

TETRAHEDRON

An Efficient Chemoenzymatic Access to Chiral 3,7-Diazabicyclo[3.3.1]nonane Derivatives

Bruno Danieli, Giordano Lesma,^{*} Daniele Passarella, Alessandra Silvani and Nunzia Viviani

Dipartimento di Chimica Organica e Industriale, Università degli Studi di Milano, Centro CNR di Studio per le Sostanze Organiche Naturali, Via Venezian, 21-20133-Milano, Italy

Received 5 May 1999; revised 13 July 1999; accepted 28 July 1999

Abstract

Enantiopure 3,7-diazabicyclo[3.3.1]nonane derivatives 4 and 5, potential precursors of quinolizidine alkaloids, were synthesised in high yields, starting from the biocatalytic asymmetrization of σ -symmetric 3,5-disubstituted piperidines. Their application to the total synthesis of the new pharmacologically active compounds 3 are also described. © 1999 Elsevier Science Ltd. All rights reserved.

Keywords: alkaloids, aminals, bicyclic heterocyclic compounds, enzyme reactions.

Quinolizidine alkaloids are widely distributed in various plant families and show an interesting range of biological and chemical properties.¹ For example, (-)-cytisine (1) and some of its derivatives act as agonists at the nicotinic receptor and have been the subject of a number of pharmacologic² and synthetic³ studies during the last few years. Another basic component of this class of natural products, (-)-sparteine (2), has recently received significant attention due to its use as a chiral chelating base for Li⁺ in asymmetric deprotonations with alkyllithium agents.⁴

A notable feature, common to these molecules, is the presence of a chiral 3,7-diazabicyclo[3.3.1]nonane system, on which one or two other rings are fused. This moiety has also been recently incorporated in a number of new pharmacologically active compounds of general structure 3,^{5,6} claimed as cholinergic agents for the prevention and treatment of Alzheimer's disease and other cerebral function disorders (Figure 1). Up to

now, access to such structures has been secured by a long sequence of reactions starting from 3-bromo-5carboxypyridine and terminating with an optical resolution step.



We report here the first enantioselective synthetic entry to $(1S,2\xi,5S)$ -2-hydroxy-7-benzyloxylcarbonyl-3,7diazabicyclo[3,3,1]nonane (4) and (1R,4R,5S)-2-oxo-4-hydroxy-7-benzyloxylcarbonyl-3,7-diazabicyclo [3.3.1]nonane (5), which we envisaged as immediate precursors of compounds of type 3 and as potential advanced intermediates in the stereocontrolled synthesis of tricyclic quinolizidine alkaloids *via* the corresponding imine and acylimine functionalities easily generated by Lewis acid treatment.

Recently, we reported⁷ the preparation of *cis*-piperidine-3,5-dimethanol monoacetates 6 and 7, in high enantiomeric excesses and yields, by means of biocatalytic asymmetrization of the corresponding σ -symmetric diol or diacetate (Scheme 1).



Prompted by the opportunity to explore the synthetic versatility of these new chiral building blocks, we first planned the preparation of 4, starting from (3S,5R)-N-benzyloxylcarbonyl-3-acetoxymethyl-5-hydroxymethyl piperidine 6, as depicted in Scheme 2.

Introduction of the amino functionality was achieved by a three step reaction sequence, starting with treatment of 6 with MsCl to afford the corresponding mesylate 8; then nucleophilic displacement using sodium azide⁸ gave 9 in high yield. Reduction of the azido group of 9 was performed with stannous chloride in methanol as solvent;⁹ the reaction was complete within 1 hour and afforded 10 in almost quantitative yield. Then the primary amino group of 10 was protected by means of (Boc)₂O to give the *N*-Boc derivative 11 in which

hydrolysis of the acetate function occurred. Swern oxidation of the hydroxymethyl moiety of 11 gave a mixture of the diastereoisomeric cyclic aminals 12 in nearly quantitative yield. Notably, this cyclization is the result of a fast favoured ring formation occurring in a minor, less stable conformation of the putative intermediate aldehyde in which the C3 and C5 substituents are axially disposed. Epimers corresponding to the structure 12 exchanged slowly at room temperature, giving rise to broad NMR signals. Removal of the *N*-Boc protecting group using TFA, followed by exposure to ice-cold, aqueous sodium hydroxide of the resulting TFA salt and immediate solvent extraction, gave the desired $(1S, 2\xi, 5S)$ -2-hydroxy-7-benzyloxylcarbonyl-3.7-diazabicyclo[3.3.1]nonane 4, in 85% yield from 12.



(a) MsCl, Py (98%); (b) NaN₃, DMF, 70 °C (92%); (c) SnCl₂, MeOH, reflux (95%);(d) Boc₂O, NaOH(aq), THF (94%); (e) (COCl)₂, DMSO, CH₂Cl₂ (90%); (f) TFA, then NaOH(aq), (85%); (g) NaBH₃CN, CH₂O, THF (71%); (h) PDC, Celite®, CH₂Cl₂ (65%);

Scheme 2

The proposed structure for **4** was further supported by performing a reduction of the masked aldehydic C-2 followed by a reductive methylation at the N-3 atom by means of NaBH₃CN in the presence of formaldehyde in THF. The high symmetry of the 3-benzyloxylcarbonyl-7-methyl-3,7-diazabicyclo[3.3.1]nonane **13** thus obtained, resulted in very simplified ¹H and ¹³C NMR spectra. On the other hand, oxidation of **4**, with PDC¹⁰ in dichloromethane afforded the (1S,5S)-2-oxo-7-benzyloxylcarbonyl-3,7-diazabycyclo[3.3.1]nonane **14**, which represents the heterocyclic motif of the above mentioned compounds of type **3**, in one of its enantiomeric forms.



For the synthesis of (1R, 4R, 5S)-5, the sequence of steps required was modified, as illustrated in Scheme 3.

(a) KMnO₄, CH₃COCH₃(75%); (b) SOCl₂, NH₃(g) (98%); (c) NaOH, THF (96%); (d) Dess-Martin periodinane, CH₂Cl₂ (82%); (e) Et₃SiH, BF₃Et₂O, CH₂Cl₂ (52%).

Scheme 3

Oxidation of **6** with KMnO₄ in anhydrous acetone led to formation of the carboxylic acid **15**, in 75% yield. Reaction of the acid chloride of **15** with gaseous NH₃ afforded quantitatively the amide **16**, which was easily converted to the corresponding deacetylated compound **17**. Oxidation of **17** was strongly dependent upon the reagent and conditions: reaction with DMSO/pyridine-SO₃ or with DMSO/(COCl)₂ gave complex mixtures of products, not easily recoverable after aqueous work up. Instead, oxidation of **17** worked smoothly using Dess-Martin periodinane¹¹ in CH₂Cl₂, yielding **82%** of lactamol **5**, as a single diastereoisomer. Close inspection of the ¹H and ¹³C NMR (APT) spectra allowed the complete attribution of signals to the whole protonic system; moreover, the configuration at C-4 was ascertained by the presence of a n.O.e. contact between the proton H-4 (δ 5.07, d) and proS H-9 (δ 2.13, br d). The n.O.e. interaction was detectable in a double-chair conformation of **5** in which H-4 and proS H-9 are juxtaposed in a synaxial disposition. Finally, a Lewis acid-catalyzed triethylsilane reduction¹² was performed on **5**, affording **ent-14** in acceptable yield.

In conclusion, we succeeded in developing a convenient enantioselective synthetic entry into 3,7diazabicyclo[3.3.1]nonane derivatives 4 and 5, functionalized at the position α to N-3 in such a way to make feasible the subsequent elaboration to tricyclic compounds. Finally, this study has contributed to finding an efficient access to 14 and ent-14 which represent the core of compounds of type 3.

Experimental¹³

General

All separations were carried out under flash chromatography (FC) conditions on silica gel 60 (230-400 mesh) using the indicated solvents. The organic extracts were dried over anhydrous Na₂SO₄ prior to solvent removal on a rotary evaporator.

Materials

(3S,5R)-3-Acetoxymethyl-5-hydroxymethyl-piperidine-1-carboxylic acid benzyl ester (6) was obtained in 78% yield and >98% ee by enzymatic acetylation of the corresponding diol as previously described.^{7b}

(3S, 5R)-3-Acetoxymethyl-5-aminomethyl-piperidine-1-carboxylic acid benzyl ester 10

Mesyl chloride (1.36 ml, 13.9 mmol) was added to a solution of 6 (3.7 g, 11.6 mmol) in pyridine (30 ml) at 0°C. After 4 h at rt, the reaction mixture was quenched by addition of sat. aq. NaHCO₁ at 0°C, acidified by 1N HCl and extracted with CH₂Cl₂. The combined organic extracts were washed with water, dried, and concentrated. FC (AcOEt : hexane 7:3) of the residue gave 8 (4.5 g, 98%): oil; Rf (AcOEt) 0.50; ¹H NMR (300 MHz, CDCl₃, 50 °C) & 7.34 (m, 5H), 5.13 (s, 2H), 4.30 (m, 2H), 4.11 (dd, 1H, J=10.0, 5.0 Hz), 4.04 (1H, dd, J=10.0, 5.0 Hz), 3.98 (dd, 1H, J=11.2, 5.7 Hz), 3.89 (dd, 1H, J=11.2, 6.8 Hz), 2.98 (s, 3H), 2.43 (t, 1H, J=12.5 Hz), 2.37 (t, 1H, J=12.5 Hz), 2.03 (s, 3H), 2.00-1.83 (m, 3H), 1.01 (q, 1H, J=11.9 Hz); EIMS m/z (relative intensity) 399 (10, M⁺), 354 (22), 339 (5), 250 (100). Sodium azide (950 mg, 14.6 mmol) was added in one portion to a solution of the mesylate 8 (2.20 g, 5.5 mmol) in DMF (30 ml) at rt. After heating at 75 °C for 8 h, the reaction mixture was partitioned between ether and water. The combined organic layers were dried and concentrated in vacuo to provide the almost pure crude azide 9 (1.70 g, 92%): oil; Rf(AcOEt) 0.65; IR (CHCl₃) 2150, 1300 cm⁻¹; ¹H NMR (300 MHz, CDCl₃, 50 °C) δ 7.35 (m, 5H), 5.12 (s, 2H), 4.26 (br d, 2H, J=12.0 Hz), 3.97 (dd, 1H, J=11.2, 5.2 Hz), 3.86 (dd, 1H, J=11.2, 6.4 Hz), 3.22 (m, 2H), 2.40 (t, 2H, J=12.6 Hz), 2.05 (s, 3H), 1.98-1.64 (m, 3H), 0.95 (q, 1H, J=12.5 Hz); EIMS m/z (relative intensity) 318 (3, M⁺-28), 183 (15), 91(100). To a stirred suspension of SnCl₂ (3.2 g, 9.53 mmol) in 33 ml of methanol was added dropwise a solution of the crude azide 9 (3.3 g, 9.53 mmol) in methanol (10 ml). The reaction was exothermic and N₂ gas was evolved. The resulting mixture was stirred at rt for 1h and then evaporated in vacuo. The residue was partitioned between ether (20 ml) and 5% NH₄OH solution (aq, 20 ml) and the precipitated inorganic material filtered off. The aq. portion was saturated with NaCl and further extracted with ether (5 x 5 ml). The combined organic layers were dried and concentrated to give, after FC (CHCl₁: methanol 19:1) of the residue, the amine 10 (2.9 g, 95 %): oil; ; R_f (CHCl₁: MeOH 9:1) 0.10; $[\alpha]_D^{25} - 5.4$ (c 1, CHCl₃); IR (CHCl₃) 3400, 1725 cm⁻¹; ¹H NMR (300 MHz, CDCl₃, 50 °C) δ 7.35 (m, 5H), 5.14 (s, 2H), 4.30 (br d, 2H, J=12.0 Hz), 4.00 (dd, 1H, J=11.4, 5.7 Hz), 3.88 (dd, 1H, J=11.4, 6.9 Hz), 2.65 (d, 2H, J=7.2 Hz), 2.40 (q, 2H, J=12.3 Hz), 2.04 (s, 3H), 2.00-1.79 (m, 2H), 1.78-1.50 (m, 1H), 0.85 (q, 1H, J=12.6 Hz); EIMS m/z (relative intensity) 320 (3, M*), 156 (48), 91(100); Anal. Calcd for C₁₇H₂₄N₂O₄: C, 63.73; H, 7.55; N, 8.74. Found: C, 63.84; H, 7.35; N, 8.76.

(3S,5R)-3-Hydroxymethyl-5-(t-butyloxycarbonyl)aminomethyl-piperidine-1-carboxylic acid benzyl ester 11

To a stirred solution of **10** (2.90 g, 9.0 mmol) in THF (20 ml) and NaOH 1N (10 ml) (Boc)₂O (2.40 g, 10.8 mmol) was added. After 6h at rt, the reaction mixture was poured into water and extracted with Et_2O ; the organic phase was washed with brine, dried and concentrated to give pure **11** (3.24 g, 94%): oil; R_f (AcOEt) 0.40; $[\alpha]_D^{25}$ – 6.2 (*c* 1, CHCl₃); ¹H NMR (300 MHz, CDCl₃, 50 °C) δ 7.35 (m, 5H), 5.12 (s, 2H), 4.50 (br s, 1H), 4.24 (m, 2H), 3.52 (dd, 1H, *J*=11.8, 5.6 Hz), 3.48 (dd, 1H, *J*=11.8, 6.0 Hz), 3.01 (t, 2H, *J*=6.4 Hz), 2.42 (m, 2H), 1.88 (br d, 1H, *J*=12.0 Hz), 1.80-1.60 (m, 2H), 1.45 (s, 9H), 0.88 (q, 1H, *J*=12.0 Hz); ¹³C NMR (CDCl₃, 75.4 MHz) δ 156.0, 155.3, 137.1, 128.5, 127.9, 127.8, 67.1, 65.3, 47.9, 47.1, 43.9, 38.6, 36.9, 31.7, 28.3; EIMS *m/z* (relative intensity) 378 (2, M⁺), 158 (77), 91(100); HRMS calcd for C₂₀H₃₀N₂O₅ 378.2155, found 378.2169.

(15,25,5)-2-Hydroxy-3-t-butyloxycarbonyl-7-benzyloxycarbonyl-3,7-diazabicyclo[3.3.1]nonane 12 (mixture of diastereomers)

At -7%C, a solution of oxalyl chloride (480 µl, 5.5 mmol) in CH₂Cl₂ (5 ml) was treated dropwise within 1h with a solution of DMSO (790 µl, 11.0 mm ol) in CH₂Cl₂ (5 ml). After stirring for 50 min, the resulting mixture was treated dropwise with a solution of 11 (1.0 g, 2.75 mmol) in CH₂Cl₂ (5 ml), stirred for 1h at -70%C and then treated with Et₃N (2 ml). After 1h at -70%C, the mixture was warmed

at rt and washed with water; the organic phase was dried and concentrated to give, after FC (AcOEt:exane 1:2) of the residue, pure **12** (880 mg, 90%): foam; R_f (AcOEt:hexane 1:1) 0.60; IR (CHCl₃) 1725, 1700 cm⁻¹; ¹H NMR (300 MHz, CDCl₃, 50 °C) δ 7.35 (m, 5H), 5.20-5.07 (m, 3H), 4.60 (br s, 1H), 4.41 (br d, 1H, *J*=12.2 Hz), 4.18 (br d, 1H, *J*=12.2 Hz), 3.05 (m, 2H), 2.74 (m, 1H), 2.45 (m, 2H), 2.15 (br d, 1H, *J*=13.1 Hz), 1.75 (m, 1H), 1.22 (br d, 1H, *J*=13.1 Hz); ¹³C NMR (CDCl₃, 75.4 MHz) δ 156.0, 155.0, 136.5, 128.4, 127.9, 127.7, 67.2, 58.7, 48.0, 47.5, 43.7, 36.3, 33.6, 28.5, 28.3; FABMS *m/z* 377 (MH⁺); Anal. Calcd for C₂₀H₂₈N₂O₅: C, 63.81; H, 7.50; N, 7.44. Found: C, 63.95; H, 7.37; N, 7.41.

(1S, 25, 5S)-2-Hydroxy-7-benzyloxycarbonyl-3, 7-diazabicyclo[3.3.1]nonane 4 (mixture of diastereomers)

At 0°C, **12** (880 mg, 2.3 mmol) was dissolved in TFA (5 ml) and stirred for 1h. Then the solution was poured into ice-cold, aq. NaOH 10% and immediatly extracted with CH₂Cl₂, to give **4** (546 mg, 85%): oil; R_f (19:1 CHCl₃/ MeOH) 0.43; $[\alpha]_D^{25} - 9.3$ (*c* 1, CHCl₃); ¹H NMR (300 MHz, CDCl₃, 50°C) δ 7.35 (m, 5H), 5.08 (m, 2H), 5.02 (s, br, 1H), 4.24 (m, 2H), 3.28 (m, 2H), 3.00 (d, br, 1H, *J* =13.0 Hz), 2.91 (d, br, 1H, *J* =13.0 Hz), 2.39 (m, 1H), 1.99 (m, 2H), 1.75 (m, 1H); ¹³C NMR (CDCl₃, 75.4 MHz) δ 156.5, 136.0, 129.0, 128.6, 68.4, 66.0, 59.2, 49.1, 44.8, 33.8, 28.9, 24.1; FABMS *m/z* 277 (MH⁺); HRMS calcd for C₁₅H₂₀N₂O₃ 276.1474, found 276.1482.

3-Benzyloxycarbonyl-7-methyl-3,7-diazabicyclo[3.3.1]nonane 13

A solution of 4 (180 mg, 0.6 mmol) in THF (5 ml) was treated with NaBH₃CN (164 mg, 2.6 mmol) and then with a solution of 37% aq. CH₂O. After stirring for 1h at rt, the mixture was partitioned between AcOEt and aq. NaHCO₃ 5%; the combined organic layers were dried and concentrated to give, after FC (CHCl₃ : MeOH 85:15), pure 13 (127 mg, 71%): oil; R_f (4:1 CHCl₃ / MeOH) 0.18; ¹H NMR (300 MHz, CDCl₃, 50°C) δ 7.35 (s, 5H), 5.13 (s, 2H), 4.23 (dd, 2H, *J* =13.5, 2.2 Hz), 3.09 (ddd, 2H, *J* =13.5, 3.5, 2.2 Hz), 2.95 (d, br, 2H, *J* =12.0 Hz), 2.24 (dt, br, 2H, *J* =12.0, 2.0 Hz), 2.14 (s, 3H), 2.05-1.84 (m, 2H), 1.81 (m, 1H), 1.65 (m, 1H); FABMS 275 *m/z* (MH⁺).

(1S,5S)-2-Oxo-7-benzyloxycarbonyl-3,7-diazabycyclo[3.3.1]nonane 14

To a solution of 4 (200 mg, 0.7 mmol) in CH₂Cl₂ (3 ml) pyridinium dichromate (376 mg, 1 mmol) was added. After stirring at rt for 3h, the mixture was diluted with Et₂O, the inorganic material was filtered off and the filtrate was evaporated, to give, after FC (AcOEt:hexane 2:1), pure 14 (118 mg, 65%): oil; R_f (2:1 EtOAc / *n*-hexane) 0.45; $[\alpha]_D = -8.4$ (*c* 1, CHCl₃); IR (CHCl₃) 1725, 1650 cm⁻¹; ¹H NMR (300 MHz, CDCl₃, 50°C) δ 7.37 (m, 5H),), 5.17 and 5.08 (AB system, J = 12.2 Hz), 4.48 (br d, 1H, J = 13.5 Hz), 4.20 (m, 1H), 3.47 (dd, 1H, J = 12.0, 6.8 Hz), 3.26 (br d, J = 12.0 Hz), 3.08 (m, 2H), 2.53 (m, 1H), 2.08 (m, 1H), 1.97 (br d, 1H, J = 14.5 Hz), 1.71 (br d, 1H, J = 14.5 Hz); FABMS *m/z* 275 (MH⁺); HRMS calcd for C₁₅H₁₈N₂O₃ 274.1317, found 274.1327.

(3R,5S)-3-Carboxyl-5-acetoxymethyl-piperidine-1-carboxylic acid benzyl ester 15

To a solution of 6 (200 mg, 0.62 mmol) in acetone (6 ml), KMnO₄ (200 mg, 1.27 mmol) was added in portions, at 0°C and the mixture was stirred for 1h at 0°C and for 2h at rt. EtOH (3 ml) was added and, after stirring for 1h, the brown precipitate was filtered through Celite[®]. The filtrate was evaporated and the residue was partitioned between Et₂O and NaHCO₃ 5%; the aq. phase was acidified with H₂SO₄ 10% and extracted with AcOEt, to give **15** (156 mg, 75%): amorphous solid; R_f (AcOEt) 0.29; $[\alpha]_D = -7.5$ (*c* 1, CHCl₃); IR (CHCl₃) 1780, 1725 cm⁻¹; ¹H NMR (300 MHz, CDCl₃, 50°C) δ 7.35 (m, 5H),), 5.12 (s, 2H), 4.46 (br d, 1H, *J* = 12.7 Hz), 4.28 (br d, 1H, *J* = 12.7 Hz), 3.98 (dd, 1H, *J* = 12.0, 5.5 Hz), 3.90 (dd, 1H, *J* = 12.0, 6.7 Hz), 2.79 (t, 1H, *J* = 12.0 Hz), 2.46 (t, 1H, *J* = 12.0 Hz), 2.18 (br d, 1H, *J* = 12.4 Hz), 2.03 (s, 3H), 1.91 (m, 2H), 1.37 (q, 1H, *J* = 12.4 Hz); ¹³C NMR (CDCl₃, 75.4 MHz) δ 177.5,

170.9, 155.1, 136.4, 128.5, 128.1, 127.9, 67.5, 65.9, 46.7, 45.4, 40.9, 35.0, 30.3, 20.7; EIMS *m/z* (relative intensity) 335 (28, M^{*}), 275 (54), 232 (100); Anal. Calcd for C₁₇H₂₁N₁O₆: C, 60.88; H, 6.32; N, 4.18. Found: C, 60.95; H, 6.47; N, 4.05.

(3R,5S)-3-Carboxyamido-5-acetoxymethyl-piperidine-1-carboxylic acid benzyl ester 16

A solution of **15** (232 mg, 0.70 mmol) in SOCl₂ (10 ml) was heated at reflux for 4h. Thionyl chloride was evaporated, the residue was dissolved in dioxane (10 ml), cooled at 0°C and saturated with gas. NH₃. After 3h at 0°C, the solution was concentrated, acidified with aq. H₂SO₄ 10% and extracted with AcOEt, to give pure **16** (229 mg, 98%): oil; R_f(AcOEt) 0.26; $[\alpha]_D = -6.0$ (*c* 1, CHCl₃); IR (CHCl₃) 1725, 1680 cm⁻¹; ¹H NMR (300 MHz, CDCl₃, 50°C) δ 7.35 (m, 5H), 5.43 (br s, 2H), 5.12 (s, 2H), 4.33 (br d, 1H, *J* = 13.0 Hz), 4.24 (br d, 1H, *J* = 13.0 Hz), 4.00 (dd, 1H, *J* = 11.2, 5.4 Hz), 3.90 (dd, 1H, *J* = 11.2, 6.8 Hz), 2.87 (dd, 1H, *J* = 13.6, 11.8 Hz), 2.53 (br t, 1H, *J* = 13.6 Hz), 2.35 (m, 1H), 2.03 (s, 3H), 2.01 (m, 1H), 1.88 (m, 1H), 1.54 (q, 1H, *J* = 12.0 Hz); ¹³C NMR (CDCl₃, 75.4 MHz) δ 175.8, 170.9, 155.2, 136.3, 128.5, 128.0, 127.7, 127.3, 67.4, 65.9, 46.6, 42.1, 34.4, 30.3, 20.7; FABMS *m/z* 335 (MH²).

(3R,5S)-3-Carboxyamido-5-hydroxymethyl-piperidine-1-carboxylic acid benzyl ester 17

To a stirred solution of **16** (150 mg, 0.45 mmol) in THF (3 ml), aq. NaOH 0.5 N (1.35 ml, 0.68 mmol) was added. After 2h at rt, usual work up gave pure **17** (121 mg, 92%): oil; R_f (CHCl₃:MeOH 9:1) 0.44; IR (CHCl₃) 1680, 1603 cm⁻¹; ¹H NMR (300 MHz, CDCl₃, 50°C) δ 7.10 (m, 5H), 6.53 (br s, 1H), 5.60 (br s, 1H), 4.89 (s, 2H), 4.07 (m, 2H), 3.57 (br s, 1H), 3.17 (m, 2H), 2.60 (t, 1H, *J* = 12.0 Hz), 2.32-2.09 (m, 2H), 1.78 (br d, 1H, *J* = 13.0 Hz), 1.46 (m, 1H), 1.16 (q, 1H, *J* = 13.0 Hz); ¹³C NMR (CDCl₃, 75.4 MHz) δ 175.5, 154.9, 136.6, 128.2, 127.7, 127.5, 66.7, 64.3, 46.6, 46.2, 42.0, 38.0, 30.8; EIMS *m/z* (relative intensity) 292 (2, M⁺), 157 (47), 91 (100); HRMS calcd for C₁₅H₂₀N₂O₄ 292.1423, found 292.1418.

(1R,4R,5S)-2-Oxo-4-hydroxy-7-benzyloxycarbonyl-3,7-diazabicyclo[3.3.1]nonane 5

Freshly prepared Dess-Martin reagent¹¹ (310 mg, 0.73 mmol) was added to a stirred solution of 17 (165 mg, 0.56 mmol) in CH₂Cl₂ (20 ml). The solution turned yellow, and after 10 min, a colourless precipitate formed. Filtration through Celite[®] after 25 min gave a pale yellow filtrate, from which part of the solvent was removed by vacuum evaporation. FC (CH₂Cl₂:MeOH 95:5) of the residue gave pure 5 (128 mg, 82%) as a single epimer (from 300 MHz ¹H NMR). 5: amorphous solid; R_f (CH₂Cl₂:MeOH 9:1) 0.46; $[\alpha]_D = -$ 3.3 (*c* 1, CHCl₃); ¹H NMR (300 MHz, CDCl₃, 50°C) δ 7.35 (s, 5H), 6.70 (br s, 1H), 5.14 and 5.06 (AB system, *J*= 12.1 Hz), 5.07 (d, 1H, *J* = 5.7 Hz), 4.54 (br d, 1H, *J* = 14.2 Hz), 4.43 (br d, 1H, *J* = 13.5 Hz), 3.11 (br d, 1H, *J* = 13.5 Hz), 2.92 (dd, 1H, *J* = 14.2, 3.0 Hz), 2.49 (m, 1H), 2.19-2.08 (m, 2H), 1.96 (ddd, 1H, *J* = 14.5, 4.4, 3.0 Hz); ¹³C NMR (CDCl₃, 75.4 MHz) δ 173.5, 168.1, 143.0, 128.4, 128.1, 78.7, 68.2, 48.0, 43.8, 37.6, 32.7, 27.8; EIMS *m/z* (relative intensity) 290 (4, M⁺), 272 (22), 166 (65), 91 (100); Anal. Calcd for C₁₅H₁₈N₂O₄: C, 62.05; H, 6.25; N, 9.65. Found: C, 62.02; H, 6.17; N, 9.48.

(1R.5R)-2-Oxo-7-benzyloxycarbonyl-3,7-diazabicyclo[3.3.1]nonane ent-14

At -40°C, Et₃SiH (40 μ l, 0.25 mmol) and BF₃OEt₂ (31 μ l, 0.25 mmol) were added dropwise to a solution of 5 (60 mg, 0.21 mmol) in 3 ml of CH₃Cl₂. The resulting mixture was warmed at rt, stirred for 5h and then treated with sat. aq. Na₂CO₃. Normal work up, followed by FC (AcOEt:hexane 2:1) of the residue gave ent-14 (28 mg, 52% yield).

Acknowledgement

This work was supported by Ministero dell'Università e della Ricerca Scientifica e Tecnologica (MURST).

References and notes

- Reviews on quinolizidine alkaloids, see: (a) The Alkaloids, G.A. Cordell, Ed., 1995, 47(1), Academic Press Inc. (b) J.P. Michael, Nat. Prod. Rep., 1997, 14(1), 21-41, and earlier references in this series.
- 2. Schmeller, T.; Sauerwein, M.; Sporer, F.; Wink, M.; Muller, W.E. J. Natural Products, 1994, 57, 1316-1319.
- Fazylov, S.; Gazaliev, A.M.; Vlasova, L. M.; Kasenov, R. Z. Zh.Obshch. Khim., 1995, 65, 877 [Chem. Abstr. 1995, 123, 288605].
- Schwerdtfeger, J.; Hoppe, D. Angew. Chem. Int. Ed. Engl., 1992, 31, 1505-1507; Zschage, O., Schwark, J.-R., Hoppe, D. Angew. Chem. Int. Ed. Engl., 1990, 29, 296-298; Muci, A.R.; Campos, K.R.; Evans, D.A. J. Am. Chem. Soc., 1995, 117, 9075-9076.
- 5. Kobayashi, K.; Orita, K.; Hamada, A.; Inaba, T.; Abe, H.; Miyazaki, S., PCT Int. Appl. WO 96 30,372 [Chem. Abstr. 1997, 126, 8107f].
- 6. Inaba, T; Abe, H.; Miyazaki, S., Jpn. Kokai Tokkio Koho JP 09, 255,679 [Chem. Abstr. 1997, 127, 293136q].
- (a) Danieli, B.; Lesma, G.; Passarella, D.; Silvani, A. Tetrahedron: Asymm., 1996, 7, 345-348; (b) Danieli, B.; Lesma, G.; Passarella, D.; Silvani, A. J. Org. Chem., 1998, 63, 3492-3496.
- 8. Gmeiner, P.; Orechner, F.; Thomas, C.; Weber, K. Tetrahedron Lett., 1995, 36, 381-382.
- 9. Maiti, S. N.; Singh, M. P.; Micetich, R.G. Tetrahedron Lett., 1986, 27, 1423-1424.
- 10. Benbow, J. W.; Schulte, G. K.; Danishefsky, S. J. Angew. Chem. Int. Ed. Engl. 1992, 31, 915-917.
- 11. Dess, D. B.; Martin, J. C. J. Am. Chem. Soc., 1991, 113, 7277-7287.
- 12. Yamazaki, H.; Horikawa, H.; Nishitani, T.; Iwasaki, T. Chem. Pharm. Bull., 1990, 38, 2024-2026.
- For typical experimental protocols see: Danieli, B.; Lesma, G.; Mauro, M.; Palmisano, G.; Passarella, D. Tetrahedron 1994, 50, 8837-8852.