

# Synthesis and crystal structure of (4S,5R)-3,4-dimethyl-5-phenyl-2-(hydroxyethylimino)-1,3-thiazolidine

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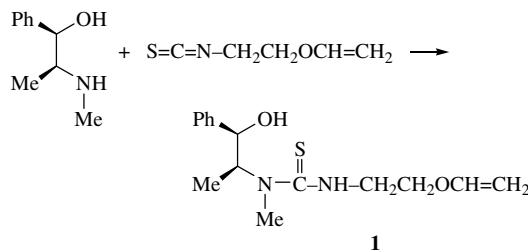
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The acid hydrolysis *l*-*N*-(*N'*-vinyloxyethylthiocarbamoyl)ephedrine has been investigated and the X-ray diffraction analysis of the synthesised (4S,5R)-3,4-dimethyl-5-phenyl-2-(hydroxyethylimino)-1,3-thiazolidine has been carried out.

Thiourea derivatives have antimicrobial, anti-inflammatory, anti-ulcerous and nematocide properties, growth regulatory activity and high insecticidal and acaricidal activities.<sup>1,2</sup> Interest in the derivatives of thioureas is caused by not only their biological activity but also the fact that they are convenient synthons in organic synthesis.<sup>3,4</sup>

We have synthesised *l*-*N*-(*N'*-vinyloxyethylthiocarbamoyl)ephedrine.<sup>5</sup> Thioamide **1** was obtained through interaction between *l*-ephedrine and vinyloxyethyl isothiocyanate in an alcoholic medium.<sup>†</sup>



There are no data on the hydrolysis of thiourea derivatives of ephedrine alkaloids in the literature. It is known that thioamides are susceptible to hydrolysis, especially in alkaline conditions; the full hydrolysis resulting in carboxylic acids, hydrogen sulfide and ammonia or amines is possible, but frequently under conditions of nitrile hydrolysis, heterogeneous ring compounds or products of oxidising splitting<sup>6</sup> are obtained.

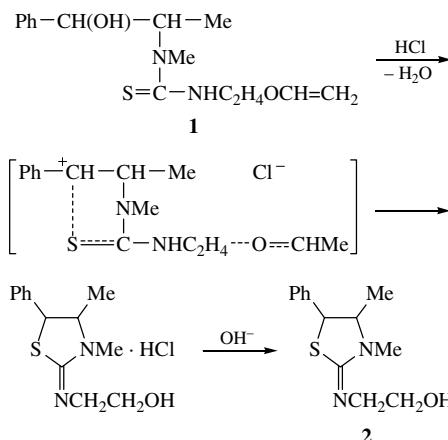
To study the reactivity of thioamide on the basis of *l*-ephedrine containing replaced oxyethyl group, we investigated the acid hydrolysis of *l*-*N*-(*N'*-vinyloxyethylthiocarbamoyl)ephedrine **1** in the presence of concentrated hydrochloric acid at ambient temperature.<sup>‡</sup>

To establish the dimensional structure of a molecule of (4S,5R)-3,4-dimethyl-5-phenyl-2-(hydroxyethylimino)-1,3-thiazolidine **2**, X-ray diffraction analysis was performed.<sup>§</sup> The structure of the molecule of **2** is presented in Figure 1.

The bond lengths and valence angles are close to the routine.<sup>7</sup> The thiazolidine ring in a molecule of **2** accepts the 4β-envelope

<sup>†</sup> The IR spectra were recorded on an AVATAR-320 instrument (in KBr tablets). The <sup>1</sup>H NMR spectra were measured on a Varian MERCURY-300 instrument at a frequency of 300 MHz in a CD<sub>3</sub>Cl solution with HMDS as the internal standard. The melting points were measured on a Boetius instrument.

<sup>‡</sup> *l*-*N*-(*N'*-Vinyloxyethylthiocarbamoyl)ephedrine **1**. 1.5 g (0.012 mol) of vinyloxyethyl isothiocyanate was added to 2 g (0.012 mol) of *l*-ephedrine in 5 ml of ethanol. The mixture was agitated at ambient temperature for 20–30 min; then, 1/3 of solvent was distilled off and allowed to stand overnight. The precipitate was filtered off and washed with diethyl ether to give 2.9 g (86%) compound **1**, mp 96–97 °C. IR ( $\nu/\text{cm}^{-1}$ ): 1530–1500 [NH–C(S)], 3400–3200 (OH). <sup>1</sup>H NMR,  $\delta$ : 0.86 (d, MeCH,  $J_{\text{HH}}$  8.4 Hz), 2.01 (s, MeN), 2.34–2.52 (m, CHMe), 4.43 (d, CH–O,  $J_{\text{HH}}$  10.6 Hz), 7.10–7.24 (m, Ph), 3.06 (t, NCH<sub>2</sub>), 3.40 (t, CH<sub>2</sub>O), 6.44 (q, CH=C,  $J_{\text{HH}}$  4.6 Hz), 4.01–4.07 (dd, C=CH<sub>2</sub>,  $J_{\text{HH}}^{\text{cis}}$  5.8 Hz,  $J_{\text{HH}}^{\text{trans}}$  11.8 Hz). Found (%): C, 61.27; H, 7.36. Calc. for C<sub>15</sub>H<sub>22</sub>O<sub>2</sub>N<sub>2</sub>S (%): C, 61.22; H, 7.48.

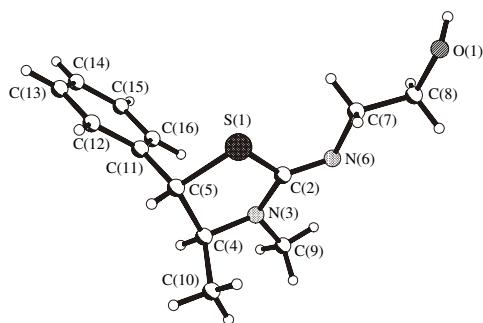


conformation, to some extent distorted from the theoretical ( $\Delta C_s^4 = 9.57^\circ$ ). The C(4) atom leaves the plane of other atoms of a cycle on  $\pm 0.49 \text{ \AA}$ ; the S(1), C(2), N(3) and C(5) atoms are coplanar to within  $\pm 0.05 \text{ \AA}$ . In a 4β-envelope conformation, the methyl group at the C(4) atom and the phenyl group at the C(5) atom of the group are oriented axially [torsion angles S(10)S(4)N(3)S(2)  $-91.72^\circ$  and C(11)C(5)C(2)N(3)  $88.54^\circ$ ]. The

<sup>§</sup> (4S,5R)-3,4-Dimethyl-5-phenyl-2-(hydroxyethylimino)-1,3-thiazolidine **2**. 10 ml of concentrated hydrochloric acid was added dropwise to 1.5 g (0.009 mol) of *l*-*N*-(*N'*-2-vinyloxyethylthiocarbamoyl)ephedrine at ambient temperature. The mixture was agitated for 3 h; then, a six-fold water volume was added, and the solvent was distilled off in a vacuum. A 40% aqueous solution of caustic soda was added to the residue, and the product was extracted with benzene. The organic layer was dried over Na<sub>2</sub>SO<sub>4</sub>, the solvent was distilled off to give 0.76 g (60%) of crystal product, mp 108–109 °C. IR ( $\nu/\text{cm}^{-1}$ ): 1680–1650 (C=N), 3500–3000 (OH). <sup>1</sup>H NMR,  $\delta$ : 0.92 (d, MeCH,  $J_{\text{HH}}$  8.6 Hz), 2.10 (s, MeN), 2.30–2.50 (m, CHMe), 4.93 (d, CHS,  $J_{\text{HH}}$  10.6 Hz), 7.00–7.15 (m, Ph), 3.14 (t, NCH<sub>2</sub>), 3.32 (t, CH<sub>2</sub>CH<sub>2</sub>). Found (%): C, 62.35; H, 7.12. Calc. for C<sub>13</sub>H<sub>18</sub>N<sub>2</sub>OS (%): C, 62.40; H, 7.20.

<sup>§</sup> Crystal data for **2** (C<sub>13</sub>H<sub>18</sub>N<sub>2</sub>OS) at 293 K: rhombic, space group P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub>,  $a = 7.0181(6)$ ,  $b = 10.891(1)$ ,  $c = 17.511(2) \text{ \AA}$ ,  $V = 1338.4(2) \text{ \AA}^3$ ,  $d_{\text{calc}} = 1.242 \text{ g cm}^{-3}$ ,  $Z = 4$ . A total of 1383 independent reflections ( $I > 2\sigma$ ) were measured using an automatic Bruker P4 four-circle diffractometer with a graphite monochromator using MoKα-radiation;  $\theta/2\theta$ -scanning,  $2\theta < 50^\circ$ . The structure was solved by a direct method and full-matrix least-squares technique in an anisotropic approximation for non-hydrogen atoms (isotropic for H atoms). Final *R* factor was 0.0391 ( $R_w = 0.1069$ ). Specification of the geometry was carried out by the SHELXL-97 program.

Atomic coordinates, bond lengths, bond angles and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre (CCDC). These data can be obtained free of charge via www.ccdc.cam.ac.uk/conts/retrieving.html (or from the CCDC, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336 033; or deposit@ccdc.cam.ac.uk). Any request to the CCDC for data should quote the full literature citation and CCDC reference number 287033. For details, see 'Notice to Authors', Mendeleev Commun., Issue 1, 2006.



**Figure 1** Crystal structure of compound 2.

methyl and hydroxyethylamine groups at the atoms N(3) and C(2) have equatorial orientations [C(5)C(4)N(3)C(9) –169.49°, C(4)N(3)C(2)N(6) 167.0°]. The 4 $\beta$ -envelope conformation was also observed in 2-*p*-bromophenyl-3,4-dimethyl-5-phenyl-1,3-oxazolidine.<sup>8</sup>

Due to the presence of substituents in the oxazolidine derivatives of *l*-ephedrine and *d*-pseudoephedrine at C(4), C(5), N(3), another favourable conformation of a cycle is the 3 $\alpha$ -envelope, in which the methyl group at N(3) has equatorial, and the other two substituents at the above atoms, pseudo-equatorial orientations. Exactly such a conformation is taken on by the majority of oxazolidine derivatives of pseudoephedrine, for example, (2*S*,4*S*,5*S*)-3,4-dimethyl-5-phenyl-2-phenylethenyl-1,3-oxazolidine.<sup>9</sup>

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