

Effect of Crosslinker Multiplicity on the Gel Point in ATRP

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ABSTRACT: The experimental gelation was studied in atom transfer radical polymerization (ATRP) of methyl acrylate (MA) with various branching reagents that structurally differ by the number of vinyl groups (multiplicity). MA was copolymerized with branching reagents containing 2, 3, 4, 5, or 6 acrylate moieties per molecule, respectively. Reactions with a constant concentration of branching vinyl groups (the same molar ratio of $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$) revealed a different gelation behavior when ethylene glycol diacrylate (2A) and trimethylolpropane triacrylate (3A) were used as crosslinkers, whereas the reactions using pentaerythritol tetraacrylate (4A) showed similar gelation behavior as compared to reactions using 3A. Additional reactions with dipentaerythritol pentaacrylate (5A) and hexafunctional acrylate crosslinkers (6A) revealed the presence of a concentration-dependant gelation behavior. On the other hand,

in reactions with the same molar concentration of various crosslinkers, gelation occurred at progressively lower MA conversions for reactions with an increasing number of vinyl groups per crosslinker. In addition, the number of unreacted pendant vinyl groups in the sols was compared for reactions with different multiplicity of the crosslinker and various ratios of $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$. Finally, a linear oligomeric crosslinker containing multiple branching vinyl groups along the side chain was used as the branching reagent for gelation. The gelation behavior during the ATRP of MA with the linear crosslinker was different as compared to the use of the aforementioned crosslinkers with a star-like architecture. © 2010 Wiley Periodicals, Inc. *J Polym Sci Part A: Polym Chem* 48: 2016–2023, 2010

KEYWORDS: ATRP; crosslinker; gelation; multiplicity

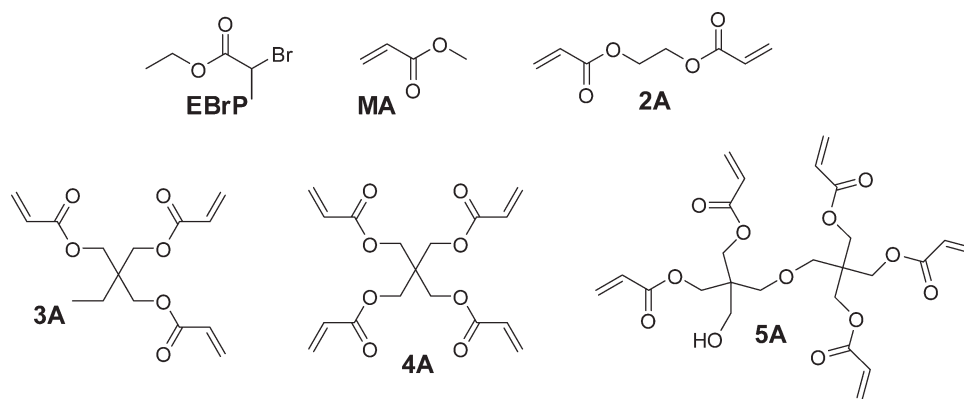
INTRODUCTION Conventional radical polymerization (RP) of a monovinyl monomer with a small amount of divinyl crosslinker is a widespread method for the synthesis of branched polymers or gels. However, the slow and continuous initiation process, fast chain propagation, and termination reactions that are inherent to conventional RP result in networks with an inhomogeneous and uncontrolled structure.^{1–4} Recently, several research groups have shown that the application of controlled radical polymerization (CRP) techniques^{5–9} produces gels with a more homogeneous structure, as a result of the fast initiation process and reversible deactivation of propagating polymer chains.^{10–13} The fast initiation reactions, relative to propagation, result in a quick conversion of nearly all initiators to growing primary chains. On the other hand, the dynamic equilibrium between the active species, bearing a radical at the polymer chain end, and the dormant species ensures a low steady concentration of radicals. During each activation/deactivation cycle, only a few monomers are added to the growing polymer chain before it is quickly deactivated to the dormant state. The dormant state allows for chain relaxation and diffusion of reagents. As a result, the probability of reaction of each vinyl species,

monomer, “unreacted” crosslinker or pendant vinyl group, is dependent on their relative concentration and reactivity. Therefore, the branched polymers or gels have a more homogeneous distribution of branching points than the polymers synthesized by RP methods at similar reaction conditions.¹¹

Recently, several research groups systematically studied the synthesis of branched polymers and/or gels by the copolymerization of monovinyl monomer and divinyl crosslinker using atom transfer radical polymerization (ATRP).^{10,14–25} Sherrington^{20,21} and Armes^{18,19,22} focused on the synthesis of branched polymers based on copolymerization of methacrylate and dimethacrylate crosslinker. Zhu and coworkers studied the homopolymerization²⁵ of ethylene glycol dimethacrylate and copolymerization^{23,24} of methyl methacrylate and dimethacrylate crosslinker using bulk ATRP. Our group applied the ATRP of methyl acrylate (MA) and ethylene glycol diacrylate as a model system to study the dependence of the experimental gel points on various parameters, because of its fast initiation process, high initiation efficiency, and low polydispersity in all performed experiments.¹⁴ The effect of several parameters was studied, for example, the molar ratio of divinyl crosslinker to initiator,¹⁴ the concentration of

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SCHEME 1 Synthesis of polymer networks by ATRP of MA and different branching reagents.

all reagents during the gelation process,¹⁵ the relative reactivity of divinyl crosslinker compared to monomer,¹⁶ and the initiation efficiency and polydispersity of the primary chains.¹⁷ Furthermore, the experimental results were compared with values obtained from Monte Carlo simulations.²⁶ Finally, an AB* inimer (initiator-monomer containing vinyl bond A and initiator fragment B* in one molecule) was also explored as a branching reagent for the synthesis of branched copolymers and gels by copolymerization with monomer and divinyl crosslinker using ATRP,²⁷ and the results were compared with gelation reactions using divinyl and trivinyl crosslinker.

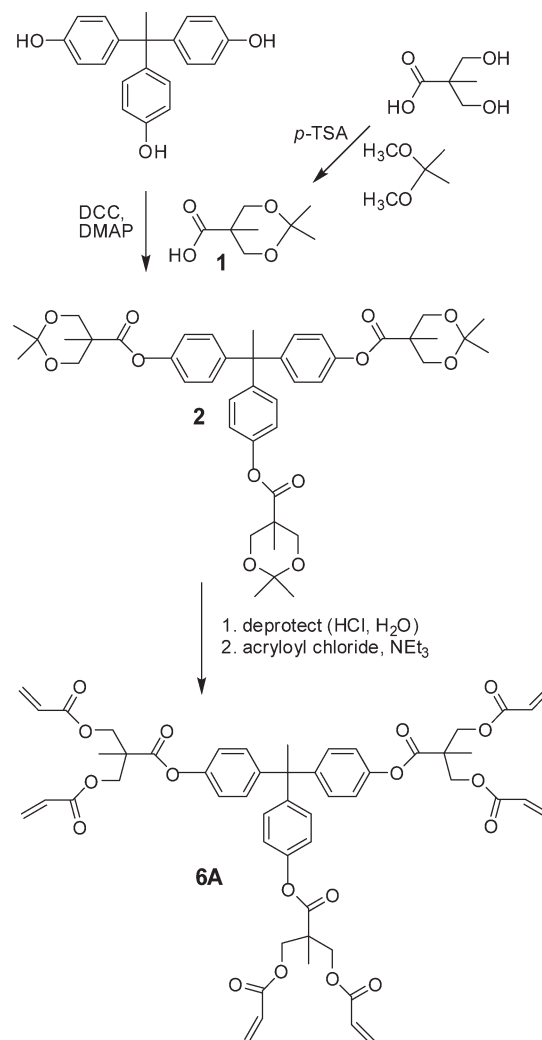
In this article, we aim to expand the systematic study on the gelation in ATRP and investigate the effect of crosslinker multiplicity, that is, the number of vinyl groups per crosslinker molecule, on the experimental gel points. Copolymerization of MA was carried out with various branching reagents: ethylene glycol diacrylate (2A), trimethylolpropane triacrylate (3A), pentaerythritol tetraacrylate (4A), dipentaerythritol pentaacrylate (5A), and a hexafunctional acrylate (6A) crosslinker (see Schemes 1 and 2). To keep the system comparable to those previously studied, the experimental gel points based on the monomer (MA) conversions were determined and systematically compared to each other for reactions where the molar ratios of branching vinyl groups to initiator ($[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$) and the crosslinker species were varied. The effect of crosslinker multiplicity was also investigated in a series of reactions where the molar concentration of the various crosslinkers was kept constant. In addition, the behavior of the aforementioned star-like crosslinkers (3A, 4A, 5A, 6A) was compared with a linear oligomeric multi-vinyl crosslinker molecule, containing an average of seven dangling vinyl groups along the polymer chain.

EXPERIMENTAL

Materials

Methyl acrylate (MA, 99%), 2-hydroxyethyl acrylate (HEA, 96%), *n*-butyl acrylate (*n*BA, 99%), and ethylene glycol diacrylate (2A, 90%) were purchased from Aldrich and purified twice by passing through a column filled with basic alumina to remove the inhibitor. The trivinyl crosslinker, trimethylolpropane triacrylate (3A, technical grade, Aldrich) and tetra-vinyl crosslinker, pentaerythritol tetraacrylate (4A, technical

grade, Aldrich) was purified via silica liquid chromatography with mixture of hexane/ethyl acetate 7/3 by volume as the eluent. Dipentaerythritol penta-/hexa-acrylate (5A, technical grade, Aldrich) was determined to have an average multiplicity of 5 by ¹H NMR, and was used as received. CuBr (98%, Acros) was purified using a modified literature procedure.²⁸



SCHEME 2 Synthesis of hexafunctional acrylate crosslinker (6A).

All other reagents: methyl 2-bromopropionate (MBP), ethyl 2-bromopropionate (EBrP), *N,N,N',N',N''*-pentamethyldiethylenetriamine (PMDETA), acryloyl chloride, *N,N'*-dicyclohexylcarbodiimide (DCC), 4-(dimethylamino)pyridine, *p*-toluenesulfonic acid monohydrate, 2,2-bis(hydroxymethyl)propionic acid, 2,2-dimethoxypropane, triethylamine, 1,1,1-tris(4-hydroxyphenyl)ethane, 4-Hydroxy-TEMPO, CuBr₂, and solvents were purchased from Aldrich with the highest purity and used as received without further purification.

Characterization

Monomer conversions were determined from the concentration of the unreacted monomer in the samples periodically removed from the reactions using a Shimadzu GC-17A gas chromatograph, equipped with a capillary column (DB-Wax, 30 m × 0.54 mm × 0.5 μm, J and W Scientific). DMF was used as internal standard for calculation of monomer conversions. After filtration through 220 nm PTFE filter, the polymer samples were separated by GPC [Polymer Standards Services (PSS) columns (guard, 10⁵, 10³, and 10² Å), with THF eluent at 35 °C, flow rate = 1.00 mL/min and differential refractive index (RI) detector (Waters, 2410)]. The apparent molecular weights and polydispersities (*M_w*/*M_n*) were determined with a calibration based on linear poly(methyl methacrylate) (polyMMA) standards using WinGPC 6.0 software from PSS. The GPC curves obtained with flat baseline were imported into the WinGPC software for calculation of their apparent molecular weights and polydispersity. ¹H NMR spectra, using CDCl₃ or acetone-*d*₆ as solvent, were measured on a Bruker Avance 300 MHz spectrometer at 27 °C.

Synthesis of Hexafunctional Acrylate (6A) Crosslinker

Synthesis of 2,2-Bis((2,2-propyl)dioxymethyl)propionic acid 1

2,2-Bis(hydroxymethyl)propionic acid (bisMPA) (10.0 g, 74.6 mmol), was reacted with 2,2-dimethoxypropane (13.8 mL, 112 mmol, 1.5 eq. to bisMPA), and *p*-TSA (0.71 g, 5 mol % to bisMPA) in 50 mL of acetone at room temperature. After 2 h, 1 mL of NH₄OH (aqueous, 30%)/EtOH (1/1) solution was added to neutralize the catalyst. The reaction mixture was concentrated by rotary evaporation, the residue was redissolved in CH₂Cl₂, and extracted twice with 20 mL of H₂O. The organic phase was dried over MgSO₄, filtered, and evaporated to yield 8.3 g (63%) of white crystals.

¹H NMR (CDCl₃): δ = 1.20 (s, 3H, —CH₃), 1.41–1.44 (d, 6H, —C(CH₃)₂), 3.65–4.20 (dd, 4H, —C(CH₂O—)₂).

Synthesis and Deprotection of 2

1,1,1-Tris(4-hydroxyphenyl)ethane (3.73 g, 12.2 mmol) and 1 (7.0 g, 40.2 mmol, 1.1 eq. to OH groups) was dissolved in 25 mL of dry CH₂Cl₂. *N,N'*-dicyclohexylcarbodiimide (DCC) (11.3 g, 1.5 eq. to OH groups) and 4-(dimethylamino)pyridine (1.13 g, 10 wt % to DCC) was added. The mixture was stirred overnight at room temperature. The formed urea was removed by filtration. The filtrate was concentrated and the resulting viscous oil was purified by silica gel column chromatography (eluens hexane/ethyl acetate 3/2). The pure fractions were concentrated, resulting in a viscous oil. After

purification, the product was dissolved in a mixture of 30 mL of THF and 30 mL of 1 M HCl (aq.). The reaction mixture was stirred for 2 h before the precipitated product was filtered and washed with water. The product was dried in vacuum at 50 °C, yielding a white solid. Yield: 3.83 g (90%).

¹H NMR (acetone-*d*₆): δ = 1.31 (s, 9H, —CH₃), 2.21 (s, 3H, —CH₃), 3.76–3.89 (m, 12H, —C(CH₂O—)₂), 7.06–7.17 (m, 12H, —OPh—).

Esterification with Acryloyl Chloride

2 (1.52 g, 2.32 mmol, 13.9 mmol OH groups), 25 mL dry dichloromethane, triethylamine (2.81 g, 3.88 mL, 27.9 mmol), and 10.0 mg of radical inhibitor 4-hydroxy-TEMPO were added to a clean, dry, round-bottom flask. The flask was placed in an ice-water bath and the mixture was cooled down to 0 °C. Acryloyl chloride (2.52 g, 2.27 mL, 27.8 mmol, 2 eq. to OH groups) was added dropwise to the reaction mixture over a period of 20 min. After 1 h, the flask was removed from the ice-water bath and the reaction mixture was stirred for 24 h at room temperature. During this period, the reaction mixture changed from a yellowish color to brown. The reaction mixture was washed successively with 100 mL of 1 M HCl, 100 mL of 1 M NaOH, and 100 mL of deionized water and then dried over anhydrous MgSO₄ for 1 h. Ten milligram of radical inhibitor 4-hydroxy-TEMPO was added to the mixture and the solvent was removed via rotary evaporation and oil pump vacuum. The final product was a viscous oil.

The structure was verified by ¹H NMR spectroscopy (CDCl₃): δ = 1.44 (s, 9H, —CH₃), 2.13 (s, 3H, —CH₃), 4.47 (s, 12H, —C(CH₂O—)₂), 5.85–5.89 (2d, 1H, —CH₂CHC(O)O), 6.08–6.18 (dd, 1H, —CH₂CHC(O)O—), 6.40–6.46 (2d, 1H, —CH₂CHC(O)O), 6.91–7.08 (m, 12H, —OPh—).

Synthesis of PolyMA-Based Gels by ATRP of MA with 2A, 3A, 4A, 5A, or 6A

A typical procedure for the ATRP of MA and 2A is briefly described, starting with a ratio of reagents [MA]₀/[2A]₀/[EBrP]₀/[CuBr]₀/[CuBr₂]₀/[PMDETA]₀ = 50/2/1/0.45/0.05/0.5, [MA]₀ = 6.0 M. A clean and dry Schlenk flask was charged with MA (3.50 mL, 0.039 mol), 2A (0.290 mL, 1.56 mmol), PMDETA (81.2 μL, 0.39 mmol), and *N,N*-dimethylformamide (DMF, 2.51 mL). The flask was deoxygenated by five freeze-pump-thaw cycles. During the final cycle, the flask was filled with nitrogen before CuBr (50.2 mg, 0.35 mmol) and CuBr₂ (8.7 mg, 0.039 mmol) were quickly added to the frozen mixture. No precautions were taken to avoid moisture condensation. The flask was sealed with a glass stopper then evacuated and back-filled with nitrogen five times before it was immersed in an oil bath at 60 °C. Four cylindrical glass tubes were initially added into the reaction system. EBrP initiator (0.101 mL, 0.78 mmol) initiator was added to start the gelation reaction. At timed intervals, samples were withdrawn via a syringe for GC measurements of monomer conversion. The system formed a gel at certain moment when the reaction fluid lost its mobility at an upside down position for 10 s. After gelation, the reaction was kept at 60 °C for another 2 days before stopping the reaction via exposure

to air. The catalyst was removed from the gel by repeated THF extraction and filtration. The experimental procedures were similar for ATRP of MA with 3A, 4A, 5A, or 6A.

Synthesis of Copolymer Poly(HEA-co-*n*BA)

A polymerization was conducted using reaction conditions $[nBA]_0/[HEA]_0/[MBP]/[CuBr]_0/[CuBr_2]_0/[PMDTA]_0 = 25/25/1/0.45/0.05/0.5$ (in 25 vol % of DMF). *n*BA (41.7 mmol, 6.0 mL), HEA (41.7 mol, 4.89 mL), Cu(II)Br₂ (0.083 mmol, 18.6 mg), PMDTA as ligand (0.75 mmol, 0.174 mL), and DMF as solvent (3.63 mL, 25 vol %) were added to a reaction flask and the mixture was bubbled with N₂ for 1 h to remove oxygen from the reaction mixture. After that, Cu(I)Br (0.749 mmol, 0.108 g) was added and the reaction flask was placed in an oil bath at 50 °C. When the reaction mixture reached the desired reaction temperature, the polymerization was started by adding MBP (1.66 mmol, 0.186 mL) as the initiator. Samples were withdrawn periodically to monitor the monomer conversion (by GC) and the average molecular weight (by GPC). The reaction was ended after 180' by cooling the reaction mixture in liquid nitrogen. GC analysis revealed a monomer conversion of 41% for HEA and 34% for *n*BA, respectively. This ratio was confirmed by ¹H NMR analysis of the pure polymer. The copolymer was separated from the monomer by precipitation in a large excess of cold distilled water, followed by centrifugation. After drying in vacuum at room temperature overnight, the pure copolymer was obtained as a highly viscous material.

Synthesis of Multivinyl Crosslinker Based on Poly(HEA-co-*n*BA)

Poly(HEA-co-*n*BA) copolymer ($M_n = 1700$ g/mol and $M_w/M_n = 1.11$) was reacted with acryloyl chloride in a similar procedure as described above, 0.1 mol % of radical inhibitor 4-hydroxy-TEMPO was added relative to vinyl groups to prevent crosslinking during the esterification reaction. The reaction product was purified by dialysis in THF, yielding a polymer with pendant vinyl groups ($M_n = 1850$ g/mol and $M_w/M_n = 1.12$). ¹H NMR spectroscopy of the pure polymer revealed an esterification yield of about 90%.

RESULTS AND DISCUSSION

The ATRP of MA monomer and various crosslinkers containing different numbers of acrylate moieties was performed using CuBr/PMDTA as the catalyst (10 mol % of CuBr₂ of total copper species was preadded in the system), ethyl-2-bromopropionate (EBRP) as the initiator with an initial molar ratio of $[MA]_0/[EBRP]_0 = 50$ in DMF ($[MA]_0 = 6.0$ M) at 60 °C. These reaction conditions have been previously reported as a good system for ATRP of acrylate-based monomers, exhibiting fast initiation kinetics, a high initiation efficiency, and controlled polymer chain growth with low polydispersity of the primary chains.¹⁴

In this article, two approaches were used to investigate the effect of crosslinker multiplicity on experimental gelation. First, the concentration of branching vinyl groups (molar ratio of $[Vinyl_{branch}]_0/[Initiator]_0$) was kept constant for series of reactions with increasing crosslinker multiplicity. In addition,

TABLE 1 Comparison of ATRP Reactions of MA Using Crosslinkers with Different Multiplicity^a

$[Vinyl_{branch}]_0/[Initiator]_0^b$	Conv _{MA,gel} ^c				
	2A	3A	4A	5A	6A
2.20	0.97	0.94	0.89	0.99	/ ^d
3.00	0.86	0.78	0.79	0.83	/ ^d
4.00	0.79	0.67	0.67	0.65	0.82
10.0	0.48	0.40	0.39	0.34	0.34

^a Experimental conditions: (1) "2A" reaction: ATRP of MA and 2A with $[MA]_0/[2A]_0/[EBRP]_0/[CuBr]_0/[CuBr_2]_0/[PMDTA]_0 = 50/A/1/0.45/0.05/0.5$; (2) "3A" reaction: ATRP of MA and 3A: $[MA]_0/[3A]_0/[EBRP]_0/[CuBr]_0/[CuBr_2]_0/[PMDTA]_0 = 50/B/1/0.45/0.05/0.5$; (3) ATRP of MA and 4A: $[MA]_0/[4A]_0/[EBRP]_0/[CuBr]_0/[CuBr_2]_0/[PMDTA]_0 = 50/C/1/0.45/0.05/0.5$; (4) ATRP of MA and 5A: $[MA]_0/[5A]_0/[EBRP]_0/[CuBr]_0/[CuBr_2]_0/[PMDTA]_0 = 50/D/1/0.45/0.05/0.5$; (5) ATRP of MA and 6A: $[MA]_0/[6A]_0/[EBRP]_0/[CuBr]_0/[CuBr_2]_0/[PMDTA]_0 = 50/E/1/0.45/0.05/0.5$. All reactions were performed with $[MA]_0 = 6.0$ M in DMF at 60 °C and stopped after 2 days to reach the complete conversions.

^b $[Vinyl_{branch}]_0/[Initiator]_0 = (2 \times A)$ for "2A" reaction, $(3 \times B)$ for "3A" reaction, $(4 \times C)$ for reaction with 4A, $(5 \times D)$ for reaction with 5A, and $(6 \times E)$ for reaction with 6A. For the comparable reactions with the same value of $[Vinyl_{branch}]_0/[Initiator]_0$, $B = 2 \times A/3$, $C = A/2$, $D = 2 \times A/5$, $E = A/3$.

^c Conversions of MA immediately before gelation. The experimental gel point was the moment when the reaction fluid lost its mobility when held at an upside down position for 10 s.

^d No gelation occurred under these conditions.

tion, the effect of crosslinker multiplicity was investigated in a series of reactions where the molar concentration of the various crosslinkers was kept at a constant value.

Gelation Reactions with Constant Molar Ratio of $[Vinyl_{branch}]_0/[Initiator]_0$

In the first series of reactions, gelation reactions with ethylene glycol diacrylate (2A) and trimethylolpropane triacrylate (3A) crosslinkers were compared. The molar ratio of vinyl groups that can introduce branching (2 per molecule for 2A; 3 for 3A, etc.) to initiator was kept at a constant value (constant $[Vinyl_{branch}]_0/[Initiator]_0$). For various molar ratios of $[Vinyl_{branch}]_0/[Initiator]_0$ ranging from 2.20 to 10, experimental gelation occurred consistently at lower MA conversions for reactions with the trivinyl crosslinker 3A, as compared to the use of divinyl crosslinker (Table 1). Upon incorporation of a trivinyl crosslinker into a growing polymer chain, two pendant vinyl groups are generated. If one of them is consumed by intramolecular cyclization, the second one is still available to react and form an intermolecular crosslinkage. In contrast, if the pendant vinyl group from an incorporated divinyl crosslinker reacts intramolecularly, this crosslinker does not contribute to the increase of the molecular weight of the branched polymers. As a result, the use of a trivinyl crosslinker for the copolymerization leads to a network with a higher extent of intermolecular crosslinkages than the use of 1.5 eq. of divinyl crosslinker.

In similar series of reactions with various molar ratios of $[Vinyl_{branch}]_0/[Initiator]_0$, gelation reactions were also conducted using a tetravinyl crosslinker (pentaerythritol tetraacrylate, 4A), a pentavinyl crosslinker (dipentaerythritol

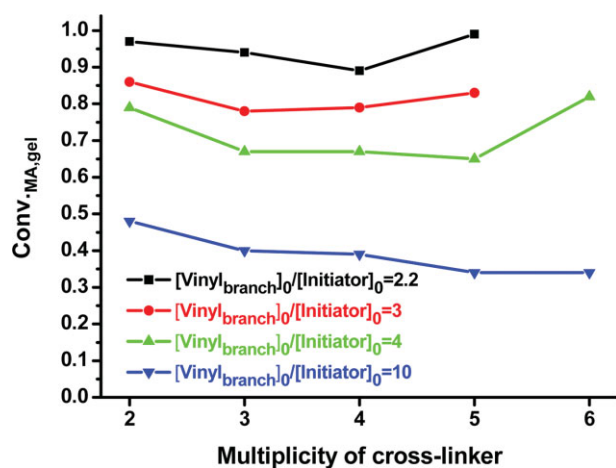


FIGURE 1 Conversion of MA at gelation for ATRP of MA and crosslinker as a function of crosslinker multiplicity for various constant molar ratios of $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$.

pentaacrylate, 5A), and a hexafunctional acrylate crosslinker (6A). It should be noted that 5A crosslinker is based on dipentaerythritol penta-/hexa-acrylate, which could contain small amounts of triester and hexafunctional compound. In our study, no further purification was performed as the average multiplicity was determined by ^1H NMR as being a pentafunctional acrylate. The synthesis of hexafunctional acrylate 6A is depicted in Scheme 2. The results of the gelation reactions are summarized in Table 1.

For reactions using the trivinyl crosslinker 3A, gelation occurred at consistently lower monomer conversion as compared to the gelation by using 2A crosslinker (Fig. 1). However, for reactions using the tetravinyl crosslinker 4A, no significant influence of the crosslinker multiplicity was observed under the same $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$ ratio, as compared to the gelation by using 3A crosslinker. When the crosslinker containing five acrylate moieties per molecule was used, a different trend was observed in comparable reactions. For low $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$ ratios (i.e., 2.2 and 3), gelation seems to be delayed with increasing multiplicity of the crosslinker. For instance, for the reaction with 5A, the experimental gel point was only observed at 99% conversion for a ratio of $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$ ratio = 2.2. In contrast, for higher $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$ ratios (i.e., 4 and 10), increased multiplicity of the crosslinker still resulted in gelation at earlier MA conversion. The effect tends to be even more clear for reactions with $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0 = 10$. This trend is confirmed for the reactions using hexafunctional crosslinker 6A. In this case, no gelation was observed for reactions with a $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$ ratio = 2.2 and 3, whereas gelation was significantly delayed for the reaction with a $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$ ratio = 4. Only for the reaction with a $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$ ratio = 10, no gelation delay was observed. We attribute these observations to a dilution effect: a constant $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$ ratio directly implies that the concentration of crosslinker in the reaction medium decreases with increasing

multiplicity of the crosslinker. This is schematically depicted in Figure 2.

For low $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$, the crosslinker concentration may drop below a critical value, resulting in a more pronounced intramolecular cyclization as the effect of a lower concentration in the reaction medium. It was previously reported that a decreasing concentration of the reaction medium results in delayed experimental gelation due to significantly increased intramolecular cyclization.¹⁵ Similarly, a decreased crosslinker concentration, originating from the increased multiplicity of the crosslinker with a constant ratio of $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$, could also lead to delayed experimental gelation. In other words, an increasing multiplicity increased the local concentration of pendant vinyl groups within one chain and increased the intramolecular cyclization.

On the other hand, if the $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$ ratio is rather high, the crosslinker concentration may be sufficient to prevent an intramolecular cyclization, resulting in earlier gelation with increasing crosslinker multiplicity.

Gelation Reactions with Constant Molar Concentration of Crosslinker

The previously discussed reactions with a constant $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$ ratio correlated the experimental gelation behavior with a dilution effect, resulting from an increasing multiplicity of the crosslinker. However, gelation reactions with a relatively high $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$ ratio still showed gelation at lower monomer conversion for increasing multiplicity of the crosslinker. To further investigate the influence of the crosslinker multiplicity in relation with its concentration, two series of gelation reactions were performed with two constant molar ratios of $[\text{monomer}]/[\text{initiator}]/[\text{crosslinker}]$ ($[\text{M}]_0/[\text{I}]_0/[\text{X}]_0 = 50/1/1$ and $50/1/2$, under similar reaction conditions as described above. In other words, the molar concentration of the crosslinker in the medium is identical for all gelation reactions, regardless of the number of branching vinyl groups per crosslinker molecule (see schematic depiction in Fig. 3).

In contrast to reactions with a constant molar ratio of $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$ for the various crosslinkers, this study avoids the influence of effects that are directly related to dilution of the crosslinker molecules in the reaction medium. For example, for the series of reactions with $[\text{M}]_0/[\text{I}]_0/[\text{X}]_0 = 50/1/1$, the $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$ ratio was 2, 3, 4, 5, and 6, for divinyl crosslinker 2A, trivinyl crosslinker 3A, tetravinyl crosslinker 4A, pentavinyl crosslinker 5A, and hexavinyl crosslinker 6A, respectively. At $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0 = 2$, no gelation was observed when 2A was used as crosslinker. The experimental gel points for these series of gelation reactions with constant concentration of crosslinker are displayed in Figure 4. For a constant concentration of the crosslinker, gelation occurred consistently at lower MA conversions for reactions with an increasing number of vinyl groups per crosslinker molecule. A similar trend was observed for a series of gelation reactions with $[\text{M}]_0/[\text{I}]_0/[\text{X}]_0 = 50/1/2$.

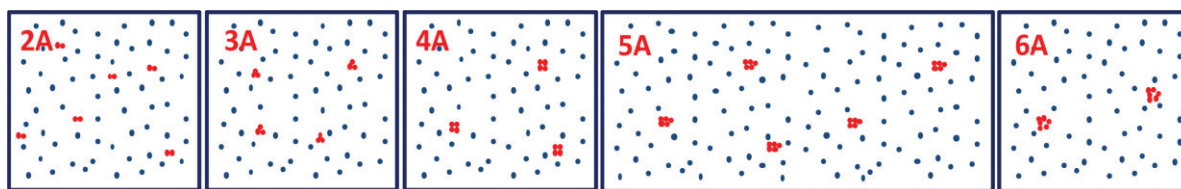


FIGURE 2 Schematic depiction of reaction medium of ATRP of MA (blue dots) and various crosslinkers with different multiplicity (in red: 2A, 3A, 4A, 5A, and 6A containing 2, 3, 4, 5, or 6 vinyl groups per molecule, respectively) for a series of gelation reactions with constant molar ratio of $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0 = 12$.

Determination of Pendant Vinyl Groups in the Sols

The reactions with various constant ratios of $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$ for various crosslinkers with different multiplicity suggested a gelation behavior related to the concentration of the crosslinker in the reaction medium. To further investigate this effect, the number of unreacted pendant vinyl groups in the sols was determined for reactions with different multiplicity of the crosslinker and various ratios of $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$. If equal reactivity of each vinyl group is assumed, the fraction of each species originating from the multivinyl crosslinker is determined by the statistical combination formula below.

$$P(K = k) = \binom{n}{k} p^k (1 - p)^{n-k}$$

$$\text{with } \binom{n}{k} = \frac{n!}{k!(n-k)!}$$

where p is conversion of M and k = number of vinyl groups reacted out of total number n .

This formula expresses the probability that k (out of total number n) vinyl groups have reacted at monomer conversion p , and defines the fraction of each species relative to the starting amount of unreacted crosslinker. For a crosslinker containing n vinyl groups, $n + 1$ species can be distinguished. For example, four different species originating from the crosslinker can be observed in the ATRP of monomer and 3A crosslinker: (i) a fraction with none of the vinyl groups reacted ($k = 0$), representing the unreacted free crosslinker; (ii) a fraction of the crosslinker with one vinyl group reacted ($k = 1$), which is incorporated in a primary chain and contains two pendant vinyl groups; (iii) a fraction of the crosslinker with two vinyl groups reacted ($k = 2$) and 1 pendant vinyl group; and (iv) a fraction of crosslinker that has fully reacted ($k = 3$). Therefore, the theoretical fraction

of the pendant vinyl groups in the sols ($f_{A,p}$) during ATRP of monomer and trivinyl crosslinker is the sum of the pendant vinyl groups present in the fractions of $k = 1$ and $k = 2$.

The theoretical and experimental fractions of pendant vinyl groups in the sols ($f_{A,p}$) as a function of MA conversion during ATRP of MA and pentavinyl crosslinker 5A were compared for a few reactions with various molar ratios of $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$ (Fig. 5). The dots in Figure 5 represent the experimental results of $f_{A,p}$ determined by ^1H NMR analysis of the sols just before gelation. Unreacted MA and free crosslinker were removed from the polymer samples by repeated precipitation/redissolving in hexane/tetrahydrofuran. ^1H NMR integration results determined the molar ratio of pendant vinyl groups to reacted MA units in the polymer backbone, which was further converted into the $f_{A,p}$ taking into account the MA conversion.¹⁴ The results in Figure 5 show that the experimental $f_{A,p}$ was in good agreement with the calculated $f_{A,p}$ for reactions with relatively low ratio of $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$ (i.e. 3 and 4). However, the experimental value of $f_{A,p}$ was higher than the theoretical value when the ratio of $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$ was 10. This difference could be attributed to the steric protection of dangling chains, resulting in the pendant vinyl groups less likely to react. For a relatively low $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$ ratio (3 and 4), the distance between the various crosslinkages would be relatively larger (lower crosslink density), compared to reactions with a higher $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0$ ratio (6 and 10; higher crosslink density). Intermolecular crosslinking reactions would be more difficult in the latter case, resulting in a higher $f_{A,p}$ than the theoretical value.

Further investigation for a molar ratio of $[\text{Vinyl}_{\text{branch}}]_0/[\text{Initiator}]_0 = 10$ was done for trivinyl crosslinker 3A, and tetra-vinyl crosslinker 4A. It was previously reported that theoretical and experimental $f_{A,p}$ values for ATRP of MA and 2A

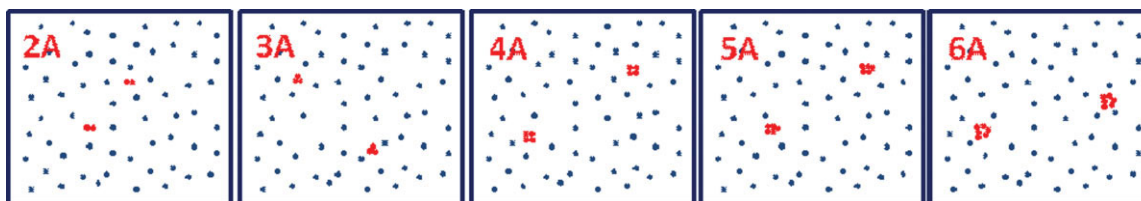


FIGURE 3 Schematic depiction of reaction medium of ATRP of MA (blue dots) and various crosslinkers with different multiplicity (in red: 2A, 3A, 4A, 5A, and 6A containing 2, 3, 4, 5, or 6 vinyl groups per crosslinker molecule, respectively) for a series of gelation reactions with a constant concentration of crosslinker.

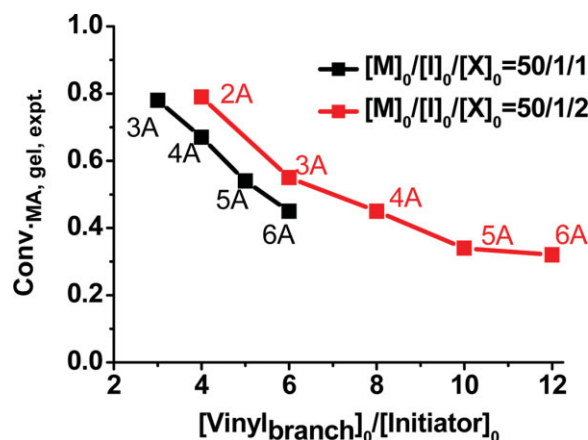


FIGURE 4 Conversion of MA at experimental gel point for ATRP of MA and crosslinker with various multiplicities, using a constant molar crosslinker concentration $[X]$.

were in a good agreement.^{14,26} For relatively low $[\text{Vinylbranch}]_0/[\text{Initiator}]_0$ ratio (i.e. 3 and 4), no significant differences between the theoretical and experimental $f_{A,p}$ for ATRP of MA and 3A and 4A were observed, just as in the case of ATRP of MA and 5A. However, for a high ratio of $[\text{Vinylbranch}]_0/[\text{Initiator}]_0$ (i.e. 10), a growing discrepancy between the theoretical and experimental $f_{A,p}$ was found with increasing multiplicity of the crosslinker (see Fig. 6).

Gelation with a Polymeric Multivinyl Crosslinker (7A)

In addition to the use of the crosslinkers with well-defined number of vinyl groups per molecule, we have also studied a gelation in the presence of an oligomeric (polymer-based) crosslinker. The crosslinker was prepared based on a random copolymer of *n*-butyl acrylate (*n*BA) and 2-hydroxyethyl acrylate (HEA), followed by the esterification of the hydroxyl groups in the side chain to acrylate moieties using acryloyl chloride. The average number of acrylate moieties per macromolecule was 7, as determined by ¹H NMR. In the reactions with a constant molar ratio of $[\text{Vinylbranch}]_0/[\text{Initiator}]_0 = 4$, the ATRP of MA and multivinyl crosslinker 7A showed no

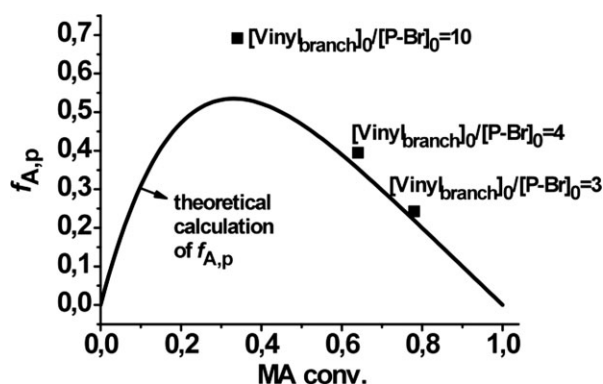


FIGURE 5 Comparison of calculated results (solid line) and experimental results (dots) of the fraction of pendant acrylate groups ($f_{A,p}$) as a function of MA conversion during the ATRP of MA and pentaacrylate crosslinker 5A.

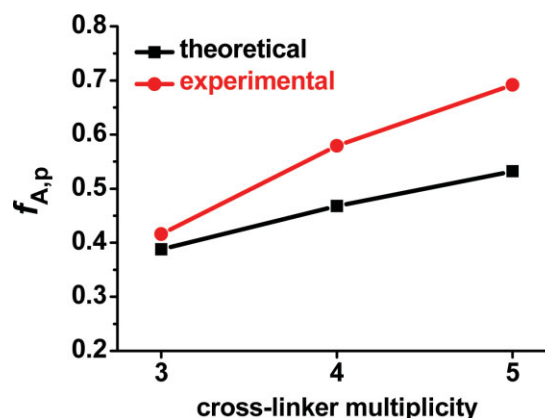


FIGURE 6 Theoretical and experimental fraction of pendant acrylate groups ($f_{A,p}$) in the sols as a function of crosslinker multiplicity for ATRP of MA and 3A, 4A, and 5A, respectively, for $[\text{Vinylbranch}]_0/[\text{Initiator}]_0$ ratio = 10. Samples were taken close to

gelation up to 91% MA conversion, in contrast to reactions with 5A (experimental gelation at 65% MA conversion) and 6A (experimental gelation at 82% MA conversion). In the reactions with a constant concentration of crosslinker ($[\text{M}]_0/[\text{I}]_0/[\text{X}]_0 = 50/1/1$), ATRP of MA and 7A showed a delayed gelation (at 99% MA conversion) in comparison to a similar reaction with the same concentration of 5A (gelation at 54% MA conversion). The difference in the gelation behavior could be attributed to the oligomeric nature of the multivinyl crosslinker and also to the linear structure of 7A. The acrylate crosslinking groups located next to each other along the backbone could facilitate intramolecular crosslinking reactions.

CONCLUSIONS

The effect of various types of crosslinkers containing different numbers of vinyl groups per molecule (2, 3, 4, 5, or 6) on the gelation behavior was studied during the ATRP of MA. At a constant ratio of branching vinyl groups to initiator, a concentration-dependant gelation for crosslinkers with different numbers of vinyl groups per molecule was observed. At a low targeted crosslink density, the experimental gelation was delayed for crosslinkers with higher number of vinyl groups per molecule (5 or 6). At a higher targeted crosslink density, increasing the number of vinyl groups per crosslinker molecule led to slightly accelerated gelation. When a constant molar concentration of various crosslinkers was used, experimental gelation occurred consistently at lower monomer conversions for reactions with a higher number of vinyl groups per crosslinker molecule. Moreover, the amount of unreacted pendant vinyl groups in the sols was determined for a series of reactions with various crosslinkers and different molar ratios of $[\text{Vinylbranch}]_0/[\text{Initiator}]_0$. The experimentally determined fraction of pendant acrylate groups ($f_{A,p}$) was significantly higher than the value theoretically predicted for gelation at a high targeted crosslink density. The discrepancy between the theoretical and experimental

$f_{A,p}$ increased with increasing number of vinyl groups per crosslinker molecule ($5A > 4A > 3A$). These results help to better understand the gelation process in controlled radical copolymerization of monomer and crosslinkers with variable multiplicities.

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