Accepted Manuscript

Preparation of 4(3H)-quinazolinones from aryldiazonium salt, nitriles and 2-aminobenzoate *via* a cascade annulation

Mani Ramanathan, Ming-Tsung Hsu, Shiuh-Tzung Liu

PII: S0040-4020(18)31568-0

DOI: https://doi.org/10.1016/j.tet.2018.12.065

Reference: TET 30051

To appear in: Tetrahedron

Received Date: 1 October 2018

Revised Date: 28 December 2018

Accepted Date: 30 December 2018

Please cite this article as: Ramanathan M, Hsu M-T, Liu S-T, Preparation of 4(*3H*)-quinazolinones from aryldiazonium salt, nitriles and 2-aminobenzoate *via* a cascade annulation, *Tetrahedron* (2019), doi: https://doi.org/10.1016/j.tet.2018.12.065.

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.



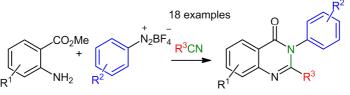
Graphical Abstract

To create your abstract, type over the instructions in the template box below. Fonts or abstract dimensions should not be changed or altered.

Preparation of 4(*3H*)-quinazolinones from aryldiazonium salt, nitriles and 2aminobenzoate *via* a cascade annulation

Leave this area blank for abstract info.

Mani Ramanathan, Ming-Tsung Hsu, Shiuh-Tzung Liu* Department of Chemistry, National Taiwan University, Taipei, Taiwan 106





Tetrahedron journal homepage: www.elsevier.com

Preparation of 4(*3H*)-quinazolinones from aryldiazonium salt, nitriles and 2aminobenzoate *via* a cascade annulation

Mani Ramanathan, Ming-Tsung Hsu and Shiuh-Tzung Liu*

Department of Chemistry, National Taiwan University, Taipei, Taiwan 106

ARTICLE INFO

ABSTRACT

Article history: Received Received in revised form Accepted Available online

Keywords: Quinazolinone Diazonium Nitrilium Nitrile Annulation One-pot synthesis of 3-aryl-4(*3H*)-quinazolinones has been realized through a cascade annulation. Reaction of aryldiazonium salt with a nitrile provides *in situ* generation of a reactive nitrilium ion, which is attacked by the amino group of 2-aminobenzoate followed by cyclization to deliver the desired product. This strategy offers a convenient and easy access to a wide range of functionalized quinazolinone.

2009 Elsevier Ltd. All rights reserved.

* Corresponding author. Tel.: +8862-3366-1661; fax: +8862-3366-8671; e-mail: stliu@ntu.edu.tw

1

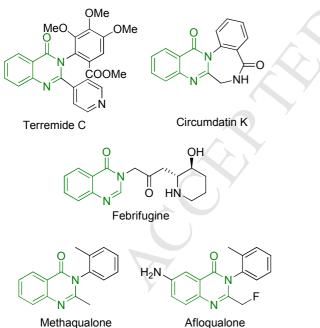
Tetrahedron

Tetrahedron ACCEPTED MA(a)/USCRIPT

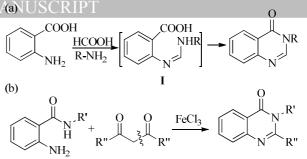
1. Introduction

Quinazolinones are prevalent structural units in a wide range of both natural products and pharmaceuticals.¹⁻² Terremide C,^{1a} Febrifugine,^{1b-c} and Circumdatin K^{1d} are naturally occurring alkaloids, whereas methaqualone³ and afloqualone⁴ are sedative and hypnotic medications (Scheme 1). In addition, quinazolinone derivatives exhibit wide spectrum of therapeutic activities such as anticancer,⁵ hypotensive,⁶ antiallergy⁷ and protein tyrosine kinase inhibitory properties.⁸ With the importance of quinazolinone derivatives in biological use, numerous synthetic approaches including methodologies with or without transition-metal catalysis leading to these heterocyclic structures have been disclosed and received significant attention. Quite a few review articles summarizing these methodologies have appeared.⁹

Methodologies for synthesis of quinazolinones are primarily based on condensation procedures. For example, microwave irradiation of a mixture of anthranilic acid, formic acid and primary amine provided 3-substituted quinazolin-4(3H)-ones (Scheme 2a). This reaction is believed to proceed through the amidine intermediate I, which subsequently underwent intramolecular amidation to yield the final product.^{10a} It should be mentioned that most of these methods often require harsh reaction conditions and use of toxic and/or corrosive reagents such as PCl₃, O=PCl₃ or related reagents, which greatly affects the sensitive functional groups. On the other hands, transition metal catalysed synthesis offers various approaches to the desired product by using easily accessed starting materials due to the versatile activity and reaction types of metal complexes.⁹ An interesting approach by Zhou and co-workers demonstrates that FeCl₃-catalyzed tandem reaction of 2-aminobenzamides with 1,3diketones offers the quinazolinone products.^{10b} This method involves a C-C bond cleavage pathway during the aromatization (Scheme 2b).

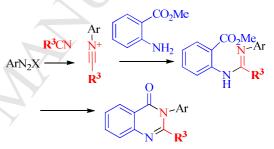


Scheme 1. Some derivatives of quinazolin-4-(3*H*)-ones



Scheme 2. Preparation of quinazolin-4(3*H*)-ones by (a) a condensation pathway and (b) a transition-metal catalysis

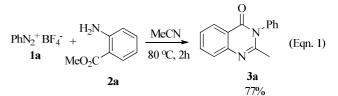
In recent studies, we have found that aryldiazonium salts can be used as the precursor for the preparation of various heterocycles.¹¹ In these syntheses, the key intermediate is *N*arylnitrilium ion generated *in situ* from the reaction of the arenediazonium salt with a nitrile molecule.¹² Such an intermediate could further react with nucleophiles to construct the heterocycles in a cascade reaction fashion. Herein, we report a mild and transition metal-free synthesis of quinazolin-4(*3H*)ones (Scheme 3). We envision that the reactive *N*-arylnitrilium intermediate could react with methyl anthranilate followed by cyclization to afford the desired product directly without the use of phosphorus trichloride or related reagents.



Scheme 3. Our approach in preparation of quinazolin-4(*3H*)-ones.

2. Results and discussion

To validate the hypothesis, we chose the reaction of phenyldiazonium tetrafluoroborate (**1a**) with methyl anthranilate (**2a**) in acetonitrile as the model example (Eqn 1). After several trials by varying the temperature, the desired product **3a** was obtained in 77% yield by carrying out the reaction at 80 °C. Running the reactions at either higher or lower temperature resulted in a lower yield of the product. Efforts to reduce the quantity of nitrile in the presence of co-solvents such as dichloroethane, toluene and methanol led to poor yields. For example, carrying out this reaction in toluene with 10 equivalent of nitrile provided the target molecule in 52% yield. In these studies, we learned that the reaction requires to proceed under anhydrous conditions. Otherwise, *N*-phenyl acetamide was obtained as the major product, presumably due to the hydrolysis of nitrilium ion.



With this optimal condition in hand, we systematically examined this annulation toward functionalized 3-aryl-4(3*H*)-quinazolinones by reacting various aryldiazonium salts, nitriles and substituted anthranilates. Gratifyingly, aryldiazonium salts

with common substitutions such as, 4-Me, 3-Me, 4-OMe, 4-tBu readily reacted with 2a to give the corresponding quinazolinones in moderate to good yields (Table 1, entries 1-5). It is worth noting that this reaction could tolerate the presence of functional groups such as halogens and esters (3f-3g), providing ample potential for further synthetic elaborations. Notably, aryldiazonium salts with sterically demanding 2,4,6-trimethyl and 3,4,5-trimethoxy substituents were also found to be amenable substrates for this reaction, furnishing the multi-substituted quinazolinones in moderate yields (Table 1, entries 9-10). In addition, this annulation protocol was found to proceed well with various readily available aliphatic, aromatic and benzylic nitriles and delivered the corresponding 2-substituted quinazolin-4(3H)ones in moderate yields (Table 1, entries 11-14). Unfortunately, malononitrile and 3-bromopropionitrile were failed to give the desired products. On the other hand, scope of this reaction was contemplated with various anthranilates. Utilization of anthranilates with diverse substitutions such as, 5-Me, 4-Br and groups participated 4,5-dimethoxy smoothly in the transformation and provided the corresponding quinazolinones in good yields (Table 1, entries 15-17).

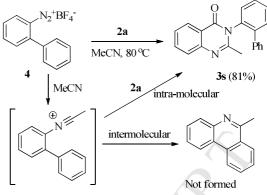
Table I. Pre	eparation of substitute	ed quinazolin-4(3H)-ones	
		D ²	

N	$\mathbb{R}_{2}^{+}\mathbb{B}_{4}^{-} \xrightarrow{R_{2}^{+}} \mathbb{N}_{2}^{+} \xrightarrow{R_{2}^{+}} \xrightarrow{R_{2}^{+}} \mathbb{N}_{2}^{+} \xrightarrow{R_{2}^{+}} \xrightarrow{R_{2}^{$	R ²
	2 h 1 1 2 3	L
Entry	Substituents in 3	Yields ^b
1	3a , $R^1 = H$, $R^2 = H$, $R^3 = CH_3$	77%
2	3b , $R^1 = H$, $R^2 = 4$ -Me, $R^3 = CH_3$	74%
3	3c , $R^1 = H$, $R^2 = 3$ -Me, $R^3 = CH_3$	55%
4	3d , $R^1 = H$, $R^2 = 4$ -OMe, $R^3 = CH_3$	53%
5	3e , $R^1 = H$, $R^2 = 4$ - <i>t</i> -Bu, $R^3 = CH_3$	81%
6	3f , $R^1 = H$, $R^2 = 4$ -Cl, $R^3 = CH_3$	33%
7	3g , $R^1 = H$, $R^2 = 4$ -CO ₂ Et, $R^3 = CH_3$	44%
8	3h , $R^1 = H$, $R^2 = 1$ -naphthyl, $R^3 = CH_3$	67%
9	3i , $R^1 = H$, $R^2 = 2,4,6$ -triMe, $R^3 = CH_3$	68%
10	3j , $R^1 = H$, $R^2 = 3,4,5$ -(MeO) ₃ , $R^3 = CH_3$	52%
11	3k , $R^1 = H$, $R^2 = 4$ -Me, $R^3 = CH_3CH_2CH_2$	56%
12	31 , $R^1 = H$, $R^2 = 4$ -Me, $R^3 = (CH_3)_2CH$	41%
13	3m , $R^1 = H$, $R^2 = 4$ -Me, $R^3 = C_6 H_5$	48%
14	3n , $R^1 = H$, $R^2 = 4$ -Me, $R^3 = CH_2C_6H_5$	53%
15	30, $R^1 = 6$ -Me, $R^2 = 4$ -Me, $R^3 = CH_3$	92%
16	3p , $R^1 = 7$ -Br, $R^2 = H$, $R^3 = CH_3$	63%
17	3q , $R^1 = 6,7$ -(OMe) ₂ , $R^2 = H, R^3 = CH_3$	68%

^a Reaction conditions: aryldiazonium salt (100 mg) and 2 (1.1 eq) in nitrile (2 mL) was heated at 80 °C for 2 h. ^b isolated yields

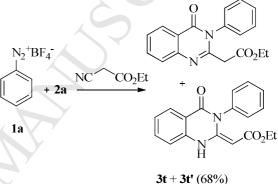
Reaction of 2-phenyl substituted aryldiazonium salt (4) with methyl anthranilate (2a) under the standard conditions was investigated to study the competition of intra- versus intermolecular pathway (Scheme 4). In contrast to our previous observations,^{11c} aryldiazonium salt 4 was reacted with 2a to furnish 3s in excellent yields instead of the formation of 6-methylphenanthridine. Apparently, the intermolecular

amination/cascade annulation of *N*-arylnitrilium ion is superior to the intramolecular arylation.



Scheme 4. Inter- versus intra-molecular pathway

On the other hand, reaction of ethyl cyanoacetate with 1a and 2a led to mixture of tautomers (3t and 3t') as an inseparable mixture (Scheme 5). This isomerization is due to the acidic hydrogen of cyanoacetate.



(inseparable 1:0.33) Scheme 5. Reaction of a nitrile with an active methylene unit

To further demonstrate the synthetic utility, we attempted the synthesis of methaqualone, a well-known sedative and hypnotic drug used for the treatment of insomnia (Eqn. 2).^{13a-b} To our delight, treatment of 0.5 gram of **1k** with 2-methylanthranilate and anhydrous acetonitrile under the optimized conditions afforded the methaqualone in 78% isolated yield. Besides spectroscopic identification, structure of compound **3r** is further

confirmed by X-ray crystallography (Figure 1).

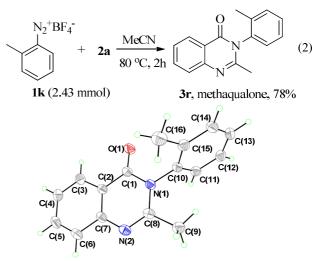


Figure 1. ORTEP plot of 3r (30% probability ellipsoids)

In summary, we have disclosed a convenient and sustainable M approach for the synthesis of quinazolin-4-(3H)-ones via a threecomponent intermolecular amination/tandem cyclization of *in situ* generated *N*-arylnitrilium ion intermediate. This method utilizes readily accessible aryldiazonium salts, nitriles and substituted anthranilates. Excellent functional group compatibility, mild reaction conditions and operational simplicity are the notable features of this report. Furthermore, this method provides an alternative to prepare diversified quinazolin-4-(3H)one scaffolds without the use of toxic reagents such as phosphorus trichloride or related reagents.

3. Experimental section

¹H and ¹³C NMR were recorded in a 400 MHz spectrometer in CDCl₃ and CD₃CN referenced to TMS. All the nitriles were dried over activated 4Å molecular sieves and solid nitriles were purchased and used without any purification. All the anilines were commercially purchased and used for diazotization. Flash chromatography was performed using silica gel 230-400 mesh. Aryldiazonium salts were prepared according to the literature procedure. In cases of known compounds, their ¹H and ¹³C NMR values were compared with the literature values. For NMR data, only major product peaks are listed. Melting points were determined on a Fargo MP-1D instrument. Unless otherwise noted, all the reactions were performed without any special precautions.

3.1. Typical Procedure for Preparation of 3.

A mixture of diazonium salt (100 mg, 1 eq) and methyl anthranilate (1.1 eq) dissolved in dry nitrile (2 mL) was placed in a sealed tube. Reaction mixture was heated to 80 $^{\circ}$ C with stirring for 2 h. Upon cooling, the reaction mixture was diluted with CH₂Cl₂ (20 mL) followed by an aqueous basic workup with aq. NaHCO₃ (5 mL). The crude product was purified by column chromatography with the eluent of ethyl acetate/ hexane (20% to 40% EtOAc:Hexane) to afford the pure product.

3.1.1 2-Methyl-3-phenylquinazolin-4(3H)-one $(3a)^{13b}$

(94 mg, 77%)as a pale yellow solid; mp 143-144 °C; IR (CaF₂): 1742, 1693 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ = 8.22 (dd, *J* = 8.0, 1.6 Hz, 1H), 7.72 (ddd, *J* = 15.4, 7.8, 1.2 Hz, 1H), 7.65 (d, *J* = 8.0 Hz, 1H), 7.54-7.45 (m, 3H), 7.44-7.40 (m, 1H), 2.22 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz): δ = 162.1, 154.1, 147.3, 137.6, 134.4, 129.9, 129.1, 127.9, 126.9, 126.6, 126.5, 120.6, 24.2; HRMS (ESI-TOF) calcd for C₁₅H₁₃N₂O [M+H]⁺ m/z = 237.1028, found 237.1030.

3.1.2 2-Methyl-3-(p-tolyl)quinazolin-4(3H)-one (3b).^{10b}

(89.8 mg, 74%) as a pale yellow solid; mp 155-156 °C; IR (KBr): 1687, 1607 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): $\delta = 8.24$ (d, J = 8.0 Hz, 1H), 7.79-7.74 (m, 2H), 7.49-7.45 (m, 1H), 7.34 (d, J = 8.0 Hz, 2H), 7.12 (d, J = 8.0 Hz, 2H), 2.43 (s, 3H), 2.32 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz): $\delta = 162.0$, 155.0, 146.5, 139.5, 134.7, 130.7, 127.6, 127.1, 126.8, 126.1, 120.6, 24.0, 21.2; HRMS (ESI-TOF) calcd for C₁₆H₁₅N₂O [M+H]⁺ m/z = 251.1184, found: 251.1193.

3.1.3 2-Methyl-3-(m-tolyl)quinazolin-4(3H)-one (3c).

(67.0 mg, 55%) as a pale yellow solid; mp 130-131 °C; IR (CaF₂): 1687, 1602 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): $\delta = 8.25$ (d, J = 7.5 Hz, 1H), 7.76-7.71 (m, 1H), 7.65 (d, J = 8.1 Hz, 1H), 7.45-7.39 (m, 2H), 7.28 (d, J = 7.6 Hz, 1H), 7.04 (d, J = 8.8 Hz, 2H), 2.40 (s, 3H), 2.23 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz): $\delta = 162.3$, 154.3, 147.5, 140.1, 137.6, 134.5, 130.0, 129.7, 128.4, 127.0, 126.7, 126.5, 124.9, 120.8, 24.3, 21.3; HRMS (ESI-TOF) calcd. for C₁₆H₁₅N₂O [M+H]⁺ m/z = 251.1184, found 251.1193.

(63.8 mg, 53%) as a yellow solid; mp 181-182 °C; IR (CaF₂): 1683, 1611 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ = 8.26-8.23 (m, 1H), 7.76-7.72 (m, 1H), 7.67 (d, *J* = 8.0, 1H), 7.46-7.42 (m, 1H), 7.15 (d, *J* = 8.9, 2H), 7.01 (d, *J* = 8.9, 2H), 3.85 (s, 3H), 2.25 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz): δ = 162.4, 160.0, 155.0, 147.0, 134.6, 130.1, 129.0, 127.1, 126.7, 126.5, 120.7, 115.2, 55.5, 24.3; HRMS (ESI-TOF) calcd for C₁₆H₁₅N₂O₂ [M+H]⁺ *m/z* = 267.1134, found 267.1131.

3.1.5 3-(4-t-Butylphenyl)-2-methylquinazolin-4(3H)-one (3e).

(96 mg, 81%) as a pale yellow solid; mp 174-175 °C; IR (CaF₂):1742, 1699 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ = 8.24 (dd, J = 8.0, 1.6 Hz, 1H), 7.73 (ddd, J = 15.2, 7.6, 1.6 Hz, 1H), 7.65 (d, J = 8.0 Hz, 1H), 7.54-7.51 (m, 2H), 7.43 (ddd, J = 14.8, 7.2, 1.2 Hz, 1H), 7.17-7.13 (m, 2H), 2.23 (s, 3H), 1.35 (s, 9H); ¹³C NMR (CDCl₃, 100 MHz): δ = 162.2, 154.6, 152.3, 147.3, 134.9, 134.5, 127.3, 127.0, 126.9, 126.6, 126.5, 120.7, 34.8, 31.3, 24.4; HRMS (ESI-TOF) calcd for C₁₉H₂₁N₂O [M+H]⁺ m/z = 293.1654, found 293.1664.

3.1.6 3-(4-Chlorophenyl)-2-methylquinazolin-4(3H)-one (3f).^{10b}

(40.2 mg, 33%) as a yellow solid; mp 175-176 °C; IR (CaF₂): 1685, 1604 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ = 8.26-8.23 (m, 1H), 7.78-7.74 (m, 1H), 7.66 (d, *J* = 8.2 Hz, 1H), 7.52 (d, *J* = 8.5 Hz, 2H), 7.48-7.44 (m, 1H), 7.20 (d, *J* = 8.5 Hz, 2H), 2.23 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz): δ = 162.2, 153.7, 147.4, 136.2, 135.4, 134.8, 130.3, 129.5, 127.1, 126.9, 126.8, 120.6, 24.4; HRMS (ESI-TOF) calcd. for C₁₅H₁₂ClN₂O [M+H]⁺ *m*/*z* = 271.0638, 273.0653; found 271.0637, 273.0652.

3.1.7 Ethyl 4-(2-methyl-4-oxoquinazolin-4-(3H)-yl)benzoate (3g).

(49.9 mg, 44%) as an orange solid; mp 188-189 °C; IR (CaF₂): 1722, 1693, 1596 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ = 8.25-8.22 (m, 3H), 7.79-7.75 (m, 1H), 7.69 (d, *J* = 8.0 Hz, 1H), 7.49-7.45 (m, 1H), 7.34 (d, *J* = 8.4 Hz, 2H), 4.41 (q, *J* = 7.2 Hz, 2H), 2.23 (s, 3H), 1.40 (t, *J* = 7.2 Hz, 3H); ¹³C NMR (CDCl₃, 100 MHz): δ = 165.4, 161.7, 153.7, 146.7, 141.3, 134.8, 131.5, 131.3, 128.2, 127.0, 126.9, 126.4, 120.4, 61.4, 24.0, 14.2; HRMS (ESI-TOF) calcd for C₁₈H₁₇N₂O₃ [M+H]⁺ *m/z* = 309.1239, found 309.1239.

3.1.8 2-Methyl-3-(naphthalen-1-yl)quinazolin-4(3H)-one (3h).

(80.1 mg, 67%) as a dark brown solid; mp 211-212 °C; IR (CaF₂): 1687, 1603 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ = 8.30-8.28 (m, 1H), 7.99 (d, *J* = 8.4, 1H), 7.94 (d, *J* = 8.4, 1H), 7.82-7.75 (m, 1H), 7.61-7.58 (m, 1H), 7.54-7.42 (m, 5H), 2.16 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz): δ = 162.1, 155.0, 147.3, 134.7, 134.5, 134.1, 129.9, 129.6, 128.7, 127.9, 127.2, 126.8, 126.8, 126.7, 126.2, 125.6, 121.3, 120.6, 23.4; HRMS (ESI-TOF) calcd for C₁₉H₁₅N₂O [M+H]⁺ *m*/*z* = 287.1184, found 287.1181.

3.1.9 2-Methyl-3-(2,4,6-trimethylphenyl)quinazolin-4(3H)-one (3i).

(82 mg, 68%) as a white solid; mp 104-105 °C; IR (CaF₂): 1678, 1610 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): $\delta = 8.27$ (dd, J = 7.6, 1.2 Hz, 1H), 7.75 (ddd, J = 15.4, 7.8, 1.2 Hz, 1H), 7.67 (d, J = 7.6 Hz, 1H), 7.44 (ddd, J = 15.2, 7.8, 1.2 Hz, 1H), 7.01 (s, 2H), 2.33 (s, 3H), 2.13 (s, 3H), 2.01 (s, 6H); ¹³C NMR (CDCl₃, 100 MHz): $\delta = 161.2$, 154.6, 147.7, 139.1, 134.6, 134.5, 133.3, 129.8, 127.2, 126.7, 126.4, 120.7, 23.1, 21.0, 17.5; HRMS (ESI-TOF) calcd for C₁₈H₁₉N₂O [M+H]⁺ m/z = 279.1497, found: 279.1493.

(60 mg, 52%) as a pale brown solid; mp 120-121 °C; IR (KBr): 1723, 1674 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): $\delta = 8.23$ (d, J = 8.0 Hz, 1H), 7.74 (t, J = 8.0 Hz, 1H), 7.66 (d, J = 8.0 Hz, 1H), 7.44 (t, J = 7.6 Hz, 1H), 6.45 (s, 2H), 3.88 (s, 3H), 3.82 (s, 6H), 2.31 (s, 2H); ¹³C NMR (CDCl₃, 100 MHz): $\delta = 162.1$, 154.5, 154.1, 147.0, 138.5, 134.7, 133.1, 127.0, 126.7, 126.5, 120.5, 105.1, 60.9, 56.2, 23.8; HRMS (ESI-TOF) calcd for C₁₈H₁₉N₂O₄ [M+H]⁺ m/z = 327.1345, found: 327.1346.

3.1.11 2-Propyl-3-(p-tolyl)quinazolin-4(3H)-one (3k).

(76.2 mg, 56%) as a yellow solid; mp 151-152 °C; IR (CaF₂): 1682, 1594 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): $\delta = 8.24$ (d, J = 8.0 Hz, 1H), 7.75-7.67 (m, 2H), 7.44-7.40 (m, 1H), 7.32 (d, J = 7.8 Hz, 2H), 7.10 (d, J = 8.0 Hz, 2H), 2.43-2.38 (m, 5H), 1.70 (q, J = 7.5 Hz, 2H), 0.85 (t, J = 7.2 Hz, 3H); ¹³C NMR (CDCl₃, 100 MHz): $\delta = 162.5$, 157.1, 147.4, 139.2, 134.6, 134.3, 130.4, 127.9, 127.0, 126.9, 126.4, 120.7, 37.6, 21.2, 20.5, 13.7; HRMS (ESI-TOF) calcd for C₁₈H₁₉N₂O [M+H]⁺ m/z = 279.1497, found 279.1501.

3.1.12 2-Isopropyl-3-(p-tolyl)quinazolin-4(3H)-one (3l).

(55.2 mg, 41%) as an orange solid; mp 151-152 °C; IR (CaF₂): 1682, 1604 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): $\delta = 8.24$ (d, J = 7.6 Hz, 1H), 7.78-7.74 (m, 2H), 7.45-7.41 (m, 1H), 7.33 (br, 2H), 7.13 (br, 2H), 2.7 (br, 1H), 2.4 (s, 3H), 1.24 (d, J = 2.0 Hz, 6H); ¹³C NMR (CDCl₃, 100 MHz): $\delta = 162.6$, 162.1, 147.3, 139.2, 134.6, 134.4, 130.5, 128.0, 127.0, 126.9, 126.5, 120.7, 32.3, 21.3; HRMS (ESI-TOF) calcd for C₁₈H₁₉N₂O [M+H]⁺ m/z = m/z 279.1497, found 279.1506.

3.1.13 2-Phenyl-3-(p-tolyl)quinazolin-4(3H)-one (**3m**).^{13c}

(72.6 mg, 48%) as a pale yellow solid; mp 185-186 °C; IR (CaF₂): 1685, 1604 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ = 8.26 (d, *J* = 7.9 Hz, 1H), 7.75-7.69 (m, 2H), 7.45-7.41 (m, 1H), 7.27-7.25 (m, 2H), 7.19-7.11 (m, 3H), 7.02 (d, *J* = 8.1 Hz, 2H), 6.94 (d, *J* = 8.2 Hz, 2H), 2.21 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz): δ = 162.3, 155.3, 147.4, 138.2, 135.5, 134.9, 134.6, 129.6, 129.2, 128.9, 128.7, 127.9, 127.6, 127.1, 120.9, 21.1; HRMS (ESI-TOF) calcd for C₂₁H₁₇N₂O [M+H]⁺ *m*/*z* = 313.1341, found 313.1335.

3.1.14 2-Benzyl-3-(p-tolyl)quinazolin-4(3H)-one (3n).^{13d}

(84.0 mg, 53%) as a yellow oil; IR (CaF₂): 1684, 1593 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ = 8.27-8.25 (m, 1H), 7.77-7.75 (m, 2H), 7.48-7.44 (m, 1H), 7.18-7.14 (m, 5H), 6.89-6.87 (m, 2H), 6.82 (d, *J* = 8.2 Hz, 2H), 3.90 (s, 2H), 2.40 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz): δ = 162.6, 155.5, 147.4, 139.1, 135.4, 134.4, 134.2, 130.0, 128.6, 128.3, 128.3, 127.2, 127.0, 126.8, 126.8, 120.9, 42.6, 21.2; HRMS (ESI-TOF) calcd. for C₂₂H₁₉N₂O [M+H]⁺ *m/z* = 327.1497, found 327.1505.

3.1.14 2,6-Dimethyl-3-(p-tolyl)quinazolin-4(3H)-one (3o).

(118.1 mg, 92%) as a pale yellow solid; mp 121-122 °C; IR (KBr): 1676, 1596 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): $\delta = 8.00$ (d, J = 0.6 Hz, 1H), 7.53-7.52 (m, 2H), 7.31 (d, J = 8.4 Hz, 2H), 7.10 (d, J = 8.2 Hz, 2H), 2.43 (s, 3H), 2.40 (s, 3H), 2.19 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz): $\delta = 162.3$, 153.4, 145.4, 139.1, 136.5, 135.8, 135.1, 130.5, 127.6, 126.4, 126.3, 120.4, 24.2, 21.1(2C); HRMS (ESI-TOF) calcd. for C₁₇H₁₇N₂O [M+H]⁺ m/z = 265.1314, found 265.1354.

3.1.15 7-Bromo-2-methyl-3-phenylquinazolin-4(3H)-one (3p).

(103 mg, 63%) as a white solid; mp 173-174 °C; IR (KBr): 1682, 1585 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ = 8.01 (d, *J* = 8.4 Hz,

7.2 Hz, 2H), 2.20 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz): $\delta = 161.7$, 155.6, 148.4, 137.4, 130.0, 129.9, 129.5, 129.4, 129.2, 128.4, 127.8, 119.5, 24.4; HRMS (ESI-TOF) calcd. for C₁₅H₁₂⁷⁹BrN₂O [M+H]⁺ m/z = 315.0133, found 315.0135; HRMS (ESI-TOF) calcd. for C₁₅H₁₂⁸¹BrN₂O [M+H]⁺ m/z = 317.0113, found 317.0122.

3.1.16 6,7-dimethoxy-2-methyl-3-phenylquinazolin-4(3H)-one (3q).

(105 mg, 68%) as a brown solid; mp 240-241 °C; IR (KBr): 1675, 1614 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 7.53-7.44 (m, 4H), 7.23 (d, *J* = 7.2 Hz, 2H), 7.06 (s, 1H), 3.97 (s, 3H), 3.93 (s, 3H), 2.18 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz): δ = 161.5, 155.1, 152.9, 148.8, 143.5, 137.8, 129.8, 129.1, 128.0, 113.8, 107.1, 105.9, 56.3, 56.2, 24.0; HRMS (ESI-TOF) calcd. for C₁₇H₁₇N₂O₃ [M+H]⁺ *m*/*z* = 297.1239, found 297.1237.

3.1.17 2-methyl-3-(o-tolyl)quinazolin-4(3H)-one (3r).^{13a}

(473 mg, 78%) as a white solid; mp 114-115 °C; IR (KBr): 1683, 1598 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz): δ 8.24 (dd, *J* = 8.0, 1.2 Hz, 1H), 7.72 (ddd, *J* = 15.2, 7.6, 1.6 Hz, 1H), 7.65 (dd, *J* = 8.0, 0.4 Hz, 1H), 7.42 (ddd, *J* = 15.0, 7.6, 0.8 Hz, 1H), 7.37-7.30 (m, 3H), 7.12 (d, *J* = 7.2 Hz, 1H), 2.14 (s, 3H), 2.09 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz): δ = 161.4, 154.1, 147.5, 136.6, 135.2, 134.4, 131.3, 129.4, 127.8, 127.5, 126.9, 126.6, 126.4, 120.6, 23.7, 17.2; HRMS (ESI-TOF) calcd. for C₁₆H₁₅N₂O [M+H]⁺ *m/z* = 251.1184, found 251.1181.

3.1.18 3-([1,1'-biphenyl]-3-yl)-2-methylquinazolin-4(3H)-one (3s).

(89 mg, 81%) as a colorless viscous oil; ¹H NMR (CDCl₃, 400 MHz): $\delta = 8.29$ -8.26 (m, 1H), 7.69 (ddd, J = 15.6, 7.6, 1.6 Hz, 1H), 7.54-7.49 (m, 4H), 7.42 (ddd, J = 15.6, 7.4, 1.2 Hz, 1H), 7.29-7.25 (m, 3H), 7.21-7.18 (m, 3H), 2.03 (s, 3H); ¹³C NMR (CDCl₃, 100 MHz): $\delta = 162.9$, 154.0, 147.4, 140.3, 137.8, 135.3, 134.4, 131.5, 129.7, 129.0, 128.8, 128.5, 128.1, 127.8, 127.1, 126.6, 126.4, 120.3, 23.9; HRMS (ESI-TOF) calcd for C₂₁H₁₇N₂O [M+H]⁺ m/z = 313.1341, found 313.1341.

3.1.19 ethyl 2-(4-oxo-3-phenyl-3,4-dihydroquinazolin-2-yl) acetate (**3t**) and ethyl 2-(4-oxo-3-phenyl-3,4-dihydroquinolin-2(1H)-ylidene)acetate (**3t**').

(108 mg, 68%) as a pale yellow viscous oil; (3t : 3t' = 1: 0.33) inseparable mixture); Compound 3t: ¹H NMR (CDCl₃, 400 MHz): $\delta = 8.28-8.26$ (m, 1H), 7.70 (ddd, J = 15.2, 7.6, 1.6 Hz, 1H), 7.71-7.69 (m, 1H), 7.56-7.53 (m, 1H), 7.52-7.51 (m, 1H), 7.51-7.50 (m, 1H), 7.49-7.48 (m, 1H), 7.27-7.25 (m, 1H), 7.24-7.23 (m, 1H), 4.07-4.03 (m, 2H), 3.58 (s, 2H), 1.16 (t, J = 7.2 Hz, 3H); ¹³C NMR (CDCl₃, 100 MHz): $\delta = 167.8$, 150.2, 147.2, 136.9, 134.6, 130.1, 129.9, 129.6, 128.5, 128.4 (2C), 127.2, 127.0, 61.5, 42.9, 14.0. Compound **3t'** ¹H NMR (CDCl₃, 400 MHz): $\delta = 12.15$ (brs, 0.33 H), 8.05-8.03 (m, 0.33 H), 7.59-7.56 (m, 0.32 H), 7.47-7.46 (m, 0.62 H), 7.22-7.21 (m, 0.33 H), 7.11 (ddd, J = 15.2, 7.8, 1.2 Hz, 0.34 H), 7.06 (d, J = 8.4 Hz, 0.34 H),4.11-4.08 (m, 0.67 H), 1.20 (t, J = 7.2 Hz, 1H); ¹³C NMR $(CDCl_3, 100 \text{ MHz}): \delta = 170.9, 162.2, 160.2, 153.6, 139.4, 135.7,$ 135.4, 129.3, 128.3, 127.3, 122.6, 121.1, 115.1, 114.8, 59.3, 14.4; HRMS (ESI-TOF) calcd for $C_{18}H_{17}N_2O_3$ [M+H]⁺ m/z = 309.1239, found 309.1234.

3.2 Crysatllography.

Crystals suitable for X-ray determination were obtained for 3r by slow evaporization of a chloroform solution at room temperature. Cell parameters were determined by a Siemens SMART CCD

diffractometer. Crystal data: $C_{16}H_{14}N_2O$, $AF_wCEP250.29$, MANUS Monoclinic, $P2_1/n$, a = 8.6926(5), b = 15.8320(5) Å, c =10.0440(5) Å, $\alpha = 90$, $\beta = 110.282(6)^{\circ}$, $\gamma = 90^{\circ}$, V = 1296.56(12)Å³, Z = 4, $D_{calcd} = 1.282 \text{ Mg/m}^3$, F(000) = 528, crystal size : 0.25 x 0.20 x 0.15 mm³, 3.8290 to 28.7710°, 7038 reflections collected, 2887 reflections [R(int) = 0.0311], Final R indices [I>2sigma(I)]: R1 = 0.0559, wR2 = 0.1543, for all data R1 =0.0799, wR2 = 0.1810, Goodness-of-fit on $F^2 = 0.937$. The structure was solved using the SHELXS-97 program¹⁴ and refined using the SHELXL-97 program¹⁵ by full-matrix leastsquares on F2 values. CCDC 1880653 contain the supplementary crystallographic data for complexes 3r, which can be obtained free of charge via http://www.ccdc.cam.ac.uk/conts/retrieving.html.

Acknowledgments

We thank the Ministry of Science and Technology for financial support (MOST-106-2113-M-002-018). "National Taiwan University Mass Spectrometry-based Proteomics Core Facility" for the measurement of ESI mass data is acknowledged.

References

1. (a) J. Peng, X.-Y. Zhang, Z.-C. Tu, X.-Y. Xu, S.-H. Qi, J. Nat. Prod. 76 (2013) 983-987;

(b) N.P. McLaughlin, P. Evans, M. Pines, Bioorg. Med. Chem. 22 (2014) 1993-2004;

(c) N.P. McLaughlin, P. Evans, P. J. Org. Chem. 75 (2010) 518-521;

(d) F. He, J. Bao, F. He, J. Bao, X.-Y. Zhang, Z.-C. Tu, Y.-M Shi, S.-H. Qi, J. Nat. Prod. 76 (2013) 1182-1186, and references therein.

- (a) U. A. Kshirsagar, Org. Biomol. Chem. 13 (2015) 9336-9352;
 (b) Maruthamuthu; S. Rajam, P. C. R. Stella, A. G. B. Dileepan, R. Ranjith, J. Chem. Pharm. Res. 8 (2016) 505-526;
 (c) G.N. Raju, K.B. Sai, V. Resshma, N. Sudarshini, P.L. Sowmya, Y. Nalini, P.R. Nadendla, J. Chem. Pharm. Res. 7 (2015) 1279-1287.
- (a) H. Hammer, B.M. Bader, C. Ehnert, C. Bundgaard, L. Bunch, K. Hoestgaard-Jensen, O.H.-U. Schroeder, J.F. Bastlund, A. Gramowski-Voss, A.A. Jensen, Mol. Pharmacol. 88 (2015) 401-420;

(b) I.R. Ager, D.R. Harrison, P.D. Kennewell, J.B. Taylor, J. Med. Chem. 20 (1977) 379-386.

- (a) Y. Yamada, M. Otsuka, J. Tani, T. Oine, Chem. Pharm. Bull. 31 (1983) 1158-1165;
 - (b) H. Fujita, I. Matsuo, Chem. Biol. Interact. 64 (1987) 139-149.(a) Y. Takase, T. Saeki, N. Watanabe, H. Adachi, S. Souda, I.
- Saito, J. Med. Chem. 37 (1994) 2106-2111;
 (b) M.-J. Hour, L.-J. Huang, S.-C. Kuo, Y. Xia, K. Bastow, Y. Nakanishi, E. Hamel, K.-H. Lee, J. Med. Chem. 43 (2000) 4479-4487;
 (c) G.M. Chinigo, M. Paige, S. Grindrod, E. Hamel, S.

Dakshanamurthy, M. Chruszcz, W. Minor, M.L. Brown, J. Med. Chem. 51 (2008) 4620-4631;

(d) M.A. Mohamed, R.R. Ayyad, T.Z. Shawer, A.A.-M. Abdel-Aziz, A.S. El-Azab, Eur. J. Med. Chem. 112 (2016) 106-113.

- S.B. Mhaske, N.P. Argade, Tetrahedron 62 (2006) 9787-9826.
 N.P. Peet, L.E. Baugh, S. Sunder, J.E. Lewis, E.H. Matthews,
- N.F. Feet, L.E. Baugh, S. Sunder, J.E. Lewis, E.H. Mathlews, E.L. Olberding, D.N. Shah, J. Med. Chem. 29 (1986) 2403-2409.
 (a) L. Orfi, J. Kokosi, G. Szasz, I. Kovesdi, M. Mak, I. Teplan, G.
- (a) L. OHI, J. ROKOSI, C. SZASZ, I. ROVESUI, M. MAR, I. FEPIAII, C. Keri, Bioorg. Med. Chem. 4 (1996) 547-551;
 (b) J. Lin, W. Shen, J. Xue, J. Sun, X. Zhang, C. Zhang, Eur. J. Med. Chem. 55 (2012) 39-48.
- (a) S. Smullen, N.P. McLaughlin, P. Evans, Bioorg. Med. Chem. 26 (2018) 2199-2220;

(b) K. Hemalatha, G. Madhumitha, Eur. J. Med. Chem. 123 (2016) 596-630;

(c) T. M. M. Maiden, J. P. A. Harrity, Org. Biomol. Chem. 14 (2016) 8014-8025;

(d) S.Y. Abbas, K.A.M. El-Bayouki, W.M. Basyouni, Synth. Commun. 46 (2016) 993-1035; US (e) I. Khan, S. Zaib, S. Batool, N. Abbas, Z. Ashraf, J. Iqbal, A. Saeed, Bioorg. Med. Chem. 24 (2016) 2361-2381;
(f) R.S. Rohokale, U.A. Kshirsagar, Synthesis 48 (2016) 1253-1268;
(g) I. Khan, A. Ibrar, W. Ahmed, A. Saeed, Eur. J. Med. Chem. 90 (2015) 124-169,
(h) L. He, H. Li, J. Chen, X.-F. Wu, RSC Adv. 4 (2014) 12065-12077, and references therein.
10. (a) K. Rad-Moghadam, M.S. Khajavi, J. Chem. Res., Synop. (1998) 702-703;
(b) G. Shen, H. Zhou, Y. Sui, Q. Liu, K. Zou, Tetrahedron Lett. 57 (2016) 587-590.
11. (a) M. Ramanathan, Y.-H. Wang, Y.-H. Liu, S.-M. Peng, Y.-C. Cheng, S.-T. Liu, J. Org. Chem. 83 (2018) 6133-6141;

- Cheng, S.-T. Liu, J. Org. Chem. 83 (2018) 6133-6141;
 (b) M. Ramanathan, Y.-H. Liu, S.-M. Peng, S.-T. Liu, Org. Lett. 19 (2017) 5840-5843, and references therein.
- 12. T. van Dijk, J.C. Slootweg, K. Lammertsma, Org. Biomol. Chem. 15 (2017) 10134-10144.
- (a) W. Phakhodee, S. Wangngae, M. Pattarawarapan, J. Org. Chem. 82 (2017) 8058-8066;
 (b) C. Bingi, K.Y. Kola, A. Kale, J.B. Nanubolu, K. Atmakur, Tetrahedron 58 (2017) 1071-1074;
 (c) H. Wang, X. Cao, F. Xiao, S. Liu, G. Deng, Org. Lett. 15 (2013) 4900-4903;
 (d) J.E.R. Sadig, R. Foster, F. Wakenhut, M.C. Willis, J. Org. Chem. 77 (2012) 9473-9486.
- G.M. Sheldrick, SHELXS-97, Acta Crystallogr., Sect. A: Found. Crystallogr. 46 (1990) 67.
- G.M. Sheldrick, SHELXL-97, University of Göttingen, Göttingen, Germany, 1997.

Supplementary Material

Spectra of quinazolinones and crystal data of **3r** are deposited as supplementary material. Supplementary data related to this article can be found at https://doi.org/xxxxxxxxxx