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## Gold-Catalyzed Stereocontrolled Oxacyclization/[4+2]-Cycloaddition Cascade of Ketone-Allene Substrates

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**Abstract:** We report the first success on the Au-catalyzed tandem oxacyclization/[4+2]-cycloaddition cascade using ketone-allene substrates to give highly substituted oxacyclics with excellent stereocontrol. In contrast to oxo-alkyne substrates, the resulting cycloadducts are isolable and efficiently produced from a reasonable scope of enol ethers.

Metal-catalyzed cycloaddition/annulation reactions are important tools to access complex molecular frameworks.<sup>1</sup> The Au- and Pt-catalyzed activation of alkynes enables the generation of unusual intermediates to react with dipolarophiles in a cycloaddition fashion.<sup>2</sup> Such reactions have attained considerable success only on oxo-alkyne substrates.<sup>1a,b</sup> In the presence of Au or Pt catalysts, 2-oxo-1-alkynylbenzenes form metal-containing benzopyriliums (I) or carbonyl ylides (I') that react with dipolarophiles to give hypothetic [4+2]- or [3+2]-cycloaddition intermediates.<sup>3,4</sup> Generation of 1,n-dipole species remains unexplored for oxo-allene substrates. This approach is mechanistically appealing because of the uncertain workability of oxonium-vinylmetal (II) or benzopyrilium (III) as 1,n-dipoles (n = 4, 5). Species II is kinetically favored by the Pt or Au- $\pi$ -allene bonding,<sup>5</sup> but its participation is absent in this work.

## Scheme 1



We prepared alkynyl acetate 1a readily from 2'-bromoacetophenone. This substrate works as a precursor to generate ketone-allene 3a via a catalytic 1,3-acyloxy shift.<sup>6</sup> As shown in Table 1, treatment of compound 1a with n-butyl vinyl ether (2a, 3 equiv) with PtCl<sub>2</sub>/ CO or AuCl<sub>3</sub> in hot dichloroethane (50-80 °C) gave an exclusive recovery of unreacted 1a. The use of PPh<sub>3</sub>AuCl/AgSbF<sub>6</sub> (3 mol %) in dichloromethane (DCM) produced ketone-allene 3a in 83% yield, but with no tractable amount of cycloadducts 4a/4a'. We were pleased to discover that ClAuP(t-Bu)2(o-biphenyl)/AgSbF6 (3 mol %) enabled the desired cyclization/cycloaddition cascade to give 4a (29%) and 4a' (56%) as a diastereomeric mixture, separable on a silica column. The stereoselectivity is greatly enhanced with ClAuP(t-Bu)<sub>2</sub>(o-biphenyl)/AgNTf<sub>2</sub>, giving only 4a' in 79% yield (entry 5); within a brief time (1 h), we obtained ketone-allene 3a in 40% yield in addition to the desired 4a' (46%, entry 6). The intermediacy of ketone-allene 3a was confirmed by a complete 
 Table 1.
 Screening of Catalytic Activity over Various Metal Catalysts



entry	substrate <sup>a</sup>	catalyst (mol %)	condition	product (yield) <sup>b</sup>
1	1a	$AuCl_3$ (5)	DCE	1a (85%)
2	1a	PtCl <sub>2</sub> /CO (5)	(80 °C, 24 h) DCE (50 °C, 20 h)	<b>1a</b> (46%) <sup>c</sup>
3	1a	Ph <sub>3</sub> PAuCl (3)/	DCM	<b>3a</b> (83%)
4	1a	$AgSbF_6$ (3) AuCIL (3)/	(30 °C, 8 h) DCM	<b>4a</b> (29%),
5	1a	AgSbF <sub>6</sub> (3) AuCIL (3)/	(30 °C, 2 h) DCM	4a' (56%) 4a (trace),
6	1a	$\begin{array}{c} \text{AgNTf}_2(3) \\ \text{AuCIL}(3) \\ \end{array}$	$(30 \ ^{\circ}C, 2.5 \ h)$ DCM $(20 \ ^{\circ}C, 1 \ h)$	4a' (79%) 3a (40%), 4a' (40%)
7	3a	AgN $\Pi_2$ (3) AuCIL (3)/	$(30  ^\circ\mathrm{C}, 1  \mathrm{n})$ DCM $(20  ^\circ\mathrm{C}, 2  \mathrm{h})$	<b>4a</b> $(46\%)$ <b>4a</b> $(trace),$
8	1a	Agivit $_2(3)$ AuCIL (5)	$(30^{\circ}C, 21)$ DCM $(30^{\circ}C, 20 h)$	<b>1a</b> (94%)
9	1a	AgNTf <sub>2</sub> $(5)$	DCM (30 °C 20 h)	<b>1a</b> (62%), <b>3a</b> (17%)
10	1a	IPrAuCl (3)/ AgNTf <sub>2</sub> (3)	DCM (30 °C, 2 h)	<b>4a</b> (48%), <b>4a'</b> (22%)

 $^{a}$ L = P(*t*-Bu)<sub>2</sub>(*o*-biphenyl), [substrate] = 0.1 M, DCM = dichloromethane, DCE = 1,2-dichloroethane.  $^{b}$  Isolated yields were reported after purification from a silica gel column.  $^{c}$  Decomposition of starting **1a** was observed in entry 2.

conversion to **4a**' with the same gold catalyst (entry 7). Control experiments indicate that  $ClAuP(t-Bu)_2(o-biphenyl)$  and  $AgNTf_2$  alone were catalytically inactive (entries 8–9). An altered chemose-lectivity was observed for IPrAuCl/AgNTf<sub>2</sub> (IPr = 1,3-bis(diiso-propyl-phenyl)imidazol-2-ylidene) that preferably gave **4a** as the major product (entry 10). Characterization of the structure of product **4a**' relies on an X-ray diffraction study of its analogue **4b**' (Table 2, entry 1).

We prepared ketone substrates 1b-1e bearing altered R<sup>1</sup> and R<sup>2</sup> substituents to examine the scope of this catalysis, as depicted in Table 2. Herein, the aldehyde substrates are not studied due to the intrinsic instability of 2-allenyl benzaldehyde intermediates.<sup>7</sup> This reaction works well with both ethyl- and *n*-butyl vinyl ether (**2a**-**2b**), and it is also efficient with alterations of the R<sup>1</sup> (Me, *n*-Bu, *i*-Bu) and R<sup>2</sup> alkyls (Me, *n*-Pr) of substrates **1a**-**1e**. In entries 2, 4, and 5, due to the poor diastereoselectivity or chromatographic inseparable property, the cycloadducts **4b/4b'** and **4c/4c'** were subject to deacylation with NaOMe in MeOH to give ketone derivatives **5b**, **6b**, and **6c** with high dr values (8.5-10.0:1). For

Table 2. Au(I)-Catalyzed Oxacyclization/Cycloaddition Cascade with Vinyl Ethers



<sup>*a*</sup> ClAuP(*t*-Bu)<sub>2</sub>(*o*-biphenyl)/AgNTf<sub>2</sub> (3 mol %), [substrate] = 0.1 M, DCM, enol ethers (3 equiv). <sup>*b*</sup> Isolated yields were reported after purification from a silica gel column. <sup>*c*</sup> The configuration at the \*C carbon is responsible for the occurrence of two diastereomers. <sup>*d*</sup> Messy mixture was obtained for portion of **1e**.

substrates **1a** and **1d–1e** bearing a bulky  $R^1 (R^1 = n$ -Bu or *i*-Bu), the same reactions gave compounds **4a'** and **4d'–4e'** with excellent dr values (>20:1), due to a large steric interaction of the other isomers **4**. The structure of representative compound **4b'** is determined by an X-ray diffraction study.<sup>8</sup>

The scope of this new synthetic method is substantially expanded with its compatibility with substituted enol ethers including 1-ethoxypro-1-ene (2c, Z/E = 1.8, 3 equiv) and cyclic enol ethers 2d-2e, as illustrated in Table 3. These enol ethers proceeded with excellent diastereoselectivity that we did not obtain any epimers bearing mutual *trans*-R<sup>3</sup> and OR<sup>4</sup> substituents. For starting ketones 1a-1c and 1-ethoxypro-1-ene (entries 1-3), their cycloadducts 8a-8c were obtained as one single diastereomer with its structure carefully determined by <sup>1</sup>H-NOE, suggesting that *cis*-enol ether **2c** is more active than its *trans* isomer. Indeed, the reaction of 2c (Z/E = 0.5, 3 equiv) with ketone 1a gave cycloadduct 8a in diminished yield (31%) together with ketone-allene 3a (61%). This new tandem cascade also works with 3,4-dihydro-2H-pyran (2d) that reacts with ketones 1a-1f smoothly (entries 5-10), giving satisfactory yields (58-82%) of the expected cycloadducts 7a', 7d'-7f' or the deacylation products 9b-9c. 2,3-Dihydrofuran is also applicable to this catalysis, and it reacts with ketone 1b to deliver compound **10b** in 53% yield (dr = 6.5:1).

We also prepared new substrates 1g-1l to examine the effects of their phenyl substituents; their reactions with 3,4-dihydro-2*H*pyran are described in Table 4. Good yields (72–87%) were obtained for cycloadducts 7g'-7h' and 7k'-7l' bearing fluoro and methoxy substituents because these  $\pi$ -donor groups stabilize proposed benzopyriliums III (*vide infra*). Hypothetic [4+2]cycloadditions of free benzopyriliums<sup>3f-h</sup> and their metal-containing analogues<sup>3</sup> are restricted to unsubstituted benzenes and methoxy derivatives. The workability with substrates 1i and 1j bearing electron-deficient benzenes highlights the high reactivity of our new benzopyrilium (III). Table 3. Au(I)-Catalyzed Oxacyclization/Cycloaddition Cascade with Substituted Enol Ethers



entry	substrates <sup>a</sup>	alkene	t (step i)	products <sup>b</sup>
1	$\mathbf{R}^1 = n \text{-} \mathbf{B} \mathbf{u},  \mathbf{R}^2 = \mathbf{M} \mathbf{e}  (1 \mathbf{a})$	$R^3 = Me, R^4 = Et$ (2c, Z/E = 1.8)	4 h	<b>8a</b> $(76\%, dr > 20:1)^c$
2	$R^1 = R^2 = Me (\mathbf{1b})$	2c	4 h	<b>8b</b> $(82\%, dr = 7.3:1)^c$
3	$\mathbf{R}^1 = \mathbf{M}\mathbf{e},  \mathbf{R}^2 = n - \Pr(1\mathbf{c})$	2c	4 h	8c $(81\%, dr > 15:1)^c$
4	$\mathbf{R}^1 = n \text{-} \mathbf{B} \mathbf{u},  \mathbf{R}^2 = \mathbf{M} \mathbf{e}  (1 \mathbf{a})$	2c (Z/E = 0.5)	4 h	<b>8a</b> $(31\%, dr > 20:1)^{c,d}$
5	$\mathbf{R}^1 = n \cdot \mathbf{B}\mathbf{u},  \mathbf{R}^2 = \mathbf{M}\mathbf{e}  (1\mathbf{a})$	$R^3, R^4 = -(CH_2)_{3^-} (2d)$	1 h	<b>7a'</b> (82%, dr > 20:1)
6	$\mathbf{R}^1 = \mathbf{R}^2 = \mathbf{Me} \ (\mathbf{1b})$	2d	2 h	<b>9b</b> $(72\%, dr = 10:1)^c$
7	$\mathbf{R}^1 = \mathbf{M}\mathbf{e},  \mathbf{R}^2 = n - \Pr\left(1\mathbf{c}\right)$	2d	1 h	<b>9c</b> $(64\%, dr = 7.5:1)^c$
8	$\mathbf{R}^1 = n \text{-} \mathbf{B} \mathbf{u},  \mathbf{R}^2 = n \text{-} \mathbf{P} \mathbf{r}  \left( 1 \mathbf{d} \right)$	2d	2 h	<b>7d'</b> (75%, dr > 20:1)
9	$\mathbf{R}^1 = i\text{-}\mathbf{B}\mathbf{u},  \mathbf{R}^2 = \mathbf{M}\mathbf{e}  \left(1\mathbf{e}\right)$	2d	3 h	$7e' (63\%, dr > 20:1)^e$
10	$R^1 = Ph, R^2 = Me (1f)$	2d	5 h	<b>7f'</b> $(58\%)$ , dr > 20:1) <sup>e</sup>
11	$\mathbf{R}^1 = \mathbf{R}^2 = \mathrm{Me} \ (\mathbf{1b})$	$R^3$ , $R^4 = -(CH_2)_{2^-} (2e)$	2 h	<b>10b</b> $(53\%)$ , dr = $(6.5:1)^{c,e}$

<sup>*a*</sup> ClAuP(*t*-Bu)<sub>2</sub>(*o*-biphenyl)/AgNTf<sub>2</sub> (3 mol %), [substrate] = 0.1 M, DCM, enol ethers (3 equiv). <sup>*b*</sup> Isolated yields were given after purification from a silica gel column. <sup>*c*</sup> The configuration at the \*C carbon is responsible for the occurrence of two diastereomers. <sup>*d*</sup> Ketone-allene **3a** was obtained in 61%. <sup>*e*</sup> Messy mixture was obtained for portion of **1b**, **1e**, or **1f**.

 Table 4.
 Au(I)-Catalyzed Oxacyclization/Cycloaddition Cascade of Substituted Aromatic Substrates

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	$X \longrightarrow n-Bu + 0 \qquad \downarrow n-Bu + 0 \qquad \coprod n-Bu + 0 \qquad n-Bu +$	Au] X	Me 7'
entry	substrates <sup>a</sup>	<i>t</i> [h]	products (yield) <sup>b</sup>
1	$\mathbf{X} = \mathbf{H},  \mathbf{Y} = \mathbf{F}  (\mathbf{1g})$	6	<b>7g</b> ' (78%, dr >20:1)
2	$X = F, Y = H(1\mathbf{h})$	8	<b>7h</b> ′ (72%, dr >20:1)
3	X = H, Y = Cl (1i)	6	7i' (42%, dr >20:1)
4	X = Cl, Y = H(1j)	6	<b>7</b> j' (47%, dr >20:1)
5	X = H, Y = OMe (1k)	2	<b>7</b> k' (87%, dr >20:1)
6	X = OMe, Y = H (11)	2	<b>7l'</b> (85%, dr >20:1)

<sup>*a*</sup> ClAuP(*t*-Bu)<sub>2</sub>(*o*-biphenyl)/AgNTf<sub>2</sub> (3 mol %), [substrate] = 0.1 M, DCM, enol ethers (3 equiv.). <sup>*b*</sup> Isolated yields were given after purification from a silica gel column.

## Scheme 2



Structural analysis of resulting cycloadducts asserts the intermediacy of benzopyrilium (III), although gold- $\pi$ -allene species **3** has cationic character located mainly at the C(1)- and C(3)-carbons.<sup>5</sup> We hypothesize a fast equilibrium between ketone-allene (**3**) and intermediate (II), but the cycloadducts expected from species (II) fail to proceed. Here, we propose a concerted mechanism (i) for the [4+2]-cycloadditions of the new benzopyrilium (III). As shown in Scheme 2, *cis*-substituted enol ether approaches the pyrilium core of species III with its R<sup>3</sup> and OR<sup>4</sup> lying away from the bulky gold-containing substituent. We envisage that the allylic gold fragment of benzopyrilium III raises the HOMO energy level at the C(4)-carbon, facilitating this concerted process.<sup>2b,9</sup> The stepwise pathway (ii) involving cationic intermediates V and V' is opposed by our observation that no epimers resulted from the conformer V' which is actually favored by steric interactions of the *cis*-R<sup>3</sup> and OR<sup>4</sup> substituents of species V. Species (III) is more useful than reported benzopyriliums<sup>3</sup> in synthetic utility, because of its isolable and stereocontrolled [4+2]-cycloadducts.



Equation 1 shows the use of this catalysis for a stereoselective synthesis of a highly oxygenated molecule. A sequential treatment of compound **6c** with *m*-CPBA (1.5 equiv), followed by the DIBAL-H cleavage of resulting acetal, gave triol derivative **11** (65%) as a single diastereomer.

In summary, we report the first success on the Au-catalyzed tandem oxacyclization/[4+2]-cycloaddition cascade using ketoneallene substrates to give highly substituted oxacyclics with excellent stereocontrol. Control experiments reveal the involvement of benzopyrilium intermediates (III) that is active for [4+2]-cycloaddition reactions.<sup>9</sup> In contrast to oxo-alkyne substrates,<sup>3</sup> the resulting cycloadducts are isolable and efficiently produced from a reasonable scope of enol ethers. Efforts to realize the asymmetric version of this catalysis is under current investigation.

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**Supporting Information Available:** Experimental procedures, characterization data of new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

## References

 Reviews for gold-catalyzed annulation and cycloaddition reactions, see:(a) Patil, N. T.; Yamamoto, Y. Chem. Rev. 2008, 108, 3395. (b) Abu Sohel, S. Md.; Liu, R.-S. *Chem. Soc. Rev.* **2009**, *38*, 2269. (c) Shapiro, N. D.; Toste, F. D. *Synlett* **2010**, 675.

- Selected examples: (a) Shapiro, N. D.; Shi, Y.; Toste, F. D. J. Am. Chem. Soc. 2009, 131, 11654. (b) Shapiro, N. D.; Toste, F. D. J. Am. Chem. Soc. 2008, 130, 9244. (c) Li, G.; Huang, X.; Zhang, L. J. Am. Chem. Soc. 2008, 130, 6944. (d) Zhang, G.; Huang, X.; Li, G.; Zhang, L. J. Am. Chem. Soc. 2008, 130, 1814. (e) Liu, F.; Yu, Y.; Zhang, J. Angew. Chem., Int. Ed. 2009, 48, 5505.
- (3) [4+2]-Cycloadducts from benzopyrilium (I) are kinetically unstable and easily rearranged to naphthalene derivatives; only allylic alcohols<sup>3e</sup> allow the interception. See selected examples: (a) Asao, N. Synlett 2006, 1645.
  (b) Asao, N.; Kasahara, T.; Yamamoto, Y. Angew. Chem., Int. Ed. 2003, 42, 3504. (c) Asao, N.; Akiwa, H.; Yamamoto, Y. J. Am. Chem. Soc. 2004, 126, 7458. (d) Dyker, G.; Hildebrandt, D.; Liu, J.; Merz, K. Angew. Chem., Int. Ed. 2003, 42, 4399. (e) Hsu, Y.-C.; Ting, C.-M.; Liu, R.-S. J. Am. Chem. Soc. 2009, 131, 2090. (f) Barluenga, J.; Vázquez-Villa, H.; Ballesteros, A.; González, J. M. Adv. Synth. Catal. 2005, 347, 526. (g) Barluenga, J.; Vázquez-Villa, H.; Ballesteros, A.; González, J. M. Org. Lett. 2003, 5, 4121. (h) Hu, Z.-L.; Qian, W.-J.; Wang, S.; Wang, S.; Yao, Z.-J. Org. Lett. 2009, 11, 4676.
- (4) For [3+2]-cycloadducts, see selected examples: (a) Kusama, H.; Funami, H.; Shido, M.; Hara, Y.; Takaya, J.; Iwasawa, N. J. Am. Chem. Soc. 2005, 127, 2709. (b) Kusama, H.; Funami, H.; Takaya, J.; Iwasawa, N. Org. Lett. 2004, 6, 605. (c) Oh, C. H.; Lee, J. H.; Lee, S. J.; Kim, J. I.; Hong, C. S. Angew. Chem., Int. Ed. 2008, 47, 7505. (d) Oh, C. H.; Lee, S. M.; Hong, C. S. S. Org. Lett. 2010, 12, 1308.
- (5) For metal-allene bonding having the following dipolar character, see selected examples: (a) Kusama, H.; Ebisawa, M.; Funami, H.; Iwasawa, N. J. Am. Chem. Soc. 2009, 131, 16352. (b) Lee, J. H.; Toste, F. D. Angew. Chem., Int. Ed. 2007, 46, 912. (c) Lemiere, G.; Gandon, V.; Cariou, K.; Hours, A.; Fukuyama, T.; Dhimane, A.-L.; Fensterbank, L.; Malacria, M. J. Am. Chem. Soc. 2009, 131, 2993.

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- (6) (a) Marion, N.; Nolan, S. P. Angew. Chem., Int. Ed. 2007, 46, 2750. (b) Zhang, L. J. Am. Chem. Soc. 2005, 127, 16804. (c) Shi, F.-Q.; Li, X.; Xia, Y.; Zhang, L.; Yu, Z.-X. J. Am. Chem. Soc. 2007, 129, 15503. (d) Schwier, T.; Sromek, A. W.; Yap, D. M. L.; Chernyak, D.; Gevorgyan, V. J. Am. Chem. Soc. 2007, 129, 9868. (e) Cordonnier, M.-C.; Blanc, A.; Pale, P. Org. Lett. 2008, 10, 1569.
- (7) Previously, we attempted to synthesize allenyl aldehydes in pure form, but its rapid decomposition in solution hampers its purification and isolation. See: Teng, T.-M.; Lin, M.-S.; Vasu, D.; Bhunia, S.; Liu, T.-A.; Liu, R.-S. *Chem.-Eur. J.* 2010, *16*, 4744.



- (8) The X-ray crystallographic data of compound 4b' are provided in the Supporting Information.
- (9) (a) Bhunia, S.; Liu, R.-S. J. Am. Chem. Soc. 2008, 130, 16488. (b) Jiménez-Núñez, E.; Raducan, M.; Lauterbach, T.; Molawi, T.; Solorio, C. R.; Echavarren, A. M. Angew. Chem., Int. Ed. 2009, 48, 6152.

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