One-pot, three-component synthesis of polyfunctionalized benzo[*h*]pyrazolo[3,4-*b*][1,6]naphthyridine and benzo[*g*]pyrazolo[3,4*b*]quinoline derivatives in the presence of silver nanoparticles (AgNPs)



Jabbar Khalafy, Fatemeh Majidi Arlan*, Ahmad Poursattar Marjani, and Vahid Sarchami

Department of Organic Chemistry, Faculty of Chemistry, Urmia University, Urmia, Irancorre

Email: majiidi@yahoo.com

Abstract

An efficient and straightforward protocol for one-pot, three-component reaction of aryl glyoxal monohydrates **1a-h**, 5-amino-1-aryl-3-methylpyrazoles **2a,b** and 4-hydroxyquinoline-2(1H)-one **3** or 2-hydroxy-1,4-naphthoquinone **4** using AgNPs as a high performance nanocatalyst in H₂O/EtOH at 60 °C afforded the corresponding polyfunctionalized benzo[*h*]pyrazolo[3,4-*b*][1,6]naphthyridines **5a-h** and benzo[*g*]pyrazolo[3,4-*b*]quinolines **6a-i**, respectively. Excellent catalytic activity, high yields, employing green media and green nanocatalyst, cost-effective and simple procedure are some notable advantages of using AgNPs as a noble metal nanocatalyst in this synthetic strategy. The structures of fused heterocycles were confirmed by their FT-IR, ¹H-NMR, and ¹³C-NMR spectral data and microanalysis.

1. INTRODUCTION

According to researchers, the heterocycles containing at least one nitrogen atom generally, have exhibited biological and pharmaceutical properties. Hence, contriving and synthesis of such aza-heterocycles and aza-aromatic compounds have been recognized worthwhile, and have become a major challenge in organic synthesis. Among these, polyfunctionalized heterocycles have considered as magic structures due to appearance of several hetero rings with wide medicinal and clinical applications. Polyfunctionalized fused heterocyclic scaffolds due to the appearance of diverse N containing rings simultaneously, have supposed to create a versatile bioactive compound so we faced an increasing prevalence of research in this field.^[1-3] As mentioned, pyrazoloquinoline derivatives are highly regarded due to their pharmaceutical and biological properties.^[4-8]

This article has been accepted for publication and undergone full peer review but has not been through the copyediting, typesetting, pagination and proofreading process which may lead to differences between this version and the Version of Record. Please cite this article as doi: 10.1002/jhet.4105

One of the most prominent and facile strategies in organic synthesis and specially, synthesis of heterocyclic compounds and natural products is multicomponent reactions (MCRs); which undoubtedly, has provided a compatible path in the synthesis by breaking or making multiple bonds through environmentally and economically useful one-pot procedures avoiding separation of intermediates or changing the solvent.^[9-16]

During recent years, chemists have organized sustainable MCRs, using green solvents alongside novel nanocatalysts have altered this strategy to a conspicuous tool. High catalytic activity, selectivity, stability, reusability and moving along green chemistry are notable features of nanocatalysts. Among them, noble metal nanocatalyst such as AgNPs due to superior physicochemical, and biological properties, environmentally benign, and low-cost have extensively investigated.^[17]

Afterwards, synthesis of heterocycles containing bioactive core framework with diverse biological and pharmaceutical activities is noteworthy and contriving economical, simple and efficient methods to achieve such compounds is much more various. Aryl glyoxals are known as an important synthon in the preparation of complex hetero scaffolds due to the presence of two active sites that has made it as 1,2-biacceptor which could be attacked by nucleophilic reagents.^[18-21]

In continuation of our previous studies on the application of aryl glyoxals in the synthesis of heterocyclic compounds,^[22-29] herein, we report a one-pot, three-component reaction of aryl glyoxal monohydrates, 5-amino-1-aryl-3-methylpyrazoles and 4-hydroxyquinoline-2(1H)-one or 2-hydroxy-1,4-naphthoquinone (lawsone) using AgNPs as nanocatalyst in H₂O/EtOH under mild conditions to afford the corresponding pyrazolonaphthyridines and pyrazoloquinolines respectively in good to high yields.

2. RESULTS AND DISCUSSION

It was found that the reaction of aryl glyoxal monohydrates **1a-h**, 5-amino-1-aryl-3methylpyrazoles **2a,b** and 4-hydroxyquinoline-2(1*H*)-one (**3**) or 2-hydroxy-1,4-naphthoquinone (**4**) using AgNPs as nanocatalyst in H₂O/EtOH (1:1) at 60 °C gave the desired benzo[*h*]pyrazolo[3,4-*b*][1,6]naphthyridines **5a-h** and benzo[*g*]pyrazolo[3,4-*b*]quinolines **6a-i** in good to high yields (Scheme 1). The products were fully characterized by their FT-IR and ¹H-NMR and ¹³C-NMR spectral data and microanalysis.



SCHEME 1 Synthesis of benzo[*h*]pyrazolo[3,4-*b*][1,6]naphthyridines **5a-h** and benzo[*g*]pyrazolo[3,4-*b*]quinolines **6a-i**

In our initial studies, we evaluated one-pot, three-component reaction of pheyl glyoxal monohydrates 1a, 5-amino-1-phenyl-3-methylpyrazole 2a, and 4-hydroxyquinoline-2(1*H*)-one **3**. The equimolar mixture, of 1a, 2a, and **3**, was tested under various conditions. The effects of solvent, catalyst and temperature were appraised for model reaction, and the results are summarized in Table 1.

Firstly, the reaction was carried out under catalyst-free conditions at room temperature but, no product was observed even after 24 h (Table 1, entry 1). Increasing the temperature of reaction up to 60 $^{\circ}$ C under catalyst-free condition in H₂O/EtOH (1:1) for 1 h, also did not give any product (Table 1, entry 2). Using 5 and 10 ppm of AgNPs as nanocatalyst, the desired products achieved in good yields (Table 1, entries 11-12). The best result was obtained in terms of yield 87% and reaction time 1 h, while 10 ppm of AgNPs were utilized (Table 1, entry 13). Employing 20 ppm of AgNPs, the yield of reaction was not improved so much (Table 1, entry 14). The model reaction was carried out using various solvents such as ethanol, acetic acid, and water to find the best solvent system. Among these solvents, H₂O/EtOH (1:1) evaluated as the best

solvent system (Table 1, entry 13). Based on achieved results, we shifted our attention toward using various ordinary catalysts for the same model reaction in H₂O/EtOH (1:1) at 60 °C. Other catalysts such as DBU, Et₃N, *p*-TSA, and *L*-proline were tested under similar reaction conditions to obtain the maximum yield and minimum reaction time. The results are mentioned in (Table 1, entries 7-10). In conclusion, the reaction was performed in the presence of catalytic amount of AgNPs in H₂O/EtOH at 60 °C, the smooth progress of reaction was clear, by monitoring the reaction completion *via* TLC. High yield of 7-benzoyl-8-methyl-10-phenyl-5,10-dihydro-6*H*-benzo[*h*]pyrazolo[3,4-*b*][1,6]naphthyridin-6-one was obtained within 1 h.

TABLE 1 Optimization of the reaction conditions for synthesis of 5a



Entry	Catalyst (mol%)	Solvent	Temperature (°C)	Time (h)	Yield (%)
1	-	H ₂ O/EtOH (1:1)	r.t.	24	_
2	-	H ₂ O/EtOH (1:1)	60	1	-
3	-	H ₂ O/EtOH (2:1)	60	1	-
4	-	H_2O	60	1	-
5	-	EtOH	60	1	-
6	-	AcOH	60	1	-
7	DBU (10 mol%)	H ₂ O/EtOH (1:1)	60	1	35
8	Et ₃ N (10 mol%)	H ₂ O/EtOH (1:1)	60	1	31
9	<i>p</i> -TSA (10 mol%)	H ₂ O/EtOH (1:1)	60	1	32
10	L-proline (10 mol%)	H ₂ O/EtOH (1:1)	60	1	40
11	Ag bulk	H ₂ O/EtOH (1:1)	60	1	37
12	AgNPs (5 ppm)	H ₂ O/EtOH (1:1)	60	1	68
13	AgNPs (10 ppm)	H ₂ O/EtOH (1:1)	60	1	87
14	AgNPs (20 ppm)	H ₂ O/EtOH (1:1)	60	1	87

15	AgNPs (10 ppm)	H ₂ O/EtOH (1:2)	60	1	68
16	AgNPs (10 ppm)	H ₂ O	60	1	56
17	AgNPs (10 ppm)	EtOH	60	1	64
18	AgNPs (10 ppm)	AcOH	60	1	70

The optimized conditions are shown in bold text.

The reaction conditions were then applied to a range of different aryl glyoxal monohydrates, to synthesize a series of products. The reaction times, melting points, and yields of the synthesized polyfunctional benzo[h]pyrazolo[3,4-b][1,6]naphthyridines **5a-h** are shown in Table 2.

TABLE 2 The yields, reaction times and melting points of compounds 5a-h

Я	$H_{1} = H_{1} = H_{1$						
	1a-h 2a 3			5a-	5a-h		
Entry	Products	R ₁	\mathbf{R}_2	Time (h)	Mp (°C)	Yield ^a (%)	
1	5a	Н	Н	1	310-311	87	
2	5b	4-Br	Н	2	300-302	68	
3	5c	4-Cl	Н	2	305-307	86	
4	5d	4-F	Н	1.5	298-300	75	
5	5e	4-Me	Н	1.5	304-306	77	
6	5 f	4-MeO	Н	1.5	303-305	79	
7	5g	3,4-(MeO) ₂	Η	1.5	268-270	81	
8	5h	$4-NO_2$	Η	2	298-300	65	

^aIsolated yield.

Further investigations, we tried to extend the scope of reactions so, 2-hydroxy-1,4-naphthoquinone **4** as dicarbonyl compound (active methylene) was employed. Interestingly, 2-hydroxy-1,4-naphthoquinone **4** in combination with aryl glyoxal monohydrates **1a-h** and 5-amino-1-aryl-3-methylpyrazoles **2a,b** provided the corresponding fused products **6a-i** in high

yields under the optimized reaction conditions as shown in Table 3. The reaction times, melting points, and yields of the synthesized benzo[g]pyrazolo[3,4-b]quinoline-5,10-diones **6a-i** are listed in Table 3.



1.5

1.5

2

1.5

2

1

1.5

272-274

284-286

275-277

295-298

272-274

263-265

248-251

88

90

93

89

88

93

87

Η

Η

Η

3-C1

3-C1

3-C1

3-C1

4-F

4-MeO

4-NO₂ 4-Cl

4-F

4-MeO

 $4-NO_2$

TABLE 3 The yields, reaction times and melting points of compounds 6a-i

^aIsolated yield.

3

4

5

6

7

8

9

6c

6d

6e

6f

6g

6h

6i

The proposed mechanism for this reaction, involves the *Knoevenage*l condensation of aryl glyoxals **1a-h** with 4-hydroxyquinolin-2(1*H*)-one **3** or 2-hydroxy-1,4-naphthoquinone **4** by loss of a water molecule to form the corresponding intermediate **7**. Following *Michael* addition of 3-methyl-1-aryl-1*H*-pyrazol-5-one **2a,b** to this intermediate will form the final products benzo[*h*]pyrazolo[3,4-*b*][1,6]naphthyridines **5a-h** and benzo[*g*]pyrazolo[3,4-*b*]quinoline-5,10-diones **6a-i**, through protonation to **9**, loss of a water molecule by intramolecular cyclization to **10**, followed by tautomerization to **11**, and final oxidation along with aromatization. (Scheme 2)



SCHEME 2 The proposed mechanism for synthesis of final products 5a-h and 6a-i

In the ¹H NMR spectra of products **5a-h** and **6a-i**, the characteristic broad singlets at around δ = 11.66-11.75 ppm, ascribed to the NH groups, and singlets at δ = 2.11-2.23 ppm are attributed to the methyl groups of pyrazole ring, which were present in all new products. In the ¹³C NMR

spectra, signals located around $\delta = 160.8-194.2$ were attributed to three different carbonyl groups. In the FT-IR (KBr) spectra, the characteristic absorptions bands at 3327-3411 cm⁻¹ and 1565-1680 cm⁻¹ could be assigned to the vibrations of NH and different carbonyl groups, respectively.

3. CONCLUSIONS

We have reported an efficient method for the synthesis of pyrazolonaphthyridines and pyrazoloquinolines through one-pot, three-component condensation reaction of aryl glyoxals, 5amino-1-aryl-3-methylpyrazoles and 4-hydroxyquinoline-2(1*H*)-one or 2-hydroxy-1,4naphthoquinone (lawsone) in the presence of AgNPs as a nobel metal nanocatalyst in EtOH/H₂O at 60 °C. The products **5a-h**, **6a**, **6d**, **6f**, **6h** and **6i** have been fully characterized by FT-IR, ¹H-NMR, and ¹³C-NMR spectral data and microanalysis, products **6b**, **6c** and **6e** have been confirmed by FT-IR and ¹H-NMR spectral data and microanalysis and also compound **6g**, characterized by FT-IR spectrum and microanalysis. The simplicity of the method, ease of product isolation, green reaction condition, and short reaction times are the advantages of this procedure. The synthesized pyrazolonaphthyridines and pyrazoloquinolines may have biological and pharmacological activities.

Experimental Section

The chemicals used in this work were purchased from Acros and Merck companies and were used without purification. Melting points were measured on an Electrothermal 9200 apparatus and are uncorrected. FT-IR (KBr) spectra were recorded on a Thermo-Nicolet (Nexus 670) spectrometer using KBr discs. ¹H (300 MHz) and ¹³C (75.5 MHz) NMR spectra were recorded on a Bruker DRX-300 Avance spectrometer in DMSO- d_6 with TMS as the internal reference. Elemental analyses were performed using a Leco Analyzer 932. The aryl glyoxals were prepared as their hydrates by oxidation of the corresponding acetophenones with SeO₂.^[30] The AgNPs as nanocatalyst was prepared according to the literature.^[31, 32]

General procedure for synthesis of benzo[*h*]pyrazolo[3,4-*b*][1,6]naphthyridines 5a-h and benzo[*g*]pyrazolo[3,4-*b*]quinoline-5,10-diones 6a-i

A mixture of 5-amino-1-aryl-3-methylpyrazoles (1 mmol), 4-hydroxyquinoline-2(1*H*)-one (1 mmol) or 2-hydroxy-1,4-naphthoquinone (1 mmol) and various aryl glyoxal monohydrates (1 mmol), in the presence of AgNPs (10 ppm) in H₂O/EtOH (1:1) (5 mL) was heated at 60° C for the appropriate time. The progress of the reaction was monitored by TLC (chloroform:*n*-hexane / 1:2). Upon the completion of the reaction, the mixture was cooled, and the precipitate was easily isolated by simple filtration and then rinsed with distilled cold water and ethanol. The pure products were obtained by recrystallization from EtOH.

The physical and spectral data for all new products as well as their elemental analyses are reported.

7-Benzoyl-8-methyl-10-phenyl-5,10-dihydro-6*H*-benzo[*h*]pyrazolo[3,4-*b*][1,6]naphthyridin-6-one (5a)

Yellow powder; mp: 310-312 °C; yield: 87% (375 mg). IR (KBr) ν_{max} : 3327, 3036, 2984, 2863, 1667, 1570, 1498, 1433, 1350, 128, 1169, 1125, 750 cm⁻¹. ¹H-NMR (300 MHz, DMSO- d_6) δ : 11.71 (s, exchanged by D₂O addition, 1H, NH), 8.76 (d, J = 7.8 Hz, 1H, Ar), 8.39 (d, J = 7.8 Hz, 2H, Ar), 7.81 (d, J = 7.2 Hz, 2H, Ar), 7.72-7.54 (m, 4H, Ar), 7.51 (t, J = 7.5 Hz, 2H, Ar), 7.40 (t, J = 7.5 Hz, 3H, Ar), 2.19 (s, 3H, CH₃) ppm. ¹³C-NMR (75.5 MHz, DMSO- d_6) δ : 193.2, 160.9, 151.9, 151.6, 146.2, 143.5, 139.1, 139.1, 137.1, 130.1, 129.8, 129.4, 129.0, 128.7, 122.0, 121.7, 120.1, 119.3, 114.7, 114.3, 14.2 ppm. Anal. Calcd for C₂₇H₁₈N₄O₂: C, 75.34; H, 4.21; N, 13.02. Found: C, 75.40; H, 4.15; N, 13.12.

7-(4-Bromobenzoyl)-8-methyl-10-phenyl-5,10-dihydro-6*H*-benzo[*h*]pyrazolo[3,4*b*][1,6]naphthyridin-6-one (5b)

Yellow powder; mp: 300-301 °C; yield: 68% (347 mg). IR (KBr) v_{max} : 3339, 2981, 2859, 1666, 1571, 1351, 1261, 1223, 1168, 1123, 1068, 756 cm⁻¹. ¹H-NMR (300 MHz, DMSO- d_6) δ : 11.75 (s, exchanged by D₂O addition, 1H, NH), 8.74 (d, J = 7.8 Hz, 1H, Ar), 8.39 (d, J = 8.1 Hz, 2H, Ar), 7.70-7.55 (m, 7H, Ar), 7.43-7.30 (m, 3H, Ar), 2.21 (s, 3H, CH₃) ppm. ¹³C-NMR (75.5 MHz, DMSO- d_6) δ : 193.1, 160.9, 151.9, 151.6, 145.4, 143.5, 139.1, 139.0, 136.1, 132.5, 131.1, 129.8, 128.3, 127.1, 122.6, 122.0, 120.1, 119.3, 114.7, 114.3, 14.2 ppm. Anal. Calcd for C₂₇H₁₇BrN₄O₂: C, 63.67; H, 3.36; N, 11.00. Found: C, 63.72; H, 3.42; N, 10.92.

7-(4-Chlorobenzoyl)-8-methyl-10-phenyl-5,10-dihydro-6*H*-benzo[*h*]pyrazolo[3,4*b*][1,6]naphthyridin-6-one (5c)

White powder; mp: 305-306 °C; yield: 86% (400 mg). IR (KBr) v_{max} : 3311, 2982, 2860, 1666, 1566, 1348, 1224, 1096, 1021, 758 cm⁻¹. ¹H-NMR (300 MHz, DMSO-*d*₆) δ : 11.75 (s, exchanged by D₂O addition, 1H, NH), 8.74 (d, *J* = 7.8 Hz, 1H, Ar), 8.39 (d, *J* = 7.8 Hz, 2H, Ar), 7.83 (d, *J* = 8.1 Hz, 2H, Ar), 7.75-7.60 (m, 3H, Ar), 7.56 (d, *J* = 8.1 Hz, 2H, Ar), 7.44-7.30 (m, 3H, Ar), 2.21 (s, 3H, CH₃) ppm. ¹³C-NMR (75.5 MHz, DMSO-*d*₆) δ : 192.9, 160.9, 151.9, 151.6, 145.5, 143.5, 139.1, 139.0, 135.8, 131.0, 129.9, 129.6, 128.5, 122.1, 121.9, 120.1, 119.9, 119.2, 114.7, 114.3, 14.2 ppm. Anal. Calcd for C₂₇H₁₇ClN₄O₂: C, 69.75; H, 3.69; N, 12.05. Found: C, 69.70; H, 3.72; N, 12.02.

7-(4-Fluorobenzoyl)-8-methyl-10-phenyl-5,10-dihydro-6*H*-benzo[*h*]pyrazolo[3,4*b*][1,6]naphthyridin-6-one (5d)

Yellow powder; mp: 298-299 °C; yield: 75% (336 mg). IR (KBr) v_{max} : 3313, 3178, 3032, 2978, 2856, 1664, 1564, 1502, 1430, 1350, 1232, 1151, 1025, 853, 760 cm⁻¹. ¹H-NMR (300 MHz, DMSO-*d*₆) δ : 11.73 (s, exchanged by D₂O addition, 1H, NH), 8.75 (d, *J* = 6.9 Hz, 1H, Ar), 8.39 (d, *J* = 6.6 Hz, 2H, Ar), 7.90 (bs, 2H, Ar), 7.65 (bs, 3H, Ar), 7.45-7.30 (m, 5H, Ar), 2.20 (s, 3H, CH₃) ppm. ¹³C-NMR (75.5 MHz, DMSO-*d*₆) δ : 192.5, 160.9, 151.9, 151.6, 145.8, 143.5, 139.0, 133.9, 133.2, 131.0, 129.8, 128.7, 122.1, 121.9, 120.4, 120.0, 119.2, 116.6, 114.6, 114.3, 14.2 ppm. Anal. Calcd for C₂₇H₁₇FN₄O₂: C, 72.31; H, 3.82; N, 12.49. Found: C, 72.40; H, 3.72; N, 12.52.

8-Methyl-7-(4-methylbenzoyl)-10-phenyl-5,10-dihydro-6*H*-benzo[*h*]pyrazolo[3,4*b*][1,6]naphthyridin-6-one (5e)

White powder; mp: 304-306 °C; yield: 77% (342 mg). IR (KBr) v_{max} : 3327, 3037, 2982, 2861, 1666, 1568, 1499, 1428, 1351, 1262, 1170, 1124, 1027, 760 cm⁻¹. ¹H-NMR (300 MHz, DMSOd₆) δ : 11.69 (s, exchanged by D₂O addition, 1H, NH), 8.75 (d, J = 7.8 Hz, 1H, Ar), 8.39 (d, J = 8.1 Hz, 2H, Ar), 7.72-7.60 (m, 5H, Ar), 7.41 (d, J = 7.8 Hz, 2H, Ar), 7.40 (t, J = 6.9 Hz, 1H, Ar), 7.31 (d, J = 8.1 Hz, 2H, Ar), 2.38 (s, 3H, CH₃), 2.23 (s, 3H, CH₃) ppm. ¹³C-NMR (75.5 MHz, DMSO-d₆) δ : 193.5, 160.9, 151.9, 151.6, 146.5, 144.6, 143.6, 139.1, 139.1, 134.8, 130.6, 130.3, 129.8, 129.4, 129.1, 128.8, 122.0, 120.1, 119.2, 114.6, 114.4, 21.9, 14.2 ppm. Anal. Calcd for C₂₈H₂₀N₄O₂: C, 75.66; H, 4.54; N, 12.60. Found: C, 75.60; H, 4.62; N, 12.65.

7-(4-Methoxybenzoyl)-8-methyl-10-phenyl-5,10-dihydro-6*H*-benzo[*h*]pyrazolo[3,4*b*][1,6]naphthyridin-6-one (5f)

Yellow powder; mp: 303-304 °C; yield: 79% (364 mg). IR (KBr) v_{max} : 3411, 2980, 2857, 1670, 1566, 1502, 1358, 1255, 1168, 1023, 756 cm⁻¹. ¹H-NMR (300 MHz, DMSO-*d*₆) δ : 11.69 (s, exchanged by D₂O addition, 1H, NH), 8.75 (d, *J* = 7.8 Hz, 1H, Ar), 8.39 (d, *J* = 7.8 Hz, 2H, Ar), 7.76 (d, *J* = 8.4 Hz, 2H, Ar), 7.66 (bt, *J* = 7.8 Hz, 3H, Ar), 7.45-7.32 (m, 3H, Ar), 7.02 (d, *J* = 8.7 Hz, 2H, Ar), 3.83 (s, 3H, OCH₃), 2.20 (s, 3H, CH₃) ppm. ¹³C-NMR (75.5 MHz, DMSO-*d*₆) δ : 192.4, 163.9, 160.8, 151.9, 151.5, 146.7, 143.7, 139.1, 132.6, 131.4, 131.0, 130.3, 122.1, 121.9, 121.7, 120.1, 119.9, 119.2, 114.7, 114.5, 114.3, 56.6, 14.2 ppm. Anal. Calcd for C₂₈H₂₀N₄O₃: C, 73.01; H, 4.38; N, 12.17. Found: C, 73.10; H, 4.42; N, 12.12.

7-(3,4-Dimethoxybenzoyl)-8-methyl-10-phenyl-5,10-dihydro-6*H*-benzo[*h*]pyrazolo[3,4*b*][1,6]naphthyridin-6-one (5g)

Yellow powder; mp: 263-265 °C; yield: 81% (397 mg). IR (KBr) v_{max} : 3330, 2987, 2917, 1667, 1589, 1504, 1430, 1349, 1267, 1131, 1024, 758 cm⁻¹. ¹H-NMR (300 MHz, DMSO- d_6) δ : 11.66 (s, exchanged by D₂O addition, 1H, NH), 8.75 (d, J = 7.5 Hz, 1H, Ar), 8.40 (d, J = 7.8 Hz, 2H, Ar), 7.75-7.30 (m, 7H, Ar), 7.12 (bd, J = 6.9 Hz, 1H, Ar), 6.92 (d, J = 8.1 Hz, 1H, Ar), 3.85 (s, 3H, OCH₃), 3.81 (s, 3H, OCH₃), 2.21 (s, 3H, CH₃) ppm. ¹³C-NMR (75.5 MHz, DMSO- d_6) δ : 192.4, 163.8, 160.8, 154.0, 151.9, 151.5, 149.5, 146.7, 143.7, 139.2, 131.0, 130.3, 129.8, 128.6, 125.2, 121.9, 120.1, 119.2, 114.5, 56.8, 56.3, 14.2 ppm. Anal. Calcd for C₂₉H₂₂N₄O₄: C, 71.01; H, 4.52; N, 11.42. Found: C, 71.10; H, 4.52; N, 11.38.

8-Methyl-7-(4-nitrobenzoyl)-10-phenyl-5,10-dihydro-6*H*-benzo[*h*]pyrazolo[3,4*b*][1,6]naphthyridin-6-one (5h)

Yellow powder; mp: 298-300 °C; yield: 65% (309 mg). IR (KBr) v_{max} : 3366, 3040, 2993, 2916, 1663, 1569, 1498, 1433, 1346, 1225, 756 cm⁻¹. ¹H-NMR (300 MHz, DMSO-*d*₆) δ : 11.81 (s, exchanged by D₂O addition, 1H, NH), 8.76 (d, *J* = 7.8 Hz, 1H, Ar), 8.40 (d, *J* = 7.8 Hz, 2H, Ar), 8.30 (d, *J* = 7.8 Hz, 2H, Ar), 8.08 (d, *J* = 8.4 Hz, 2H, Ar), 7.67 (t, *J* = 7.2 Hz, 3H, Ar), 7.45-7.30

(m, 3H, Ar), 2.23 (s, 3H, CH₃) ppm. ¹³C-NMR (75.5 MHz, DMSO- d_6) δ : 192.8, 161.1, 151.9, 151.7, 150.6, 144.7, 143.4, 141.3, 139.0, 139.0, 131.1, 129.9, 129.6, 125.7, 125.5, 123.7, 122.0, 120.2, 119.3, 114.8, 114.3, 14.3 ppm. Anal. Calcd for C₂₇H₁₇N₅O₄: C, 68.21; H, 3.60; N, 14.73. Found: C, 68.15; H, 3.65; N, 14.62.

7-Benzoyl-8-methyl-10-phenyl-5*H*-benzo[*h*]pyrazolo[3,4-*b*]quinoline-5,6(10H)-dione (6a)

Orange powder; mp: 283-286 °C; yield: 89% (396 mg). IR (KBr) v_{max} : 3064, 1675, 1566, 1446, 1282, 1221, 753, 383 cm⁻¹. ¹H-NMR (300 MHz, DMSO- d_6) δ : 8.71 (d, J = 7.5 Hz, 1H, Ar), 8.27 (d, J = 7.8 Hz, 2H, Ar), 8.06 (d, J = 7.2 Hz, 1H, Ar), 7.92-7.73 (m, 3H, Ar), 7.70 (t, J = 7.5 Hz, 2H, Ar), 7.65 (t, J = 7.8 Hz, 2H, Ar), 7.54 (t, J = 7.5 Hz, 2H, Ar), 7.42 (t, J = 7.5 Hz, 1H, Ar), 2.13 (s, 3H, CH₃) ppm. ¹³C-NMR (75.5 MHz, DMSO- d_6) δ : 194.2, 178.2, 177.5, 162.3, 153.8, 151.0, 147.2, 144.4, 138.6, 136.4, 135.9, 135.7, 134.7, 133.2, 132.1, 129.9, 129.5, 129.3, 128.8, 127.2, 121.4, 120.7, 113.9, 13.8 ppm. Anal. Calcd for C₂₈H₁₇N₃O₃: C, 75.84; H, 3.86; N, 9.48. Found: C, 75.75; H, 3.82; N, 9.35.

7-(4-Chlorobenzoyl)-8-methyl-10-phenyl-5*H*-benzo[*h*]pyrazolo[3,4-*b*]quinoline-5,6(10*H*)dione (6b)

Yellow powder; mp: 279-282 °C; yield: 85% (391 mg). IR (KBr) v_{max} : 3078, 1673, 1565, 1493, 1426, 1283, 1220, 1090, 759 cm⁻¹. ¹H-NMR (300 MHz, DMSO-*d*₆) δ : 8.71 (d, *J* = 8.1 Hz, 1H, Ar), 8.28 (d, *J* = 7.8 Hz, 2H, Ar), 8.07 (d, *J* = 7.5 Hz, 1H, Ar), 7.99 (t, *J* = 7.8 Hz, 1H, Ar), 7.91 (d, *J* = 7.8 Hz, 2H, Ar), 7.71 (t, *J* = 7.8 Hz, 1H, Ar), 7.65 (t, *J* = 7.5 Hz, 2H, Ar), 7.60 (d, *J* = 8.1 Hz, 2H, Ar), 7.43 (t, *J* = 6.9 Hz, 1H, Ar), 2.15 (s, 3H, CH₃) ppm. Anal. Calcd for C₂₈H₁₆ClN₃O₃: C, 70.37; H, 3.37; N, 8.79. Found: C, 70.30; H, 3.32; N, 8.85.

7-(4-Fluorobenzoyl)-8-methyl-10-phenyl-5*H*-benzo[*h*]pyrazolo[3,4-*b*]quinoline-5,6(10*H*)dione (6c)

Yellow powder; mp: 272-274 °C; yield: 88% (408 mg). IR (KBr) v_{max} : 3076, 2926, 1680, 1576, 1498, 1426, 1285, 1230, 1156, 761 cm⁻¹. ¹H-NMR (300 MHz, DMSO- d_6) δ : 8.72 (d, J = 7.8 Hz, 1H, Ar), 8.28 (d, J = 8.4 Hz, 2H, Ar), 8.07 (d, J = 7.5 Hz, 1H, Ar), 7.99 (t, J = 6.9 Hz, 1H, Ar), 7.90 (d, J = 7.8 Hz, 2H, Ar), 7.71 (t, J = 8.1 Hz, 1H, Ar), 7.66 (t, J = 8.1 Hz, 2H, Ar), 7.44 (d, J =

7.5 Hz, 2H, Ar), 7.36 (t, J = 8.4 Hz, 1H, Ar), 2.15 (s, 3H, CH₃) ppm. Anal. Calcd for $C_{28}H_{16}FN_3O_3$: C, 72.88; H, 3.50; N, 9.11. Found: C, 72.75; H, 3.60; N, 9.15.

7-(4-Methoxybenzoyl)-8-methyl-10-phenyl-5*H*-benzo[*h*]pyrazolo[3,4-*b*]quinoline-5,6(10*H*)dione (6d)

Yellow powder; mp: 284-286 °C; yield: 90% (428 mg). IR (KBr) v_{max} : 3068, 2932, 1672, 1586, 1497, 1450, 1266, 1162, 762 cm⁻¹. ¹H-NMR (300 MHz, DMSO- d_6) δ : 8.71 (d, J = 7.8 Hz, 1H, Ar), 8.27 (d, J = 7.8 Hz, 2H, Ar), 8.05 (t, J = 7.5 Hz, 1H, Ar), 7.92 (d, J = 7.5 Hz, 1H, Ar), 7.85 (t, J = 7.8 Hz, 2H, Ar), 7.70 (t, J = 7.5 Hz, 1H, Ar), 7.65 (t, J = 7.5 Hz, 2H, Ar), 7.42 (d, J = 7.2 Hz, 1H, Ar), 7.04 (d, J = 9 Hz, 2H, Ar), 3.84 (s, 3H, OCH₃), 2.14 (s, 3H, CH₃) ppm. ¹³C-NMR (75.5 MHz, DMSO- d_6) δ : 192.5, 178.3, 177.5, 164.3, 159.2, 158.1, 153.8, 147.5, 144.6, 138.6, 137.6, 136.5, 135.7, 133.1, 132.0, 131.7, 129.9, 129.0, 128.7, 127.1, 121.4, 120.5, 114.8, 56.1, 13.8 ppm. Anal. Calcd for C₂₉H₁₉N₃O₄: C, 73.56; H, 4.04; N, 8.87. Found: C, 73.65; H, 4.02; N, 8.80.

8-Methyl-7-(4-nitrobenzoyl)-10-phenyl-5*H*-benzo[*h*]pyrazolo[3,4-*b*]quinoline-5,6(10*H*)dione (6e)

Yellow powder; mp: 275-277 °C; yield: 93% (456 mg). IR (KBr) v_{max} : 3071, 2926, 1678, 1530, 1431, 1346, 1284, 1218, 757, 687 cm⁻¹. ¹H-NMR (300 MHz, DMSO- d_6) δ : 8.74 (d, J = 6 Hz, 1H, Ar), 8.32 (d, J = 9 Hz, 2H, Ar), 8.29 (d, J = 8.1 Hz, 2H, Ar), 8.15 (d, J = 8.7 Hz, 2H, Ar), 8.08 (d, J = 8.4 Hz, 1H, Ar), 7.94 (bt, J = 8.4 Hz, 1H, Ar), 7.72 (t, J = 7.5 Hz, 1H, Ar), 7.67 (t, J = 7.5 Hz, 2H, Ar), 7.45 (bt, J = 7.8 Hz, 1H, Ar), 2.17 (s, 3H, CH₃) ppm. Anal. Calcd for C₂₈H₁₆N₄O₅: C, 68.85; H, 3.30; N, 11.47. Found: C, 68.93; H, 3.32; N, 11.52.

7-(4-Chlorobenzoyl)-10-(3-chlorophenyl)-8-methyl-5*H*-benzo[*h*]pyrazolo[3,4-*b*]quinoline-5,6(10*H*)-dione (6f)

Yellow powder; mp: 295-298 °C; yield: 89% (458 mg). IR (KBr) v_{max} : 3081, 2926, 1678, 1575, 1477, 1455, 1277, 1223, 1162, 1091, 772 cm⁻¹. ¹H-NMR (300 MHz, DMSO- d_6) δ : 8.62 (d, J = 7.8 Hz, 1H, Ar), 8.32 (t, J = 8.7 Hz, 1H, Ar), 8.24 (d, J = 7.8 Hz, 1H, Ar), 8.05 (d, J = 7.8 Hz, 1H, Ar), 7.89 (bd, J = 7.8 Hz, 2H, Ar), 7.70 (t, J = 7.5 Hz, 1H, Ar), 7.64 (d, J = 8.4 Hz, 1H, Ar), 7.60 (d, J = 8.4 Hz, 2H, Ar), 7.43 (bd, J = 8.1 Hz, 1H, Ar), 2.12 (s,

3H, CH₃) ppm. ¹³C-NMR (75.5 MHz, DMSO- d_6) δ : 192.9, 177.9, 177.4, 162.3, 153.9, 151.2, 146.3, 145.0, 139.7, 136.1, 135.6, 134.5, 134.1, 133.1, 132.2, 131.6, 131.1, 129.8, 129.7, 128.8, 126.6, 126.4, 120.9, 120.5, 114.2, 13.8 ppm. Anal. Calcd for C₂₈H₁₅Cl₂N₃O₃: C, 65.64; H, 2.95; N, 8.20. Found: C, 65.68; H, 2.90; N, 8.25.

10-(3-Chlorophenyl)-7-(4-fluorobenzoyl)-8-methyl-5*H*-benzo[*h*]pyrazolo[3,4-*b*]quinoline-5,6(10*H*)-dione (6g)

Yellow powder; mp: 288-290 °C; yield: 88% (438 mg). IR (KBr) *v*_{max}: 3109, 2927, 1675, 1564, 1477, 1443, 1279, 1226, 774 cm⁻¹. Anal. Calcd for C₂₈H₁₅ClFN₃O₃: C, 67.82; H, 3.05; N, 8.47. Found: C, 67.87; H, 3.02; N, 7.35.

10-(3-Chlorophenyl)-7-(4-methoxybenzoyl)-8-methyl-5*H*-benzo[*h*]pyrazolo[3,4-*b*]quinoline-5,6(10*H*)-dione (6h)

Yellow powder; mp: 263-265 °C; yield: 93% (474 mg). IR (KBr) v_{max} : 3058, 2929, 1670, 1583, 1448, 1258, 1165, 1022, 770 cm⁻¹. ¹H-NMR (300 MHz, DMSO-*d*₆) δ : 8.63 (d, *J* = 7.8 Hz, 1H, Ar), 8.32 (d, *J* = 8.1 Hz, 1H, Ar), 8.29 (s, 1H, Ar), 8.04 (d, *J* = 7.8 Hz, 1H, Ar), 7.90 (t, *J* = 7.5 Hz, 1H, Ar), 7.84 (d, *J* = 8.7 Hz, 2H, Ar), 7.69 (t, *J* = 7.5 Hz, 1H, Ar), 7.64 (t, *J* = 8.1 Hz, 1H, Ar), 7.43 (d, *J* = 8.7 Hz, 1H, Ar), 7.03 (d, *J* = 8.7 Hz, 2H, Ar), 3.32 (s, 3H, OCH₃), 2.11 (s, 3H, CH₃) ppm. ¹³C-NMR (75.5 MHz, DMSO-*d*₆) δ : 192.3, 178.1, 177.4, 164.4, 162.3, 153.9, 151.2, 147.5, 145.1, 139.7, 136.3, 135.7, 134.1, 133.1, 132.1, 131.6, 129.0, 128.8, 126.6, 126.4, 120.6, 120.4, 119.2, 114.8, 114.3, 56.1, 13.7 ppm. Anal. Calcd for C₂₉H₁₈ClN₃O₄: C, 68.58; H, 3.57; N, 8.27. Found: C, 68.65; H, 3.62; N, 8.25.

10-(3-Chlorophenyl)-8-methyl-7-(4-nitrobenzoyl)-5*H*-benzo[*h*]pyrazolo[3,4-*b*]quinoline-5,6(10*H*)-dione (6i)

Yellow powder; mp: 248-251 °C; yield: 87% (457 mg). IR (KBr) v_{max} : 3093, 1675, 1571, 1531, 1452, 1344, 1218, 771 cm⁻¹. ¹H-NMR (300 MHz, DMSO- d_6) δ : 8.67 (d, J = 7.8 Hz, 1H, Ar), 8.63 (s, 1H, Ar), 8.32 (bd, J = 8.4 Hz, 3H, Ar), 8.14 (t, J = 8.7 Hz, 2H, Ar), 7.93 (t, J = 8.1 Hz, 1H, Ar), 7.92 (d, J = 7.8 Hz, 1H, Ar), 7.72 (t, J = 7.5 Hz, 1H, Ar), 7.67 (d, J = 7.8 Hz, 1H, Ar), 7.47 (d, J = 7.2 Hz, 1H, Ar), 2.15 (s, 3H, CH₃) ppm. ¹³C-NMR (75.5 MHz, DMSO- d_6) δ : 192.9, 177.8, 177.5, 162.3, 154.0, 151.0, 145.6, 144.9, 139.9, 136.0, 135.7, 134.1, 133.2, 132.3, 131.7,

130.6, 128.8, 126.8, 124.7, 121.1, 120.6, 120.6, 119.3, 114.1, 13.91 ppm. Anal. Calcd for C₂₈H₁₅ClN₄O₅: C, 64.32; H, 2.89; N, 10.71. Found: C, 64.29; H, 2.82; N, 10.75.

The compounds **6b**, **6c**, **6e** and **6g** did not show either ¹H-NMR and ¹³C-NMR or both due to very low solubilities in DMSO- d_6 .

ACKNOWLEDGEMENTS

The authors gratefully acknowledge the financial assistance from Urmia University.

REFERENCES

- [1] Y. Hao, X.-P. Xu, T. Chen, L.-L. Zhao, S.-J. Ji, Org. Biomol. Chem. 2012, 10, 724.
- [2] M. W. Embrey, J. S. Wai, T. W. Funk, C. F. Homnick, D. S. Perlow, S. D. Young, J. P. Vacca, D. J. Hazuda, P. J. Felock, K. A. Stillmoc, M. V. Witmer, G. Moyer, W. A. Schleif, L. J. Gabryelski, L. Jin, I. Chen, J. D. Ellis, B. K. Wong, J. H. Lin, Y. M. Leonard, N. N. Tsou, L. Zhuang, *Bioorg. Med. Chem. Lett.* **2005**, *15*, 4550.
- [3] X.-J. Tu, W.-J. Hao, Q. Ye, S.-S. Wang, B. Jiang, G. Li, S.-J. Tu, J. Org. Chem. 2014, 79, 11110.
- [4] R. R. Crenshaw, G. M. Luke, P. Smirnoff, J. Med. Chem. 1976, 19, 262.
- [5] S. Radl, V. Zikan, F. Smejkal, Collect. Czech. Chem. Commun. 1985, 50, 1057.
- [6] S. T. Selvi, V. Nadaraj, S. Mohan, R. Sasi, M. Hema, *Bioorg. Med. Chem.* 2006, 14, 3896.
- [7] J. M. Arif, M. Kunhi, M. P. Subramanian, A. A. Bekhit, O. A. El-Sayed, K. Al-Hussein,
 H. Y. Aboul-Enein, F. M. Al-Khodairy, *Int. J. Biomed Sci.* 2007, *3*, 194.
- [8] S. F. Thakor, D. M. Patel, M. P. Patel, R. G. Patel, Saudi. Pharma. J. 2007, 15, 48.
- [9] A. Domling, W. Wang, K. Wang, *Chem. Rev.* **2012**, *112*, 3083.
- [10] A. Domling, Chem. Rev. 2006, 106, 17.
- [11] J. Zhu, H. Bienayme, *Multicomponent Reactions*, Wiley-VCH, Weinheim, 2005.
- [12] D. Tejedor, F. Garcia-Tellado, Chem. Soc. Rev. 2007, 36, 484.
- [13] D. M. D'Souza, T. J. J. Mueller, Chem. Soc. Rev. 2007, 36, 1095.
- [14] B. B. Toure, D. G. Hall, Chem. Rev. 2009, 109, 4439.

- [15] N. R. Candeias, F. Montalbano, P. M. S. D. Cal, P. M. P. Gois, *Chem. Rev.* 2010, *110*, 6169.
- [16] R. Cioc, E. Ruijter, R. Orru, Green Chem. 2014, 16, 2958.
- [17] X.-Y. Dong, Z.-W. Gao, K.-F. Yang, W.-Q. Zhang, L.-W. Xu, Catal. Sci. Technol. 2015, 5, 2554.
- [18] B. Eftekhari-Sis, M. Zirak, A. Akbari, Chem. Rev. 2013, 113, 2958.
- [19] A. Nouri, A. Poursattar Marjani, J. Khalafy, J. Heterocycl. Chem. 2019, 56, 2912.
- [20] A. Nouri, A. Poursattar Marjani, J. Khalafy, N. Etivand, *Res. Chem. Intermed.* 2020, 46, 3025.
- [21] R. Javahershenas, F. Majidi Arlan, R. H. Prager, J. Khalafy, ARKIVOC 2020, i, 117.
- [22] J. Khalafy, F. Majidi Arlan, S. Soleimani Chalanchi, J. Heterocycl. Chem. 2018, 55, 149.
- [23] N. Etivand, J. Khalafy, A. Poursattar Marjani, Res. Chem. Intermed. 2019, 45, 3379.
- [24] A. Poursattar Marjani, J. Khalafy, S. Akbarzadeh, S. Afri. J. Chem. Soc. 2019, 72, 160.
- [25] A. Poursattar Marjani, J. Khalafy, P. Eslamipour, M. Ahmadi Sabegh, Iran J. Chem. Chem. Eng. 2019, 38, 51.
- [26] A. Poursattar Marjani, J. Khalafy, F. Majidi Arlan, E. Eyni, ARKIVOC, 2019, v, 1.
- [27] A. Poursattar Marjani, J. Khalafy, A. Farajollahi, J. Heterocycl. Chem. 2019, 56, 268.
- [28] N. Etivand, J. Khalafy, M. G. Dekamin, Synthesis 2020, 52, 1707.
- [29] M. Aslanpanjeh, A. Poursattar Marjani, J. Khalafy, N. Etivand, *Res. Chem. Intermed.* 2020, 46, 165.
- [30] F. Majidi Arlan, J. Khalafy, R. Maleki, Chem. Heterocycl. Compd. 2018, 54, 51.
- [31] M. A. Karimi, M. A. Mozaheb, A. Hate-Mehrjardi, H. Tavallali, A. M. Attaran, G. Deilamy-Rad, *Scientia Iranica*, 2015, 22, 2736.
- [32] T. Dutta, N. N. Ghosh, M. Das, R. Adhikary, V. Mandal, A. P. Chattopadhyay, J. Environ. Chem. Eng. 2020, 8, 104019.