

CHEMISTRY

Synthesis of Selenium Analogs of 1-Azabicyclo[3.3.1]nonane

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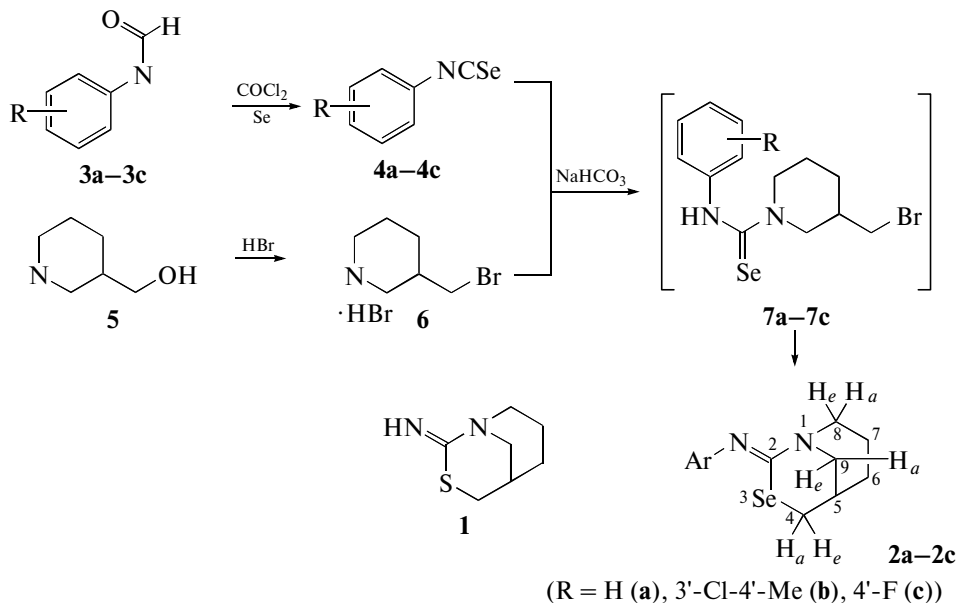
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The synthesis of organoselenium compounds, especially selenium-containing heterocycles, have lately attracted considerable attention due to their chemical properties and a wide spectrum of biological activity [1]. In particular, they show antioxidant [2], anti-inflammatory [3], and antiviral [4] properties. Isoselenocyanates are efficient reagents for the preparation of selenium-containing heterocyclic compounds on account of their high reactivity, low toxicity, and relative stability [5]. For example, isoselenocyanates were used in the synthesis of selenazolidines and selenadiazines [6].

We reported recently the synthesis of bicyclic tetrahydrothiazines (**1**), which can be considered as cyclic derivatives of isothiourrea [7]. This paper deals with the synthesis of previously unknown selenium-containing heterocyclic compounds.

Selenium analogs (**2a–2c**) of such bicyclic molecules in the series of 1-azabicyclo[3.3.1]nonane were obtained by the condensation of aryl isoselenocyanates (**4a–4c**) with 3-bromomethylpiperidine (**6**). Aryl isoselenocyanates (**4a–4c**) were prepared from the corresponding formamides (**3a–3c**) by refluxing in toluene with phosgene and selenium (powder) in the presence of triethylamine [8]. 3-Bromomethylpiperidine (**6**) was synthesized in quantitative yield from the corresponding alcohol (**5**) by heating at reflux with hydrobromic acid. The condensation of the initial compounds was carried out in an ethanol solution in the presence of sodium hydrogen carbonate. Resulting selenourea (**7a–7c**) immediately undergoes intramolecular alkylation at the selenium atom, transformation into isoselenourea, and ring closure to give a bicyclic structure of N-aryl(3-selena-1-azabicyclo[3.3.1]non-2-yliden)amine (**2a–2c**).



The obtained bicyclic compounds (**2a–2c**) were found to show high inhibitory activity in the glutamate-stimulated $^{45}\text{Ca}^{2+}$ uptake by rat cerebral cortex synaptosomes, which affects pathological processes in various neurodegenerative diseases [9]. Thus, it was shown for compound **2b** that $K_{\text{inh}} = 5.6\%$ (K_{inh} is the percentage of $^{45}\text{Ca}^{2+}$ uptake by rat cerebral cortex synaptosomes relative to control, control = 100%) and $IC_{50} = 34.7 \mu\text{M}$ are at the level of those for compounds used in the treatment of Alzheimer disease, for example Memantin ($K_{\text{inh}} = 8\%$, $IC_{50} = 22.9 \mu\text{M}$).

EXPERIMENTAL

^1H NMR spectra were recorded on a Bruker CXP-200 spectrometer (Germany), chemical shifts are given on the δ scale using Me_4Si as a reference. Melting points were determined with the use of a Boetius hot-stage apparatus and they were uncorrected. Solutions were concentrated by rotary evaporation in a vacuum of a water-jet pump.

Synthesis of *N*-aryl(3-selena-1-azabicyclo[3.3.1]non-2-yliden)amines (**2a–2c**) (general procedure).

Aryl isoselenocyanate **4a–4c** (0.01 mol) and 3-bromomethylpiperidine **6** (2.56 g, 0.01 mol) were dissolved in 30 mL of methanol. A solution of sodium hydrogen carbonate (1.85 g, 0.022 mol) in a minimal amount of water was added dropwise to the resultant solution and the mixture was stirred at ambient temperature for 1 h. The reaction mixture was diluted with water (20 mL), the methanol was evaporated, and the residue was extracted with chloroform ($3 \times 10 \text{ mL}$). The combined organic extracts were dried with sodium sulfate. The drying agent was removed by filtration, the filtrate was concentrated to give *N*-aryl(3-selena-1-azabicyclo[3.3.1]non-2-yliden)amine (**2a–2c**).

***N*-Phenyl(3-selena-1-azabicyclo[3.3.1]non-2-yliden)amine 2a.** Colorless crystals, mp 90–92°C, yield 95.2%.

^1H NMR (CDCl_3 , δ , ppm): 7.35 (m, 2H, H_{arom}), 7.12 (tt, 1H, J 1.3, 7.3 Hz, H_{arom}), 6.88 (m, 2H, H_{arom}), 4.28 (dm, 1H, J 13.3 Hz, $\text{C}(9)\text{H}_e$), 3.69 (dq, 1H, J 2.1, 13.7 Hz, $\text{C}(8)\text{H}_e$), 3.26 (m, 3H, $\text{C}(4)\text{H}_e$, $\text{C}(8)\text{H}_a$, $\text{C}(9)\text{H}_a$), 2.87 (ddd, 1H, J 0.5, 3.0, 11.6 Hz, $\text{C}(4)\text{H}_a$), 2.33 (dd, 1H, J 2.8, 10.9 Hz, $\text{C}(5)\text{H}$), 1.82 (m, 3H, $\text{C}(6)\text{H}_e$, $\text{C}(7)\text{H}_2$), 1.51 (m, 1H, $\text{C}(6)\text{H}_a$).

***N*-(3'-Chloro-4'-methylphenyl)(3-selena-1-azabicyclo[3.3.1]non-2-yliden)amine 2b.** Brown oil, yield 88.4%.

^1H NMR (CDCl_3 , δ , ppm): 7.14 (d, 1H, J 8.0 Hz, H_{arom}), 6.80 (d, 1H, J 2.0 Hz, H_{arom}), 6.64 (dd, 1H, J 2.0, 8.0 Hz, H_{arom}), 4.21 (dm, 1H, J 12.2 Hz, $\text{C}(9)\text{H}_e$), 3.63 (dq, 1H, J 1.9, 13.7 Hz, $\text{C}(8)\text{H}_e$), 3.21 (m, 3H, $\text{C}(4)\text{H}_e$, $\text{C}(8)\text{H}_a$, $\text{C}(9)\text{H}_a$), 2.84 (dd, 1H, J 2.8, 11.6 Hz, $\text{C}(4)\text{H}_a$), 2.33 (s, 3H, CH_3), 2.30 (dd, 1H, J 2.8, 10.9 Hz, $\text{C}(5)\text{H}$), 1.81 (m, 3H, $\text{C}(6)\text{H}_e$, $\text{C}(7)\text{H}_2$), 1.47 (m, 1H, $\text{C}(6)\text{H}_a$).

***N*-(4-Fluorophenyl)(3-selena-1-azabicyclo[3.3.1]non-2-yliden)amine 2c.** Colorless crystals, mp 106–108°C, yield 93.5%.

^1H NMR (CDCl_3 , δ , ppm): 7.03 (m, 2H, H_{arom}), 6.82 (m, 2H, H_{arom}), 4.25 (dd, 1H, J 2.3, 12.8 Hz, $\text{C}(9)\text{H}_e$), 3.67 (dd, 1H, J 2.2, 13.7 Hz, $\text{C}(8)\text{H}_e$), 3.26 (m, 3H, $\text{C}(4)\text{H}_e$, $\text{C}(8)\text{H}_a$, $\text{C}(9)\text{H}_a$), 2.88 (dd, 1H, J 2.7, 11.5 Hz, $\text{C}(4)\text{H}_a$), 2.33 (dd, 1H, J 2.8, 10.8 Hz, $\text{C}(5)\text{H}$), 1.89 (m, 3H, $\text{C}(6)\text{H}_e$, $\text{C}(7)\text{H}_2$), 1.51 (m, 1H, $\text{C}(6)\text{H}_a$).

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REFERENCES

1. Nogueira, C.W., Zeni, G., and Rocha, J.B.T., *Chem. Rev.*, 2004, vol. 104, pp. 6255–6286.
2. Hossain, S.U., Sharma, A.K., Ghosh, S., and Bhattacharya, S., *Eur. J. Med. Chem.*, 2010, vol. 45, pp. 1–3273.
3. Abdel-Hafez, S.H., *Europ. J. Med. Chem.*, 2008, vol. 43, pp. 1971–1977.
4. Jeong, L.S., Choi, Y.N., Tosh, D.K., et al., *J. Bioorg. Med. Chem.*, 2008, vol. 16, pp. 9891–9897.
5. López, O., Maza, S., Ulgar, V., et al., *Tetrahedron*, 2009, vol. 65, pp. 2550–2566.
6. Xie, Y., Liu, J., and Li, J., *Tetrahedron Lett.*, 2011, vol. 52, pp. 932–935.
7. Zefirova, O.N., Moisseeva, N.N., Proshin, A.N., et al., *Mendeleev Commun.*, 2011, vol. 21, pp. 247–249.
8. Proshin, A.N., Serkov, I.V., and Bachurin, S.O., *Dokl. Chem.*, 2010, vol. 430, part 1, pp. 8–10.
9. Lipton, S., *Nat. Rev. Drug. Discov.*, 2006, vol. 5, pp. 160–170.