

SYNTHESIS AND IN VITRO ANTIMALARIAL ACTIVITY OF SULFONE ENDOPEROXIDES

Mario D. Bachi,^{*a} Edward E. Korshin,^a Poonsakdi Ploypradith,^b
Jared N. Cumming,^b Suji Xie,^c Theresa A. Shapiro,^c and Gary H. Posner^b

^a*Department of Organic Chemistry, The Weizmann Institute of Science, Rehovot 76100, Israel.*

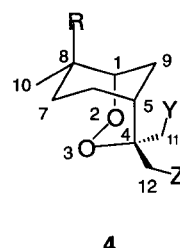
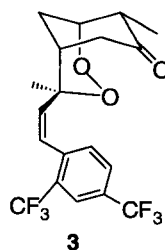
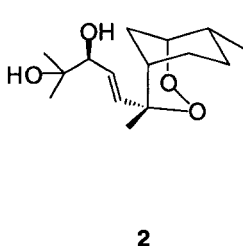
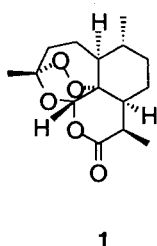
^b*Department of Chemistry, School of Arts and Sciences,
The Johns Hopkins University, Baltimore, MD 21218, U.S.A.*

^c*Department of Medicine, School of Medicine, The Johns Hopkins University, Baltimore, MD 21205, U.S.A.*

Received 26 January 1998; accepted 2 March 1998

Abstract: A series of 4,8-dimethyl-4-phenylsulfonylmethyl-2,3-dioxabicyclo[3.3.1]nonanes, carrying a variety of substituents at position-8 (**4**) were prepared by a short and efficient method from *R*-(+)-limonene. Key reactions include thiol oxygen cooxidation, and alkylation and acylation of a sterically hindered *tertiary* alcohol compatible with the endoperoxy functionality. Some of compounds **4**, which are structurally related to yingzhaosu A (**2**), were found to exhibit in vitro antimalarial activity comparable to that of artemisinin (**1**) and superior to that of arteflene (**3**). © 1998 Elsevier Science Ltd. All rights reserved.

The devastating consequences of malaria are causing wide international concern.¹ A promising approach for treating malaria deriving from chloroquine-resistant parasites is based on the development of new drugs which incorporate in their molecular structure an endoperoxide functionality.^{2–5} Artemisinin (**1**), which is being used as a drug in China, has inspired researchers to design and study various other antimalarial trioxanes.^{5–9} A structurally simpler endoperoxide, yingzhaosu A (**2**), was isolated from an antimalarial Chinese folk medicine and was subsequently obtained by total synthesis.^{10,11} The synthesis, antimalarial screening and clinical trials of 7-oxo-2,3-dioxabicyclo[3.3.1]nonanes, bearing at C(4) alkyl or alkenyl substituents as represented by arteflene (**3**) were described.^{12,13} The expectation that other compounds, like yingzhaosu A and arteflene (**3**), containing the 2,3-dioxabicyclo[3.3.1]nonane system as a central molecular feature may exhibit antimalarial activity led us to design, synthesize and screen sulfur-containing 2,3-dioxabicyclo[3.3.1]nonane derivatives of type **4** (Y = PhS and Z = H, or Y = H and Z = PhS) and **4** (Y = PhSO₂ and Z = H, or Y = H and Z = PhSO₂). The presence of a sulfonyl group is not alien to antimalarial compounds, and its compatibility with antimalarial activity of 2,3-dioxabicyclo[3.3.1]nonane pharmacophore is reported herein.^{14,15}



Epimeric sulfide endoperoxides **6a,b**, themselves very poor antimalarials, served as starting materials for the preparation of a series of highly active compounds (Table 1).¹⁷ Endoperoxides **6a,b** were obtained from *R*-(+)-limonene in a one pot process which involves thiol oxygen cooxidation of the terpene, followed by selective reduction of the resulting endoperoxide-hydroperoxide (Scheme 1).¹⁷ Oxidation of **6a,b** with 2.5 equivalents of MCPBA followed by chromatography afforded sulfones **7a** and **7b**.¹⁷ These compounds exhibit significant antimalarial activity indicating that compounds of type **4** in which Y or Z represent a PhSO₂ group exhibit higher antimalarial activity than their PhS analogs (Table 1).¹⁸

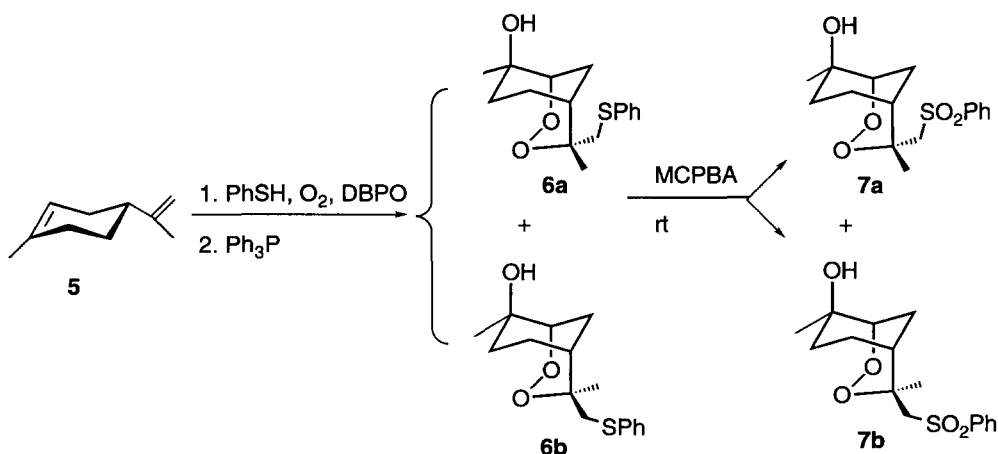
Table 1. Antimalarial Activity of 2,3-dioxabicyclo[3.3.1]nonanes **4** Against Chloroquine-Sensitive *P. falciparum* (NF 54) in vitro^a

Compound	Absolute Configuration	R	Y	Z	IC ₅₀ (nM)
6a	1 <i>R</i> , 4 <i>R</i> , 5 <i>R</i> , 8 <i>R</i>	OH	PhS	H	>2500
6b	1 <i>R</i> , 4 <i>S</i> , 5 <i>R</i> , 8 <i>R</i>	OH	H	PhS	
7a	1 <i>R</i> , 4 <i>R</i> , 5 <i>R</i> , 8 <i>R</i>	OH	PhSO ₂	H	55
7b	1 <i>R</i> , 4 <i>S</i> , 5 <i>R</i> , 8 <i>R</i>	OH	H	PhSO ₂	89
8a	1 <i>R</i> , 4 <i>R</i> , 5 <i>R</i> , 8 <i>R</i>	OCH ₂ C ₆ H ₄ OMe- <i>p</i>	PhSO ₂	H	14
10a	1 <i>R</i> , 4 <i>R</i> , 5 <i>R</i> , 8 <i>R</i>	OAc	PhSO ₂	H	17
10b	1 <i>R</i> , 4 <i>S</i> , 5 <i>R</i> , 8 <i>R</i>	OAc	H	PhSO ₂	17
13a	1 <i>R</i> , 4 <i>R</i> , 5 <i>R</i> , 8 <i>R</i>	OC(O)C(O)OEt	PhSO ₂	H	170
13b	1 <i>R</i> , 4 <i>S</i> , 5 <i>R</i> , 8 <i>R</i>	OC(O)C(O)OEt	H	PhSO ₂	140
14a	1 <i>R</i> , 4 <i>R</i> , 5 <i>R</i> , 8 <i>R</i>	OC(O)C(O)NBn ₂	PhSO ₂	H	21
14b	1 <i>R</i> , 4 <i>S</i> , 5 <i>R</i> , 8 <i>R</i>	OC(O)C(O)NBn ₂	H	PhSO ₂	81
17a	1 <i>R</i> , 4 <i>R</i> , 5 <i>R</i> , 8 <i>R</i>	OC(O)CH ₂ Ac	PhSO ₂	H	46
17b	1 <i>R</i> , 4 <i>S</i> , 5 <i>R</i> , 8 <i>R</i>	OC(O)CH ₂ Ac	H	PhSO ₂	73
Artemisinin (1)					9.3; 16 ^b
Arteflene (3)					71; 110 ^b

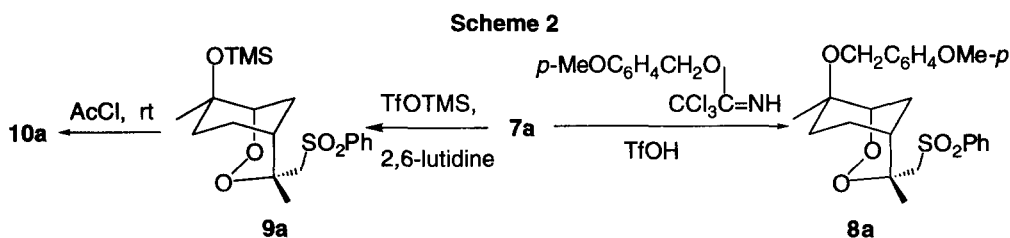
^aAntimalarial activity was determined by the modified method of Desjardins²⁶ and Milhous²⁷ as described in reference 28. The standard deviation for each set of quadruplets was an average of 9% (≤53%) of the mean. R² values for the fitted curves were ≥0.982.

^bData taken from reference 13.

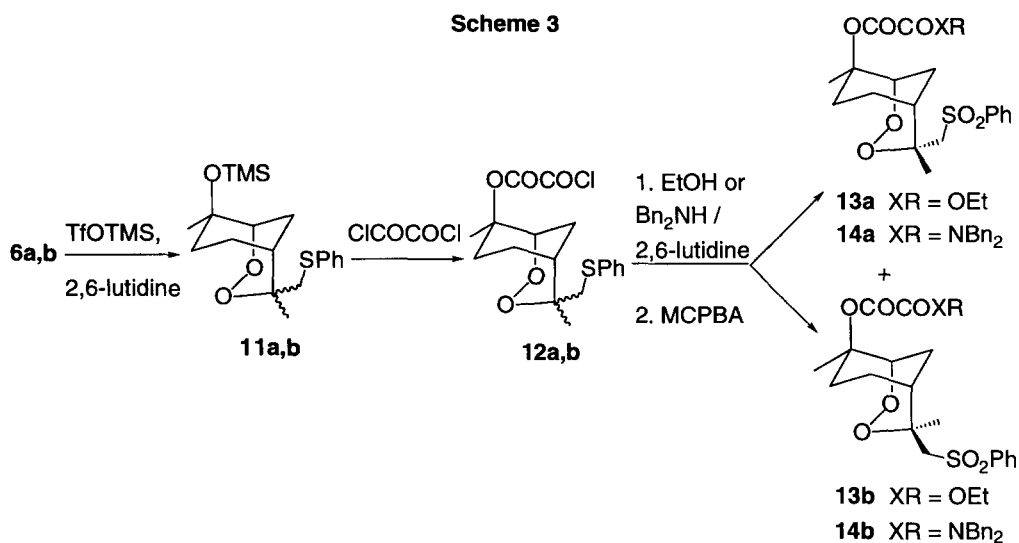
Scheme 1



Assuming that decreasing polarity and increasing lipophilicity may be associated with increase of antimalarial activity ^{8,19–22} derivatives **8a**, **10**, **13–14**, and **17** were prepared.^{23,24} Thus, alcohol **7a** (0.85 mmol) in ether (suspension in 3 mL, 0 °C) was treated with *O*-(*p*-methoxybenzyl) trichloroacetimidate²⁵ (4.2 mmol) in CH₂Cl₂ (1.5 mL) followed by TfOH (0.043 mmol) in ether (0.43 mL) (Scheme 2). After 12 h addition of imidate and TfOH was repeated and the mixture was stirred until consumption of **7a**, to give after standard workup derivative **8a** (46%). Acetyl derivative **10a** was best prepared (95%, from **7a**) by silylation of hydroxysulphone **7a** (0.70 mmol) with TfOTMS (1.5 mmol) and 2,6-lutidine (1.8 mmol) in CH₂Cl₂, followed by treatment of the resulting TMS derivative **9a** with acetyl chloride (3 mL, 45 h, rt) (Scheme 2). This method of

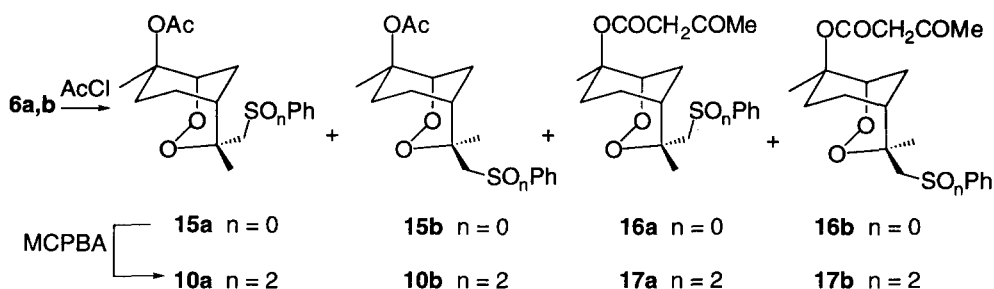


acylation was also applied for the preparation of derivatives **13–14** (Scheme 3). Silylation of epimeric hydroxy sulfides **6a,b** (0.52 mmol) afforded TMS-ethers **11a,b** which were treated with oxalyl chloride (3 mL) to give chlorides **12a,b**. Treatment of **12a,b** with EtOH or Bn₂NH and 2,6-lutidine, followed by oxidation of the sulfide group, afforded respectively the sulfone esters **13a** and **13b** (64%) or sulfone amides **14a** and **14b** (76%).



Although direct acylation of free *tertiary* alcohols **6a,b** afforded acyl derivatives in lower yields than acylation of the TMS-ethers **9** or **11**, it provided, through a secondary reaction, additional interesting antimalarial endoperoxides (Scheme 4). Thus, addition of AcCl (15.6 mmol) in CH₂Cl₂ (5 mL) to hydroxy sulfides **6a,b** (3.92 mmol), pyridine (19.5 mmol) and DMAP (0.4 mmol) in CH₂Cl₂ (30 mL) at 0 °C and then rt (12 h) afforded the sulfide acetates **15a,b** (53%) and sulfide acetoacetates **16a,b** (13%). Oxidation of these acylation products with MCPBA afforded sulfone acetates **10a** and **10b** (97%) and sulfone acetoacetates **17a** and **17b** (67%).

Scheme 4



The data for antimalarial activity *in vitro* summarized in Table 1 indicates that, except for the case of derivatives **13a** and **13b**, blocking the free hydroxy group in **7** is associated with increase in antimalarial activity. Furthermore, it was found that compounds of the “a” series are usually slightly more reactive than their corresponding C-4 epimers of the “b” series.²⁹

In conclusion, thiol oxygen cooxidation of *R*-(+)-limonene (**5**), followed by alkylation or acylations of a sterically hindered *tertiary* alcohol under conditions compatible with the peroxide function of the 2,3-dioxabicyclo[3.3.1]nonane system, provided a series of readily available and potent antimalarial agents. 4-Phenylsulfonylmethyl-2,3-dioxabicyclo[3.3.1]nonanes **8a**, **10a**, **10b**, and **14a** exhibit *in vitro* antimalarial activity comparable to that of arteflene (**2**),^{12,13} of the drug artemisinin (**1**) and of 1,2,4-trioxanes structurally related to (**1**).^{4–9,19–21,30–33}

Acknowledgment: This research was supported by grant No. 94-102 from the United States-Israel Binational Science Foundation (BSF), Jerusalem, Israel. The U.S. NIH and the Burroughs Wellcome Fund supported the antimalarial *in vitro* studies at Johns Hopkins.

References and Notes

- Butler, D.; Maurice, J.; Clare, O. *Nature* **1997**, 386, 535.
- Jung, M. *Curr. Med. Chem.* **1994**, 1, 46.
- Butler, A. R.; Wu, Y.-L. *Chem. Soc. Rev.* **1992**, 85.
- Meshnick, S. R.; Taylor, T. E.; Kamchonowngpaisan, S. *Microbiological Reviews* **1996**, 60, 301.
- Cumming, J. N.; Ploypradith, P.; Posner, G. H. *Adv. Pharmacol.* **1997**, 37, 253.

6. Jefford, C. W.; Kohmoto, S.; Jaggi, D.; Timari, G.; Rossier, J.-C.; Rudaz, M.; Barbuzzi, O.; Gerard, D.; Burger, U.; Kamalaprija, D.; Mareda, J.; Bernardinelli, G.; Manzanara, I.; Canfield, C. J.; Fleck, S. L.; Robinson, B. L.; Peters, W. *Helv. Chim. Acta* **1995**, *78*, 647.
7. Posner, G. H.; Park, S. B.; González, L.; Wang, D.; Cumming, J. N.; Klinedinst, D.; Shapiro, T. A.; Bachi, M. D. *J. Am. Chem. Soc.* **1996**, *118*, 3537.
8. Avery, M. A.; Mehrotra, S.; Johnson, T. L.; Bonk, J. D.; Vroman, J. A.; Miller, R. *J. Med. Chem.* **1996**, *39*, 4149.
9. Avery, M. A.; Gao, F.; Chong, W. K. M.; Hendrickson, T. F.; Inman, W. D.; Crews, P. *Tetrahedron* **1994**, *50*, 957.
10. Liang, X. T.; Yu, D. Q.; Wu, W. L.; Deng, H. C. *Acta Chim. Sin.* **1979**, *37*, 215.
11. Xu, X. X.; Zhu, J.; Haung, D.; Zhou, W. *Tetrahedron Lett.* **1991**, *32*, 5785.
12. Hofheinz, W.; Burgin, H.; Gocke, E.; Jaquet, C.; Masciadri, R.; Schmid, G.; Stohler, H.; Urwyler, H. *Trop. Med. Parasitol.* **1994**, *45*, 261.
13. Jaquet, C.; Stohler, H. R.; Chollet, J.; Peters, W. *Trop. Med. Parasitol.* **1994**, *45*, 266.
14. Rosenthal, P. J.; Olson, J. E.; Lee, G. K.; Palmer, J. T.; Klaus, J. L.; Rasnick, D. *Antimicrob. Agents Chemother.* **1996**, *40*, 1600.
15. For antimalarial activity of sulfone trioxanes see reference 16.
16. Posner, G. H.; O'Dowd, H.; Caferro, T.; Cumming, J. N.; Ploypradith, P.; Xie, S.; Shapiro, T. A. *Tetrahedron Lett.*, in press.
17. Reaction mechanism and experimental protocol for the reactions described in Scheme 1 are detailed in: Bachi, M. D.; Korshin, E. E. *Synlett.* **1998**, 122.
18. A similar effect is described in reference 16.
19. Avery, M. A.; Gao, F.; Chong, W. K. M.; Mehrotra, S.; Milhous, W. K. *J. Med. Chem.* **1993**, *36*, 4264.
20. Posner, G. H.; Oh, C. H.; Gerena, L.; Milhous, W. K. *J. Med. Chem.* **1992**, *35*, 2459.
21. Posner, G. H.; McGarvey, D.; Oh, C. H.; Kumar, N.; Meshnick, S. R.; Asawamahasadka, W. *J. Med. Chem.* **1995**, *38*, 607.
22. Posner, G. H.; Oh, C. H.; Gerena, L.; Milhous, W. K. *Heteroatom. Chem.* **1995**, *6*, 105.
23. All new compounds described in Table 1 were fully characterized by detailed NMR analysis that included NOE difference experiments: 1-D spectra (^1H , $^{13}\text{C}/\text{DEPT}$), and 2-D NMR spectra ($^1\text{H}/^1\text{H}$ COSY, $^1\text{H}/^{13}\text{C}$ HMQC), and by elemental microanalysis or CI HRMS.
24. Representative spectral data for compound **8a**: ^1H NMR (400 MHz, CDCl_3 , δ): 1.38 (s, 3H, Me^{10}), 1.53 (br.s, 3H, Me^{11}), 1.81 (dddd, 1H, $J = 14.0, 14.0, 6.4$ and 3.3 Hz, H_a^6), 1.85 (m, 1H, H_e^6), 1.94 (br.dd, 1H, $J = 14.0, 5.0$ Hz, H_e^7), 2.11 (ddd, 1H, $J = 14.0, 14.0$ and 6.4 Hz, H_a^7), 2.17 (m, 2H, $\text{H}_e^9 + \text{H}_a^9$), 2.27 (br.dddd, 1H, $J \approx 6.4, 6.4, 3.2$ and 3.2 Hz, H_e^5), 3.29 (d, 1H, $J = 14.3$ Hz, H^{12}), 3.81 (s, 3H, MeO), 3.83 (m, 1H, $J \approx 3.0$ and 3.0 Hz, H_e^1), 4.24 (br.d, 1H, $J = 14.3$ Hz, H^{12}), 4.29 and 4.41 (ABq, 2H, $J = 10.7$ Hz, $\text{CH}^{17}\text{H}^{17'\text{O}}$), 6.88 (ddd, 2H), 7.24 (br.d, 2H), 7.58 (dddd, 2H), 7.66 (dddd, 1H), 7.95 (br.d, 2H); ^{13}C NMR (100 MHz, CDCl_3 , δ): 22.16 (Me^{10}), 22.92 (Me^{11}), 23.53 (C^6), 24.57 (C^9), 29.86 (C^5), 30.76 (C^7), 61.10 (C^{12}), 62.91 (C^{17}), 75.94 (C^8), 80.84 (C^1), 82.71 (C^4), 113.78, 127.55, 128.72, 129.33, 131.26, 133.68, 141.17, 158.92. CI HRMS: obsd 447.18690, calcd for $\text{C}_{24}\text{H}_{31}\text{O}_6\text{S}$ ($M + 1$) 447.18414.

25. Nakajima, N.; Horita, K.; Abe, R.; Yonemitsu, O. *Tetrahedron Lett.* **1988**, 29, 4139.
26. Desjardins, R. E.; Canfield, C. J.; Haynes, J. D.; Chulay, J. D. *Antimicrob. Agents Chemother.* **1979**, 16, 710.
27. Milhous, W. K.; Weatherly, N. F.; Bowdre, J. H.; Desjardins, R. *Antimicrob. Agents Chemother.* 1985, 27, 525.
28. Posner, G. H.; Gonzales, L.; Cumming, J. N.; Klinedinst, D.; Shapiro, T. A. *Tetrahedron* **1997**, 53, 37.
29. For differentiation by ¹H NMR between epimers at C(4) **a** and **b** see reference 17.
30. Posner, G. H.; Ployapradith, P.; Hapangama, W.; Wang, D.; Cumming, J. N.; Dolah, P.; Kensler, T. W.; Klinedinst, D.; Shapiro, T. A.; Zheng, Q. Y.; Murray, C. K.; Pilkington, L. G.; Jayasinghe, L. R.; Bray, J. F.; Daughenbauth, R. *Bioorg. Med. Chem.* **1997**, 5, 1257.
31. Avery, M. A.; Bonk, J. D.; Chong, W. K. M.; Mehrotra, S.; Miller, R.; Milhous, W.; Goins, D. K.; Venkatesan, S.; Wyandt, C.; Khan, I.; Avery, B. A. *J. Med. Chem.* **1995**, 38, 5038.
32. Jung, M.; Lee, S. *Heterocycles* **1997**, 45, 1055.
33. Jefford, C. W.; Velarde, J. A.; Bernardinelli, G.; Bray, D. H.; Warhurst, D. C.; Milhous, W. K. *Helv. Chim. Acta* **1993**, 76, 2775.