HIV Protease Inhibitors Part 2: [3+2] Cycloaddition, Isomerization; and Ring Expansion en route to 4,5-Substituted Cyclohexenones

Emmanuel Demont,*1 Andrew Eatherton, Christopher S. Frampton, Irfan Kahn, Sally Redshaw

Roche Discovery Welwyn, Broadwater Road, Welwyn Garden City, Hertfordshire AL7 3AY, UK Fax 44(1438)782088; E-mail: emmanuel.h.demont@gsk.com *Received 12 January 2004*

Abstract: 4,5-Substituted cyclohexanone **10** and its derivatives are carbocyclic analogues of Indinavir **3** and are expected to have antiviral activity. Early attempts to obtain these compounds via a diastereoselective [3+2] cycloaddition between **19** and **14** failed due to the sensitivity of the cycloadduct **24**. It proved possible to obtain **30** from the α , β -unsaturated ester **27**: [3+2] cycloaddition, isomerization, and ring expansion provided α , β -unsaturated ketone **31** from ester **26** in good yields. Further transformations of **31** gave the hydroxyethylamino inhibitor analogues of Indinavir **3**.

Key words:: [3+2] cycloaddition, cyclopentene, ring expansion, cyclohexenone, HIV protease inhibitor

In the preceding paper² we have described a route to 4substituted carbocyclic Phe-Pro mimetics 1 using diastereoselective Diels–Alder reactions. We also hoped to develop a route to 5-substituted compounds 2 since molecular modelling, as well as marketed HIV protease inhibitors such as Indinavir 3, indicate that substituents at this position should be tolerated and may have important effects on potency and pharmacokinetic properties (Figure 1).



Figure 1

Attempts to prepare 5-functionalized carbocycles using Diels–Alder chemistry were not successful. The dienophile **4** (Scheme 1) did not react with dienes 5^3 or 6^4 under either thermal conditions or Lewis acid catalysis to give

SYNLETT 2004, No. 4, pp 0684–0687 Advanced online publication: 29.01.2004 DOI: 10.1055/s-2004-815439; Art ID: D24303ST © Georg Thieme Verlag Stuttgart · New York the α -functionalized ketone 7; instead only the starting materials were recovered (Figure 2).





It proved difficult to oxidize in situ the silyl enol ether 8^5 formed by reaction of 4 with 2-(trimethylsilyloxy)-1,3butadiene, whilst use of Danishefsky's diene under similar conditions lead only to decomposition. Attempts to effect conversion of 7 to 8 using chiral amide⁶ bases such as 9^7 were also unsuccessful (Scheme 1).





We decided then to modify our strategy and to access ketone **10** by ring expansion. Ozonolysis of cyclopentene **11** followed by crotonization of the intermediate keto-aldehyde should lead, after epimerization of the carbon α to the ester, to the more stable *trans*-isomer **10** (Scheme 2). It was hoped that **11** could be obtained from **12**, after isomerization of the double bond and reduction of the subsequent vinyl sulfone. **12** would result from reaction of allylic sulfone **14**⁸ with lactone **13**.

We chose as our starting material the epoxide 15,⁹ for which a large-scale synthesis was already known (Scheme 3). Attempts to react 15 with methyl propionate under Yamaguchi's conditions¹⁰ gave the acetylene 17 together with substantial amounts of the ketene 18^{11} (as a





Scheme 2 Retrosynthetic analysis.

2:1 mixture of isomers), from which the acetylene was difficult to separate. We decided then to synthesise **17** via functionalization of alkyne **16**. Reduction of **17** using Lindlar's catalyst followed by lactonization¹² gave **19**¹³ in good overall yield.

Lactone **19** reacts kinetically with the anions of allylic sulfones to give stereoselectively compounds such as **20**, whilst addition of nitromethane¹⁴ gives only the thermodynamically favoured *cis*-isomer **21** (Scheme 4). On the other hand, addition of azide¹⁵ affords a 1:1 mixture of isomers. Diels–Alder reaction of **19** and **7** was found to be stereoselective and gave, after acidic workup, bicyclic adduct **23**, presumably due to kinetic protonation of the concave enol ether ether **22**. Interestingly, this reaction can be considered, in addition to hydrogenation, as a way to access highly functionalized all-*cis* cyclohexanes, in which one of the alkyl chains is in the thermodynamically unfavored pseudo-axial position.

Reaction of **19** with sulfone **14** was also stereoselective and gave **24** as a single isomer whose stereochemistry was determined by X-ray crystallography (Figure 3).¹⁶

The 6-membered lactone adopts a highly strained boat conformation. Treatment of **24** with potassium *tert*-butoxide gave vinyl sulfone **25** in 93% yield. Reduction of sulfone **25** proved difficult and lead to decomposition, mainly because of the strain of the bicyclic structure. We showed, for example, that treatment of **24** with samarium iodide gave the corresponding lactol in 70% yield, releasing the strain of the boat conformation of the starting material.¹⁷ Unfortunately, all attempts to saponify lactones **24** or **25** gave prior or concomitant opening of the phthalimide group. Attempts to remove the phthalimide group in order to replace it by a more robust protecting group, even using very mild conditions,¹⁸ were unsuccessful.



Scheme 4 Conditions: a) allyl sulfone (1.2 equiv), LDA, THF, -78 °C. b)CH₃NO₂, DBU, r.t. c) 7 (3 equiv), Et₂AlCl (3 equiv), THF, r.t. d) 14 (1.2 equiv), LDA, THF, -78 °C. e) *t*-BuOK, *t*-BuOH–THF, 0 °C.

Synlett 2004, No. 4, 684-687 © Thieme Stuttgart · New York



Figure 3 ORTEP diagram of compound 24.

The length of the synthesis of 19^{19} together with the problem of the choice of a good protecting group²⁰ for the side chain nitrogen led us to move towards a shorter approach using 27 as a Michael acceptor (Scheme 5).

Treatment of **27** with the lithium salt of sulfone **14** gave, in high yield²¹ **28**, which was isomerized to **29** prior to reduction.²² Ozonolysis of **30** followed by aldolizationelimination provided the desired ketone **31** in good yield. Ketone **31** was then easily transformed to **32** by reduction of the double bond, saponification and coupling to *tert*butyl amine. The transformations previously described² on the isomeric 4-ketone were also applicable to **32**: for example, reduction of **32** using Luche's²³ conditions followed by methylation gave ether **33** in good yields and with high diastereoselectivity.

We have thus developed efficient syntheses of both structures of both 4- and 5-substituted carbocyclic mimetics of scissile Phe-Pro. The biological activities of HIV protease inhibitors derived from these intermediates will be reported in due course.

References

- (1) Present address: GlaxoSmithKline. The Frythe, Welwyn, AL6 9AR Hertfordshire, UK.
- (2) Crackett, P.; Demont, E.; Eatherton, A.; Frampton, C. S.; Gilbert, J.; Kahn, I.; Redshaw, S.; Watson, W. Synlett 2004, DOI: 10.1055/s-2004-81548.
- (3) Martin, V. A.; Murray, D. H.; Pratt, N. E.; Zhao, Y.; Albizati, K. F. J. Am. Chem. Soc. 1990, 112, 6965.
- (4) Nicolaou, K. C.; Postema Maarten, H. D.; Miller, N. D.; Yang, G. Angew. Chem. Int. Ed. 1997, 36, 2821.
- (5) (a) Adam, W.; Hadjiarapoglou, L.; Wang, X. *Tetrahedron Lett.* **1989**, *30*, 6497. (b) Stankovic, S.; Espenson, J. H. *J. Org. Chem.* **1998**, *63*, 4129.
- (6) (a) For an example of selective deprotonation of non-symmetrical ketones using a chiral amide base see: Sobukawa, M.; Nakajima, M.; Koga, K. *Tetrahedron: Asymmetry* 1990, *1*, 295. (b) For a recent review on the use of chiral amide bases see: O'Brien, P. J. Chem. Soc., Perkin Trans. 1 1998, 1439; and references therein.
- (7) Overberger, C. G.; Marullo, N. P.; Hiskey, R. G. J. Am. Chem. Soc. 1961, 83, 1374.
- (8) Ghera, E.; Yechezkel, T.; Hassner, A. J. Org. Chem. 1996, 61, 4959.
- (9) Parkes, K. E. B.; Bushnell, D. J.; Crackett, P. H.; Dunson, S. J.; Freeman, A. C.; Gunn, M. P.; Hopkins, R. A.; Lambert, R. W.; Martin, J. A.; Merrett, J. H.; Redshaw, S.; Spurden, W. C.; Thomas, G. J. *J. Org. Chem.* **1994**, *59*, 3656.
- (10) Yamagushi, M.; Hirao, I. *Tetrahedron Lett.* **1983**, 24, 391– 394.
- (11) Use of propiolate trialkylester could have avoided obtaining this by-product but the large scale synthesis of these compounds was considered difficult. See for example: Baldwin, J. E.; Pritchard, G. J.; Rathmell, R. E. Synth. Commun. 2000, 30, 3833.
- (12) For the procedures used to synthesize 17 see: (a) Brimble,
 M. A.; Edmonds, M. K.; Williams, G. M. *Tetrahedron* 1992,
 48, 6455. (b) Marshall, J. A.; Andrews, R. C.; Lebioda, L. J. Org. Chem. 1987, 52, 2378.
- (13) This type of compound can be obtained by olefin metathesis but the high dilution of these reaction is not compatible with the scale of reaction envisioned in this synthesis, see: Ghosh, A. K.; Cappiello, J.; Shin, D. *Tetrahedron Lett.* **1998**, *39*, 4651.



Scheme 5 Conditions: a) DIBAL-H, PhCH₃, -78 °C. b) Ph₃PCHCO₂CH₃, THF, r.t. c) 15 (1.2 equiv), LDA, THF, -78 °C. d) MeONa, MeOH, 0 °C. e) Na/Hg 5%, MeOH, KH₂PO₄, 0 °C. f) O₃, MeOH-CH₂Cl₂, -78 °C; PPh₃, -78 °C to r.t. g) PTSA, PhCH₃, Dean–Stark, reflux. h) H₂, Pd/C, EtOAc, RT. i) LiOH, THF–water, 0 °C. j) HOBT, EDAC·HCl, *t*-BuNH₂, CH₂Cl₂, 0 °C to r.t. k) NaBH₄, CeCl₃, MeOH, -10 °C to r.t. (ds> 90/10). j) NaH, DMF, 0 °C; MeI, 0 °C to r.t.

Synlett 2004, No. 4, 684-687 © Thieme Stuttgart · New York

Downloaded by: University of Illinois. Copyrighted material.

- (14) Plummer, J. S.; Emery, L. A.; Stier, M. A.; Suto, M. J. *Tetrahedron Lett.* **1993**, *34*, 7529.
- (15) Lakshmipathi, P.; Rama Rao, A. V. *Tetrahedron Lett.* **1997**, *38*, 2551.
- (16) CCDC No. 217035 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/conts/ retrieving.html (or from the CCDC 12 Union Road Cambridge CB2 1EZ UK; fax:+44 1223 336033; e-mail: deposit@ccdc.cam.ac.uk)
- (17) SmI₂ does not usually reduce ester. See Molander, G. A. Org. React. **1994**, 46, 211–367; and references cited therein.
- (18) Kukojla, S.; Lammert, S. R. J. Am. Chem. Soc. **1975**, 97, 5583.

- (19) An other approach was attempted using literature precedent, see: (a) Hanessian, S.; Haeil, P.; Rui-Yang, Y. *Synlett* 1997, 353. (b) Nicoll-Griffith, D. A.; Weiler, L. *Tetrahedron* 1991, 47, 2733.
- (20) N,N-Dibenzyl amine was considered to be a reasonable solution. For an excellent review on N,N-dibenzylamino compounds see: Reetz, M. T. Chem. Rev. 1999, 99, 1121.
- (21) The reaction of **14** with oxazolidinone **4** was non-selective. Adduct **28** was obtained as a 1:1 mixture of *trans*-isomers.
- (22) The direct reduction of allylic sulfone **28** gave a mixture of alkenes.
- (23) Gemal, A. L.; Luche, J.-L. J. Am. Chem. Soc. 1981, 103, 5454.