## Synthesis of Enantiopure 1-Benzyl-2,3-disubstituted Piperazines from Enantiopure *p*-Toluenesulfinimines

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**Abstract:** The treatment of enantiopure *N*-sulfinyl-*N'*-benzyldiaminoalcohols with diethyl oxalate and sodium methoxide followed by reduction with BH<sub>3</sub> affords enantiopure 1-benzyl-2,3disubstituted piperazines. A related sequence produces substituted monoketopiperazines in good yields.

**Key words:** piperazines, amino alcohols, sulfinimines, chiral auxiliaries, asymmetric synthesis

The piperazine ring is truly ubiquitous in molecules involved in the regulation of a wide variety of biological processes. Indeed, the piperazine scaffold is found in HIV-protease inhibitors such as indinavir,<sup>1</sup> potent antiproliferative agents such as ectenaiscidin 743,<sup>2</sup> compounds acting at different membrane receptors such as arylpiperazines, powerful 5-HT<sub>1A</sub> ligands,<sup>3</sup> as well as in some members of the dragmacines,<sup>4</sup> a family of marine natural products with a broad spectrum of biological activities. On the other hand, the structurally related mono- and diketopiperazines are gaining importance as farnesyl transferase inhibitors,<sup>5</sup> and as conformationally restricted peptidomimetics,6 among several biological activities.7 In addition, recent findings have revealed piperazines and diketopiperazines as efficient chiral ligands in asymmetric catalysis.8

Despite the increasing interest on these compounds, the existing routes to prepare chiral piperazines are scarce,<sup>9</sup> and not fully applicable to the straightforward synthesis of highly substituted derivatives. This is particularly true for non-symmetrical 2,3-disubstituted piperazines for which the simplest route entails reduction of 2,3-disubstituted diketopiperazines in turn prepared from the corresponding vicinal diamines. In fact, the synthesis of suitably substituted enantiopure vicinal diamines may be quite challenging,<sup>10</sup> and this severely limits the viability of this approach.

Within a program directed to the discovery of new therapeutically valuable piperazine derivatives,<sup>11</sup> and taking advantage of our previous experience in the asymmetric synthesis of non-symmetrical vicinal diamine derivatives,<sup>12</sup> we now disclose a straightforward entry to a new family of homochiral piperazines and ketopiperazines, with carbon and nitrogen atoms differently functionalized.

We have previously reported the asymmetric synthesis of enantiopure 1,3-imidazolidines C through the diastereoselective stepwise condensation between readily available *p*-tolylsulfinimines  $\mathbf{A}$ ,<sup>13</sup> and glycine iminoester enolates  $\mathbf{B}$ in the presence of boron trifluoride.<sup>12b</sup> Under these conditions, enantiopure *p*-tolylsulfinimines display a moderate to high facial selectivity rendering the corresponding Nsulfinylimidazolidines in good to excellent diastereomeric excess. Ensuing reductive cleavage of the aminal moiety expediently transforms our 1,3-imidazolidines C into differentially protected vicinal diamines **D** in good yields. (Scheme 1). The availability of these substrates, along with the aforementioned relevance of piperazines prompted us to examine the transformation of a series of our vicinal diamines 1 into the corresponding ketopiperazines and piperazines.



Scheme 1 Synthesis of *N*-sulfinyldiaminoalcohols.<sup>a</sup> *Conditions*: (a)  $BF_3$ ·OEt<sub>2</sub>, -78 °C, THF, (60–93%); (b) LiAlH<sub>4</sub>, Et<sub>2</sub>O, r.t., (68–85%).

Our initial efforts to obtain homochiral diketopiperazines were focussed on substrates 1a-e,<sup>14</sup> originally obtained from glycine-derived enolates (Scheme 2, R<sup>2</sup> = H). After testing different experimental conditions we found that the treatment of 1a (R<sup>1</sup> = Et) with diethyl oxalate in CH<sub>2</sub>Cl<sub>2</sub> afforded 2,3-diketopiperazine 2a in moderate yields (60%), along with a small amount of morpholinedione 3a derived from N,O ring closure. After chromatographic separation we observed that 3a slowly converted into 2a upon standing in a solution of MeOH, (5 days) and that a catalytic amount of NaOMe accelerated (1 h) this conversion. Substrates 1b and 1c paralleled this behavior, after removal of the diketopiperazine by filtration. Seek-

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ing to increase the efficiency of the process, we added a solution of NaOMe in MeOH to the reaction mixture, and this promoted in situ nucleophilic displacement at sulfur, rendering methyl *p*-toluenesulfinate along with the desired diketopiperazine in excellent yields. This procedure was also satisfactorily extended to aromatic substrates (**1d** and **1e**) to produce diketopiperazines **2d** and **2e**.<sup>15</sup>

After considerable experimentation,<sup>16</sup> diketopiperazines **2a–e** were efficiently transformed into the corresponding enantiopure 2,3-disubstituted piperazines **4a–e** in good yields (60–84%), upon treatment with borane dimethyl sulfide complex.<sup>17</sup> These hydroxymethylpiperazines are amenable to selective protection at the secondary nitrogen as benzyloxycarbonyl derivatives as illustrated by the preparation of hydroxymethylpiperazine **5**.



Scheme 2 Synthesis of enantiopure hydroxymethylpiperazines and ketopiperazines.<sup>a</sup> *Conditions*: (a)  $(CO_2Et)_2$ , NaOMe,  $CH_2Cl_2$ , MeOH, r.t.; (b) i)  $(CO_2Et)_2$ ,  $CH_2Cl_2$ , r.t.; ii) NaOMe, MeOH, r.t.; (c) NaOMe, MeOH, r.t.; (d) BH<sub>3</sub>·SMe<sub>2</sub>, THF, reflux; (e) CbzCl, NaOH,  $CH_2Cl_2$ , r.t.; (f) TFA, MeOH, r.t.

To broaden the scope of these procedures we considered vicinal diamine 1f, originally obtained from alanine (Scheme 2).<sup>14</sup> Unfortunately, the presence of a quaternary carbon in its skeleton ( $R^2 = CH_3$ ) prevented any cyclization from taking place. Thus, when compound 1f was treated with diethyl oxalate, cyclic derivatives 2 or 3 were not observed, even in the presence of NaOMe, instead labile open-chain acylated intermediates were detected. To enhance the nucleophilicity of the substrate we prepared desulfinylated diamine 6, and the treatment of 6 with diethyloxalate (Scheme 2), surprisingly, afforded hydroxymethylimidazoline 7 (68%). Subsequent reductive cleavage of the imidazoline ring produced exclusively Nbenzyl-N'-methyldiaminoalcohol 8 in fair yield (55%). The generality of these serendipitous findings remains to be tested.

To address the preparation of monoketopiperazines from our substrates,<sup>18</sup> the primary alcohol of **1a** and **1f** was protected uneventfully as a silvl ether in excellent yields, affording 9a and 9b, respectively (Scheme 3). Not unexpectedly, 9b did not undergo N-acylation with ClCH<sub>2</sub>COCl under mild conditions, presumably due to the quaternary carbon hindering the amino moiety and, at this stage, no further efforts were conducted on this substrate. In contrast, 9a underwent smooth acylation to give chloroacetamide 10 that cyclized to N-sulfinyl-N'-benzyl monoketopiperazine 11 in excellent yield, often accompanied by small amounts of a desulfinylated product, tentatively assigned as imine 12, presumably derived from 11 by enolate formation and elimination of the sulfur moiety. Treatment of **11** with NaH in refluxing THF afforded an excellent yield of imine **12** that was fully characterized.<sup>19</sup> We believe that **11** and **12** hold considerable potential for further selective manipulations at 'C-5' and 'C-6'.



Scheme 3 Synthesis of monoketopiperazines.<sup>a</sup> *Conditions*: (a) TBDMSCl, imidazole, cat. DMAP,  $CH_2Cl_2$ , r.t., 90 min; (b) ClCH<sub>2</sub>COCl, EtOAc, sat. NaHCO<sub>3</sub>, (1:1), 0 °C to r.t., 2 h; (c) Cs<sub>2</sub>CO<sub>3</sub>, DMF, 65 °C, 2 h; (d) NaH, THF, 0 °C to reflux, 1 h.

In summary, readily available enantiopure *N*-sulfinyl-*N'*benzyl diaminoalcohols have been expediently transformed into a variety of homochiral 2,3-disubstituted piperazines with adequate functionalization for subsequent transformations. In addition, a simple entry to more functionalized monoketopiperazines from the same precursors has been outlined. We are currently exploring the scope, limitations and further developments of the methodology.

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$$\begin{split} & R_{\rm f} = 0.10 \; (40\% \; \text{MeOH}{-}\text{Et}_2\text{O}). \; [\alpha]^{20}{}_{\rm D} - 14.1 \; (c\; 0.91, \; \text{CHCl}_3). \\ ^1\text{H}\; \text{NMR}\; (\text{CDCl}_3, \; 300 \; \text{MHz}): \; \delta = 0.92 \; (t, \; 3 \; \text{H}, \; J = 7.3 \; \text{Hz}), \\ & 1.48 \; (\text{sept}, \; 1 \; \text{H}, \; J = 7.6 \; \text{Hz}), \; 1.72 \; (m, \; 1 \; \text{H}), \; 2.19 \; (m, \; 1 \; \text{H}), \\ & 2.29 \; (dt, \; 1 \; \text{H}, \; J = 9.9, \; 3.0 \; \text{Hz}), \; 2.70 \; (m, \; 1 \; \text{H}), \; 2.78-2.96 \; (m, \\ & 3 \; \text{H}), \; 3.29 \; (d, \; 1 \; \text{H}, \; J = 13.3 \; \text{Hz}), \; 3.65 \; (dd, \; 1 \; \text{H}, \; J = 11.7, \; 1.6 \\ & \text{Hz}), \; 4.02 \; (d, \; 1 \; \text{H}, \; J = 13.3 \; \text{Hz}), \; 4.05 \; (dd, \; 1 \; \text{H}, \; J = 11.8, \; 3.2 \\ & \text{Hz}), \; 7.26 \; (m, \; 5 \; \text{H}). \; ^{13}\text{C}\; \text{NMR}\; (\text{CDCl}_3, \; 50 \; \text{MHz}): \; \delta = 10.3, \\ & 25.1, \; 43.1, \; 50.5, \; 58.3, \; 59.1, \; 63.3, \; 127.0, \; 128.3 \; (2 \; \text{C}), \; 128.7 \\ & (2 \; \text{C}), \; 138.4. \; \text{IR}(\text{film}): \; 3306, \; 3057, \; 3021, \; 2958, \; 1490, \; 1452, \\ & 1027, \; 739, \; 699 \; \text{cm}^{-1}. \; \text{MS}\; (\text{ES}): \; 235 \; [\text{M} + 1]^+ \; (100\%). \end{split}$$

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2 H, J = 8.1 Hz). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 50 MHz):  $\delta = -5.3, -5.2,$ 10.3, 18.1, 21.3, 22.5, 25.8 (3 C), 39.7, 48.6, 57.8, 59.1, 62.4, 125.6 (2 C), 128.0, 128.8 (2 C), 128.9 (2 C), 129.8 (2 C), 136.6, 139.3, 141.8, 165.5. IR(film): 2963, 2855, 1732, 1646, 1435, 1260, 1090, 1018, 799 cm<sup>-1</sup>. MS (ES): 501 [M (100%), 502 [M + 2]<sup>+</sup>, 503 [M + 3]<sup>+</sup>, 523 [M + Na]<sup>+</sup>. Synthesis of (+)-(5R,6S)-1-benzyl-6-(t-butyldimethylsilyloxymethyl)-5-ethyl-5,6-dihydro-1H-pyrazin-2-one, 12. From a cold (0 °C) suspension of 4 equiv of NaH in anhyd THF (5 mL/mmol of NaH), with a solution of 11 (72 mg, 0.144 mmol) in THF (5 mL/mmol) after stirring at r.t. (1 h) and at reflux (1 h), a crude product was obtained after standard extractive isolation. Purification by chromatography on silica gel (30-80% Et<sub>2</sub>O-hexane) afforded 43 mg (0.119 mmol, 83%) of 12 as a colorless oil. Data of 12:  $R_f = 0.38 \ (80\% \ Et_2O-hexane). \ [\alpha]_D^{20} + 188.3 \ (c \ 1.23, c)^{20}$ CHCl<sub>3</sub>). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta = 0.05$  (s, 6 H), 0.54 (t, 3 H, J = 7.4 Hz), 0.86 (s, 9 H), 0.86–0.98 (m, 1 H), 1.40– 1.49 (m, 1 H), 3.27 (ap t, 1 H, J = 6.5 Hz), 3.55 (dd, 1 H, J = 10.2, 6.9 Hz), 3.63 (dd, 1 H, J = 10.2, 5.8 Hz), 3.78 (dd, 1 H, J = 7.7, 5.5 Hz), 3.92 (d, 1 H, J = 14.3 Hz), 5.34 (d, 1 H, J = 14.3 Hz), 7.26–7.34 (m, 5 H), 7.70 (d, 1 H, J = 1.5 Hz). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 50 MHz):  $\delta = -5.6, -5.5, 10.2, 18.1,$ 25.8 (4 C), 48.4, 55.2, 59.4, 63.0, 128.1, 128.8 (2 C), 129.1 (2 C), 136.1, 155.1 (2 C). IR(film): 3400, 3030, 2929, 2857, 1674, 1632, 1454, 1361, 1258, 1101, 908, 838, 804, 779, 704 cm<sup>-1</sup>. MS (ES): 361 [M + 1]<sup>+</sup> (100%), 743 [2 M + Na]<sup>+</sup>.