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Total Synthesis of (—)-Chamobtusin A

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ABSTRACT

The first asymmetric total synthesis of a structurally unique alkaloid, chamobtusin A (1), is described. The route features a novel aziridine formation from the 1,2-oxazine derivative and a palladium-mediated annulation of the vinylaziridine intermediate.

Chamobtusin A (1), isolated by Tan and co-workers from *Chamaecyparis obtusa* cv. *tetragon* in 2007, is the first alkaloid from the order Pinales, as well as the first diterpene alkaloid from the family *Cupressacea*. Its structure was established mainly on the basis of 2D NMR techniques (Figure 1) and was confirmed by single-crystal X-ray diffraction analysis, but the absolute configuration has not been assigned. To date, two syntheses of racemic chamobtusin A have been reported independently by Watanabe's group² and our group³ using an approach that featured an intramolecular aza-Michael reaction to construct the key nitrogen-bearing quaternary center. We now report the first enantioselective total synthesis of (—)-chamobtusin A (1).

Our retrosynthetic analysis for chamobtusin A is shown in Scheme 1. Chamobtusin A (1) would be derived from perhydrobenzoindole 2 by manipulation of the keto alcohol side chain and the 2*H*-pyrrole moiety. The key C-8 stereocenter of 2 could be introduced through a stereoselective allylation of tricyclic 1,2-oxazine 3 bearing the siloxy group at C-9, which could be derived from octahydronaphthalenone 4.

As depicted in Scheme 2, our synthesis commenced with the preparation of the known α -hydroxy ketone 5^4 from the known ketone 4, which was obtained in optically pure

Figure 1. Structure of chamobtusin A (1).

form from (S)-(+)-Wieland-Miescher ketone⁶ in seven steps, following the published procedure.⁷ Oxidation of the obtained 5 with Dess-Martin periodinane and treatment of the resulting diketone with LHMDS followed by TBSCl trapping of the enolate gave the corresponding *tert*-butyl-dimethylsilyl enol ether 6, which was successively treated with vinyl Grignard reagent and tetrabutylammonium fluoride (TBAF) in THF to produce the desired vinyl alcohol 7 as a single diastereomer in 77% yield over four steps from 5. Having successfully constructed the requisite C-9 stereocenter, we set out to synthesize the tricyclic 1, 2-oxazine 3. Thus, 7 was subjected to hydroboration with dicyclohexylborane (Chx₂BH)⁸ followed by oxidation of the alkylborane intermediate to afford primary alcohol 8 in

⁽¹⁾ Zhang, Y.-M.; Tan, N.-H.; Lu, Y.; Chang, Y.; Jia, R.-R. Org. Lett. 2007, 9, 4579–4581.

⁽²⁾ Kuzuya, K.; Mori, N.; Watanabe, H. Org. Lett. 2010, 12, 4709-4711.

⁽³⁾ Suzuki, H.; Aoyagi, S. Chem. Commun. 2011, 47, 7878–7879.

⁽⁴⁾ Ihara, M.; Toyota, M.; Fukumoto, K. J. Chem. Soc., Perkin Trans. 1 1986, 2151–2161.

⁽⁵⁾ Hatzellis, K.; Pagona, G.; Spyros, A.; Demetzos, C.; Katerinopoulos, H. E. J. Nat. Prod. 2004, 67, 1996–2001.

⁽⁶⁾ Buchschacher, P.; Fürst, A.; Gutzwiller, J. Organic Syntheses; Wiley: New York, 1990; Collect. Vol. VII, pp 368–372.

⁽⁷⁾ Rubottom, G. M.; Gruber, J. M.; Juve, H. D., Jr.; Charleson, D. A. Org. Synth. 1986, 64, 118–126.

^{(8) (}a) Pelter, A.; Smith, K.; Brown, H. C. *Borane Reagents*; Academic Press: London, 1988. (b) Kabalka, G. W.; Yu, S.; Li, N.-S. *Tetrahedron Lett.* **1997**, *38*, 5455–5458.

Scheme 1. Retrosynthetic Analysis for Chamobtusin A (1)

65% yield. Treatment of **8** with *N*-hydroxyphthalimide under Mitsunobu conditions gave **9** in quantitive yield. Deprotection of the phthalimide group of **9** by treating with hydrazine monohydrate, heating of the formed hemiaminal with acetic acid in EtOH, and exposure to TMSOTf and 2,6-lutidine resulted in an 84% yield of the oxime ether **3**.

Scheme 2. Preparation of 1,2-Oxazine 3

With the desired oxime ether 3 in hand, we next focused on stereoselective construction of the pyrrolidine moiety and elaboration of the necessary side chain of perhydrobenzoindole 2 (Scheme 3). Thus, in order to stereoselectively introduce the allyl group at C-8, 3 was subjected to

Scheme 3. Formation of Vinylaziridine Derivative 12

reaction with allylmagnesium chloride (4 equiv) in THF at 0 °C. To our surprise, however, this reaction provided the unexpected spiro aziridine 11 in 58% yield, presumably via initial abstraction of an allylic hydrogen by an appropriately oriented proximal allyl component on the nitrogen of allylated intermediate complex 10, followed by simultaneous aziridine ring formation/N—O bond cleavage. ¹⁰ The stereostructure of 11 was confirmed by X-ray crystallographic analysis of the *N*,*O*-dinosyl derivative 12, derived in 85% yield from 11.

Our next task was to elaborate the tricyclic framework of chamobtusin A. For this, we envisioned use of a palladium-mediated ring-expansion, 11 in which a π -allylpalladium intermediate 13 derived from vinylaziridine derivative 12 by palladium-catalyzed hydrogenolysis, 12 would serve as a precursor for an intramolecular S_N2 -like amination (Scheme 4). Thus, when 12 was subjected to the standard conditions for palladium-catalyzed hydrogenolysis ($Pd_2(dba)_3CHCl_3$, PBu_3 , formic acid, Et_3N), the cascade reaction occurred to generate the desired perhydrobenzoindole 2, albeit in low yield (21%). The use of PPh_3 as a ligand, however, greatly increased the yield of 2 to 92%.

Subsequent transformation of the C-8 allyl moiety into the 3-hydroxy-3-methyl-2-butanone side chain was accomplished by means of our previously developed sequence

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⁽⁹⁾ Treatment of 7 with borane reagents such as 9-BBN, disiamylborane, or thexylborane gave low yields or complex mixtures.

⁽¹⁰⁾ For related examples of the preparation of aziridines from oximes, see: (a) Kotera, K.; Takano, Y.; Matsuura, A.; Kitahonoki, K. *Tetrahedron* **1970**, *26*, 539–556. (b) Freeman, J. P. *Chem. Rev.* **1973**, 73, 283–292

⁽¹¹⁾ For related palladium-mediated annulation of vinylaziridines, see: Lowe, M. A.; Ostovar, M.; Ferrini, S.; Chen, C. C.; Lawrence, P. G.; Fontana, F.; Calabrese, A. A.; Aggarwal, V. K. *Angew. Chem., Int. Ed.* **2011**, *50*, 6370–6374.

^{(12) (}a) Tsuji, J.; Minami, I.; Shimizu, I. *Synthesis* **1986**, 623–627. (b) Mandai, T.; Matsumoto, T.; Kawada, M.; Tsuji, J. *J. Org. Chem.* **1992**, *57*, 1326–1327. (c) Tsuji, J.; Mandai, T. *Synthesis* **1996**, 1–24.

Scheme 4. Synthesis of Perhydrobenzoindole 2

(Scheme 5). Dihydroxylation of 2 with OsO₄ and NMO, followed by NaIO₄ cleavage of the resulting diol, provided aldehyde 14 in 80% yield. Wittig reaction gave alkene 15 in 86% yield. At this stage, the silvl protecting group of 15 was removed, and the resulting tertiary alcohol was treated with Burgess' reagent¹³ in THF at reflux to cleanly afford dihydropyrrole 16 in 92% yield for these two steps. Regioselective dihydroxylation of the side chain of 16, in preference to the ring double bond with OsO₄, followed by TPAP oxidation of the resulting diol gave keto alcohol 17 in 72% yield over two steps. Finally, the nosyl group was removed using Fukuyama's conditions, 14 and the resulting secondary amine was oxidized with iodosobenzene¹⁵ in CH₂Cl₂ at room temperature, furnishing (–)-chamobtusin A (1) in 78% yield over two steps. The spectroscopic data (¹H and ¹³C NMR, and IR) and optical rotation $[\alpha]^{25}$ _D -233.2 (c 0.14, MeOH) (lit. 1 [α] 24 _D -220.1 (c 0.24, MeOH))] of synthetic 1 were in accord with those reported for the natural product.

In summary, we have completed the first asymmetric synthesis and determined the absolute configuration

Scheme 5. Synthesis of (–)-Chamobtusin A

of (–)-chamobtusin A. The key features of this synthesis include a novel aziridine formation from the corresponding 1,2-oxazine derivative and a palladium-mediated annulation of the vinylaziridine derivative.

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Supporting Information Available. Experimental procedures and compound characterization data including X-ray data (CIF). This material is available free of charge via the Internet at http://pubs.acs.org.

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⁽¹³⁾ Atkins, G. M.; Burgess, E. M. J. Am. Chem. Soc. 1968, 90, 4744–4745.

^{(14) (}a) Kurosawa, W.; Kan, T.; Fukuyama, T. Org. Synth. 2002, 79, 186–195. (b) Kan, T.; Fukuyama, T. Chem. Commun. 2004, 353–359.

^{(15) (}a) Muller, P.; Gilabert, D. M. *Tetrahedron* **1988**, *44*, 7171–7175. (b) Larsen, J.; Jorgensen, K. A. *J. Chem. Soc., Perkin Trans.* **2 1992**, 1213–1217. (c) Han, G.; LaPorte, M. G.; Folmer, J. J.; Werner, K. M.; Weinreb, S. M. *J. Org. Chem.* **2000**, *65*, 6293–6306.

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