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Synthesis of N-(2-(Methylamino)ethyl) Derivatives of 2H-Phthalazin-1-ones

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SYNTHESIS OF *N*-(2-(METHYLAMINO)ETHYL) DERIVATIVES OF 2*H*-PHTHALAZIN-1-ONES

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GRAPHICAL ABSTRACT



Abstract A series of new alkyl, tosyl, acetyl, and tert-butoxycarbonyl derivatives of 2-(2-aminoethyl)-phthalazinones were efficiently synthesized by reaction of lactams with N-Boc-, N-acetyl-, or N,O-ditosyl derivatives of N-methylethanolamine in the presence of MeONa or under Mitsunobu reaction conditions. Selected compounds were converted into corresponding 2-[2-(methylamino)ethyl]phthalazinones in good yields.

Keywords Alkylation; amines; Mitsunobu reaction; phthalazinones; rotamers

INTRODUCTION

2*H*-Pyridazin-3-ones and their benzo- (*phthalazinones*) and pyrido- (*pyridopir-ydazinones*) derivatives demonstrate a wide spectrum of biological properties. Compounds of these type show an interesting pharmacological action (e.g., adenosine A₁ receptor antagonist I,^[1] antinociceptive agents II,^[2–4] α_1 -adrenoreceptors antagonist III,^[5,6] and nonprostanoid PGI₂ agonist IV^[7]; Fig. 1).

Especially interesting group of 2*H*-pyridazin-3-one derivatives are compounds substituted at the lactam nitrogen atom by ω -aminoalkyl group. These type of

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Figure 1. Structures of biologically active 2H-pyridazin-3-one derivatives I-V.

compounds can exhibit analgesic as well as nonsteroidal anti-inflammatory activity Va-c.^[8-11]

In our previous articles we have described methodologies for the synthesis of novel (dimethylamino)methyl- and 2-(dimethylamino)ethyl- derivatives of phthalazinones and pyridopyridazinones (Scheme 1).^[11,12] The synthetic methods applied in the preparation of these compounds were based on the Mannich reaction and on the reaction of lactams 1 with 2-chloro-N, N-dimethylethanamine.

RESULTS AND DISCUSSION

In this article, we present methods for the preparation of 2-(*N*-methylamino)ethyl- derivatives of phthalazinones. Starting lactams **1** were synthesized from the appropriate 3-hydroxyisoindolinones or ketoacids upon reaction with hydrazine monohydrate, according to previously described methods.^[11–14]

Our preliminary studies have shown that preparation of 2-(ω -aminoalkyl) phthalazinones by reaction of corresponding 2-halogenethylamines **4** (Scheme 2) with lactams **1** is not effective. If primary or secondary ω -halogenalkylamines **4a,b** were used to react with phthalazinones **1**, the outcomes were rather poor.^[15]



Scheme 1. Synthesis of (N,N-dimethylamino)alkyl phthalazinones 2 and 3.^[11,12]

N-Alkylated products **5** were formed in trace amounts only, if at all. The formation of compounds **5** was observed by NMR spectroscopy and they were not isolated in individual form. Therefore, practical application of the ω -halogenalkylamines as alkylating agents is limited and determined by the degree of substitution of amine nitrogen atom. In practice, this methodology plays a significant role in preparation of 2-(*N*,*N*-dialkylamino)ethyl-lactam derivatives type **3**,^[11] solely (Scheme 1).

Based on these results, we focused on the synthesis of selected *N*-(2-aminoethyl)- phthalazinone derivatives substituted on the amine nitrogen atom by alkyl, tosyl, acetyl, and *tert*-butoxycarbonyl group. These derivatives can be effortlessly converted into the desired secondary amines (*N*-methylamines).^[16,17] Easily accessible derivatives of 2-(methylamino)ethanol, types **6**, **7**, and **8** (Fig. 2),^[18–20] were used as ω -aminoethylating agents.

Synthesis of *N*-2-(methylamino)ethyl lactams type **10** was carried out using two approaches as shown in Scheme 3. Initially, benzenesulfonamides **9** were synthesized, as precursors of corresponding 2-(methylamino)ethyl derivatives **10**.

Phthalazinones 1 after treatment with MeONa were converted into amides 9 by reaction with ditosylated 2-(methylamino)ethanol, in 40–64% yields. In the second



Scheme 2. Studies of synthesis of 2-(N-methylamino)ethyl- and 2-aminoethylphthalazinones 5.

stage benzenesulfonamides **9** were hydrolyzed to corresponding amines **10** by heating with concentrated H_2SO_4 (110 °C). Desired amines were obtained in 56–78% yields. The structures of compounds **9** and **10** were determined by infrared (IR), ¹H NMR, and elemental analysis or mass spectroscopy (Table 1).

Unexpectedly, synthesis of amine **10d** via hydrolysis of appropriate *N*-tosyl derivative with H_2SO_4 ended in failure. For this reason, the synthesis of amine **10d** was carried out using the Mitsunobu reaction^[21,22] as a key step, followed by deprotecion of the amine group. Alkylation of phthalazinone **1** with *N*-Boc-protected 2-(metylamino)ethanol using standard Mitsunobu conditions (TPP, DEAD) gave carbamic acid derivatives **11e** in satisfactory yields (Scheme 3). Cleavage of the Boc protecting group of compound **11e** with hydrochloric acid at rt gave the secondary amine **10d** in 55% yield. This protocol was successfully applied to



Figure 2. Structures of ω-aminoethylating agents.



Scheme 3. Synthesis of compounds 9, 10, and 11. Reagents and conditions: (i) (a) MeONa/MeOH, Δ 30 min, (b) TsOCH₂CH₂N(Me)Ts (6), Δ 7 h; (ii) H₂SO₄, Δ 6 h; (iii) Mitsunobu procedure: (a) TPP, DEAD, -10 °C, (b) HOCH₂CH₂N(Me)R³ 7 or 8, rt 20 h; (iv) HCl (20%_{aq}), MeOH, rt 30 h.

Entry	Compound	\mathbb{R}^1	\mathbb{R}^2	R ³	Yield (%)
1	9a	Н	Н		50
2	9b	Me	Н		45
3	9c	Ph	Н		64
4	9d	4-(MeO)-C ₆ H ₄	Н		47
5	9e	4-Cl-C ₆ H ₄	Н		40
6	9f	2Py	MeO		45
7	10a	H	Н		78
8	10b	Me	Н	_	65
9	10c	Ph	Н		56
10	10d	2Py	MeO	_	55
11	11a	Н	Н	Ac	30
12	11b	Н	Н	Boc	45
13	11c	Ph	Н	Ac	55
14	11d	Ph	Н	Boc	50
15	11e	2Py	MeO	Boc	55

 Table 1. Synthesis of compounds 9, 10, and 11

preparation of N-(*tert*-butoxycarbonyl) derivatives **11b,d** and N-acetyl analogs **11a,c** using N-(2-hydroxyethyl)-N-methylacetamide as a starting material, too. Therefore, presented methods for synthesis of N-methylaminoethyl derivatives of phthalazinones are complementary to one another but application of Mitsunobu methodology allows the reaction to be carried out under milder conditions.

CONCLUSIONS

In conclusion, we described an efficient synthesis of new *N*-substituted phthalazinones derivatives containing (2-{methyl[(4-methylphenyl)sulfonyl]amino}ethyl)- and {2-[(*tert*-butoxy carbonyl)(methyl)amino]ethyl}- moiety by reaction of series phthalazinones with readily accessible 2-(methylamino)ethanol derivatives. This, coupled with an effective conversion of compounds 9 and 11 to the corresponding 2-[2-(methylamino)ethyl]-2*H*-phthalazin-1-ones 10, should allow to access a wide variety of these types of compounds.

EXPERIMENTAL

The melting points were determined on a Boetius hot-stage apparatus and are uncorrected. ¹H NMR spectra were recorded at 200 MHz and ¹³C NMR spectra at 50 MHz on a Varian Gemini 200 BB spectrometer with tetramethylsilane (TMS) as an internal reference. IR spectra were recorded on a Nexus FT-IR spectrometer. Mass spectra analyses were performed on a MAT95-Finnigan mass spectrometer. The analytical thin-layer chromatography tests (TLC) were carried out on Merck silica gel plates (Kiselgel 60 F₂₅₄, layer thickness 0.2 mm) and the spots were visualized using an ultraviolet lamp.

Commercially available (Aldrich, Fluka) hydrazine monohydrate, N, N, N', N'-tetramethylethylenediamine, triphenylphosphite (TPP), diethyl azodicarboxylate (DEAD), 2-acetylbenzoic acid, and 2-benzoilbenzoic acid were used without further purification. Tetrahydrofuran (THF) was distilled from sodium benzophenone ketyl immediately before use. ω -Aminoethylating agents **6**, **7**, and **8** were prepared by standard methods^[18–20] from the commercially available 2-(methylamino)ethanol.

2*H*-Phthalazin-1-ones **1** were prepared according to an already reported procedure from appropriate 3-hydroxy-1*H*-isoindolin-1-ones or ketoacids.^[11–14]

General Procedure for Preparation of Benzenesulfonamides 9

2H-Phthalazin-1-one 1 (0.0153 mol) was added to a solution of sodium methoxide (0.0165 mol) in dry MeOH (70 ml). The mixture was heated to boiling for 30 min. Afterward, toluene-4-sulfonic acid 2-[methyl-(toluene-4-sulfonyl)-amino]ethyl ester (6) (0.0230 mol) was added and heating was continued for the next 7 h. After this time, the mixture was cooled to ambient temperature. The separated product was filtrated off, washed with dry methanol, and purified by crystallization or column chromatography.

The FTIR spectra of the benzenesulfonamides **9** displayed characteristic absorption of the C=O in the region $1640-1650 \text{ cm}^{-1}$. Besides, bands of SO₂ between 1150 and 1340 cm^{-1} were observed. The ¹H NMR spectra of amides **9** showed the presence of two singlets at 2.31–2.34 ppm and 2.91–2.93 ppm corresponding to methyl group of 4-toluenesulfonyl moiety and N-Me protons, respectively. The signals of the ethylene bridge were displayed as two triplets at 4.36–4.50 and 3.51–3.59 ppm.

4,*N*-Dimethyl-*N*-[2-(1-oxo-4-phenyl-1*H*-phthalazin-2-yl)-ethyl] benzenesulfonamide (9c)

Yield 64%; mp 128–131 °C (MeOH); FT-IR (KBr, cm⁻¹) ν : 1654 (C=O), 1336, 1163 (SO₂); ¹H NMR (200 MHz, CDCl₃, ppm) δ : 8.46–8.51 (m, 1H, 8-ArH),

7.82–7.72 (m, 3H, Ph, ArH), 7.67–7.52 (m, 7H; Ph, Ar, TsH), 7.15 (m, 2H, TsH), 4.48 (t, 2H, CH₂, J = 6.0 Hz), 3.59 (t, 2H, CH₂, J = 6.0 Hz), 2.93 (s, 3H, NMe), 2.31 (s, 3H, TsMe); ¹³C NMR (50 MHz, CDCl₃, ppm) δ : 159.1, 142.9, 135.0, 134.9, 132.7, 131.2, 129.4, 129.0, 128.5, 127.9, 127.0, 126.6, 48.1, 34.8, 21.4. Anal. calcd. for (C₂₄H₂₃N₃O₃S): C, 66.49; H, 5.35; N, 9.69; S, 7.40%. Found: C, 66.45; H, 5.42; N, 9.73; S, 7.47%.

General Procedure for Preparation of Acetyl- and *tert*-Butoxycarbonyl-Derivatives of 2-[(Methylamino)ethyl]-2*H*-phthalazin-1-ones 11 (Mitsunobu Procedure)

The Mitsunobu reaction was carried out under argon. DEAD (0.0102 mol, solution in toluene $c \approx 40\%$) was slowly added to a stirred solution of TPP (0.0102 mol) in dry THF (10 ml) at -10 °C. Then a solution of phthalazinone 1 (0.0068 mol) in THF (44 ml) was added dropwise. The whole lot was mixed for 15 min at -10 °C and next the appropriate derivative of *N*-methylethanolamine 7 or 8 (0.00748 mol) in THF (5 ml) was added at -10 °C. The mixture was stirred during 2 h at -10 °C, after which time the reaction mixture was warmed to ambient temperature and stirred in this condition for 20 h. All volatile materials were removed under reduced pressure, ethyl ether (20 ml) was added to the residue, and the whole lot was stirred for 0.5 h at ambient temperature. The separate white solid was collected by flirtation and washed with ether, and the filtrate was evaporated to dryness. The residue was subjected to column chromatography to give the pure product 11.

¹H NMR spectra of compounds **11** showed doubled singlets of methyl groups connected with nitrogen atoms and derived from acetyl or *tert*-butoxycarbonyl moieties. This fact may indicate that amides **11** coexist as mixtures of rotamers.^[23]

N-Methyl-*N*-[2-(1-oxo-4-phenyl-1*H*-phthalazin-2-yl)-ethyl]-acetamide (11c)

Yield: 55%; mp 138–140 °C; R_f =0.12 (CH₂Cl₂-acetone = 9:1); FT-IR (KBr, cm⁻¹) ν: 1652, 1649 (C=O); ¹H NMR (200 MHz, CDCl₃, ppm, mixture of rotamers) δ: 8.53–8.50 (m, 1H, ArH), 7.81–7.72 (m, 3H, ArH), 7.61–7.50 (m, 5H, ArH), 4.50 (m, 2H, CH₂), 3.93–3.77 (m, 2H, CH₂), 3.04, 3.02 (2 × s, 3H, NMe, two rotamers), 2.02, 1.97 (2 × s, 3H, Ac-Me, two rotamers); ¹³C NMR (50 MHz, CDCl₃, ppm, mixture of rotamers) δ: 171.1, 170.9, 159.4, 159. 3, 147.8, 147.1, 135.1, 134.8, 133.1, 132.8, 131.7, 131.3, 129.4, 129.3, 129.0, 128.7, 128.6, 127.1, 126.7, 49.0, 48.6, 46.2, 36.7, 33.5, 21.7, 20.9. Anal. calcd. for C₁₉H₁₉N₃O₂: C, 71.01; H, 5.96; N, 13.07%. Found: C, 70.88; H, 5.99; N, 12.81%.

General Procedure for Preparation of 2-[(Methylamino)ethyl]-2*H*-phthalazin-1-ones 10

Method A: Hydrolysis of benzenesulfonamides 9. The benzenesulfonamide **9** (0.0106 mol) and H_2SO_4 (98%, 0.0318 mol) were heated up to 100–110 °C over a period of 6 h. After this time the reaction mixture was cooled to ambient temperature, alkalized with 20% aqueous solution of NaOH, and next extracted with CH_2Cl_2 (3 × 30 ml). The combined extracts were dried over MgSO₄ and concentrated to dryness. The amine was separated by column chromatography.

Method B: Hydrolysis of *tert*-butyl methyl[2-(1-oxo-1*H*-phthalazin-2-yl) ethyl]-carbamates 11. Carbamate 11 (0.00275 mol) was added to a solution of HCl_{aq} (20%, 36 ml). The mixture was stirred for 30 h at room temperature. Next reaction mixture was alkalized with 20% aqueous solution of NaOH and extracted with CH_2Cl_2 (3 × 50 ml). The combined extracts were dried over MgSO₄ and concentrated under vacuum. The amine was isolated by column chromatography.

In the case of amines **10** FTIR spectra showed NH bands at $\approx 3320 \text{ cm}^{-1}$ and strong C=O bands at 1652 cm^{-1} . In the ¹H NMR spectra of amines **10** (Scheme 3) the signals of the 2-(methylamino)ethyl moiety were displayed as two triplets at ≈ 4.42 and $\approx 3.11 \text{ ppm}$ corresponding to CH₂ protons of ethylene chain, whereas the protons of NMe and NH groups exhibited as two singlets between 1.52 and 2.51 ppm.

2-[2-(Methylamino)-ethyl]-4-phenyl-2H-phthalazin-1-one (10c). Yield: 56% (method A); mp 96–98 °C; R_f =0.01 (AcOEt–MeOH = 1:1 next MeOH); FT-IR (KBr, cm⁻¹) ν : 1652 (C=O), 3332 (NH). ¹H NMR (200 MHz, CDCl₃, ppm) δ : 8.57–8.53 (m, 1H, ArH), 7.82–7.76 (m, 3H, ArH), 7.62–7.53 (m, 6H, ArH), 4.49 (t, 2H, CH₂, J=6.1 Hz), 3.16 (t, 2H, CH₂, J=6.1 Hz), 2.51 (s, 3H, NMe), 2.29 (s, 1H, NH); ¹³C NMR (50 MHz, CDCl₃, ppm) δ : 159.2, 146.9, 135.0, 132.6, 131.2, 129.3, 128.9, 128.4, 127.0, 126.5, 50.3, 50.2, 36.1. Anal. calcd. for (C₁₇H₁₇N₃O): C, 73.10; H, 6.13; N, 15.04%. Found: C, 73.11; H, 5.98; N, 15.11%.

6-Methoxy-2-[2-(methylamino)-ethyl]-4-(pyridin-2-yl)-2*H*-phthalazin-1one (10d). Yield: 55% (method B); mp 94–96 °C; R_f =0.0 (AcOEt next MeOH); FT-IR (KBr, cm⁻¹) ν: 1652 (C=O), 3316 (NH); ¹H NMR (200 MHz, CDCl₃, ppm) δ: 8.75–7.24 (m, 1H, ArH), 8.42 (d, 1H, ArH, *J*=8.8 Hz), 7.95–7.86 (m, 3H, ArH), 7.42–7.28 (m, 2H, ArH), 4.44 (t, 2H, CH₂, *J*=6.1 Hz), 3.88 (s, 3H, OMe), 3.10 (t, 2H, CH₂, *J*=6.1 Hz), 2.47 (s, 3H, NMe), 1.52 (brs, 1H, NH); ¹³C NMR (50 MHz, CDCl₃, ppm) δ: 162.9, 159.3, 154.9, 148.4, 143.3, 137.1, 130.4, 128.9, 124.2, 123.4, 121.9, 120.2, 108.6, 55.5, 50.3, 50.2, 36.0. Anal. calcd. for (C₁₇H₁₈N₃O₂): C, 65.79; H, 5.85; N, 18.05%. Found: C, 65.52; H, 5.78; N, 17.93%.

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SUPPORTING INFORMATION

Full experimental details, ¹H and ¹³C NMR, and IR spectra can be accessed on the publisher's website.

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