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# Imidazo[1,2- $\alpha$ ]pyridines possess adenosine A<sub>1</sub> receptor affinity for the potential treatment of cognition in neurological disorders.

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### Abstract

Previous research has shown that bicyclic 6:5-fused heteroaromatic compounds with two N-atoms have variable degrees of adenosine  $A_1$  receptor antagonistic activity. Prompted by this imidazo[1,2- $\alpha$ ]pyridine analogues were synthesized and evaluated for their adenosine A<sub>1</sub> and A<sub>2A</sub> receptor affinity via radioligand binding studies and subjected to a GTP shift assay to determine its adenosine  $A_1$ receptor agonistic or antagonistic functionality. Imidazo[1,2- $\alpha$ ]pyridine, the parent scaffold, was found devoid of affinity for the adenosine  $A_1$  and  $A_{2A}$  receptors. The influence of substitution on position C2 showed no improvement for either adenosine  $A_1$  or  $A_{2A}$  receptor affinity. The addition of an amino or a cyclohexylamino group to position C3 also showed no improvement of adenosine  $A_1$  or  $A_{2A}$  receptor affinity. Surprisingly *para*-substitution on the phenyl ring at position C2 in combination with a cyclohexylamino group at position C3 led to adenosine A1 receptor affinity in the low micromolar range with compound 4d showing: (1) the highest affinity for the adenosine  $A_1$  receptor with a  $K_i$  value of 2.06  $\mu$ M and (2) adenosine A<sub>1</sub> receptor antagonistic properties. This pilot study concludes that *para*-substituted 3-cyclohexylamino-2-phenyl-imidazo[1,2- $\alpha$ ]pyridine analogues represent an interesting scaffold to investigate further structure-activity relationships in the design of novel imidazo[1,2- $\alpha$ ]pyridine-based adenosine A<sub>1</sub> receptor antagonists for the treatment of neurodegenerative disorders.

Keywords: Alzheimer's disease, Parkinson's disease, imidazo $[1,2-\alpha]$ pyridine analogues, Adenosine A<sub>1</sub> and A<sub>2A</sub> receptor antagonists

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Adenosine has many physiological functions throughout the body as well as a role in numerous nervous system disorders including cognitive disorders, epilepsy, ischemia, stroke, Alzheimer's disease (AD) and Parkinson's disease (PD) [1]. The effects of adenosine in neurodegenerative disorders are predominately mediated by the A<sub>1</sub> adenosine receptors (ARs) expressed in the hippocampus and the A<sub>2A</sub> ARs expressed in the striatum [2–4]. The A<sub>2A</sub> ARs are important for motor function [5] and display neuroprotective [6] and antidepressant effects [7], whereas the A<sub>1</sub> ARs play an important role in cognitive function [8]. Antagonism of the A<sub>1</sub> AR represents a beneficial strategy in the treatment of AD and PD since they have cognitive enhancing abilities that are still inadequately treated with available treatment in both disorders. Furthermore, the A<sub>1</sub> and A<sub>2A</sub> ARs exert a synergistic effect on motor function as the A<sub>1</sub> ARs enhance the presynaptic release of dopamine (DA) [9], while the A<sub>2A</sub> AR antagonists have valuable potential as treatment agents as it will synergistically improve motor function via A<sub>1</sub> and A<sub>2A</sub> AR antagonism (PD), enhance cognitive function through A<sub>1</sub> AR antagonism (AD and PD) and alleviate depressive symptoms via A<sub>2A</sub> AR antagonism.

Compounds with affinity for the  $A_1$  and  $A_{2A}$  ARs are generally divided into xanthine derivatives and the amino-substituted heterocyclic compounds (non-xanthine) [12]. The xanthine derivatives are one of the most researched AR classes and consist of a bicyclic fused 6:5-membered N-containing ring system. Various non-xanthine heterocyclic compounds have previously been documented to possess affinity for the  $A_1$  AR with the majority of these compounds considered as fused bi-or-tri-heterocyclic systems that are extensions of the xanthine scaffold [13]. The pyrazolo[1,5- $\alpha$ ]pyridines, imidazo[1,2- $\alpha$ ]pyridines and benzimidazoles (Figure 1) currently constitutes the largest group of non-xanthine 6:5fused heteroaromatic  $A_1$  AR antagonists [13]



Figure 1. The general scaffolds of bicyclic 6:5 fused N-containing heteraromatic compounds.

Compounds containing the imidazo  $[1,2-\alpha]$  pyridine ring system have been shown to possess a broad range of useful pharmacological properties, including antibacterial, antifungal, anthelmintic, antiviral, antiprotozoal, anti-inflammatory, anticonvulsant, anxiolytic, hypnotic (e.g., zolpidem), gastrointestinal, antiulcer and immunomodulatory activities [14-17]. Until recently imidazo[1,2- $\alpha$ )pyridines have only been described in patent literature for their selective A<sub>1</sub> AR antagonistic properties [18]. However, a study conducted by Reutlinger and co-workers [19] identified N-benzyl-2-phenylimidazo[1,2- $\alpha$ ]pyridin-3-amine as a potential A<sub>1</sub> AR ligand with 89% binding at 100  $\mu$ M (Figure 2). Based on the latter finding, the imidazo[1,2- $\alpha$ ]pyridines scaffold represent an interesting class of 6:5-fused bicyclic compounds and prompted the current pilot study to investigate the potential of several structurally related 3-cyclohexylamino-2-phenylimidazo[1,2-a]pyridine analogues possessing affinity for the  $A_1$  and  $A_{2A}$  ARs. Furthermore, in order to find new insights into the structural requirements for AR binding, substitution at position C2 and C3 was explored (Figure 2).





N-benzyl-2-phenylimidazo[1,2-α]pyridin-3-amine



**Figure 2.** N-benzyl-2-phenylimidazo[1,2- $\alpha$ ]pyridin-3-amine used as lead to investigate 3-cyclohexylamino-2-phenylimidazo[1,2- $\alpha$ ]pyridine analogues.

The selected imidazo[1,2- $\alpha$ ]pyridine analogues were either obtained commercially (1, 2a–d and 3) or synthesized (4a–i) according to literature procedures. Compound 1, imidazo[1,2- $\alpha$ ]pyridine, is seen as the parent compound of the study (Figure 1). The structure-activity relationships of the imidazo[1,2- $\alpha$ ]pyridine scaffold was further explored via substitution at the C2 position alone (2phenylimidazo[1,2- $\alpha$ ]pyridines, 2a–d) and in combination with position C3 (2-phenylimidazo[1,2- $\alpha$ ]pyridin-3-amine, 3 and 3-cyclohexylamino-2-phenylimidazo[1,2- $\alpha$ ]pyridines, 4a–i). The C2 substitution explored the unsubstituted phenyl ring (2a) as well as *para*-substitution which included OH- (2b), OCH<sub>3</sub>- (2c) and Br- (2d). The simultaneous substitution of C2 and C3 included C2 phenyl substitution combined with a C3 amino- (3) and C3 cyclohexyl amino-group (4a). The effect of *para*substitution at the C2 phenyl ring was further explored with the C3 cyclohexyl amino series and included OH- (4b) OCH<sub>3</sub>- (4c), CH<sub>3</sub>- (4d), Br- (4e), Cl- (4f), F- (4g) CF<sub>3</sub>- (4h) and NO<sub>2</sub>- (4i) (Figure 3; Table 1)



Figure 3. The imidazo $[1,2-\alpha]$  pyridine scaffold and suggested structural modifications.

Compounds 1, 2a–d and 3 were commercially available from Sigma-Aldrich<sup>®</sup> and subsequently used without any further purification. The synthetic route of the known synthesized imidazo[1,2- $\alpha$ ]pyridine-based derivatives (4a–i) is outlined in Scheme 1. Compounds 4a–i was successfully synthesized via a modified catalyst- and solvent-free three-compound procedure described in the literature [20]. Briefly, a mixture of 2-aminopyridine, the appropriate aldehyde and cyclohexylisocyanide was refluxed with no solvent at 120°C (4a–g and 4i) or 60°C (4h) until completion of the reaction, as indicated by TLC. The crude compounds were purified by recrystallization from either hexane:ethyl acetate (4a, 4d, 4e, 4f, 4g, 4h) or ethyl acetate:ethanol (4b, 4c, 4i) (Scheme 1). Compounds (4a–i) were successfully synthesized and the structures confirmed by NMR spectrometry and supported by MS results. Both the <sup>1</sup>H and <sup>13</sup>C NMR of each test compound corresponded with the proposed structures and the spectra reported in the literature [21].



X= H, OH, OCH<sub>3</sub>, CH<sub>3</sub>, Br, Cl, F, CF<sub>3</sub>, NO<sub>2</sub>

Scheme 1: The catalyst- and solvent-free synthetic procedure that was utilized to obtain the corresponding imidazo[1,2- $\alpha$ ]pyridines. Reagents and conditions: (a) heated at 120°C (4a, 4b, 4c, 4d, 4e, 4f, 4g, 4i) or 60°C (4h), (b) reflux for the appropriate time.

The  $A_1$  and  $A_{2A}$  AR affinity of all the test compounds (1, 2a-d, 3 and 4a-i) was determined via radioligand binding assays via a previously described procedure [22]. In short, the A1 AR affinity was determined in the presence of the radioligand [<sup>3</sup>H]-8-cyclopentyl-1,3-dipropylxanthine ([<sup>3</sup>H]DPCPX) with rat whole brain membranes expressing the A1 AR, while the A2A AR affinity was performed in the presence of 5'-N-[<sup>3</sup>H]-ethylcarboxamideadenosine ([<sup>3</sup>H]NECA) as radioligand with rat striatal membranes expressing the  $A_{2A}$  ARs. In order to minimize [<sup>3</sup>H]NECA's binding to the  $A_1$  AR the  $A_1$ AR agonist N<sup> $\circ$ </sup>-cyclopentyladenosine (CPA) was also included in the A<sub>2A</sub> AR radioligand binding experiment. Nonspecific binding was defined by the addition of 100 µM CPA. CPA was used as reference compound and its assay results confirmed validity of the radioligand binding assays. The  $K_i$ values were obtained by determining the  $IC_{50}$  values from sigmoidal-dose response curves by means of the Graphpad Software Inc. package. The corresponding  $K_i$  value for the competitive inhibition by the test compounds of [<sup>3</sup>H]DPCPX ( $K_d = 0.36$  nM) or [<sup>3</sup>H]NECA ( $K_d = 15.3$  nM) were subsequently calculated from the  $IC_{50}$  values [22]. The sigmoidal-dose response curves were obtained by plotting the specific binding (i.e. the eight concentrations of each test compound ranging between  $0 \ \mu M$  and 100  $\mu$ M) versus the logarithm of the test compound's concentrations [22]. The *in vitro* A<sub>1</sub> and A<sub>2A</sub> AR affinity results of the test compounds and CPA are summarized in Table 1.

### Table 1

The dissociation constant values ( $K_i$  values) for the binding of the imidazo[1,2- $\alpha$ ]pyridine analogues to rat adenosine A<sub>1</sub> and A<sub>2A</sub> receptors



 $K_i \pm \text{SEM} (\mu \text{M})^a / \% \text{ displacement}^b$ 

Compd	X	A <sub>1</sub> vs [ <sup>3</sup> H]DPCPX	A <sub>2A</sub> vs [ <sup>3</sup> H]NECA	$A_1 + GTP^c$ vs	GTP
)				[ <sup>3</sup> H]DPCPX	Shift <sup>d</sup>
1	-	> 100 (100%) <sup>b</sup>	> 100 (75%) <sup>b</sup>	-	-
2a	Н	$> 100 (49\%)^{b}$	$> 100 (52\%)^{b}$	-	-
<b>2b</b>	OH	> 100 (47%) <sup>b</sup>	> 100 (85%) <sup>b</sup>	-	-
2c	OCH <sub>3</sub>	> 100 (76%) <sup>b</sup>	> 100 (61%) <sup>b</sup>	-	-
2d	Br	> 100 (86%) <sup>b</sup>	$> 100 (87\%)^{b}$	-	-
3	-	$> 100 (52\%)^{b}$	$> 100 (86\%)^{b}$	-	-

		L	L.					
<b>4</b> a	Н	$> 100 (57\%)^{6}$	$> 100 (77\%)^{6}$	-	-			
<b>4</b> b	OH	$5.53 \pm 0.86^{a}$	> 100 (27%) <sup>b</sup>	-	-			
4c	OCH <sub>3</sub>	$7.61 \pm 1.25^{a}$	$> 100 (55\%)^{b}$	-	-			
<b>4d</b>	$CH_3$	$2.06 \pm 0.08^{a}$	> 100 (83%) <sup>b</sup>	$2.00 \pm 0.24^{a}$	0.97			
<b>4e</b>	Br	$3.90 \pm 0.65^{a}$	$> 100 (54\%)^{b}$	-	-			
<b>4f</b>	Cl	$> 100 (61\%)^{b}$	$> 100 (64\%)^{b}$	-	-			
4g	F	> 100 (63%) <sup>b</sup>	$> 100 (90\%)^{b}$	-	-			
4h	$CF_3$	> 100 (76%) <sup>b</sup>	> 100 (74%) <sup>b</sup>	-				
<b>4</b> i	$NO_2$	> 100 (99%) <sup>b</sup>	$> 100 (69\%)^{b}$	-	-			
СРА		0.015 <sup>e</sup>	0.331 <sup>e</sup>	0.99 <sup>e</sup>	6.48 °			
<sup>a</sup> All $K_i$ values determined in triplicate and expressed as mean $\pm$ SEM.								

An  $K_i$  values determined in unpricate and expressed as mean  $\pm$  SEIVI.

<sup>b</sup> Percentage displacement of the radioligand at the indicated concentration.

 $^{c}$  GTP shift assay, where the 100  $\mu M$  GTP was added to the  $A_{1}$  AR radioligand binding assay.

<sup>d</sup> GTP shifts calculated by dividing the  $K_i$  in the presence of GTP by the  $K_i$  in the absence of GTP.

<sup>e</sup> Literature values obtained from references [22].

Compound **4d** was further evaluated via a GTP shift assay to determine the functionality of the test compounds towards the  $A_1$  AR. The GTP shift assay was carried out as described previously with rat whole brain membranes and [<sup>3</sup>H]DPCPX in the absence and presence of 100  $\mu$ M GTP (Table 1; Figure 4). Nonspecific binding was defined by the addition of 10  $\mu$ M DPCPX (unlabeled). A compound with a calculated GTP shift of approximately 1 is considered an antagonist, in turn the presence of GTP affects the competition curve of an agonist and shifts the curve to the right [22].



**Figure 4.** The sigmoidal-dose response curves of compound **4d** (Panel A) and CPA (Panel B, reference agonist) displaying the binding affinity to  $A_1$  ARs in the absence (-) and presence (+) of GTP. (A) GTP shift of 0.97 calculated for the  $A_1$  AR antagonist compound **4d**, and (B) GTP shift of 6.48 calculated for the  $A_1$  AR agonist CPA.

Previous research [19] identified N-benzyl-2-phenylimidazo[1,2- $\alpha$ ]pyridin-3-amine as a lead for designing compounds with A<sub>1</sub> AR affinity (Figure 2). Imidazo[1,2- $\alpha$ ]pyridine (1), the parent scaffold of this pilot study, is unsubstituted at positions C2 and C3 and was devoid of A<sub>1</sub> and A<sub>2A</sub> AR affinity. The influence of C2 substitution on A<sub>1</sub> and A<sub>2A</sub> AR affinity was investigated by comparing compound

1 to compound  $2\mathbf{a}-\mathbf{d}$  containing a/an (un)substituted C2 phenyl ring. Although the specific binding percentages of these compounds improved compared to compound 1, they did not display improved A<sub>1</sub> and A<sub>2A</sub> AR affinity. The addition of an amino group at position C3 (3 *vs.* 2a) also showed no improvement for the A<sub>1</sub> and A<sub>2A</sub> AR affinity. Alternatively the combination of a *para*-substituted phenyl ring at position C2 together with a cyclohexylamino group at position C3 displayed A<sub>1</sub> AR affinity in the low micromolar range (4a vs. 4b, 4c, 4d, and 4e). However, no significant A<sub>2A</sub> AR affinity was obtained. It seems that the more bulky electron donating groups (-OH, -OCH<sub>3</sub>, -CH<sub>3</sub>) displayed good A<sub>1</sub> AR affinity, while the electron withdrawing groups (-Cl, -F, -CF<sub>3</sub>, -NO<sub>2</sub>) showed no A<sub>1</sub> AR affinity with the exception of the bromo group (4e) that possessed good A<sub>1</sub> AR affinity (Table 1).

The compound documented with the best A<sub>1</sub> AR affinity, was compound **4d** with a *para*-methyl substituent ( $K_i = 2.06 \mu$ M). Compound **4e** (*para*-bromo substituent) displayed the second highest A<sub>1</sub> AR affinity ( $K_i = 3.90 \mu$ M), followed by the *para*-hydroxy (**4b**) and *para*-methoxy (**4c**) analogues, exhibiting  $K_i$  values of 5.53  $\mu$ M and 7.61  $\mu$ M respectively for the A<sub>1</sub> AR. Unfortunately, none of these structural modifications favored A<sub>2A</sub> AR affinity. Furthermore, selective A<sub>1</sub> AR affinity was only obtained with *para*-substitution of the phenyl ring at position C2 in combination with a cyclohexylamino substitution on position C3, thus requiring substitution of both position C2 and C3 to obtain A<sub>1</sub> AR affinity

In order to demonstrate if the compound possessing the highest  $A_1$  AR binding affinity, compound 4d, acted as an antagonist or agonist, a GTP shift experiment was performed. Generally, a rightward shift of the binding curve in the presence of GTP (due to an uncoupling of the  $A_1$  AR from its  $G_i$  protein) is expected for an  $A_1$  AR agonist. In the case of an  $A_1$  AR antagonist, no significant shift is anticipated in the presence of GTP [22,23]. As expected compound 4d showed no significant shift of the binding curve in the presence of GTP, thus compound 4d may be considered an antagonist of the  $A_1$  AR (Table 1; Figure 4). Based on the structural similarity of compound 4d to compounds 4b, 4c and 4e, the latter compounds are also expected to act as  $A_1$  AR antagonists

For future optimization of the 3-cyclohexylamino-imidazo[1,2- $\alpha$ ]pyridine scaffold the results of a study by Novellino and co-workers [24] gave valuable insight for future scaffold modifications. They explored the A<sub>1</sub> AR affinity of novel N-alkyl- and N-acyl-(7-substituted-2-phenylimidazo[1,2- $\alpha$ ][1,3,5]triazin-4-yl)amines (ITAs). Their findings (Figure 5) highlighted the importance of a CH<sub>2</sub> spacer and CO linker. The CO linker enhanced A<sub>1</sub> AR affinity; however the affinity of the A<sub>2A</sub> AR was diminished. When an acetyl group was introduced at the N4 position, it reduced the binding affinity of the A<sub>1</sub> AR, while the binding affinity of the A<sub>2A</sub> AR remained unchanged. The A<sub>1</sub> AR affinity was improved when a CO-cyclohexyl and CO-phenyl group was introduced although the binding affinity of the A<sub>2A</sub> AR showed no improvement. The addition of a methylene spacer between the N4 position and the acetyl group enhanced A<sub>1</sub> AR affinity. When a CO-cyclopentyl was introduced, it enhanced the affinity for the A<sub>1</sub> AR and affinity for the A<sub>2A</sub> AR was gained. Substitution of the C7 position with a phenyl moiety rather than a methyl group combined with the CO-cyclopentyl maintained affinity for the A<sub>1</sub> AR, while A<sub>2</sub> AR was further increased [24].



Figure 5. Structural modification to ITA analogues to gain A1 and A2A AR affinity [24].

A possible explanation for our low AR affinity can be ascribed, in part, to the research done by Gillespie and co-workers [25], where the increased number of nitrogens in the heterocyclic ring (from a pyridine to a pyrimidine) enhanced both  $A_1$  and  $A_{2A}$  AR affinities. They compared the affinities of triazine, pyrimidine and pyridine scaffolds and observed that the pyridine scaffold was sevenfold less potent than the triazine and 45-fold less potent than the corresponding aminopyrimidine, thereby concluding that two nitrogens in the ring are optimum for both  $A_1$  and  $A_{2A}$  AR affinity [25,26]. This trend was also observed in the ITA analogues synthesized by Novellino and co-workers [24], where the 6:5 fused bicyclic rings containing four nitrogen atoms, displayed better AR affinity in comparison to the current investigated imidazo[1,2- $\alpha$ ]pyridine analogues containing only two nitrogens in the fused heterocyclic rings

Therefore we hypothesize that the introduction of additional nitrogens in the heterocyclic ring may increase both  $A_1$  and  $A_{2A}$  AR affinity (Figure 6). Moreover the findings of Novellino and co-workers [24] can be used for further structural modifications (Figure 6) to 3-cyclohexylamino-2-(4'-methylphenyl)imidazo[1,2- $\alpha$ ]pyridine (4d) to optimize the scaffold for improved  $A_1$  and  $A_{2A}$  AR affinity (Figure 6). These modifications include the replacement of the cyclohexyl ring with a cyclopentyl ring to increase both  $A_1$  and  $A_{2A}$  AR affinity and the insertion of a CO group and a methylene spacer between the NH and the CO group to increase  $A_1$  AR affinity.



Figure 6. Proposed future structural modifications to compound 4d.

In conclusion, the newly proposed structural optimization of the investigated imidazo[1,2- $\alpha$ ]pyridine analogues may be implemented in future studies to potentially improve the A<sub>1</sub> AR affinity and possibly gain A<sub>2A</sub> AR binding. Among the test compounds, **4d** possessing a *para*-substituted methyl functional group, was identified as the best selective A<sub>1</sub> AR antagonist ( $K_i = 2.06 \mu$ M) and may find therapeutic relevance to enhance PD-associated cognitive dysfunction. Since the study provided proof of concept that the imidazo[1,2- $\alpha$ ]pyridine derivatives possess A<sub>1</sub> AR affinity, this scaffold can thus be used for further structure-activity relationship studies to design novel imidazo[1,2- $\alpha$ ]pyridinebased A<sub>1</sub> AR antagonists for the potential treatment of cognition in neurological disorders.

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### Highlights:

- A<sub>1</sub> receptors are considered drug targets for neurological disorders.
- Known 3-cyclohexylamino-2-phenyl-imidazo[1,2-α]pyridines were synthesized.
- Para-substituted analogs favor A<sub>1</sub> affinity with diminished A<sub>2A</sub> affinity.
- Para-methyl-substituted compound 4d possess highest K<sub>i</sub> value of 2.06 μM.

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