Subscriber access provided by Iowa State University | Library

Structure Ligation Relationship of Amino Acids for the Amination Cross-Coupling Reactions

Gargi Nikhil Vaidya, Arif Khan, Hansa Verma, Sanjeev Kumar, and Dinesh Kumar

J. Org. Chem., Just Accepted Manuscript • Publication Date (Web): 08 Feb 2019

Downloaded from http://pubs.acs.org on February 8, 2019

Just Accepted

Note

"Just Accepted" manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides "Just Accepted" as a service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. "Just Accepted" manuscripts appear in full in PDF format accompanied by an HTML abstract. "Just Accepted" manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are citable by the Digital Object Identifier (DOI®). "Just Accepted" is an optional service offered to authors. Therefore, the "Just Accepted" Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the "Just Accepted" Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these "Just Accepted" manuscripts.



is published by the American Chemical Society. 1155 Sixteenth Street N.W., Washington, DC 20036

Published by American Chemical Society. Copyright © American Chemical Society. However, no copyright claim is made to original U.S. Government works, or works produced by employees of any Commonwealth realm Crown government in the course of their duties.

Structure Ligation Relationship of Amino Acids for the Amination Cross-Coupling Reactions

Gargi Nikhil Vaidya, Arif Khan, Hansa Verma, Sanjeev Kumar, Dinesh Kumar*

Department of Medicinal Chemistry,

National Institute of Pharmaceutical Education and Research (NIPER) – Ahmadabad, Palaj, Gandhinagar-382355, Gujarat, India



ABSTRACT: The **S**tructure Ligation Relationship (SLR) of amino acids (AA) for the cross-coupling aminations was examined. While AA ligated C–N cross-coupling under Pd- and Ni- catalysis were minor or ineffective, the AA ligated Cu-catalyzed C–N cross-couplings were promising particularly with the use of L-Methionine. The role –NH₂, -CO₂H, and –S- of L-Methionine were investigated and found critical for their ligation efficiency. The finding was compatible with aromatic as well as aliphatic amines including tautomerizable *N*-heteroarenes.

Introduction of nitrogen functionality into carbocyclic frameworks has been the hot area of research owing to the ubiquity of amino compounds in pharmaceuticals, agrochemicals, and functional materials.¹ While traditional approaches to construct C–N bonds often require multi-steps procedure, the metal-catalyzed C–N bond formations has opened a new avenue in context of efficiency and applicability.² In this regard, the cross coupling of amine nucleophiles with prefunctionalized arene electrophiles (Ullmann reaction³ and Buchwald–Hartwig reaction⁴) or nucleophiles (Chan–Lam reaction⁵) constitute the fundamental building blocks (Scheme 1). Recently, direct C–H aminations⁶ emerged as valuable strategy for C–N bond formation complementing the classical cross coupling procedures.

Scheme 1: Metal-catalyzed cross coupling C–N Bond Formation reactions.

Ullmann & Goldberg coupling Ar - X + HNBase Ar - NBuchwald & Hartwig coupling Ar - X + HN PdBase Ar - NChan-Lam coupling $Ar - B(OH)_2 + HN$ CuBase Ar - N CuBase

Amino acids (AAs), the fundamental building blocks of life, have been exploited well in organic synthesis and catalysis including asymmetric reactions⁷ and direct C-H functionalizations.⁸ The ease of availability and its derivatizations (cost-effectiveness), tolerance toward aerial oxidation & moisture, and lack of general toxicity is the key advantages. The use of amino acid ligands in crosscoupling reactions including C-N bond formation is documented,⁹ however their general Structure Ligation Relationship (SLR) or Structure Efficiency Relationship (SER) have been not investigated. In this context, the present work aims to evaluate AA ligations (SLR) on C–N bond formation under Cu-, Pd-, and Ni-catalysis and subsequently translate the optimal catalytic combinations for the general amination cross-couplings.

The study begun with evaluating different AAs for the C–N bond formation under metal catalysis (Scheme 2). For each AA, three parallel reactions were performed, representing Ni-, Pd-, and Cu-catalysis respectively. The reason being choosing indole as amine source is the challenges lie ahead in controlling selectivity as it represents a reactant with three distinct nucleophilic sites (NH, C₃–H, and C₂–H). We initially examined the cross-coupling of iodobenzene (1a) with indole (2a), employing Ni (II) as metal precursor. No C–N bond formation (3a) was observed in any case, the starting 2a was found intact (no C-arylation was observed as well).



Scheme 2. Evaluation of AAs ligation for C-N cross-

conditions. ^bIsolated yield. ^cNumber in parenthesis indicates the isolated yield of C_3 -arylated indole. Ni (II) = Ni(acac)₂; Cu (I) = Cul; Pd (II) = PdCl₂

Scheme 3. Ligation distinction of elected AAs for C–N cross-coupling reactions under metal catalysis.^{*a,b*}



Next, the AA ligation for C–N bond formation was examined under Pd (II)-catalysis. The results were quite interesting as we observed the formation of both, the C–N bond (see, L-Arg, L-Lys, L-Pro) and C–C bond (see, L-Asp), however the yields were very poor (< 10%). The

scope of AAs ligation for C–N bond formations under Cu (I)-catalysis was examined subsequently. The formation of C–N bond was observed with variable yields of **3a** in different AAs such as Gly (62%), L-Val (79%), L-Leu (81%), L-Met (92%), L-Arg (63%), L-Lys (83%), L-Tyrosine (72%), and L-Proline (76%). In order to obtain the ligation distinction between these elected AAs, cross coupling of **1a** with 1(2H)-phthalazinone (**4a**) were performed (Scheme 3). The results were promising, the use of L-Met was found distinctly superior over others, with 78% yield of **5a**.

The above studies concluded that AA ligation suits best with Cu-catalysis for C–N cross-coupling reactions, the L-Met being the distinctly superior over others notable AAs such as Gly, L-Val, L-Leu, L-Lys, L-Arg, L-Tyr, and L-Pro. The AA-ligated Pd-catalysis was capable of C–N bond formations; however selectivity was an issue as it forms the competitive side products (C–arylation) significantly. The AAs as ligand were ineffective for C–N cross-coupling reaction under Ni-catalysis.

For the catalytic system consisting of Cu (I)/ α -amino acid, rate and yields is induced by the structure of the α amino acid. The greater ability of L-Met, a sulfurcontaining AA, to promote coupling reactions might be dependent on the presence of thioether sulfur donors in its structure, in addition to the N and O donor groups, which may lead to more active and stable Cu (I) species in the reaction medium. To gain insight further, structural analogs of L-Met such as N-Acetyl-L-Met (L-Met-NAc), L-Met ethyl ester (L-Met-OEt), L-Met sulfoxide [L-Met-(S=O)], Seleno-L-Met (L-Met-Se), and L-Norleucine were investigated under optimized conditions to elucidate the role of -NH2, -CO2H, and -S- (Scheme 4). The yield of 3a was significantly decreased in case of L-Met-NAc (21%), L-Met-OEt (22%), L-Met-Se (20%), and L-norLeu (28%) suggesting the simultaneous involvement of free -NH2, -CO₂H, and –S-. The L-Met-(S=O) was ligable to provide excellent yield (78%), however inferior compared to L-Met (85 %). A similar yield was obtained using isomeric D-Met (83%) indicating chirality of AAs insignificant.

Scheme 4. Investigation of role of L-Met as ligand



Ligation of L-Met for the C–N cross-couplings with other metal-catalysis such as iron, manganese, rutheni-



um, rhodium were re-investigated; however, it was discouraging as no desired **3a** was formed in any case as shown in scheme 5.

Scheme 5. Re-investigation of L-Met ligation for cross-coupling amination reaction under Fe-, Mn-, Ru-, Rh- catalysis.



We took this opportunity to discover a general high yielding protocol for the synthesis of high value *N*-aryl indoles. A full optimization of different reaction parameters revealed the use of 5 mol % of CuI with 10 mol% of L-Met in presence of K_2CO_3 (2 equiv) at 100 °C in DMF was optimal with 92% yield of **3a**. Variations in the Cu-source using L-Met as ligand resulted inferior results (see ESI).

Scheme 6. C–N cross-couplings of various amines under optimized conditions unless otherwise noted.



With optimized conditions, the scope of developed protocol was examined. As summarized in Scheme 6,

various indoles bearing e-withdrawing groups, e-releasing groups, halogens etc. were reacted well with electronically different iodobenzenes affording excellent yields of *N*-arylated indoles. Wide range of functional groups (–OMe, –OBn, –F, –Br, –NO₂,–CF₃, –OCH₂O–) tolerated well, validating the robustness of protocol. The applicability of this protocol was further extended to other biologically relevant *N*-heteroarenes. Gratifyingly, *N*-arylation of electron-deficient tautomerizable bioactive *N*-heteroarenes such as phthalazine **5a**, quinazoline **5b**, and pyridine **5c** proceeded well with excellent yields and chemoselectivity. The efficiency of the developed protocol with aliphatic amines was showcased using representative examples (**5d** and **5e**) with excellent yields.

Scheme 7. Demonstration of gram scale reaction



To demonstrate the scalability and utility of the protocol, the reaction was also performed on a gram scale. Treatment of **2a** (1 g, 8.54 mmol) with **1a** under optimized condition resulted the formation of **3a** in 91%. A 52% of **3a** was obtained using 1 mol% of CuI and 2 mol% of L-Met (Scheme 7).

In conclusion, the present work reports the scope and limitations of the amino acids (AAs) ligation for the crosscoupling amination reactions under commonly exercised metal (Cu-, Pd-, & Ni-) catalysis. While AAs were ineffective ligand for C-N cross-coupling under Ni-catalysis, their ligation under Pd- and Cu-catalysis were capable for C-N bond construction. However, AA-ligated C-N crosscoupling under Pd-catalysis is markedly inferior compared to Cu-catalysis in term of chemoselectivity and yields. While several AAs were found effective for C-N cross-couplings under Cu-catalysis, the use of L-Met is distinctly superior and versatile. The finding was applied to construct the varieties of N-aryl indoles in excellent yields with high functional group tolerance. Notably, the methodology was found effective for C-N cross-couplings using electron deficient tautomerizable N-heteroarenes and aliphatic amines as well. The mechanistic studies indicated the simultaneous involvement of the NH2, -CO₂H, and –S- of L-Met is critical for effective ligation.

EXPERIMENTAL SECTION

Unless otherwise noted, all manipulations (reactions) were carried out in oven-dried glasswares under nitrogen conditions. All commercially obtained reagents/solvents were used as received; chemicals were purchased from Alfa Aesar[®], Sigma-Aldrich[®], Acros[®], and were used as received without further purification. ¹NMR and ¹³C NMR

spectra were recorded on Brukar 400 MHz and 100 spectrometers respectively using TMS as an internal standard. Both ¹H and ¹³C NMR chemical shifts were reported in parts per million downfield from tetramethylsilane (δ = o). Data for ¹H NMR are reported as chemical shift (δ ppm) with the corresponding integration values. Coupling constants (*J*) are reported in hertz (Hz). Standard abbreviations indicating multiplicity were used as follows: s (singlet), b (broad), d (doublet), t (triplet), q (quartet) and m (multiplet). Data for ¹³C NMR spectra are reported in terms of chemical shift (δ ppm). High-resolution mass spectra of unknown products were obtained at Agilent QTOFs spectrometer in positive (ESI⁺) ion mode.

1

2

3

4

5

6

7

8

9

10

11

12

13

14

15

16

17

18

19

20

21

22

23

24

25

26

27

28

29

30

31

32

33

34

35

36

37

38

39

40

41

42

43

44

45

46

47

48

49

50

51

52

53

54

55 56

57 58 59

60

Representative procedure for C-N bond formation: To dried tube equipped with a stir bar, CuI (1.90 mg, 0.01 mmol, 5 mol%), L-Met (2.98 mg, 0.02 mmol, 10 mol%), Indole 2a (23.43 mg, 0.2 mmol), iodobenzene 1a (33.57 µL, 61.20 mg, 0.3 mmol, 1.5 equiv), K2CO3 (55.28 mg, 0.4 mmol, 2.0 equiv) followed by DMF (1 mL) was added and the resulting reaction mixture was stirred at 100 °C (oil bath). After 24 h, the reaction mixture was cooled to rt, adsorbed on to silica gel and purified by column chromatography using petroleum ether (PE) and ethyl acetate (EA) (6/1, $R_f = 0.52$) as mobile phase to get analytically pure 3a^{10a} (35.57 mg, 92%) as white solid; ¹H NMR (400 MHz, CDCl₃): δ 7.93 (d, J = 7.1 Hz, 1H), 7.79 (d, J = 8.2 Hz, 1H), 7.70 - 7.65 (m, 4H), 7.54-7.49 (m, 2H), 7.47 – 7.38 (m, 2H), 6.90 (d, J = 3.2 Hz, 1H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ 140.00, 136.04, 129.81, 129.57, 128.15, 126.61, 124.52, 122.60, 121.38, 120.61, 110.75, 103.83; HRMS (ESI-TOF) m/z: $[M + H]^+$ Calcd for C₁₄H₁₂N 194.0964, Found 194.0966.

5-Methoxy-1-phenyl-1*H***-indole**^{10b} (**3b**): White solid (40.18 mg, 90%); purified by column chromatography using PE/EA: 9/1, R_f = 0.58; ¹H NMR (400 MHz, CDCl₃): δ 7.48 – 7.39 (m, 5H), 7.33 – 7.23 (m, 2H), 7.12 (d, *J* = 2.5 Hz, 1H), 6.86 (dd, *J* = 9.0, 2.5 Hz, 1H), 6.57 (dd, *J* = 3.3, 0.8 Hz, 1H), 3.83 (s, 3H); ¹³C [¹H] NMR (101 MHz, CDCl₃) δ δ154.7, 140.0, 131.1, 130.0,129.71, 128.4, 126.3, 124.08, 112.6, 111.5, 103.4, 102.8, 55.9; HRMS (ESI-TOF) *m*/*z*: [M + H]⁺ Calcd for C₁₅H₁₄NO 224.1075, Found 224.1078.

5-Nitro-1-phenyl-1H-indole^{10C} (**3c**): White solid (50.40 mg, 72%); purified by column chromatography using PE/EA: 9/1, R_f = 0.42; ¹H NMR (400 MHz, CDCl₃): δ 8.63 (d, J = 2.3 Hz, 1H), 8.10 (dd, J = 9.1, 2.3 Hz, 1H), 7.59 – 7.44 (m, 7H), 6.85 (d, J = 3.3 Hz, 1H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ 142.2, 138.7, 138.6, 131.3, 130.0, 128.5, 127.8, 124.7, 118.3, 117.9, 110.5, 105.6; HRMS (ESI-TOF) *m/z*: [M + H]⁺ Calcd for C₁₄H₄N₂O₃239.0821, Found 239.0807.

5-(Benzyloxy)-1-phenyl-1H-indole (3d): White solid (52.68 mg, 88%); purified by column chromatography using PE/EA: 6/1, R_f = 0.45; ¹H NMR (400 MHz, CDCl₃): δ 7.57 - 7.46 (m, 7H), 7.45 - 7.38 (m, 2H), 7.39 - 7.30 (m, 3H), 7.24 (d, J = 2.4 Hz, 1H), 6.99 (dd, J = 8.9, 2.5 Hz, 1H), 6.62 (dd, J = 3.3, 0.9 Hz, 1H), 5.16 (s, 2H); ¹³C {¹H} NMR (101 MHz, CDCL₃) δ 153.7, 139.9, 137.6, 131.2, 129.8, 129.6, 128.6, 128.4, 127.8, 127.6, 126.3, 124.1, 113.2, 111.4, 104.3, 103.3, 70.8;

HRMS (ESI-TOF) m/z: $[M + H]^+$ Calcd for $C_{21}H_{18}NO$ 300.1383, Found 300.1387.

1-(4-Nitrophenyl)-1H-indole^{10d} (**3e**): White solid (39.23 mg, 82%); purified by column chromatography using PE/EA: 9/1, R_f= 0.45; NMR (400 MHz, CDCl₃): δ 8.35 (d, *J* = 9.0 Hz, 2H), 7.69 (d, *J* = 7.0 Hz, 1H), 7.63 (d, *J* = 8.9 Hz, 3H), 7.36 (d, *J* = 3.4 Hz, 1H), 7.25 (dtd, *J* = 24.0, 7.1, 1.2 Hz, 2H), 6.76 (dd, *J* = 3.4, 0.9 Hz, 1H); ¹³C {¹H} NMR (101 MHz, CDCl₃) δ 145.2, 145.0, 135.3, 130.1, 127.1, 125.5, 123.5, 123.3, 121.7, 121.6, 110.5, 106.2; HRMS (ESI-TOF) *m/z*: [M + H]⁺ Calcd for C₁₄H₁N₂O₂239.0821, Found 239.0807.

1-(4-Fluorophenyl)-5-methoxy-1H-indole^{10e} (3f): White solid (41.00 mg, 85%); purified by column chromatography using PE/EA: 8/1, $R_f = 0.57$; ¹H NMR (400 MHz, CDCl₃): δ 7.44 – 7.38 (m, 2H), 7.36 (d, J = 9.0 Hz, 1H), 7.24 (d, J = 3.2 Hz, 1H), 7.21 – 7.15 (m, 2H), 7.13 (d, J = 2.5 Hz, 1H), 6.87 (dd, J = 9.0, 2.5 Hz, 1H), 6.59 (dd, J = 3.2, 0.9 Hz, 1H), 3.86 (s, 3H): ¹³C {¹H} NMR (101 MHz, CDCl₃): δ 160.9 (d, J = 246.0 Hz), 154.6, 136.0 (d, J = 2.9 Hz), 131.3, 129.7, 128.5, 125.9 (d, J = 8.4 Hz), 116.5 (d, J = 22.8 Hz), 112.6, 111.1, 103.3, 102.7, 55.9; HRMS (ESI-TOF) *m/z*: [M + H]⁺ Calcd for C₁₅H₁₃FNO 242.0981, Found 242.0977.

5-(Benzyloxy)-1-(4-fluorophenyl)-1H-indole (3g): White solid (56.47 mg, 89%); purified by column chromatography using PE/EA: 19/1, R_f = 0.56;¹H NMR (400 MHz, CDCl₃): δ 7.52 – 7.31 (m, 9H), 7.24 – 7.17 (m, 3H), 6.97 (d, *J* = 8.9 Hz, 1H), 6.60 (s, 1H), 5.14 (s, 2H); ¹³C {¹H} NMR (101 MHz, CDCl₃) δ 160.9 (d, *J* = 245.8 Hz), 153.7, 137.5, 135.9, 131.4, 129.6, 128.5, 128.4, 127.7, 127.4, 125.8 (d, *J* = 8.6 Hz), 116.4 (d, *J* = 22.9 Hz), 113.3, 110.9, 104.3, 103.2, 70.8; HRMS (ESI-TOF) *m/z*: [M + H]⁺ Calcd for C₂₁H₁₇FNO 318.1289, Found 318.1293.

1-(4-Fluorophenyl)-5-nitro-1H-indole^{10f} (**3h**): White solid (61.72 mg, 82%); purified by column chromatog-raphy using PE/EA: 9/1, $R_f = 0.47$; ¹H NMR (400 MHz, CDCl₃): δ 8.64 (d, J = 2.3 Hz, 1H), 8.11 (dd, J = 9.1, 2.3 Hz, 1H), 7.57 – 7.35 (m, 4H), 7.27 (t, J = 8.4 Hz, 2H), 6.86 (d, J = 3.4 Hz, 1H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ 161.8 (d, J = 248.3 Hz), 142.2, 138.9, 134.6 (d, J = 3.1 Hz), 131.4, 128.3, 126.6 (d, J = 8.5 Hz), 118.4, 118.1, 116.9 (d, J = 23.1 Hz), 110.2, 105.7; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₁₄H₁₀FN₂O₂ 257.0726, Found 257.0727.

5-Fluoro-1-(4-fluorophenyl)-1H-indole^{10g} (**3i**): White solid (37.13 mg, 81%); purified by column chromatography using PE/EA: 9/1, R_f = 0.5; ¹H NMR (400 MHz, CDCl₃): δ 7.39–7.26 (m, 4H), 7.24 (d, *J* = 3.3 Hz, 1H), 7.15 (t, *J* = 8.5 Hz, 2H), 6.92 (td, *J* = 9.0, 2.6 Hz, 1H), 6.59 (d, *J* = 3.2 Hz, 1H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ 161.2 (d, *J* = 246.7 Hz), 159.5, 157.1, 135.7 (d, *J* = 3.0 Hz), 132.8, 129.6, 126.2 (d, *J* = 8.5 Hz), 116.6 (d, *J* = 22.8 Hz), 111.0 (d, *J* = 9.0 Hz), 110.7, 105.9 (d, *J* = 23.5 Hz), 103.5 (d, *J* = 4.5 Hz); HRMS (ESI-TOF) *m/z*: [M + H]⁺Calcd for C₁₄H₁₀F₂N 230.0781, Found 230.0785

1

2

3

4

5

6

7

8

9

10

11

12

13

14

15

16

17

18

19

20

21

22

23

24

25

26

27

28

29

30

31

32

33

34

35

36

37

38

39

40

41

42

43

44

45

46

47

48

49

50

51

52

53

54

55 56

57 58 59

60

5-Methoxy-1-(4-(trifluoromethyl)phenyl)-1H-

indole (**3***j*): White solid (45.83 mg, 79%); purified by column chromatography using PE/EA: 19/1, R_f = 0.56; ¹H NMR (400 MHz, CDCl₃): δ 7.78 (d, *J* = 8.3 Hz, 2H), 7.61 (d, *J* = 8.3 Hz, 2H), 7.51 (d, *J* = 9.0 Hz, 1H), 7.33 (d, *J* = 3.3 Hz, 1H), 7.17 (d, *J* = 2.5 Hz, 1H), 6.93 (dd, *J* = 9.0, 2.5 Hz, 1H), 6.67 (d, *J* = 3.3 Hz, 1H), 3.90 (s, 3H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ 154.9, 142.9, 130.6, 130.4, 127.85 (q, *J* = 32.6 Hz), 127.82, 126.8 (q, *J* = 3.8 Hz), 124.0 (q, *J* = 275 Hz), 123.3, 112.9, 111.2, 104,6, 103.0, 55.8; HRMS (ESI-TOF) *m*/*z*: [M + Na]⁺ Calcd for C₁₆H₁₂F₃NONa 314.0763, Found 314.0769.

5-Bromo-1-(4-fluorophenyl)-1H-indole^{10h} (3k): White solid (50.80 mg, 85%); purified by column chromatography using PE/EA: 19/1, R_f = 0.57; ¹H NMR (400 MHz, CDCl₃): δ 7.82 (s, 1H), 7.43 (dd, J = 8.9, 4.7 Hz, 2H), 7.35 – 7.30 (m, 2H), 7.28 (d, J = 3.3 Hz, 1H), 7.22 (t, J = 8.5 Hz, 2H), 6.62 (d, J = 3.2 Hz, 1H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ 161.3 (d, J = 247.0 Hz), 135.4 (d, J = 3.1 Hz), 134.8, 130.8, 129.2, 126.2 (d, J = 8.5 Hz), 125.3, 123.6, 116.6 (d, J = 22.8 Hz), 113.6, 11.7, 103.1; HRMS (ESI-TOF) *m*/*z*: [M + H]⁺Calcd for C₁₄H₁₀BrFN 298.9981, Found 298.9985.

5-Methoxy-1-(4-nitrophenyl)-1H-indole¹⁰ⁱ (31): White solid (41.81 mg, 78%); purified by column chromatography using PE/EA: 4/1, R_f = 0.47; ¹H NMR (400 MHz, CDCl₃): δ 8.40 (d, J = 9.1 Hz, 2H), 7.67 (d, J = 8.4 Hz, 2H), 7.56 (d, J = 8.9 Hz, 1H), 7.36 (d, J = 3.4 Hz, 1H), 7.15 (d, J = 2.6 Hz, 1H), 6.94 (dd, J = 8.9, 2.6 Hz, 1H), 6.70 (d, J = 3.4 Hz, 1H), 3.89 (s, 3H); ¹³C {¹H} NMR (101 MHz CDCl₃): δ 155.3, 145.4, 144.8, 130.9, 130.3, 127.5, 125.6, 122.8, 113.2, 111.4, 106.0, 103.3, 55.8; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₁₅H₁₃N₂O₃ 269.0926, Found 269.0926

5-(Benzyloxy)-1-(4-nitrophenyl)-1H-indole (3m): White solid (55.50 mg, 83%); purified by column chromatography using PE/EA: 12/1, R_f = 0.5; ¹H NMR (400 MHz, CDCl₃): δ 8.37 (d, *J* = 8.9 Hz, 2H), 7.63 (d, *J* = 8.9 Hz, 2H), 7.56 (d, *J* = 9.0 Hz, 1H), 7.50 (d, *J* = 7.1 Hz, 2H), 7.42 (t, *J* = 7.3 Hz, 2H), 7.38 – 7.32 (m, 2H), 7.23 (d, *J* = 2.4 Hz, 1H), 7.03 (dd, *J* = 9.0, 2.5 Hz, 1H), 6.70 (d, *J* = 3.3 Hz, 1H), 5.15 (s, 2H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ 154.5, 145.3, 144.8, 137.3, 130.9, 130.4, 128.6, 128.0, 127.6, 127.5, 125.6, 122.8, 113.9, 111.4, 106.1, 104.9, 70.7; HRMS (ESI-TOF) *m/z*: [M + H]⁺ Calcd for C₂₁H₁₇N₂O₃ 345.1234, Found 345.1236.

5-Fluoro-1-(4-nitrophenyl)-1*H***-indole (3n):** White solid (43.55 mg, 85%); purified by column chromatography using PE/EA: 6/1, $R_f = 0.5$; ¹H NMR (400 MHz, CDCl₃): δ 8.43 (d, *J* = 9.0 Hz, 2H), 7.69 (d, *J* = 9.0 Hz, 2H), 7.58 (dd, *J* = 9.0, 4.3 Hz, 1H), 7.44 (d, *J* = 3.4 Hz, 1H), 7.36 (dd, *J* = 9.1, 2.5 Hz, 1H), 7.06 (td, *J* = 9.0, 2.6 Hz, 1H), 6.76 (dd, *J* = 3.3, 0.8 Hz, 1H); ¹³C {¹H} NMR (101 MHz CDCl₃): δ 158.6 (d, *J* = 237.5 Hz), 145.1 (d, *J* = 26.3 Hz), 131.8, 130.7 (d, *J* = 10.2 Hz), 128.7, 125.6, 123.4, 111.8, 111.5, 111.3 (d, *J* = 9.6 Hz), 106.6 (d, *J* = 23.5 Hz), 105.9 (d, *J* = 4.4 Hz); HRMS (ESI-TOF) *m*/*z*: [M + Na]⁺ Calcd for C₁₄H₉FN₂O₂Na 279.0546, Found 279.0549.

1-(Benzo[d][1,3]dioxol-5-yl)-5-bromo-1*H*-indole

(30): White solid (51.00 mg, 81%); purified by column chromatography using PE/EA: 30/1, $R_f = 0.5$; ¹H NMR (400 MHz, CDCl₃): δ 7.82 (d, J = 1.9 Hz, 1H), 7.38 – 7.21 (m, 4H), 6.97 – 6.88 (m, 3H), 6.60 (dd, J = 3.2, 0.8 Hz, 1H), 6.09 (s, 2H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ 148.5, 146.6, 135.0, 133.4, 130.6, 129.4, 125.1, 123.5, 118., 113.4, 111. 9, 108.7, 106.2, 102.5, 101.8; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for $C_{15}H_{11}BrNO_{2}$ 315.9968, Found 315.9972.

1-(Benzo[d][1,3]dioxol-5-yl)-5-methoxy-1H-indole

(**3p**): White solid (43.83 mg, 82%); purified by column chromatography using PE/EA: 30/1, R_f = 0.5; ¹H NMR (400 MHz, CDCl₃): δ 7.39 (d, *J* = 8.9 Hz, 1H), 7.24 (d, *J* = 3.2 Hz, 1H), 7.13 (d, *J* = 2.5 Hz, 1H), 6.97 (d, *J* = 1.7 Hz, 1H), 6.92 (d, *J* = 2.0 Hz, 2H), 6.88 (dd, *J* = 9.0, 2.6 Hz, 1H), 6.57 (d, *J* = 3.1 Hz, 1H), 6.05 (s, 2H), 3.88 (s, 3H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ 154.5, 148.3, 146.2, 134.1, 131.5, 129.5, 128.7, 117.7, 112.5, 111.2, 108.6, 106.1, 102.7, 101.7, 55.9, 29.7; HRMS (ESI-TOF) *m*/*z*: [M + H]⁺ Calcd for C₁₆H₁₄NO₃ 268.0968, Found 268.0969.

2-Phenylphthalazin-1(*2H*)-**one**^{11a} (**5a**): White solid (36.00 mg, 81%); purified by column chromatography using PE/EA: 3/1, $R_f = 0.40$; ¹H NMR (400 MHz, CDCl₃): δ 8.50 (dt, J = 7.6, 1.0 Hz, 1H), 8.28 (d, J = 0.8 Hz, 1H), 7.88 – 7.76 (m, 2H), 7.76 – 7.70 (m, 1H), 7.67 (dd, J = 8.6, 1.2 Hz, 2H), 7.49 (t, J = 7.8 Hz, 2H), 7.38 (t, J = 7.4 Hz, 1H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ 159.2, 141.9, 138.5, 133.5, 132.0, 129.5, 128.8, 128.6, 127.8, 127.3, 126.2, 125.7; HRMS (ESI-TOF) *m*/*z*: [M + H]⁺ Calcd for C₁₄H₁₁N₂O 223.0871, Found 223.0873.

3-Phenylquinazolin-4(3*H***)-one^{11b} (5b):** White solid (35.55 mg, 80%); purified by column chromatography using PE/EA: 1/1, $R_f = 0.52$; ¹H NMR (400 MHz, CDCl₃): δ 8.37 (d, *J* = 7.5 Hz, 1H), 8.14 (s, 1H), 7.84 – 7.74 (m, 2H), 7.59 – 7.52 (m, 3H), 7.52 – 7.47 (m, 1H), 7.46 – 7.40 (m, 2H); ¹³C {¹H} NMR (101 MHz, ,CDCl₃): δ 160.7, 147.7, 146.1, 137.5, 134.6, 129.7, 129.2, 127.7, 127.5, 127.2, 127.0, 122.3; HRMS (ESI-TOF) *m/z*: $[M + H]^+$ Calcd for C₁₄H₁₁N₂O 223.0871, Found 223.0873.

1-Phenylpyridin-2(1*H***)-one^{11C} (5c):** White solid (28.00 mg, 82%); purified by column chromatography using PE/EA: 1/9, $R_f = 0.52$; ¹H NMR (400 MHz, CDCl₃): δ 7.49 (t, *J* = 7.4 Hz, 2H), 7.45 – 7.35 (m, 4H), 7.33 (d, *J* = 6.9 Hz, 1H), 6.67 (d, *J* = 9.2 Hz, 1H), 6.24 (td, *J* = 6.7, 1.4 Hz, 1H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ 162.4, 140.9, 139.9, 138.0, 129.4, 128.5, 126.5, 121.9, 105.9; HRMS (ESI-TOF) *m/z*: [M + H]⁺ Calcd for C₁₁H₁₀NO 172.0762, Found 172.0763.

4-Phenylmorpholine^{11d} (**5d**): White solid (23.50 mg, 72%); purified by column chromatography using PE/EA: 4/1, R_f = 0.4;¹H NMR (400 MHz, CDCl₃): δ7.29 (dd, *J* = 8.7, 7.3 Hz, 2H), 6.91 (dd, *J* = 18.4, 7.7 Hz, 3H), 3.87 (t, *J* = 9.6 Hz, 4H), 3.16 (t, *J* = 10 Hz, 4H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ 151.2, 129.2, 120.1, 115.7, 66.9, 49.4; HRMS (ESI-TOF) *m*/*z*: [M + H]⁺Calcd for C₁₀H₁₄NO 164.1075, Found 164.1077.

58 59

60

Page 6 of 8

N-(**3**-**Phenylpropyl)aniline**¹¹¹² (**5e**): White solid (33.80 mg, 75%); purified by column chromatography using PE/EA: 9/1, R_f = 0.60; ¹H NMR (400 MHz, CDCl₃): δ 7.35 – 7.27 (m, 2H), 7.24 – 7.15 (m, 5H), 6.71 (t, *J* = 7.3 Hz, 1H), 6.61 (d, *J* = 7.7 Hz, 2H), 3.15 (t, *J* = 6.9 Hz, 2H), 2.68 (t, *J* = 7.5 Hz, 2H), 1.92 – 1.58 (m, 4H); ¹³C {¹H} NMR (101 MHz, CDCl₃): δ 145.9, 139.7, 126.7, 125.9, 125.8, 123.3, 114.6, 110.2, 41.3, 33.1, 26.6, 26.4; HRMS (ESI-TOF) *m*/*z*: [M + H]⁺ Calcd for C₁₅H₁₈N 212.1439, Found 212.1436

3-Phenyl-1H-indole^{nf} as white solid; purified by column chromatography using PE/EA: 4/1, R_f = 0.35; ¹H NMR (400 MHz CDCl₃): δ 8.16 (s, 1H, NH), 7.93 (d, *J* = 6.0 Hz, 1H), 7.66 (d, *J* = 7.2 Hz, 1H), 7.46-7.38 (m, 3H), 7.32 (d, *J* = 2.0 Hz, 1H), 7.30-7.17 (m, 4H); ¹³C {¹H} NMR (100 MHz, CDCl₃): δ 136.7, 135.6, 128.7, 127.5, 126.0, 125.7, 122.5, 121.8, 120.4, 119.8, 116.3, 111.4; HRMS (ESI-TOF) *m/z*: [M + H]⁺Calcd for C₁₄H₁₂N 194.0964, Found 194.0967.

Representative procedure for the gram-scale reaction: To dried tube equipped with a stir bar, CuI (81.32 mg, 0.43 mmol, 5 mol%), L-Met (126.82 mg, 0.85 mmol, 10 mol%, 2 equiv), Indole **2a** (1.0 g, 8.54 mmol), iodobenzene **1a** (1.43 mL, 2.61 g, 12.81 mmol, 1.5 equiv), K_2CO_3 (2.36 g, 17.08 mmol, 2.0 equiv) followed by DMF (10 mL) was added and the resulting reaction mixture was stirred at 100 °C (oil bath). After 24 h, the reaction mixture was cooled to rt, worked-up with water-EtOAC system, dried using anhydrous MgSO₄, adsorbed on to silica gel and purified by column chromatography (PE/EA: 6/1, R_f = 0.52) to get analytically **3a** (1.48 g, 91%) as white solid.

ASSOCIATED CONTENT

Supporting Information. Few experimental details and copy of ¹H NMR & ¹³C spectra. This material is available free of charge via the Internet at http://pubs.acs.org.

AUTHOR INFORMATION

Corresponding Author

*Email: dineshk@niperahm.ac.in

ORCID: Dinesh Kumar: <u>0000-0003-2680-881X</u>

Notes

The authors declare no competing financial interest

ACKNOWLEDGMENT

We acknowledge the generous financial support from Department of Pharmaceuticals (DoP), Ministry of Chemical & Fertilizer, Gol. DK hearty acknowledge Prof. Kiran Kalia, Director NIPER-A for her constant support, encouragement, and motivation. DK also gratefully acknowledge the DST-SERB for the award of Ramanujan Fellowship (File No. SB/S2/RJN-135/2017). Authors thank Ms. Rajeshwari Rathod for recoding the HRMS spectra.

REFERENCES

(1) (a) Ricci, A. Amino Group Chemistry: From Synthesis to the Life Sciences; Wiley-VCH: Weinheim, **2008**. (b) Lawrence, S. A.,

Amines: Synthesis Properties and Applications. Cambridge University Press: Cambridge, **2004**; pp 265. (c) Craig, P. N. In Comprehensive Medicinal Chemistry; Drayton, C. J., Ed.; Pergamon Press: New York, 1991. (c) Hili, R.; Yudin, A. K. Making Carbon-Nitrogen Bonds in Biological and Chemical Synthesis. *Nat. Chem. Biol.* **2006**, *2*, 284-287. (d) Liang, M.; Chen, J. Arylamine Organic Dyes for Dye-Sensitized Solar Cells. *Chem. Soc. Rev.* **2013**, *42*, 3453-3488.

(2) Recent review: (a) Park, Y.; Kim, Y.; Chang, S. Transition Metal-Catalyzed C-H Amination: Scope, Mechanism, and Applications. *Chem. Rev.* **2017**, *117*, 9247-9301. (b) Hendrick, C. E.; Wang, Q. Emerging Developments using Nitrogen-Heteroatom Bonds as Amination Reagents in the Synthesis of Aminoarenes. *J. Org. Chem.* **2017**, *82*, 839-847. (c) Ruiz-Castillo, P.; Buchwald, S. L. Applications of Palladium-Catalyzed C-N Cross-Coupling Reactions. *Chem. Rev.* **2016**, *116*, 12564-12649. (d) Jiao, J.; Murakami, K.; Itami, K. Catalytic Methods for Aromatic C-H Amination: An Ideal Strategy for Nitrogen-Based Functional Molecules. *ACS Catal.* **2016**, *6*, 610-633. (e) Kim, H.; Chang, S. Transition-Metal-Mediated Direct C-H Amination of Hydrocarbons with Amine Reactants: The Most Desirable but Challenging C-N Bond-Formation Approach. *ACS Catal.* **2016**, *6*, 2341-2351.

(3) (a) Li, Y.; Peng, J.; Chen, X.; Mo, B.; Li, X.; Sun, P.; Chen, C. Copper-Catalyzed Synthesis of Multisubstituted Indoles through Tandem Ullmann-Type C–N Formation and Crossdehydrogenative Coupling Reactions. J. Org. Chem. 2018, 83, 5288-5294. (b) Lo, Q. A.; Sale, D.; Braddock, D. C.; Davies, R. P. Mechanistic and Performance Studies on the Ligand-Promoted Ullmann Amination Reaction. ACS Catal. 2018, 8, 101-109. (c) Zhou, W.; Fan, M.; Yin, J.; Jiang, Y.; Ma, D. Cul/Oxalic Diamide Catalyzed Coupling Reaction of (Hetero)Aryl Chlorides and Amines J. Am. Chem. Soc. 2015, 137, 11942-11945. (d) Ziegler, D. T.; Choi, J.; Muñoz-Molina, J. M. Bissember, A. C.; Peters, J. C.; Fu, G. C. A Versatile Approach to Ullmann C-N Couplings at Room Temperature: New Families of Nucleophiles and Electrophiles for Photoinduced, Copper-Catalyzed Processes. J. Am. Chem. Soc. 2013, 135, 13107-13112. (e)Giri, R.; Hartwig, J. F. Cu(I)-Amido Complexes in the Ullmann Reaction: Reactions of Cu(I)-Amido Complexes with Iodoarenes with and without Autocatalysis by Cul. J. Am. Chem. Soc. 2010, 132, 15860-15863. (f) Gujadhur, R. K.; Bates, C. G.; Venkataraman, D. Formation of Aryl-Nitrogen, Aryl-Oxygen, and Aryl-Carbon Bonds Using Well-Defined Copper(I)-Based Catalysts. Org. Lett. 2001, 3, 4315-4317. (g) Bhunia, S.; Pawar, G. G.; Kumar, S. V.; Jiang, Y.; Ma, D. Selected copper-based reactions for C-N, C-O, C-S, and C-C Bond Formation. Angew. Chem. Int. Ed. 2017, 56, 16136-16179. (h) Ma, D.; Cai Q. L-Proline Promoted Ullmann-Type Coupling Reactions of Aryl Iodides with Indoles, Pyrroles, Imidazoles or Pyrazoles. Synlett 2004, 128-130. (i) Liu, J.; Yan, J.; Qin, D.; Cai, Q. An Unexpected Inversion of Enantioselectivity in a Copper-Catalyzed Intramolecular- Desymmetric Aryl C-N Coupling Reaction. Synthesis 2014, 46, 1917-1923.

(4) (a) Dennis, J. M.; White, N. A.; Liu, R. Y.; Buchwald, S. L. Breaking the Base Barrier: An Electron-Deficient Palladium Catalyst Enables the Use of a Common Soluble Base in C-N Coupling. *J. Am. Chem. Soc.* **2018**, *140*, 4721-4725. (b) Ayothiraman, R.; Rangaswamy, S.; Maity, P.; Simmons, E. M.; Beutner, G. L.; Janey, J.; Eastgate, M.D.; Vaidyanathan, R. Zinc Acetate-Promoted Buchwald-Hartwig Couplings of Heteroaromatic Amines. *J. Org. Chem.* **2017**, *82*, 7420-7427. (c) Zhang, B.; Zhang, X.; Hao, J.; Yang, C. Direct Approach to N-Substituted-2-Fluoroindoles by Sequential Construction of C–N Bonds from gem-Difluorostyrenes. *Org. Lett.* **2017**, *19*, 1780-1783. (d) Ramirez-Lopez, P.; Ros, A.; Romero-Arenas, A.; Iglesias-Siguenza, J.;

1

3

4

5

7

Fernandez, R.; Lassaletta, J. M. Synthesis of IAN-type N,N-Ligands via Dynamic Kinetic Asymmetric Buchwald-Hartwig Amination. J. Am. Chem. Soc. 2016, 138, 12053-12056. (e) Valente, 2 C.; Calimsiz, S.; Hoi, K. H.; Mallik, D.; Sayah, M.; Organ, M. G. The Development of Bulky Palladium NHC Complexes for the Most-Challenging Cross-Coupling Reactions. Angew. Chem., Int. Ed. 2012, 51, 3314-3334. (f) Hartwig, J. F. Evolution of a Fourth 6 Generation Catalyst for the Amination and Thioetherification of Aryl Halides. Acc. Chem. Res. 2008, 41, 1534-1544. (g) Hartwig, J. F. Carbon-heteroatom bond formation catalysed by organome-8 tallic complexes. Nature 2008, 455, 314-322. (h) Surry, D. S.; 9 Buchwald, S. L. BiarylPhosphane Ligands in Palladium-Catalyzed 10 Amination. Angew. Chem., Int. Ed. 2008, 47, 6338-6361. (i) Shek-11 har, S.; Ryberg, P.; Hartwig, J. F.; Mathew, J. S.; Blackmond, D. 12 G.; Strieter, E. R.; Buchwald, S. L. Reevaluation of the Mecha-13 nism of the Amination of Aryl Halides Catalyzed by BINAP-14 Ligated Palladium Complexes. J. Am. Chem. Soc. 2006, 128, 3584-3591. 15

(5) (a) Derosa, J.; O'Duill, M. L.; Holcomb, M.; Boulous, M. N.; 16 Patman, R. L.; Wang, F.; Tran-Dube, M.; McAlpine, I.; Engle, K. 17 M. J. Copper-Catalyzed Chan-Lam Cyclopropylation of Phenols 18 and Azaheterocycles. Org. Chem. 2018, 83, 3417-3425. (b) Miras, 19 H. N.; Isidro-Llobet, A.; Sproules, S.; Watson, A. J. B. Spectro-20 scopic Studies of the Chan-Lam Amination: A Mechanism-21 Inspired Solution to Boronic Ester Reactivity. J. Am. Chem. Soc.2017, 139, 4769-4779. (c) Liu, S.; Zu, W.; Zhang, J.; Xu, L. 22 Chemoselective N-Arylation of Aminobenzamides via Copper 23 Catalysed Chan-Evans-Lam Reactions. Org. Biomol. Chem. 2017, 24 15, 9288-9299. (d) Levitskiv, O. A.; Grishin, Y.K.; Sentyurin, V.v. 25 V.; Magdesieva, T. V. Copper-Assisted Amination of Boronic 26 Acids for Synthesis of Bulky Diarylamines: Experimental and 27 DFT Study. Chem. Eur. J. 2017, 23, 12575-12584. (e) Dar'in, D.; Krasavin, M. The Chan-Evans-Lam N-Arylation of 2-28 Imidazolines. J. Org. Chem. 2016, 81, 12514-12519. (f) Moon, P. J.; 29 Halperin, H. M.; Lundgren, R. J. Oxidative Coupling of Aryl 30 Boron Reagents with sp(3)-Carbon Nucleophiles: The Enolate 31 Chan-Evans-Lam Reaction. Angew. Chem., Int. Ed. 2016, 55, 1894-32 1898. (g) Yoo, W.-J.; Tsukamoto, T.; Kobayashi, S. Visible-Light-33 Mediated Chan-Lam Coupling Reactions of Aryl Boronic Acids and Aniline Derivatives. Angew. Chem., Int. Ed. 2015, 54, 6587-34 6587. (h) Moon, S.-Y.; Nam, J.; Rathwell, K.; Kim, W.-S. Copper-35 Catalyzed Chan-Lam Coupling between Sulfonyl Azides and 36 Boronic Acids at Room Temperature. Org. Lett. 2014, 16, 338-341. 37 (i) Bruneau, A.; Brion, J.-D.l; Alami, M.; Messaoudi, S. Stereose-38 lective Copper-Catalyzed Chan-Lam-Evans N-Arylation of Glu-39 cosamines with Arylboronic acids at Room Temperature. Chem. 40 Commun. 2013, 49, 8359-8361.

(6) (a) Margrey, K. A.; Levens, A.; Nicewicz, D. A. Direct Aryl 41 C-H Amination with Primary Amines Using Organic Photoredox 42 Catalysis. Angew. Chem., Int. Ed. 2017, 56, 15644-15648. (b) Wild-43 ing, M. J. T.; Iovan, D. A.; Wrobel, A.T.; Lukens, J. T.; MacMillan, 44 S. N.; Lancaster, K. M.; Betley, T. A. Direct Comparison of C-H 45 Bond Amination Efficacy through Manipulation of Nitrogen-46 Valence Centered Redox: Imido versus Iminyl. J. Am. Chem. Soc. 47 2017, 139, 14757-14766. (c) Alt, I. T.; Guttroff, C.; Plietker, B. Iron-Catalyzed Intramolecular Aminations of C(sp³)-H Bonds 48 in AlkylarylAzides. Angew. Chem., Int. Ed. 2017, 56, 10582-10586. 49 (d) Bagh, B.; Broere, D. L. J.; Sinha, V.; Kuijpers, P. F.; van Leest, 50 N. P.; de Bruin, B.; Demeshko, S.; Siegler, M. A.; van der Vlugt, J. 51 I. Catalytic Synthesis of N-Heterocycles via Direct C(sp3)-H 52 Amination Using an Air-Stable Iron(III) Species with a Redox-53 Active Ligand. J. Am. Chem. Soc. 2017, 139, 5117-5124. (e) Bai, H.-54 Y.; Ma, Z.-G.; Yi, M.; Lin, J.-B.; Zhang, S.-Y. Palladium-Catalyzed Direct Intermolecular Amination of Unactivated Methylene 55

C(sp³)-H Bonds with Azodiformates via Bidentate-Chelation Assistance. ACS Catal. 2017, 7, 2042-2046. (f) Tang, C.; Zou, M.; Liu, J.; Wen, X.; Sun, X.; Zhang, Y.; Jiao, N. Rh-Catalyzed Direct Amination of Unactivated C(sp3)-H bond with Anthranils Under Mild Conditions. Chem. Eur. J. 2016, 22, 11165-11169. (g) Alt, I. T.; Plietker, B. Iron-Catalyzed Intramolecular C(sp2)-H Amination. Angew. Chem. Int. Ed. 2016, 55, 1519-1522. (h) Tran, L.D.; Roane, J.; Daugulis, O. Directed amination of non-acidic arene C-H bonds by a copper-silver catalytic system. Angew. Chem., Int. Ed. 2013, 52, 6043-6046. (i) Hennessy, E. T.; Betley, T. A. Complex N-Heterocycle Synthesis via Iron-Catalyzed, Direct C-H Bond Amination. Science 2013, 340, 591-594.

(7) (a) An, Z.; Guo, Y.; Zhao, L.; Li, Z.; He, J. L-Proline-Grafted Mesoporous Silica with Alternating Hydrophobic and Hydrophilic Blocks to Promote Direct Asymmetric Aldol and Knoevenagel-Michael Cascade Reactions. ACS Catal. 2014, 4, 2566-2576. (b) Hayashi, Y.; Yamaguchi, J.; Sumiya, T.; Hibino, K.; Shoji, M. Direct Proline-Catalyzed Asymmetric α-Aminoxylation of Aldehydes and Ketones. J. Org. Chem. 2004, 69, 5966-5973.

(8) Shi, B.-F.; Maugel, N.; Zhang, Y.-H. Yu, J.-Q. Pd^{II}-Catalyzed Enantioselective Activation of C(sp²)-H and C(sp³)-H Bonds Using Monoprotected Amino Acids as Chiral Ligands. Angew. Chem., Int. Ed. 2008, 16, 4882-4886.

(o) (a) Veisi, H.; Biabri, P. M.; Falahi, H. L- Arginine as a Base and Ligand for the Palladium-Catalyzed C-C and C-N Cross-Coupling Reactions in Aqueous Media. Tetrahedron Lett. 2017, 58, 3482-3486. (b) Ma, D.; Cai, Q. Copper/Amino Acid Catalyzed Cross-Couplings of Aryl and Vinyl Halides with Nucleophiles. Acc. Chem. Res. 2008, 41, 1450-1460. (c) Deng, W.; Liu, L.; Zhang, C.; Liu, M.; Guo, Q.-X., Copper-Catalyzed Cross-Coupling of Sulfonamides with Aryl Iodides and Bromides Facilitated by Amino Acid Ligands. Tetrahedron Lett. 2005, 46, 7295-7298. (d) Ma, D.; Cai, Q.; Zhang, H. Mild Method for Ullmann Coupling Reaction of Amines and Aryl Halides. Org. Lett. 2003, 5, 2453-2455. (e) Elliott, E.-C.; Maggs, J. L.; Park, B. K.; O'Neilla, P. M.; Stachulski, A. V. Convenient Syntheses of Halodibenz[b,f]azepines and Carbamazepine Analogues via N-Arylindoles. Org. Biomol. Chem. 2013, 11, 8426-8434. (f) Zhang, H.; Cai, Q.; Ma, D. Amino Acid Promoted CuI-Catalyzed C-N Bond Formation between Aryl Halides and Amines or N-Containing Heterocycles. J. Org. Chem. 2005, 70, 5164-5173. (g) Griffin, J. H.; Kellogg, R. M. L-Cysteine-, L-methionine- and D-Penicillamine-Derived Ligands for Transition Metal Catalyzed Carbon-Carbon Bond-Forming Reactions. J. Org. Chem. 1985, 50, 3261-3266.

(10) (a) Ma, H.-C.; Jiang, X.-Z.N-Hydroxyimides as Efficient Ligands for the Copper-Catalyzed N-Arylation of Pyrrole, Imidazole, and Indole.J.Org. Chem. 2007, 72, 8943-8946. (b) Monguchi, Y.; Marumoto, T.; Takamatsu, H.; Sawama, Y.; Sajiki, H. Palladium on Carbon-Catalyzed One-Pot N-Arylindole Synthesis: Intramolecular Aromatic Amination, Aromatization, and Intermolecular Aromatic Amination. Adv Synth. Catal. 2014, 356, 1866-1872. (c) Antilla, J. C.; Klapars, A.; Buchwald, S. L. The Copper-Catalyzed N-Arylation of Indoles. J. Am. Chem. Soc. 2002, 124, 11684-11688. (d) Sharma, R. K.; Gaur, R.; Yadav, M.; Rathi, A. K.; Pechousek, J.; Petr, M.; Zboril, R.; Gawande, M. B. Maghemite-Copper Nanocomposites: Applications for Ligand-Free Cross-Coupling (C-O, C-S, and C-N) Reactions. ChemCatChem. 2015, 7, 3495-3502. (e) Hartwig, J. F.; Kawatsura, M.; Hauck, S. I.; Shaughnessy, K. H.; Alcazar-Roman, L. M. Room-Temperature Palladium-Catalyzed Amination of Aryl Bromides and Chlorides and Extended Scope of Aromatic C-N Bond Formation with a Commercial Ligand. J. Org. Chem. 1999, 64, 5575-5580. (f) Perregaard, J.; Arnt, J.; Boegesoe,

56

57 58

K. P.; Hyttel, J.; Sanchez, C.Noncataleptogenic, centrally acting dopamine D-2 and serotonin 5-HT2 antagonists within a series of 3-substituted 1-(4-fluorophenyl)-1H-indoles. J. Med. Chem. 1992, 35, 1092-1101. (g) Old, D. W.; Harris, M. C.; Buchwald, S. L.Efficient Palladium-Catalyzed N-Arylation of Indoles. Org. Lett. 2000, 2, 1403-1406. (h) Balle, T.; Perregaard, J.; Ramirez, M. T.; Larsen, A. K.; Søby, K. K.; Liljefors, T.; Andersen, K. Synthesis and Structure-Affinity Relationship Investigations of 5-Heteroaryl-Substituted Analogues of the Antipsychotic Sertindole. A New Class of Highly Selective & Adrenoceptor Antagonists. J. Med. Chem. 2003, 46, 265-283. (i) Kumar, S.; Rathore, V.; Verma, A.; Prasad, Ch. D.; Kumar, A.; Yadav, A.; Jana, S.; Sattar, M.; Meenakshi, Kumar, S.KOtBu-Mediated Aerobic Transition-Metal-Free Regioselective β-Arylation of Indoles: Synthesis of β-(2-/4-Nitroaryl)-indoles. *Org. Lett.* **2015**, *17*, 82-85. (11) (a) Wang, H.; Cai, J.; Huang, H.; Deng, G.-J.Palladium-CatalyzedPhthalazinone Synthesis Using Paraformaldehyde as Carbon Source. Org. Lett. 2014, 16, 5324-5327. (b) Natte, K.; Neumanna, H.; Wu, X.-F. Pd/C as an efficient heterogeneous catalyst for carbonylative four-component synthesis of 4(3H)quinazolinones. Catal. Sci. Technol. 2015, 5, 4474-4480. (c) Lv, X.; Bao, W. A β-Keto Ester as a Novel, Efficient, and Versatile Ligand for Copper(I)-Catalyzed C-N, C-O, and C-S Coupling Reactions. J. Org. Chem. 2007, 72, 3863-3867. (d) wasaki, T.; Yamashita, K.; Kuniyasu, H.; Kambe, N. Co-Catalyzed Cross-Coupling Reaction of Alkyl Fluorides with Alkyl Grignard Reagents. Org. Lett.2017, 19, 3691-3694. (e) Chen, Z.; Zeng, H.; Girard, S. A.; Wang, F.; Chen, N.; Li, C.-J. Formal Direct Cross-Coupling of Phenols with Amine. Angew. Chem. Int. Ed. 2015, 54, 14487-14491. (f) Singsardar, M.; Chakraborty, A.; Jana, S.; Hajra, A. Metal-Free Synthesis of Indoles from Arylhydrazines and Nitroalkenes at Room Temperature. Chemistry Select, 2017, 2, 8893-8897.