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IRAK-4 inhibitors. Part III: A series of imidazo[1,2-a]pyridines

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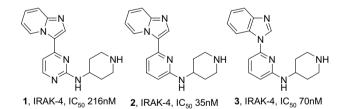
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Dedicated to the memory of our friend and colleague David Rainey.

Abstract—Following the identification of a potent IRAK inhibitor through routine project cross screening, a novel class of IRAK-4 inhibitor was established. The SAR of imidazo[1,2-*a*]pyridino-pyridines and benzimidazolo-pyridines was explored. © 2008 Elsevier Ltd. All rights reserved.

In the preceding article we demonstrated how a cross screening hit from another kinase programme yielded the potent imidazo[1,2-*a*]pyridino-pyrimidine IRAK-4 inhibitor 1.¹ This led to lead compounds 2 (imidazo[1,2-*a*]pyridino-pyridine) and 3 (benzimidazolo-pyridine) as our preferred scaffolds for optimisation.



Initially the SAR of the 5,6-fused bicyclic binding group was studied on the benzimidazole analogue series due to the simplicity of the chemistry and readily available substituted starting materials.² A series of analogues were prepared from commercially available benzimidazoles and 2,6-difluoro or 2,6-dibromopyridine by sequential S_NAr displacement reactions and separation of regioisomers as required (Scheme 1).

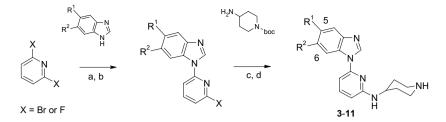
Substitution in the 6-position (\mathbf{R}^2) resulted in significant increase in IRAK-4 potency (5- to 25-fold) whereas substitution at the 5-position (\mathbf{R}^{1}) reduced potency irrespective of whether the 6-position was substituted (Table 1). With this in mind we embarked on the more involved synthesis of imidazo[1,2-a]pyridino-pyridines substituted in the regio-equivalent position (Scheme 2 and Table 2). A variety of groups were tolerated in this position, many giving low nanomolar enzyme inhibition potencies (selected examples shown, 12-24). The SAR was not very discernable except that the bulky tertiary amides were less potent than the primary amide (23 < 24 < 22). Homology model docking experiments of this series imply that the 6-substituent points towards the ATP-ribose region.¹ It is plausible that the flat SAR might be a result of the displacement of a bound water molecule from this site, thus gaining an entropic advantage in binding which would be largely independent of spatial complementarily or bonding interactions.

We also focused attention on the pyridine 2-position substituent. Much of the SAR could be probed by employing parallel synthesis (Scheme 3). The key bromopyridine intermediate **25** was prepared via Stille coupling³ of the tri-*n*-butylstannane formed in situ from trans-metallation of the Grignard derived from 3-bromo imidazo[1,2-*a*]pyridine. A variety of N-linked derivatives were prepared using an automated serial microwave reactor,⁴ either by direct S_NAr chemistry (method b) or by using Buchwald palladium catalysed

Keywords: IRAK; IRAK-4; IRAK-4 inhibitor; Kinase; Kinase inhibitor; Imidazo[1,2-*a*]pyridine; Inflammation; Anti-inflammatory; Ligand efficiency; IL-1; TNF-alpha; TNFα; Immunity.

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Scheme 1. Synthesis of substituted benzimidazolo-pyridines 3–11. Reagents and conditions: (a) NaH, NMP, 180 °C, microwave (33–90%); (b) HPLC separation of regioisomers as required; (c) 4-amino-1-Boc piperidine, NMP, 140 °C, microwave (13–75%); (d) HCl, Et₂O, DCM.

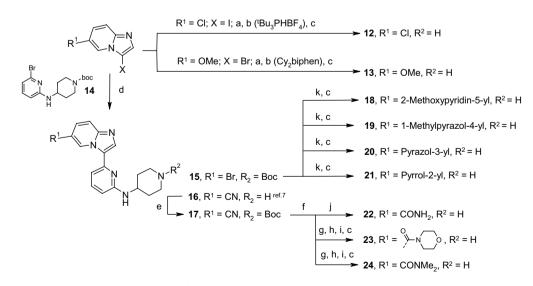
Table 1. IRAK-4 enzyme inhibition for benzimidazolo-pyridines 3-11

Compound	\mathbb{R}^1	R ²	IRAK-4 inhibition, IC ₅₀ (nM)			
3	Н	Н	70			
4 5	H OMe	OMe H	15 208			
6	OMe	OMe	181			
7	Н	Cl	3			
8	Cl	Н	112			
9	Cl	Cl	166			
10	Н	Me	10			
11	Me	Н	1085			

couplings (method c).⁵ Bifunctional bis-amines were suitably protected with *tert*-butoxycarbonyl groups. The products were purified by mass directed preparative HPLC⁶ following deprotection step (d) where required.

The series of aniline and heteroaryl analogues (26–29) showed some interesting SAR (e.g., 2-PhCN > Ph \sim 2-*N*-methylpyrazol-3-yl > 4-PhCN), albeit with modest potencies (Table 3). Some examples of secondary linked alkylamines indicated a preference for a free -NH-/NH₂ spaced approximately 2-3 carbon atoms from the linking amino group (e.g., 32 > 31 and 34 > 33). The IRAK-4 potency varied markedly with quite subtle changes to the positioning of the distal amino group. For example, the bicyclic analogue 34 was an order of magnitude less potent than 4-aminopiperidine, 2, whilst azetidine 37 and 3-(S)-piperidyl 39 had comparable potency. The 3-linked pyrrolidine rac-38 and methylene spaced 2-linked pyrrolidine rac-36 were \sim 10-fold less potent than 2 and the methylene spaced 4-linked piperidine 35 is 100-fold less potent. Tertiary N-linked alkylamines (40-43) demonstrated that a 2-aminopyridine free NH was not an essential prerequisite to binding, consistent with our earlier findings.¹ However, these compounds do not represent especially potent IRAK-4 inhibitors. Once again, the precise positioning and constraints of the distal free amino NH/NH₂ appeared to be important for good IRAK potency (rac-42 > 41).

Having assessed the SAR of substituents at different ends of the molecule the obvious next step was to combine preferred substituents in the same structure (Table

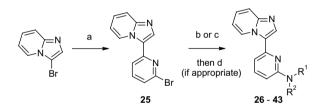


Scheme 2. Synthesis of 6-substituted imidazo[1,2-*a*]pyridin-3-yl-pyridin-2-ylamines, 12–24. Reagents and conditions: (a) *i*-PrMgCl, THF, -78 °C; *n*-Bu₃SnCl, -78 °C to rt; 2,6-dibromopyridine, Pd(PPh₃)₄, 140 °C, microwave (11–21%); (b) 4-amino-1-Boc piperidine, NaO^tBu, Pd(OAc)₂; *t*-Bu₃PHBF₄ DME, 140 °C, microwave (47%) or 2-(dicyclohexylphosphino)biphenyl (Cy₂biphen), toluene, 140 °C, microwave (44%); (c) HCl, Et₂O, DCM, MeOH (40–63%); (d) *i*-PrMgCl, THF, -78 °C; *n*-Bu₃SnCl, -78 °C to rt; 14, Pd(PPh₃)₄, reflux or 140 °C, microwave;⁷ (e) Boc₂O, Et₃N, DCM (61%); (f) KOH, THF, H₂O, 140 °C, microwave; (g) 20% (aq) H₂SO₄, 140 °C, microwave (57%); (h) Boc₂O, NaOH, dioxane, H₂O (96%); (i) morpholine or dimethylamine (2.0 M in THF), EDCI, HOBt, DCM (38–43%); (j) TFA, DCM (57%); (k) ArB(OH)₂ or ArB(OCMe₂)₂, Na₂CO₃, *n*-Bu₄NBr, Pd(PPh₃)₄, DME, H₂O, 140 °C, microwave (47–50%).

 Table 2. IRAK-4 enzyme inhibition for 6-substituted imidazo-[1,2a]pyridin-3-yl-pyridin-2-ylamines, 12–24

Table 3.	IRAK-4 enzyme inhibition for 2-amino substituted imidazo-	
pyridino	-pyridines 26–43	

Compound	\mathbf{R}^1	IRAK-4 inhibition, IC ₅₀ (nM)
12	Cl	1
13	OMe	6
16	CN	4
18	Meo	8
19	N.N.	7
20	∠_N H	2
21	NH	63
22	H ₂ N	5
23	N N	263
24	N 0	49



Scheme 3. Parallel synthesis of 2-amino substituted imidazopyridinopyridines 26–43. Reagents and conditions: (a) *i*-PrMgCl, THF, -78 °C; *n*-Bu₃SnCl, -78 °C to rt; 2,6-dibromopyridine, Pd(PPh₃)₄; 66 °C (71%); (b) R¹R²NH, Et₃N, *n*-BuOH, 150 °C, microwave; or (c) R¹R²NH, Pd(OAc)₂, Cy₂biphen, NaO'Bu, toluene, 150 °C, microwave; (d) HCl, Et₂O, DCM (77–97%).

4). The compounds were prepared by constructing the substituted imidazopyridine from condensing functionalised 2-aminopyridines with chloroacetaldehyde⁸ and halogenating regioselectively in the 3-position with *N*-iodo or *N*-bromosuccinmide.⁹ Once again, Stille crosscoupling and Buchwald conditions were used to construct the final compounds (Scheme 4).

In the examples given it was apparent that the SAR was not additive. Having a substituent in the imidazopyridine 6-position generally increased potency by 4- to 500-fold (44–47). This perhaps suggested the relative dominance of the 6-substituent in terms of binding over the pendant 2-amino heterocycles. However, compound 48 is an example where this is not the case. In the absence of a 6-substituent ($R^1 = H$, 39), potency is a respectable 19 nM. Introduction of the nitrile group in the 6-position dropped potency by ~5-fold. This high-

Compound	NR ¹ R ²	IRAK-4 inhibition, IC ₅₀ (nM)		
26	· · · · · · · · · · · · · · · · · · ·	1156		
27	· NH CN	433		
28	CN H	(12% inhibition at 10 μ M)		
29	N-N N-N H	1134		
30	N H N=N	497		
31	N N N	8100		
32	N NH ₂	1300		
33	H N H	914		
34	H NH H	225		
35	Ĩ. Ŋ	3030		
rac-36		313		
37	NH N	39		
rac-38	· N H	545		
abs-39	N ¹ ¹	19		
abs-40	H NH H	126		
41	NNH2	1880		
rac-42	N NH2	535		
43	N N NH	487		

 Table 4. IRAK-4 inhibition of hybridized substituted imidazo[1,2-a]pyridino-pyridines 44-48

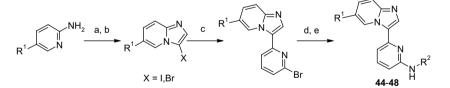
Compound	\mathbb{R}^1	\mathbb{R}^2	IRAK-4, IC ₅₀ (nM)			
44	Cl	NH	6			
rac-45	Cl	NH	1			
<i>rac</i> -46	Cl	NH	9			
47	OMe	\NH	7			
48	CN		95			

lights subtleties in SAR, where optimising two remote parts of a molecule independently sometimes results in non-additive effects.

Compounds that exhibited significant IRAK-4 potency were submitted into a screening cascade to build up a fuller profile and to prioritise compounds for in vivo studies (Table 5). This included an IRAK-1 enzyme assay.¹⁰ IRAK-1 is a related kinase homologue that is implicated in the same signalling pathway and hence a dual inhibitor of both IRAK-4 and IRAK-1 could offer synergistic benefits.¹¹ A variety of functional cell assays were also used to assess potency, such as LPS-induced TNF α cytokine release assay.¹² Potency in the IRAK enzyme assays tracked reasonably well with cytokine release in the cell assays. Good cellular IC₅₀s of 10s–100s nanomolar were often observed. The physicochemical profile of this series of compounds was generally good, with few solubility issues,¹³ and the in vitro DMPK looked promising with low to moderate microsomal clearance in both human and rat. Cytochrome P_{450} liabilities were not a major issue, with many compounds exhibiting only very weak CYP inhibition.¹⁴

We were encouraged by the fact that this series of compounds exhibited high enzyme potencies (frequently <10 nM at 10 μ M ATP), whilst still retaining low molecular weights (300–350). This represents high Ligand Efficiency (LE)¹⁵ and suggests further optimisation could be achieved, designing in selectivity elements whilst retaining IRAK potency and desirable physicochemical properties. For example, compound **16** has a LE of 0.52 (Table 5). As a point of reference the average kinase inhibitor has a LE of 0.3.¹⁶ The calculated Log *P* and PSA are all respectably low (*a*Log *P* < 3, PSA < 85Å²), giving plenty of scope for further elaboration.¹⁷

We have described how an active hit identified through routine cross screening was developed into a new class of IRAK-4 inhibitor. The SAR of the new lead series was explored with guidance from the IRAK-4 homology model developed in-house. Some highly potent (low nanomolar) IRAK-4 inhibitors with good cellular TNFα inhibition were prepared and were shown to have good drug-like properties and encouraging in vitro DMPK profiles. Initial kinase panel profiling and cell screening has indicated that some degree of off-target activity is still evident with this series, despite departing from the aminopyrimidine motif. However, the scope for fine tuning and optimising this potent and efficient class of IRAK inhibitor could lead to the generation of new therapeutic agents or useful tool compounds based on the blockade of the IL-1 driven immune response.



Scheme 4. Synthesis of hybridised substituted imidazo[1,2-*a*]pyridino-pyridines 44–48. Reagents and conditions: (a) chloroacetaldehyde, NaHCO₃, H₂O (45–55%); (b) NIS or NBS, MeCN (18–64%); (c) i-PrMgCl, THF, -78 °C; *n*-Bu₃SnCl, 66 °C; 2,6-dibromopyridine, Pd(PPh₃)₄, 150 °C, microwave; (28–44%); (d) Boc-R₂–NH₂, Pd(OAc)₂, dicyclohexyl-1-biphenylphosphine, NaO'Bu, microwave 140 °C toluene; (e) HCl, DCM, Et₂O, MeOH.

Table 5. Selected in vitro/cellular potency, in vitro DMPK data and calculated properties

Compound	MWt	2D	$a \operatorname{Log} P^{a}$	IRAK-4	Ligand	IRAK-1	TNFα	Mic. Clint,	Mic. Clint,	CYP 2D6	CYP 3A4
		PSA ^a	(Å ²)	IC ₅₀ (nM)	efficiency (LE) ^b	IC ₅₀ (nM)	inhibition (nM)	(hu) at 0.5 μM ^c	(rat) at 0.5 μM ^c	inhibition (µM)	inhibition (µM)
13	323	68.1	1.7	6	0.52	293	220	19.5	5.3	41% at 50 µM	31
16	318	82.6	1.6	4	0.54	96	200	8.3	39.9	42	22
44	300	55.5	2.3	6	0.48	195	54	13	0.01	21% at 10 µM	11
rac-45	314	60.1	2.3	1	0.48	23	16	19.2	49.3	15	12
rac-46	328	60.1	2.9	9	0.47	175	65	29.5	78.1	1.8	8.6

^a alog P and 2D PSA calculated with Scitegic Pipeline Pilot software. http://www.scitegic.com.

^b Ligand efficiencies approximated from IC₅₀ values: LE (kcal mol⁻¹) = { $[-RT \ln IC_{50} (mol L^{-1})]/#$ of non-hydrogen atoms}.¹⁵

Acknowledgements

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- 7. Under the high temperature (140 °C) Suzuki coupling conditions compound 16 ($R_1 = CN$) was formed directly with loss of the Boc protecting group whilst compound 15 ($R_1 = Br$) retained the Boc group. This may be due to relative substrate stabilities or subtle experimental variations between the two reactions. In any case, reinstallation and removal of the Boc groups was synthetically trivial.

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