

Palladium-Catalyzed C–S Bond Cleavage with Allenoates: Synthesis of Tetrasubstituted 2-Alkenylfuran Derivatives

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S Supporting Information

ABSTRACT: Palladium-catalyzed C-S cleavage of tetrasubstituted internal alkene α -oxo ketene dithioacetals was realized with allenoates as the coupling partners, efficiently affording tetrasubstituted 2-alkenylfuran derivatives with excellent regioselectivity under mild conditions. Allenoates acted as C1 synthons in the desulfurative [4 + 1] annulation.

he furan motif is a key structural unit in many bioactive \bot molecules, natural products, and pharmaceuticals,¹ and furan compounds can be used as the important building blocks in organic synthesis.² To access multisubstituted furan derivatives, two strategies can be applied: (a) functionalization of an existing furan ring and (b) direct construction of a substituted furan ring by ring closure of suitable precursor compounds. Although direct C-H functionalization of furans seems to be an attractive method to prepare multisubstituted furans,³ this method remains a great challenge in the reaction scope, efficiency, and chemoselectivity. As for the ring closure method, much progress has recently been made by means of the cyclization reactions of allenes. In this regard, cycloisomerization of allenyl ketones has been applied for the construction of multisubstituted furans by Marshall,^{4a} Hashmi,^{4b} Ma,^{4c-f} Gevorgyan,^{4g,h} and others.^{4i-m} Palladiumcatalyzed cycloisomerization of homoallenyl amides was developed to furnish 2-aminofurans.⁵ Intramolecular cyclization of the allene precursors, that is, propargylic alcohols⁶ and γ -acyloxybutynoates,⁷ was also used to synthesize multisubstituted furans. Owing to their intrinsic reactivity,⁸ allenes usually act as C2 and C3 synthons in the synthesis of heterocyclic compounds through cycloaddition reactions. However, such cycloaddition reactions have rarely been applied for furan synthesis from allenes.¹⁰ So far, only a few examples have been documented for allenes serving as the C1 synthons in the synthesis of vinyl-substituted heterocycles,¹ and transition-metal-catalyzed cycloaddition by means of allenes as the C1 synthons has not yet been achieved for the construction of furan derivatives.

 α -Oxo ketene dithioacetals have recently demonstrated their diversity in organic synthesis.¹² However, they have usually been used as substrates in the trisubstituted form which is obtained by hydrolysis of the corresponding tetrasubstituted diketone ketene dithioacetals under strong acidic conditions (Scheme 1a). Direct C-H alkylation,¹³ alkenylation,¹





arylation,¹⁵ and other C–H functionalization reactions¹⁶ of trisubstituted α -oxo ketene dithioacetals have been achieved, and transition-metal-catalyzed construction of heterocyclic compounds was also reported.¹⁷ In comparison to trisubstituted α -oxo ketene dithioacetals, their tetrasubstituted analogues have been paid much less attention, and only a few reports have been documented for the desulfurative synthesis of tetrasubstituted alkenes,¹⁸ carbocycles,¹⁹ and N-²⁰ and O-containing heterocycles.²¹ We recently reported ironcatalyzed oxidative C-H/C-H cross-coupling of trisubstituted α -oxo ketene dithioacetals with carbonyl methylenes, affording tetrasubstituted furans (Scheme 1b).^{17a} Thus, we reasonably envisioned that tetrasubstituted 2-alkenylfurans might be accessed through the cyclization of tetrasubstituted α -oxo ketene dithioacetals with allenes as the C1 synthons. Herein, we disclose palladium-catalyzed $\begin{bmatrix} 4 + 1 \end{bmatrix}$ annulation of

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tetrasubstituted α -oxo ketene dithioacetals with allenoates for the synthesis of tetrasubstituted 2-alkenylfurans (Scheme 1c).

Initially, the reaction of α, α -diacetyl ketene di(methylthio)acetal (1a) with benzyl buta-2,3-dienoate (2a) was conducted to screen the reaction conditions (eq 1) (see the Supporting



Information for details). Under a nitrogen atmosphere, in the presence of 10 mol % of $Pd(PPh_3)_4$ catalyst, 2 equiv of copper(I) thiophene-2-carboxylate (CuTC) as the sacrifacial additive, and 2 equiv of Cs_2CO_3 base, the reaction of 1a with 2a in a 1:3 molar ratio was undergone in diethyl ether at 25 °C for 12 h, affording tetrasubstituted 2-alkenylfuran 3a in 71% yield. Elevating the temperature to 30 °C led to more efficient formation of 3a. It is noteworthy that a base is not necessary for the reaction. Lowering the amount of CuTC diminished the reaction efficiency. The control experiments revealed that the reaction could not efficiently proceed in the absence of $Pd(PPh_3)_4$ or CuTC. On a 0.3 mmol scale the target product 3a was isolated in 74% yield. The ¹H NMR analysis of the reaction mixture of 1a and 2a revealed exclusive formation of the stereospecific product 3a.

Next, the scope of tetrasubstituted α -oxo ketene dithioacetals 1 was investigated under the optimized conditions (Scheme 2). The ethylthio-featuring α, α -diacetyl ketene di(ethylthio)acetal (1b) reacted less efficiently than 1a to form the target product 3b (59%). Although two different carbonyl groups, i.e., acetyl and benzoyl, were present in ketene dithioacetals 1c, only the acetyl group was involved in the reaction with 2a, and the reaction exclusively gave product 3c (67%). 2-(Bis(methylthio)methylene)cyclohexane-1,3dione (1d) exhibited no reactivity to 2a. The α -acetyl- α ester ketene dithioacetals 1e-i efficiently underwent the reactions with 2a, affording products 3e-i in 71-81% yields, while α -amide and α -cyano-functionalized α -acetyl ketene dithioacetals 1j and 1k could not efficiently react to yield 3j (25%) and 3k (34%), respectively. α -Phenyl- α -acetyl ketene dithioacetals 11 also reacted efficiently with 2a to form 31 (77%). It is noteworthy that α -ester-functionalized α -alkanoyl ketene di(methylthio)acetals 1m-q reacted well with 2a to produce products 3m-q in 73-83% yields. Cyclopropyl and cyclobutyl moieties facilitated the formation of 3p (83%) and **3q** (80%), respectively, whereas the *tert*-butyl group in α -ester- α -pivaloyl ketene dithioacetal (1r) completely inhibited the reaction with 2a, exhibiting a remarkable steric effect.

Then, the protocol generality was explored by extending the substrate scope to α -aroyl ketene dithioacetals of type 1. Notably, α -ester-functionalized α -benzoyl ketene dithioacetal **1s** efficiently underwent the reaction to generate **3s** (84%). The analogues of **1s**, that is, substituted α -benzoyl ketene dithioacetals **1t**-**z2**, exhibited various reactivities to form the target furan products **3t**-**z2** in 69-81% yields. Bulky α -(2-naphthoyl) ketene dithioacetal **1z3** reacted well with **2a** to give **3z3** (74%), showing no obvious steric effect from the 2-naphthyl group. α -Heteroaroyl (2-furoyl and 2-thienoyl) ketene dithioacetals **1z4** and **1z5** also reacted efficiently with **2a** to form **3z4** (76%) and **3z5** (74%), respectively. It should be noted that reacting α, α -dibenzoyl ketene dithioacetal **1z6** with allenoate **2a** led to the target product **3z6** in 77% yield.





^{*a*}Conditions: **1** (0.3 mmol), **2a** (0.9 mmol), Pd(PPh₃)₄ (0.03 mmol), CuTC (0.6 mmol), ether (3 mL), 0.1 MPa N₂, 30 °C, 12 h. Yields refer to the isolated products. ^{*b*}Pd(PPh₃)₄ (0.06 mmol).

The protocol generality was further explored by using various allenoates 2 as the coupling partners (Scheme 3). α -Benzoyl- α -ester ketene dithioacetal 1s reacted with ethyl buta-2,3-dienoate (2b) to afford the target product 4a in 73% yield. Compound 1s also efficiently reacted with allenoates 2b-f to give tetrasubstituted 2-alkenylfurans 4b-e (70-76%), and variation of the alkyl ester groups in the allenoates had no obvious impact on the reaction efficiency. Phenyl buta-2,3dienoate (2g) exhibited a moderate reactivity, and its reaction with 1s gave product 4f in 50% yield. Both ethyl 2-methylbuta-2,3-dienoate (2h) and 2-benzylbuta-2,3-dienoate (2i) showed an obvious steric effect on the formation of the target products 4g (50%) and 4h (44%). The ¹H NMR analysis of the crude product 4g or 4h before separation was made, but the proton NMR signals were too complicated to distinguish the product stereoselectivity. We only obtained the stereospecific products 4g and 4h by silica gel column chromatography. Both the

Scheme 3. Scope of Allenoates $(2)^{a}$



^{*a*}Conditions: 1 (0.3 mmol), 2 (0.9 mmol), Pd(PPh₃)₄ (0.03 mmol), CuTC (0.6 mmol), ether (3 mL), 0.1 MPa N₂, 30 °C, 12 h. Yields refer to the isolated products. ^{*b*}60 °C, in a 25 mL sealed tube.

reactions of α -acetyl ketene dithioacetals 1a with 2e and 1f with 2b proceeded smoothly to form products 4i (70%) and 4j (67%), respectively. The molecular structures of compounds 3e and 4g were further confirmed by the X-ray single-crystal crystallographic determinations (see the SI for details).

A comparative evaluation of the reactivities was made between the tetra- and trisubstituted α -oxo ketene dithioacetals 1 and 5. It was found that trisubstituted α -benzoyl ketene di(methylthio)acetals **5a**-**c** exhibited a reactivity lower than their tetrasubstituted analogues 1s, 1w, and 1z, and their reactions with allenoate 2a formed the corresponding trisubstituted 2-alkenylfurans **6a**-**c** in 50–58% yields (eq 2).



In order to verify the role of the dialkylthio functionality in 1, 3-((methylthio)methylene)pentane-2,4-dione (7a) and 3-benzylidenepentane-2,4-dione (7b) were reacted with 2a under the standard conditions. No reaction was observed to form the desired products 8a and 8b (eq 3), while the



corresponding α, α -diacetyl ketene di(methylthio)acetal (1a) reacted with 2a to give the target product 3a in 74% yield (Scheme 2). These results have unambiguously revealed that such a di(alkylthio) functionality is crucial for α -oxo ketene dithioacetals of type 1 to undergo the palladium-catalyzed [4 + 1] annulation with allenoates. Allene derivatives 9a and 9b were used to replace 2a in the annulation reaction, but they could not undergo the same annulation reactions, implicating the crucial role of an ester group at the terminus of the allene backbone (eq 3).

To demonstrate the utility of the synthetic protocol, a gramscale reaction of 1s with 2a was performed, and the target product 3s was obtained in 72% yield (eq 5). A two-step



procedure was developed to transform 2-alkenylfuran 3v to lactone 11^{22} (59%), which has been proved to be a key structural motif in some bioactive molecules, natural products, and pharmaceuticals (eq 6).²³ With *meta*-chloroperoxybenzoic acid (*m*-CPBA) as the oxidant, 2-alkenylfuran 3s was readily oxidized to the corresponding sulfone 12 in 89% yield (eq 7).

A plausible mechanism is proposed in Scheme 4. The copper(I) additive initially activates one of the C-S bonds in

Scheme 4. Proposed Reaction Mechanism



1a through coordination with the sulfur atom by assistance of the directing α -acetyl group, forming species **A**. Pd(0) species is then inserted into the activated C–S bond to yield Pd(II) species **B** in which both Pd–C and Pd–S bonds are formed. Interaction of species **B** with allenoate **2a** generates Pd(II) species **C** through insertion of the terminal alkenyl of the

allene substrate into the Pd–C bond. β -H elimination occurs to produce intermediate D^{24} and is followed by C=C insertion into the Pd–H bond, resulting in species E.^{24b} Loss of the CuTC methylthiol adduct generates cyclopalladate species F which then undergoes reductive elimination to afford the target product 3a and regenerate the catalytically active Pd(0) species, accomplishing a catalytic cycle.

In summary, efficient palladium-catalyzed, copper-mediated [4 + 1] annulation of tetrasubstituted α -oxo ketene dithioacetals and allenoates has been realized to synthesize tetrasubstituted 2-alkenylfurans under mild conditions. In the process, allenoates acted as effective C1 synthons. The present protocol provides a convenient route to tetrasubstituted 2-alkenylfurans.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.or-glett.8b02253.

Experimental materials and procedures, NMR of compounds, and X-ray crystallographic analysis for compounds **3e** and **4g** (PDF)

Accession Codes

CCDC 1813518 and 1842016 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

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Notes

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