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Pd(II)-Catalyzed alkyne annulation through allylic isomerization: synthesis of spiro-cyclopentadiene pyrazolones†

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The Pd(μ)-catalyzed activation of Csp²–H bond and double alkyne annulation which proceeds *via* allylic isomerization is reported for the first time. This reaction of antipyrines with alkynes provides an efficient synthetic route for the biologically important spiro-cyclopentadiene pyrazolones. In the presence of Lawesson's reagent, this Pd(μ)-catalyzed annulation reaction affords another spiro-cyclopentadiene pyrazolone which displays very good fluorescence properties.

In the past few decades, transition-metal-catalyzed hetero atom directed C-H bond activation and oxidative alkyne annulation reactions have emerged as a powerful tool for the construction of a wide range of heterocycles.¹ Jiao and co-workers developed a Pd-catalyzed reaction to activate two C(sp²)-H bonds and alkyne annulation for the synthesis of aromatic carbocycles.² Nakao and co-workers developed a Ni-catalyzed alkyne annulation reaction of formamides via functionalization of one $C(sp^3)$ -H bond and one $C(sp^2)$ -H bond (Scheme 1, eqn (1)).³ Lam and co-workers performed the alkyne annulation via C(sp²)-H bond activation, sequential keto enol tautomerization and C(sp³)-H bond activation for the synthesis of spirocycles (Scheme 1, eqn (2)).⁴ Yao *et al.* reported a Rh-catalyzed alkyne annulation via enol directed formal C(sp³)-H bond activation of α -arylidene pyrazolones for the synthesis of spiro-pyrazolones (Scheme 1, eqn (3)).⁵ In continuation of our work on metalcatalyzed reactions,⁶ herein, we report an unprecedented alkyne annulation reaction which involves the activation of one olefinic C(sp²)-H bond, and the annulation proceeds via the allyl bond isomerization (Scheme 1, eqn (4)). This Pd-catalyzed annulation reaction of antipyrines and alkynes provides an efficient route to access 4-spiro-5-pyrazolones. The

4-spiro-5-pyrazolones and their derivatives have attracted considerable attention recently, due to their interesting biological activities. These compounds exhibit miticidal activity, inhibit type-4 phosphodiesterases and they are also used as PPAR α antagonists (Fig. 1).⁷ Therefore, a new efficient method to access this core structure is highly desirable. Furthermore, we developed an unprecedented Pd(II)-catalyzed Lawesson's reagent mediated cascade annulation reaction of antipyrines with alkynes for the synthesis of spiro-pyrazolones which proceeds *via* 1,2-methyl shift (Scheme 1, eqn (4)).

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Initially, antipyrine **1a** and alkyne **2a** were chosen to optimize the reaction conditions for the synthesis of the spiro compound **3aa**. As shown in Table S1 (ESI[†]), among the metal complexes studied for this annulation reaction, $Pd(OAc)_2$ catalyst along with the additive $Cu(OAc)_2$, provided the highest yield of **3aa** (48%) in 1,4-dioxane (entry 3). Then, studies on solvents revealed toluene and ^tAmOH as the most suitable solvents that provided 74% and 79% yields of **3aa**, respectively (entries 9 and 11). These optimized reaction conditions were then tested to study the substrate scope of various substituted pyrazolones **1a–k** with alkyne **2a** which is shown in Table 1.



Scheme 1 Examples of various alkyne annulation reactions.

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The pyrazolones possessing electron-donating and electronwithdrawing substituents such as methyl, isopropyl, fluoro and chloro on the 2-phenyl ring **1b**, **1d–g** were tested with **2a**. These reactions provided the spiro compounds **3ba**, **3da–ga** with yields in the range of 56–70%, irrespective of the position of the substituent on the phenyl ring. Similarly, dimethyl



substituted pyrazolone **1c** was also found to be a good substrate to provide **3ca** in 71% yield. Then, the annulation reaction of pyrazolones containing substituents at 5-position of the pyrazolone ring was studied. Thus, the studied typical pyrazolones substituted at 5-position with *n*-propyl and iso-propyl groups **1h–j** were also found to be suitable substrates to perform this annulation reaction to afford **3ha–ja** in 62–67% yields. The spiro compound **3ha** was a mixture of both the geometrical isomers (E:Z = 3:1) which was determined by the NOE spectrum. Finally, pyrazolone substituted with a benzyl group at 1-position **1k** was tested for the annulation reaction with **2a** to afford **3ka** in 77% yields. The structures of all the compounds were determined with the help of NMR spectroscopy and finally confirmed by single X-ray crystallography studies of compound **3aa.**⁸

Then the scope of the alkynes 2b-i was studied with antipyrine (1a) as shown in Table 2. The symmetrically substituted alkynes, substituted with electron-releasing and electronwithdrawing substituents such as methyl, methoxy and fluoro at 4-position of the phenyl ring 2b-d afforded the spiro compounds 3ab-ad with yields in the range of 66-76%. Similarly, symmetrical disubstituted alkyne substituted with an electronwithdrawing fluoro substituent at the 3-position of the phenyl ring 2e was also found to be a suitable substrate for this reaction to provide 3ae in 73% yield. The reaction of the aryl and alkyl group substituted alkyne 2f with 1a was highly regioselective to provide only one isomer of the spiro compound 3af. However, alkyne 2g substituted with a phenyl group and an ethyl group afforded a 3:1 mixture of spiro compound 3ag. The regioselectivity of these unsymmetrical alkynes is similar to the previously reported metal-catalyzed alkyne annulation reactions, where the sterically hindered phenyl ring containing centre of the alkyne binds with the metal during the insertion of alkyne in the C-M bond, leading to the preferential formation of one isomer.9 Finally, the dialkyl substituted alkynes 2h-i were tested for this annulation reaction, and they also turned out to be suitable substrates for this annulation reaction to provide spiro compounds 3ah-ai, though

Table 2 Scope of alkynes 2b-i with 1a^a Pd(OAc)₂ (5.0 mol %) Cu(OAc)₂,H₂C (1.0 equiv) ^tAmOH, 90 °C Ρh R¹ air. 8 h Рń 1a 2b-i 3ab-ai 3ab (R¹ = R² = 4-Me-Ph, 76%) 3ac (R¹ = R² = 4-MeO-Ph, 70%) **3ad** ($R^1 = R^2 = 4$ -F-Ph, 66%) **3ae** (R¹ = R² = 3-F-Ph, 73%) 3af (R¹ = Me, R² = Ph, 58%) **3ag** (R¹ = Et, R² = Ph, 55%, r.r. = ~ 3:1)^b **3ah** (R¹= R² = *n*-Propyl, 55%) **3ai** ($\mathbb{R}^1 = \mathbb{R}^2 = n$ -Butyl, 52%)

^{*a*} Reaction conditions: **1** (0.5 mmol), **2a** (1.0 mmol), Pd(OAc)₂ (5.0 mol%) and Cu(OAc)₂·H₂O (1.0 equiv.) in ^{*t*}AmOH (5.0 mL) at 90 °C for 8 h under air. ^{*b*} For clarity one molecule is shown (for details see ESI).

^{*a*} Reaction conditions: **1a** (0.5 mmol), **2** (1.0 mmol), $Pd(OAc)_2$ (5.0 mol%) and $Cu(OAc)_2 \cdot H_2O$ (1.0 equiv.) in ^{*t*}AmOH (5.0 mL) at 90 °C for 8 h under air. ^{*b*} Major isomer is shown, ratio of regioisomers (r.r.) was determined by ¹H NMR.

with comparatively less yields (52–55%). The terminal alkynes, trimethyl(phenylethynyl)silane, methyl 3-phenylpropiolate and 1-methyl-2,5-diphenyl-1,2-dihydro-3*H*-pyrazol-3-one were not found to be suitable substrates for this reaction.

To see the synthetic utility of the spiro product, we performed some of the transformation reactions of 3aa. As shown in Scheme 2, the hydrogenation reaction of 3aa in the presence of Pd/C as the catalyst provided a mixture of hydrogenated products 4a and 4b, which could be separated using silica gel column chromatography. Hydroboration of 3aa with borane complex and subsequent reduction of the resultant boron complex in the presence of Pd/C in methanol provided the same exocyclic double bond reduced product 4a. Interestingly, the bromination reaction of 3aa provided a geminal dibromo spiro compound 4c. This dibromo compound 4c was formed from 3aa via bromination-dehydrobromination and subsequent repeated bromination-dehydrobromination of the resulted monobromo olefin. This dibromo compound would be a suitable synthon for further functionalization. Next, we attempted to transform the pyrazolone 3aa to its corresponding thiopyrazolone by treatment with Lawesson's reagent in toluene. Interestingly, instead of providing the thiopyrazolone, this reaction provided the spiro-cyclopentadiene pyrazolone 4aa in 84% yield (Scheme 3). Further studies revealed that 4aa could also be synthesized in one pot from 1a and 2a, using the optimized reaction conditions and toluene as the solvent, followed by the addition of Lawesson's reagent into the reaction mixture (Scheme 3). Owing to the importance of these spiro-pyrazolones,⁷ we studied these novel one-pot and two-steps reactions to synthesize some of the representative spiro-pyrazolones which is shown in Scheme 3. The one pot synthesis provided inferior yields of spiro compounds (42-59%) as compared to the two-steps reaction (64-84%). The structure of these compounds was determined using NMR spectroscopy and confirmed using X-ray crystallography studies of compound 4aa.¹⁰ Moreover, treatment of 1a alone under standard conditions in D₂O for 4 h provided deuterated 1a, where the fourth C-H bond of 1a is exchanged with D atom (35%).

Recently, the polyarylated compounds have attracted considerable attention owing to their notable photochemical properties, which make them useful compounds for organic semiconductors. Therefore, a preliminary study of the optical properties of all the synthesized compounds was carried out.¹¹ The absorption bands of all the synthesized spiro-pyrazolones appeared in the region 320–360 nm. Among all the compounds studied, only compounds shown in Scheme 3 exhibited intense fluorescence in the range from 438 to 457 nm with quantum efficiency (Φ) of 48% (4aa), 75% (4ab), 64% (4ad), 52% (4ca), 61% (4da), 30% (4fa) and 47% (4ga) (Fig. 2).



Scheme 2 Transformation of 4-spiro-5-pyrazolone.



Scheme 3 LR-mediated synthesis of spiro-pyrazolones. ^a Reaction conditions: **1** (0.5 mmol), **2** (1.0 mmol), Pd(OAc)₂ (5.0 mol%) and Cu(OAc)₂·H₂O (1.0 equiv.) in toluene (5.0 mL), 90 °C for 8 h under air, then Lawesson's reagent (0.5 mmol), 90 °C for 6 h. ^b Yield from two-step reaction performed in toluene starting with **3** at 90 °C for 6 h.

A plausible mechanism for the formation of **3aa** is proposed in Scheme 4. Initially, direct activation of the olefinic $C(sp^2-H)$ bond of antipyrine forms the Pd(II) complex **A**. Then, metalalkyne coordination followed by insertion of the alkyne into the Pd–C bond results in the Pd(II) complex **B**. Subsequent allylic isomerization affords π -allylpalladium species **C**.¹² Insertion of one more molecule of an alkyne into the Pd–C bond of **C** affords a six membered Pd-complex **D**. Finally, reductive elimination and subsequent oxidation of the metal in the presence of air and Cu(OAc)₂ provides **3aa** and the active catalyst. Again, [3+2] cycloaddition of **3aa** with LR generates **E**, which on subsequent 1,2-methyl shift followed by elimination of phosphorous sulphide **G**¹³ finally led to the formation of **4aa**.

In conclusion, we have developed a new Pd-catalyzed alkyne annulation reaction, where annulation proceeds *via* the activation of one olefinic $C(sp^2-H)$ bond followed by the rearrangement of one allylic $C(sp^3-H)$ bond. This annulation reaction of antipyrines and disubstituted alkynes provides an efficient method for the synthesis of 4-spiro-5-pyrazolones which are the key structural motif of several bioactive compounds.



Fig. 2 Fluorescence $(2.0 \times 10^{-5} \text{ M}, \text{ excitation at 340 nm})$ spectra of spiropyrazolones **4** in toluene using quinine sulfate (QS) as the standard.



Moreover, we have developed another unprecedented Lawesson's reagent mediated and Pd-catalyzed annulation reaction for the synthesis of spiro-cyclopentadiene pyrazolones which exhibited fluorescence properties.

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Conflicts of interest

There are no conflicts to declare.

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