52} the lung,^{46,53} the cardiovascular system,^{54–58} the cells of the immune system,^{59–61} and bone tissue,⁶² and S1P receptor activation has been implicated in many pathological situations such as autoimmunity, inflammation, cardiovascular disorders, and cancer.^{3,60,61,63–66}

S1P plays an important role in the motility of various cell types,^{62,67,68} in particular of the immune system.^{69–72} Lymphocytes continuously circulate through the body, thereby monitoring the body for microbial intruders or aberrant cells and home to lymph nodes and Peyer's patches.⁷³ In order to leave these secondary lymphoid organs and return to

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circulation, lymphocytes follow the gradient of S1P that exists between lymph and blood. This chemotactic event is driven by the S1P/S1P₁ receptor signaling axis.^{69,74–76} Synthetic S1P₁ agonists have been shown to trigger internalization of the S1P₁ receptor, thereby abolishing the lymphocyte's ability to sense the S1P gradient.^{76,77} As a consequence, lymphocytes are sequestered to lymph nodes and no longer migrate to other organs. Interfering with lymphocyte trafficking by S1P₁ receptor modulation represents an attractive approach to treat a variety of lymphocyte dependent autoimmune diseases, as demonstrated in several clinical trials. In contrast, activation of the S1P₃ receptor is deemed undesirable, as it has been associated with effects on heart rate,^{78–84} vaso- and bronchoconstriction,^{82,85} hypertension,⁸⁴ and enhancement of fibrosis^{86–88} in animal studies.

In addition to the peripheral immunomodulatory action of S1P₁ receptor agonists, more recent studies identified modulation of the S1P₁, S1P₃, or S1P₅ receptors on neural cells like astrocytes and oligodendrocytes as a nonimmunologic central nervous system (CNS) effect possibly contributing to the efficacy of synthetic S1P receptor agonists in multiple sclerosis.^{6,89–95} Following these arguments, brain penetration of a synthetic S1P₁ receptor agonist may be beneficial for a compound's efficacy in CNS related diseases such as multiple sclerosis. On the other hand, for the treatment of peripheral diseases such as psoriasis or Crohn's disease, CNS penetration of a S1P receptor agonist is not needed or may even be undesired. Hence, the utility of a compound for different diseases may depend on its ability to penetrate brain tissue, and we therefore characterized some of our most advanced compounds with respect to their potential to penetrate the brain.

Previously we described the discovery of a novel series of selective S1P₁ receptor agonists incorporating a 5-isobutyl substituted thiophene.⁹⁶ A prototypical example, compound **1** (Figure 1), had EC₅₀ values of 0.7 and 320 nM in a GTPγS

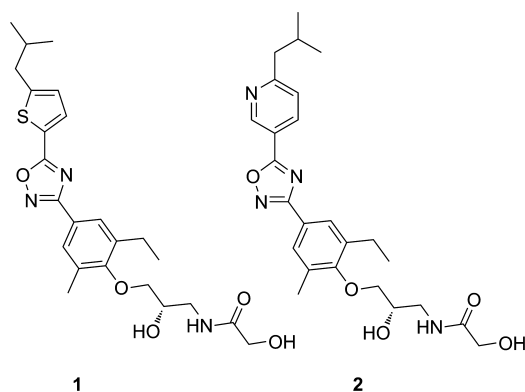


Figure 1. Structures of thiophene based S1P₁ receptor agonist **1** and a first 3-pyridine analogue **2**.

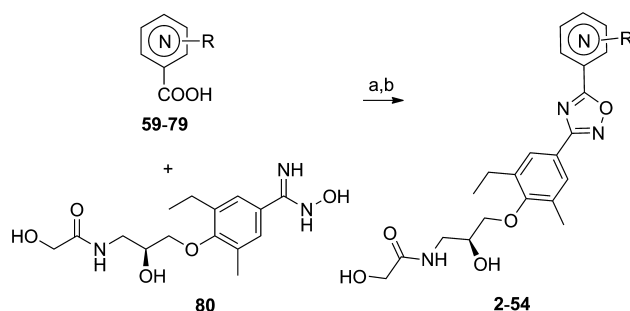
assay for S1P₁ and S1P₃, respectively. With a clogP value of 3.97 this thiophene derivative is rather lipophilic. As high lipophilicity has been shown to be associated with compound liabilities such as target promiscuity, hERG inhibition, poor aqueous solubility, metabolic instability, or high protein binding,^{97–101} we embarked on a program in which the thiophene head of our S1P₁ receptor agonists was replaced by a pyridine. A first such analogue, pyridine derivative **2**, has a clogP of 2.74 and showed an EC₅₀ value of 5.7 nM for S1P₁,

making the pyridine an attractive alternative to the thiophene in **1**. In the following sections we describe our detailed studies aiming at the discovery of pyridine containing S1P₁ receptor agonists meeting the following requirements: the compound has a lower clogP than thiophene **1** (ideally <3), is highly potent on S1P₁ (EC₅₀ < 5 nM) with an affinity of >250 nM for S1P₃ and a more than 100-fold selectivity against this receptor, and shows maximal lymphocyte count reduction for at least 24 h when administered at a dose of 10 mg/kg to Wistar rats. For compounds fulfilling these criteria the ability to penetrate the brain would then be measured to assess their utility in different diseases.

RESULTS AND DISCUSSION

Synthesis. The target compounds **2–54** for this study were usually prepared by coupling each of the pyridine carboxylic acids **59–79** with *N*-hydroxybenzamidines **80** followed by cyclizing the hydroxyimide ester intermediate to the desired oxadiazole (Scheme 1). *N*-Hydroxybenzamidines **80** was

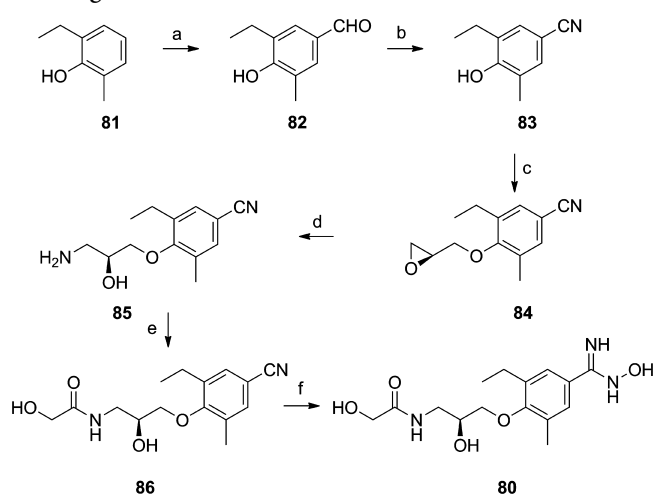
Scheme 1. Assembling the Central Oxadiazole Ring^a



^aReagents and conditions: (a) HOBt, EDC HCl, THF, rt, 18 h; or TBTU, Hünig's base, DMF, rt, 1 h; (b) dioxane, 80 °C, 18 h, 6–76% (two steps).

prepared by formylating phenol **81** in a Duff reaction to give benzaldehyde **82** (Scheme 2). Reacting **82** with hydroxylamine hydrochloride in NMP under microwave irradiation furnished benzonitrile **83** which was then alkylated with (*R*)-glycidol under Mitsunobu conditions. Epoxide **84** was reacted with ammonia in methanol to give amino alcohol **85**. Coupling of **85** with glycolic acid to give **86** established the side chain attached to the benzene ring. Finally, nitrile **86** was reacted with hydroxylamine hydrochloride to afford the desired building block **80**. In some cases the polar side chain attached to the para position of the phenyl moiety was introduced after the oxadiazole ring had been established. In this case, the appropriate pyridine carboxylic acid was reacted with 3-ethyl-*N*,4-dihydroxy-5-methylbenzamidines and the side chain was established in analogy to the steps outlined in Scheme 2 (for details see Supporting Information). For details on the synthesis of compounds **55–58** (Table 7) see Supporting Information.

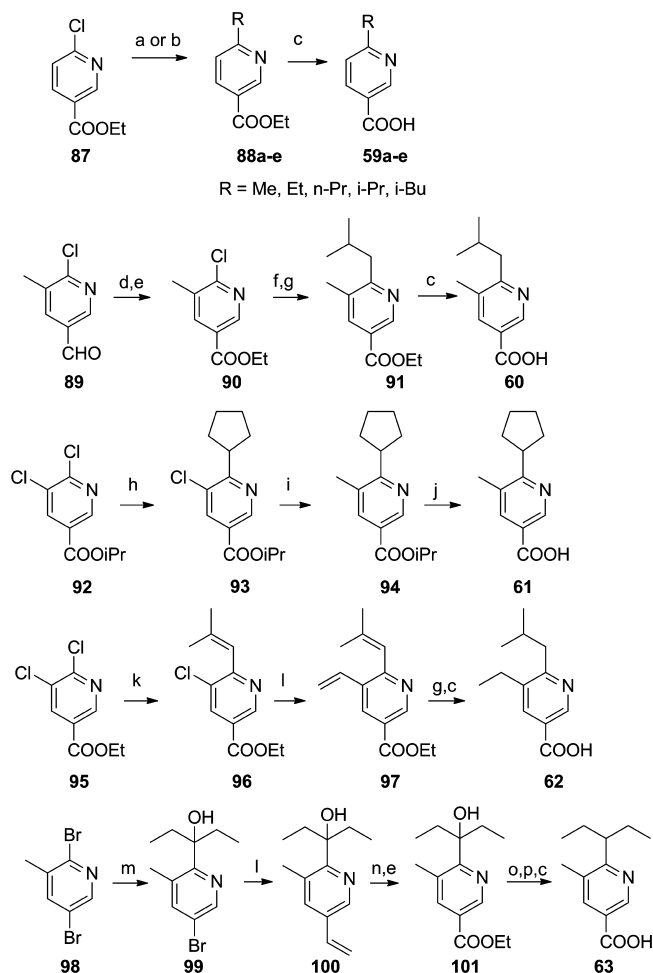
Prototypical syntheses of the various pyridinecarboxylic acids are exemplified in Schemes 3, 4, 5, and 6. Alkyl groups could be introduced to the pyridines by making use of transition metal catalyzed cross-coupling reactions developed by Suzuki,^{102–104} Negishi,¹⁰⁴ and Fürstner.^{105,106} The choice of the reaction type was generally driven by the availability of the corresponding alkyl electrophile. Alkylamines were introduced either by simple thermal displacement or by a Pd-catalyzed Buchwald–Hartwig

Scheme 2. Preparation of the *N*-Hydroxybenzamidine Building Block 80^a

^aReagents and conditions: (a) hexamethylenetetraamine, HOAc, H₂O, 120 °C, Dean–Stark 2–3 h, 53–97%; (b) HONH₂ HCl, NMP, microwave 80–100 °C, 0.5–3 h; 86–95%; (c) (*R*)-glycidol, PPh₃, DEAD, THF, 0–20 °C, 18 h, 72–86%; (d) 7 N NH₃ in MeOH, 65 °C, 18 h (sealed vessel), quantitative (crude); (e) glycolic acid, EDC HCl, HOBT, rt, 18 h, 51–90%; (f) HONH₂ HCl, NaHCO₃ or Et₃N, MeOH or EtOH 60–80 °C, 18 h, 60–87%.

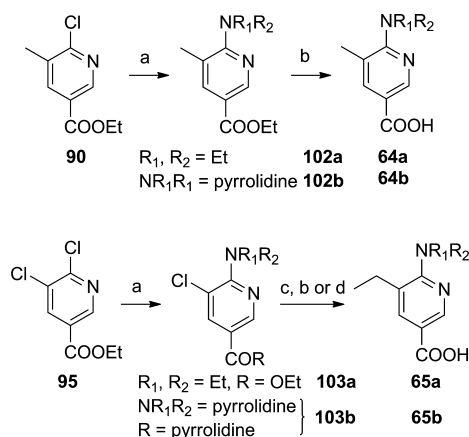
reaction.¹⁰⁷ For more details see Supporting Information. Scheme 3 illustrates the synthesis of some nicotinic acids. 6-Alkylnicotinic acid esters **88a–e** were obtained by reacting ethyl 6-chloronicotinate **87** either with the appropriate alkenyl boronic ester under Suzuki conditions followed by catalytic hydrogenation or with the appropriate Grignard reagent under Fürstner conditions. Ester cleavage under acidic conditions afforded the 6-alkylnicotinic acid building blocks **59a–e**. 6-Alkyl-5-methylnicotinic acids (e.g., **60**) were prepared in a similar fashion starting from ethyl 5-methyl-6-chloronicotinate **90** employing 2,4,6-tri-(2-alkenyl)cyclotriboroxanepyrindine complexes or alkenylboronic acid pinacol esters under Suzuki conditions.¹⁰⁸ Catalytic hydrogenation of the ethyl 6-alkylnicotinate intermediate furnished the corresponding ethyl 6-alkyl-5-methylnicotinate (e.g., **91**) which upon acidic ester cleavage afforded the desired nicotinic acid building block (e.g., **60**). Ethyl nicotinate **90** was prepared by oxidizing aldehyde **89**¹⁰⁹ with hydrogen peroxide and subsequent esterification of the resulting nicotinic acid carboxylic acid. In an alternative approach, isopropyl 5,6-dichloronicotinate **92** was first reacted with cyclopentylmagnesium bromide at room temperature in the presence of Fe(acac)₃ to give isopropyl 5-chloro-6-cyclopentylnicotinate **93** which was then converted to nicotinate **94** using dimethylzinc at 75–85 °C. Saponification yielded nicotinic acid **61**. Reacting ethyl 5,6-dichloronicotinate **95** with 2,4,6-tri-(2-methylprop-1-en-1-yl)cyclotriboroxanepyrindine complex under aqueous Suzuki conditions furnished 5-chloronicotinate **96** which was then subjected to anhydrous Suzuki conditions in the presence of 2,4,6-trivinylcyclotriboroxanepyrindine complex to afford 5,6-dialkenylnicotinate **97**. Hydrogenation and ester cleavage furnished 5-ethyl-6-isobutylnicotinic acid **62**.

In the case of the 3-pentyl derivative **63** the above approaches were unsuccessful. We therefore followed an alternative strategy. Thus, 2,5-dibromo-3-methylpyridine **98** was lithiated using 1.1 equiv of *n*-butyllithium and then reacted

Scheme 3. Preparation of 6-Alkylnicotinic Acids^a

^aReagents and conditions: (a) 2-alkenyl-4,4,5,5-tetramethyl-1,3,2-dioxaborolane, Pd(PPh₃)₄, PPh₃, 2 M aq K₂CO₃, 80 °C, 18 h, 66%, then 5 bar of H₂, Pd/C, methanol, 18 h, 46%; (b) alkenylmagnesium bromide, Fe(acac)₃, THF, NMP, –75 °C, 1–4 h, 47–93%; (c) 4–7 N aq HCl, 65–80 °C, 18–24 h, 91–97%; (d) 50% H₂O₂, HCOOH, 0 °C, 15 h, 47–87%; (e) EtOH, H₂SO₄, or TMSCl, reflux 18–42 h, 65–92%; (f) 2,4,6-tri-(2-methylpropenyl)cyclotriboroxanepyrindine complex or alkenylboronic acid pinacol ester, PPh₃, Pd(PPh₃)₄, DME or dioxane, 2 M aq K₂CO₃, 90 °C, 20 h, 71–73%; (g) Pd/C, H₂, MeOH, THF, rt, 15 h, 61–99%; (h) cyclopentylmagnesium bromide, Fe(acac)₃, THF, NMP, 0 °C, 1 h, rt, 18 h, 78%; (i) Me₂Zn, Pd(dppf)Cl₂, dioxane, 75–85 °C, 3–18 h, 36–58%; (j) 2 M aq LiOH, MeOH or EtOH, rt, 1–2 h, 83–88%; (k) 2,4,6-tri-(2-methylpropenyl)-cyclotriboroxanepyrindine complex, Ph₃P, Pd(PPh₃)₄, DME, 2 M aq K₂CO₃, 100 °C, 4.5 h, 92%; (l) 2,4,6-trivinylcyclotriboroxanepyrindine complex, Cs₂CO₃, tri-*tert*-butylphosphine, Pd₂(dba)₃, or PPh₃, Pd(PPh₃)₄, dioxane, 100 °C, 16–40 h, 59–67%; (m) *n*-BuLi, 3-pentanone, THF, –70 °C, 2 h, 37%; (n) KMnO₄, acetone, rt, 24 h, quantitative crude; (o) Burgess reagent, dioxane, 80 °C, 1 h microwave, 76%; (p) 1 bar of H₂, Pd/C, THF/methanol 1:1, rt, 4 h, 80%.

with 3-pentanone to form 5-bromopyridine **99**, which was converted to vinylpyridine **100** using trivinylcyclotriboroxanepyrindine complex under Suzuki conditions. The olefin in **100** was cleaved by KMnO₄ and the resulting acid was esterified to give nicotinate **101**. Treating **101** with Burgess reagent under microwave irradiation effected dehydration, and the corresponding unsaturated product was hydrogenated using H₂ on

Scheme 4. Preparation of 6-Aminonicotinic Acids^a

^aReagents and conditions: (a) $\text{R}_1\text{R}_2\text{NH}$, 80–100 °C, 1 day to 3 weeks, 85–99%; (b) 4–7 N aq HCl, 65–80 °C, 18–24 h, quantitative; (c) Et_2Zn , $\text{Pd}(\text{dppf})\text{Cl}_2$, dioxane, 75 °C, 18–24 h, 84% to quantitative; (d) 3 N aq NaOH, MeOH, 75 °C, 18 h, 79%.

Pd/C. Saponification of this material finally furnished the desired 6-(3-pentyl)nicotinic acid derivative **63**.

6-Aminonicotinic acids **64a,b** were obtained from their esters **102a,b** which were prepared by heating 6-chloronicotinate **90** in a sealed vessel with a large excess of the corresponding amine (Scheme 4). Alternatively, 5,6-dichloronicotinate **95** was reacted with diethylamine to form ethyl 5-chloro-6-aminonicotinate **103a**. When pyrrolidine was used as the amine component, the ester in **95** was transformed to the corresponding pyrrolidine amide **103b**. Pd-catalyzed Negishi coupling of **103a,b** with diethylzinc and subsequent ester hydrolysis afforded the 6-amino-5-ethylnicotinic acids **65a,b**.

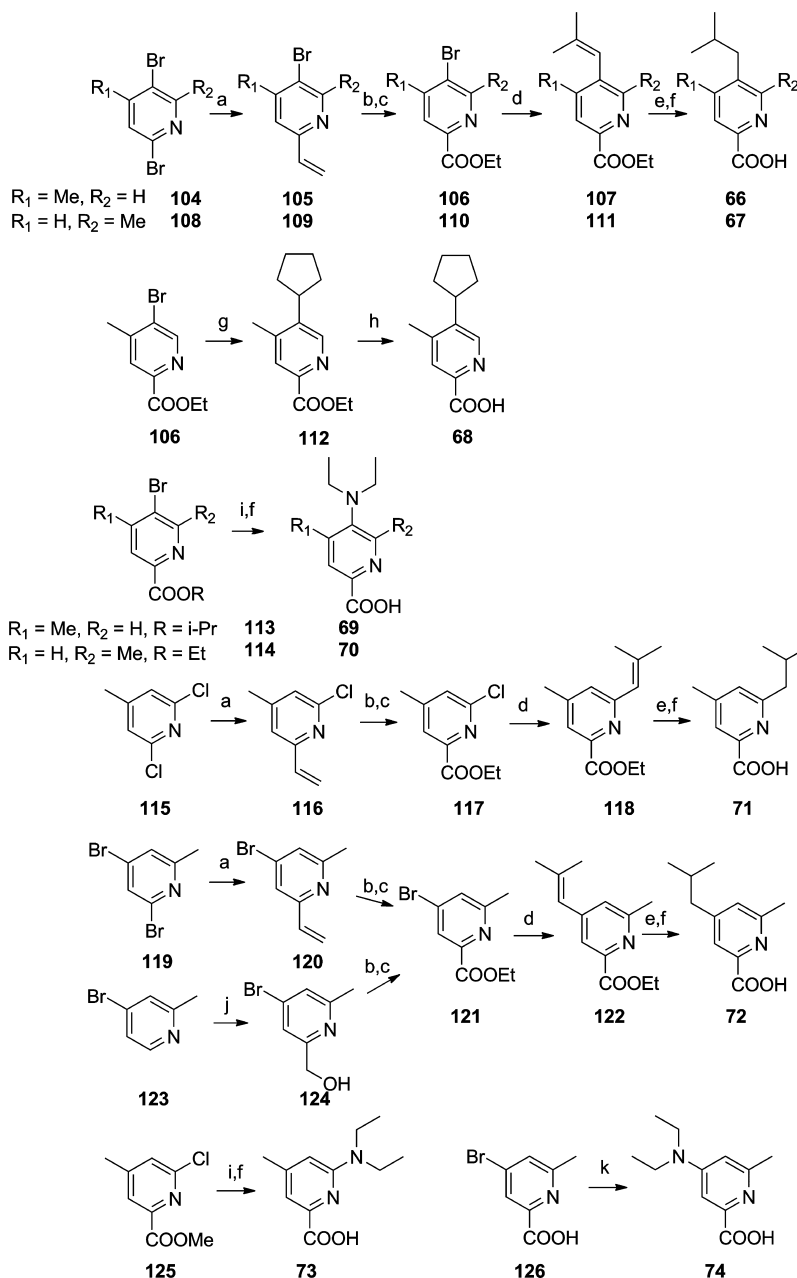
The synthesis of prototypical 4,5-, 5,6-, and 4,6-disubstituted picolinic acids is outlined in Scheme 5. Reacting 2,5-dibromopyridine **104** or **108** with trivinylcyclotrioxanepyrindine complex selectively furnished the corresponding 2-vinylpyridines **105** and **109** in good yields. Potassium permanganate mediated oxidation established the carboxylic acids which were esterified to deliver **106** and **110**. A second Suzuki reaction using 2,4,6-tri(2-methylpropenyl)-cyclotrioxanepyrindine complex produced the 5-alkenylpicolinic acid derivatives **107** and **111**. Catalytic hydrogenation and subsequent ester cleavage completed the synthesis of the two picolinic acids **66** and **67**. Negishi coupling of 5-bromo-4-methylpicolinate **106** with cyclopentylzinc bromide furnished ester **112** which was cleaved under basic conditions to give cyclopentyl-4-methylpicolinic acid **68**. 5-Diethylaminopicolinic acids **69** and **70** were prepared by reacting 5-bromopicolinate **113** and **114**, respectively, with diethylamine under Buchwald–Hartwig conditions and subsequent ester hydrolysis. Although the yields of these cross-coupling reactions were rather low, sufficient material could be isolated for the purpose of our studies. The preparation of 4,6-disubstituted picolinic acids followed a strategy similar to the one pursued for the 3,4- and 2,3-disubstituted picolinic acids **66** and **67**. Hence, 2,6-dichloro-4-methylpyridine **115** was first converted to 2-chloro-4-methyl-6-vinylpyridine **116** and then oxidized and esterified to give 2-chloropicolinate **117**. Suzuki cross-coupling of this material with 2,4,6-tri(2-methylpropenyl)-cyclotrioxanepyrindine complex furnished the 6-isobutenylpicolinate **118** which, after hydrogenation and ester hydrolysis, produced picolinic acid **71**.

The isomeric picolinic acid **72** was prepared following the same strategy starting from 2,4-dibromopicoline **119**. Thus, a vinyl group was introduced to position 6 of **119** via Suzuki reaction to give **120** which was then oxidized and esterified to form picolinate **121**. Suzuki reaction using 2,4,6-tri(2-methylpropenyl)-cyclotrioxanepyrindine complex, catalytic hydrogenation, and saponification furnished the desired compound **72**. Ethyl 4-bromo-6-methylpicolinate **121** could also be prepared via an alternative route making use of the Minisci reaction.^{110–112} To this end, 4-bromopicoline **123** was treated with ammonium peroxydisulfate in methanol containing a small amount of sulfuric acid to give hydroxymethylpyridine **124** in moderate yield. Oxidation of the primary alcohol with KMnO_4 and esterification of the resulting acid gave picolinate **121**. The 6-diethylaminopicolinic acid **73** was obtained by Buchwald reaction of 6-chloropicolinate **125** and subsequent saponification. The isomeric acid **74** was obtained by thermal displacement of the 4-bromo substituent of acid **126**. To enhance purification, the acid was temporarily transformed to its ethyl ester.

Finally, Scheme 6 illustrates the synthesis of 2-substituted and 2,6-disubstituted isonicotinic acids. The preparation of the 2-substituted representatives **75a–h** started from 2-haloisonicotinic ester **127** and involved a Negishi or a Fürstner type cross-coupling reaction with the appropriate alkylzinc or alkylmagnesium halide, respectively, and subsequent ester cleavage. The 6-methylisonicotinic acids **76a–g** were prepared in a similar fashion via Negishi or Suzuki reaction of the 2-halo-6-methylisonicotinate **129**. 2-Alkyl-6-ethylnicotinic acids **77a–d** were prepared starting from ethyl 2,6-dichloroisonicotinate **131** by two consecutive Negishi couplings. Although the bulkier alkyl group was introduced first, the monochloroisonicotinates **132a–d** were usually contaminated with 10–25% of the corresponding double alkylation product. In the case of Negishi reactions employing 3-pentylzinc bromide, formation of a significant amount of the isomeric 2-(pentan-2-yl)isonicotinates clearly hampered the purification of the reaction products. Hence, Suzuki reaction and subsequent catalytic hydrogenation as employed to prepare 6-methylisonicotinic esters **130a–e** often gave better results when compared to a Negishi or Fürstner cross-coupling approach.

2-Amino-6-methylisonicotinic acids **78a,b** and **79** were prepared starting from 2-chloro-6-methyl- or 2,6-dichloroisonicotinic acid ester **134** or **136**, respectively. A large excess of the amine was usually employed when 2-chloro-6-methylisonicotinate **134** was reacted under thermal, noncatalyzed conditions. However, the use of a large excess of pyrrolidine led to the formation of the corresponding amide (e.g., **135b**) which then had to be cleaved under basic conditions to give the desired acid (**78b**). A much smaller excess of the amine could be used if the reaction was carried out under Pd-catalyzed Buchwald conditions (e.g., **135a**). In addition, 1 equiv of diethylamine was sufficient to efficiently convert *tert*-butyl 2,6-dichloroisonicotinate **136** to the corresponding monochloride **137**. 6-Chloropicolinate **137** was then treated with dimethylzinc to afford the 6-methyl substituted isonicotinate **138** which upon ester cleavage delivered the isonicotinic acid **79**.

In Vitro SAR Discussion. As mentioned in the introduction, we wanted to derive the overall lipophilicity and amphiphilic nature of our S1P₁ agonists such as **1** (Figure 1). In our previous studies,^{113,96} we learned that a large variety of polar side chains are tolerated in the 4-position of the phenyl ring of **1**. This offered a starting point to improve the

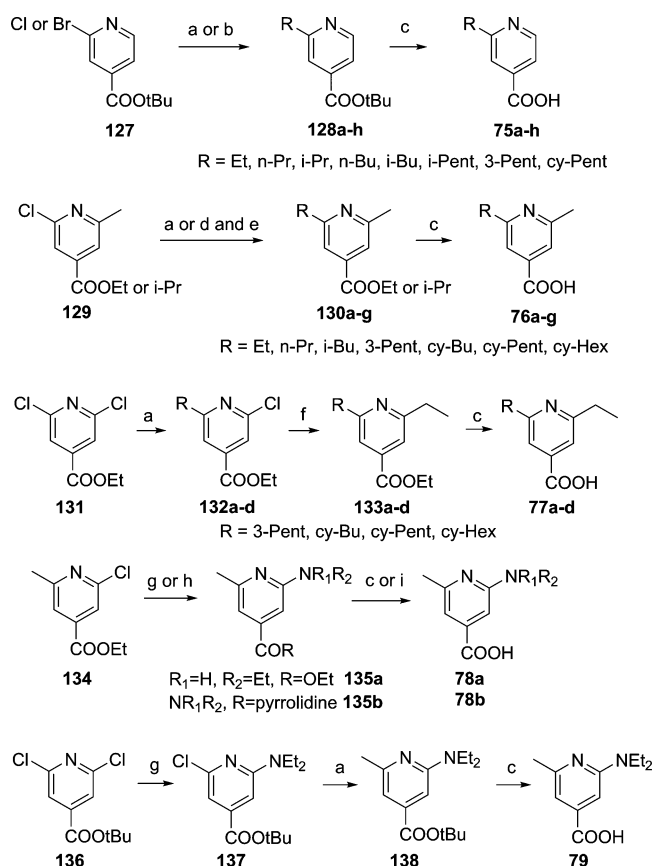
Scheme 5. Preparation of Picolinic Acids^a

^aReagents and conditions: (a) 2,4,6-trivinylcyclotriboroxanepyrindine complex, 2 M aq K_2CO_3 , $\text{Pd}(\text{PPh}_3)_4$, DME, 80 °C, 15 h, 49–99%; (b) KMnO_4 , acetone, water, rt, 3 days, quantitative crude; (c) EtOH , H_2SO_4 or TMSCl , 70 °C, 18 h, 40–95%; (d) 2,4,6-tri(2-methylpropenyl)cycloboroxanepyrindine complex, $\text{Pd}(\text{PPh}_3)_4$, 2 M aq K_2CO_3 , DME, 80 °C, 6 h, 37–68%; (e) H_2 , Pd/C , THF, EtOH , rt, 15 h, 95%; (f) 6–7 N aq HCl , 65 °C, 48 h, 64–94%; (g) cyclopentylzinc bromide, $\text{Pd}(\text{dppf})\text{Cl}_2$, dioxane, 65 °C, 18 h, 28–70% (h) 2 M aq LiOH , dioxane, 75 °C, 5 h, 15–88%; (i) Et_2NH , $\text{Pd}(\text{OAc})_2$, xanthphos, Cs_2CO_3 , dioxane, 90–110 °C, 18–24 h, 1–9%; (j) MeOH , H_2SO_4 , ammonium peroxydisulfate, 75 °C, 2 h, 19–28%; (k) Et_2NH , BuOH , reflux 16 h; EtOH , H_2SO_4 , 70 °C, 18 h; 6 N aq HCl , 65 °C, 18 h, 40% (three steps).

compounds' physicochemical properties. In a next step, we explored whether polar groups were also tolerated in the headgroup. In this account, we focus our discussion on the replacement of the thiophene by a pyridine. For the following SAR discussion the (*S*)-3-(3-ethyl-4-(2-hydroxy-3-(2-hydroxyacetamido)propoxy)-5-methylphenyl)-1,2,4-oxadiazole part was kept constant, as this moiety reliably brought about high affinity and selectivity for S1P_1 and high *in vivo* efficacy.

A set of 3-pyridines representing close analogues of compound **1** is compiled in Table 1. As evident from the clogP values listed in Table 1, even pyridines bearing large and

lipophilic substituents such as the 3-pentyl derivative **11** or the cyclopentylpyridines **12** and **14** are less lipophilic than thiophene **1**. The closest analogue of thiophene **1**, the isobutylpyridine **2**, had a lower clogP (2.75 vs 3.97) and showed an EC_{50} of 5.7 nM for S1P_1 , demonstrating that replacing the thiophene in **1** by a pyridine reduces the compound's lipophilicity while retaining its affinity for the S1P_1 receptor. The $\text{S1P}_{1/3}$ selectivity ratio is comparable for the two compounds **1** and **2**. The potency on S1P_1 and S1P_3 remained unaffected when the isobutyl chain in **2** was replaced by an *n*-propyl (**4**) or isopropyl (**5**) group. However, when the alkyl

Scheme 6. Preparation of Pyridine-4-carboxylic Acids^a

^aReagents and conditions: (a) alkylzinc bromide or chloride, or dialkylzinc, Pd(dppf)Cl₂, dioxane, 75–80 °C, 2–24 h, 21–84%; (b) alkylmagnesium bromide or chloride, Fe(acac)₃, THF, NMP, –75 °C, 1 h, rt, 1 h, 26–40%; (c) 6–7 M aq HCl, 60–95 °C, 3–20 h, 37% to quantitative; (d) 2,4,6-trialkylcycloboroxanepyrindine complex, PPh₃, Pd(PPh₃)₄, 2 M aq K₂CO₃, DME, 75–90 °C, 6–20 h, 50–95%; (e) H₂, Pd/C, THF, MeOH, rt, 5–16 h, 98%; (f) Et₂Zn, Pd(dppf)Cl₂, dioxane, 75–80 °C, 1–2 h, 79–82%; (g) diethylamine, 70–100 °C, 2–72 h, 95% to quantitative; (h) ethylamine, Cs₂CO₃, Pd(OAc)₂, Xanthphos, dioxane, 90 °C, 15 h, 44–47%; (i) 3 N aq NaOH, EtOH, 60 °C, 72 h; 87%.

group was further shortened to an ethyl (**6**) or a methyl (**7**) group, a significant potency loss was observed. As shown with compounds **8** and **10**, introducing a methyl group to position 5 of the pyridine clearly enhanced the compound's affinity for S1P₁ and S1P₃. An ethyl group in position 5 brought a further potency gain on both receptors, and several compounds (e.g., **13**, **14**, **16**) reached EC₅₀ values close to 0.1 nM for S1P₁ and about 25 nM for S1P₃. While the (*R*)-enantiomer of **2**, compound **3**, was significantly less potent in particular on S1P₁, the two enantiomers **8** and **9** showed an almost identical affinity profile. As shown with compound **11**, replacing the isobutyl chain in **8** by a 3-pentyl group had no significant effect on S1P₁ but significantly increased the compound's selectivity against S1P₃. Interestingly, the cyclopentyl derivative **12** was clearly less selective against S1P₃ when compared to its open chain analogue **11**. Replacing the 3-pentyl group in **11** by diethylamine to give aminopyridine **15** was well tolerated by S1P₁ and slightly improved the compound's affinity for S1P₃. Conversely, exchanging the cyclopentyl group in **12** by a pyrrolidine to give **17** resulted in a marked affinity loss on both

Table 1. SAR of 3-Pyridines (Nicotinic Acid Derivatives)

compd	R ₁	R ₂	clogP	EC ₅₀ [nM] ^a	
				S1P ₁	S1P ₃
1	<i>b</i>		3.97	0.7	320
2	isobutyl	H	2.75	5.7	1710
3^c	isobutyl	H	2.75	89	5740
4	<i>n</i> -propyl	H	2.41	4.6	1770
5	isopropyl	H	2.37	3.9	2900
6	ethyl	H	1.94	23	>10000
7	methyl	H	1.59	248	>10000
8	isobutyl	Me	3.06	1.0	318
9^c	isobutyl	Me	3.06	1.9	519
10	isopropyl	Me	2.68	0.1	429
11	pent-3-yl	Me	3.61	0.9	2630
12	cyclopentyl	Me	3.20	0.1	25
13	isobutyl	Et	3.42	0.1	29
14	cyclopentyl	Et	3.56	0.2	24
15	diethylamino	Me	2.83	2.1	616
16	diethylamino	Et	3.18	0.3	30
17	<i>N</i> -pyrrolidine	Me	2.24	23	1230
18	<i>N</i> -pyrrolidine	Et	2.59	2.8	176

^aEC₅₀ values as determined in a GTPγS assay using membranes of CHO cells expressing either S1P₁ or S1P₃.¹¹ EC₅₀ values represent the geometric mean of at least three independent measurements. ^bFor structure, see Figure 1. ^c(*R*)-Enantiomer.

receptors. As observed with the alkyl substituted pyridines, the 5-ethyl analogues **16** and **18** were clearly more potent on S1P₁ and S1P₃ when compared to their 5-methyl analogues **15** and **17**, respectively.

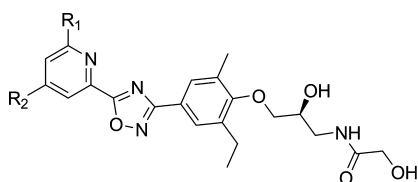
In a next step we turned our attention to replacing the thiophene in **1** by a number of substituted 2-pyridines bearing two substituents in either an ortho or a meta relationship. A few examples illustrating their SAR are listed in Tables 2 and 3. As judged by their clogP values, the most lipophilic 2-pyridines in

Table 2. SAR of 4,5- and 5,6-Disubstituted Pyridin-2-yl Derivatives (Picolinic Acid Derivatives)

compd	R ₁	X	Y	clogP	EC ₅₀ [nM] ^a	
					S1P ₁	S1P ₃
19	isobutyl	N	CH	3.02	0.2	154
20	isobutyl	CH	N	3.12	4.4	3080
21	cyclopentyl	N	CH	3.15	0.2	106
22	cyclopentyl	CH	N	3.26	0.8	860
23	diethylamino	N	CH	2.41	0.2	617
24	diethylamino	CH	N	2.52	9.5	5960

^aEC₅₀ values as determined in a GTPγS assay using membranes of CHO cells expressing either S1P₁ or S1P₃.¹¹ EC₅₀ values represent the geometric mean of at least three independent measurements.

Table 3. SAR of 4,6-Disubstituted Pyridin-2-yl Derivatives (Picolinic Acid Derivatives)



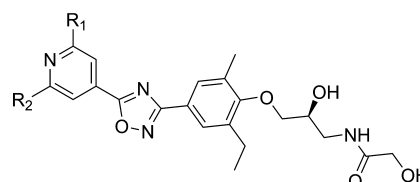
compd	R ₁	R ₂	clogP	EC ₅₀ [nM] ^a	
				S1P ₁	S1P ₃
25	isobutyl	methyl	3.12	0.5	125
26	methyl	isobutyl	3.12	1.3	139
27	cyclopentyl	methyl	3.26	0.2	866
28	methyl	cyclopentyl	3.26	0.9	258
29	diethylamino	methyl	2.89	1.3	429
30	methyl	diethylamino	2.52	54	5240

^aEC₅₀ values as determined in a GTPγS assay using membranes of CHO cells expressing either S1P₁ or S1P₃.¹¹ EC₅₀ values represent the geometric mean of at least three independent measurements.

Tables 2 and 3 (e.g., 22 and 28) are less lipophilic than thiophene 1. The 4,5-disubstituted 2-pyridines 19, 21, and 23 are all highly potent S1P₁ agonists with affinities and selectivity profiles very close to those of their 3-pyridine analogues 8, 12, and 15, respectively (Table 1). Compound 21 appears to be somewhat more selective against S1P₃ when compared to its analogue 12. The 5,6-disubstituted 2-pyridines 20, 22, and in particular 24 were less potent on both S1P receptors when compared to the corresponding 3-pyridine analogues 8, 12, and 15, respectively. In the series of the 4,6-disubstituted 2-pyridines, compounds 25–28 were highly potent S1P₁ agonists, and no relevant difference in the affinity profile was seen between isomers (i.e., 25 vs 26, 27 vs 28). This is in contrast to diethylaminopyridines 29 and 30 where the 6-methyl substituted isomer 30 was significantly less potent than 4-methyl isomer 29. A similar trend was observed between compounds 23 and 24.

The observation that incorporation of 4,6-disubstituted 2-pyridines led to potent and selective S1P₁ agonists prompted us to extend our study to 4-pyridine derivatives. Table 4 compiles some prototypical examples illustrating their SAR. As observed with the 2- and the 3-pyridine derivatives, the 4-pyridines were generally less lipophilic when compared to thiophene 1 and only examples incorporating rather large lipophilic substituents (e.g., as in 48, 51) showed a clogP as high as the one of thiophene 1. The least lipophilic compound of this series, the unsubstituted 4-pyridine 31, gave only weak activity on the S1P₁ receptor. However, the compound's affinity for S1P₁ very rapidly improved with increasing length of a 2-alkyl substituent, and the 2-ethylpyridine 33 already represented a highly potent and selective S1P₁ agonist. A further increase in the length of the alkyl group brought a less profound gain in S1P₁ affinity. The *n*-butyl (36) and in particular the isopentyl (38) derivative were in fact less potent when compared to their isomers 37 and 39, indicating that the ideal alkyl substituent may be branched but should not exceed more than three consecutive carbons in the chain. In general, there is a trend for compounds 32–40 to become more potent on S1P₃ with increasing size of the alkyl substituent. As for S1P₁, the isopentylpyridine 38 appears to mark the turn of this trend.

Table 4. SAR of 2- and 2,6-(Di)Substituted Pyridin-4-yl Derivatives (Isonicotinic Acid Derivatives)



compd	R ₁	R ₂	clogP	EC ₅₀ [nM] ^a	
				S1P ₁	S1P ₃
31	H	H	1.17	1110	>10000
32	methyl	H	1.59	86	>10000
33	ethyl	H	1.94	2.1	9950
34	<i>n</i> -propyl	H	2.41	0.7	1120
35	isopropyl	H	2.37	0.3	3940
36	<i>n</i> -butyl	H	2.87	1.8	700
37	isobutyl	H	2.75	0.2	757
38 ^b	isopentyl	H	3.21	8.4	3730
39	pent-3-yl	H	3.30	0.1	1340
40	cyclopentyl	H	2.88	0.1	736
41	ethyl	Me	2.36	0.3	1532
42	<i>n</i> -propyl	Me	2.83	0.4	310
43	isobutyl	Me	3.17	0.1	33
44	pent-3-yl	Me	3.71	0.2	46
45	cyclobutyl	Me	2.98	0.1	24
46	cyclopentyl	Me	3.30	0.2	47
47	cyclohexyl	Me	3.62	1.0	149
48	pent-3-yl	Et	4.07	0.2	3.6
49	cyclobutyl	Et	3.34	0.1	5.0
50	cyclopentyl	Et	3.65	0.3	2.4
51	cyclohexyl	Et	3.98	2.0	25
52	ethylamino	Me	2.07	0.4	868
53	diethylamino	Me	2.93	0.6	352
54	<i>N</i> -pyrrolidinyl	Me	2.34	1.9	307

^aEC₅₀ values as determined in a GTPγS assay using membranes of CHO cells expressing either S1P₁ or S1P₃.¹¹ EC₅₀ values represent the geometric mean of at least three independent measurements.
^bRacemate.

Compounds 41–47 bearing an additional 6-methyl substituent were all highly potent S1P₁ agonists, indicating that the additional methyl group was well tolerated by this receptor. The 2-ethyl-6-methyl derivative 41 appears to be slightly more potent when compared to the corresponding des-methyl analogue 33. In the case of the *n*-propylpyridine 42, the affinity gain on S1P₁ was not significant. The additional 6-methyl group markedly increased the activity of compounds 41–47 on S1P₃ (compare 44 with 39, 46 with 40). A further gain in potency on S1P₃ was observed when the 6-methyl group was replaced by an ethyl group. As a consequence, compounds 48–51 were almost equipotent on both S1P receptors.

The activity of the alkylaminopyridines 52 and 53 on S1P₁ was nearly identical to the one of the corresponding alkylpyridine analogues 42 and 44, while the pyrrolidine derivative 54 was less potent than its cyclopentane analogue 46. On the other hand, the aminopyridines had lower clogP values and showed a clearly reduced affinity for S1P₃ when compared to the corresponding alkylpyridines (compare 53 with 44 or 54 with 46), making them a particularly attractive subclass of compounds.

In brief, all substituted pyridine isomers constituted potent S1P₁ agonists. More specifically, we have seen that in the 3-

pyridine series potent S1P₁ agonists were obtained if the substituent at position 6 of the pyridine consisted of at least three carbon atoms. An additional substituent in position 5 improved the affinity for both S1P₁ and S1P₃ at least 5-fold. In the series of the 2-pyridine derivatives, compounds incorporating a 4,5-, a 5,6-, or a 4,6-substitution pattern have been studied. While the 4,5- and the 4,6-disubstituted 2-pyridine compounds led to highly potent S1P₁ agonists, compounds with a 5,6-disubstitution pattern were slightly less active. In the 4-pyridines, an ethyl substituent in position 2 was sufficient to obtain highly potent S1P₁ agonists. A larger substituent in position 2 and/or an additional substituent in position 6 mainly improved the compound's affinity for S1P₃. The amino-4-pyridine derivatives had lower clogP values and were usually more selective against S1P₃ when compared to the corresponding alkylpyridine analogues.

Pharmacokinetics and Pharmacodynamics. In the body, S1P₁ agonists lead to sequestration of circulating lymphocytes to lymph nodes and lymphoid tissue. Blood lymphocyte numbers can be measured easily by means of hemocytometry and therefore represent an attractive biomarker to assess a compound's in vivo activity. Hence, the ability to reduce the blood lymphocyte count (LC) was studied with several of the highly potent S1P₁ agonists discussed above. While a high potency on S1P₁ was required for a compound to be included in the pharmacodynamic experiment, we did not restrict these studies to highly selective compounds only in order to obtain a representative data set for each pyridine isomer subseries. The compounds were administered orally at a dose of 10 mg/kg to Wistar rats and the blood LC was measured shortly before and 3, 6, and 24 h after compound administration. A LC reduction of $\geq 60\%$ was considered to be the maximal effect observable under the experimental conditions. A sustained 24 h plasma exposure resulting in sustained maximal LC reduction appeared desirable from a compound efficacy and safety perspective. In humans, a transient heart rate reduction after oral dosing of (selective) S1P₁ receptor agonists has been observed.^{8,114–116} Model studies in guinea pigs suggested that the S1P₁ receptor is rapidly desensitized upon activation by a S1P₁ receptor agonist and sustained plasma concentrations of the agonist protect the animal from second dose effects on heart rate.¹¹⁷ Table 5 summarizes our results.

At 3 h all compounds tested displayed maximal LC reduction, indicating a rapid onset of action irrespective of the pyridine isomer series the compound belongs to. In general, maximal LC reduction was maintained for 24 h. Notable exceptions were the two closely related 2-pyridines **26** and **28** and the 4-pyridine **39**. Although these three compounds were highly potent on S1P₁, they did not significantly lower LC at 24 h.

In the LC experiments with the cyclopentyl substituted 2-pyridines **21** and **28**, compound concentrations were determined in plasma (Table 6). The data show a good PK–PD correlation. In line with its long duration of action, compound **21** still showed nearly micromolar plasma concentrations at 24 h. On the other hand, compound **28** reached micromolar concentrations 3 and 6 h after administration but was more rapidly cleared and almost completely removed from circulation at 24 h. Metabolic stability testing using rat liver microsomes (RLM) revealed that compound **28** is less stable when compared to the corresponding isomer **21** (Tables 5 and 6), explaining its shorter duration of action in

Table 5. Effect on Blood LC after Oral Administration of 10 mg/kg to Wistar Rats and in Vitro Intrinsic Clearance Measured in Rat Liver Microsomes

compd	type	% LC after dose			CL _{int} in RLM ^a [(μL/min)/mg protein]
		3 h	6 h	24 h	
8^b	3-pyridine	–68	–61	–69	nd
11	3-pyridine	nd	nd	nd	253
12	3-pyridine	–75	–81	–85	8
13^b	3-pyridine	–55	–61	–66	nd
15	3-pyridine	–59	–61	–67	56
19	2-pyridine	–70	–74	–65	29
21	2-pyridine	–68	–68	–71	29
25	2-pyridine	–69	–74	–61	17
26	2-pyridine	–64	–67	–20	48
27	2-pyridine	–67	–69	–50	21
28	2-pyridine	–67	–63	9	85
39	4-pyridine	–71	–71	7	112
40	4-pyridine	–70	–73	–71	40
41	4-pyridine	–67	–74	–81	2.0
42	4-pyridine	–61	–67	–73	5.7
43	4-pyridine	–62	–65	–71	27
44	4-pyridine	–59	–63	–53	65 ^c
46	4-pyridine	–67	–73	–76	17 ^d
47	4-pyridine	–57	–59	–67	nd
50	4-pyridine	–62	–67	–68	nd
51	4-pyridine	–66	–65	–70	nd
53	4-pyridine	–64	–64	–70	1.0 ^e
54	4-pyridine	–73	–76	–82	nd

^aIntrinsic clearance in the presence of rat liver microsomes (RLM) at 1 μM compound concentration. ^bCompound tested as racemate. ^cIntrinsic clearance at 0.5 and 0.1 μM **44** was 107 and 233 μL/(min mg), respectively. ^dIntrinsic clearance at 0.5 and 0.1 μM **46** was 20 and 12 μL/(min mg), respectively. ^eIntrinsic clearance at 0.5 and 0.1 μM **53** was 5 and 13 μL/(min mg), respectively.

Table 6. Effect on Lymphocyte Count (LC) and Plasma Concentration after Oral Administration of 10 mg/kg Compounds **21 and **28** to Wistar Rats (*n* = 6)**

compd	3 h		6 h		24 h		CL _{int} in RLM ^a [(μL/min)/mg protein]
	plasma concn [nM]	LC [%]	plasma concn [nM]	LC [%]	plasma concn [nM]	LC [%]	
21	4910	–68	3690	–68	800	–71	29
28	1890	–67	1070	–63	9	9	85

^aIntrinsic clearance determined in rat liver microsomes at 1 μM compound concentration. Experiments at 0.5 and 0.1 μM gave comparable results indicating that metabolic processes are not saturated under the assay conditions.

vivo. Similar differences observed between the microsomal stability of the two isobutyl-2-pyridines **19** and **26** may also explain the corresponding differences in the compounds' PD profiles. Rat microsomal stability data revealed that the pent-3-ylpyridine **39** is less stable than its cyclopentyl analogue **40** (Table 5), offering an explanation of why compound **39** showed a shorter duration of action relative to analogue **40**. A similar difference in microsomal stability between pent-3-ylpyridines **11** and **44** and their cyclopentyl analogues **12** and **46**, respectively, indicated that the pent-3-yl residue in general leads to metabolically less stable compounds than the cyclopentyl analogues (Table 5).

Table 7. Pharmacokinetic Data of Compounds 44, 46 and 53

compd	species ^b	pharmacokinetic parameters ^a								
		oral dosing				intravenous dosing				
		dose [mg/kg]	C_{\max} [ng/mL] (t_{\max} [h])	AUC_{0-24h} [ng h/mL]	F [%]	dose [mg/kg]	AUC_{0-inf} [ng h/mL]	Cl [(mL/min)/kg]	V_{ss} [L/kg]	$t_{1/2}$ [h]
44	rat	10	804 (0.5)	1760	126 ^c	0.5	69.9	120	3.4	0.6
		1	BLQ							
46	rat	1	49 (3.0)	436	39	0.5	566	16	7.3	5.8
53	rat	10	1090 (2.0)	16500 ^d	52	1	3200	5.4	3.5	8.3
	dog	3	546 (1.0)	3250	32	1	3350	5.0	1.2	3.4
	cynomolgus	1	97 (2.0)	866	16	1	5450	3.1	1.1	4.8

^a F = bioavailability; C_{\max} = maximal concentration reached in plasma; t_{\max} = time at which C_{\max} was reached, $t_{1/2}$ = half-life determined with iv experiment, Cl = clearance, V_{ss} = volume of distribution. ^bOral dosing in three Wistar rats, two Beagle dogs, or three cynomolgus monkeys; iv dosing in two Wistar rats, two Beagle dogs, or three cynomolgus monkeys; for experimental details, see Supporting Information. ^cBioavailability exceeds 100% because of saturation of the metabolism of compound 44 in the 10 mg/kg oral dose experiment. ^d AUC_{0-32h} .

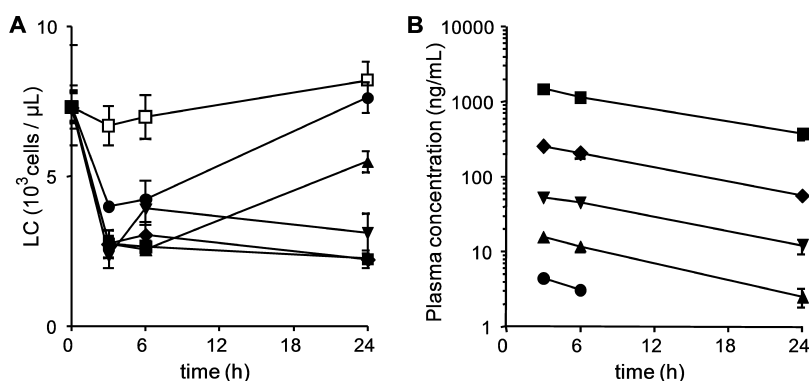


Figure 2. LC (A) and plasma concentrations (B) of a dose response experiment with amino pyridine 53 in Wistar rats: □, vehicle ($n = 5$); ●, 0.1 mg/kg ($n = 5$); ▲, 0.3 mg/kg ($n = 5$); ▼, 1 mg/kg ($n = 5$); ◆, 3 mg/kg ($n = 5$); ■, 10 mg/kg ($n = 5$). (A) LC fully recovered 96 h after administration of all doses tested (data not shown). (B) 24 h after administration of 0.1 mg/kg and 96 h after administration of 0.3, 1, 3, and 10 mg/kg plasma concentrations were below the limit of quantification (BLQ) of 1.5 ng/mL.

Interestingly, the enhanced in vitro clearance of 65 ($\mu\text{L}/\text{min}$)/mg of the pent-3-yl derivative 44 was not clearly reflected in a significantly shorter duration of action in vivo. Pharmacokinetic assessment of 44 revealed that this compound was indeed cleared very rapidly from the body with a clearance value in the range of liver blood flow (Table 7). When the compound was administered orally at a dose of 1 mg/kg, no significant exposure in the plasma was detected. At a dose of 10 mg/kg, however, plasma concentrations reached 804 ng/mL and therefore clearly exceeded dose proportionality, suggesting possible saturation of compound metabolism. A closer inspection of the in vitro clearance revealed that the intrinsic clearance of compound 44 was already saturated under the assay conditions chosen (1 μM compound concentration). Repeating the microsomal stability assay at lower concentrations of 44 showed that the unsaturated in vitro clearance is >250 ($\mu\text{L}/\text{min}$)/mg (see Table 5). From substrate depletion studies using RLM the Michaelis constant (K_M) was estimated to be <0.5 μM (see Supporting Information). Pent-3-ylpyridine 44 therefore is best characterized as a high clearance compound for which the metabolic process is saturated at relatively low concentrations. The pharmacokinetic behavior of the pent-3-yl derivative 44 is in clear contrast to the one of its close analogues 46 and 53 (Table 7). Cyclopentyl analogue 46 not only had a lower in vitro and in vivo clearance than 44 but also showed no sign of blocking its own metabolism in vitro at concentrations up to 1 μM (Tables 5 and 7). Interestingly, replacing the pent-3-yl group in 44 by a diethylamino moiety to

give 53 completely abolished the metabolic instability of 44. RLM data of 53 indicated very low intrinsic clearance ($K_M = 0.06 \mu\text{M}$), although saturation was occurring at 1 μM (Table 5). A PK experiment in the rat revealed that compound 53 was rapidly and well absorbed, slowly cleared, and easily distributed into tissue (Table 7).

In brief, several compounds listed in Table 4 meet the requirements we set forth for a compound to qualify for more detailed studies. For instance, compounds 15, 40, 41, 42, 53 were all highly potent on S1P₁, displayed an EC_{50} of >250 nM on S1P₃, and were >100-fold selective against this receptor. In particular, the last three compounds were further characterized by a very low in vitro clearance and produced a maximal LC reduction in the rat lasting for at least 24 h. In the following paragraphs we chose compound 53 to illustrate the additional experiments that were carried out to assess the compounds' viability as candidates for preclinical development.

First, a clear dose response of compound 53 on LC could be established in the rat (Figure 2A). At a dose of 0.1 mg/kg, aminopyridine 53 showed a substantial reduction of LC at 3 and 6 h. Maximal LC count reduction was achieved for 6 h with a dose of 0.3 mg, and a dose of 3 mg/kg maximally reduced LC for at least 24 h. At all doses studied, LC completely recovered within 96 h after compound administration. As shown in Figure 2B, plasma concentrations were dose proportional over the dose range of 0.1–10 mg/kg at each time point measured. The exposure data further suggest that a plasma concentration of

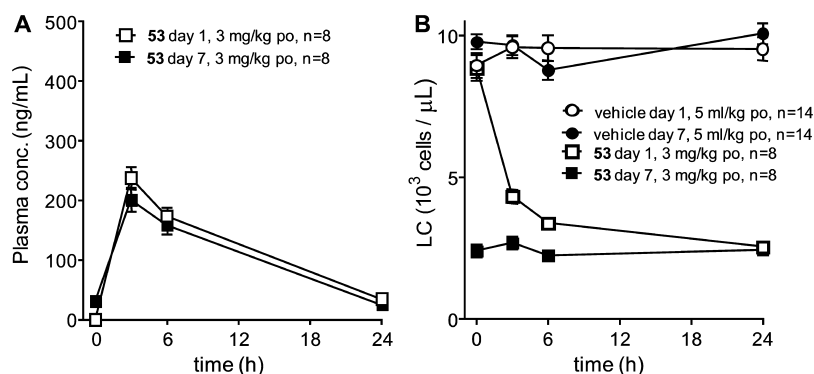
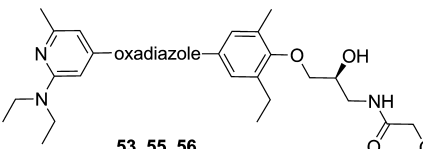
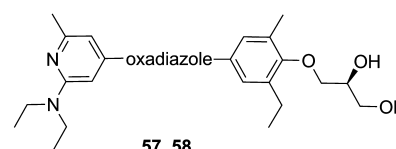







Figure 3. Plasma concentrations (A) and blood LC (B) at days 1 and 7 after repeated daily dosing of 3 (mg/kg)/day compound 53 to Wistar rats.

Table 8. Plasma and Brain Concentrations of Compounds 53, 55, 56, 57, and 58 after Oral Administration to Wistar Rats^a

	
53, 55, 56	57, 58

comp ound	oxadiazole	clogP	2 h		6 h		24 h	
			plasma [ng/mL]	brain [ng/mL]	plasma [ng/mL]	brain [ng/mL]	plasma [ng/mL]	brain [ng/mL]
53		2.93	2170	269	1775	653	412	487
55		2.93	875	140	918	369	230	301
56		3.00	815	46	510	52	8	3
57		3.86	608	435	254	271	22	34
58		3.93	820	600	315	215	4	2

optimal for brain penetration.^{120,122} In view of these observations, aminopyridine **53** clearly classified as a compound with a low potential to cross the BBB, as it has a PSA of 134 Å², a M_w of 497, and three hydrogen bond donors and scored only 2.9 out of 6 points using the CNS MPO tool. On the other hand, the smallest cross section of the aminopyridine **53** as calculated according to Seelig et al.¹²⁷ is 56 Å² and thus is clearly below the proposed maximal projection area of 80 Å² still allowing good permeability.¹¹⁸ Plasma and brain concentrations of **53** were then measured after oral administration of 10 mg/kg to Wistar rats (Table 8). In line with the previous PK experiment, plasma concentration of **53** reached a C_{max} of 2170 ng/mL very rapidly and was still about 400 ng/mL at 24 h. Interestingly, high concentrations of **53** were also found in brain tissue. At first sight, this result appeared to contradict the above predictions. In the brain, however, C_{max} was reached much later ($t_{max} \approx 6$ h) than in plasma ($t_{max} = 2$ h). At 24 h brain concentrations were even slightly higher than in plasma, indicating that for compound **53** crossing the BBB is a rather slow process in both directions. Hence, the predictions were indeed correct in the sense that **53** crosses the BBB only slowly. The high and sustained plasma exposure of **53** compensated for the low permeability, i.e., the slow BBB passage of the compound. A similar result was obtained with the [1,2,4]oxadiazole isomer **55**. As with **53**, sustained high plasma concentrations allowed reaching significant brain concentrations despite slow permeation across the BBB. In contrast, the [1,3,4]oxadiazole **56** showed a more rapid decline of the plasma concentrations which clearly hampered building up of significant brain concentrations. (For a detailed discussion of property differences between oxadiazole isomers see Boström et al.¹²⁸) Compared to **53**, the two glycerol derivatives **57** and **58** are characterized by a reduced molecular weight (440) and PSA (105 Å²) and a smaller number of hydrogen bond donors. With a CNS MPO score of 3.2 and 3.6 the glycerol derivatives **57** and **58** respectively show a more desirable property profile for brain penetration. Indeed, brain concentrations of the two glycerol derivatives **57** and **58** followed plasma concentrations very closely confirming a more rapid crossing of the BBB of these two compounds. As a consequence of the enhanced permeability of **57** and **58**, significant brain concentrations were observed at 2 and 6 h after compound administration but no sustained 24 h brain exposure was achieved. The different behavior of compounds **53** and **56** with respect to brain penetration makes them interesting tools to assess the contribution of S1P₁ agonism to the treatment efficiency in diseases affecting the brain. In particular, the above studies suggest that compound **53** might be well suited for the treatment of diseases for which brain penetration is important, while compound **56** may be useful for diseases for which brain penetration is not necessary or even undesirable.

CONCLUSIONS

Aiming at reducing the lipophilicity of thiophene based S1P₁ agonists such as compound **1**, we replaced the thiophene in **1** by a 2-, 3-, or 4-pyridine. This led to the discovery of a new series of less lipophilic S1P₁ receptor agonists with all three pyridine isomers furnishing potent compounds. Increasing the size of a first alkyl or alkylamino substituent in the para- or meta-position to the oxadiazole improved the compound's affinity rapidly for the S1P₁ and to a lesser extent for the S1P₃ receptor. When a second alkyl group was attached to the

pyridine, the compound's potency on S1P₁ and S1P₃ was further enhanced. This time, the affinity gain was more pronounced on S1P₃. Representatives from all three pyridine isomer series maximally reduced blood LC for at least 24 h when administered at a dose of 10 mg/kg to Wistar rats. Enhanced in vitro clearance in rat liver microsomes may explain the shorter duration of action observed for compounds **26**, **28**, and **39**. Compound **44** showed a high clearance in vitro and in vivo. However, the metabolism was saturated at relatively low compound concentrations, explaining why at a dose of 10 mg/kg this compound was still able to maximally reduce LC for almost 24 h. From the above studies several compounds appeared interesting for further characterization, as they were less lipophilic than thiophene **1**, displayed high affinity and selectivity for S1P₁, and sustained maximal reduction of blood LC in the rat. As an example, aminopyridine **53** has a clogP which is 1 log unit lower than the one of thiophene **1**, showed an EC₅₀ of 0.6 nM on S1P₁, and was more than 500-fold selective against S1P₃. In addition, this compound was highly efficacious in reducing blood LC in Wistar rats. In the following, the aminopyridine **53** was chosen to illustrate the more detailed characterization process. A dose response experiment revealed that for **53** the ED₅₀ for maximal LC reduction at 24 h after administration is about 0.5 mg/kg in the rat. Furthermore, compound **53** showed favorable PK properties in the Wistar rat and Beagle dog, showed acceptable PK behavior in the cynomolgus monkey, and is considered suitable for once daily dosing in humans. High concentrations of **53** were found in rat brain tissue after oral administration, making this compound an interesting candidate for further studies in particular in CNS related diseases or disorders.

EXPERIMENTAL SECTION

Chemistry. All reagents and solvents were used as purchased from commercial sources (Sigma-Aldrich, Switzerland; Lancaster Synthesis GmbH, Germany; Acros Organics, USA). Moisture sensitive reactions were carried out under an argon atmosphere. Progress of the reactions was followed either by thin-layer chromatography (TLC) analysis (Merck, 0.2 mm silica gel 60 F₂₅₄ on glass plates) or by LC–MS.

LC–MS parameters were the following: Finnigan MSQ plus or MSQ surveyor (Dionex, Switzerland), with HP 1100 binary pump and DAD (Agilent, Switzerland), column Zorbax SB-AQ, 3.5 μm, 120 Å, 4.6 mm × 50 mm (Agilent), gradient 5–95% acetonitrile in water containing 0.04% of trifluoroacetic acid within 1 min, flow of 4.5 mL/min. t_R is given in min. The asterisk (*) denotes basic conditions: gradient 5–95% acetonitrile in water containing 0.04% of ammonium hydroxide within 1 min, flow of 4.5 mL/min; 40 °C. t_R is given in min. UV detection was at 230, 254, and 280 nm.

Purity of all final compounds was checked by an additional LC–MS analysis on a Waters Acquity UPLC system equipped with an ACQ-PDA detector, an ACQ-ESL detector, and an ACQ-SQ detector: column ACQUITY UPLC BEH C18 1.7 μm, 2.1 mm × 50 mm; gradient 2–98% acetonitrile containing 0.045% formic acid in water containing 0.05% formic acid over 1.8 min; flow of 1.2 mL/min; 60 °C. According to these LC–MS analyses, final compounds showed a purity of >95% (UV at 230 and at 214 nm).

Chiral integrity was proven by HPLC (chiral stationary phase). Hardware was from UltiMate instrument series (Dionex): HPG-3200SD binary pump, WPS-3000 autosampler, TCC-3200 thermostated column compartment, DAD-3000 detector, SRD-3400 degasser, ValveMate 2 (Gilson) solvent valves. Column, solvent, and retention time (t_R) are as indicated, DEA = diethylamine, TFA = trifluoroacetic acid, at 25 °C, flow 1 mL/min. No racemization was observed during the synthesis of the target compounds.

LC–HRMS parameters were the following: analytical pump Waters Acquity binary, Solvent Manager, MS, SYNAPT G2 MS, source

temperature of 150 °C, desolvation temperature of 400 °C, desolvation gas flow of 400 L/h; cone gas flow of 10 L/h, extraction cone of 4 RF; lens 0.1 V; sampling cone 30; capillary 1.5 kV; high resolution mode; gain of 1.0, MS function of 0.2 s per scan, 120–1000 amu in full scan, centroid mode. Lock spray: keucine enkephalin, 2 ng/mL (556.2771 Da), scan time of 0.2 s with interval of 10 s and average of 5 scans; DAD: Acquity UPLC PDA detector. Column was an Acquity UPLC BEH C18 1.7 μ m, 2.1 mm \times 50 mm from Waters, thermostated in the Acquity UPLC column manager at 60 °C. Eluents were the following: water + 0.05% formic acid; B, acetonitrile + 0.05% formic acid. Gradient was 2–98% B over 3.0 min. Flow was 0.6 mL/min. Detection was at UV 214 nm. For MS, t_R is given in min.

Purity of all target compounds was assessed using the two independent LC–MS methods described above: (1) a Zorbax SB-AQ, 5 μ m, 120 Å, 4.6 mm \times 50 mm (Agilent) column, eluting with a gradient of 5–95% acetonitrile in water containing 0.04% of trifluoroacetic acid, within 1 min, flow of 4.5 mL/min; (2) an ACQUITY UPLC BEH C18 1.7 μ m, 2.1 mm \times 50 mm column, eluting with a gradient of 2–98% acetonitrile containing 0.045% formic acid in water containing 0.05% formic acid over 1.8 min; flow of 1.2 mL/min. In addition, important compounds were analyzed by LC–HRMS as described above. Purity and identity of the target compounds were further corroborated by NMR spectroscopy, and chiral integrity was proven by HPLC using chiral stationary phases. No racemization/epimerization was observed during the synthesis of the target compounds. According to these LC–MS analyses, final compounds showed a purity of $\geq 95\%$ (UV at 230 and at 214 nm).

For NMR spectroscopy, instruments used were the following: Varian Oxford, ^1H (300 MHz) or ^{13}C (75 MHz); Bruker Avance II, 400 MHz UltraShield, ^1H (400 MHz), ^{13}C (100 MHz). Chemical shifts are reported in parts per million (ppm) relative to tetramethylsilane (TMS), and multiplicities are given as s (singlet), d (doublet), t (triplet), q (quartet), quint (quintuplet), h (hextet), hept (heptuplet), or m (multiplet). br = broad, and coupling constants are given in Hz. Several compounds have been prepared in a combinatorial library format on a 15–50 mmol scale. For those compounds ^1H NMR spectra were acquired using nondeuterated 10 mM DMSO stock solutions submitted for biological testing.¹²⁹ The solvent and water signals were suppressed by irradiation at 2.54 and 3.54 ppm, respectively. As a consequence, signal integrals close to those frequencies are not always accurate. The numbers of protons given in the description represent observed values.

Compounds were purified by flash column chromatography (CC) on silica gel 60 (Fluka Sigma-Aldrich, Switzerland), by preparative TLC glass plates coated with silica gel 60 F₂₅₄ (0.5 mm), by preparative HPLC (Waters XBridge Prep C18, 5 μ m, OBD, 19 mm \times 50 mm, or Waters X-terra RP18, 19 mm \times 50 mm, 5 μ m, gradient of acetonitrile in water containing 0.4% of formic acid, flow of 75 mL/min), or by MPLC (Labomatic MD-80-100 pump, linear UVIS-201 detector, column 350 mm \times 18 mm, Labogel-RP-18-5s-100, gradient 10% methanol in water to 100% methanol).

clogP values were calculated using an algorithm developed in-house and published online at openmolecules.org.¹³⁰

In Vitro Potency Assessment. Data (EC_{50}) are given as geometric mean values (X_{geo}) with geometric standard deviation (σ_g). The upper and lower 95% confidence limits are calculated as $X_{\text{geo}}\sigma_g^2$ and $X_{\text{geo}}/\sigma_g^2$, respectively (results not shown).

GTP γ S binding assays were performed in 96-well polypropylene microtiter plates in a final volume of 200 μ L. Membrane preparations of CHO cells expressing recombinant human S1P₁ or S1P₃ receptors were used. Assay conditions were 20 mM Hepes, pH 7.4, 100 mM NaCl, 5 mM MgCl₂, 0.1% fatty acid free BSA, 1 or 3 μ M GDP (for S1P₁ or S1P₃, respectively), 2.5% DMSO, and 50 pM ^{35}S -GTP γ S. Test compounds were dissolved and diluted and preincubated with the membranes, in the absence of ^{35}S -GTP γ S, in 150 μ L of assay buffer at room temperature for 30 min. After addition of 50 μ L of ^{35}S -GTP γ S in assay buffer, the reaction mixture was incubated for 1 h at room temperature. The assay was terminated by filtration of the reaction mixture through a Multiscreen GF/C plate, prewetted with ice-cold 50 mM Hepes, pH 7.4, 100 mM NaCl, 5 mM MgCl₂, 0.4% fatty acid free

BSA, using a cell harvester. The filter plates were then washed with ice-cold 10 mM Na₂HPO₄/NaH₂PO₄ (70%/30%, w/w) containing 0.1% fatty acid free BSA. Then the plates were dried at 50 °C and sealed, 25 μ L of MicroScint20 was added, and membrane-bound ^{35}S -GTP γ S was determined on the TopCount. Specific ^{35}S -GTP γ S binding was determined by subtracting nonspecific binding (the signal obtained in the absence of agonist) from maximal binding (the signal obtained with 10 μ M S1P). The EC_{50} of a test compound is the concentration of a compound inducing 50% of specific binding.

Intrinsic metabolic clearance (CL_{int}) was determined by substrate depletion experiments, with a default starting concentration of 1 μ M in the presence of 0.5 mg/mL rat liver microsomes (RLM) in 100 mM sodium phosphate buffer at pH 7.4. Incubations were initiated by the addition of an NADPH regenerating system containing D-glucose 6-phosphate, NADPH, and glucose 6-phosphate dehydrogenase. All incubations were conducted by shaking reaction mixtures under air at 37 °C. Aliquots (0.1 mL) were removed at 0, 2.5, 5, 10, and 15 min and terminated by addition to 0.1 mL of methanol on ice. After protein precipitation by centrifugation, the remaining concentrations were analyzed by liquid chromatography coupled to mass spectrometry (LC–MS/MS). CL_{int} was calculated from the concentration remaining versus time, fitted to a first order decay constant versus time. K_M values from substrate consumption experiments were determined by plotting the first order decay constants versus the substrate concentration on a linear–log plot and fitting the following equation:¹³¹

$$\text{decay constant} = \text{CL}_{\text{int}}[\text{S}] \rightarrow 0 \left\{ \frac{[\text{S}]}{[\text{S}] + K_M} \right\}$$

Male Wistar rats (RccHan:WIST) were obtained from Harlan (Venray, The Netherlands) and used for pharmacokinetic experiments after an acclimatization period of at least 7 days. The body weight of the rats was about 250 g at the day of the experiment. Two days prior to dosing, rats were anesthetized via inhalation of the gas anesthetic isoflurane (4–5% for induction and 1.5–3% for maintenance) in 100% O₂. Buprenorphine was dosed as analgesic at 0.03 mg/kg sc half an hour before the operation. Catheters were implanted into jugular vein and carotid artery under aseptic conditions to allow for multiple serial blood sampling. Animals foreseen for oral dosing did not undergo surgery, but blood samples were taken sublingually under light anesthesia with isoflurane. Compounds were administered intravenously via the tail vein at doses of 1 mg/kg body weight formulated as solutions in an aqueous mixed micellar vehicle based on phospholipids and bile acids (mixed micelles). Oral administration at doses of 10 mg/kg was performed by gavage. Oral formulations were dispersions prepared by addition of a DMSO stock solution of the compounds to succinylated gelatin (7.5% w/v) in water.

For pharmacokinetics in the dog, the pharmacokinetic profile of compound 53 was determined in the fasted male Beagle dog ($n = 2$ dogs). The compound was administered iv at a dose of 1 mg/kg as a solution in mixed micelles and orally at a dose of 3 mg/kg as a dispersion in succinylated gelatin (7.5% w/v) in water.

For pharmacokinetics in the monkey, the pharmacokinetic profile of compound 53 was determined in the male cynomolgus monkey ($n = 3$). The compound was administered iv as a short infusion at a dose of 1 mg/kg as a solution in mixed micelles and orally at a dose of 1 mg/kg as a dispersion in 0.25% methyl cellulose in water with 0.05% Tween 80.

For bioanalysis and pharmacokinetics, serial blood samples of 0.25 mL each were taken from each individual animal to obtain a complete concentration vs time profile per animal. Blood samples were taken predose and at 30 min and 1, 2, 3, 4, 6, 8, and 24 h postdose into vials containing EDTA as anticoagulant. For the iv applications, additional samples were obtained 2, 10, and 20 min after dosing. Plasma was prepared from blood samples by centrifugation and stored at –20 °C and analyzed for drug concentrations using LC–MS/MS after protein precipitation. Pharmacokinetic parameters were estimated with the WinNonlin software (Pharsight Corporation, Mountain View, CA, USA) using noncompartmental analysis.

For brain penetration in the rat, at 2, 6, and 24 h after dosing, male Wistar rats ($n = 2$) were anaesthetized with 5% isoflurane and sacrificed by opening the diaphragm. A blood sample was taken. Plasma was prepared, and the brain was slowly perfused with 10 mL of 0.9% NaCl through the carotid. The whole brain was then removed and homogenized in an equal volume of ice-cold 0.1 M Na phosphate buffer, pH 7.4, using a IKA-WERKE Ultra-Turrax T25 tissue homogenizer for 10 s, and the brain homogenate was snap frozen in liquid nitrogen. Drug concentrations were then determined as described for plasma, using a calibration curve from blanco brain homogenate.

The in vivo efficacy of the target compounds was assessed by measuring the circulating lymphocytes after oral administration of 3–100 mg/kg of a target compound to normotensive male Wistar rats. The animals were housed in climate-controlled conditions with a 12 h light/dark cycle and had free access to normal rat chow and drinking water. Blood was collected before and 3, 6, and 24 h after drug administration. Full blood was subjected to hematology using Beckman Coulter Ac-T 5diff CP (Beckman Coulter International SA, Nyon, Switzerland). The effect on lymphocyte count (% LC) was calculated for each animal as the difference between LC at a given time point and the predose value (=100%). All data are presented as the mean \pm SEM. Statistical analyses were performed by analysis of variance (ANOVA) using Statistica (StatSoft) and the Student–Newman–Keuls procedure for multiple comparisons. The null hypothesis was rejected when $p < 0.05$. Because of interindividual variability and the circadian rhythm of the number of circulating lymphocytes, a compound showing relative changes in the range of –20% to +40% is considered inactive. A lymphocyte count (LC) reduction in the range of –60% to –75% represents the maximal effect to be observed under the conditions of the experiment. For formulation, the compounds were dissolved in DMSO. This solution was added to a stirred solution of succinylated gelatin (7.5% w/v) in water. The resulting milky suspension containing a final concentration of 5% of DMSO was administered to the animals by gavage. A mixture of 95% of succinylated gelatin (7.5% w/v) in water and 5% of DMSO served as vehicle.

6-Isobutyl-5-methylnicotinic Acid (60). (a) Phosphoroxylchloride (183 mL, 2 mol) was heated at 90 °C, and a mixture of commercially available 2-methyl-2-butenenitrile (73 g, 0.9 mol) and DMF (154 mL, 2 mol) was added slowly while keeping the temperature at 100–110 °C. The mixture was stirred at 110 °C for 15 h, cooled to room temperature, and diluted with DCM (500 mL). The mixture was cooled at 0 °C and carefully quenched with water (500 mL). The phases were separated, and the aqueous phase was extracted with DCM (total of 800 mL). The combined organic extracts were dried (Na_2SO_4), filtered, and evaporated. The residue was crystallized from cyclohexane to provide 6-chloro-3-formyl-5-methylpyridine **89**¹⁰⁹ (28.3 g, 20%) as slightly yellow crystals. LC–MS: $t_R = 0.76$ min, $[M + 1]^+ = 156.14$. ^1H NMR ($\text{DMSO}-d_6$): δ 10.09 (s, 1 H), 8.78 (s, 1 H), 8.24 (s, 1 H), 2.43 (s, 3 H).

(b) A solution of **89** (10 g, 64 mmol) in formic acid (200 mL) was cooled at 0 °C, and an aqueous 50 wt % solution of H_2O_2 in water (9.6 mL, 360 mmol) was added at this temperature. The mixture was stirred at 0 °C for 15 h, carefully diluted with water (200 mL), and extracted with DCM (8 \times 100 mL). The combined organic extracts were washed with 1 M aqueous HCl (100 mL) (checked for remaining peroxide), dried (MgSO_4), filtered, and evaporated. The residue was dried to give 6-chloro-5-methylnicotinic acid (9.56 g, 87%). LC–MS: $t_R = 0.72$ min, $[M + 1]^+ = 172.0$. ^1H NMR (CDCl_3): δ 8.94 (d, $J = 2.0$ Hz, 1 H), 8.23 (d, $J = 2.3$ Hz, 1 H), 2.49 (s, 3 H).

(c) A solution of 6-chloro-5-methylnicotinic acid (13.85 g, 80.75 mmol) in dry EtOH (200 mL) containing some drops of concentrated H_2SO_4 was stirred at reflux for 2 days. The solution was cooled to room temperature, the solvent evaporated, the residue dissolved in EA (200 mL) and washed with a solution of saturated aqueous Na_2CO_3 (2 \times 80 mL), 1 M aqueous KHSO_4 (2 \times 80 mL), and brine (50 mL). The organic phase was dried over MgSO_4 , filtered, and evaporated to give 6-chloro-5-methylnicotinic acid ethyl ester **90** (12.65 g, 79%) as a solid. LC–MS: $t_R = 0.92$ min, $[M + 1]^+ = 200.10$. ^1H NMR (CDCl_3) δ

1.43 (t, $J = 7.0$ Hz, 3 H), 2.46 (s, 3 H), 4.43 (q, $J = 7.3$ Hz, 2 H), 8.16 (m, 1 H), 8.84 (d, $J = 2.0$ Hz, 1 H).

(d) To a solution of **90** (4.98 g, 24.9 mmol), 2,4,6-tri(2-methylpropenyl)cyclotriphosphazene complex (5.74 g, 17.7 mmol, prepared in analogy to a procedure given by F. Kerins and D. F. O'Shea¹⁰⁸), and PPh_3 (1.15 g, 4.4 mmol) in DME (60 mL), a solution of 2 M aqueous K_2CO_3 (20 mL) was added. The mixture was degassed and flushed with N_2 before $\text{Pd}(\text{PPh}_3)_4$ (460 mg, 0.4 mmol) was added. The mixture was stirred at 90 °C for 20 h before it was cooled to room temperature, diluted with EA (150 mL), and washed with saturated aqueous NaHCO_3 (2 \times 50 mL). The organic extract was dried over MgSO_4 , filtered, and evaporated. The crude product was purified by FC (SiO_2 , heptane–EA) to give 5-methyl-6-(2-methylpropenyl)nicotinic acid ethyl ester (3.98 g, 73%) as an orange oil. LC–MS: $t_R = 0.72$ min, $[M + 1]^+ = 220.15$. ^1H NMR (CDCl_3): δ 9.04 (d, $J = 1.9$ Hz, 1 H), 8.04 (d, $J = 1.6$ Hz, 1 H), 6.37 (s, 1 H), 4.41 (q, $J = 7.1$ Hz, 2 H), 2.34 (s, 3 H), 2.008 (s, 3 H), 2.006 (s, 3 H), 1.42 (t, $J = 7.1$ Hz, 3 H).

(e) 5-Methyl-6-(2-methylpropenyl)nicotinic acid ethyl ester (3.98 g, 18.2 mmol) was dissolved in THF (100 mL) and MeOH (100 mL). Pd/C (500 mg, 10% Pd) was added as a slurry in THF (5 mL), and the mixture was stirred under 1 atm of H_2 at room temperature for 15 h. The catalyst was filtered off and the filtrate was evaporated to give 6-isobutyl-5-methylnicotinic acid ethyl ester **91** (3.76 g, 93%) as a colorless oil. LC–MS: $t_R = 0.75$ min, $[M + 1]^+ = 222.15$. ^1H NMR (CDCl_3) δ 0.97 (d, $J = 6.8$ Hz, 6 H), 1.42 (t, $J = 7.3$ Hz, 3 H), 2.20 (hept, $J = 6.8$ Hz, 1 H), 2.38 (s, 3 H), 2.75 (d, $J = 7.0$ Hz, 2 H), 4.41 (q, $J = 7.3$ Hz, 2 H), 8.03 (d, $J = 1.8$ Hz, 1 H), 9.00 (d, $J = 2.0$ Hz, 1 H).

(f) A solution of **91** (3.75 g, 16.95 mmol) in 12.5% aqueous HCl (50 mL) was stirred at 65 °C for 24 h before the solvent was evaporated. The residue was dried under high vacuum to give 6-isobutyl-5-methylnicotinic acid hydrochloride **60** (3.55 g, 91%) as a white powder. LC–MS: $t_R = 0.57$ min, $[M + 1]^+ = 194.25$. ^1H NMR (CDCl_3): δ 9.23 (s, 1 H), 8.77 (s, 1 H), 3.17 (d, $J = 7.5$ Hz, 2 H), 2.62 (s, 3 H), 2.35 (hept, $J = 6.3$ Hz, 1 H), 1.09 (d, $J = 6.6$ Hz, 6 H).

6-Cyclopentyl-5-methylnicotinic Acid (61). (a) A solution of 5,6-dichloronicotinic acid (50 g, 260 mmol) and TMSCl (8.49 g, 781 mmol) in isopropanol (500 mL) was stirred at 60 °C for 18 h. The mixture was concentrated, and the residue was partitioned between EA and saturated aqueous NaHCO_3 solution. The organic extract was separated, dried over MgSO_4 , filtered, and concentrated. The crude product was purified on silica gel, eluting with heptane/EA 9:1 to give 5,6-dichloronicotinic acid isopropyl ester **92** (56.2 g, 92%) as a white solid. LC–MS: $t_R = 1.01$ min, $[M + 1]^+ = 233.89$. ^1H NMR ($\text{DMSO}-d_6$): δ 8.85 (d, $J = 2.0$ Hz, 1 H), 8.49 (d, $J = 2.0$ Hz, 1 H), 5.18 (hept, $J = 6.3$ Hz, 1 H), 1.35 (d, $J = 6.3$ Hz, 6 H).

(b) To a solution of **92** (2.15 g, 9.19 mmol) in THF (30 mL) were added NMP (3 mL) and $\text{Fe}(\text{acac})_3$ (162 mg, 0.459 mmol). The mixture was cooled to 0 °C before cyclopentylmagnesium bromide (9.2 mL of 2 M solution in diethyl ether) was added. The dark red to black reaction mixture was stirred at 0 °C for 1 h, then at room temperature for 18 h. The reaction was quenched carefully by adding water (30 mL), then extracted twice with EA (2 \times 60 mL). The combined organic extracts were dried over MgSO_4 , filtered, and concentrated. The crude product was purified by CC on silica gel, eluting with heptane/EA 9:1 to give isopropyl 5-chloro-6-cyclopentyl-nicotinate **93** (1.93 g, 78%) which was still contaminated with 5,6-dichloronicotinic acid isopropyl ester as a pale yellow oil. LC–MS: $t_R = 0.87$ min, $[M + 1]^+ = 268.2$. ^1H NMR ($\text{DMSO}-d_6$): δ 8.95 (d, $J = 1.8$ Hz, 1 H), 8.21 (d, $J = 1.9$ Hz, 1 H), 5.16 (hept, $J = 6.3$ Hz, 1 H), 3.66 (quint, $J = 7.8$ Hz, 1 H), 1.95–2.05 (m, 2 H), 1.74–1.88 (m, 4 H), 1.59–1.73 (m, 2 H), 1.34 (d, $J = 6.2$ Hz, 6 H).

(c) To a solution of the above **93** (1.93 g, 7.21 mmol) and $\text{Pd}(\text{dppf})\text{Cl}_2$ (60 mg, 74 μmol) in dioxane (30 mL) was added dimethylzinc (13.3 mL of a 1.2 M solution in toluene). The mixture was stirred at 75 °C for 18 h before the reaction was quenched at room temperature by carefully adding water (50 mL). The mixture was extracted twice with EA (2 \times 100 mL). The combined organic extracts were dried over MgSO_4 , filtered, and concentrated. The crude product was purified by CC on silica gel, eluting with heptane/EA 9:1 to give

isopropyl 6-cyclopentyl-5-methylnicotinate **94** (600 mg, 36%, containing the corresponding methyl ester as impurity) as a pale yellow oil. LC–MS: $t_R = 0.61$ min, $[M + 1]^+ = 248.31$. ^1H NMR (CDCl_3): δ 9.01 (s, 1 H), 7.99 (s, 1 H), 5.27 (hept, $J = 6.5$ Hz, 1 H), 3.40 (quint, $J = 7.5$ Hz, 1 H), 2.41 (s, 3 H), 1.95–2.07 (m, 2 H), 1.83–1.95 (m, 4 H), 1.65–1.78 (m, 2 H), 1.38 (d, $J = 6.2$ Hz, 6 H).

(d) A solution of the above **94** (600 mg, 2.43 mmol) in MeOH (8 mL) and 2 M aqueous LiOH (2 mL) was stirred at room temperature for 2 h. The mixture was concentrated, diluted with water (15 mL) and EA (40 mL), and then neutralized by adding 2 N aqueous HCl (approximately 2 mL). The organic phase was separated, and the aqueous phase was extracted three more times with EA (3 \times 40 mL). The organic extracts were combined, dried over MgSO_4 , filtered, concentrated, and dried to give the title compound (412 mg, 83%) as a pale yellow oil. LC–MS: $t_R = 0.35$ min, $[M + 1]^+ = 206.39$. ^1H NMR (CDCl_3): δ 9.14 (s, 1 H), 8.13 (s, 1 H), 3.45 (quint, $J = 7.3$ Hz, 1 H), 2.46 (s, 3 H), 2.01–2.11 (m, 2 H), 1.87–2.01 (m, 4 H), 1.64–1.84 (m, 2 H).

5-Cyclopentyl-4-methylpicolinic Acid (68). (a) Under argon, $\text{Pd}(\text{dppf})\text{Cl}_2$ (161 mg, 0.198 mmol) and cyclopentylzinc bromide (39 mL of a 0.5 M solution in THF) were added to a solution of ethyl 5-bromo-4-methylpicolinate **107** (4.73 g, 19.4 mmol) in dioxane (250 mL). The mixture was stirred at 65 °C for 2 h before another portion of $\text{Pd}(\text{dppf})\text{Cl}_2$ (161 mg, 0.198 mmol) and cyclopentylzinc bromide (19.5 mL) was added. Stirring was continued at 65 °C for 18 h. The mixture was cooled to room temperature before the reaction was quenched by carefully adding water (200 mL) and saturated aqueous NaHCO_3 solution (150 mL). The mixture was extracted six times with EA (6 \times 100 mL) and DCM (100 mL). The combined organic extracts were dried over MgSO_4 , filtered, and concentrated to give crude ethyl 5-cyclopentyl-4-methylpicolinate **113** (3.96 g) as an orange oil containing about 10% of the corresponding cyclopentyl ester. LC–MS: $t_R = 0.85$ min, $[M + 1]^+ = 234.05$. Cyclopentyl ester, LC–MS: $t_R = 0.95$ min, $[M + 1]^+ = 274.06$.

(b) A solution of **113** (3.96 g, 17.0 mmol) in dioxane (2 mL) and 2 M aqueous LiOH (50 mL) was stirred at 75 °C for 5 h. The mixture was cooled to room temperature before it was extracted with EA. The aqueous phase was acidified by adding 1 N HCl and then extracted with EA. The second organic extract was dried over MgSO_4 , filtered, and concentrated. The crude product was purified by MPLC (ODS-AQ), eluting with a gradient of methanol in water to give **68** (532 mg, 15%) as a beige solid. LC–MS: $t_R = 0.37$ min, $[M + 1]^+ = 206.31$. ^1H NMR ($\text{DMSO}-d_6$): δ 8.52 (s, 1 H), 7.92 (s, 1 H), 3.21–3.31 (m, 1 H), 2.44 (s, 3 H), 1.99–2.11 (m, 2 H), 1.75–1.88 (m, 2 H), 1.56–1.75 (m, 4 H).

2-Methyl-6-(pentan-3-yl)isonicotinic Acid (76d). (a) To a suspension of 2-chloro-6-methylisonicotinic acid (20.0 g, 117 mmol) in isopropanol (80 mL) was added H_2SO_4 (5 mL) dropwise. The mixture became warm (40 °C). The mixture was stirred for 24 h at room temperature, then at 90 °C for 28 h before the solvent was removed in vacuo. The residue was dissolved in diethyl ether (200 mL), washed with saturated aqueous NaHCO_3 solution (3 \times 50 mL) followed by brine (3 \times 50 mL), dried over Na_2SO_4 , filtered, and concentrated to give 2-chloro-6-methylisonicotinic acid isopropyl ester (21.0 g) as a colorless oil which slowly crystallizes. LC–MS: $t_R = 0.97$ min, $[M + 1]^+ = 214.05$. ^1H NMR (CD_3OD): δ 7.64 (s, 1 H), 7.61 (s, 1 H), 5.21 (hept, $J = 6.2$ Hz, 1 H), 2.54 (s, 3 H), 1.37 (d, $J = 6.3$ Hz, 6 H).

(b) A solution of 2-chloro-6-methylisonicotinic acid isopropyl ester (2.0 g, 9.36 mmol) in dioxane (75 mL) was degassed and put under argon before $\text{Pd}(\text{dppf})\text{Cl}_2$ (229 mg, 0.281 mmol) was added. At room temperature, a 0.5 M solution of 1-ethylpropylzinc bromide in THF (46.8 mL, 23.4 mmol) was added dropwise to the mixture. The mixture was stirred at 80 °C for 16 h before the reaction was quenched by adding ice-cold water (200 mL). A precipitate forms, and the mixture was diluted with EA (200 mL) and filtered through Celite. The filtrate was transferred into a separatory funnel. The organic phase was collected, and the aqueous phase was extracted with EA (120 mL). The combined organic extracts were dried over MgSO_4 , filtered, and concentrated. The crude product was purified by CC on silica gel,

eluting with heptane/EA 9:1 to 4:1 to give isopropyl 2-(1-ethylpropyl)-6-methylisonicotinate **130d** (1.6 g, 69%) as a yellow oil containing a few percent of isopropyl 2-methyl-6-(pentan-2-yl)-isonicotinate. LC–MS: $t_R = 0.79$ min, $[M + 1]^+ = 250.14$. ^1H NMR ($\text{DMSO}-d_6$): δ 0.70 (t, $J = 7.3$ Hz, 6 H), 1.33 (d, $J = 6.3$ Hz, 6 H), 1.58–1.70 (m, 4 H), 2.51 (s, 3 H), 2.55–2.63 (m, 1 H), 5.15 (hept, $J = 5.8$ Hz), 7.39 (s, 1 H), 7.49 (s, 1 H).

(c) A solution of **130d** (1.54 g, 6.18 mmol) in 25% aqueous HCl (60 mL) was stirred at 65 °C for 16 h. The solvent was removed in vacuo and the residue was dissolved in dioxane and concentrated again to give 2-(1-ethylpropyl)-6-methylisonicotinic acid **76d** as hydrochloride salt (1.70 g) in the form of a brownish solid. LC–MS: $t_R = 0.62$ min, $[M + 1]^+ = 208.52$. ^1H NMR (CD_3OD): δ 8.23 (s, 1 H), 8.19 (s, 1 H), 3.00 (d, $J = 7.4$ Hz, 2 H), 2.88 (s, 3 H), 2.10–2.25 (m, 1 H), 1.05 (d, $J = 6.6$ Hz, 6 H).

2-Cyclopentyl-6-methylisonicotinic Acid (76f). (a) Under argon, $\text{Pd}(\text{dppf})\text{Cl}_2$ (200 mg, 0.245 mmol) was added to a solution of 2-chloro-6-methylisonicotinic acid ethyl ester **129** (4.80 g, 24.0 mmol) in dioxane (60 mL). A solution of cyclopentylzinc chloride (50 mL, 24.0 mmol, ~2 M solution in THF) was added dropwise. The mixture was stirred at 75 °C for 2 h before it was cooled to room temperature, carefully diluted with water, and extracted twice with EA. The combined organic extracts were dried over MgSO_4 , filtered, and concentrated. The crude product was purified by CC on silica gel, eluting with heptane/EA 9:1 to give 2-cyclopentyl-6-methylisonicotinic acid ethyl ester **130f** (3.96 g, 71%) as an oil. LC–MS: $t_R = 0.72$ min, $[M + 1]^+ = 234.11$. ^1H NMR (CDCl_3): δ 7.56 (s, 1 H), 7.52 (s, 1 H), 4.42 (q, $J = 7.1$ Hz, 2 H), 3.18–3.29 (m, 1 H), 2.61 (s, 3 H), 2.06–2.18 (m, 2 H), 1.66–1.93 (m, 6 H), 1.43 (t, $J = 7.1$ Hz, 3 H).

(b) A solution of **130f** (3.96 g, 17.0 mmol) in 25% aqueous HCl (50 mL) was stirred at 75 °C for 16 h. The solvent was removed in vacuo and the remaining residue was dried under high vacuum to give **76f** as a hydrochloride salt (4.12 g, quantitative) in the form of a white solid. LC–MS: $t_R = 0.54$ min, $[M + 1]^+ = 206.08$. ^1H NMR ($\text{DMSO}-d_6$): δ 7.90 (s, 1 H), 7.85 (s, 1 H), 3.79 (s br, 2 H), 3.41–3.53 (m, 1 H), 2.73 (s, 3 H), 2.07–2.17 (m, 2 H), 1.63–1.89 (m, 6 H).

2-Cyclopentyl-6-ethylisonicotinic Acid (77c). The title compound (7.19 g) was prepared in analogy to **77a** (see Supporting Information). LC–MS: $t_R = 0.59$ min, $[M + 1]^+ = 220.00$. ^1H NMR ($\text{DMSO}-d_6$): δ 7.85 (s, 1 H), 7.83 (s, 1 H), 3.47–3.57 (m, 1 H), 3.02 (q, $J = 7.5$ Hz, 2 H), 2.06–2.16 (m, 2 H), 1.71–1.88 (m, 4 H), 1.64–1.71 (m, 2 H), 1.29 (t, $J = 7.5$ Hz, 3 H).

2-(Diethylamino)-6-methylisonicotinic Acid (79). (a) A mixture of *tert*-butyl 2,6-dichloroisonicotinate **136** (11.3 g, 45.4 mmol) and diethylamine (3.32 g, 45.4 mmol) was stirred at 100 °C for 72 h in a sealed vessel. The mixture was cooled to room temperature, diluted with water (50 mL), and extracted three times with EA (3 \times 100 mL). The combined organic extracts were dried over MgSO_4 , filtered, and concentrated. The crude product was purified by CC on silica gel, eluting with heptane/EA 9:1 to give *tert*-butyl 2-(diethylamino)isonicotinate **137** (13.3 g, quantitative) as a pale yellow oil. LC–MS: $t_R = 1.15$ min, $[M + 1]^+ = 285.05$. ^1H NMR ($\text{DMSO}-d_6$): δ 6.88 (d, $J = 0.9$ Hz, 1 H), 6.80 (d, $J = 0.8$ Hz, 1 H), 3.50 (q, $J = 7.0$ Hz, 4 H), 1.54 (s, 9 H), 1.11 (t, $J = 7.0$ Hz, 6 H).

(b) To solution of **137** (6.70 g, 23.5 mmol) and $\text{Pd}(\text{dppf})\text{Cl}_2$ (196 mg, 240 μmol) in dioxane (50 mL) under argon was added dimethylzinc (70.6 mL, 1 M solution in THF). The mixture was stirred at 75 °C for 16 h before it was cooled to room temperature. The reaction was quenched by carefully adding water. The mixture was extracted twice with EA. The combined organic extracts were dried over MgSO_4 , filtered, and concentrated. The crude product was purified by CC on silica gel, eluting with heptane/EA to give *tert*-butyl 2-(diethylamino)-6-methylisonicotinate **138** (2.54 g, 41%) as a yellow oil. LC–MS: $t_R = 0.81$ min, $[M + 1]^+ = 265.09$.

(c) A solution of **138** (2.53 g, 9.57 mmol) in 25% aqueous HCl (50 mL) was stirred at 80 °C for 24 h. The mixture was concentrated and dried under high vacuum to give the hydrochloride salt of **79** (2.45 g, quantitative) as a yellow oil. LC–MS: $t_R = 0.48$ min, $[M + 1]^+ = 209.42$. ^1H NMR ($\text{DMSO}-d_6$): δ 7.17 (s, 1 H), 6.97 (s, 1 H), 3.70 (q br, $J = 7.0$ Hz, 2 H), 2.56 (s, 3 H), 1.17 (t, $J = 6.8$ Hz, 6 H).

***N*-(*S*)-3-[2-Ethyl-4-(*N*-hydroxycarbamidomethyl)-6-methylphenoxy]-2-hydroxypropyl-2-hydroxyacetamide (80).** (a) To an ice-cold solution of H_2SO_4 (150 mL) in water (250 mL) was added 2-ethyl-6-methylaniline (15.0 g, 111 mmol). The solution was treated with ice (150 g) before a solution of NaNO_2 (10.7 g, 155 mmol) in water (150 mL) and ice (50 g) was added dropwise. The mixture was stirred at 0 °C for 1 h. Then 50% aqueous H_2SO_4 (200 mL) was added and stirring was continued at room temperature for 18 h. The mixture was extracted with DCM. The organic extracts were dried over MgSO_4 and evaporated. The crude product was purified by CC on silica gel, eluting with heptane/EA 9:1 to give 2-ethyl-6-methylphenol **81** (8.6 g, 57%) as a crimson oil. LC–MS: t_R = 0.89 min. ^1H NMR (CDCl_3): δ 7.03–6.95 (m, 2 H), 6.80 (t, J = 7.6 Hz, 1 H), 4.60 (s, 1 H), 2.64 (q, J = 7.6 Hz, 2 H), 2.25 (s, 3 H), 1.24 (t, J = 7.6 Hz, 3 H).

(b) A solution of 2-ethyl-6-methylphenol **81** (200 g, 1.47 mol) and hexamethylenetetraamine (206 g, 1.47 mol) in acetic acid (1600 mL) and water (264 mL) was heated to reflux. The condensate was removed using a Dean–Stark apparatus until about 1200 mL of condensate was collected. The reaction mixture was cooled to room temperature, and water (1000 mL) was added. The thick suspension was filtered and the collected solid was dried at 60 °C under vacuum (10 mbar) to give 3-ethyl-4-hydroxy-5-methylbenzaldehyde **82** (191 g, 79%) as an orange solid. LC–MS: t_R = 0.86 min, $[\text{M} + 1 + \text{CH}_3\text{CN}]^+ = 206.27$. ^1H NMR (CDCl_3): δ 9.84 (s, 1 H), 7.59 (s, 1 H), 7.57 (s, 1 H), 4.64 (s br, 1 H), 2.71 (q, J = 7.5 Hz, 2 H), 2.34 (s, 3 H), 1.30 (t, J = 7.5 Hz, 3 H).

(c) A solution of **82** (5.32 g, 32.4 mmol) and hydroxylamine hydrochloride (3.38 g, 48.6 mmol) in NMP (35 mL) was stirred for 3 h at 80 °C under microwave irradiation (300 W, continuous cooling).¹³² The mixture was diluted with water and extracted twice with diethyl ether. The organic extracts were washed with 2 N aqueous HCl, saturated aqueous NaHCO_3 solution, and brine. The organic extracts were combined, dried over Na_2SO_4 , filtered, and concentrated. The crude product was purified by CC on silica gel, eluting with heptane/EA 3:2 to give 3-ethyl-4-hydroxy-5-methylbenzonitrile **83** as a pale yellow solid (4.80 g, 92%). LC–MS: t_R = 0.90 min. ^1H NMR (CDCl_3): δ 1.24 (t, J = 7.6 Hz, 3 H), 2.26 (s, 3 H), 2.63 (q, J = 7.6 Hz, 2 H), 5.19 (s, 1 H), 7.30 (s, 2 H).

(d) To a solution of **83** (5.06 g, 31.4 mmol) in THF (80 mL) were added PPh_3 (9.06 g, 34.5 mmol) and (*R*)-glycidol (2.29 mL, 34.5 mmol). The mixture was cooled to 0 °C before DEAD in toluene (15.8 mL, 34.5 mmol) was added. The mixture was stirred for 18 h while warming up to room temperature. The solvent was evaporated and the crude product was purified by CC on silica gel, eluting with heptane/EA 7:3 to give (*S*)-3-ethyl-5-methyl-4-(oxiran-2-ylmethoxy)-benzonitrile **84** (5.85 g, 86%) as a yellow oil. LC–MS: t_R = 0.96 min, $[\text{M} + 42]^+ = 259.08$. ^1H NMR (CDCl_3): δ 7.38 (s, 1 H), 7.35 (s, 1 H), 4.12–4.19 (m, 1 H), 3.73–3.80 (m, 1 H), 3.36–3.42 (m, 1 H), 2.90–2.96 (m, 1 H), 2.68–2.77 (m, 3 H), 2.34 (s, 3 H), 1.26 (t, J = 7.6 Hz, 3 H).

(e) The above epoxide **84** (5.85 g, 26.9 mmol) was dissolved in 7 N NH_3 in methanol (250 mL), and the solution was stirred at 65 °C for 18 h. The solvent was evaporated to give crude (*S*)-4-(3-amino-2-hydroxypropoxy)-3-ethyl-5-methylbenzonitrile **85** (6.23 g, quantitative) as a yellow oil. LC–MS: t_R = 0.66 min, $[\text{M} + 1]^+ = 235.11$. ^1H NMR ($\text{DMSO}-d_6$): δ 7.54 (s, 2 H), 4.96 (s br, 1 H), 3.77–3.83 (m, 1 H), 3.68–3.77 (m, 2 H), 2.59–2.75 (m, 4 H), 2.28 (s, 3 H), 1.58 (s br, 2 H), 1.17 (t, J = 7.5 Hz, 3 H).

(f) To a solution **85** (6.23 g, 26.6 mmol) were added glycolic acid (2.43 g, 31.9 mmol), HOBt (4.31 g, 31.9 mmol), and EDC hydrochloride (6.12 g, 31.9 mmol). The mixture was stirred at room temperature for 18 h before it was diluted with saturated aqueous NaHCO_3 and extracted twice with EA. The combined organic extracts were dried over MgSO_4 , filtered, and concentrated. The crude product was purified by CC with DCM containing 8% of methanol to give (*S*)-*N*-[3-(4-cyano-2-ethyl-6-methylphenoxy)-2-hydroxypropyl]-2-hydroxyacetamide **86** (7.03 g, 90%) as a yellow oil. LC–MS: t_R = 0.74 min, $[\text{M} + 1]^+ = 293.10$. ^1H NMR (CDCl_3): δ 1.25 (t, J = 7.5 Hz, 3 H), 2.32 (s, 3 H), 2.69 (q, J = 7.5 Hz, 2 H), 3.48–3.56 (m, 3 H), 3.70–

3.90 (m, 3 H), 4.19 (s, br, 3 H), 7.06 (m, 1 H), 7.36 (s, 1 H), 7.38 (s, 1 H).

(g) To a solution of **86** (19.6 g, 67 mmol) in methanol (500 mL) were added hydroxylamine hydrochloride (9.32 g, 134 mmol) and NaHCO_3 (11.3 g, 134 mmol).¹³³ The resulting suspension was stirred at 65 °C for 18 h. The mixture was filtered, and the filtrate was concentrated. The residue was dissolved in water (20 mL) and EA (300 mL). The aqueous phase was separated and extracted three times with EA. The combined organic extracts were dried over MgSO_4 , filtered, concentrated, and dried to give the title compound **80** as a white solid (18.9 g, 87%). LC–MS: t_R = 0.51 min, $[\text{M} + 1]^+ = 326.13$. ^1H NMR ($\text{DMSO}-d_6$): δ 1.17 (t, J = 7.4 Hz, 3 H), 2.24 (s, 3 H), 2.62 (q, J = 7.4 Hz, 2 H), 3.23 (m, 1 H), 3.43 (m, 1 H), 3.67 (m, 2 H), 3.83 (s, 2 H), 3.93 (m, 1 H), 5.27 (s br, 1 H), 5.58 (s br, 1 H), 5.70 (s, 2 H), 7.34 (s, 1 H), 7.36 (s, 1 H), 7.67 (m, 1 H), 9.46 (s br, 1 H).

Method A: (*S*)-*N*-(3-(4-(5-(2-Cyclopentyl-6-ethylpyridin-4-yl)-1,2,4-oxadiazol-3-yl)-2-ethyl-6-methylphenoxy)-2-hydroxypropyl)-2-hydroxyacetamide (50). To a solution of 2-cyclopentyl-6-ethylisonicotinic acid hydrochloride **77c** (187 mg, 0.73 mmol) in DMF (20 mL) was added Hünig's base (284 mg, 2.19 mmol) followed by TBTU (210 mg, 0.65 mmol). The mixture was stirred at room temperature for 10 min before *N*-(*S*)-3-[2-ethyl-4-(*N*-hydroxycarbamidomethyl)-6-methylphenoxy]-2-hydroxypropyl-2-hydroxyacetamide **80** (251 mg, 0.77 mmol) was added. Stirring was continued at room temperature for 2 h. The mixture was diluted with EA (100 mL) and washed three times with saturated NaHCO_3 solution (2 \times 30 mL). The organic extract was concentrated, and the residue was dissolved in dioxane (20 mL). The mixture was stirred at 100 °C for 16 h before it was concentrated. The crude product was purified on preparative HPLC (Waters XBridge, gradient of MeCN in water containing 0.5% NH_3) to give **53** (281 mg, 76%) as beige wax. LC–MS: t_R = 0.66 min, $[\text{M} + 1]^+ = 509.28$. ^1H NMR (CDCl_3): δ 7.87 (s, 1 H), 7.86 (s, 1 H), 7.76 (s, 1 H), 7.73 (s, 1 H), 7.21 (t, J = 5.9 Hz, 1 H), 4.18–4.25 (m, 3 H), 3.82–3.93 (m, 2 H), 3.74–3.82 (m, 2 H), 3.60 (s br, 1 H), 3.49–3.58 (m, 1 H), 3.27–3.39 (m, 1 H), 2.95 (q, J = 7.5 Hz, 2 H), 2.75 (q, J = 7.5 Hz, 2 H), 2.39 (s, 3 H), 2.13–2.21 (m, 2 H), 1.73–1.95 (m, 6 H), 1.39 (t, J = 7.6 Hz, 3 H), 1.32 (t, J = 7.5 Hz, 3 H). ^{13}C NMR (CDCl_3): δ 174.5, 174.0, 168.8, 166.9, 164.3, 157.3, 137.6, 131.60, 131.55, 128.3, 126.6, 122.5, 116.8, 116.2, 74.2, 69.9, 62.1, 48.1, 42.3, 33.6, 31.4, 25.8, 22.8, 16.4, 14.8, 13.8. LC–HRMS: t_R = 2.21 min, $[\text{M} + \text{H}]^+ = 509.2764$, found = 509.2762.

Method B: *N*-(*S*)-3-[4-[5-(2-Diethylamino-6-methylpyridin-4-yl)]1,2,4]oxadiazol-3-yl]-2-ethyl-6-methylphenoxy)-2-hydroxypropyl-2-hydroxyacetamide (53). For an alternative, large scale synthesis, see Schmidt et al.¹³⁴ (a) A solution of 2-(diethylamino)-6-methylisonicotinic acid hydrochloride **79** (6.96 g, 28.4 mmol), Hünig's base (11.0 g, 85.3 mmol), and TBTU (9.13 g, 28.4 mmol) in DCM (100 mL) was stirred at room temperature for 10 min before 3-ethyl-*N*,4-dihydroxy-5-methylbenzimidamide (**5.52** g, 28.4 mmol) was added. Stirring was continued at room temperature for 1 h. The mixture was diluted with DCM, washed with saturated aqueous NaHCO_3 solution, dried over MgSO_4 , filtered, and concentrated. The residue was dissolved in dioxane (100 mL), and the mixture was stirred at 100 °C for 18 h. The solvent was evaporated and the crude product was purified by CC on silica gel, eluting with heptane/EA 5:1 to give 4-(5-(2-(diethylamino)-6-methylpyridin-4-yl)-1,2,4-oxadiazol-3-yl)-2-ethyl-6-methylphenol (6.40 g, 61%) as a yellow solid. LC–MS: t_R = 0.90 min, $[\text{M} + 1]^+ = 367.16$. ^1H NMR (CDCl_3): δ 7.85 (s, 2 H), 7.10 (s, 1 H), 7.01 (s, 1 H), 4.98 (s, 1 H), 3.63 (q, J = 7.0 Hz, 4 H), 2.73 (q, J = 7.4 Hz, 2 H), 2.49 (s, 3 H), 2.36 (s, 3 H), 1.33 (t, J = 7.5 Hz, 3 H), 1.25 (t, J = 6.9 Hz, 6 H).

(b) To a solution of 4-(5-(2-(diethylamino)-6-methylpyridin-4-yl)-1,2,4-oxadiazol-3-yl)-2-ethyl-6-methylphenol (580 mg, 1.58 mmol) in isopropanol (15 mL) and 3 M aqueous NaOH (3 mL) was added (*R*)-epichlorohydrin (439 mg, 4.75 mmol), and the mixture was stirred at room temperature for 24 h before another portion of (*R*)-epichlorohydrin (439 mg, 4.75 mmol) was added. Stirring was continued for 24 h. The mixture was diluted with EA (100 mL) and washed with saturated aqueous NaHCO_3 solution. The organic extract was dried over MgSO_4 , filtered, and concentrated. The crude product

was purified by CC on silica gel using heptane/EA 4:1 to give (S)-N,N-diethyl-4-(3-(3-ethyl-5-methyl-4-(oxiran-2-ylmethoxy)phenyl)-1,2,4-oxadiazol-5-yl)-6-methylpyridin-2-amine (450 mg, 67%) as a yellow oil. LC-MS: t_R = 0.94 min, $[M + 1]^+$ = 423.19. 1H NMR ($CDCl_3$): δ 7.88 (s, 1 H), 7.87 (s, 1 H), 7.10 (s, 1 H), 7.01 (s, 1 H), 4.18–4.27 (m, 1 H), 3.92 (d, J = 5.2 Hz, 2 H), 3.58–3.70 (m, 6 H), 2.77 (q, J = 7.6 Hz, 2 H), 2.50 (s, 3 H), 2.41 (s, 3 H), 1.33 (t, J = 7.5 Hz, 3 H), 1.25 (t, J = 7.0 Hz, 6 H).

(c) A solution of (S)-N,N-diethyl-4-(3-(3-ethyl-5-methyl-4-(oxiran-2-ylmethoxy)phenyl)-1,2,4-oxadiazol-5-yl)-6-methylpyridin-2-amine (5.70 g, 13.5 mmol) in 7 M NH_3 in MeOH (100 mL) was stirred in a sealed vessel at 60 °C for 24 h. The mixture was concentrated and the crude product was purified by CC on silica gel, eluting with DCM/MeOH 10:1 containing a small amount of 7 M NH_3 in MeOH to give (S)-1-amino-3-(4-(5-(2-(diethylamino)-6-methylpyridin-4-yl)-1,2,4-oxadiazol-3-yl)-2-ethyl-6-methylphenoxy)propan-2-ol (2.33 g, 39%) as a yellow oil. LC-MS: t_R = 0.70 min, $[M + 1]^+$ = 440.25. 1H NMR δ : 7.88 (s, 1 H), 7.87 (s, 1 H), 7.10 (s, 1 H), 7.00 (s, 1 H), 3.97–4.05 (m, 1 H), 3.85–3.93 (m, 2 H), 3.62 (q, J = 7.0 Hz, 4 H), 3.23 (s, 1 H), 3.03 (dd, J_1 = 12.8 Hz, J_2 = 3.8 Hz, 1 H), 2.93 (dd, J_1 = 12.8 Hz, J_2 = 7.0 Hz, 1 H), 2.77 (q, J = 7.5 Hz, 2 H), 2.49 (s, 3 H), 2.40 (s, 3 H), 1.32 (t, J = 7.5 Hz, 3 H), 1.24 (t, J = 7.0 Hz, 6 H).

(d) To a solution of (S)-1-amino-3-(4-(5-(2-(diethylamino)-6-methylpyridin-4-yl)-1,2,4-oxadiazol-3-yl)-2-ethyl-6-methylphenoxy)propan-2-ol (2.33 g, 5.30 mmol) in DCM (25 mL) were added glycolic acid (443 mg, 5.83 mmol) and HOBt (788 mg, 5.83 mmol). The mixture was stirred for 10 min before EDC-HCl (1.12 g, 5.83 mmol) was added. The mixture was stirred at room temperature for 1 h before it was diluted with water and brine. The mixture was extracted twice with EA. The combined organic extracts were dried over $MgSO_4$, filtered, and concentrated. The crude product was purified by CC on silica gel, eluting with DCM/MeOH 10:1 to give **56** (2.20 g, 83%) as a yellow foam. LC-MS: t_R = 0.78 min, $[M + 1]^+$ = 498.22. HPLC with chiral stationary phase (Chiralpak AD-H 250 mm \times 4.6 mm i.d., 5 μ m; 85% hexane, 15% ethanol containing 0.1% DEA): t_R = 12.3 min, 100% ((R)-enantiomer, t_R = 10.2 min). 1H NMR ($DMSO-d_6$): δ 7.79 (s, 2 H), 7.70 (t, J = 5.5 Hz, 1 H), 7.07 (s, 1 H), 6.99 (s, 1 H), 5.56 (t, J = 5.5 Hz, 1 H), 5.31 (d, J = 5.1 Hz, 1 H), 3.93–4.00 (m, 1 H), 3.84 (d, J = 5.2 Hz, 2 H), 3.70–3.81 (m, 2 H), 3.58 (q, J = 6.5 Hz, 4 H), 3.39–3.48 (m, 1 H), 3.20–3.29 (m, 1 H), 2.73 (q, J = 7.3 Hz, 2 H), 2.43 (s, 3 H), 2.35 (s, 3 H), 1.22 (t, J = 7.5 Hz, 3 H), 1.16 (t, J = 6.9 Hz, 6 H). LC-HRMS: t_R = 1.86 min, $[M + H]^+/z$ = 498.2716, found = 498.2717.

N-((S)-3-{2-Ethyl-4-[5-(6-isobutyl-5-methylpyridin-3-yl)-1,2,4]oxadiazol-3-yl}-6-methylphenoxy)-2-hydroxypropyl)-2-hydroxyacetamide (8). **8** was obtained with method A, using nicotinic acid **60**. Colorless resin (17 mg, 20%). LC-MS: t_R = 0.83 min, $[M + 1]^+$ = 483.23. HPLC with chiral stationary phase (Chiralpak AD-H 250 \times 4.6 mm i.d., 5 μ m; 80% heptane containing 0.05% DEA, 20% ethanol containing 0.05% DEA): t_R = 12.9 min, 98% ((R)-enantiomer, t_R = 11.3 min, 2%). 1H NMR ($DMSO-d_6$): δ 9.08 (d, J = 2.0 Hz, 1 H), 8.30 (d, J = 1.7 Hz, 1 H), 7.79 (s, 2 H), 7.72 (t, J = 5.8 Hz, 1 H), 5.58 (t, J = 5.7 Hz, 1 H), 5.33 (d, J = 5.2 Hz, 1 H), 3.93–4.01 (m, 1 H), 3.84 (d, J = 5.7 Hz, 2 H), 3.70–3.79 (m, 2 H), 3.40–3.48 (m, 1 H), 3.20–3.29 (m, 1 H), 2.69–2.78 (m, 4 H), 2.42 (s, 3 H), 2.34 (s, 3 H), 2.19 (hept, J = 6.8 Hz, 1 H), 1.22 (t, J = 7.5 Hz, 3 H), 0.94 (d, J = 6.6 Hz, 6 H). LC-HRMS: t_R = 1.93 min, $[M + H]^+/z$ = 483.2607, found = 483.2612.

N-((S)-3-[4-[5-(6-Cyclopentyl-5-methylpyridin-3-yl)]1,2,4]-oxadiazol-3-yl)-2-ethyl-6-methylphenoxy)-2-hydroxypropyl)-2-hydroxyacetamide (12). **12** was obtained with method A, using nicotinic acid **61**. Pale yellow oil (105 mg, 73%). LC-MS: t_R = 0.75 min, $[M + 1]^+$ = 495.31. 1H NMR ($CDCl_3$): δ 9.20 (d, J = 2.0 Hz, 1 H), 8.20 (d, J = 1.5 Hz, 1 H), 7.87 (s, 1 H), 7.85 (s, 1 H), 7.17 (t, J = 5.8 Hz, 1 H), 4.18–4.25 (m, 3 H), 3.76–3.93 (m, 3 H), 3.40–3.57 (m, 2 H), 2.74 (q, J = 7.5 Hz, 2 H), 2.49 (s, 3 H), 2.38 (s, 3 H), 2.00–2.11 (m, 2 H), 1.88–2.00 (m, 4 H), 1.70–1.81 (m, 2 H), 1.31 (t, J = 7.5 Hz, 3 H). LC-HRMS: t_R = 2.11 min, $[M + H]^+/z$ = 495.2607, found = 495.2609.

N-((S)-3-[4-[5-(5-Cyclopentyl-4-methylpyridin-2-yl)]1,2,4]-oxadiazol-3-yl)-2-ethyl-6-methylphenoxy)-2-hydroxypropyl)-

2-hydroxyacetamide (21). **21** was obtained with method A, using picolinic acid **68**. Beige resin (79 mg, 22%). LC-MS: t_R = 0.75 min, $[M + 1]^+$ = 495.34. 1H NMR (CD_3OD): δ 8.62 (s, 1 H), 8.18 (s, 1 H), 8.00 (t, J = 6.0 Hz, 1 H), 7.90 (s, 1 H), 7.87 (s, 1 H), 4.11–4.19 (m, 1 H), 4.04 (s, 2 H), 3.84–3.92 (m, 2 H), 3.63–3.71 (m, 1 H), 3.37–3.52 (m, 2 H), 2.81 (q, J = 7.5 Hz, 2 H), 2.55 (s, 3 H), 2.41 (s, 3 H), 2.13–2.24 (m, 2 H), 1.68–1.99 (m, 6 H), 1.32 (t, J = 7.5 Hz, 3 H). LC-HRMS: t_R = 1.95 min, $[M + H]^+/z$ = 495.2607, found = 495.2609.

N-((S)-3-(2-Ethyl-4-[5-[2-(1-ethylpropyl)-6-methylpyridin-4-yl]]1,2,4]oxadiazol-3-yl)-6-methylphenoxy)-2-hydroxypropyl)-2-hydroxyacetamide (44). **44** was obtained with method A, using isonicotinic acid **76d**. Pale yellow solid (836 mg, 34%). LC-MS: t_R = 0.94 min, $[M + 1]^+$ = 497.19. HPLC with chiral stationary phase (Chiralpak AD-H 250 mm \times 4.6 mm i.d., 5 μ m; 80% heptane containing 0.05% DEA, 20% ethanol containing 0.05% DEA): t_R = 9.5 min, 100% ((R)-enantiomer, t_R = 7.5 min). 1H NMR ($CDCl_3$): δ 7.90 (s, 1 H), 7.89 (s, 1 H), 7.75 (s, 1 H), 7.67 (s, 1 H), 6.99 (t, J = 5.5 Hz, 1 H), 4.19–4.27 (m, 3 H), 3.92 (dd, J_1 = 9.6 Hz, J_2 = 4.7 Hz, 1 H), 3.85 (dd, J_1 = 9.5 Hz, J_2 = 6.3 Hz, 1 H), 3.80 (ddd, J_1 = 13.8 Hz, J_2 = 6.0 Hz, J_3 = 2.5 Hz, 1 H), 3.54 (ddd, J_1 = 14.1 Hz, J_2 = 7.3 Hz, J_3 = 5.5 Hz, 1 H), 3.32 (d, J = 4.5 Hz, 1 H), 2.76 (q, J = 7.5 Hz, 2 H), 2.70 (s, 3 H), 2.55 (s br, 1 H), 2.41 (s, 3 H), 1.80 (quint, J = 7.4 Hz, 4 H), 1.33 (t, J = 7.6 Hz, 3 H), 0.85 (t, J = 7.4 Hz, 6 H). ^{13}C NMR ($CDCl_3$): δ 174.4, 173.8, 168.9, 166.7, 159.3, 157.3, 137.7, 131.6, 131.5, 128.4, 126.6, 122.5, 118.3, 116.9, 74.2, 70.0, 62.1, 51.5, 42.3, 28.1, 24.6, 22.9, 16.5, 14.8, 12.1. LC-HRMS: t_R = 2.01 min, $[M + H]^+/z$ = 497.2764, found = 497.2767.

N-((S)-3-[4-[5-(2-Cyclopentyl-6-methylpyridin-4-yl)]1,2,4]-oxadiazol-3-yl)-2-ethyl-6-methylphenoxy)-2-hydroxypropyl)-2-hydroxyacetamide (46). **46** was obtained with method A, using isonicotinic acid **76f**. Colorless oil (40 mg, 47%). LC-MS: t_R = 0.83 min, $[M + 1]^+$ = 495.30. HPLC with chiral stationary phase (Chiralpak AD-H 250 mm \times 4.6 mm i.d., 5 μ m; 80% heptane containing 0.05% DEA, 20% ethanol containing 0.05% DEA): t_R = 13.6 min, 100% ((R)-enantiomer, t_R = 10.5 min). 1H NMR ($CDCl_3$): δ 7.89 (s, 1 H), 7.88 (s, 1 H), 7.76 (s, 1 H), 7.73 (s, 1 H), 7.03 (t, J = 5.5 Hz, 1 H), 4.18–4.27 (m, 3 H), 3.91 (dd, J_1 = 9.6 Hz, J_2 = 4.7 Hz, 1 H), 3.85 (dd, J_1 = 9.5 Hz, J_2 = 6.3 Hz, 1 H), 3.80 (ddd, J_1 = 14.2 Hz, J_2 = 6.7 Hz, J_3 = 3.3 Hz, 1 H), 3.49–3.58 (m, 1 H), 3.38 (s br, 1 H), 3.25–3.35 (m, 1 H), 2.76 (q, J = 7.6 Hz, 2 H), 2.68 (s, 3 H), 2.40 (s, 3 H), 2.12–2.22 (m, 2 H), 1.73–1.95 (m, 6 H), 1.33 (t, J = 7.5 Hz, 3 H). ^{13}C NMR ($CDCl_3$): δ 174.4, 173.3, 168.9, 167.0, 159.2, 157.3, 137.7, 131.7, 131.6, 128.4, 126.7, 122.5, 118.2, 116.0, 74.1, 70.2, 62.2, 48.2, 42.3, 33.7, 25.8, 24.6, 22.9, 16.5, 14.8. LC-HRMS: t_R = 1.98 min, $[M + H]^+/z$ = 495.2607, found = 495.2600.

■ ASSOCIATED CONTENT

Supporting Information

Additional experimental details on the synthesis and characterization of all target compounds and all intermediates not given above as well as data and a brief comment on mean arterial blood pressure and heart rate effects of compound **53** in spontaneously hypertensive rats are given. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Notes

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■ ABBREVIATIONS USED

AUC, area under the curve; BuLi, butyllithium; CC, column chromatography; DCM, dichloromethane; DEAD, diethyl azodicarbonylate; DMF, dimethylformamide; EA, ethyl acetate; EDC, 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide; HOBt, *N*-hydroxybenzotriazole; LC, lymphocyte count; LC-MS, (high pressure) liquid chromatography combined with mass spectrometry; PK, pharmacokinetics; PD, pharmacodynamics; RLM, rat liver microsome; SAR, structure–activity relationship; TBTU, *O*-(benzotriazol-1-yl)-*N,N,N',N'*-tetramethyluronium tetrafluoroborate; THF, tetrahydrofuran; S1P, sphingosine 1-phosphate

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