



Article

Formation of Methylene Linkage for N-Heterocycles: Sequential C-H and C-O Bond Functionalization of Methanol with Cosolvent Water

Na Li, Jinku Bai, Xiaolin Zheng, and Honghua Rao

J. Org. Chem., Just Accepted Manuscript • DOI: 10.1021/acs.joc.9b00729 • Publication Date (Web): 15 May 2019

Downloaded from http://pubs.acs.org on May 15, 2019

Just Accepted

"Just Accepted" manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides "Just Accepted" as a service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. "Just Accepted" manuscripts appear in full in PDF format accompanied by an HTML abstract. "Just Accepted" manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are citable by the Digital Object Identifier (DOI®). "Just Accepted" is an optional service offered to authors. Therefore, the "Just Accepted" Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the "Just Accepted" Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these "Just Accepted" manuscripts.



Formation of Methylene Linkage for *N*-Heterocycles: Sequential C–H and C–O Bond Functionalization of Methanol with Cosolvent Water

Na Li,^{+,I} Jinku Bai,^{+,I} Xiaolin Zheng⁺ and Honghua Rao*,^{+,‡}

[†]Department of Chemistry, Capital Normal University, Beijing, 100048, P. R. China

*Key Laboratory of Bioorganic Phosphorus Chemistry and Chemical Biology (Ministry of Education), Department of Chemistry, Tsinghua University, Beijing 100084, P. R. China

Supporting Information Placeholder

TOC:

Sequential C-H/C-O functionalization of methanol with Fe-catalysis
 Easy access to pharmaceutically important methylene heterodiment

ABSTRACT: An iron-catalyzed methylene forming strategy is disclosed through sequential C–H and C–O bond functionalization of methanol with cosolvent water. This protocol provides an easy and novel access to methylene tethered imidazo[1,2-a]pyridine and 2-aminopyridine analogues in a sustainable manner, and represents a complementary approach to traditional methylene forming strategies.

INTRODUCTION

Nitrogen-containing heterocycles are ubiquitous structural motifs of numerous natural products and pharmaceuticals, thus their syntheses and late-stage functionalizations have become an important focus for organic and medicinal chemists.² Specifically, use of covalent linkage to tether stable and highly accessible Nheterocycles has substantially expanded access to diverse pharmaceutically valuable molecules.³ For instance, some welldeveloped drugs such as esomeprazole,4 alosetron,5 baquiloprim,6 and motesanib7 were found in the series of methylene-tethered Nheterocycles (e.g., imidazole and 2-aminopyridine analogues) (Figure 1). Therefore, various methods have been developed to construct the methylene-linked N-heterocycles, one of which prepositiones methylene linker on either N-heterocycle, thus suffers from operational complexity, poor step- and atomeconomy.8 Another noteworthy strategy by using small molecules to serve as the methylene synthon has been actively pursued, which involves direct metalation of C-H bonds on either N-heterocycle. With this strategy, the commonly used solvent N,Ndimethylacetamide has been successfully explored to synthesize methylene bisimidazo[1,2-a]pyridines.9 Furthermore, feedstocks such as formaldehyde (via nucleophilic addition and subsequent dehydration reaction)10 or the toxic and explosive diazomethane (via transition metal such as Pd-catalyzed migratory insertion reaction¹¹) are capable of installing the methylene linkage for limited N-heterocycles. However, as a simple and easilyhandling C1-precursor, methanol is not frequently used as the methylene source thus far.

Recently, as an emerging field in C-H functionalizations, cross-

Figure 1. Some pharmaceuticals with methylene-tethered heterocyclic pairs.

dehydrogenative coupling (CDC) between C-H bonds (especially the intrinsically inert sp3 C-H bond in small molecules) has become a highly attractive yet challenging C-C bond-forming strategy due to its step- and atom-economic features.¹² One typical example is the CDC reaction between methanol C-H and pyridyl C-H bond (also known as the Minisci reaction¹³), which has provided a useful means to forge C-C bonds. This protocol has captured considerable attentions from synthetic chemists, although more catalytic systems should be exploited to showcase its generality (e.g., CDC coupling of methanol C-H bond with electron-sufficient substrates) and practicality (e.g., CDC coupling of methanol C-H bond under milder reaction conditions, cheaper catalytic systems, greener reaction mediums).¹⁴ Moreover, to expand the utilization of CDC reaction with methanol, various methods can be applied in late-stage functionalization of the hydroxymethyl products (mainly resulting involving transformations of the hydoxyl group). Apart from traditional functional group transformation strategies, 15 an appealing approach is the direct amination/amidation of hydroxyl group through a hydrogen-borrowing process¹⁶ via Ir-,¹⁷ Ru-,¹⁸ or other transitionmetal-catalysis.¹⁹ Given the state of art, we questioned that whether a general and sustainable protocol could be devised with methanol as the methylene synthon for selective linkage of *N*-heterocycles through a consequtive C–H and C–O bond functionalization of methanol. Herein, we present an iron-catalyzed methylene-forming strategy from methanol with cosolvent water, which can be applied to tether imidazo[1,2-a]pyridine and 2-aminopyridine via sequential C–C and C–N bond formation (Scheme 1).

Scheme 1. Iron-catalyzed methylene formation from methanol (this work)

RESULTS AND DISCUSSION

Our efforts on developing this methylene-forming strategy from methanol began with the evaluation of the reaction with 2phenylimidazo[1,2-a]pyridine (1a) and 2-aminopyridine (2a) (Table 1). We found that the treatment of **1a** and **2a** with catalyst FeCp₂ (25 mol %) and oxidant TBHP (70% aqueous solution, 2.5 equiv) in methanol at 100 °C under an argon atmosphere for 24 h gave the desired methylene-linked heterodimer 3aa (confirmed by X-ray diffraction; please refer to pS1 in the Supporting Information) in 44% yield (Table 1, entry 1). Fortunately, a marked increase in efficiency was observed when the model reaction was performed in MeOH/ H_2O (7:3 v/v ratio), affording **3aa** in 75% yield (entry 2). Seeking to improve on this result, FeCp*2, dppf, FeCl2 and FeCl3 were examined, among which FeCp2 afforded 3aa in the highest 72% yield (entries 2-6). Remarkably, it was found that oxidant significantly influenced the efficiency of the reaction: TBHP (5.5 M in decane) decreased the reaction efficiency slightly, while DTBP, CHP or DCP reduced the yield dramatically (entries 7-10); and surprisingly, (BzO)₂ or BzOOBu^t completely inhibited the reaction (entries 11 and 12). Meanwhile, changing the volume ratio of MeOH to H2O was found to be detrimental to this reaction (entries 13 and 14). Afterwards, a variety of N,N-, P,P- and P,Nbidentate ligands were tested (entries 15-22). Nevertheless, a further improvement was observed when using 2,2'-bipyridine (L_1) as the ligand (entry 15, 80% yield). More endeavors to increase the reaction efficiency were also attempted. For instance, screening of reaction temperature and time achieved a superior result (85% yield) at 90 °C for 30 h (entries 23 and 24), while performing the reaction under an air atmosphere did not (entry 25). Notably, no or only 59% of 3aa was produced when running the reaction in the absence of oxidant or catalyst (entries 26 and 27), thus demonstrating that the combination of catalyst FeCp2 and oxidant TBHP is crucial for this transformation.

With the optimized reaction conditions in hand, we began an exploration of the scope of 2-aminopyridine component for this methylene-forming protocol. As shown in Scheme 2, electronneutral and electron-rich 2-aminopyridines bearing alkyl and methoxyl functionality may be employed in good to excellent yields (3aa-ae, 62-82% yield; CD₂-3aa, 53% yield), while electrondeficient ones containing fluoro, chloro, bromo groups readily reacted with 1a to afford the desired products in good yields (3af-aj, 53-70% yield). The tolerance of these functionalities with coupling capability potentially allows for late-stage functionalization of

Table 1. Screening of the reaction conditions^a

entry	Catalyst	Oxidant	Solvent	Ligand	yield (%) ^b
1	FeCp ₂	TBHPaq	MeOH	-	44
2	FeCp ₂	TBHPaq	MeOH/H ₂ O (7:3)	-	75
3	FeCp*2	TBHPaq	MeOH/H ₂ O (7:3)	-	58
4	dppf	TBHPaq	MeOH/H ₂ O (7:3)	-	48
5 ^c	FeCl ₂	TBHPaq	MeOH/H ₂ O (7:3)	-	58
6	FeCl ₃	TBHPaq	MeOH/H ₂ O (7:3)	-	65
7	FeCp ₂	TBHPdec	MeOH/H ₂ O (7:3)	-	72
8	FeCp ₂	DTBP	MeOH/H ₂ O (7:3)	-	24
9	FeCp ₂	CHP	MeOH/H ₂ O (7:3)	-	47
10	FeCp ₂	DCP	MeOH/H ₂ O (7:3)	-	31
11	FeCp ₂	$(BzO)_2$	MeOH/H ₂ O (7:3)	-	-
12	FeCp ₂	$BzOOBu^t$	MeOH/H ₂ O (7:3)	-	-
13	FeCp ₂	TBHPaq	MeOH/H ₂ O (8:2)	-	58
14	FeCp ₂	TBHPaq	MeOH/H ₂ O (6:4)	-	59
15	FeCp ₂	TBHPaq	MeOH/H ₂ O (7:3)	L ₁	80
16	FeCp ₂	TBHPaq	MeOH/H ₂ O (7:3)	L_2	70
17	FeCp ₂	TBHPaq	MeOH/H ₂ O (7:3)	L ₃	71
18	FeCp ₂	TBHPaq	MeOH/H ₂ O (7:3)	L_4	42
19	FeCp ₂	TBHPaq	MeOH/H ₂ O (7:3)	L_5	43
20	FeCp ₂	TBHPaq	MeOH/H ₂ O (7:3)	L ₆	76
21	FeCp ₂	TBHPaq	MeOH/H ₂ O (7:3)	L ₇	76
22	FeCp ₂	TBHPaq	MeOH/H ₂ O (7:3)	L ₈	42
23 ^c	$FeCp_2$	TBHPaq	MeOH/H ₂ O (7:3)	L ₁	80
24 ^d	FeCp ₂	TBHPaq	MeOH/H ₂ O (7:3)	L ₁	85
25 ^e	FeCp ₂	TBHPaq	MeOH/H ₂ O (7:3)	L ₁	72
26 ^f	FeCp ₂	-	MeOH/H ₂ O (7:3)	L ₁	-
27	-	TBHPaq	MeOH/H ₂ O (7:3)	L ₁	59

"Reaction conditions: **1a** (0.2 mmol), **2a** (0.6 mmol), Catalyst (25 mol %), Oxidant (2.5 equiv), Solvent (2.0 mL), Ligand (25 mol %), 100 °C, 24 h, under argon. ^bYields determined by ¹H NMR analysis of the crude reaction mixture using mesitylene as an internal standard. Reaction was conducted at 90 °C. ^dReaction was carried out for 30 h. ^cReaction was conducted under air. Cp = cyclopentadienyl, Cp* = pentamethylcyclopentadienyl, TBHP^{aq} = tert-butyl hydroperoxide (70% aqeous solution), TBHP^{dec} = *tert*-butyl hydroperoxide (5.5 M in decane), DTBP = *tert*-butyl peroxide, CHP = cumene hydroperoxide (techanicial grade, 80%), DCP = dicumyl peroxide, $L_1 = 2,2'$ -bipyridine, $L_2 = 4,4'$ -dimethyl-2,2'-bipyridine, $L_3 = 2,2'$ -biquinoline, L_4 = bathophenanthroline, $L_5 = 1,10$ -phenanthroline, $L_6 = DavePhos$, $L_7 = 1,4$ -bis(diphenylphosphino)-butane, $L_8 = N_1N_1N'_1N'_1N'_2$ -tetramethylethylenediamine.

the products. Besides, cyano and ester groups were also tolerated to provide the desired products in useful yields (3ak and 3al, 30 and 42% yield). Notably, the efficiency of the reaction was impeded by ortho substituent (to amino group on pyridyl unit) (cf. 3am and 3ab/3ac, 52% versus 72%/62% yield). And 2-aminopyridine with polysubstituents succeeded in generating the desired products in good yields (3an and 3ao, 64 and 53% yield). Additionally, we were pleased to find that the extended aromatic system (e.g., 2-aminoquinoline) and pharmaceutically relevant heterocycles (e.g., pyrimidyl) were competent coupling partners to provide an encouraging quantity of the desired products (3ap and 3aq, 35 and 35% yield).

Scheme 2. Scope for 2-aminopridine component

^aGeneral reaction conditions: **1a** (0.2 mmol), **2a-r** (0.6 mmol), FeCp₂ (0.05 mmol, 25 mol %), TBHP^{aq} (2.5 equiv), MeOH/H₂O (2.0 mL, 7:3 v/v ratio), **L**₁ (0.05 mmol, 25 mol %), 90 °C, 30 h, under argon atmosphere. Yields of isolated products are given.

Scheme 3. Scope for 2-arylimidazo 1,2-a pyridine component

^aGeneral reaction conditions: **1b-w** (0.2 mmol), **2a** (0.6 mmol), FeCp₂ (0.05 mmol, 25 mol %), TBHP^{aq} (2.5 equiv), MeOH/H₂O (2.0 mL, 7:3 v/v ratio), **L**₁ (0.05 mmol, 25 mol %), 90 °C, 30 h, under argon atmosphere. Yields of isolated products are given.

We next sought to explore the generality of this protocol with respect to 2-arylimidazo [1,2-a] pyridine component. As shown in Scheme 3, substrate containing either electron-donating (e.g., Me, Et; 3ba-ea, 41-65% yield) or -withdrawing groups (e.g., F, Cl and CO₂Me; **3fa-ja**, 41-78% yield) on pyridyl unit readily underwent the titled reaction in moderate to good yields. Furthermore, electron-donating groups on phenyl unit (e.g., Me, Pr and OMe; 3la-ma, 45-61% yield) generally displayed higher efficiency than electron-withdrawing groups (e.g., Cl, Br and CN; 3na-pa, 40-43% yield). The steric limit of substituent on phenyl unit was also investigated, but no detrimental effect on coupling efficiency was observed when 2'-methylated substrate 1q was employed (cf. 3ka and 3qa). Moreover, reactants bearing both substituent on pyridyl and phenyl unit were smoothly converted to the desired products in moderate to good yields (3ra-ta, 47-62% yield). And substrates with oxidation-vulnerable aromatic rings (e.g., thienyl and naphthyl; 3ua and 3va, 31 and 45% yield) were examined to evaluate the power of this methylene-forming protocol, and fortunately they were tolerated in this transformation, albeit in diminished yields. Finally, 2-tert-butylimidazo[1,2-a]pyridine was also found to be suitable for this transformation (3wa, 33% yield).

Scheme 4. Gram-scale synthesis of product 3aa.

As far as we know, the coupling products may bind or interact with the modularory sites on a large number of different gamma-aminobutryric acid (GABA) receptor complexes, indicating that there are excellent prospects for developing more selective drugs for the treatment of central nervous system (CNS) disorders. 20,21 We thereby evaluated the scalability of this methylene-forming protocol by performing reaction on a 6 mmol scale. The reaction of 2-phenylimidazo[1,2-a]pyridine (1a) and 2-aminopyridine (2a) in methanol smoothly furnished product 3aa in 73% yield without significant decrease in efficiency (versus 82% for the reaction on a 0.2 mmol scale for 3aa) (Scheme 4), thus showing its great potential in practical synthesis.

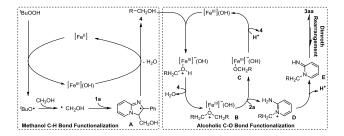
Further investigations were performed to gain some insights into the reaction mechanism (Scheme 5). We first conducted experiments to explore the existence of the radical intermediates. When a radical scavenger, (2,2,6,6-tetramethyl-piperidin-1-yl)oxyl (TEMPO) or butylated hydroxytoluene (BHT) was added into the reaction of 1a and 2a under standard conditions, nearly no desired product 3aa was observed (Scheme 5a), implying that radical intermediates are involved under this iron-catalysis. Additionally, in the absence of 2a, 1a under standard conditions could capture the radical intermediate to afford the hydroxymethylation product 4 in 79% yield, therefore suggesting the generation of a hydroxymethyl radical during the reaction (Scheme 5b).

Subsequently, exposure **4** and **2a** to the standard conditions afforded **3aa** in 86% yield, indicating that product **4** is most probably involved in the reaction (Scheme 5c). On the other hand, as benzyl alcohol analogues are prone to be transformed into aldehyde through transition-metal involved dehydrogenation²² or hydrogenborrowing process,¹⁶ product **4** was thereby subjected into the standard conditions; however, only trace amount of aldehyde **5** was

observed (Scheme 5d). Besides, the desired coupling reaction of 5 with 2a did not occur either, with more than 90% recovery of the starting material 5 (Scheme 5e). These results demonstrate that there is little chance to form the C–N bond if dehydrogenation or hydrogen-borrowing process of 4 is involved in the titled reaction. Furthermore, another pathway to 3aa involving the in-situ oxidation of methanol to formaldehyde and a subsequent Mannich-type reaction^{21a,23} of either 1a, 2a and formaldehyde or 1a and imine 6 (generated from condensation of formaldehyde and 2a) may be excluded as well (Scheme 5f-g).

Scheme 5. Investigation into the reaction mechanism

Scheme 6. Plausible mechanism



Based on these experimental observations, a plausible mechanism for this reaction is proposed in Scheme 6. Initiation occurs by reducing TBHP with FeCp₂ to generate Fe^{III}(OH)complex and the tert-butoxyl radical,24 which then abstracts a hydrogen from methanol to putatively afford the hydroxymethyl radical. This ensuing radical could be well-suited to add to 2phenylimidazo [1,2-a] pyridine **1a** to give radical A_1^{25} which is believed to undergo direct oxidation by Fe^{III}(OH)-complex and deprotonation to give product 4. Afterwards, product 4 may be rapidly converted to dimeric ether by eliminating water with the assistance of Fe^{III}(OH)-complex.²⁶ Presumably, the polarization of ether with $Fe^{III}(OH)$ -complex (by forming intermediate **B**) may generate an incipient allyl-type carbocation, which can result in a nucleophilic attack reaction in the presence of nucleophile 2a to provide intermediates C and D. 12i,26e Subsequently, deprotonation of D can generate intermediate E, followed by the Dimroth

rearrangment to afford the final product **3aa.**²⁷ Meanwhile, upon reaction of proton, **C** can release **4** and Fe^{III}(OH)-complex to accomplish the catalytic cylcle of alcoholic C–O bond functionalization.

CONCLUSION

In summary, we have reported an iron-catalyzed sequentional C-H/C-O bond functionalization of methanol with cosolvent water, generating the methylene linker to bridge pharmaceutically important 2-arylimidazo[1,2-a]pyridine and 2-aminopyridine fragments through a consecutive C-C and C-N bond-forming process. The power of this protocol has been fully exemplified by the substantial structural diversity of the resulting pharmaceutically important methylene heterodimers, as well as by the green reaction medium, mild reaction conditions, operational simplicity, and the numerous opportunities for late-stage functionalization of the resulting methylene heterodimers. Perhaps most important, this strategy employs methanol as the methylene synthon, thus representing an uncommon utilization of "methanol chemistry". Further exploration on tethering other heterocycles is currently underway in our laboratory. We expect this protocol to be widely adopted by the synthetic and drug discovery community as a sustainable and complementary platform for the construction of methylene heterodimers.

EXPERIMENTAL SECTION

General Methods. All reactions were carried out under an argon atmosphere unless otherwise stated. All work-up and purification procedures were carried out with reagent-grade solvents. Flash column chromatography was performed on silica gel (300-400 mesh) with an appropriate solvent system (see details below). Melting point was recorded on SGW X-4B. ¹H and ¹³C-NMR spectra were recorded on Varian 600 MHz and Bruker 400 MHz spectrometers in CDCl₃ or d_6 -DMSO solutions and chemical shifts (δ , ppm) were determined with internal solvent signal as reference (CDCl₃: 7.26 for 1 H-NMR and 77.0 for 13 C-NMR; d_{6} -DMSO: 2.50 for ¹H -NMR and 39.5 for ¹³C-NMR). NMR data are reported as following: chemical shift, multiplicity (s = singlet, d = doublet, dd = doublet of doublets, t = triplet, td = doublet of triplets, q = quartet, m = multiplet, br = broad signal), coupling constant (Hz), and integration. High-resolution mass spectra were recorded on Thermo Scientific LTQ Orbitrap Discovery (Bremen, Germany) (mass analyzer type: linear ion trap). Materials were purchased from Alfa-Aesar, Acros, Aldrich, Aladdin, Energy-Chemical, Bide Pharmatech. Ltd., and Ouhe-Chemicals. Unless otherwise noted, commercial reagents were used without further purifications. All compounds and their ¹H and ¹³C NMR spectra are provided in Supporting Information.

Preparation of imidazo[1,2-a]pyridine derivatives 1. **Method A**²⁸: To an ethanol solution containing 2-bromoacetophenones (5.024 mmol, 1.0 equiv) and 2-aminopyridines (6.280 mmol, 1.25 equiv) was added NaHCO₃ (7.837 mmol, 1.56 equiv). After The reaction mixture was stirred at room temperature for 6 hours. After completion of the reaction, the resulting mixture was diluted with water (15 ml) and extract with ether (3 × 20 ml). The combined organic layer was washed with brine (25 ml), dried with anhydrous MgSO₄, concentrated under vacuum to give the crude product, which was purified by silica gel

column with petroleum ether/ethyl acetate as the eluent to give the analytical pure 2-arylimidazo[1,2-*a*]pyridine **1a-m** and **1p-s** in 45-90% yields. The ¹H and ¹³C NMR data of **1a-m** and **1p-s** were in accordance with literature.²⁵

Method B²⁹: An clean oven-dried 10 mL round bottom vessel was charged with acetophenones (1.0 mmol), 2-amino pyridines (1.2 mmol), CuI (0.2 mmol) and 1,4-dioxane (3.0 mL). Then the vessel was sealed, placed into an oil bath and heated at 100 °C for 4 h. The resulting mixture was cooled to room temperature, filtered through a short silica gel pad and washed with ethyl acetate. The above solution was evaporated under vacuum, and the residue was purified by silica gel column with petroleum ether/ethyl acetate as the eluent to give the analytically pure product 2-phenylimidazo[1,2-a]pyridine **1n-o** and **1t-u** in 40-75% yield. The ¹H and ¹³C NMR data of **1n-o** and **1t-u** were in accordance with literature. ²⁵

Methyl 2-phenylimidazo[1,2-a]pyridine-6-carboxylate (1j). Prepared with **Method A**. ¹H NMR (600 MHz, CDCl₃) δ 8.89 (s, 1 H), 7.96 (d, J = 7.08 Hz, 2 H), 7.91 (s, 1 H), 7.72 (dd, J = 9.42, 1.74 Hz, 1 H), 7.62 (d, J = 9.48 Hz, 1 H), 7.45 (t, J = 7.59 Hz, 2 H), 7.36 (t, J = 7.35 Hz, 1 H), 3.95 (s, 3 H). ¹³C{¹H} NMR (150 MHz, CDCl₃) δ 165.3, 147.6, 146.1, 133.0, 129.7, 128.8, 128.5, 126.2, 124.4, 116.7, 116.5, 108.9, 52.4.

Typical procedures for the synthesis of methylene-tethered 2-arylimidazo[1,2-a]pyridine and **2-aminopyridine.** An ovendried round bottom reaction vessel was charged with imidazo[1,2-a]pyridine **1** (0.2 mmol), 2-aminopyridine **2** (0.6 mmol), 2,2'-bpy (7.8 mg, 0.05 mmol, 25 mol %), and FeCp₂ (9.3 mg, 0.05 mmol, 25.0 mol %). After the vessel was filled with argon, TBHP (70 wt. % in H₂O, 69.0 *u*L, 0.5 mmol), MeOH/H₂O (2.0 mL, 7:3 v/v ratio) was added by syringe under argon, and the reaction mixture was stirred at room temperature for 5 min. Then the vessel was sealed, placed into an oil bath and heated to 90 °C. After 30 h, the resulting mixture was cooled to room temperature, filtered through a short silica gel pad and washed with ethyl acetate. The above solution was concentraed under vacuum, and the residue was purified by silica gel column chromatography to give the analytically pure product.

N-((2-Phenylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3aa). Nhite solid, m.p. 173.7-174.4 °C; Isolated yield 82% (49.3 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. H NMR (600 MHz, CDCl₃) 8 8.16 (d, J = 6.12 Hz, 2 H), 7.77 (d, J = 6.60 Hz, 2 H), 7.62 (d, J = 9.00 Hz, 1 H), 7.42 (t, J = 7.65 Hz, 3 H), 7.35 (t, J = 7.41 Hz, 1 H), 7.20 (t, J = 7.89 Hz, 1 H), 6.78 (t, J = 6.81 Hz, 1 H), 6.65 (td, J = 6.15, 2.04 Hz, 1 H), 6.46 (d, J = 8.34 Hz, 1 H), 5.01 (d, J = 4.98 Hz, 2 H), 4.71 (s, 1 H). NMR (100 MHz, CDCl₃) 8 158.0, 147.8, 145.0, 144.5, 137.4, 133.9, 128.6, 128.4, 127.9, 124.8, 124.3, 117.4, 117.0, 113.5, 112.4, 108.4, 35.5. HR-ESI-MS [M+H]+ calcd for C₁9H₁7N₄ 301.1448; found 301.1446.

N-((2-Phenylimidazo[1,2-a]pyridin-3-yl)methyl-d₂)pyridin-2-amine (CD_2 -3aa). White solid, m.p. 170.3-171.4 °C. Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2; Isolated yield 53% (32.1 mg) 1 H NMR (600 MHz, CDCl₃) δ 8.18 (t, J = 6.66 Hz, 2 H), 7.78 (d, J = 7.56 Hz, 2 H), 7.64 (d, J = 9.00 Hz, 1 H), 7.45-7.41 (m, 3 H), 7.36 (t, J = 7.44 Hz, 1 H), 7.21 (td, J = 7.89, 1.26 Hz, 1 H), 6.80 (t, J = 6.75 Hz, 1 H), 6.66 (td, J = 6.15, 2.10 Hz, 1 H), 6.45 (d, J = 8.34 Hz, 1 H), 4.65 (s, 1 H). HR-ESI-

MS $[M+H]^+$ calcd for $C_{19}H_{15}N_4D_2$ 303.1573; found 303.1569.

4-Methyl-N-((2-phenylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3ab). ^{21b} Light brown solid, m.p. 165.1-165.8 °C; Isolated yield 72% (45.3 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) δ 8.20 (d, J = 6.84 Hz, 1 H), 8.01 (d, J = 5.28 Hz, 1 H), 7.79 (d, J = 6.66 Hz, 2 H), 7.65 (d, J = 9.06 Hz, 1 H), 7.46 (t, J = 7.65 Hz, 2 H), 7.38 (t, J = 7.41 Hz, 1 H), 7.22 (t, J = 7.89 Hz, 1 H), 6.81 (t, J = 6.60 Hz, 1 H), 6.50 (d, J = 3.90 Hz, 1 H), 6.24 (s, 1 H), 5.01 (d, J = 5.04 Hz, 2 H), 4.75 (br, s, 1 H), 2.20 (s, 3 H). ¹³C{ ¹H} NMR (100 MHz, CDCl₃) δ 157.6, 149.7, 145.8, 145.1, 144.5, 134.0, 128.7, 128.5, 128.0, 125.0, 124.4, 117.3, 116.7, 115.1, 112.5, 108.6, 35.5, 21.2. HR-ESI-MS [M+H]+ calcd for C₂₀H₁₉N₄ 315.1604; found 315.1596.

5-Methyl-N-((2-phenylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3ac). White solid, m.p. 190.4-191.2 °C; Isolated yield 62% (38.9)mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) δ 8.18 (d, I = 6.90 Hz, 1 H), 7.97 (d, I = 1.38 Hz, 1 H), $7.77 \text{ (dd, } J = 8.31, 1.50 \text{ Hz}, 2 \text{ H)}, 7.63 \text{ (d, } J = 9.06 \text{ Hz}, 1 \text{ H)}, 7.43 \text{ (t, } J = 9.06 \text{ Hz$ J = 7.68 Hz, 2 H), 7.35 (td, J = 7.41 Hz, 1 H), 7.26 (dd, J = 8.19, 2.40 Hz, 1 H), 7.21-7.19 (m, 1 H), 6.78 (t, *J* = 6.81, 1.20 Hz, 1 H), 6.39 (d, *J* = 8.40 Hz, 1 H), 4.98 (d, *J* = 5.22 Hz, 2 H), 4.65 (t, *J* = 5.25 Hz, 1 H), 2.21 (s, 3 H). ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 156.2, 146.9, 145.1, 144.5, 138.9, 134.0, 128.7, 128.5, 128.0, 124.9, 124.5, 122.5, 117.4, 117.2, 112.5, 108.1, 35.8, 17.5. HR-ESI-MS $[M+H]^+$ calcd for $C_{20}H_{19}N_4$ 315.1604; found 315.1600.

4-Ethyl-N-((2-phenylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3ad). Brown solid, m.p. 156.9-157.8 °C; Isolated yield 75% (49.4 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) δ 8.17 (d, J = 6.90 Hz, 1 H), 7.99 (d, J = 5.40 Hz, 1 H), 7.75 (d, J = 7.50 Hz, 2 H), 7.61 (d, J = 9.00 Hz, 1 H), 7.41 (t, J =7.56 Hz, 2 H), 7.34 (t, J = 7.41 Hz, 1 H), 7.18 (t, J = 7.83 Hz, 1 H),6.76 (t, J = 6.72 Hz, 1 H), 6.50 (d, J = 5.22 Hz, 1 H), 6.25 (s, 1 H),5.06 (br, s, 1 H), 4.97 (d, J = 4.68 Hz, 2 H), 2.47 (q, J = 7.64 Hz, 2 H), 1.15 (t, J = 7.62 Hz, 3 H). ${}^{13}C\{{}^{1}H\}$ NMR (100 MHz, CDCl₃) δ 158.1, 155.0, 145.95, 145.0, 144.4, 133.9, 128.6, 128.4, 127.9, 124.9, 124.4, 117.3, 116.9, 114.0, 112.4, 107.2, 35.6, 28.3, 14.1. HR-ESI-MS $[M+H]^+$ calcd for $C_{21}H_{21}N_4$ 329.1761; found 329.1760.

4-Methoxy-N-((2-phenylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3ae). White solid, m.p. 159.1-159.6 °C; Isolated yield 71% (46.9 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) 8 8.19 (d, J = 6.84 Hz, 1 H), 7.95 (d, J = 6.06 Hz, 1 H), 7.79 (dd, J = 8.13, 1.50 Hz, 2 H), 7.64 (d, J = 9.00 Hz, 1 H), 7.45 (t, J = 7.68 Hz, 2 H), 7.37 (t, J = 7.41 Hz, 1 H), 7.23-7.20 (m, 1 H), 6.80 (td, J = 6.81, 1.26 Hz, 1 H), 6.27 (dd, J = 6.00, 2.22 Hz, 1 H), 5.88 (d, J = 2.16 Hz, 1 H), 5.00 (d, J = 4.92 Hz, 2 H), 4.87 (br, s, 1 H), 3.70 (s, 3 H). 13 C{ 1 H} NMR (100 MHz, CDCl₃) 8 167.3, 159.7, 148.4, 145.1, 144.5, 134.0, 128.7, 128.5, 128.0, 124.9, 124.4, 117.4, 116.8, 112.4, 102.7, 91.3, 54.9, 35.8. HR-ESI-MS [M+H]+ calcd for C₂₀H₁₉N₄O 331.1553; found 331.1552.

5-Fluoro-N-((2-phenylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3af). White solid, m.p. 163.2-163.9 °C; Isolated yield 60% (38.3 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz,

CDCl₃) δ 8.16 (d, J = 6.90 Hz, 1 H), 8.04 (d, J = 3.00 Hz, 1 H), 7.75 (dd, J = 8.25, 1.50 Hz, 2 H), 7.64 (d, J = 9.00 Hz, 1 H), 7.42 (t, J = 7.65 Hz, 2 H), 7.35 (t, J = 7.35 Hz, 1 H), 7.24-7.20 (m, 2 H), 6.81 (td, J = 6.81, 1.14 Hz, 1 H), 6.46 (dd, J = 9.00, 3.48 Hz, 1 H), 4.98 (d, J = 5.10 Hz, 2 H), 4.67 (br, s, 1 H). 13 C{ 1 H} NMR (100 MHz, CDCl₃) δ 154.8, 153.8 (d, J_{CF} = 241.1 Hz), 144.8, 144.1, 134.3 (d, J_{CF} = 24.4 Hz), 133.6, 128.6, 128.3, 125.4 (d, J_{CF} = 20.6 Hz), 125.1, 124.3, 116.20, 117.2, 117.0, 112.5, 108.9 (d, J_{CF} = 4.0 Hz), 35.9. HR-ESI-MS [M+H]⁺ calcd for C₁₉H₁₆N₄F 319.1354; found 319.1351.

4-Chloro-N-((2-phenylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3ag). White solid, m.p. 210.5-211.5 °C; Isolated yield 63% (42.3 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) δ 8.17 (d, J = 6.84 Hz, 1 H), 8.03 (d, J = 5.64 Hz, 1 H), 7.74 (d, J = 7.56 Hz, 2 H), 7.63 (d, J = 9.00 Hz, 1 H), 7.40 (t, J = 7.56 Hz, 2 H), 7.33 (t, J = 7.47 Hz, 1 H), 7.23 (t, J = 8.04 Hz, 1 H), 6.83 (t, J = 6.78 Hz, 1 H), 6.65 (d, J = 5.52 Hz, 1 H), 6.56 (s, 1 H), 5.25 (br, s, 1 H), 4.99 (d, J = 4.80 Hz, 2 H). 13 C 1 H 1 NMR (100 MHz, CDCl 3) δ 158.2, 149.7, 144.7, 143.9, 139.6, 133.3, 128.7, 128.3, 128.1, 125.4, 124.6, 117.0, 116.9, 112.8, 112.6, 106.5, 35.4. HR-ESI-MS [M+H] $^{+}$ calcd for C_{19} H $_{16}$ N $_{4}$ Cl 335.1058; found 335.1054.

S-Chloro-N-((2-phenylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3ah). ^{21b} Light brown solid, m.p. 211.2-213.1 °C; Isolated yield 70% (46.7 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) δ 8.17 (d, J = 6.90 Hz, 1 H), 8.13 (d, J = 2.52 Hz, 1 H), 7.75 (d, J = 6.90 Hz, 2 H), 7.64 (d, J = 9.00 Hz, 1 H), 7.42 (t, J = 7.71 Hz, 2 H), 7.39 (dd, J = 8.76, 2.58 Hz, 1 H), 7.34 (t, J = 7.38 Hz, 1 H), 7.23 (td, J = 7.91, 1.26 Hz, 1 H), 6.83 (td, J = 6.83, 1.14 Hz, 1 H), 6.48 (d, J = 8.76 Hz, 1 H), 5.00 (d, J = 4.86 Hz, 2 H), 4.88 (br, s, 1 H). ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 156.5, 146.2, 144.5, 143.5, 137.3, 133.0, 128.7, 128.4, 128.2, 125.6, 124.6, 120.5, 117.2, 117.0, 113.0, 109.6, 35.4. HR-ESI-MS [M+H]+ calcd for C₁₉H₁₆N₄Cl 335.1058; found 335.1056.

6-Chloro-N-((2-phenylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3ai). White solid, m.p. 174.2-175.1 °C; Isolated yield 60% (40.0 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. 1H NMR (600 MHz, CDCl₃) δ 8.17 (d, I = 6.78 Hz, 1 H), 7.70 (d, I = 7.26 Hz, 2 H), 7.58 (d, J = 9.06 Hz, 1 H), 7.39-7.31 (m, 4 H), 7.19 (t, J = 7.86 Hz, 1.58 Hz)1 H), 6.79 (t, J = 6.78 Hz, 1 H), 6.65 (d, J = 7.56 Hz, 1 H), 6.41 (d, I = 8.10 Hz, 1 H), 5.11 (s, 1 H), 4.96 (d, I = 4.98 Hz, 2 H). ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 158.2, 149.6, 144.6, 143.9, 139.5, 133.2, 128.6, 128.3, 128.0, 125.4, 124.5, 117.0, 116.8, 112.7, 112.5, 106.4, 35.3. HR-ESI-MS [M+H]+ calcd for C₁₉H₁₆N₄Cl 335.1058; found 335.1050.

6-Bromo-N-((2-phenylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3aj). White solid, m.p. 203.3-204.1 °C; Isolated yield 53% (40.2 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) δ 8.23 (d, J = 6.84 Hz, 1 H), 7.71 (dd, J = 8.28, 1.62 Hz, 2 H), 7.61 (d, J = 9.06 Hz, 1 H), 7.39 (t, J = 7.65 Hz, 2 H), 7.32 (t, J = 7.41 Hz, 1 H), 7.25 (t, J = 7.80 Hz, 1 H), 7.22-7.19 (m, 1 H), 6.83-6.80 (m, 2 H), 6.45 (d, J = 8.10 Hz, 1 H), 5.10 (br, s, 1 H), 4.97 (d, J = 5.04 Hz, 2 H). 13 C{ 1 H} NMR (100 MHz, CDCl₃) δ 158.3, 145.0, 144.5, 140.3, 139.4, 133.8, 128.7, 128.2, 127.9, 125.0, 124.4,

117.2, 116.6, 116.4, 112.5, 106.7, 35.5. HR-ESI-MS $[M+H]^+$ calcd for $C_{19}H_{16}N_4Br$ 379.0553; found 379.0553.

2-((2-Phenylimidazo[1,2-a]pyridin-3-yl)methylamino) isonicotinonitrile (3ak). White solid, m.p. 184.4-185.4 °C; Isolated yield 30% (19.5 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) δ 8.27 (dd, J = 5.22, 0.78 Hz, 1 H), 8.07 (d, J = 6.84 Hz, 1 H), 7.66 (d, J = 6.84 Hz, 2 H), 7.58 (d, J = 9.00 Hz, 1 H), 7.36 (t, J = 7.44 Hz, 2 H), 7.31 (t, J = 7.35 Hz, 1 H), 7.21 (t, J = 7.32 Hz, 1 H), 6.82-6.77 (m, 3 H), 5.60 (br, s, 1 H), 4.98 (d, J = 4.80 Hz, 2 H). 13 C{¹H} NMR (100 MHz, CDCl₃) δ 158.2, 1447, 143.9, 140.3, 139.4, 133.2, 128.7, 128.3, 128.1, 125.4, 124.6, 117.0, 116.8, 116.5, 112.8, 106.7, 35.4. HR-ESI-MS [M+H]* calcd for C_{20} H₁₆N₅ 326.1400; found 326.1397.

Methyl 2-((2-phenylimidazo[1,2-a]pyridin-3-yl)methylamino)isonicotinate (3al). Light yellow solid, m.p. 163.8-164.2 °C; Isolated yield 42% (30.2 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. 1 H NMR (600 MHz, CDCl₃) δ 8.24 (d, J = 5.22 Hz, 1 H), 8.11 (d, J = 6.90 Hz, 1 H), 7.75 (d, J = 7.56 Hz, 2 H), 7.59 (d, J = 9.06 Hz, 1 H), 7.40 (t, J = 7.59 Hz, 2 H), 7.33 (t, J = 7.41 Hz, 1 H), 7.18 (t, J = 7.92 Hz, 1 H), 7.14 (dd, J = 5.28, 1.32 Hz, 1 H), 7.04 (s, 1 H), 6.77 (t, J = 6.69 Hz, 1 H), 5.17 (br, s, 1 H), 5.02 (d, J = 4.86 Hz, 2 H), 3.87 (s, 3 H). 13 C{ 1 H} NMR (100 MHz, CDCl₃) δ 166.0, 158.6, 148.8, 145.0, 144.5, 138.7, 133.7, 128.6, 128.3, 128.0, 125.0, 124.2, 117.3, 116.7, 112.5, 112.2, 108.4, 52.5, 35.6 . HR-ESI-MS [M+H]+ calcd for C_{21} H₁₉N₄O₂ 359.1503; found 359.1498.

3-Methyl-N-((2-phenylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3am). Brown solid, m.p. 150.2-152.1 °C; Isolated yield 52% (32.6)mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) δ 8.26 (d, J = 6.78 Hz, 1 H), 8.12 (dd, J = 5.13, 1.80 Hz, 1 H), 7.83 (d, J = 6.90 Hz, 2 H), 7.71 (d, J = 9.06 Hz, 1 H), 7.46 (t, J = 9.06 Hz, 1 H), J = 9.06 Hz, 1 H), J = 9.06 Hz, J = 9.06= 7.68 Hz, 2 H), 7.38 (t, J = 7.41 Hz, 1 H), 7.28-7.24 (m, 2 H), 6.84(td, J = 6.81, 1.20 Hz, 1 H), 6.64 (dd, J = 6.15, 1.20 Hz, 1 H), 5.20(d, J = 5.04 Hz, 2 H), 4.28 (s, 1 H), 2.03 (s, 3 H). ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 155.2, 144.1, 143.5, 142.4, 136.3, 132.2, 127.8, 127.5, 127.2, 124.6, 123.7, 116.7, 116.3, 116.0, 112.6, 111.9, 34.3, 16.0. HR-ESI-MS [M+H]⁺ calcd for C₂₀H₁₉N₄ 315.1604; found 315.1596.

4,6-Dimethyl-N-((2-phenylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3an). White solid, m.p. 163.8-164.1 °C; Isolated yield 64% (42.1 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) δ 8.25 (d, J = 6.90 Hz, 1 H), 7.79 (d, J = 7.08 Hz, 2 H), 7.64 (d, J = 9.00 Hz, 1 H), 7.46 (t, J = 7.59 Hz, 2 H), 7.38 (t, J = 7.44 Hz, 1 H), 7.22 (t, J = 7.89 Hz, 1 H), 6.81 (t, J = 6.78 Hz, 1 H), 6.37 (s, 1 H), 6.04 (s, 1 H), 4.97 (d, J = 4.86 Hz, 2 H), 4.92 (s, 1 H), 2.38 (s, 3 H), 2.15 (s, 3 H). 13 C 1 H 1 NMR (100 MHz, CDCl₃) δ 158.0, 156.6, 148.9, 145.1, 144.6, 134.2, 128.7, 128.5, 127.9, 124.7, 124.5, 117.4, 117.3, 114.5, 112.2, 104.8, 35.9, 24.2, 21.0. HR-ESI-MS [M+H]+ calcd for C₂₁H₂₁N₄ 329.1761; found 329.1758.

5-Fluoro-4-methyl-N-((2-phenylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3ao). Light yellow solid, m.p. 210.6-211.5 °C; Isolated yield 53% (35.2 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. 1 H NMR (600 MHz, CDCl₃) δ 8.13 (d, J = 6.66 Hz, 1 H), 7.91 (d, J = 1.50 Hz, 1 H), 7.74 (d, J = 7.02 Hz, 2 H), 7.61 (d, J = 9.06 Hz, 1 H), 7.42 (t, J

= 7.62 Hz, 2 H), 7.35 (t, J = 7.41 Hz, 1 H), 7.19 (td, J = 7.88, 1.26 Hz, 1 H), 6.78 (td, J = 6.75, 1.20 Hz, 1 H), 6.26 (d, J = 5.04 Hz, 1 H), 4.93 (d, J = 5.04 Hz, 2 H), 4.57 (s, 1 H), 2.18 (s, 3 H). 13 C{ 1 H} NMR (100 MHz, CDCl₃) δ 154.7, 153.5 (d, J_{CF} = 239.0 Hz), 145.1, 144.5, 136.3 (d, J_{CF} = 17.1 Hz), 134.0, 133.7 (d, J_{CF} = 25.9 Hz), 128.7, 128.4, 128.0, 124.9, 124.4, 117.4, 117.1, 112.4, 109.7 (d, J_{CF} = 1.7 Hz), 36.1, 14.6 (d, J_{CF} = 3.2 Hz). HR-ESI-MS [M+H]* calcd for C_{20} H₁₈N₄F 333.1510; found 333.1507.

N-((2-Phenylimidazo[1,2-a]pyridin-3-yl)methyl) quinolin-2-amine (3ap). White solid, m.p. 186.5-187.3 °C; Isolated yield 35% (24.5 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) δ 8.29 (d, J = 6.84 Hz, 1 H), 7.82 (d, J = 8.82 Hz, 1 H), 7.78 (t, J = 8.76 Hz, 3 H), 7.63 (d, J = 7.92 Hz, 1 H), 7.60-7.57 (m, 2 H), 7.40 (t, J = 7.56 Hz, 2 H), 7.33 (t, J = 7.35 Hz, 1 H), 7.27 (t, J = 7.29 Hz, 1 H), 7.16 (t, J = 7.80 Hz, 1 H), 6.73 (t, J = 6.78 Hz, 1 H), 6.68 (d, J = 8.76 Hz, 1 H), 5.25 (d, J = 4.92 Hz, 2 H), 5.15 (s, 1 H). 13 C{ 1 H} NMR (100 MHz, CDCl₃) δ 156.2, 147.5, 145.0, 144.4, 137.5, 133.9, 129.7, 128.6, 128.3, 127.9, 127.5, 126.2, 124.9, 124.7, 122.60, 122.5, 117.4, 117.2, 112.4, 112.3, 34.8. HR-ESI-MS [M+H]+ calcd for C_{23} H₁₉N₄ 351.1604; found 351.1601.

N-((2-Phenylimidazo[1,2-a]pyridin-3-yl)methyl) pyrimidin-2-amine (3aq). Light yellow solid, m.p. 119.4-120.1 °C; Isolated yield 35% (21.1 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. H NMR (600 MHz, CDCl₃) δ 8.19 (d, J = 6.90 Hz, 1 H), 8.15 (br, s, 1 H), 7.80 (d, J = 6.90 Hz, 2 H), 7.67 (d, J = 9.00 Hz, 1 H), 7.46 (t, J = 7.56 Hz, 2 H), 7.38 (t, J = 7.38 Hz, 1 H), 7.22 (td, J = 7.85, 1.26 Hz, 1 H), 6.80 (td, J = 6.83, 1.20 Hz, 1 H), 6.54 (t, J = 4.83 Hz, 1 H), 5.93 (s, 1 H), 5.08 (d, J = 5.10 Hz, 2 H). Clab NMR (100 MHz, CDCl₃) δ 162.1, 158.1, 145.0, 144.5, 133.7, 128.8, 128.5, 128.2, 125.2, 124.4, 117.4, 116.9, 112.7, 111.4, 35.2. HR-ESI-MS [M+H]+ calcd for $C_{18}H_{16}N_5$ 302.1400; found 302.1398.

N-((8-*Methyl-2-phenylimidazo*[1,2-*a*]*pyridin-3-yl*)*methyl*)*pyridin-2-amine* (3*ba*). Light brown solid, m.p. 159.1-159.8 °C; Isolated yield 51% (31.4 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) δ 8.15 (dd, J = 5.16, 1.74 Hz, 1 H), 8.03 (d, J = 6.78 Hz, 1 H), 7.76 (dd, J = 8.19, 1.50 Hz, 2 H), 7.44-7.40 (m, 3 H), 7.33 (t, J = 7.41 Hz, 1 H), 7.01 (d, J = 6.84 Hz, 1 H), 6.71 (t, J = 6.84 Hz, 1 H), 6.66-6.64 (m, 1 H), 6.47 (d, J = 8.34 Hz, 1 H), 4.97 (d, J = 4.92 Hz, 2 H), 4.84 (br, s, 1 H), 2.66 (s, 3 H). 13 C{ 1 H} NMR (100 MHz, CDCl₃) δ 158.0, 147.6, 145.4, 143.9, 137.5, 134.0, 128.6, 128.5, 127.8, 127.3, 123.8, 122.1, 117.2, 113.4, 112.5, 108.4, 35.6, 17.1. HR-ESI-MS [M+H]+ calcd for C₂₀H₁₉N₄ 315.1604; found 315.1601.

N-((7-Methyl-2-phenylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3ca). Light yellow solid, m.p. 167.6-168.7 °C; Isolated yield 41% (25.7 mg). Eluent: petroleum ether/ethyl acetate/chlorofomr/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) 8 8.16 (d, J = 3.18 Hz, 1 H), 8.02 (d, J = 6.90 Hz, 1 H), 7.75 (d, J = 7.08 Hz, 2 H), 7.44-7.39 (m, 3 H), 7.37 (s, 1 H), 7.33 (t, J = 7.32 Hz, 1 H), 6.65 (td, J = 6.09, 2.04 Hz, 1 H), 6.60 (dd, J = 6.96, 1.68 Hz, 1 H), 6.48 (d, J = 8.28 Hz, 1 H), 4.97 (d, J = 4.92 Hz, 2 H), 4.76 (s, 1 H), 2.38 (s, 3 H). 13 C{ 1 H} NMR (100 MHz, CDCl₃) 8 158.2, 147.9, 145.4, 144.0, 137.4, 135.9, 134.0, 128.6, 128.3, 127.8, 123.5, 116.4, 115.7, 115.0, 113.5, 108.4, 35.5, 21.3. HR-ESI-MS [M+H]* calcd for C₂0H₁9N₄ 315.1604; found 315.1601.

N-((6-Methyl-2-phenylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3da).^{21b} Yellow solid, m.p. 166.4-167.2 °C; Isolated yield (40.9 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. 1H NMR (600 MHz, CDCl₃) δ 8.17 (d, J = 3.24 Hz, 1 H), 7.95 (s, 1 H), 7.75 (d, J = 7.56Hz, 2 H), 7.53 (d, J = 9.12 Hz, 1 H), 7.44 (td, J = 7.77, 1.92 Hz, 1 H), 7.41 (t, I = 7.59 Hz, 2 H), 7.33 (t, I = 7.44 Hz, 1 H), 7.05 (dd, I= 9.12, 1.68 Hz, 1 H), 6.66 (dd, J = 6.06, 2.10 Hz, 1 H), 6.51 (d, J = 6.06, 2.10 Hz, 1 H)8.34 Hz, 1 H), 4.95 (d, J = 4.74 Hz, 2 H), 4.86 (s, 1 H), 2.28 (s, 3 H). ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 157.1, 146.5, 142.81, 142.78, 136.6, 132.6, 127.7, 127.4, 127.3, 126.9, 121.4, 121.1, 115.7, 115.4, 112.5, 107.5, 34.6, 17.3. HR-ESI-MS [M+H]+ calcd for C₂₀H₁₉N₄ 315.1604; found 315.1595.

N-((7-Ethyl-2-phenylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3ea). Yellow solid, m.p. 166.7-167.6 °C; Isolated yield 46% (30.1)mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) δ 8.17 (d, J = 4.92, 1.86 Hz, 1 H), 8.06 (d, J = 7.02 Hz, 1 H),7.76 d, I = 7.08 Hz, 2 H), 7.44-7.40 (m, 4 H), 7.34 (t, I = 7.35Hz, 1 H), 6.65 (td, J = 5.73, 2.40 Hz, 2 H), 6.45 (d, J = 8.28 Hz, 1 H), 4.98 (d, J = 4.98 Hz, 2 H), 4.67 (s, 1 H), 2.69 (q, J = 7.56 Hz, 2H), 1.27 (t, J = 7.56 Hz, 3 H). ${}^{13}C\{{}^{1}H\}$ NMR (100 MHz, CDCl₃) δ 158.3, 148.0, 145.6, 144.2, 142.0, 137.4, 134.2, 128.6, 128.3, 127.7, 123.7, 116.4, 114.3, 114.0, 113.5, 108.4, 35.6, 28.4, 14.4. HR-ESI-MS $[M+H]^+$ calcd for $C_{21}H_{21}N_4$ 329.1761; found 329.1758.

N-((6-Fluoro-2-phenylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3fa). Yellow solid, m.p. 169.5-170.4 °C; Isolated yield 42% (26.7)Eluent: petroleum ether/ethyl mg). acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) δ 8.24 (dd, J = 4.38, 1.98 Hz, 1 H), 8.17 (dd, J = 5.46, 1.86 Hz, 1 H), 7.73 (dd, J = 8.31, 1.62 Hz, 2 H), 7.58 (dd, J = 9.78, 5.10 Hz, 1 H), 7.45-7.42 (m, 3 H), 7.36 (t, *J* = 7.38 Hz, 1 H), 7.13-7.10 (m, 1 H), 6.67 (td, J = 6.15, 1.20 Hz, 1 H), 6.48 (d, J = 8.28 Hz, 1 H), 5.00 (d, J = 5.16 Hz, 2 H), 4.81 (s, 1 H). ${}^{13}C\{{}^{1}H\}$ NMR (100 MHz, CDCl₃) δ 157.8, 153.2 (d, J_{C-F} = 235.6 Hz), 147.3, 145.7 (d, $I_{CF} = 2.14 \text{ Hz}$), 142.6, 137.8, 133.6, 128.7, 128.3, 128.1, 118.7 (d, $I_{CF} = 2.14 \text{ Hz}$) $_F = 2.16 \text{ Hz}$), 117.7 (d, $J_{C-F} = 9.02 \text{ Hz}$), 116.9 (d, $J_{C-F} = 25.46 \text{ Hz}$), 113.7, 111.5 (d, $J_{C-F} = 41.0 \text{ Hz}$), 108.7, 35.4. HR-ESI-MS [M+H]⁺ calcd for $C_{19}H_{16}N_4F$ 319.1354; found 319.1347.

N-((8-Chloro-2-phenylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3ga). White solid, m.p. 218.5-219.5 °C; Isolated yield 41% (27.3)mg). Eluent: ether/ethyl petroleum acetate/chloroform/methanol 10:5:2:2. 1H NMR (600 MHz, CDCl₃) δ 8.14 (d, J = 6.84 Hz, 1 H), 8.12 (d, J = 4.26 Hz, 1 H), 7.72 (d, J = 7.08 Hz, 2 H), 7.45 (td, J = 7.82, 1.92 Hz, 1 H), 7.38 (t, J)= 7.44 Hz, 2 H), 7.33 (t, J = 7.29 Hz, 1 H), 7.25 (d, J = 7.80 Hz, 1 Hz)H), 6.70 (t, J = 7.08 Hz, 1 H), 6.66 (t, J = 6.12 Hz, 1 H), 6.54 (d, J =8.40 Hz, 1 H), 5.37 (s, 1 H), 5.00 (d, I = 5.04 Hz, 2 H). ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 157.8, 147.2, 145.4, 143.3, 137.9, 133.4, 128.8, 128.4, 128.3, 126.4, 122.7, 120.7, 117.9, 117.6, 113.8, 108.8, 35.5. HR-ESI-MS [M+H]⁺ calcd for C₁₉H₁₆N₄Cl 335.1058; found 335.1055.

N-((7-Chloro-2-phenylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3ha). Light yellow solid, m.p. 163.7-164.3 °C; Isolated yield 63% (42.0 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) δ 8.15 (d, J = 7.14 Hz, 2 H), 7.70 (d, J = 6.96 Hz, 2 H), 7.56 (s, 1 H), 7.45-7.40 (m, 3 H), 7.35 (t, J = 7.41 Hz, 1 H), 6.76 (dd, J

= 7.29, 2.16 Hz, 1 H), 6.66 (td, J = 6.17, 0.84 Hz, 1 H), 6.49 (d, J = 8.34 Hz, 1 H), 5.04 (s, 1 H), 4.99 (d, J = 4.74 Hz, 2 H). 13 C{ 1 H} NMR (100 MHz, CDCl₃) δ 157.7, 147.1, 145.1, 144.7, 137.9, 133.4, 131.6, 128.7, 128.3, 128.2, 125.0, 117.4, 116.1, 114.0, 113.6, 108.7, 35.2. HR-ESI-MS [M+H] $^{+}$ calcd for $C_{19}H_{16}N_{4}Cl$ 335.1058; found 335.1050.

N-((6-Chloro-2-phenylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3ia). Light yellow solid, m.p. 172.2-173.4 °C; Isolated yield 78% (52.3 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) 8 8.37 (d, J = 1.98 Hz, 1 H), 8.16 (dd, J = 4.83, 2.04 Hz, 1 H), 7.73 (d, J = 7.02 Hz, 2 H), 7.55 (d, J = 9.42 Hz, 1 H), 7.46-7.42 (m, 3 H), 7.37 (t, J = 7.38 Hz, 1 H), 7.16 (dd, J = 9.54, 1.92 Hz, 1 H), 6.68 (td, J = 6.17, 1.07 Hz, 1 H), 6.49 (d, J = 8.34 Hz, 1 H), 5.02 (br, s, 1 H), 4.99 (d, J = 3.66 Hz, 2 H). ¹³C{¹H} NMR (100 MHz, CDCl₃) 8 157.7, 147.2, 145.3, 143.3, 137.9, 133.4, 128.7, 128.3, 128.2, 126.3, 122.7, 120.7, 117.8, 117.6, 113.7, 108.7, 35.4. HR-ESI-MS [M+H]+ calcd for C¹9H¹6N₄Cl 335.1058; found 335.1057.

Methyl 2-phenyl-3-((pyridin-2-ylamino)methyl)imidazo [1,2-a] pyridine-6-carboxylate (3ja). White solid, m.p. 153.6-154.7 °C; Isolated yield 45% (32.2 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) δ 9.16 (s, 1 H), 8.18 (d, J = 3.00 Hz, 1 H), 7.77 (d, J = 7.08 Hz, 2 H), 7.73 (dd, J = 9.48, 1.68 Hz, 1 H), 7.59 (d, J = 9.36 Hz, 1 H), 7.45-7.42 (m, 3 H), 7.37 (t, J = 7.29 Hz, 1 H), 6.66 (dd, J = 6.15, 2.04 Hz, 1 H), 6.47 (d, J = 8.28 Hz, 1 H), 5.06 (d, J = 5.16 Hz, 2 H), 4.90 (br, s, 1 H), 3.90 (s, 3 H). 13 C{ 1 H} NMR (100 MHz, CDCl₃) δ 165.4, 157.8, 147.6, 146.1, 145.5, 137.7, 133.5, 129.2, 128.8, 128.5, 128.4, 124.6, 118.7, 116.6, 116.3, 113.7, 108.6, 52.4, 35.2. HR-ESI-MS [M+H] $^{+}$ calcd for C₂₁H₁₉N₄O₂ 359.1503; found 359.1497.

N-((2-p-Tolylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin -2-amine (3ka).^{21b} Light yellow solid, m.p. 169.1-169.7 °C; Isolated yield 61% (38.3)mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) δ 8.19 (d, J = 6.78 Hz, 1 H), 8.16 (d, J = 5.22, 1.80 Hz, 1 H), 7.66 (d, J = 7.74 Hz, 2 H), 7.64 (d, J = 9.30 Hz, 1 H), 7.43 (t, J= 8.73 Hz, 1 H), 7.23 (d, J = 8.10 Hz, 2 H), 7.20 (d, J = 8.28 Hz, 1 Hz)H), 6.79 (t, J = 6.78 Hz, 1 H), 6.66 (t, J = 6.12 Hz, 1 H), 6.49 (d, J =8.34 Hz, 1 H), 5.01 (d, J = 4.80 Hz, 2 H), 4.90 (s, 1 H), 2.38 (s, 3 H). ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 158.2, 148.0, 145.0, 144.6, 137.7, 137.4, 131.1, 129.4, 128.2, 124.7, 124.3, 117.3, 116.7, 113.5, 112.3, 108.4, 35.6, 21.3. HR-ESI-MS [M+H]⁺ calcd for C₂₀H₁₉N₄ 315.1604; found 315.1601.

N-((2-(4-Isopropylphenyl)imidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (31a). Light yellow solid, m.p. 148.9-149.5 °C; Isolated yield 57% (39.0 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) δ 8.15 (d, J = 3.36 Hz, 1 H), 8.13 (d, J = 6.84 Hz, 1 H), 7.68 (d, J = 7.86 Hz, 2 H), 7.60 (d, J = 9.06 Hz, 1 H), 7.42 (td, J = 7.74, 1.92 Hz, 1 H), 7.27 (d, J = 7.86 Hz, 2 H), 7.17 (t, J = 7.89 Hz, 1 H), 6.74 (t, J = 6.72 Hz, 1 H), 6.64 (t, J = 6.09 Hz, 1 H), 6.47 (d, J = 8.22 Hz, 1 H), 4.99 (d, J = 4.98 Hz, 2 H), 4.76 (s, 1 H), 2.93 (heptet, J = 6.91 Hz, 1 H), 1.27 (d, J = 7.02 Hz, 6 H). 13 C{ 1 H} NMR (100 MHz, CDCl₃) δ 158.1, 148.6, 147.9, 144.9, 144.5, 137.3, 131.3, 128.2, 126.7, 124.6, 124.3, 117.2, 116.7, 113.4, 112.2, 108.4, 35.5, 33.8, 23.9. HR-ESI-MS [M+H] $^{+}$ calcd for C₂₂H₂₃N₄343.1917;

found 343.1914.

N-((2-(4-Methoxyphenyl)imidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3ma). Light yellow solid, m.p. 144.9-145.6 °C; Isolated yield 45% (29.7 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. H NMR (600 MHz, CDCl₃) δ 8.17 (d, J = 3.24 Hz, 1 H), 8.14 (d, J = 6.84 Hz, 1 H), 7.70 (d, J = 8.70 Hz, 2 H),7.61 (d, J = 9.06 Hz, 1 H), 7.43 (td, J = 7.76, 1.86 Hz, 1 H), 7.20-7.17 (m, 1 H), 6.95 (d, J = 8.70 Hz, 2 H), 6.77 (td, J = 6.78, 1.14 Hz, 1 H), 6.65 (td, J = 6.17, 0.96 Hz, 1 H), 3.83 (s, 3 H). J = 8.28 Hz, 1 H), 4.98 (d, J = 4.98 Hz, 2 H), 4.80 (s, 1 H), 3.83 (s, 3 H). J = 8.26 Hz, 1 H), 8.14 (100 MHz, CDCl₃) δ 159.5, 158.1, 147.8, 144.7, 144.0, 137.5, 129.5, 126.2, 124.9, 124.3, 117.0, 116.3, 114.1, 113.5, 112.4, 108.5, 55.2, 35.5. HR-ESI-MS [M+H]+ calcd for $C_{20}H_{19}N_4O$ 331.1553; found 331.1551.

N-((2-(4-Chlorophenyl)imidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3na). ^{21b} White solid, m.p. 151.6-152.9 °C; Isolated yield 43% (28.7 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) δ 8.17 (t, J = 2.70 Hz, 1 H), 8.14 (t, J = 6.81 Hz, 1 H), 7.68 (t, J = 7.14 Hz, 2 H), 7.60 (t, J = 7.56 Hz, 1 H), 7.45 (td, J = 7.71, 1.80 Hz, 1 H), 7.36 (t, J = 7.41 Hz, 2 H), 7.23-7.19 (m, 1 H), 6.81-6.77 (m, 1 H), 6.67 (t, J = 6.18 Hz, 1 H), 6.50 (dd, J = 8.37, 3.60 Hz, 1 H), 4.97 (d, J = 4.98 Hz, 2 H), 4.78 (s, 1 H). ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 158.0, 147.9, 145.0, 143.2, 137.6, 133.9, 132.4, 129.5, 128.8, 125.2, 124.4, 117.3, 117.2, 113.7, 112.6, 108.5, 35.4. HR-ESI-MS [M+H]+ calcd for C₁9H₁6N₄Cl 335.1058; found 335.1057.

N-((2-(4-Bromophenyl)imidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3oa). White solid, m.p. 165.3-166.5 °C; Isolated yield 40% (30.2 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) 8 8.12 (dd, J = 5.10, 1.80 Hz, 1 H), 8.11 (d, J = 6.90 Hz, 1 H), 7.59 (d, J = 8.22 Hz, 2 H), 7.57 (d, J = 9.12 Hz, 1 H), 7.48 (d, J = 8.52 Hz, 2 H), 7.44 (td, J = 7.71, 1.92 Hz, 1 H), 7.19 (td, J = 7.92, 1.26 Hz, 1 H), 6.77 (td, J = 6.80, 1.20 Hz, 1 H), 6.66 (td, J = 6.17, 0.90 Hz, 1 H), 6.52 (d, J = 8.28 Hz, 1 H), 4.95-4.94 (m, 3 H). 13 C{ 1 H} NMR (100 MHz, CDCl₃) 8 157.9, 147.7, 144.9, 143.0, 137.6, 132.7, 131.7, 129.7, 125.2, 124.3, 122.2, 117.3, 117.2, 113.6, 112.6, 108.5, 35.4. HR-ESI-MS [M+H]+ calcd for C₁9H₁6N₄Br 379.0553; found 379.0548.

4-(3-((Pyridin-2-ylamino)methyl)imidazo[1,2-a]pyridin-2-yl)benzonitrile (3pa). Light yellow solid, m.p. 137.6-138.1 °C; Isolated yield 41% (26.7 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. H NMR (600 MHz, CDCl₃) δ 8.17 (d, J = 5.16 Hz, 1 H), 8.14 (d, J = 6.66 Hz, 1 H), 7.84 (d, J = 7.68 Hz, 2 H), 7.61 (d, J = 7.98 Hz, 2 H), 7.58 (d, J = 9.06 Hz, 1 H), 7.45 (td, J = 7.73, 1.86 Hz, 1 H), 7.23 (t, J = 7.86 Hz, 1 H), 6.80 (t, J = 6.75 Hz, 1 H), 6.68 (t, J = 6.12 Hz, 1 H), 6.53 (d, J = 8.28 Hz, 1 H), 4.97 (d, J = 5.16 Hz, 2 H), 4.95 (s, br, 1 H). 13 C{ 1 H} NMR (100 MHz, CDCl₃) δ 157.9, 148.0, 145.3, 142.2, 138.5, 137.6, 132.4, 128.5, 125.6, 124.5, 118.9, 118.4, 117.6, 113.9, 113.0, 111.2, 108.6, 35.4. HR-ESI-MS [M+H]⁺ calcd for C₂₀H₁₆N₅ 326.1400; found 326.1394.

N-((2-o-Tolylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin -2-amine (3qa). Light brown solid, m.p. 146.3-147.1 °C; Isolated yield 59% (37.0 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. 1 H NMR (600 MHz, CDCl₃) δ 8.23 (d, J = 6.78 Hz, 1 H), 8.06 (d, J = 5.40 Hz, 1 H),

7.60 (d, J = 8.88 Hz, 1 H), 7.35 (t, J = 7.80 Hz, 1 H), 7.31-7.26 (m, 3 H), 7.19 (t, J = 8.22 Hz, 2 H), 6.77 (t, J = 6.69 Hz, 1 H), 6.58 (t, J = 6.09 Hz, 1 H), 6.32 (d, J = 8.46 Hz, 1 H), 4.78 (d, J = 5.10 Hz, 2 H), 4.72 (s, 1 H), 2.32 (s, 3 H). 13 C{ 1 H} NMR (100 MHz, CDCl₃) 8 158.1, 147.7, 145.0, 144.8, 137.62, 137.57, 133.3, 130.7, 130.4, 128.4, 125.5, 124.70, 124.66, 118.1, 117.4, 113.5, 112.3, 108.1, 35.3, 20.3. HR-ESI-MS [M+H]⁺ calcd for C_{20} H₁₉N₄ 315.1604; found 315.1598.

N-((6-Methyl-2-p-tolylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3ra). White solid, m.p. 167.5-168.2 °C; Isolated yield 47% mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2.1H NMR (600 MHz, CDCl₃) δ 8.16 (d, J = 4.14 Hz, 1 H), 7.89 (s, 1 H), 7.62 (d, J = 7.74Hz, 2 H), 7.50 (d, J = 9.12 Hz, 1 H), 7.43 (td, J = 7.77, 1.92 Hz, 1 H), 7.19 (d, J = 7.74 Hz, 2 H), 7.02 (dd, J = 9.06, 1.56 Hz, 1 H), 6.65 (td, *J* = 6.15, 2.04 Hz, 1 H), 6.52 (d, *J* = 8.34 Hz, 1 H), 4.91-4.88 (m, 3 H), 2.36 (s, 3 H), 2.26 (s, 3 H). ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 158.0, 156.6, 148.9, 145.1, 144.6, 134.2, 128.7, 128.5, 127.9, 124.7, 124.5, 117.4, 117.3, 114.5, 112.2, 104.8, 35.9, 24.2, 21.0. HR-ESI-MS [M+H]⁺ calcd for C₂₁H₂₁N₄ 329.1761; found 329.1757.

N-((2-(4-Methoxyphenyl)-7-methylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3sa). Light red solid, m.p. 146.9-147.6 °C; Isolated yield 55% (37.8 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) δ 8.18 (d, J = 3.18 Hz, 1 H), 8.01 (d, J = 6.90 Hz, 1 H), 7.69 (d, J = 8.70 Hz, 2 H), 7.43 (td, J = 7.77, 1.92 Hz, 1 H), 7.36 (s, 1 H), 6.95 (d, J = 8.76 Hz, 2 H), 6.65 (td, J = 6.12, 2.10 Hz, 1 H), 6.60 (dd, J = 7.05, 1.68 Hz, 1 H), 6.48 (d, J = 8.34 Hz, 1 H), 4.94 (d, J = 4.86 Hz, 2 H), 4.71 (s, 1 H), 3.83 (s, 3 H), 2.38 (s, 3 H). 13 C{ 1 H} NMR (100 MHz, CDCl₃) δ 159.4, 158.2, 147.9, 145.3, 143.9, 137.4, 135.8, 129.5, 126.6, 123.4, 115.7, 115.6, 114.9, 114.1, 113.5, 108.3, 55.3, 35.6, 21.3. HR-ESI-MS [M+H]* calcd for C_{21} H₂₁N₄O 345.1710; found 345.1707.

N-((2-(4-Methoxyphenyl)-6-methylimidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3ta). Light yellow solid, m.p. 166.6-167.5 °C; Isolated yield 62% (42.6 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) δ 8.16 (dd, J = 5.22, 1.02 Hz, 1 H), 7.84 (s, 1 H), 7.65 (d, J = 8.82 Hz, 2 H), 7.46 (d, J = 9.12 Hz, 1 H), 7.44 (td, J = 7.73, 1.98 Hz, 1 H), 7.00 (dd, J = 9.12, 1.68 Hz, 1 H), 6.90 (d, J = 8.76 Hz, 2 H), 6.65 (td, J = 6.11, 0.90 Hz, 1 H), 6.54 (d, J = 8.28 Hz, 1 H), 4.98 (br, s, 1 H), 4.88 (d, J = 4.68 Hz, 2 H), 3.81 (s, 3 H), 2.25 (s, 3 H). 13 C{ 1 H} NMR (100 MHz, CDCl₃) δ 159.4, 158.2, 148.0, 145.4, 143.9, 137.4, 135.8, 129.5, 126.6, 123.5, 115.7, 115.6, 114.9, 114.1, 113.5, 108.4, 55.3, 35.7, 21.4. HR-ESI-MS [M+H]+ calcd for C₂1H₂1N₄O 345.1710; found 345.1706.

N-((2-(Thiophen-2-yl)imidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3ua). ^{21b} Light brown solid, m.p. 152.2-153.4 °C; Isolated yield 31% (18.9 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) 8 8.19 (td, J = 5.04, 2.22 Hz, 2 H), 7.57 (d, J = 9.06 Hz, 1 H), 7.44-7.41 (m, 2 H), 7.33 (d, J = 4.98, 1.08 Hz, 1 H), 7.17 (t, J = 7.92 Hz, 1 H), 7.07 (t, J = 4.35 Hz, 1 H), 6.76 (t, J = 6.81 Hz, 1 H), 6.66 (td, J = 6.15, 1.98 Hz, 1 H), 6.51 (d, J = 8.28 Hz, 1 H), 5.09 (d, J = 5.04 Hz, 2 H), 4.85 (s, 1 H). ¹³C{¹H} NMR (100 MHz, CDCl₃) 8 157.7, 146.9, 145.0, 138.8, 138.0, 136.7, 127.8, 125.9, 125.25, 125.21, 124.4, 117.1, 116.4, 113.5, 112.6, 108.8, 35.2. HR-ESI-MS

 $[M+H]^+$ calcd for $C_{17}H_{15}N_4S$ 307.1012; found 307.1006.

N-((2-(Naphthalen-2-yl)imidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2-amine (3va). Yellow solid, m.p. 182.6-183.7 °C; Isolated yield 45% (31.5 mg). Eluent: petroleum ether/ethyl acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) δ 8.22 (s, 1 H), 8.17 (dd, J = 5.25, 2.46 Hz, 1 H), 8.15 (d, J = 6.84 Hz, 1 H), 7.90 (d, J = 8.52 Hz, 1 H), 7.85 (d, J = 8.58 Hz, 1 H), 7.81 (d, J = 5.82 Hz, 2 H), 7.63 (d, J = 9.00 Hz, 1 H), 7.46-7.45 (m, 2 H), 7.42 (t, J = 6.99 Hz, 1 H), 7.17 (t, J = 7.92 Hz, 1 H), 6.74 (t, J = 6.78 Hz, 1 H), 6.66 (t, J = 6.15 Hz, 1 H), 6.49 (d, J = 8.34 Hz, 1 H), 5.07 (d, J = 4.80 Hz, 2 H), 4.93 (s, 1 H). 13 C{ 1 H} NMR (100 MHz, CDCl₃) δ 158.1, 147.9, 145.1, 144.3, 137.6, 133.5, 133.0, 131.3, 128.40, 128.36, 127.7, 127.5, 126.3, 126.2, 125.1, 124.4, 117.5, 117.4, 113.7, 112.6, 108.5, 35.6. HR-ESI-MS [M+H]+ calcd for C_{23} H₁₉N₄351.1604; found 351.1601.

N-((2-(tert-Butyl)imidazo[1,2-a]pyridin-3-yl)methyl)pyridin-2amine (3wa).21b White solid, m.p. 196.3-197.8 °C; Isolated yield 33% petroleum (18.6) Eluent: ether/ethyl mg). acetate/chloroform/methanol 10:5:2:2. ¹H NMR (600 MHz, CDCl₃) δ 8.18 (d, J = 5.22 Hz, 1 H), 8.03 (d, J = 6.78 Hz, 1 H), 7.60 (d, J = 9.00 Hz, 1 H), 7.43 (td, J = 7.77, 1.92 Hz, 1 H), 7.14 (t, J = 7.77, I = 7.77J = 7.83 Hz, 1 H), 6.74 (t, J = 6.75 Hz, 1 H), 6.65 (t, J = 6.09 Hz, 1 Hz)H), 6.44 (d, J = 8.34 Hz, 1 H), 4.98 (d, J = 4.80 Hz, 2 H), 4.30 (s, 1 H), 1.50 (s, 9 H). ${}^{13}C{}^{1}H$ NMR (100 MHz, CDCl₃) δ 158.1, 153.7, 148.1, 143.6, 137.4, 124.0, 123.6, 117.2, 115.4, 113.5, 112.1, 108.3, 36.0, 33.4, 31.2. HR-ESI-MS [M+H]⁺ calcd for C₁₇H₂₁N₄ 281.1761; found 281.1754.

Gram-scale synthesis of product 3aa. An oven-dried round bottom reaction vessel was charged with 2-phenylimidazo[1,2-a]pyridine **1a** (6 mmol), 2-aminopyridine **2a** (18 mmol), 2,2'-bpy (234.3 mg, 1.5 mmol, 25 mol %), and FeCp₂ (279.1 mg, 1.5 mmol, 25 mol %). After the vessel was filled with argon, TBHP (70 wt. % in H₂O, 2.07 mL, 15 mmol), MeOH/H₂O (60 mL, 7:3 v/v ratio) was added by syringe under argon, and the reaction mixture was stirred at room temperature for 5 min. Then the vessel was sealed, placed into an oil bath and heated to 90 °C. After 48 h, the resulting mixture was cooled to room temperature, filtered through a short silica gel pad and washed with ethyl acetate. The above solution was concentraed under vacuum, and the residue was purified by silica gel column chromatography to give the analytically pure product **3aa** (73%; 1.3156 g).

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications websites at DOI: 10.1021/acs.joc.xxxxxxxx.

X-ray crystallographic structure and data for **3aa** and Copies of ¹H and ¹³C NMR spectra for the products (PDF)

AUTHOR INFORMATION

Corresponding Author

*E-mail: raohonghua@tsinghua.org.cn

Author Contributions

N.L. and J.B. contributed equally to this work.

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENT

We thank the NSFC (21871185, 21402128), Beijing Natural Science Foundation (2172015), Key Laboratory of Bioorganic Phosphorus Chemistry and Chemical Biology (Ministry of Education, P. R. China), and Capital Normal University for generous financial support. We thank Hong Cui in our group for reproducing the result of **3aa**.

REFERENCES

- (1) (a) Balaban, A. T.; Oniciu, D. C.; Katritzky, A. R. Aromaticity as a Cornerstone of Heterocyclic Chemistry. *Chem. Rev.* **2004**, *104*, 2777. (b) Buckingham, J.; Baggaley, K. H.; Roberts, A. D.; Szabó, L. F. *Dictionary of Alkaloids*, 2nd ed.; CRC Press: London, 2010.
- (2) Brahmachari, G. Green Synthetic Approaches for Biologically Relevant Heterocycles; Elsevier: Oxford, 2015.
- (3) (a) Spicer, J. A.; Gamage, S. A.; Rewcastle, G. W.; Finlay, G. J.; Bridewell, D. J. A.; Baguley, B. C.; Denny, W. A. Bis(phenazine-1-carboxamides): Structure–Activity Relationships for a New Class of Dual Topoisomerase I/II-Directed Anticancer Drugs. J. Med. Chem. 2000, 43, 1350. (b) Brahmachari, G. Chemistry and Pharmacology of Naturally Occurring Bioactive Compounds; CRC Press: London, 2010. (c) Hegedus, L. S.; Söderberg, B. C. G. Transition Metals in the Synthesis of Complex Organic Molecules, 3rd ed.; University Science Books: USA, 2010.
- (4) Johnson, T. J.; Hedge, D. D. Esomeprazole: a Clinical Review. *Am. J. Health-Syst. Pharm.* **2002**, *59*, 655.
- (5) Balfour, J. A. B.; Goa, K. L.; Perry, C. M. Alosetron. *Drugs* **2000**, *59*, 511.
- (6) McHardy, N. Baquiloprim for Treating Protozoal Infections. *Eur. Pat. Appl.* EP19910300061, **1991**.
- (7) Schilder, R. J.; Sill, M. W.; Lankes, H. A.; Gold, M. A.; Mannel, R. S.; Modesitt, S. C.; Hanjani, P.; Bonebrake, A. J.; Sood, A. K.; Godwin, A. K.; Hu, W.; Alpaugh, R. K. A Phase II Evaluation of Motesanib (AMG 706) in the Treatment of Persistent or Recurrent Ovarian, Fallopian Tube and Primary Peritoneal Carcinomas: A Gynecologic Oncology Group Study. *Gynecol. Oncol.* **2013**, *129*, 86.
- (8) For selected examples, see: (a) Flaherty, A.; Trunkfield, A.; Barton, W. Palladium-Catalyzed Cross-Coupling of B-Benzvl-9borabicyclo [3.3.1] nonane To Furnish Methylene-Linked Biaryls. Org. Lett. **2005**, 7, 4975. (b) Hall, A.; Billinton, A.; Brown, S. H.; Chowdhury, A.; Clayton, N. M.; Giblin, G. M. P.; Gibson, M.; Goldsmith, P. A.; Hurst, D. N.; Naylor, A.; Peet, C. F.; Scoccitti, T.; Wilson, A. W.; Winchester, W. of Sodium 6-[(5-Chloro-2-{[(4-chloro-2 $fluor ophenyl) methyl] oxy\} phenyl) methyl] -2-pyridine carboxylate$ (GSK269984A) an EP₁ Receptor Antagonist for the Treatment of Inflammatory Pain. Bioorg. Med. Chem. Lett. 2009, 19, 2599. (c) Dreher, S. D.; Lim, S.-E.; Sandrock, D. L.; Molander, G. A. Suzuki-Miyaura Cross-Coupling Reactions of Primary Alkyltrifluoroborates with Aryl Chlorides. J. Org. Chem. 2009, 74, 3626.
- (9) Modi, A.; Ali, W.; Patel, B. K. N,N-Dimethylacetamide (DMA) as a Methylene Synthon for Regioselective Linkage of Imidazo[1,2-a]pyridine. Adv. Synth. Catal. **2016**, 358, 2100.
- (10) For selected example, see: N'Nang Obiang, E. O.; Genta-Jouve, G.; Gallard, J.-F.; Kumulungui, B.; Mouray, E.; Grellier, P.; Evanno, L.; Poupon, E.; Champy, P.; Beniddir, M. A. Pleiokomenines A and B: Dimeric Aspidofractinine Alkaloids Tethered with a Methylene Group. *Org. Lett.* **2017**, *19*, 6180.
- (11) For selected example, see: Xu, S.; Chen, R.; Fu, Z.; Zhou, Q.; Zhang, Y.; Wang, J. Palladium-Catalyzed Formal [4+1] Annulation via Metal Carbene Migratory Insertion and $C(sp^2)$ –H Bond Functionalization. *ACS Catal.* **2017**, *7*, 1993.
- (12) For representative reviews, see: (a) Sun, C.-L.; Li, B.-J.; Shi, Z.-J. Direct C–H Transformation via Iron Catalysis. *Chem. Rev.* **2011**, *111*, 1293. (b) Li, B. J.; Shi, Z.-J. From $C(sp^2)$ –H to $C(sp^3)$ –H: Systematic Studies on Transition Metal-Catalyzed Oxidative C–C Formation. *Chem. Soc. Rev.* **2012**, *41*, 5588. (c) Shi, Z.; Zhang, C.; Tang, C.; Jiao, N. Recent Advances

- in Transition-metal Catalyzed Reactions Using Molecular Oxygen as the Oxidant. Chem. Soc. Rev. 2012, 41, 3381. (d) Arochiam, P. B.; Bruneau, C.; Dixneuf, P. H. Ruthenium(II)-Catalyzed C-H Bond Activation and Functionalization. Chem. Rev. 2012, 112, 5879. (e) Mousseau, J. J.; Charette, A. A. B. Direct Functionalization Processes: A Journey from Palladium to Copper to Iron to Nickel to Metal-Free Coupling Reactions. Acc. Chem. Res. **2013**, *46*, 412. (f) Lin, B.; Dixneuf, P. H. *sp*² C–H Bond Activation in Water and Catalytic Cross-Coupling Reactions. Chem. Soc. Rev. 2013, 42, 5744. (g) Wu, Y.; Wang, J.; Mao, F.; Kwong, F. Y. Palladium-Catalyzed Cross-Dehydrogenative Functionalization of C(sp2)-H Bonds. Chem. Asian. J. 2014, 9, 26. (h) Girard, S. A.; Knauber, T.; Li, C.-J. The Cross-Dehydrogenative Coupling of C_{sp}³–H Bonds: A Versatile Strategy for C–C Bond Formations. Angew. Chem. Int. Ed. 2014, 53, 74. (i) Bauer, I.; Knölker, H.-J. Iron Catalysis in Organic Synthesis. Chem. Rev. 2015, 115, 3170. (j) Lei, A.; Shi, W.; Liu, C.; Liu, W.; Zhang, H.; He, C. Oxidative Cross-Coupling Reactions; Wiley-VCH: Weinheim, 2017.
- (13) Minisci, F.; Bernardi, R.; Bertini, F.; Galli, R.; Perchinummo, M. Nucleophilic Character of Alkyl Radicals—VI: A New Convenient Selective Alkylation of Heteroaromatic Bases. *Tetrahedron* **1971**, *27*, 3575.
- (14) For representative examples, see: (a) Katz, R. B.; Mistry, J.; Mitchell, M. B. An Improved Method for the Mono-Hydroxymethylation of Pyridines. A Modification of the Minisci Procedure. *Synth. Commun.* **1989**, *19*, 317. (b) Biyouki, M. A. A.; Smith, R. A. J.; Bedford, J. J.; Leader, J. P. Hydroxymethylation and Carbamoylation of Di-And Tetramethylpyridines Using Radical Substitution (Minisci) Reactions. *Synth. Commun.* **1998**, *28*, 3817. (c) Clerici, A.; Ghilardi, A.; Pastori, N.; Punta, C.; Porta, O. A New One-Pot, Four-Component Synthesis of 1,2-Amino Alcohols: TiCl₃/*t*-BuOOH-Mediated Radical Hydroxymethylation of Imines. *Org. Lett.* **2008**, *10*, 5063. (d) Huff, C. A.; Cohen, R. D.; Dykstra, K. D.; Streckfuss, E.; DiRocco, D. A.; Krska, S. W. Photoredox-Catalyzed Hydroxymethylation of Heteroaromatic Bases. *J. Org. Chem.* **2016**, *81*, 6980.
- (15) Larock, R. C. Comprehensive Organic Transformations: a Guide to Functional Group Preparations, 2nd ed.; Wiley-VCH: Weinheim, 1999.
- (16) For reviews, see: (a) Hamid, M. H. S. A.; Slatford, P. A.; Williams, J. M. J. Borrowing Hydrogen in the Activation of Alcohols. *Adv. Synth. Catal.* **2007**, *349*, 1555. (b) Nixon, T. D.; Whittlesey, M. K.; Williams, J. M. J. Transition Metal Catalysed Reactions of Alcohols Using Borrowing Hydrogen Methodology. *Dalton Trans.* **2009**, 753. (c) Dobereiner, G. E.; Crabtree, R. H. Dehydrogenation as a Substrate-Activating Strategy in Homogeneous Transition-Metal Catalysis. *Chem. Rev.* **2010**, *110*, 681.
- (17) For the first example on Ir-catalyzed amination of alcohols with amines, see: Fujita, K.-I.; Yamamoto, K.; Yamaguchi, R. Oxidative Cyclization of Amino Alcohols Catalyzed by a Cp*Ir Complex. Synthesis of Indoles, 1,2,3,4-Tetrahydroquinolines, and 2,3,4,5-Tetrahydro-1-benzazepine. *Org. Lett.* **2002**, *4*, 2691.
- (18) For the first example on Ru-catalyzed amination of alcohols with amines, see: Hamid, M. H. S. A.; Williams, J. M. J. Ruthenium Catalysed *N*-Alkylation of Amines with Alcohols. *Chem. Commun.* **2007**, 725.
- (19) For selected examples on Fe-catalysis, see: (a) Fujita, K.-I.; Komatsubara, A.; Yamaguchi, R. N-Alkylation of Carbamates and Amides with Alcohols Catalyzed by a Cp*Ir Complex. Tetrahedron 2009, 65, 3624. (b) Gonzalez-Arellano, C.; Yoshida, K.; Luque, R.; Gai, P. L. Highly Active and Selective Supported Iron Oxide Nanoparticles in Microwave-Assisted N-Alkylations of Amines with Alcohols. Green Chem. 2010, 12, 1281. (c) Watson, A. J. A.; Maxwell, A. C.; Williams, J. M. J. Borrowing Hydrogen Methodology for Amine Synthesis under Solvent-Free Microwave Conditions. J. Org. Chem. 2011, 76, 2328. For selected examples on Cu-catalysis, see: (a) Likhar, P. R.; Arundhathi, R.; Kantam, M. L.; Prathima, P. S. Amination of Alcohols Catalyzed by Copper-Aluminium Hydrotalcite: A Green Synthesis of Amines. Eur. J. Org. Chem. 2009, 5383. (b) He, J.; Yamaguchi, K.; Mizuno, N. Selective Synthesis of Secondary Amines via N-Alkylation of Primary Amines and Ammonia with Alcohols by Supported Copper Hydroxide Catalysts. Chem. Lett. 2010, 39, 1182. (c) Martínez-Asencio, A.; Ramón, D. J.; Yus, M. N-Alkylation of Poor Nucleophilic Amines and Derivatives with Alcohols by a Hydrogen Autotransfer Process Catalyzed by

Copper(II) Acetate: Scope and Mechanistic Considerations. Tetrahedron , 67, 3140. (d) Li, F.; Shan, H.; Kang, Q.; Chen, L. Regioselective N-Alkylation of 2-Aminobenzothiazoles with Benzylic Alcohols. Chem. Commun. 2011, 47, 5058. For selected examples on Pd-catalysis, see: (a) Corma, A.; Ròdenas, T.; Sabater, M. J. A Bifunctional Pd/MgO Solid Catalyst for the One-Pot Selective N-Monoalkylation of Amines with Alcohols. Chem. Eur. J. 2010, 16, 254; (b) Zhang, Y.; Qi, X.; Cui, X.; Shi, F.; Deng, Y. Palladium Catalyzed N-Alkylation of Amines with Alcohols. Tetrahedron Lett. 2011, 52, 1334. For selected examples on Ag-catalysis, see: (a) Shimizu, K.; Nishimura, M.; Satsuma, A. γ-Alumina-Supported Silver Cluster for N-Benzylation of Anilines with Alcohols. ChemCatChem 2009, 1, 497. (b) Cui, X.; Zhang, Y.; Shi, F.; Deng, Y. Organic Ligand-Free Alkylation of Amines, Carboxamides, Sulfonamides, and Ketones by Using Alcohols Catalyzed by Heterogeneous Ag/Mo Oxides. Chem. Eur. J. 2011, 17, 1021. For selected examples on Au-catalysis, see: He, L.; Lou, X.-B.; Ni, J.; Liu, Y.-M.; Cao, Y.; He, H.-Y.; Fan, K.-N. Efficient and Clean Gold-Catalyzed One-Pot Selective N-Alkylation of Amines with Alcohols. Chem. Eur. J. 2010, 16, 13965. For selected examples on Os-catalysis, see: Bertoli, M.; Choualeb, A.; Lough, A. J.; Moore, B.; Spasyuk, D.; Gusev, D. G. Osmium and Ruthenium Catalysts for Dehydrogenation of Alcohols. Organometallics 2011, 30, 3479.

- (20) Fang, Q. K.; Wu, F. X.; Grover, P. T.; Hopkins, S. C.; Campbell, U.; Chytil, M.; Spear, K. L. Imidazo[1,2-a]pyridine Compounds: Cross-Reference to Related Applications. PCT/US2009/044525, **2009**.
- (21) For some examples on the aminomethylation of imidazo[1,2-a]pyridine to generate analogues of the desired products, see: (a) Sanaeishoar, H.; Nazarpour, R.; Mohave, F. Novel One-pot Pseudo Four Component Reaction: Expeditious Synthesis of Functionalized Imidazo[1,2-a]pyridines. RSC Adv. 2015, 5, 68571. (b) Patel, Om P. S.; Nandwana, M. K.; Sah, A. K.; Kumar, A. Metal-free Synthesis of Aminomethylated Imidazoheterocycles: Dual Role of tert-Butyl Hydroperoxide as Both an Oxidant and a Methylene Source. Org. Biomol. Chem. 2018, 16, 8620.
- (22) Ma, X.; Liu, F.; Li, Z.; Cao, S.; Rao, H. Tetra-*n*-butylammonium Bromide: A Simple but Efficient Organocatalyst for Alcohol Oxidation under Mild Conditions. *Adv. Synth. Catal.* **2014**, *356*, 1741. And references cited therein
- (23) Li, C.-J. Organic Reactions in Aqueous Media with a Focus on Carbon–Carbon Bond Formations: A Decade Update. *Chem. Rev.* **2005**, 105, 3095.
- (24) Wertz, S.; Leifert, D.; Studer, A. Cross Dehydrogenative Coupling via Base-Promoted Homolytic Aromatic Substitution (BHAS): Synthesis of Fluorenones and Xanthones. *Org. Lett.* **2013**, *15*, 928.
- (25) Su, H.; Wang, L.; Rao, H.; Xu, H. Iron-Catalyzed Dehydrogenative sp³–sp² Coupling via Direct Oxidative C–H Activation of Acetonitrile. *Org. Lett.* **2017**, *19*, 2226.
- (26) For Fe^{III}-catalyzed formation of dimeric ether from the corresponding alcohols, see: (a) Sabhi, P.; Iranpoor, N.; Behbahani, F. K. Selective and Efficient Alcoholyses of Allylic, Secondary- and Tertiary Benzylic Alcohols in the Presence of Iron(III). *Tetrahedron* 1998, 54, 943. (b) Jana, U.; Biswas, S.; Maiti, S. A Simple and Efficient FeCl₃-catalyzed Direct Alkylation of Active Methylene Compounds with Benzylic and Allylic Alcohols under Mild Conditions. *Tetrahedron Lett.* 2007, 48, 4065. (c) Jana, U.; Maiti, S.; Biswas, S. An FeCl₃-catalyzed Highly C3-Selective Fiedel–Crafts Alkylation of Indoles with Alcohols. *Tetrahedron Lett.* 2007, 48, 7160. (d) Jana, U.; Maiti, S.; Biswas, S. An Efficient FeCl₃-catalyzed Amidation Reaction of Secondary Benzylic and Allylic Alcohols with Carboxamides or *p*-Toluenesulfonamide. *Tetrahedron Lett.* 2008, 49, 858. (e) Jana, U.; Biswas, S.; Maiti, S. Iron(III)-Catalyzed Addition of Benzylic Alcohols to Aryl Alkynes–A New Synthesis of Substituted Aryl Ketones. *Eur. J. Org. Chem.* 2008, 5798.
- (27) (a) Gündel, W.-H. Study on Quaternary Pyridinium Reaction of 3-Cyanopyridinium Salts on Addition of Alkoxides. *Z. Naturforsch. B J. Chem. Sci.* **1980**, 35*b*, 490. (b) Nakaike, Y.; Hayashi, D.; Nishiwaki, N.; Tobe, Y.; Ariga, M. Formylnitroenamines: Useful Building Blocks for Nitrated Pyridones and Aminopyridines with Functional Groups. *Org. Biomol. Chem.*

- **2009**, *7*, 325. (c) Dissanayake, A. A.; Staples, R. J.; Odom, A. L. Titanium-Catalyzed, One-Pot Synthesis of 2-Amino-3-cyano- pyridines. *Adv. Synth. Catal.* **2014**, 356, 1811. (d) Duan, S.; Place, D.; Perfect, H. H.; Ide, N. D.; Maloney, M.; Sutherland, K.; Wiglesworth, K. E. P.; Wang, K.; Olivier, M.; Kong, F.; Leeman, K.; Blunt, J.; Draper, J.; McAuliffe, M.; O'Sullivan, M.; Lynch, D. Palbociclib Commercial Manufacturing Process Development. Part I: Control of Regioselectivity in a Grignard-Mediated S_NAr Coupling. *Org. Process Res. Dev.* **2016**, 20, 1191.
- (28) Takizawa, S.-Y.; Nishida, J.-I.; Tsuzuki, T.; Tokito, S.; Ymashita, Y. Phosphorescent Iridium Complexes Based on 2-Phenylimidazo[1,2-a]pyridine Ligands: Tuning of Emission Color toward the Blue Region and Application to Polymer Light-Emitting Devices. *Inorg. Chem.* **2007**, *46*, 4308
- (29) Kumar, G. S.; Ragini, S. P.; Kumar, A. S.; Meshram, H. M. A Copper-catalyzed Multi-component Reaction Accessing Fused Imidazoheterocycles *via* C–H Functionalization. *RSC Adv.* **2015**, *5*, 51576.