ARTICLE

www.rsc.org/obc

# Bifurcated, modular syntheses of chiral annulet triazacyclononanes

Gilles Argouarch,† Graham Stones, Colin L. Gibson,\* Alan R. Kennedy and David C. Sherrington

Department of Pure & Applied Chemistry, University of Strathclyde, 295 Cathedral Street, Glasgow, UK G1 1XL. E-mail: c.l.gibson@strath.ac.uk

Received 1st September 2003, Accepted 13th October 2003 First published as an Advance Article on the web 6th November 2003

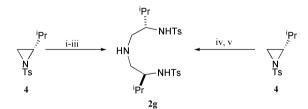
Three chiral 2,6-disubstituted tri-*N*-methyl azamacrocycles have been prepared by modular methods. These macrocycles were accessed from three chiral 1,4,7-triazaheptanes intermediates that were prepared by two independent routes. The first of these routes involved the benzylamine opening of chiral tosyl aziridines followed by debenzylation but was problematic on solubility grounds. A second, more effective, route was developed which avoided debenzylation by using ammonia in the nucleophilic opening of chiral tosyl aziridines.

## Introduction

The continued interest in the synthesis of 1,4,7-triazacyclononane derivatives is maintained by the use of transition metal complexes of these azamacrocycles to catalyse a variety of processes. Transition metal complexes of 1,4,7-triazacyclononane derivatives have been investigated as biomimetics of manganese catalase, 1,2 Photosystem II, 1,3 and hemocyanin. These metal complexes have also been used to catalyse organic transformations such as the hydrolytic cleavage of RNA, DNA and peptides as well as the oxidation of sulfides, alcohols, alkanes 10 and alkenes. In particular, the epoxidation of alkenes has generated considerable interest and stereoselective 2 as well as enantioselective processes have been described using chiral analogues of 1,4,7-triazacyclononane by Beller et al., Bolm et al. 14,15 and ourselves.

Despite the great interest in triazacyclononanes and their synthesis, the preparation of chiral variants where the stereochemistry is associated with the carbon backbone are less common. Thus, chiral triazacyclononanes with one,17 two (C-2,3) 18 and three 15 stereocentres on the macrocyclic ring, have been prepared. Recently, ourselves (1a-c), Kim et al. (1e) 19 and Watkinson and co-workers (1b,d) 20 have reported the preparation of 2,6-disubstituted triazacyclononanes 1a-e. These three independent reports utilised the Richman-Atkins cyclisation of chiral 4-substituted 1,7-ditosyl-1,4,7-triazaheptanes 2a-e with ethylene glycol ditosylate 3 in the key macrocyclisation step (Scheme 1). The key intermediate 4substituted 1,7-ditosyl-1,4,7-triazaheptanes 2a-e were prepared from 1,7-ditosyl-1,4,7-triazaheptanes **2f-h** which, in turn, were accessed by nitrogen based nucleophilic opening of the chiral tosyl aziridines e.g. 4 (Scheme 2). While Kim et al. used azide opening of chiral tosyl aziridine 4, we and Watkinson and co-workers opted to use benzylamine. In our hands, the subsequent debenzylation of 2i-k and analogues was problematic because of solubility problems. Herein, we report full experimental details of our syntheses of azamacrocycles 1a-c and their detosylation and N-methylation from our previous communication.<sup>16</sup> In addition, an alternative route of choice, avoiding the debenzylation problems involving the opening of chiral tosyl aziridines with ammonia is described. In some cases, this alternative route leads to improved overall yields in comparison with the use of benzylamine aziridine opening and subsequent debenzylation.

Scheme 1 Reagents and conditions: CsCO<sub>3</sub>, DMF, 80 °C or NaH, DMF.

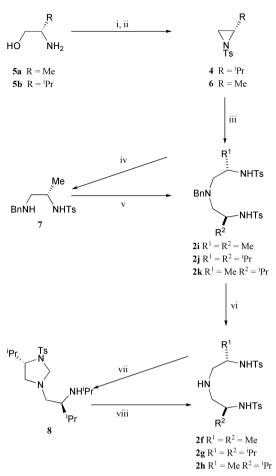


Scheme 2 Reagents and conditions: i, NaN<sub>3</sub>; ii, H<sub>2</sub>, Pd–C, MeOH; iii, 4; iv, 0.5 equiv. BnNH<sub>2</sub>, MeOH; v, H<sub>2</sub>, Pd(OH)<sub>2</sub>, MeOH.

# **Results and discussion**

The synthesis of chiral azamacrocycles **1a**–**c** began with the conversion of alaninol **5a** and valinol **5b** into the known tosyl aziridines **6**<sup>21</sup> and **4**<sup>22</sup> via tosylation and cyclisation through a modification of the method of Moberg and co-workers (Scheme 3).<sup>23</sup> It had been shown that sulfonamide aziridines could be opened with primary amines to give bis(adducts) under the appropriate conditions.<sup>22a,24</sup> This nucleophilic attack generally occurs at the least substituted carbon. Thus, reaction of tosyl aziridines **6** and **4** with 0.5 of an equivalent of benzylamine in methanol afforded the bis(adducts) **2i** and **2j** in 86 and 67%, respectively.<sup>25,26</sup> Alternatively, treatment of tosyl aziridine **6** with excess benzylamine in methanol at room temperature afforded the mono adduct **7** in 85% yield. Concomitantly with our communication, <sup>16</sup> Watkinson and co-workers <sup>25</sup> reported a similar approach to the mono adduct **7** (56%) and analogues

<sup>†</sup> *Present address*: UMR CNRS 6509, Organometalliques et Catalyse, University de Rennes, Campus de Beaulieu, 35042 Rennes, France.



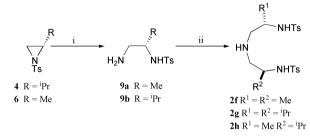
Scheme 3 Reagents and conditions: i, TsCl, pyr; ii, NaH, THF; iii, 0.5 equiv. BnNH<sub>2</sub>, MeOH; iv, excess BnNH<sub>2</sub>, MeOH; v, 6 MeOH; vi, H<sub>2</sub>, Pd(OH)<sub>2</sub>, MeOH or H<sub>2</sub>, Pd–C, AcOH; vii, H<sub>2</sub>, Pd(OH)<sub>2</sub>, MeOH–CH<sub>2</sub>Cl<sub>3</sub>; viii, 3 M HCl, MeOH.

using acetonitrile at reflux. The subsequent reaction of mono adducts such as 7 with alternative tosyl aziridines was an attractive prospect because it might allow a divergent approach to  $C_1$  symmetric chiral precursors. Accordingly, reaction of mono adduct 7 with tosyl aziridine 4 in methanol afforded the non-symmetrical bis(adduct) 2k in 45% yield.

Ourselves and Watkinson and co-workers have shown that 2i<sup>16</sup> and 2j,<sup>20</sup> respectively, are not appropriate substrates for Richman-Atkins cyclisation because of nucleophilic interference by the benzyl protected amine. Accordingly, hydrogenolytic removal of the benzyl group from 2i and 2j using palladium hydroxide in methanol afforded the free amines 2f and 2g in 86 (65.2% overall from 5a) and 71% yield (42.9% overall from 5b), respectively. However, solubility problems were experienced in these debenzylation procedures. Indeed, in larger scale (ca. 1 g) debenzylation of 2g, solubility problems were acute and required the addition of dichloromethane as co-solvent which afforded the novel imidazolidine 8 in 78% vield. Subsequent, acid hydrolysis of imidazolidine 8 afforded an alternative synthesis of amine 2g (43.3% overall yield from **5b**). These solubility issues were severe for the debenzylation of 2k and this deprotection was unsuccessful using palladium hydroxide in methanol. This process was achieved using palladium on charcoal with acetic acid as the solvent which afforded the non symmetric amine 2h in 40% yield (13.5% overall from 5a).

The above problems with solubility in the debenzylations of benzylamines 2i–k forced us to re-evaluate routes to amines 2f–h. The use of ammonia to open tosyl aziridines can yield mixtures of mono, bis and tris(adducts).<sup>23,26</sup> However, Taylor and co-workers have developed a two step process for the selective preparation of bis(adducts) *via* the mono ammonia

opened tosyl aziridines.<sup>22a,27</sup> Accordingly, treatment of tosyl aziridine 6 with ammonia saturated methanol at 0 °C afforded the mono adduct 9a in 95% yield (Scheme 4). Subsequent reaction of 9a with tosyl aziridine 6 in toluene at reflux provided an alternative route to **2f** in 61% yield (51.1% overall from **5a**). Analogously, ammonia opening of tosyl aziridine 4 afforded the mono adduct 9b in 89% which then reacted with tosyl aziridine 4 to provide 2g in 56% (44.9% overall from 5b). 22a,27 The non-symmetric amine 2h was accessed by reaction of mono adduct 9a with tosyl aziridine 4 in 88% yield (73.7% overall from 5a). These alternative routes to the amines 2f-h via the mono adducts 9 (Scheme 4) were the methods of choice as there were no solubility issues making the syntheses much easier. Moreover, the route using ammonia opening of aziridine was significantly more efficient than the route using benzylamine (Scheme 3) in the case of **2h** (70.3 vs. 13.5 % overall from **5a**).

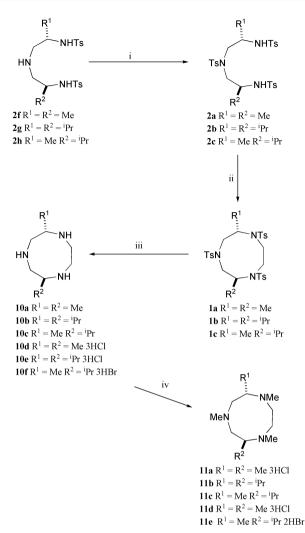


Scheme 4 Reagents and conditions: i, NH<sub>3</sub>, MeOH, 0 °C; ii, 4 or 6,  $C_6H_4CH_3$ ,  $\Delta$ .

With effective routes to the secondary amine tosamides **2f**–**h** in hand we turned our attention to facilitating macrocyclisation which required suppressing the nucleophilicity of the central amine function. In Richman–Atkins macrocyclisations, <sup>28,29</sup> the amino functionalities are commonly protected as their sulfonamide derivatives. Accordingly, treatment of amines **2f**–**h** with tosyl chloride in pyridine furnished the fully protected tristosamides **1a**–**c** (Scheme 5).

Richman–Atkins cyclisations to form aza crowns generally involve the reaction of the dianion of a bissulfonamide with a bistosylate ester in anhydrous DMF. The sodium dianion of the bissulfonamide is commonly prepared prior to cyclisation with sodium hydride in DMF or sodium ethoxide in ethanol. 28,29 Alternatively, the caesium salt can be generated *in situ* using caesium carbonate in DMF. We have generally found that the use of sodium hydride in DMF is a more effective procedure to carry out the macrocyclisations to afford C-substituted 1,4,7-triazacyclononane tristosamides. 30 Therefore, macrocyclisation of the tristosamides 2a–c with sodium hydride in DMF with ethylene glycol ditosylate 3 smoothly afforded the triazacyclononane tristosamides 1a–c in over 70% yield.

The subsequent steps from tristosamides 1a-c required the removal of the N-tosyl protecting groups and subsequent N-alkylation. N-Tosyl groups have been removed from C-alkyl substituted triazacyclononanes by three popular deprotection procedures. Thus, Mason et al. and Weatherburn et al. have used hydrobromic acid in glacial acetic acid at reflux to deprotect a C-methyl substituted triazacyclononane tristosamide. 17a,d However, we have found that this method gives incomplete deprotection of C-alkyl triazacyclononane tristosamides, forming instead a mixture of mono and bistosamide macrocycles.30 Alternatively, Parker and co-workers have removed the tosyl protecting groups in a 2-(4-benzamidobutyl)-1,4,7-triazacyclononane tristosamide by lithium in ammonia or by heating in concentrated sulfuric acid. 17e We have found that the lithium in ammonia reductive deprotection to be very efficient in the case of C-alkyl macrocycles. 30 Accordingly, lithium in ammonia detosylation of tosamides 1a-c provided the free amines 10a-c. In the case of 10b and 10c the process was reasonably efficient with the formation of products in 66 and 87% yield, respect-



Scheme 5 Reagents and conditions: i, TsCl, Pyr; ii, 3, NaH, DMF, 80 °C; iii, Li, NH<sub>3</sub>, EtOH; iv, CH<sub>2</sub>O, HCO<sub>2</sub>H, Δ.

ively. The deprotection of the dimethyl substituted macrocycle 1a was achieved in a disappointing yield of 13% and was a function of the water solubility of triamine 10a. The C<sub>2</sub> symmetric macrocycles 10a and 10b were additionally characterised as their respective hydrochloride salts 10d and 10e while the non-symmetric macrocycle 1c was also characterised as the hydrobromide salt 10f. The use of sulfuric acid deprotection was investigated in the case of deprotection of the non-symmetric macrocycle 1c which afforded a lower yield of the free amine 10c (63%).

The final step in our syntheses required the N-methylation of macrocycles 10a-c. After a brief model study of the methylation of 1,4,7-triazacyclononane using butyllithium and methyl iodide<sup>31</sup> (45%) or Eschweiler-Clarke (82%)<sup>32</sup> it was decided to use the later N-methylation procedure on the grounds of efficiency and purity of the product. Furthermore, the use of Eschweiler-Clarke N-methylation proved to be particularly advantageous in the case of dimethyl macrocycle 10a. In this case simply taking the crude aqueous extract after lithium in ammonia detosylation of 1a and subjecting it to these N-methylation conditions afforded the dimethyl macrocycle 11a in 44% over two steps. As water solubility was not an issue with 10b and 10c, Eschweiler-Clarke N-methylation of the isolated macrocycles gave the requisite novel macrocycles 11b and 11c in 66 and 90% yields, respectively. The N-methyl macrocycles 11a and 11c were additionally characterised as their salts 11d and 11e, respectively. The crystalline nature of hydrobromide salt 11e allowed an X-ray analysis to be carried out and allowed both the stereochemistry and the absolute structure to be confirmed unequivocally by single crystal diffraction (Fig. 1).

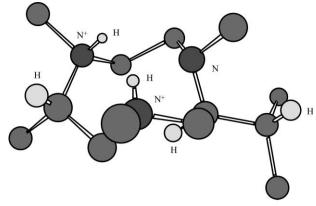


Fig. 1 Chem3D representation of the molecular structure of 11e with only the key hydrogen atoms included for clarity.

Unexpectedly, this material was the dihydrobromide salt 11e rather than the anticipated trihydrobromide salt. The dihydrobromide salt 11e results from protonation of the less sterically encumbered N1 and N4 while N7, adjacent to the isopropyl group remains as the free base.

#### **Conclusions**

Two independent and modular routes have been developed for the syntheses of chiral 1,4,7-triazaheptanes as intermediates in the preparation of chiral 1,4,7-triazamacrocycles. The first modular route to 1,4,7-triazaheptanes involved the benzylamine nucleophilic opening of chiral tosyl aziridines followed by hydrogenolytic debenzylation. Solubility problems in the debenzylation step were overcome by developing an alternate route that avoided the need for this deprotection step. These more amenable routes involved the nucleophilic opening of chiral tosyl aziridines with ammonia, which led, in some cases, to increased efficiencies. The highly modular nature of these routes readily allowed the preparation of C2 symmetric and dissymmetric 1,4,7-triazaheptanes. These key intermediates were then used in the preparation of three chiral 2,6-disubstituted 1,4,7-trimethyl triazamacrocycles in 8-9 steps overall and 12.77-29.9% overall yields.

# **Experimental**

### Instrumentation

Melting points were determined on a Reichert 7905 hot stage and are uncorrected. Specific rotations were measured at 20 °C in a 1 cm<sup>3</sup> cell with a pathlength of 10 cm using a Perkin-Elmer 341 polarimeter . The  $[a]_D$  values are given in  $10^{-1}$  deg cm<sup>2</sup> g<sup>-1</sup> and the concentrations are given in g/100 cm<sup>3</sup>. <sup>1</sup>H-nmr spectra were recorded on Bruker WM-250, Jeol 270, or Bruker AMX-400 spectrometers in the indicated solvents operating at 250, 270 or 400 MHz, respectively. <sup>13</sup>C-NMR spectra were obtained on the same instruments operating at 62.89, 67.80, and 100 MHz, respectively. The following abbreviations were used: s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet; dd, doublet of doublets; dq, doublet of quartets; sep, septet. Coupling constants were recorded in Hz. Infra-red (IR) spectra were recorded on a Nicolet Impact 400D FTIR spectrometer either as liquid films, KBr discs or as a 1–2% solution (CCl<sub>4</sub>). Mass spectra were recorded on a Jeol JMS AX505 spectrometer at Strathclyde or at the EPSRC National Mass Spectrometry service, Swansea. Microanalyses were performed by the microanalytical service at Strathclyde.

# General methods

Anhydrous reactions were carried out under an atmosphere of nitrogen in oven dried glassware (140 °C). Anhydrous solvents

were obtained using standard procedures: ethanol (Mg(OEt)<sub>2</sub>), pyridine (predried over KOH, distilled from CaH<sub>2</sub>), THF (K metal), toluene (Na metal) and triethylamine (CaH<sub>2</sub>). All other reagents were used as supplied. Flash column chromatography was performed according to the procedure of Still *et al.*<sup>33</sup> using silica gel (230–400 mesh).

#### **Experimental procedures**

(2S)-2-Methyl-1-[(4-methylphenyl)sulfonyl]aziridine 6. Using (S)-alaninol 5a and the method of Moberg and co-workers,<sup>2</sup> the title compound was obtained as an off white solid (4.4 g, 20.8 mmol, 88%); mp 60-62 °C (lit.21 mp 58-59 °C); Found C, 56.87; H, 6.1; N, 6.61; S, 15.48%; MH<sup>+</sup> m/z 212.0736; Calculated for C<sub>10</sub>H<sub>14</sub>NO<sub>2</sub>: C, 56.85; H, 6.2; N, 6.63; S, 15.17%; m/z 212.0659;  $[a]_D$  +30.3 (c = 1.02, CHCl<sub>3</sub>) (lit.<sup>21</sup> +29.6 (c = 1.02, CHCl<sub>3</sub>));  $v_{\text{max}}$  (KBr, cm<sup>-1</sup>) 3049 (w, C<sub>6</sub>H<sub>4</sub>), 2962 (s, CH), 2928 (m, CH), 2873 (w, CH), 1319 (s, SO<sub>2</sub>NH), 1159 (s, SO<sub>2</sub>NH);  $\delta_{\rm H}$  (400 MHz, CDCl<sub>3</sub>) 1.26 (d, J 5.6, 3H, -CH<sub>3</sub>), 2.02 (d, J 5.8, 1H,  $TsNCH_2$ ), 2.45 (s, 3H,  $-C_6H_4-CH_3$ ), 2.62 (d, J 7.0, 1H,  $TsNCH_2$ ), 2.82 (m, 1H, TsNCH), 7.34 (d, J 8.0, 2H,  $-C_6H_4$ -CH<sub>3</sub>), 7.82 (d, J 8.0, 2H,  $-C_6H_4$ –CH<sub>3</sub>);  $\delta_C$  (100 MHz, CDCl<sub>3</sub>) 17.0 ( $-CH_3$ ), 21.8 ( $-C_6H_4-CH_3$ ), 35.0 ( $TsNCH_2$ ), 36.1 (TsNCH), 128.0 (Ar-CH), 129.9 (Ar-CH), 136.6 (Ar-C), 144.6 (Ar-C).

(2S)-2-Isopropyl-1-[(4-methylphenyl)sulfonyl]aziridine 4. Analogously, (S)-valinol **5b** afforded the title compound as an off white solid (3.3 g, 13.6 mmol, 90%); mp 85-87 °C (lit.27 75–77 °C); Found C, 60.5; H, 7.2; N, 5.8; S, 13.4%; MH+ m/z 240.1051; Calculated for  $C_{12}H_{27}O_2SN$ : C, 60.2; H, 7.2; N, 5.9; S, 13.4%; m/z 240.1058;  $[a]_D$  +15.9 (c = 0.95, CHCl<sub>3</sub>) (lit.<sup>27</sup> +12.2  $(c = 1.2, \text{CHCl}_3)$ ;  $v_{\text{max}}$  (KBr, cm<sup>-1</sup>) 3049 (w, C<sub>6</sub>H<sub>4</sub>), 2962 (s, CH), 2928 (m, CH), 2873 (w, CH), 1319 (s, SO<sub>2</sub>NH), 1159 (s,  $SO_2NH$ );  $\delta_H$  (400 MHz, CDCl<sub>3</sub>) 0.80 (d, J 6.7, 3H, -CH<sub>3</sub>), 0.90  $(d, J 6.7, 3H, -CH_3), 1.42 (m, 1H, CH(CH_3)_2), 2.10 (d, J 4.6,$ 1H,  $TsNCH_2$ ), 2.45 (s, 3H,  $-C_6H_4-CH_3$ ), 2.52 (m, 1H, TsNCH), 2.61 (d, J 8.0, 1H, TsNC $H_2$ ), 7.34 (d, J 8.0, 2H,  $-C_6H_4$ –CH<sub>3</sub>); 7.83 (d, J 8.0, 2H,  $-C_6H_4$ –CH<sub>3</sub>);  $\delta_C$  (100 MHz, CDCl<sub>3</sub>) 19.2  $(-CH_3)$ , 19.7  $(-CH_3)$ , 21.8  $(-C_6H_4-CH_3)$ , 30.3  $(CH(CH_3)_2)$ , 32.9 (TsNCH<sub>2</sub>), 46.4 (TsNCH<sub>2</sub>); 128.3 (Ar-CH), 129.7 (Ar-CH), 135.4 (Ar-C), 144.6 (Ar-C).

 $N-\{(1S)-2-[Benzyl((2S)-\{(4-methylphenyl)sulfonyl]amino}\}$ propyl)amino]-1-methylethyl}-4-methylbenzenesulfonamide 2i. To a solution of (2S)-2-methyl-1-[(4-methylphenyl)sulfonyl]aziridine 6 (1.8 g, 8.53 mmol) in methanol (20 cm<sup>3</sup>) was added benzylamine (456 mg, 4.26 mmol). The solution was stirred for 4 days at room temperature, upon completion of this period the solvent was evaporated to give a crude brown solid. Purification by column chromatography on silica (hexane: EtOAc from 2:1 to 1:1) afforded a colourless solid (1.94 g, 3.67 mmol, 86%); mp 129 °C (lit. for hemi hydrate 25 91-92 °C). Found C, 60.84; H, 6.46; N, 7.74%; MH<sup>+</sup> m/z 530.2153; Calculated for  $C_{27}H_{35}O_4S_2N_3$ : C, 61.22; H, 6.66; N, 7.94%; m/z 530.2147; [a]<sub>D</sub> -13.8 (c = 1.00, CHCl<sub>3</sub>) (lit. for hemi hydrate<sup>25</sup> -43.2 (c = 0.5, CHCl<sub>3</sub>));  $v_{\text{max}}$  (KBr, cm<sup>-1</sup>) 3450–3100 (bm, NHTs), 3064 (w, ArH), 3030 (w, ArH), 2970, 2930, 2820 (all w, CH), 1332 (s,  $NSO_2$ ), 1162 (s,  $NSO_2$ ), 1094 (s), 815 (m, ArH);  $\delta_H$  (250 MHz,  $CDCl_3$ ) 0.91 (d, J 6.3, 6H,  $-CH_3$ ), 2.15–2.41 (m, 4H,  $NCH_2$ ), 2.33 (s, 6H,  $-C_6H_4-CH_3$ ), 3.08 (d, J 13.5, 1H,  $NCH_aH_b$ ), 3.36 (m, 2H, NCH), 3.47 (d, J 13.5, 1H, NCH<sub>a</sub>H<sub>b</sub>), 5.24 (bs, 2H,  $SO_2NH$ ), 7.02–7.05 (m, 2H, ArH), 7.19–7.31 (m, 7H, ArH), 7.76 (d, J 8.1, 4H,  $-C_6H_4$ –CH<sub>3</sub>);  $\delta_C$  (100 MHz, CDCl<sub>3</sub>) 19.8  $(-CH_3)$ , 21.6  $(-C_6H_4CH_3)$ , 47.1  $(NCH_2)$ , 57.9  $(NCH_2)$ , 59.6 (TsNCH), 127.2, 127.4, 128.5, 129.3, 129.7 (all Ar-CH), 137.5, 138.1, 143.3 (all Ar–*C*).

((1S)-1-{[Benzyl((2S)-3-methyl-2-{[(4-methylphenyl)sulf-onyl]amino}butyl)amino]methyl}-2-methylpropyl)-4-methylbenzenesulfonamide 2j. The title compound was obtained in

an analogous fashion to 2i using (2S)-2-isopropyl-1-[(4-methylphenyl)sulfonyl]aziridine 4 (2.2 g, 14.2 mmol) and benzylamine (492 mg, 4.60 mmol). Purification by column chromatography on silica (hexane: ethyl acetate 2:1) followed by recrystallisation (hexane: dichloromethane 2:1) afforded a colourless solid (1.8 g, 3.08 mmol, 67%); mp 123-125 °C (lit.25 135-136 °C); Found MH<sup>+</sup> m/z 586.2773; Calculated for  $C_{31}H_{43}O_4S_2N_3 m/z$  586.2773;  $[a]_D - 12.2$  (c = 1.00, CHCl<sub>3</sub>) (lit.<sup>25</sup> -37.6 (c = 0.5, CHCl<sub>3</sub>));  $v_{\text{max}}$  (KBr, cm<sup>-1</sup>) 3266 (s, NHTs), 3034 (w, C<sub>6</sub>H<sub>4</sub>), 2966 (s, CH), 2927 (m, CH), 2876 (w, CH), 1325 (s, SO<sub>2</sub>NH), 1160 (s, SO<sub>2</sub>NH);  $\delta_{\rm H}$  (400 MHz, CDCl<sub>3</sub>) 0.65 (d, J 8.2, 6H,  $-CH_3$ ), 0.67 (d, J 7.0, 6H,  $-CH_3$ ), 1.89 (m, 2H,  $CH(CH_3)_2$ , 2.30 (d, J 5.7, 2H,  $CH_2N$ ), 2.34 (d, J 5.7, 2H,  $CH_2N$ ), 2.41 (s, 6H,  $-C_6H_4-CH_3$ ) 2.48 (m, 2H,  $NCH_2$ ), 3.47 (m, 2H, TsNHCH), 7.23 (m, 5H,  $-C_6H_5$ ) 7.28 (m, 4H,  $-C_6H_4$ -CH<sub>3</sub>), 7.81 (d, J 8.3, 4H,  $-C_6H_4$ -CH<sub>3</sub>);  $\delta_C$  (100 MHz, CDCl<sub>3</sub>) 17.5  $(-CH_3)$ , 18.0  $(-CH_3)$ , 21.7  $(2 \times -C_6H_4-CH_3)$ , 29.9  $(CH(CH_3)_2)$ , 54.3 (2 × NCH<sub>2</sub>), 56.2 (NCH<sub>2</sub>), 58.6 (2 × TsNCH), 127.1  $(2 \times Ar-CH)$ , 127.4  $(2 \times Ar-CH)$ , 128.5  $(2 \times Ar-CH)$ , 129.6  $(2 \times Ar-CH)$ , 129.7  $(2 \times Ar-CH)$ , 137.9  $(2 \times Ar-C)$ , 139.0  $(2 \times Ar - C)$ , 143.1 (Ar - C).

N-[(1S)-2-(Benzylamino)-1-methylethyl]-4-methylbenzenesulfonamide 7. To a solution of (2S)-2-methyl-1-[(4-methylphenyl)sulfonyl]aziridine 6 (422 mg, 2 mmol) in methanol (10 cm<sup>3</sup>) was added benzylamine (1 g, 9.34 mmol). The solution was stirred for 2 days at room temperature, upon completion of this period the solvent was evaporated. Purification by column chromatography on silica (hexane : EtOAc from 1 : 4) afforded a colourless oil (542 mg, 1.70 mmol, 85%); Found C, 63.72; H, 6.91; N, 9.2; S, 10.18%; MH+ m/z 319.1482; Calculated for  $C_{17}H_{22}O_2SN_2C$ , 64.12; H, 6.69; N, 8.8; S, 10.07%; m/z 319.1480;  $[a]_D$  +2.6 (c = 0.70, CHCl<sub>3</sub>) (lit.<sup>25</sup> -4.8 (c = 0.5, CHCl<sub>3</sub>));  $v_{\text{max}}$  $(CCl_4, cm^{-1})$  3710–3460 (bw, NH), 3460–3130 (s, SO<sub>2</sub>NH), 3080, 3060, 3030 (all s, ArH), 2975, 2930, 2870, 2850 (all s, CH), 1600, 1490, 1450 (all s, Ar), 1320 (s, SO<sub>2</sub>N), 1150 (s, SO<sub>2</sub>N), 1095 (s), 820 (s), 740 (s), 700 (s);  $\delta_{\rm H}$  (250 MHz, CDCl<sub>3</sub>) 1.06 (d, J 6.5, 3H,  $-CH_3$ ), 2.36 (s, 3H,  $-C_6H_4-CH_3$ ), 2.38–2.57 (m, 2H, NCH<sub>2</sub>), 3.23 (m, 1H, TsNHCH), 3.53 (s, 2H, NCH<sub>2</sub>), 7.15-7.32 (m, 7H, ArH), 7.70 (d, J 8, 2H,  $-C_6H_4-CH_3$ );  $\delta_C$  (100 MHz,  $CDCl_3$ ) 20.0 (-CH<sub>3</sub>), 21.9 (-C<sub>6</sub>H<sub>4</sub>-CH<sub>3</sub>), 49.2 (NCH<sub>2</sub>), 53.6  $(NCH_2)$ , 54.1 (2 × TsNCH), 127.6 (Ar–CH), 127.7 (Ar–CH), 128.4 (Ar-CH), 128.9 (Ar-CH), 130.1 (Ar-CH), 138.0 (Ar-C), 140.2 (Ar-C), 143.7 (Ar-C).

N-{(1S)-2-[Benzyl((2S)-3-methyl-2-{[(4-methylphenyl)sulfonyl]amino}butyl)amino]-1-methylethyl}-4-methylbenzenesulfonamide 2k. To a solution of (2S)-2-isopropyl-1-[(4methylphenyl)sulfonyl]aziridine 4 (433 mg, 1.81 mmol) in methanol (10 cm³) was added N-[(1S)-2-(benzylamino)-1methylethyl]-4-methylbenzenesulfonamide 7 (576 mg, 1.81 mmol). The solution was stirred for 4 days at RT and the solvent was evaporated. The crude product was purified by column chromatography on silica (hexane: EtOAc 3:1) to give the unreacted aziridine 4 in a first fraction (206 mg, 0.86 mmol, 48%) and in a second fraction the title compound as a colourless solid (452 mg, 0.81 mmol, 45%); mp 143-145 °C; Found C, 62.54; H, 7.16; N, 7.39%; MH<sup>+</sup> m/z 558.2444; C<sub>29</sub>H<sub>39</sub>N<sub>3</sub>O<sub>4</sub>S<sub>2</sub> requires C, 62.45; H, 7.05; N, 7.54%; m/z 558.2460;  $[a]_D = -12.8$  $(c = 0.50, \text{CHCl}_3); \nu_{\text{max}} (\text{KBr}, \text{cm}^{-1}) 3270 \text{ (s, SO}_2\text{NH)}, 3060 \text{ and}$ 3030 (w, ArH), 2960 and 2930 (m, CH), 1600 (w, Ar), 1445 (s), 1320 (s, SO<sub>2</sub>N), 1160 (s, SO<sub>2</sub>N), 810 (m, ArH), 750 (m, ArH), 700 (m, ArH);  $\delta_{\rm H}$  (400 MHz, CDCl<sub>3</sub>) 0.64 (d, J 6.8, 3H, -CH<sub>3</sub>), 0.67 (d, J 6.9, 3H,  $-CH_3$ ), 1.04 (d, J 6.3, 3H,  $-CH_3$ ), 1.81 (m, 1H,  $CH(CH_3)_2$ ), 2.27–2.32 (m, 2H,  $CH_2N$ ), 2.39 (s, 3H,  $-C_6H_4$ –  $CH_3$ ), 2.42 (s, 3H,  $-C_6H_4-CH_3$ ), 2.43–2.49 (m, 2H,  $CH_2N$ ), 3.19 (d, J 13.3, 1H,  $CH_aH_bPh$ ), 3.43 (m, 1H, TsNHCH), 3.53 (m, 1H, TsNHCH), 3.68 (d, J 13.3, 1H, CH<sub>a</sub>H<sub>b</sub>Ph), 4.85 (bs, 1H, TsNHCH), 5.48 (bs, 1H, TsNHCH), 7.16–7.30 (m, 9H, ArH), 7.80–7.85 (m, 4H, ArH);  $\delta_{\rm C}$  (100 MHz, CDCl<sub>3</sub>) 17.5 (–CH<sub>3</sub>),

[(1S)-2-Amino-1-methylethyl]-4-methylbenzenesulfonamide

9a. To a saturated solution of ammonia in methanol (300 cm<sup>3</sup>) at 0 °C was added, dropwise, with stirring a solution of (2S)-2methyl-1-[(4-methylphenyl)sulfonyl]aziridine 6 (4.8 g, 22.77 mmol) in methanol (100 cm³) over a period of 2 h. Ammonia was bubbled continuously through the reaction mixture during addition and for 30 min after it was completed. The solution flask was stoppered and allowed to stand for 24 h. The volatiles were evaporated under reduced pressure to afford the title compound as a colourless solid (4.95 g, 21.69 mmol, 95%) (toluene); mp 128-130 °C; Found C, 52.8; H, 7.38; N, 12.17; S, 14.36%;  $MH^{+}$  m/z 229.0994;  $C_{10}H_{16}N_{2}O_{2}S$  requires C, 52.61; H, 7.06; N, 12.27; S, 14.04%; m/z 229.1011;  $[a]_D$  14.9 (c=1, CHCl<sub>3</sub>);  $\nu_{max}$ (KBr, cm<sup>-1</sup>) 3350 (s, NH<sub>2</sub>), 3290 (m, NHTs), 3055 (s,  $C_6H_4$ ), 2980 (s, CH), 2878 (s, CH), 2626 (s, CH), 1600 (s), 1493 (s), 1457 (s), 1311 (s,  $SO_2NH$ ), 1156 (s,  $SO_2NH$ ), 1097 (s), 815 (s,  $C_6H_4$ );  $\delta_{\rm H}$  (400 MHz, CDCl<sub>3</sub>) 0.99 (d, J 6.6, 3H, -CH<sub>3</sub>), 2,40 (s, 3H,  $C_6H_4-CH_3$ ), 2.49 (dd, J 13, 6.97, 1H,  $CH_2NH_2$ ), 2.66 (dd, J 13, 4.49, 1H, CH<sub>2</sub>NH<sub>2</sub>), 3.14–3.22 (m, 1H, CHNHTs), 7.27 (d, J 7.98, 2H,  $-C_6H_4-CH_3$ ), 7.75 (d, J 7.98, 2H,  $-C_6H_4-CH_3$ );  $\delta_{\rm C}$  (100 MHz, CDCl<sub>3</sub>) 18.99 (-CH<sub>3</sub>), 21.64 (-C<sub>6</sub>H<sub>4</sub>-CH<sub>3</sub>), 47.31 (CH<sub>2</sub>NH<sub>2</sub>), 51.87 (TsNCH), 127.19 (2 × Ar-CH), 129.81 $(2 \times Ar - CH)$ , 138.13 (Ar - C), 143.36 (Ar - C).

[(1S)-1-(Aminomethyl)-2-methylpropyl]-4-methylbenzenesulfonamide 9b. Compound 9b was prepared in an analogous fashion to 9a using (2S)-2-isopropyl-1-[(4-methylphenyl)sulfonyl]aziridine 4 (1.4 g, 5.8 mmol) which afforded a colourless solid (1.35 g, 5.2 mmol, 89%); mp 85-87 °C (lit.22a 87-88 °C); Found MH<sup>+</sup> m/z 257.1334;  $C_{12}H_{20}O_2SN_2$  requires m/z 257.1323;  $[a]_D$  –15.9  $(c = 1, CHCl_3)$  (lit.<sup>22a</sup> +12.0  $c = 1, C_6H_6)$ ;  $v_{\text{max}}$  (KBr, cm<sup>-1</sup>) 3396 (s, NH<sub>2</sub>), 3357 (s, NH<sub>2</sub>), 3297 (s, NHTs), 3070 (s, C<sub>6</sub>H<sub>4</sub>), 2959 (s, CH), 2873 (s, CH), 2778 (m, CH), 1311 (s, SO<sub>2</sub>NH), 1148 (s, SO<sub>2</sub>NH);  $\delta_{\rm H}$  (400 MHz, CDCl<sub>3</sub>) 0.79 (d, J 6.9, 3H,  $-CH_3$ ), 0.81 (d, J 6.9, 3H,  $-CH_3$ ), 1.77 (m, 1H,  $CH(CH_3)_2$ , 2.42 (s, 3H,  $-C_6H_4-CH_3$ ), 2.58 (dd, J 13.2, 4.7, 1H,  $-CH_2NH_2$ ), 2.67 (dd, J 13.2, 6.0, 1H,  $-CH_2NH_2$ ), 2.96 (m, 1H, CHNHTs), 7.29 (d, J 8.0, 2H, -C<sub>6</sub>H<sub>4</sub>-CH<sub>3</sub>), 7.78 (d, J 8.0, 2H,  $-C_6H_4$ -CH<sub>3</sub>);  $\delta_C$  (100 MHz, CDCl<sub>3</sub>) 18.7 (-CH<sub>3</sub>), 18.9  $(-CH_3)$ , 30.0  $(-C_6H_4-CH_3)$ , 30.4  $(CH(CH_3)_2)$ , 42.5  $(CH_2NH_2)$ 61.2 (TsNCH), 127.2 (2 × Ar-CH), 129.7 (2 × Ar-CH), 138.4 (Ar-C), 143.3 (Ar-C).

4-Methyl-N-{(1S)-1-methyl-2-[((2S)-2-{[(4-methylphenyl)sulfonyl] amino]propyl)amino]ethyl}benzenesulfonamide Method (a), via debenzylation of  $N-\{(1S)-2-[benzyl((2S)-a)]\}$ {(4-methylphenyl)sulfonyl]amino}propyl)amino]-1-methylethyl}-4-methylbenzenesulfonamide 2i. N-{(1S)-2-[Benzyl((2S)-{(4-methylphenyl)sulfonyl]amino}propyl)amino]-1-methylethyl}-4-methylbenzenesulfonamide 2i (1.62 g, 3.06 mmol) and Pd(OH)<sub>2</sub> (163 mg, 10 mol% Pd) in MeOH (30 cm<sup>3</sup>) under 1 atm H<sub>2</sub> at RT. After 1 h, the hydrogenation was complete and the solvent was evaporated. Purification of the crude product on silica (EtOAc) gave a colourless solid (1.15 g, 2.62 mmol, 86%); mp 131°C; Found C, 54.56; H, 6.63; N, 9.46%; MH+ m/z 440.1673; C<sub>20</sub>H<sub>29</sub>N<sub>3</sub>O<sub>4</sub>S<sub>2</sub> requires C, 54.65; H, 6.65; N, 9.56%; m/z 440.1678;  $[a]_D = -15.4$  (c = 1, CHCl<sub>3</sub>);  $v_{max}$  (KBr, cm<sup>-1</sup>) 3260 (s, SO<sub>2</sub>NH), 3060 and 3039 (w, ArH), 2970, 2930 and 2870 (all s, CH), 1600 (m, Ar), 1460 (s), 1440 (s), 1320 (s, SO<sub>2</sub>N), 1150 (s,  $SO_2N$ ), 820 (s, ArH);  $\delta_H$  (400 MHz, CDCl<sub>3</sub>) 0.95 (d, J 6.6, 6H, -CH<sub>3</sub>), 2.30-2.38 (m, 4H, -CH<sub>2</sub>NH), 2.40 (s, 6H,  $-C_6H_4-CH_3$ ), 3.23 (m, 2H, -CHTsN), 5.20 (bs, 2H, TsNH), 7.29 (d, J 7.9, 4H,  $-C_6H_4-CH_3$ ), 7.77 (d, J 7.9, 4H,  $-C_6H_4-CH_3$ )

CH<sub>3</sub>);  $\delta_{\rm C}$  (100 MHz, CDCl<sub>3</sub>) 19.4 (-*C*H<sub>3</sub>), 21.6 (-*C*H<sub>3</sub>), 49.1 (N*C*H<sub>2</sub>), 54.1 (N*C*H), 127.2 (Ar-*C*H), 129.9 (Ar-*C*H), 137.9 (Ar-*C*), 143.4 (Ar-*C*).

Method (b), via ring opening of (2S)-2-methyl-1-[(4methylphenyl)sulfonyl]aziridine 6 with [(1S)-2-amino-1-methylethyl]-4-methylbenzenesulfonamide 9a. A solution of [(1S)-2amino-1-methylethyl]-4-methylbenzenesulfonamide 9a (1.5 g, 6.6 mmol) in anhydrous toluene (25 cm<sup>3</sup>) was heated to reflux under an inert atmosphere. To this solution was added (2S)-2methyl-1-[(4-methylphenyl)sulfonyl]aziridine 6 (1.4 g, 6.6 mmol) dropwise over a 2 h period. The reaction mixture was heated at reflux temperature for an additional 48 h. On completion of this period the toluene was removed under reduced pressure and the residue was dissolved in dichloromethane (10 cm<sup>3</sup>). The dichloromethane extract was washed with 3 M HCl (×2, 15 cm<sup>3</sup>) dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and evaporated to give a light brown solid. Purification by column chromatography on silica (hexane: EtOAc: CH2Cl2 2:3:5) afforded a colourless solid (1.9 g, 4.0 mmol, 60%) with identical spectroscopic data to that recorded above.

4-Methyl-N-((1S)-2-methyl-1-{[((2S)-3-methyl-2{[(4-methylphenyl)sulfonyl]amino}butyl)amino]methyl}propyl)benzenesulfonamide 2g. Method (a), via debenzylation of ((1S)-1-{[benzyl- $((2S)-3-methyl-2-\{[(4-methylphenyl)sulfonyl]amino\}butyl)$ amino]methyl}-2-methylpropyl)-4-methylbenzenesulfonamide 2j. of  $((1S)-1-\{[benzyl((2S)-3-methyl-2-\{[(4-$ Debenzylation methylphenyl)sulfonyl]amino}butyl)amino|methyl}-2-methylpropyl)-4-methylbenzenesulfonamide 2j (1 g, 1.79 mmol) in an analogous fashion to the preparation of 2f followed by column chromatography on silica (hexane: EtOAc: CH<sub>2</sub>Cl<sub>2</sub> 2:3:5) gave the title compound as a colourless solid (695 mg, 1.4 mmol, 78%); mp 121–123 °C (lit.<sup>22a</sup> 121–122 °C); Found MH<sup>+</sup> m/z 496.2317;  $C_{24}H_{37}O_4S_2N_3$  requires m/z 496.2304;  $[a]_D$  -21.8  $(c = 0.5, \text{CHCl}_3)$  (lit. 27 – 17.0  $(c = 1.1, \text{CHCl}_3)$ );  $v_{\text{max}}$  (KBr, cm<sup>-1</sup>) 3289 (s, NHTs), 3230 (s, NH), 2957 (s, CH), 2810 (m, CH), 1322 (s, SO<sub>2</sub>NH), 1159 (s, SO<sub>2</sub>NH);  $\delta_{\rm H}$  (400 MHz, CDCl<sub>3</sub>) 0.65 (m, 12H, -CH<sub>3</sub>), 1.66 (m, 2H, CH(CH<sub>3</sub>)<sub>2</sub>), 2.33 (m, 2H, CH<sub>2</sub>NH), 2.41 (s, 6H, -C<sub>6</sub>H<sub>4</sub>-CH<sub>3</sub>), 2.45 (m, 2H, CH<sub>2</sub>NH), 3.02 (m, 2H, CHNH), 4.96 (br, 1H, NH), 7.29 (d, J 8.0, 4H,  $-C_6H_4$ –CH<sub>3</sub>), 7.76 (d, J 8.0, 4H,  $-C_6H_4$ –CH<sub>3</sub>);  $\delta_{\rm C}$  (100 MHz, CDCl<sub>3</sub>) 18.5  $(-CH_3)$ , 18.7  $(-CH_3)$ , 21.6  $(2 \times -C_6H_4-CH_3)$ , 30.4  $(CH(CH_3)_2)$ ,  $49.7 (2 \times NCH_2)$ ,  $58.8 (2 \times TsNCH)$ ,  $127.2 (4 \times Ar-CH)$ , 129.8 $(4 \times Ar - CH)$ , 138.4  $(2 \times Ar - C)$ , 143.3  $(2 \times Ar - C)$ .

Method (b), via hydrolysis of N-[(1S)-1-({(4S)-4-isopropyl-*3-[(4-methylphenyl)sulfonyl]imidazolidinyl}methyl)-2-methyl*propyl]-4-methylbenzenesulfonamide 8.  $((1S)-1-\{[Benzyl((2S)-1-(1S)-1-($ 3-methyl-2-{[(4-methylphenyl)sulfonyl]amino} butyl)amino]methyl}-2-methylpropyl)-4-methylbenzenesulfonamide 2j (3.0 g, 5.3 mmol) and Pd(OH)<sub>2</sub> (285 mg, 10 mol% Pd) were placed in MeOH (50 cm<sup>3</sup>) and dichloromethane (50 cm<sup>3</sup>) under 1 atm H<sub>2</sub>. After 3 h there was no further uptake of hydrogen and the reaction was stopped. The solvent was evaporated and the crude material was purified by column chromatography on silica (hexane : EtOAc : CH<sub>2</sub>Cl<sub>2</sub> 2 : 3 : 5) to afford N-[(1S)- $1-(\{(4S)-4-\text{isopropyl}-3-[(4-\text{methylphenyl})\text{sulfonyl}]\text{imidazol}$ idinyl}methyl)-2-methylpropyl]-4-methylbenzenesulfonamide 8 as a colourless solid (1.8 g, 3.6 mmol, 68%); mp 110–111 °C; Found MH<sup>+</sup> m/z 508.22816;  $C_{25}H_{38}O_4S_2N_3$  requires m/z508.23038;  $[a]_D = -31.0 (c = 1, CHCl_3). \nu_{max} (KBr, cm^{-1}) 3285 (s, cm^{-1})$ NHTs), 2969 (s, CH), 2823 (m, CH), 1342 (s, SO<sub>2</sub>NH), 1162 (s,  $SO_2NH$ );  $\delta_H$  (400 MHz, CDCl<sub>3</sub>) 0.6 (d, J 6.9, 3H, -CH<sub>3</sub>), 0.65  $(d, J 6.9, 3H, -CH_3), 0.84 (d, J 6.7, 3H, -CH_3), 0.65 (d, J 6.7, J 6$ 3H,  $-CH_3$ ), 1.60 (m, 1H,  $CH(CH_3)_2$ ), 2.01 (m, 1H,  $CH(CH_3)_2$ ),  $2.02 \text{ (m, 2H, } CH_2N), 2.41 \text{ (s, 6H, } -C_6H_4-CH_3), 2.47 \text{ (m, 2H, }$  $CH_2NTs$ ), 2.83 (m, 1H,  $CH_aH_bNTs$ ), 3.51 (m, 1H,  $CH_aH_bNTs$ ), 3.62 (d, J 8.4, 1H, CHNTs), 4.0 (d, J 8.4, 1H, CHNTs), 7.29 (m, 4H,  $-C_6H_4$ –CH<sub>3</sub>), 7.76 (m, 4H,  $-C_6H_4$ –CH<sub>3</sub>);  $\delta_C$  (100 MHz, CDCl<sub>3</sub>) 16.50 (-CH<sub>3</sub>), 17.7 (-CH<sub>3</sub>), 17.9 (-CH<sub>3</sub>), 19.3 (-CH<sub>3</sub>), 29.0 (2 ×  $-C_6H_4-CH_3$ ), 31.4 (CH(CH<sub>3</sub>)<sub>2</sub>), 53.4 (NCH<sub>2</sub>), 53.7

 $(NCH_2)$ , 57.4 (TsNCH), 63.8 (TsNCH) 71.5 (TsNCH<sub>2</sub>N), 127.3 (2 × Ar–CH), 127.9 (2 × Ar–CH), 129.8 (2 × Ar–CH), 129.9 (2 × Ar–CH), 135.3 (Ar–C), 137.9 (Ar–C), 143.5 (Ar–C), 143.9 (Ar–C).

To a solution of N-[(1S)-1-({(4S)-4-isopropyl-3-[(4-methylphenyl)sulfonyl]imidazolidinyl}methyl)-2-methylpropyl]-4methylbenzenesulfonamide 8 (130 mg, 0.26 mmol) in methanol (5 cm<sup>3</sup>) was added 3 M aqueous hydrochloric acid (5 cm<sup>3</sup>) and the reaction was stirred for 7 h. The reaction was cooled to 0 °C and neutralised to pH 10 with 2 M aqueous sodium hydroxide and extracted with dichloromethane (×2, 10 cm<sup>3</sup>). The combined organic extracts were concentrated under reduced pressure and the residue taken up in ethanol (5 cm<sup>3</sup>). To the resulting solution was added solid sodium hydroxide (256 mg, 6.4 mmol. 25 equiv.) and this mixture was heated to reflux for 22 h. At the completion of this period the volatiles were removed under reduced pressure. The residue was suspended in water (10 cm<sup>3</sup>) and this was acidified to pH 8 with 12 M aqueous hydrochloric acid. The aqueous phase was extracted with dichloromethane (×3, 10 cm<sup>3</sup>) then dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and evaporated. Purification by column chromatography on silica (hexane : EtOAc : CH<sub>2</sub>Cl<sub>2</sub> 2 : 3 : 5) gave 4-methyl-N-((1S)-2methyl-1- $\{[((2S)-3-methyl-2)](4-methyl-1-1)\}$ phenyl)sulfonyl]amino}butyl)amino]methyl}propyl)benzenesulfonamide 2g as a colourless solid (116.7 mg, 0.24 mmol, 92%); Found C, 58.25; H, 7.60; N, 8.25; S, 13.09%; MH+ m/z 496.2295; C<sub>24</sub>H<sub>37</sub>N<sub>3</sub>O<sub>4</sub>S<sub>2</sub> requires C, 58.15; H, 7.52; N, 8.48; S,

Method (c), via ring opening of (2S)-2-isopropyl-1-[(4-methylphenyl)sulfonyl]aziridine 4 with (1S)-1-(aminomethyl)-2-methylpropyl]-4-methylbenzenesulfonamide 9b. The title compound was obtained in an analogous fashion to the preparation of compound 2f using [(1S)-1-(aminomethyl)-2-methylpropyl]-4-methylbenzenesulfonamide 9b (1.3 g, 5 mmol) and (2S)-2-isopropyl-1-[(4-methylphenyl)sulfonyl]aziridine 4 (1.2 g, 5 mmol) followed by purification by column chromatography on silica (hexane: EtOAc: CH<sub>2</sub>Cl<sub>2</sub> 2: 3: 5) gave the title compound as a white solid (1.4 g, 2.8 mmol, 56%) with spectroscopic data identical to that recorded above.

12.94%; m/z 496.2304. The remaining spectroscopic data was

identical to that recorded above.

# 4-Methyl-*N*-{(1*S*)-1-methyl-2-[((2*S*)-3-methyl-2-{[(4-methyl-phenyl)sulfonyl]amino}butyl)amino]ethyl}benzenesulfonamide

**2h.** Method (a), via debenzylation of N-{(1S)-2-[benzyl((2S)-3-methyl-2-{[(4-methylphenyl)sulfonyl]amino}butyl)amino]-1methylethyl}-4-methylbenzenesulfonamide  $[Benzyl((2S)-3-methyl-2-{[(4-methylphenyl)sulfonyl]amino}$ butyl)amino]-1-methylethyl}-4-methylbenzenesulfonamide 2k (300 mg, 0.54 mmol) and palladium on charcoal (57 mg, 10 mol% Pd) in AcOH (5 cm<sup>3</sup>) were hydrogenated under 1 atm H<sub>2</sub> at RT for 12 h. After evaporation of the solvent, the crude product was filtered through a short pad of silica (EtOAc: EtOH 1:1). After evaporation of the solvent, 2 M aqueous sodium hydroxide was added to the amine which was then extracted with EtOAc, washed with 2 M NaOH and brine, and dried over MgSO<sub>4</sub>. Purification by flash column chromatography on silica (EtOAc) gave an oil (100 mg, 0.21 mmol, 40%); Found MH<sup>+</sup> m/z 468.1981; Calculated for C<sub>22</sub>H<sub>34</sub>N<sub>3</sub>O<sub>4</sub>S<sub>2</sub> m/z461.1991;  $[a]_D = -13.8 (c = 1.00; CHCl_3); v_{max} (CCl_4, cm^{-1}) 3275$ (s, NH), 3064 and 3043 (w, ArH). 2965 (s, CH), 2933 (m, CH), 2878 (m, CH), 1463 (s), 1492 (s), 1326 (s, SO<sub>2</sub>NH), 1161 (s,  $SO_2NH$ ), 815 (m, ArH);  $\delta_H$  (400 MHz, CDCl<sub>3</sub>) 0.74 (d, J 6.9, 3H,  $CH_3$ ), 0.76 (d, J 6.9, 3H,  $CH_3$ ), 0.96 (d, J 6.5, 3H,  $CH_3$ ), 1.67 (m, 1H,  $CH(CH_3)_2$ ), 2.29–2.46 (m, 4H,  $CH_2NH$ ), 2.41 (s, 6H,  $-C_6H_4-CH_3$ ), 3.02 (m, 1H, CHNH), 3.22 (m, 1H, CHNH), 5.05 (bs, 2H, NH), 7.28–7.32 (m, 4H,  $-C_6H_4$ –CH<sub>3</sub>), 7.75–7.78 (m, 4H,  $-C_6H_4$ -CH<sub>3</sub>);  $\delta_C$  (100 MHz, CDCl<sub>3</sub>) 18.4 (-CH<sub>3</sub>), 18.7  $(-CH_3)$ , 19.5  $(-CH_3)$ , 21.6  $(2 \times -C_6H_4-CH_3)$ , 30.6  $(CH(CH_3)_2)$ , 49.0 (NCH<sub>2</sub>), 49.5 (NCH<sub>2</sub>), 54.4 (TsNCH), 58.9 (TsNCH), 127.2 (Ar-CH), 127.3)Ar-CH), 129.8 (Ar-CH), 129.9

(Ar–CH), 138.0 (Ar–C), 138.3 (Ar–C), 143.4 (Ar–C), 143.5 (Ar–C).

Method (b), via ring opening of (2S)-2-isopropyl-1-[(4methylphenyl) sulfonyl] aziridine 4 with [(1S)-2-amino-1-methylethyl]-4-methylbenzenesulfonamide 9a. The title compound was obtained in an analogous fashion to compound 2f using [(1S)-2-amino-1-methylethyl]-4-methylbenzenesulfonamide 9a (1.458 g, 6.396 mmol) and (2S)-2-isopropyl-1-[(4-methylphenyl)sulfonyl]aziridine 4 (1.526 g, 6.38 mmol) and flash column chromatography on silica (CH<sub>2</sub>Cl<sub>2</sub>: MeOH 92: 8) afforded a colourless gum (2.621 g, 5.61 mmol, 88%); mp 30-32.5 °C; Found MH<sup>+</sup> m/z 468.2018;  $C_{22}H_{34}N_3O_4S_2$  requires m/z468.1991;  $[a]_D = -10.2 (c = 1.01; CHCl_3); \nu_{max} (CCl_4, cm^{-1}) 3275$ (s, NH) 3064 and 3043 (w, ArH), 2965 (s, CH), 2933 (m, CH), 2878 (m, CH), 1463 (s), 1429 (s), 1326 (s, SO<sub>2</sub>NH), 1161 (s,  $SO_2NH$ ), 815 (m, ArH);  $\delta_H$  (400 MHz, CDCl<sub>3</sub>) 0.72 (d, J 6.9,  $3H_{1}-CH_{2}$ , 0.75 (d, J 6.9,  $3H_{2}-CH_{3}$ ), 0.93 (d, J 6.5,  $3H_{2}-CH_{3}$ ), 1.66 (m, 1H,  $CH(CH_3)_2$ ), 2.35 (m, 4H,  $-CH_2NH$ ), 2.40 (s, 6H,  $-C_6H_4-CH_3$ ), 3.00 (m, 1H, CHNHTs), 3.21 (m, 1H, CHNHTs), 7.28 (m, 4H,  $-C_6H_4-CH_3$ ), 7.76 (m, 4H,  $-C_6H_4-CH_3$ );  $\delta_C$  (100 MHz, CDCl<sub>3</sub>) 18.4 (-CH<sub>3</sub>), 18.7 (-CH<sub>3</sub>), 19.5 (-CH<sub>3</sub>), 21.6  $(2 \times -C_6H_4-CH_3)$ , 30.5 (CH(CH<sub>3</sub>)<sub>2</sub>), 48.97 (NCH<sub>2</sub>), 49.4 (NCH<sub>2</sub>), 54.4 (TsNCH), 58.8 (TsNCH), 127.2 (Ar-CH), 127.3)Ar-CH), 129.8 (Ar-CH), 129.9 (Ar-CH), 138.0 (Ar-C), 138.4 (Ar–C), 143.4 (Ar–C), 143.4 (Ar–C).

4-Methyl-N-{(1S)-1-methyl-2-[[(4-methylphenyl)sulfonyl]- $((2S)-2-\{[(4-methylphenyl)sulfonyl]amino\}propyl)amino]ethyl\}$ benzenesulfonamide 2a. To a solution of 4-methyl-N- $\{(1S)$ -1methyl-2- $[((2S)-2-\{[(4-methylphenyl)sulfonyl]amino]propyl)$ aminolethyl}benzenesulfonamide 2f (1.3 g, 2.8 mmol) in pyridine (20 cm<sup>3</sup>) at 0 °C under a nitrogen atmosphere was added p-toluenesulfonyl chloride (550 mg, 2.8 mmol) in batches over 30 min. The orange solution was allowed to come to room temperature and stirred for a further 6 h whereby the colour changed from orange to red. The reaction was quenched with ice (~100 g) and conc. HCl (25 cm<sup>3</sup>). This mixture was extracted with dichloromethane (×2, 25 cm<sup>3</sup>), and the combined organic extracts were dried (Na2SO4), filtered and evaporated to afford a crude dark brown oil. Purification by column chromatography on silica (hexane : EtOAc : CH<sub>2</sub>Cl<sub>2</sub> 4 : 1 : 5) afforded a yellow solid (1.5 g, 2.5 mmol, 89%); mp 68-70 °C; Found C, 54.5; H, 5.9; N, 6.8%;  $MH^+$  m/z 594.1760;  $C_{27}H_{35}O_6S_3N_3$ requires C, 54.6; H, 5.9; N, 7.1%; m/z 594.1766; [a]<sub>D</sub> -42.3  $(c = 1, \text{CHCl}_3); \nu_{\text{max}} (\text{KBr}, \text{cm}^{-1}) 3275 \text{ (s, N} \text{HTs)}, 3034 \text{ (w, C}_6 \text{H}_4),$ 2977 (m, CH), 2926 (m, CH), 2874 (m, CH), 1331 (s, SO<sub>2</sub>NH), 1159 (s,  $SO_2NH$ );  $\delta_H$  (400 MHz, CDCl<sub>3</sub>) 0.96 (d, J 6.6, 6H,  $-CH_3$ ), 2.46 (s, 6H,  $-C_6H_4-CH_3$ ), 2.48 (s, 3H,  $-C_6H_4-CH_3$ ), 2.84 (dd, J 14.8, 6.2, 2H, -CH<sub>2</sub>NTs), 3.16 (dd, J 14.8, 6.2, 2H,  $-CH_2NTs$ ), 3.40 (m, 2H, -CHNHTs), 7.31 (d, J 8.3, 4H,  $-C_6H_4$  $CH_3$ ), 7.36 (d, J 8.3, 2H,  $-C_6H_4-CH_3$ ), 7.69 (d, J 8.3, 2H,  $-C_6H_4-CH_3$ ), 7.77 (d, J 8.3, 4H,  $-C_6H_4-CH_3$ );  $\delta_C$  (100 MHz, CDCl<sub>3</sub>) 19.2 (2 ×  $-CH_3$ ), 21.7 ( $-C_6H_4-CH_3$ ), 21.7 (2 ×  $-C_6H_4-CH_3$ )  $CH_3$ ), 48.7 (2 × TsN $CH_2$ ), 60.6 (2 × TsNCH), 127.3 (2 × Ar– CH), 127.6 (4  $\times$  Ar–CH), 129.9 (4  $\times$  Ar–CH), 130.2 (4  $\times$  Ar– CH), 135.21 (2  $\times$  Ar-C), 137.8 (2  $\times$  Ar-C), 143.7 (Ar-C), 144.3 (Ar-C).

4-Methyl-N-[(1S)-2-methyl-1-({((2S)-3-methyl-2-{[(4-methylphenyl)sulfonyl]amino}butyl)[(4-methylphenyl)sulfonyl]-amino}methyl)propyl]benzenesulfonamide 2b. Compound 2b was prepared in an identical manner to 2a using 4-methyl-N-((1S)-2-methyl-1-{[((2S)-3-methyl-2{[(4-methylphenyl)-sulfonyl]amino}butyl)amino]methyl} propyl)benzenesulfonamide 2g (1.4 g, 2.8 mmol). Purification by column chromatography on silica (hexane : EtOAc :  $CH_2Cl_2$  4 : 1 : 5) afforded a yellow solid (1.5 g, 2.3 mmol, 82%); mp 209–211 °C (lit.  $^{20}$  mp 205–207 °C); Found C, 57.3; H, 6.7; N, 6.5; S, 14.8%; MH+ m/z 650.2404;  $C_{31}H_{43}O_6S_3N_3$  requires C, 57.3; H, 6.8; N, 6.2; S, 14.7%; m/z 650.2405;  $[a]_D$  -73.1 (c = 1,  $CHCl_3$ ) (lit.  $^{20}$  -30.4

 $\begin{array}{l} (c=0.5,\,\mathrm{CHCl_3});\; \nu_{\mathrm{max}}\; (\mathrm{KBr},\,\mathrm{cm^{-1}})\; 3250\; (\mathrm{s},\,NH\mathrm{Ts}),\; 3044\; (\mathrm{w},\,\mathrm{C}_{\mathrm{6}\mathrm{H_4}}),\; 2961\; (\mathrm{m},\,\mathrm{CH}),\; 2928\; (\mathrm{m},\,\mathrm{CH}),\; 2874\; (\mathrm{m},\,\mathrm{CH}),\; 1325\; (\mathrm{s},\,\mathrm{SO_2NH}),\; 1162\; (\mathrm{s},\,\mathrm{SO_2NH});\; \delta_{\mathrm{H}}\; (400\;\mathrm{MHz},\,\mathrm{CDCl_3})\; 0.58\; (\mathrm{d},\,J\,6.9,\,\mathrm{GH},\,-\mathrm{C}H_3),\; 0.74\; (\mathrm{d},\,J\,6.9,\,\mathrm{GH},\,-\mathrm{C}H_3),\; 1.84\; (\mathrm{m},\,\mathrm{2H},\,\mathrm{C}H(\mathrm{CH_3})_2),\; 2.42\; (\mathrm{s},\,\mathrm{GH},\,-\mathrm{C}_{\mathrm{H_4}}-\mathrm{C}H_3),\; 2.46\; (\mathrm{s},\,\mathrm{3H},\,-\mathrm{C}_{\mathrm{6}\mathrm{H_4}}-\mathrm{C}H_3),\; 3.01\; (\mathrm{m},\,\mathrm{4H},\,-\mathrm{C}H_2\mathrm{NHTs}),\; 3.18\; (\mathrm{m},\,\mathrm{2H},\,-\mathrm{C}H\mathrm{NHTs}),\; 5.00\; (\mathrm{d},\,J\,8.0,\,\mathrm{2H},\,\mathrm{NHTs}),\; 7.28\; (\mathrm{d},\,J\,8.0,\,\mathrm{4H},\,-\mathrm{C}_{\mathrm{6}\mathrm{H_4}}-\mathrm{CH_3}),\; 7.36\; (\mathrm{d},\,J\,8.0,\,\mathrm{2H},\,-\mathrm{C}_{\mathrm{6}\mathrm{H_4}}-\mathrm{CH}_3),\; 7.67\; (\mathrm{d},\,J\,8.3,\,\mathrm{2H},\,-\mathrm{C}_{\mathrm{6}\mathrm{H_4}}-\mathrm{CH_3}),\; 7.74\; (\mathrm{d},\,J\,8.3,\,\mathrm{4H},\,-\mathrm{C}_{\mathrm{6}\mathrm{H_4}}-\mathrm{CH}_3);\; \delta_{\mathrm{C}}\; (100\;\mathrm{MHz},\,\mathrm{CDCl_3})\; 16.6\; (2\times-\mathrm{C}\mathrm{H_3}),\; 17.9\; (2\times-\mathrm{C}\mathrm{H_3}),\; 21.7\; (3\times-\mathrm{C}_{\mathrm{6}\mathrm{H_4}}-\mathrm{C}\mathrm{H_3}),\; 28.5\; (2\times\mathrm{CH}(\mathrm{CH_3})_2),\; 51.0\; (2\times\mathrm{TsNCH_2}),\; 57.2\; (2\times\mathrm{TsNCH}),\; 127.4\; (2\times\mathrm{Ar-CH}),\; 127.7\; (3\times\mathrm{Ar-CH}),\; 129.7\; (3\times\mathrm{Ar-CH}),\; 130.2\; (3\times\mathrm{Ar-CH}),\; 134.8\; (2\times\mathrm{Ar-C})\; 137.7\; (2\times\mathrm{Ar-C}),\; 143.5\; (2\times\mathrm{Ar-C}). \end{array}$ 

4-Methyl-*N*-((1*S*)-1-methyl-2-{((2*S*)-3-methyl-2-{[(4-methylphenyl)sulfonyl]amino}butyl)[(4-methylphenyl)sulfonyl]amino}ethyl)benzenesulfonamide 2c. Compound 2c was prepared in an identical manner to 2a using 4-methyl-N-{(1S)-1-methyl-2- $[((2S)-3-methyl-2-\{[(4-methyl-phenyl)sulfonyl]amino\}butyl)$ aminolethyl}benzenesulfonamide 2h (1.916 mg, 4.1 mmol). Purification by silica flash column chromatography (CH<sub>2</sub>Cl<sub>2</sub>: EtOAc 94: 6) afforded a colourless solid (1.838 g, 2.96 mmol, 72%); mp 165–166 °C; Found C, 55.77; H, 6.12; N, 6.66; S, 15.65%; MH<sup>+</sup> m/z 622.2112; C<sub>29</sub>H<sub>40</sub>N<sub>3</sub>O<sub>6</sub>S<sub>3</sub> requires C, 56.02; H, 6.32; N, 6.76; S, 15.47%; m/z 622.2079;  $[a]_D$  -67 (c = 0.505, CHCl<sub>3</sub>);  $v_{\text{max}}$  (CCl<sub>4</sub>, cm<sup>-1</sup>) 3281 (w, NHTs), 2965, 2927 and 2873 (m, all CH), 1164 (w,  $SO_2N$ ), 908 (s), 739 (s);  $\delta_H$  (400 MHz,  $CDCl_3$ ) 0.51 (d, J 6.9, 3H,  $-CH_3$ ), 0.69 (d, J 6.9, 3H,  $-CH_3$ ),  $0.95 (d, J 6.5, 3H, -CH_3), 1.74 (m, 1H, CH(CH_3)_2), 2.39 (s, 3H, -CH_3), 1.74 (m, 1H, CH_3)_2), 2.39 (s, 3H, -CH_3)_2)$  $-C_6H_4-CH_3$ ), 2.41 (s, 3H,  $-C_6H_4-CH_3$ ), 2.43 (s, 3H,  $-C_6H_4-CH_3$ )  $CH_3$ ), 2.70 (dd, J 14.9, 6.2, 1H, -CHHNHTs), 2.91 (dd, J 14.1, 6.2, 1H, -CHHNHTs), 3.12 (m, 3H, -CHNHTs), 3.35 (m, 1H, -CHNHTs), 5.08 (d, J 6.2, 1H, NHTs), 5.44 (bs, 1H, NHTs), 7.27 (m, 6H,  $-C_6H_4-CH_3$ ), 7.69 (m, 12H,  $-C_6H_4-CH_3$ );  $\delta_{\rm C}$  (100 MHz, CDCl<sub>3</sub>) 16.15 (-CH<sub>3</sub>), 18.42 (-CH<sub>3</sub>), 19.43  $(-CH_3)$ , 21.66  $(2 \times -C_6H_4-CH_3)$ , 21.70  $(-C_6H_4-CH_3)$  28.25  $(CH(CH_3)_2)$ , 48.99  $(TsNCH_2)$ , 51.50  $(TsNCH_2)$ , 55.27 (TsNCH), 56.97 (TsNCH), 127.4, 127.37, 127.72, 129.82. 129.88 and 130.15 (all Ar-CH), 134.86, 137.55, 137.86 (all Ar-C), 143.52, 143.69, 144.30 (all Ar-C).

(2S,6S)-2,6-Dimethyl-1,4,7-tris[(4-methylphenyl)sulfonyl]-1,4,7-triazacyclononane 1a. Hexane washed NaH (42 mg, 1.7 mmol) in DMF (1 cm<sup>3</sup>) was added to a solution of 4-methyl- $N-\{(1S)-1-\text{methyl}-2-[[(4-\text{methylphenyl})\text{sulfonyl}]((2S)-2-\{[(4-\text{methyl$ sulfonyl]amino}propyl)amino]ethyl}benzenesulfonamide 2a (500 mg, 0.84 mmol) in DMF (8 cm<sup>3</sup>) at room temperature. The resulting solution was heated to 80 °C and a solution of ethyleneglycol ditosylate 3 (343 mg, 0.92 mmol) in DMF (2 cm<sup>3</sup>) was added dropwise over 1 h. During this period the colour of the solution changed from brown to black. The reaction was stirred overnight at 80 °C and at completion was quenched by addition of water (10 cm<sup>3</sup>). The DMF was removed under reduced pressure (12 mmHg) and the residue dissolved in EtOAc (15 cm<sup>3</sup>). The organic layer was washed with water (×2, 20 cm<sup>3</sup>), dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and evaporated to give a brown solid. Purification by column chromatography on silica (hexane: EtOAc: CH<sub>2</sub>Cl<sub>2</sub>4:1:5) afforded a colourless solid (374 mg, 0.60 mmol, 71%); mp 209-211 °C; Found C, 56.0; H, 6.0; N, 6.5%; MH+ m/z 620.1926; C<sub>29</sub>H<sub>37</sub>O<sub>6</sub>S<sub>3</sub>N<sub>3</sub> requires C, 56.2; H, 6.0; N, 6.8%; m/z 620.1923;  $[a]_D$  +95.3 (c = 1, CHCl<sub>3</sub>);  $v_{\text{max}}$  (KBr, cm<sup>-1</sup>) 3064 (w, C<sub>6</sub>H<sub>4</sub>), 2924 (w, CH), 1335 (s, SO<sub>2</sub>NH), 1155 (s, SO<sub>2</sub>NH);  $\delta_{\rm H}$  (400 MHz,  $CDCl_3$ ) 0.69 (d, J 6.5, 6H,  $-CH_3$ ), 2.43 (s, 6H,  $-C_6H_4-CH_3$ ), 2.46 (s, 3H,  $-C_6H_4-CH_3$ ), 3.24 (d, J 13.8, 2H,  $-CH_2NTs$ ), 3.52 (m, 6H, -CH<sub>2</sub>NTs), 4.15 (m, 2H, -CHNTs), 7.30 (d, 8.0, 4H,  $-C_6H_4$ -CH<sub>3</sub>), 7.35 (d, J 8.0, 2H,  $-C_6H_4$ -CH<sub>3</sub>), 7.65 (d, J 8.3, 4H,  $-C_6H_4-CH_3$ ), 7.78 (d, J 8.3, 2H,  $-C_6H_4-CH_3$ );  $\delta_C$  (100 MHz, CDCl<sub>3</sub>) 14.9 ( $-CH_3$ ), 21.7 (3 ×  $-C_6H_4-CH_3$ ), 45.6  $(4 \times TsNCH_2)$ , 53.8  $(2 \times TsNCH)$ , 127.2  $(3 \times Ar-CH)$ , 127.7

(3 × Ar–*C*H), 129.9 (3 × Ar–*C*H), 130.0 (3 × Ar–*C*H), 137.1 (2 × Ar–*C*), 143.6 (2 × Ar–*C*), 143.9 (2 × Ar–*C*).

(2S,6S)-2,6-Diisopropyl-1,4,7-tris[(4-methylphenyl)sulfonyl]-1,4,7-triazacyclononane 1b. Compound 1b was prepared in an identical manner to 1a from 4-methyl-N-[(1S)-2-methyl-1- $(\{((2S)-3-methyl-2-\{[(4-methylphenyl)sulfonyl]amino\}butyl)-$ [(4-methylphenyl)sulfonyl]amino}methyl)propyl]benzenesulfonamide 2b (800 mg, 1.2 mmol) and washed sodium hydride (103 mg, 2.5 mmol). Purification by column chromatography on silica (hexane: EtOAc: CH<sub>2</sub>Cl<sub>2</sub> 4:1:5) afforded a colourless solid (514 mg, 0.87 mmol, 73%); mp 210-212 °C (lit.20 mp 214–215 °C); Found MH<sup>+</sup> m/z 676.2540; C<sub>33</sub>H<sub>46</sub>O<sub>6</sub>S<sub>3</sub>N<sub>3</sub> requires m/z 676.2549;  $[a]_D = -76.1$  (c = 1, CHCl<sub>3</sub>) (lit.<sup>20</sup>  $[a]_D = -7.4$  $(c = 0.5, \text{CHCl}_3)); \nu_{\text{max}} (\text{KBr}, \text{cm}^{-1}) 2972 (\text{w}, \text{CH}), 2880 (\text{w}, \text{CH}),$ 2720 (w, CH), 1338 (s, SO<sub>2</sub>NH), 1172 (s, SO<sub>2</sub>NH);  $\delta_{\rm H}$  (400 MHz,  $CDCl_3$ ) 0.38 (br, 6H,  $-CH_3$ ), 0.88 (d, J 6.2, 6H,  $-CH_3$ ), 1.01 (m, 2H,  $CH(CH_3)_2$ ), 2.40 (s, 6H,  $-C_6H_4-CH_3$ ), 2.47 (s, 3H,  $-C_6H_4-CH_3$ )  $CH_3$ ), 3.21 (m, 4H,  $-CH_2NTs$ ), 3.40 (m, 2H, -CHNTs), 3.66 (m, 4H, -CH<sub>2</sub>NTs), 3.78 (m, 2H, -CHNTs), 7.28 (d, J 8.0, 4H,  $-C_6H_4-CH_3$ ), 7.37 (d, J 8.0, 2H,  $-C_6H_4-CH_3$ ), 7.73 (m, 6H,  $-C_6H_4-CH_3$ );  $\delta_C$  (100 MHz, CDCl<sub>3</sub>) 20.2 (2 × -CH<sub>3</sub>), 20.6  $(2 \times -CH_3)$ , 21.6  $(3 \times -C_6H_4-CH_3)$ , 21.7  $(2 \times CH(CH_3)_2)$ , 45.7 (4 × TsNCH<sub>2</sub>), 64.6 (2 × TsNCH), 127.7 (4 × Ar–CH),  $128.2 (4 \times Ar-CH), 129.8 (4 \times Ar-CH), 129.9 (2 \times Ar-C),$  $143.6 (2 \times Ar - C), 144.1 (2 \times Ar - C).$ 

(2S,6S)-2-Isopropyl-6-methyl-1,4,7-tris[(4-methylphenyl)sulfonyl]-1,4,7-triazacyclonane 1c. Compound 1c was prepared in an identical manner to 1a from 4-methyl-N-((1S)-1-methyl- $2-\{((2S)-3-\text{methyl}-2-\{[(4-\text{methylphenyl})\text{sulfonyl}]\text{amino}\}\text{butyl}\}$ [(4-methylphenyl)sulfonyl]amino}ethyl)benzenesulfonamide 2c (820.6 mg, 1.32 mmol) and washed sodium hydride (108.6 mg, 2.71 mmol). Purification by column chromatography on silica (EtOAc: CH<sub>2</sub>Cl<sub>2</sub> 2: 98) afforded a colourless solid (618 mg, 0.955 mmol, 72%), an analytical sample was obtained by purification on alumina (CH<sub>2</sub>Cl<sub>2</sub>) mp 105-106 °C; Found C, 57.14; H, 6.40; N, 6.41; S, 15.15%; MH<sup>+</sup> m/z 648.2260; C<sub>31</sub>H<sub>41</sub>N<sub>3</sub>O<sub>6</sub>S<sub>3</sub> requires C, 57.47; H, 6.38; N, 6.49; S, 14.85%; m/z 648.2236; [a]<sub>D</sub> = 80.8 (c = 0.505 CHCl<sub>3</sub>));  $v_{\text{max}}$  (KBr, cm<sup>-1</sup>) 3063 (w, ArH), 3029 (w, ArH), 2967 (m, CH), 1452 (m), 1392 (m), 1336 (s, SO<sub>2</sub>N), 1155 (s, SO<sub>2</sub>N), 1089 (s), 982 (s), 822 (m, ArH);  $\delta_{\rm H}$  (400 MHz,  $CDCl_3$ ) 0.33 (d, J 6.8, 3H,  $-CH_3$ ), 0.63 (d, J 6.8, 3H,  $-CH_3$ ), 0.89 (d, J 6.5, 3H,  $-CH_3$ ), 1.47 (m, 1H,  $CH(CH_3)_2$ ), 2.41 (s, 3H,  $-C_6H_4-CH_3$ ), 2.43 (s, 3H,  $-C_6H_4-CH_3$ ), 2.47 (s, 3H,  $-C_6H_4-CH_3$ )  $CH_3$ ), 3.37 (m, 5H, 2 ×– $CH_2$ NTs + –CHNTs), 3.72 (m, 4H, 2 ×  $-CH_2NHTs$ ), 4.20 (bs, 1H, CHNTs), 7.29 (m, 4H,  $-C_6H_4-CH_3$ ), 7.36 (d, J 8, 2H,  $-C_6H_4$ –CH<sub>3</sub>), 7.67 (m, 4H,  $-C_6H_4$ –CH<sub>3</sub>), 7.76 (d, J 8, 2H,  $-C_6H_4$ –CH<sub>3</sub>);  $\delta_C$  (100 MHz, CDCl<sub>3</sub>) 15.22 ( $-CH_3$ ),  $20.19 (-CH_3)$ ,  $20.55 (-CH_3)$ ,  $21.66 (2 \times -C_6H_4-CH_3)$ , 21.69 $(-C_6H_4-CH_3)$ , 29.3  $(CH(CH_3)_2)$ , 44.60  $(TsNCH_2)$ , 47.06  $(TsNCH_2)$ , 53.81  $(TsNCH_2)$  64.6 (2 × TsNCH), 127.23 (Ar-CH), 127.63 (Ar-CH), 127.75 (Ar-CH), 129.81 (Ar-CH), 130.02 (Ar-CH), 137.18 (Ar-C), 143.60 (Ar-C), 143.75 (Ar-C), 143.88 (Ar-C).

(2S,6S)-2,6-Dimethyl-1,4,7-triazacyclononane 10a. To a solution of (2S,6S)-2,6-dimethyl-1,4,7-tris[(4-methylphenyl)-sulfonyl]-1,4,7-triazacyclononane 1a (912 mg, 1.5 mmol) in THF (25 cm³) and EtOH (4.5 cm³, 80 mmol) was condensed dry NH₃ (200 cm³) at -78 °C. To this solution was added lithium metal (514 mg, 74 mmol) in small portions to give an intense blue colour. The reaction mixture was allowed to warm to room temperature overnight. Water was added (10 cm³) and the solution was acidified (pH 1) with conc. HCl (1 cm³). The aqueous solution was extracted with dichloromethane (×2, 10 cm³). The aqueous phase was made basic (pH 14) by addition of solid NaOH (~500 mg). The basic solution was extracted with dichloromethane (×4, 10 cm³), EtOAc (×2, 10 cm³) and the combined organic phases were dried (Na₂SO₄), filtered and

evaporated to give a light yellow oil (38 mg, 0.24 mmol, 16%); Found MH<sup>+</sup> m/z 158.1654;  $C_8H_{19}N_3$  requires m/z 158.1657;  $[a]_D$  +41.3 (c=1, CHCl<sub>3</sub>);  $\nu_{\rm max}$  (CCl<sub>4</sub>, cm<sup>-1</sup>) 3291 (br, w, NH), 2872 (s, CH), 2809 (s, CH);  $\delta_{\rm H}$  (400 MHz, CDCl<sub>3</sub>) 1.13 (d, J 6.5, 6H, -C $H_3$ ), 2.69 (m, 2H, NHC $H_2$ ), 2.95 (m, 6H, NHC $H_2$ ), 3.01 (m, 2H, NHC $H_3$ ), 3.51 (br, 3H, NH);  $\delta_C$  (100 MHz, CDCl<sub>3</sub>) 18.3 (2 × -CH<sub>3</sub>), 42.5 (2 × NHCH<sub>2</sub>), 50.7 (NHCH<sub>2</sub>), 51.9 (NHCH<sub>2</sub>), 53.6 (2 × NHCH).

(2*S*,6*S*)-2,6-Dimethyl-1,4,7-triazacyclononane trihydrochloride 10d. To a solution of (2*S*,6*S*)-2,6-dimethyl-1,4,7-triazacyclononane 10a (38 mg, 0.14 mmol) in EtOH (2 cm³) was added conc. HCl (0.08 cm³) at room temperature with rapid stirring. To this solution was added ether (10 cm³) and the white precipitate that formed was removed by filtration and dried under reduced pressure to give the hydrochloride salt as a colourless solid (49 mg, 0.19 mmol, 79%); mp 133–135 °C; Found MH<sup>+</sup> *m*/*z* 158.1654; C<sub>8</sub>H<sub>19</sub>N<sub>3</sub> requires *m*/*z* 158.1657; [a]<sub>D</sub> +33.1 (c = 1, MeOH);  $\nu$ <sub>max</sub> (KBr, cm<sup>-1</sup>) 2963 (s, CH), 2871 (s, CH), 2805 (s, CH);  $\delta$ <sub>H</sub> (400 MHz, CD<sub>3</sub>OD) 1.33 (d, *J* 6.6, 6H, –C*H*<sub>3</sub>), 2.94 (m, 2H, NHC*H*<sub>2</sub>), 3.19 (dd, *J* 14.6, 4.2 2H, NHC*H*<sub>2</sub>), 3.31 (m, 2H, NHC*H*<sub>2</sub>), 3.47 (m, 2H, NHC*H*<sub>2</sub>), 3.68 (m, 2H, NHC*H*);  $\delta$ <sub>C</sub> (100 MHz, CD<sub>3</sub>OD) 16.1 (–*C*H<sub>3</sub>), 17.8 (–*C*H<sub>3</sub>), 40.9 (2 × NH*C*H<sub>2</sub>), 51.6 (2 × NH*C*H<sub>2</sub>), 57.8 (2 × NH*C*H).

(2*S*,6*S*)-2,6-Diisopropyl-1,4,7-triazacyclononane 10b. Compound 10b was prepared in an identical fashion to 10a using (2*S*,6*S*)-2,6-diisopropyl-1,4,7-tris[(4-methylphenyl)sulfonyl]-1,4,7-triazacyclononane 1b (400 mg, 0.59 mmol), EtOH (1.8 cm³, 32 mmol) and lithium metal (206 mg, 29 mmol) to give a light yellow oil (92 mg, 0.43 mmol, 72%); Found MH<sup>+</sup> *m/z* 214.2283; C<sub>12</sub>H<sub>27</sub>N<sub>3</sub> requires *m/z* 214.2283; [*a*]<sub>D</sub> +79.2 (*c* = 1, CHCl<sub>3</sub>);  $\nu_{\text{max}}$  (CCl<sub>4</sub>, cm<sup>-1</sup>) 3316 (w, NH), 2958 (m, CH), 2871 (m, CH);  $\delta_{\text{H}}$  (400 MHz, CDCl<sub>3</sub>) 0.88 (d, *J* 6.7, 6H, -C*H*<sub>3</sub>), 0.96 (d, *J* 6.7, 6H, -C*H*<sub>3</sub>), 1.56 (m, 2H, C*H*(CH<sub>3</sub>)<sub>2</sub>), 2.44 (m, 4H, NHC*H*<sub>2</sub>), 2.64 (m, 2H, NHC*H*<sub>2</sub>), 2.76 (m, 2H, NHC*H*<sub>2</sub>), 3.01 (d, *J* 8.9, 2H, NHC*H*), 3.25 (br, 3H, NH);  $\delta_{\text{C}}$  (100 MHz, CDCl<sub>3</sub>) 19.7 (2 × -CH<sub>3</sub>), 19.9 (2 × -CH<sub>3</sub>), 31.9 (2 × CH(CH<sub>3</sub>)<sub>2</sub>), 43.9 (2 × NHCH<sub>2</sub>), 48.7 (2 × NHCH<sub>2</sub>), 58.9 (2 × NHCH).

(2S,6S)-2,6-Diisopropyl-1,4,7-triazacyclononane monohydrochloride 10e. Compound 10e was prepared in an analogous fashion to 10d using (2S,6S)-2,6-diisopropyl-1,4,7-triazacyclononane 10b (92 mg, 0.43 mmol) to afford a colourless solid (98 mg, 0.39 mmol, 90%); mp 136–138 °C; Found MH+ m/z 214.2283;  $C_{12}H_{27}N_3$  requires m/z 214.2283;  $[a]_D$  +58.7 (c = 1, MeOH);  $\nu_{max}$  (KBr, cm<sup>-1</sup>) 3416 (w, NH); 2963 (s, CH), 2766 (s, CH), 2650 (s, CH);  $\delta_H$  (400 MHz, D<sub>2</sub>O) 0.94 (d, J 6.8, 6H, –C $H_3$ ), 1.99 (m, 2H, CH(CH<sub>3</sub>)<sub>2</sub>), 2.98 (m, 2H, NHC $H_2$ ), 3.31 (m, 2H, NHC $H_2$ ), 3.48 (m, 6H, NHC $H_2$ ), 3.48 (m, 2H, NHC $H_3$ );  $\delta_C$  (100 M, D<sub>2</sub>O) 18.1 (2 × –CH<sub>3</sub>), 19.0 (2 × –CH<sub>3</sub>), 29.6 (2 × CH(CH<sub>3</sub>)<sub>2</sub>), 40.9 (2 × NHC $H_2$ ), 45.5 (2 × NHC $H_2$ ), 60.3 (2 × NHC $H_3$ ).

(2S,6S)-2-Isopropyl-6-methyl-1,4,7-triazacyclononane 10c. Compound 10c was prepared in an analogous fashion to 10a using (2S,6S)-2-isopropyl-6-methyl-1,4,7-tris[(4-methylphenyl)-sulfonyl]-1,4,7-triazacyclonane 1c (559.7 mg, 0.865 mmol), EtOH (2.74 cm³) and lithium wire (242.7 mg, 34.97 mmol) to afford a light yellow oil (140.1 mg, 0.756 mmol, 87%); Found MH+ m/z 186.1968; C<sub>10</sub>H<sub>24</sub>N<sub>3</sub> requires m/z 186.1970;  $v_{max}$  (liq. film, cm $^{-1}$ ) 3630–3055 (bm, NH), 2953 (s, CH), 2921 (s, CH), 2870 (s, CH), 1466 (m), 1370 (m), 1160 (m), 1121 (m);  $\delta_{H}$  (400 MHz, CDCl<sub>3</sub>) 0.89 (d, J 6.7, 3H,  $-CH_3$ ), 0.95 (d, J 6.7, 3H,  $-CH_3$ ), 1.01 (d, J 6.6, 3H,  $-CH_3$ ), 1.54 (sep, J 6.7, 1H,  $CH(CH_3)_2$ ), 2.20 (bs, 3H, 3 × NH), 2.40 (m, 3H, 3 × CHN), 2.58 (m, 2H, 2 × CHN), 2.83 (m, 4H, 4 × CHN), 2.91 (m, 1H, CHN);  $\delta_{C}$  (100 MHz, CDCl<sub>3</sub>) 17.51 ( $-CH_3$ ), 19.36 ( $-CH_3$ ), 19.41 ( $-CH_3$ ), 31.94 ( $CH(CH_3)_2$ ), 40.46 ( $CH_2$ N), 44.85

(CH<sub>2</sub>N), 49.03 (CH<sub>2</sub>N), 49.55 (CH<sub>2</sub>N), 49.90 (CHN), 58.36 (CHN).

(2S,6S)-2-Isopropyl-6-methyl-1,4,7-triazacyclononane hydrobromide 10f. To a stirred solution of (2S,6S)-2-isopropyl-6-methyl-1,4,7-triazacyclonane **10c** (38.6 mg, 0.209 mmol) in ether (2 cm<sup>3</sup>) at 0 °C was added hydrobromic acid (48%, 0.1 cm<sup>3</sup>, 0.883 mmol) to give a tan coloured solid. The solid was washed with ether (×3, 3 cm<sup>3</sup>) followed by azeotropic removal of water by evaporation of added toluene (×4, 5 cm<sup>3</sup>). The residue was twice precipitated from EtOH (1 cm<sup>3</sup>) by the addition of ether (2 cm<sup>3</sup>) which gave a cream solid (58.9 mg, 0.138 mmol, 66%) mp 195–198 °C; Found C, 27.03; H, 5.85; N, 9.02%; C<sub>10</sub>H<sub>25</sub>N<sub>3</sub>· 3HBr·H<sub>2</sub>O requires C, 26.93; H, 6.33; N, 9.42%; [a]<sub>D</sub> +30.4  $(c = 0.565, \text{MeOH}); v_{\text{max}} (\text{KBr}, \text{cm}^{-1}) 3645 - 3200 (\text{sbr}, \text{NH}), 2961$ (s, CH), 2766 (s), 2651 (s);  $\delta_{\rm H}$  (400 MHz, D<sub>2</sub>O) 0.99 (d, J 6.8, 3H,  $-CH_3$ ), 1.08 (d, J 6.8, 3H,  $-CH_3$ ), 1.33 (d, J 6.68, 3H,  $-CH_3$ ), 1.97 (sep, J 6.8, 1H,  $CH(CH_3)_2$ ), 3.02 (m, 2H,  $2 \times CHNH$ ), 3.23–3.71 (m, 7H, 7 × CHNH), 3.91 (m, 1H, CHNH);  $\delta_{\rm C}$  (100 MHz, D<sub>2</sub>O) 14.68 (-CH<sub>3</sub>), 18.31 (-CH<sub>3</sub>), 19.11 (-CH<sub>3</sub>), 30.87 (CH(CH<sub>3</sub>)<sub>2</sub>), 38.29 (CH<sub>2</sub>N), 42.75 (CH<sub>2</sub>N), 46.41 (CH<sub>2</sub>N), 46.86 (CH<sub>2</sub>N), 51.8 (CHN), 58.65 (CHN).

(2S,6S)-1,2,4,6,7-Pentamethyl-1,4,7-triazacyclononane 11a. To a solution of (2S,6S)-2,6-dimethyl-1,4,7-tris[(4-methylphenyl)sulfonyl]-1,4,7-triazacyclononane 1a (912 mg, 1.5 mmol) in THF (25 cm<sup>3</sup>) and EtOH (4.5 cm<sup>3</sup>, 80 mmol) was condensed dry NH<sub>3</sub> (200 cm<sup>3</sup>) at -78 °C. To this solution was added lithium metal (512 mg, 74 mmol) in small portions to give an intense blue colour. The reaction mixture was allowed to warm to room temperature overnight. Water was added (5 cm<sup>3</sup>) and the solution was acidified (pH 1) with conc. HCl (1 cm<sup>3</sup>). The aqueous solution was extracted with dichloromethane ( $\times 2$ , 10 cm<sup>3</sup>). The aqueous phase was made basic (pH 14) by addition of solid NaOH (~500 mg). The basic solution was extracted with dichloromethane (×4, 10 cm<sup>3</sup>) and ethyl acetate (×2, 10 cm<sup>3</sup>). The resulting aqueous solution was acidified (pH 1) with conc. HCl (1 cm<sup>3</sup>) and the volatiles were removed under reduced pressure (2 mmHg). This residue was dissolved in H<sub>2</sub>O (1 cm<sup>3</sup>) and the solution was neutralised with solid NaOH (168 mg). To this solution was added formaldehyde (37%, 1.1 cm<sup>3</sup>, 13.5 mmol) and formic acid (90%, 1.7 cm<sup>3</sup>, 30 mmol). The solution was heated to reflux (bath temp. 90 °C) under a nitrogen atmosphere for 20 h. After cooling to room temperature the reaction was acidified (pH 1) with conc. HCl (1 cm<sup>3</sup>) and the volatiles were removed under reduced pressure. The aqueous solution was extracted with dichloromethane ( $\times$ 2, 10 cm<sup>3</sup>). The aqueous phase was made basic (pH 14) by addition of solid NaOH (~500 mg). The basic solution was extracted with dichloromethane (×4, 10 cm<sup>3</sup>), and the combined organic phases were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and evaporated to give light brown oil (131 mg, 0.66 mmol, 44%); Found MH<sup>+</sup> m/z 200.2121; C<sub>9</sub>H<sub>21</sub>N<sub>3</sub> requires m/z 200.2127;  $[a]_D$  +42.3  $(c = 1, \text{CHCl}_3); \nu_{\text{max}} (\text{CCl}_4, \text{cm}^{-1}) 2966 \text{ (m, CH)}, 2777 \text{ (m, CH)};$  $\delta_{\rm H}$  (400 MHz, CDCl<sub>3</sub>) 0.74 (d, J 6.1, 6H, -CH<sub>3</sub>), 2.05 (d, J 10.1, 2H, NCH<sub>2</sub>), 2.18 (m, 2H, NCH<sub>2</sub>), 2.31 (s, 9H, NCH<sub>3</sub>), 2.68 (d, J 11.3, 2H, NCH<sub>2</sub>), 2.92 (m, 2H, NCH<sub>2</sub>), 2.96 (m, 2H, NCH);  $\delta_{\rm C}$  (100 MHz, D<sub>2</sub>O) 11.6 (2 × -CH<sub>3</sub>), 41.3 (NCH<sub>3</sub>), 45.9  $(NCH_3)$ , 51.3  $(NCH_3)$ , 53.4  $(2 \times NCH_2)$ , 54.9  $(2 \times NCH_2)$ , 60.4  $(2 \times NCH)$ .

(2S,6S)-1,2,4,6,7-Pentamethyl-1,4,7-triazacyclononane tri-hydrochloride 11d. To a solution of (2S,6S)-1,2,4,6,7-pentamethyl-1,4,7-triazacyclononane 11a (131 mg, 0.66 mmol) in EtOH (2 cm³) was added conc. HCl (0.19 cm³) at room temperature with rapid stirring. To this solution was added ether (10 cm³) and the white precipitate that formed was removed by filtration and dried under reduced pressure to give the hydrochloride salt (202 mg, 0.65 mmol, 98%); mp 181–183 °C; Found

 $MH^{+}$  m/z 200.2127;  $C_{12}H_{27}N_{3}$  requires m/z 200.2126;  $[a]_{D}$  +36.1  $(c = 1, MeOH); v_{max} (KBr, cm^{-1}) 3400 (s, NH, br), 2956 (m,$ CH), 2814 (m, CH);  $\delta_{\rm H}$  (400 MHz, D<sub>2</sub>O) 1.25 (d, J 6.1, 6H,  $-CH_3$ ), 2.71 (m, 3H, NC $H_3$ ), 2.80 (m, 3H, NC $H_3$ ), 2.94 (m, 3H,  $NCH_3$ ), 3.04 (m, 6H,  $NCH_2$ ), 3.45 (m, 2H,  $NCH_2$ ), 3.76 (m, 2H, NCH),  $\delta_{\rm C}$  (100 MHz, D<sub>2</sub>O) 9.4 (-CH<sub>3</sub>), 11.1 (-CH<sub>3</sub>), 37.1 (NCH<sub>3</sub>), 41.8 (NCH<sub>3</sub>), 44.6 (NCH<sub>3</sub>), 45.5 (NCH<sub>2</sub>), 51.6  $(NCH_2)$ , 53.0  $(NCH_2)$ , 55.8  $(NCH_2)$ , 58.9  $(2 \times NCH)$ .

(2S,6S)-2,6-Diisopropyl-1,4,7-trimethyl-1,4,7-triazacyclo-

**nonane 11b.** A stirred solution of (2S,6S)-2,6-diisopropyl-1,4,7triazacyclononane 10b (91 mg, 0.42 mmol) and formaldehyde (37%, 0.3 cm<sup>3</sup>, 3.9 mmol) and formic acid (90%, 0.4 cm<sup>3</sup>, 9.2 mmol) was heated to reflux (bath temp. 90 °C) under a nitrogen atmosphere for 20 h. After cooling to room temperature the reaction was acidified (pH 1) with conc. HCl (1 cm<sup>3</sup>) and the volatiles were removed under reduced pressure (12 mmHg). The aqueous solution was extracted with dichloromethane (×2, 10 cm<sup>3</sup>). The aqueous phase was made basic (pH 14) by addition of solid NaOH (~500 mg). The basic solution was extracted with dichloromethane (×4, 10 cm<sup>3</sup>), and the combined organic phases were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and evaporated to give light brown oil (73 mg, 0.28 mmol, 66%); Found MH<sup>+</sup> m/z 256.2752; C<sub>15</sub>H<sub>33</sub>N<sub>3</sub> requires m/z 256.2753;  $[a]_D$  +38.4  $(c = 0.5, \text{CHCl}_3); \nu_{\text{max}} (\text{CCl}_4, \text{cm}^{-1}) 2955 \text{ (m, CH)}, 2870 \text{ (m, CH)};$  $\delta_{\text{H}} (400 \text{ MHz}, \text{CDCl}_3) 0.86 \text{ (d, } J 2.3, 6H, -CH_3), 0.88 \text{ (d, } J 2.3,$ 6H, -CH<sub>3</sub>), 1.64 (m, 2H, CH(CH<sub>3</sub>)<sub>2</sub>), 2.19 (m, 3H, NCH<sub>3</sub>), 2.33 (m, 3H, NCH<sub>3</sub>), 2.35 (m, 3H, NCH<sub>3</sub>), 2.40 (m, 4H, NCH<sub>3</sub>), 2.49 (s, 2H, NCH<sub>3</sub>), 2.64 (m, 2H, NCH<sub>2</sub>), 2.82 (m, 2H, NCH);  $\delta_{\rm C}$  (100 MHz, CDCl<sub>3</sub>) 20.9 (2 × -CH<sub>3</sub>), 21.2 (2 × -CH<sub>3</sub>), 30.7  $(2 \times CH(CH_3)_2)$ , 40.0  $(2 \times NCH_3)$ , 45.2  $(NCH_3)$ , 46.1  $(NCH_2)$ , 46.2 (NCH<sub>2</sub>), 52.0 (NCH<sub>2</sub>), 55.7 (NCH<sub>2</sub>), 67.2 (2 × NCH).

(2S,6S)-2-Isopropyl-1,4,6,7-tetramethyl-1,4,7-triazacyclo**nonane 11c.** A stirred solution of (2S,6S)-2-isopropyl-6-methyl-1,4,7-triazacyclononane **10c** (133.1 mg, 0.718 mmol) and formaldehyde (37%, 0.5 cm<sup>3</sup>, 6.67 mmol) and formic acid (90%, 0.67 cm<sup>3</sup>, 15.59 mmol) was heated to reflux under a nitrogen atmosphere for 20 h. After cooling to 0 °C, the reaction was acidified (pH 1) via the addition of 1 M aqueous HCl (5 cm<sup>3</sup>) and concentrated in vacuo (1 mmHg). The residue was dissolved in water (15 cm<sup>3</sup>) and washed with ether (×3, 15 cm<sup>3</sup>) and then basified (pH 14) by the addition of solid NaOH. The basic aqueous solution was extracted with dichloromethane (×4, 15 cm<sup>3</sup>) and the combined organic extracts were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and evaporated to afford a yellow oil (147.1 mg, 0.647 mmol, 90%); Found MH $^+$  m/z 228.2441;  $C_{13}H_{29}N_3$  requires 228.2439;  $v_{\text{max}}$  (liq. film, cm<sup>-1</sup>) 2954 (s), 2928 (s), 2870 (s, all CH), 2835 (s), 2790 (s), 2765 (s), 1450 (m), 1363 (m);  $\delta_{\rm H}$  (400 MHz, CDCl<sub>3</sub>) 0.74 (d, J 6.4, 3H, -CH<sub>3</sub>), 0.79 (d, J 6.7, 6H,  $-CH_3$ ), 1.57 (m, 1H,  $CH(CH_3)_2$ ), 2.09 (m, 2H, CHN), 2.25 (m, 2H, CHN), 2.28 (s, 6H, NCH<sub>3</sub>), 2.41 (s, 3H, NCH<sub>3</sub>), 2.46 (m, 1H, CHN), 2.57 (m, 1H, CHN), 2.78 (m, 1H, CHN), 2.85 (m, 1H, CHN), 2.88 (m, 1H, CHN);  $\delta_{\rm C}$  (100 MHz, CDCl<sub>3</sub>)  $12.51 (-CH_3), 20.96 (-CH_3), 21.25 (-CH_3), 30.68 (CH(CH_3)_2),$ 41.95 (-CH<sub>3</sub>N), 45.67 (-CH<sub>3</sub>N), 50.24 (-CH<sub>3</sub>N), 54.39  $(-CH_2N)$ , 56.30  $(-CH_2N)$ , 56.5 (-CHN), 60.6  $(-CH_2N)$ , 65.76

(2S,6S)-2-Isopropyl-1,4,6,7-tetramethyl-1,4,7-triazacyclo**nonane dihydrobromide 11d.** To a stirred solution of (2S,6S)-2isopropyl-1,4,6,7-tetramethyl-1,4,7-triazacyclononane 11c (74 mg, 0.325 mmol) dissolved in ether (3 cm<sup>3</sup>) at 0 °C was added hydrobromic acid (48%, 0.165 cm<sup>3</sup>, 1.45 mmol) to afford a brown oil. The volatiles were removed and the residue was dissolved in ethanol (1 cm<sup>3</sup>) and ether (3 cm<sup>3</sup>) was added to afford a cream solid which was recrystallised twice from ethanol to give a colourless solid (77 mg, 0.199 mmol, 61%); mp 250-253 °C; Found C, 39.86; H, 8.26; N, 10.88%; C<sub>13</sub>H<sub>28</sub>N<sub>3</sub>·2HBr requires C, 40.12; H, 8.03; N, 10.8%;  $[a]_D$  +34.6 (c = 0.52,

MeOH);  $v_{\text{max}}$  (CCl<sub>4</sub>, cm<sup>-1</sup>) 3625–3170 (bm, NH<sup>+</sup>), 3060 (s), 2995 (s, CH), 2980 (s, CH), 2910 (s, CH), 2900-2710 (bs), 1470 (s), 1383 (s), 1341 (s), 1050 (s), 1044 (s), 974 (s);  $\delta_{\rm H}$  (400 MHz,  $D_2O$ ) 1.0 (d, J 5.8, 3H,  $-CH_3$ ), 1.18 (d, J 6.0, 3H,  $-CH_3$ ), 1.42 (d, J 6.0, 3H,  $-CH_3$ ), 2.17 (m, 1H,  $CH(CH_3)_2$ ), 2.70 (s, 3H,  $NCH_3$ ), 2.92 (m, 1H, NCH), 3.06 (s, 3H, NCH<sub>3</sub>), 3.16 (s, 3H, NCH<sub>3</sub>), 3.25 (m, 3H, NCH), 3.48 (m, 3H, NCH), 3.74 (m, 1H, NCH), 3.86 (m, 1H, NCH), 4.08 (m, 1H, NCH);  $\delta_{\rm C}$  (100 MHz, D<sub>2</sub>O) 9.60 (-CH<sub>3</sub>), 19.25 (-CH<sub>3</sub>), 21.45 (-CH<sub>3</sub>), 28.21 (CH(CH<sub>3</sub>)<sub>2</sub>), 36.30 (-CH<sub>3</sub>N), 42.25 (-CH<sub>3</sub>N), 44.55 (-CH<sub>3</sub>N), 47.68  $(-CH_2N)$ , 52.02  $(-CH_2N)$ , 55.58  $(-CH_2N)$ , 57.35 (-CHN), 57.80 (-CHN), 58.37 (-CH<sub>2</sub>N), 59.57 (-CHN).

#### Crystal data for 11e

Measurements were made on a Nonius Kappa CCD diffractometer at 150 K: C<sub>13</sub>H<sub>31</sub>Br<sub>2</sub>N<sub>3</sub>, orthorhombic, space group  $P2_12_12_1$ , a = 7.5110(2), b = 12.0412(5), c = 19.5980(7) Å,  $V = 1772.47(11) \text{ Å}^3$ , Z = 4, MoK $\alpha$  radiation,  $\lambda = 0.71073 \text{ Å}$ ,  $\mu = 4.565 \text{ mm}^{-1}$ . Data were corrected for absorption with a multi-scan method. Final refinement to convergence on  $F^2$  and with 177 parameters gave R = 0.0349 (3241 observed data),  $R_{\rm w} = 0.0685$  (all 3844 unique reflections) and GOF = 1.039, Flack parameter -0.030(11).

CCDC reference number 219446.

See http://www.rsc.org/suppdata/ob/b3/b310492c/ for crystallographic data in CIF or other electronic format.

# Acknowledgements

We thank the EPSRC for funding through a ROPA award (to G. A., grant GR/M52477) and studentship (to G. S.). We also thank Stuart Cook and Dr Terry Lowdon (Hickson & Welch) for valuable discussions. We are indebted to the EPSRC National Mass Spectrometry Service, Swansea for running some mass spectra, and to the EPSRC National Crystallography Service, University of Southampton, for data collection on 11e.

### References

- 1 For reviews of biomimetic manganese catalysts see: R. Hage, Recl. Trav. Chim. Pays-Bas., 1996, 115, 385-395; K. Wieghardt, Angew. Chem., Int. Ed. Engl., 1989, 28, 1153-1172.
- 2 U. Bossek, M. Saher and T. Weyhermüller and K. Wieghardt, J. Chem. Soc., Chem. Commun., 1992, 1780-1782; K. Wieghardt, U. Bossek, B. Nuber, J. Weiss, J. Bonvoisin, M. Corbella, S. E. Vitols
- and J. J. Girerd, J. Am. Chem. Soc., 1988, 110, 7398–7411.

  3 U. Bossek, T. Weyhermüller, K. Wieghardt, B. Nuber and J. Weiss, J. Am. Chem. Soc., 1990, 112, 6387-6388; K. Wieghardt, U. Bossek, Bonvoisin, P. Beauvillain, J. J. Girerd, B. Nuber, J. Weiss and Heinze, Angew. Chem., Int. Ed. Engl., 1986, 25, 1030-1031; K. Wieghardt, U. Bossek and W. Gebert, *Angew. Chem., Int. Ed. Engl.*, 1983, **22**, 328–329.
- 4 W. B. Tolman, Acc. Chem. Res., 1997, 97, 227-237; K. Wieghardt, U. Bossek, D. Ventur and J. Weiss, J. Chem. Soc., Chem. Commun., 1985, 347-349.
- 5 K. P McCue and J. R. Morrow, Inorg. Chem., 1999, 38, 6136-6142; E. L. Hegg, K. A. Deal, L. Kiessling and J. Burstyn, Inorg. Chem., 1997, 36, 1715-1718.
- 6 C. Sissi, P. Rossi, F. Felluga, F. Fromaggio, M. Palumbo, P. Tecilla, C. Tonniolo and P. Scrimin, J. Am. Chem. Soc., 2001, 123, 3169–3170; A. Ehmann, D. Chafin, K.-M. Lee and J. J. Hayes, Nucleic Acid Res., 1998, 26, 2086-2091; E. L. Hegg and J. Burstyn, Inorg. Chem., 1996, 35, 7474-7481.
- 7 E. L. Hegg and J. Burstyn, J. Am. Chem. Soc., 1995, 117, 7015-7016. 8 J. Brinksma, R. La Crois, B. L. Feringa, M. I. Donnoli and C. Rosini, *Tetrahedron Lett.*, 2001, **42**, 4049–4052; D. H. R. Barton,
- W. Li and J. Smith, *Tetrahedron Lett.*, 1998, **39**, 7055–7058. 9 W.-H. Cheung, W.-Y. Yu, W.-P. Yip, N.-Y. Zhu and C.-M. Che, J. Org. Chem., 2002, **67**, 7716–7723
- H. Benhur, S. Sabne, S. S. Deshpande, D. Srinivas and Sivasanker, J. Mol. Catal. A: Chem., 2002, 185, 71-80; J. R. Lindsay Smith and G. B. Shul'pin, Tetrahedron Lett., 1998, 39, 4909-4912.

(-CHN).

- 11 R. Hage, J. E. Iburg, J. Kerschner, J. H. Koek, E. L. M. Lempers, R. J. Martens, U. S. Racherla, S. W. Russell, T. Swathoff, H. R. P van Vliet, J. B. Warnaar, L. van der Wolf and B. Krijnen, *Nature*, 1994, 369, 637–638; S. H. Jureller, J. L. Kerschner and R. Humphreys, *US Patent*, US 5,329,024, 1994 ( *Chem. Abstr.*, 1995, 122, 9845z)); D. De Vos and T. Bein, *Chem. Commun.*, 1996, 917–918; D. E. De Vos, J. L. Meinershagen and T. Bein, *Angew. Chem., Int. Ed. Engl.*, 1996, 35, 2211–2213; D. E. De Vos and T. Bein, *J. Organomet. Chem.*, 1996, 520, 195–200; A. Grenz, S. Ceccarelli and C. Bolm, *Chem. Commun.*, 2001, 1726–1727.
- 12 D. E. De Vos, B. F. Sels, M. Reynaers, Y. V. Subba Rao and P. A. Jacobs, *Tetrahedron Lett.*, 1998, 39, 3221–3224; A. Berkesssel and C. A. Sklorz, *Tetrahedron Lett.*, 1999, 40, 7965–7968.
- 13 M. Beller, A. Tafesh, R. W. Fischer and B. Scharbert, German Patent, Ger. DE 195 523 890, 1996 ( (Chem. Abstr., 1997, 126, 74725g)).
- 14 C. Bolm, D. Kadereit and M. Valacchi, Synlett., 1997, 687–688.
- 15 C. Bolm, N. Meyer, G. Raabe, T. Weyhermüller and E. Bothe, *Chem. Commun.*, 2000, 2435–2436.
- 16 G. Argouarch, C. L. Gibson, G. Stones and D. C. Sherrington, Tetrahedron Lett., 2002, 43, 3795–3798.
- 17 (a) S. F. Mason and R. D. Peacock, *Inorg. Chim. Acta*, 1976, 19, 75–77; (b) A. F. Drake, R. Kuroda and S. F. Mason, *J. Chem. Soc., Dalton Trans.*, 1979, 1095–1100; (c) M. Nonoyama, *Transition Met. Chem.*, 1980, 5, 269–271; (d) P. G. Graham and D. C. Weatherburn, *Aust. J. Chem.*, 1983, 36, 2349–2354; (e) J. P. L. Cox, A. S. Craig, I. M. Helps, K. J. Jankowski, D. Parker, M. A. W. Eaton, A. T. Millican, K. Millar, N. R. A. Beeley and B. A. Boyce, *J. Chem. Soc., Perkin Trans. 1*, 1990, 2567–2576; (f) J. H. Koek, E. W. M. J. Kohlen, S. W. Russell, L. van der Wolf, P. F. ter Steeg and J. C. Hellemons, *Inorg. Chim. Acta*, 1999, 295, 189–199; (g) L. J. Charbonnière, N. Wiebel and R. F. Ziessel, *J. Org. Chem.*, 2002, 67, 3933–3936.
- M. Beller, A. Tafesh, R. W. Fischer and B. Scharbert, German Patent, Ger. DE 195 23 891, 1995 ( (Chem. Abstr., 1997, 126, 69346e));
   S. W. Golding, T. W. Hambley, G. A Lawrence,

- S. M. Luther, M. Maeder and P. Turner, *J. Chem. Soc., Dalton Trans.*, 1999, 1875–1980; S. Pulacchini, K. F. Sibbons, K. Shastri, M. Motevalli, M. Watkinson, H. Wan, A. Whiting and A. Lightfoot, *J. Chem. Soc., Dalton Trans.*, 2003, 2043–2052.
- 19 B. M. Kim, S. M. So and H. J. Choi, Org. Lett., 2002, 4, 949-952.
- 20 E. W. Scheuermann, F. Ronketti, M. Motevalli, D. V. Griffiths and M. Watkinson, New J. Chem., 2002, 26, 1054–1059.
- 21 G. W. Daub, D. A. Heerding and L. A. Overman, *Tetrahedron*, 1988, 44, 3919–3930.
- 22 (a) R. Alul, M. B. Cleaver and J.-S. Taylor, *Inorg. Chem.*, 1992, 31, 3636–3636; (b) S. Itsuno, K. Tanaka and K. Ito, *Chem. Lett.*, 1986, 1113–1136.
- 23 M. Cernerud, H. Adolfsson and C. Moberg, *Tetrahedron: Asymmetry*, 1997, **8**, 2655–2662.
- 24 M. Cernerud, I. Skrinning, I. Bérgère and C. Moberg, *Tetrahedron: Asymmetry*, 1997, 8, 3437–3441.
- 25 J. E. W. Scheuermann, G. Ilashenko, D. V. Griffiths and M. Watkinson, *Tetrahedron: Asymmetry*, 2002, 13, 269–272.
- 26 F. Lake and C. Moberg, Eur. J. Org. Chem., 2002, 3179-3188.
- 27 J.-Q. Wang, M. Zhong and G.-Q. Lin, Chin. J. Chem., 1998, 16, 65–77.
- 28 T. J. Atkins, J. E. Richman and W. F. Oettle, Org. Synth., Coll. Vol. III, 1955, 652–662; J. E. Richman and T. J. Atkins, J. Am. Chem. Soc., 1974, 96, 2268–2270.
- 29 D. Parker, Macrocyclic Synthesis A Practical Approach, Oxford University Press, Oxford, 1996, ch. 1.
- 30 G. Stones, G. Argouarch, A. R. Kennedy, D. C. Sherrington and C. L. Gibson, *Org. Biomol. Chem.*, 2003, 1, 2357–2363.
- 31 C. F. G. C. Geraldes, A. D. Sherry, M. P. M. Marques, M. C. Alpoim and S. Cortes, J. Chem. Soc., Perkin Trans. 2, 1991, 137–146.
- 32 W. Eschweiler, *Chem. Ber.*, 1905, **38**, 880–887; H. T. Clarke, H. B. Gillispie and S. Z. Weishaus, *J. Am. Chem. Soc.*, 1933, **55**, 4571–4587; S. H. Pine and B. L. Sanchez, *J. Org. Chem.*, 1971, **36**, 829–833.
- 33 W. C. Still, M. Kahn and A. Mitra, J. Org. Chem., 1978, 43, 2923–2925.