

# Enantioselective Iridium-Catalyzed Hydrogenation of 3,4-Disubstituted Isoquinolines\*\*

Lei Shi, Zhi-Shi Ye, Liang-Liang Cao, Ran-Ning Guo, Yue Hu, and Yong-Gui Zhou\*

The past decade has witnessed rapid progress in the field of asymmetric hydrogenation of aromatic compounds, a transformation, which is regarded as one of the most straightforward means for accessing enantiopure cyclic compounds.<sup>[1]</sup> Extensive research has significantly expanded the substrate scope of this reaction, and substrates such as quinolines,<sup>[2]</sup> quinoxalines,<sup>[3]</sup> indoles,<sup>[4]</sup> furans,<sup>[5]</sup> pyrroles,<sup>[6]</sup> pyridines,<sup>[7]</sup> imidazoles,<sup>[8]</sup> and aromatic carbocycles can now be transformed through asymmetric hydrogenation.<sup>[9]</sup> Despite achievements made, the asymmetric hydrogenation of isoquinoline still remains an important unmet challenge. Hydrogenation reactions involving this substrate have been plagued by catalyst deactivation owing to the strong coordinating ability of the substrate and the product. So far, only one example of an enantioselective hydrogenation of isoquinoline has been reported by our research group.<sup>[10]</sup> *N*-protected 1-substituted 1,2-dihydroisoquinolines were obtained in moderate yield and enantioselectivity in the presence of stoichiometric amounts of chloroformate as the substrate activator (Scheme 1). However, several obvious limitations remain, such as the need for a stoichiometric amount of activating reagent and inorganic base, and that current methods only lead to products containing one stereogenic center, which is usually the C1 position. Given the prevalence of the chiral 1,2,3,4-tetrahydroisoquinoline motif in natural alkaloids and

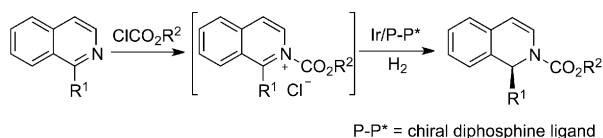
pharmaceutical molecules,<sup>[11]</sup> the development of an efficient method for the direct hydrogenation of isoquinolines is highly desirable. Herein, we describe a highly efficient direct enantioselective iridium-catalyzed hydrogenation of 3,4-disubstituted isoquinolines.

Recent results from our research group<sup>[2a]</sup> and that of others<sup>[7b,12]</sup> have demonstrated that iodine can significantly improve the performance of an iridium catalyst in asymmetric hydrogenation. We wanted to investigate whether isoquinoline could be amenable to asymmetric hydrogenation catalyzed by an iodine-activated iridium complex. Initially, ethyl 3-methylisoquinoline-4-carboxylate **1a** was chosen as model substrate. Upon exposure to 500 psi H<sub>2</sub> in the presence of a chiral iridium complex, which is generated in situ from [Ir(cod)Cl]<sub>2</sub>/(*R*)-synphos and iodine at 50°C, isoquinoline **1a** underwent enantioselective hydrogenation to afford product **2a** with full conversion, excellent diastereoselectivity (d.r. > 20:1) and moderate enantioselectivity (59% *ee*; Table 1, entry 2); when iodine was omitted, only the 1,2-hydrogenation product was observed (Table 1, entry 1).

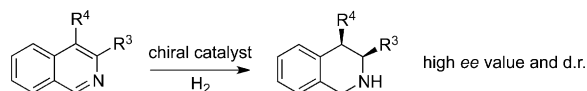
Encouraged by this promising result, we initially investigated the effect of the identity of the solvent on the substrate conversion and enantioselectivity. The substrate conversion was, in most solvents, uniformly good, whereas the *ee* value of **2a** exhibited a dramatic dependence upon the solvent identity (Table 1, entries 2–5). The use of toluene as the solvent was the most beneficial in terms of the enantioselectivity of the hydrogenation (80% *ee*, Table 1, entry 6). Next, the effect of the nature of the additive was investigated using various halogen sources (Table 1, entries 6–10). Each additive promoted this transformation, thus leading to full conversion of substrate and similar enantioselectivity. Among these additives, the use of 1-bromo-3-chloro-5,5-dimethyl-hydantoin (BCDMH) led to the isolation of product with slightly superior *ee* value (83% *ee*; Table 1, entry 10). The effect of the nature of the ligand on the reaction was then investigated by employing BCDMH as the halogen source in combination with iridium catalysts that were generated from [Ir(cod)Cl]<sub>2</sub> and a diverse array of commercially available ligands (Table 1, entries 10–13). Disappointingly, no ligand gave a better result than the ligand used in the initial screening of reaction conditions (**L1**).

Dynamic kinetic resolution (DKR), which is a powerful tool for accessing enantioenriched compounds, has been successfully applied in asymmetric hydrogenation.<sup>[13]</sup> In our previous research on asymmetric hydrogenation of 2,3-disubstituted quinolines and indoles, an interesting DKR phenomenon was also observed.<sup>[2f,4b]</sup> For the asymmetric hydrogenation of 3,4-disubstituted isoquinolines, a dynamic kinetic resolution process was involved (see below). In

previous work : substrate activation



this work: catalyst activation



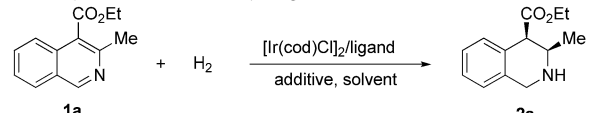
**Scheme 1.** Asymmetric hydrogenation of isoquinoline.

[\*] Dr. L. Shi, Z.-S. Ye, L.-L. Cao, R.-N. Guo, Y. Hu, Prof. Y.-G. Zhou  
State Key Laboratory of Catalysis  
Dalian Institute of Chemical Physics  
Chinese Academy of Sciences  
Dalian 116023 (China)  
E-mail: ygzhou@dicp.ac.cn

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**Table 1:** Optimization of the hydrogenation reaction.<sup>[a]</sup>



1a + H<sub>2</sub>  $\xrightarrow[\text{additive, solvent}]{[\text{Ir}(\text{cod})\text{Cl}]_2/\text{ligand}}$  2a

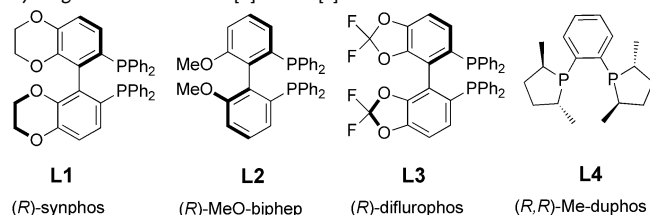
1a: CC(=O)Oc1ccc2c(c1)c(c[nH]2)C  
2a: CC(=O)O[C@@H]1Cc2ccccc2[C@H]1C

DBDMH, R<sup>1</sup> = Br, R<sup>2</sup> = Br  
BCDMH, R<sup>1</sup> = Br, R<sup>2</sup> = Cl

Entry	Ligand	H <sub>2</sub> [psi]	Solvent	Additive	ee [%] <sup>[b]</sup>
1 <sup>[c]</sup>	L1	500	THF	none	–
2	L1	500	THF	I <sub>2</sub>	59
3	L1	500	DCM	I <sub>2</sub>	52
4	L1	500	EtOAc	I <sub>2</sub>	67
5	L1	500	benzene	I <sub>2</sub>	70
6	L1	500	toluene	I <sub>2</sub>	80
7	L1	500	toluene	NBS	77
8	L1	500	toluene	NCS	77
9	L1	500	toluene	DBDMH	76
10	L1	500	toluene	BCDMH	83
11	L2	500	toluene	BCDMH	75
12	L3	500	toluene	BCDMH	32
13	L4	500	toluene	BCDMH	53
14	L1	100	toluene	BCDMH	87
15	L1	40	toluene	BCDMH	89
16 <sup>[d]</sup>	L1	40	toluene	BCDMH	91
17 <sup>[e]</sup>	L1	40	toluene	BCDMH	93

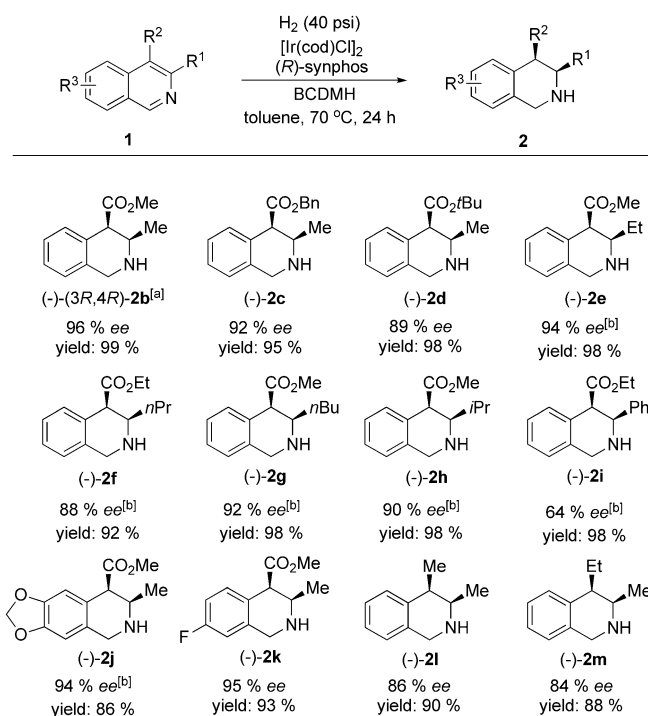
[a] Reaction conditions: 0.2 mmol of **1a**, 1.0 mol % [Ir(cod)Cl]<sub>2</sub>, 2.2 mol % ligand, 10 mol % additive, 3.0 mL of solvent, 50 °C. Reaction conversion and d.r. were determined by <sup>1</sup>H NMR spectroscopy. In all cases, the reaction conversion was > 95 % and the d.r. > 20:1.

[b] Determined by HPLC using a chiral stationary phase. [c] Only 1,2-hydrogenation occurred. [d] 60 °C. [e] 70 °C.



general, efficient DKR requires rapid interconversion of two enantiomeric intermediates. In asymmetric hydrogenations of the type described herein, such a condition can usually be ensured by conducting the reaction at high temperature and by using low hydrogen pressure. As expected, further improvement of enantioselectivity was realized when a lower hydrogen pressure was used (Table 1, entries 14 and 15). Simultaneously lowering the hydrogen pressure and raising the reaction temperature led to further increases in enantioselectivity and, ultimately, product was obtained in 93 % ee when the hydrogenation was carried out at 70 °C and at a hydrogen pressure of 40 psi (Table 1, entry 17).

As a demonstration of the practicality of this reaction, various 3,4-disubstituted tetrahydroisoquinolines were accessed with complete conversion from substrate and with high ee value (up to 96 % ee) when the optimized reaction conditions were used (Scheme 2). Notably, the size of the isoquinoline substituents influenced the enantioselectivity of reaction. The presence of a bulky ester at the C4 position or

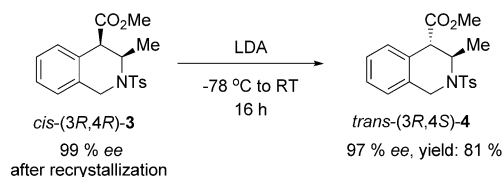


**Scheme 2.** Iridium-catalyzed enantioselective hydrogenation of 3,4-disubstituted isoquinolines. [a] The absolute configuration was determined by single-crystal X-ray diffraction analysis of the corresponding N-tosyl derivative. [b] The reaction was carried out at 80 °C. cod = 1,5-cyclooctadiene.

the presence of a longer alkyl group at the C3 position caused a small decrease in product ee value, presumably because the isomerization of the imine intermediate was relatively slow. For a similar reason, the 3-phenyl-substituted tetrahydroisoquinolines **2i** was obtained with only moderate 64 % ee; raising the reaction temperature from 70 °C to 80 °C can lead to satisfactory enantioselectivity (**2e–j** except for **2i**). Moreover, 3,4-dialkyl-substituted tetrahydroisoquinolines **2l** and **2m** were prepared with good ee value. This result suggests that the presence of an ester group is not necessary for achieving satisfactory results, thus further highlighting the generality of this method.

To demonstrate that the chiral *cis*-disubstituted products could be used for preparing the corresponding *trans* compounds, compound **3** was treated with lithium diisopropylamide (LDA) to give the *trans* epimer **4** (Scheme 3), which is generally difficult to obtain through direct asymmetric hydrogenation.

To explore the mechanism further, the hydrogenation reaction was carried out at room temperature and was then



**Scheme 3.** LDA-mediated conversion of *cis*-(3R,4R)-**3** into epimeric *trans*-(3R,4S)-**4**.



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- [14] CCDC 876983 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif).

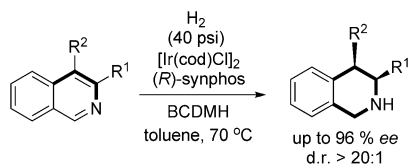
## Communications



### Asymmetric Hydrogenation

L. Shi, Z.-S. Ye, L.-L. Cao, R.-N. Guo,  
Y. Hu, Y.-G. Zhou\* ——— ■■■■—■■■■

Enantioselective Iridium-Catalyzed  
Hydrogenation of 3,4-Disubstituted  
Isoquinolines



**Reining in the outliers:** An efficient approach for enantioselective hydrogenation of 3,4-disubstituted isoquinolines was successfully developed. When isoquinolines are treated with [Ir(cod)Cl]<sub>2</sub>/ (R)-synphos in the presence of 1-bromo-3-chloro-5,5-dimethyl-hydantoin (BCDMH), the chiral 3,4-disubstituted tetrahydroisoquinoline derivatives are obtained with *ee* values as high as 96% (see scheme; cod = 1,5-cyclooctadiene).