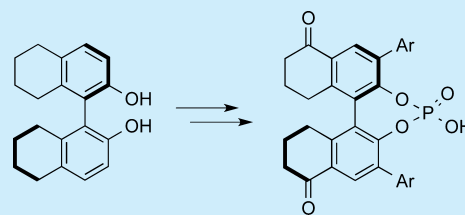


Synthesis and Evaluation of 5,5'-Bitetralone-Based Chiral Phosphoric Acids

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Supporting Information

ABSTRACT: A new type of phosphoric acid bearing a 5,5'-bitetralone scaffold was synthesized from BINOL and was shown to be a highly effective catalyst as illustrated in the asymmetric transfer hydrogenation of 2-phenylquinoline and the Friedel–Crafts reaction of 2,2,2-trifluoroacetophenone.



Chiral Brønsted acids have been shown to be highly effective catalysts for various organic transformations.¹ In particular, BINOL-based chiral phosphoric acids (**1** and **2**) (Figure 1) have attracted significant attention since Akiyama,

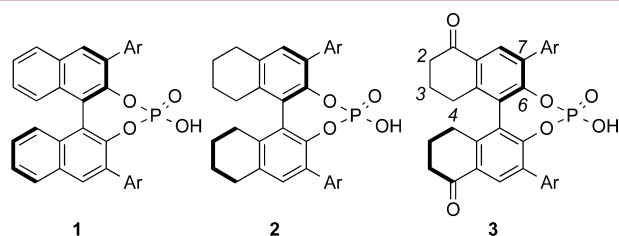
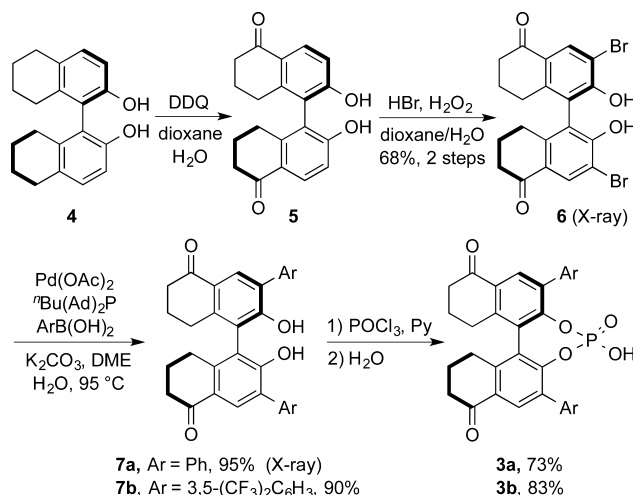


Figure 1. Chiral phosphoric acids.

Terada, and co-workers reported them a decade ago.^{2,3} A variety of asymmetric processes can be catalyzed by these phosphoric acids.^{4,5} In our previous studies on chiral acid-catalyzed asymmetric electrophilic addition reactions of olefins, promising results were also obtained with BINOL-based chiral phosphoric acids.⁶ The catalyst acidity was found to be a crucial factor for the reaction outcome. It was envisioned that a catalyst with enhanced acidity while retaining the BINOL skeleton would benefit the reaction. Along this line, phosphoric acids bearing a 5,5'-bitetralone scaffold (**3**) were of interest since the introduction of the carbonyl group would enhance the acidity while the BINOL chiral framework would still be maintained. Herein, we report our preliminary studies on this subject.

Three phosphoric acids with commonly used aryl substituents were prepared. The synthesis of 5,5'-bitetralone-based chiral phosphoric acids **3a** and **3b** are outlined in Scheme 1. The oxidation of (R)-H₈-BINOL (**4**)⁷ with DDQ in dioxane–H₂O at room temperature gave 5,5'-bitetralone-6,6'-diol (**5**),

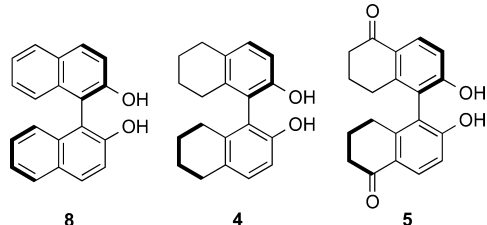
Scheme 1. Synthesis of Phosphoric Acids **3a**, **3b**

which was subsequently brominated with HBr and H₂O₂⁸ to afford compound **6** in 68% yield over two steps. The structure of **6** was confirmed by X-ray diffraction (see Supporting Information). The Ph and 3,5-(CF₃)₂-Ph groups were introduced via Suzuki coupling⁹ of dibromide **6** with the corresponding arylboronic acids to give compounds **7a** and **7b** in 95% and 90% yield, respectively (Scheme 1). Phosphoric acids **3a** and **3b** were obtained in 73% and 83% yield, respectively, by treating **7a** and **7b** with POCl₃ in pyridine, followed by the one-pot hydrolysis.¹⁰

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5,5'-Bitetralone-6,6'-diol (**5**), as a new chiral scaffold, could be potentially useful for other chiral ligands and catalysts. Its configurational stability under various conditions¹¹ was investigated along with BINOL (**8**) and H₈-BINOL (**4**). As shown in Table 1, compound **5** was configurationally stable up

Table 1. Studies on the Configurational Stability of Compounds 4, 5, and 8^a



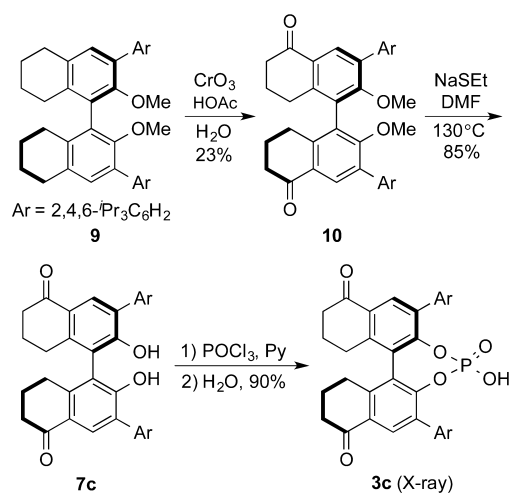
entry	conditions	ee (%) (8)	ee (%) (4)	ee (%) (5)
1	starting material	>99	>99	>99
2	NMP, 150 °C, 24 h	42	>99	>99
3	NMP, 180 °C, 24 h	1	99	99
4	NMP, 200 °C, 24 h	0	96	92
5	6 N HCl, 100 °C, 24 h	9	>99	>99
6	10% NaOH, 100 °C, 24 h	85	>99	>99

^aFor more details, see Supporting Information.

to 180 °C. The ee decreased to 92% at 200 °C for 24 h. No detectable racemization was observed when **5** was subjected to 6 N HCl in dioxane at 100 °C for 24 h or 10% NaOH in dioxane at 100 °C for 24 h. Comparable configurational stability was observed for H₈-BINOL (**4**). However, when **4** was subjected to the acidic and basic conditions, significant amounts of impurities were formed as judged by ¹H NMR and chiral HPLC analysis of the crude reaction mixture. BINOL (**8**) racemized readily at 150 °C and under the acidic conditions. The ee also dropped to 85% when **8** was treated with 10% NaOH at 100 °C for 24 h.¹¹

The synthesis of 2,4,6-*i*-Pr₃Ph substituted phosphoric acid **3c** is described in Scheme 2. Diketone **10** was obtained in 23% yield from known methyl ether **9**¹² via oxidation with CrO₃ in HOAc–H₂O.¹³ No desired product (**10**) was obtained when **9** was treated with DDQ. Compound **10** was demethylated with NaSEt in DMF at 130 °C to give compound **7c** in 85% yield.¹⁴

Scheme 2. Synthesis of Phosphoric Acid 3c



Treatment of **7c** with POCl₃ in pyridine and subsequent hydrolysis led to phosphoric acid **3c** in 90% yield.¹⁰

The X-ray structure of **3c** is shown in Figure 2. The torsion angle between the two phenyl planes of tetralone in **3c** is

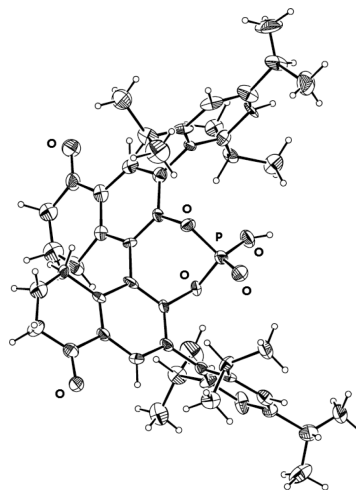


Figure 2. X-ray crystal structure of phosphoric acid **3c** (cocrystal CH₂Cl₂ was omitted for clarity).

62.00°, which is larger than that in BINOL-based **1c** (49.78°–51.29°)^{10,15} and H₈-BINOL-based **2c** (57.70°) (Figure 3), indicating that phosphoric acid **3c** may have a larger chiral cavity, which could be beneficial for certain transformations.

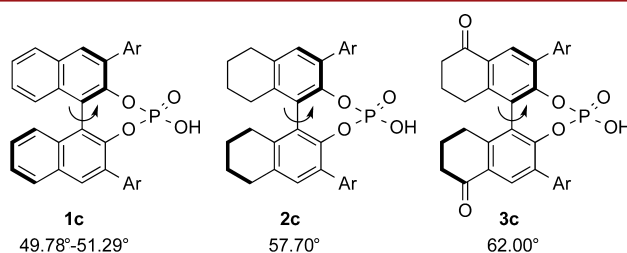
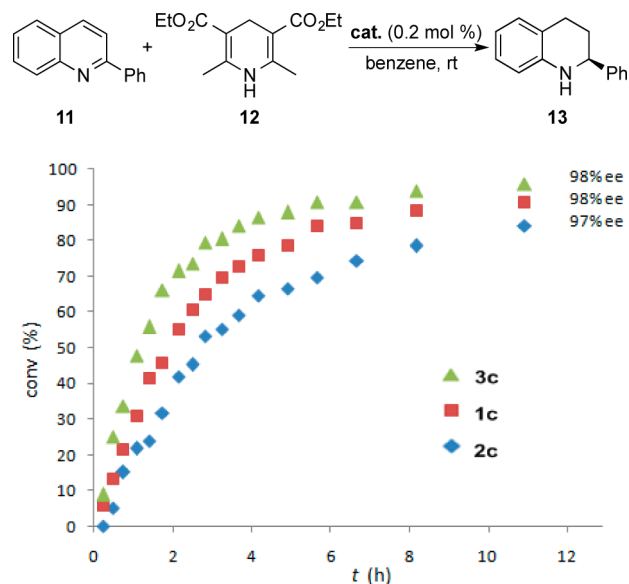


Figure 3. Torsion angles of phosphoric acids **1c**, **2c**, and **3c** (Ar = 2,4,6-*i*-Pr₃C₆H₂).

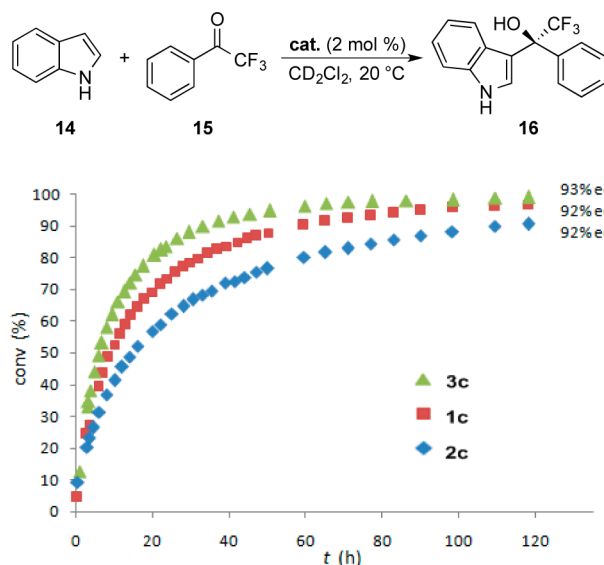
The catalytic properties of phosphoric acid **3c** was evaluated with previously reported asymmetric transfer hydrogenation of 2-phenylquinoline (**11**) with the Hantzsch dihydropyridines (**12**) (Scheme 3).^{5g,15a,16,17} The reaction was carried out in benzene at room temperature with 0.2 mol % phosphoric acids **1c**, **2c**, and **3c**. Catalyst **3c** was found to be slightly more active than **1c** and **2c** while the same high enantioselectivity was achieved. A similar behavior was observed for catalyst **3c** in the asymmetric Friedel–Crafts reaction^{18,19} of 2,2,2-trifluoroacetophenone (**15**) with indole (**14**) (Scheme 4). These results indicated that the introduction of the carbonyl group enhanced the activity of the phosphoric acid but did not deteriorate the enantioselectivity, as initially hoped. The potential of the newly synthesized phosphoric acids in asymmetric catalysis awaits further exploration.

In summary, we have developed a new type of phosphoric acid bearing a 5,5'-bitetralone scaffold from BINOL. As illustrated in asymmetric transfer hydrogenation of 2-phenylquinoline and the Friedel–Crafts reaction of 2,2,2-trifluoroacetophenone, the newly synthesized phosphoric acid catalyst displayed the same enantioselectivity as the corresponding

Scheme 3. Asymmetric Transfer Hydrogenation of 2-Phenylquinoline 11



Scheme 4. Asymmetric Friedel–Crafts Reaction of 2,2,2-Trifluoroacetophenone 15



BINOL and H₈-BINOL-based phosphoric acids, but with enhanced catalytic activity. The larger torsion angle and stronger acidity associated with 5,5'-bitetralone-based phosphoric acid could provide additional opportunities for asymmetric transformations. Furthermore, the 5,5'-bitetralone scaffold provides a novel chiral framework for the development of new chiral ligands and catalysts. All these studies are currently underway.

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.5b02400.

Experimental procedures, characterization data, X-ray structures, HPLC data for the determination of enantiomeric excess, and NMR spectra (PDF)

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Notes

The authors declare no competing financial interest.

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■ REFERENCES

- (1) For leading reviews on chiral Brønsted acid catalysis, see: (a) Schreiner, P. R. *Chem. Soc. Rev.* **2003**, 32, 289. (b) Doyle, A. G.; Jacobsen, E. N. *Chem. Rev.* **2007**, 107, 5713. (c) Yamamoto, H.; Payette, N. In *Hydrogen Bonding in Organic Synthesis*; Pihko, P. M., Ed.; Wiley-VCH Verlag GmbH & Co. KGaA: Weinheim, 2009; pp 73–140. (d) Kampen, D.; Reisinger, C. M.; List, B. *Top. Curr. Chem.* **2009**, 291, 395.
- (2) Akiyama, T.; Itoh, J.; Yokota, K.; Fuchibe, K. *Angew. Chem., Int. Ed.* **2004**, 43, 1566.
- (3) Uraguchi, D.; Terada, M. *J. Am. Chem. Soc.* **2004**, 126, 5356.
- (4) For leading reviews on phosphoric acid catalyzed reactions, see: (a) Akiyama, T. *Chem. Rev.* **2007**, 107, 5744. (b) Terada, M. *Chem. Commun.* **2008**, 4097. (c) Zamfir, A.; Schenker, S.; Freund, M.; Tsogoeva, S. B. *Org. Biomol. Chem.* **2010**, 8, 5262. (d) Yu, J.; Shi, F.; Gong, L.-Z. *Acc. Chem. Res.* **2011**, 44, 1156. (e) Rueping, M.; Kuenkel, A.; Atodiresei, I. *Chem. Soc. Rev.* **2011**, 40, 4539. (f) Čorić, I.; Vellalath, S.; Müller, S.; Cheng, X.; List, B. *Top. Organomet. Chem.* **2012**, 44, 165. (g) Parmar, D.; Sugiono, E.; Raja, S.; Rueping, M. *Chem. Rev.* **2014**, 114, 9047.
- (5) For leading references on phosphoric acids and phosphoramides, see: (a) Akiyama, T.; Saitoh, Y.; Morita, H.; Fuchibe, K. *Adv. Synth. Catal.* **2005**, 347, 1523. (b) Rowland, G. B.; Zhang, H.; Rowland, E. B.; Chennamadhavuni, S.; Wang, Y.; Antilla, J. C. *J. Am. Chem. Soc.* **2005**, 127, 15696. (c) Storer, R. I.; Carrera, D. E.; Ni, Y.; MacMillan, D. W. C. *J. Am. Chem. Soc.* **2006**, 128, 84. (d) Nakashima, D.; Yamamoto, H. *J. Am. Chem. Soc.* **2006**, 128, 9626. (e) Chen, X.-H.; Xu, X.-Y.; Liu, H.; Cun, L.-F.; Gong, L.-Z. *J. Am. Chem. Soc.* **2006**, 128, 14802. (f) Li, G.; Liang, Y.; Antilla, J. C. *J. Am. Chem. Soc.* **2007**, 129, 5830. (g) Guo, Q.-S.; Du, D.-M.; Xu, J. *Angew. Chem., Int. Ed.* **2008**, 47, 759. (h) Cheon, C. H.; Yamamoto, H. *J. Am. Chem. Soc.* **2008**, 130, 9246. (i) Vellalath, S.; Čorić, I.; List, B. *Angew. Chem., Int. Ed.* **2010**, 49, 9749. (j) Xu, F.; Huang, D.; Han, C.; Shen, W.; Lin, X.; Wang, Y. *J. Org. Chem.* **2010**, 75, 8677. (k) Čorić, I.; Müller, S.; List, B. *J. Am. Chem. Soc.* **2010**, 132, 17370. (l) Rueping, M.; Nachtsheim, B. J.; Koenigs, R. M.; Ieawsuwan, W. *Chem. - Eur. J.* **2010**, 16, 13116. (m) Shapiro, N. D.; Rauniyar, V.; Hamilton, G. L.; Wu, J.; Toste, F. D. *Nature* **2011**, 470, 245. (n) Momiyama, N.; Konno, T.; Furiya, Y.; Iwamoto, T.; Terada, M. *J. Am. Chem. Soc.* **2011**, 133, 19294. (o) Čorić, I.; List, B. *Nature* **2012**, 483, 315. (p) Stemper, J.; Isaac, K.; Duret, V.; Retailleau, P.; Voituriez, A.; Betzer, J.-F.; Marinetti, A. *Chem. Commun.* **2013**, 49, 6084. (q) Neel, A. J.; Hehn, J. P.; Tripet, P. F.; Toste, F. D. *J. Am. Chem. Soc.* **2013**, 135, 14044.
- (6) (a) Huang, D.; Wang, H.; Xue, F.; Guan, H.; Li, L.; Peng, X.; Shi, Y. *Org. Lett.* **2011**, 13, 6350. (b) Guan, H.; Wang, H.; Huang, D.; Shi, Y. *Tetrahedron* **2012**, 68, 2728. (c) Li, L.; Li, Z.; Huang, D.; Wang, H.; Shi, Y. *RSC Adv.* **2013**, 3, 4523.

- (7) Cram, D. J.; Helgeson, R. C.; Peacock, S. C.; Kaplan, L. J.; Domeier, L. A.; Moreau, P.; Koga, K.; Mayer, J. M.; Chao, Y.; Siegel, M. G.; Hoffman, D. H.; Sogah, G. D. Y. *J. Org. Chem.* **1978**, *43*, 1930.
- (8) Tillu, V. H.; Shinde, P. D.; Bedekar, A. V.; Wakharkar, R. D. *Synth. Commun.* **2003**, *33*, 1399.
- (9) Bartoszek, M.; Beller, M.; Deutsch, J.; Klawonn, M.; Köckritz, A.; Nemati, N.; Pews-Davtyan, A. *Tetrahedron* **2008**, *64*, 1316.
- (10) Klussmann, M.; Ratjen, L.; Hoffmann, S.; Wakchaure, V.; Goddard, R.; List, B. *Synlett* **2010**, 2189.
- (11) (a) Kyba, E. P.; Gokel, G. W.; de Jong, F.; Koga, K.; Sousa, L. R.; Siegel, M. G.; Kaplan, L.; Sogah, G. D. Y.; Cram, D. J. *J. Org. Chem.* **1977**, *42*, 4173. (b) Yudin, A. K.; Martyn, J. P.; Pandiaraju, S.; Zheng, J.; Lough, A. *Org. Lett.* **2000**, *2*, 41. (c) Chen, Y.; Yekta, S.; Yudin, A. K. *Chem. Rev.* **2003**, *103*, 3155.
- (12) (a) Romanov-Mikhailidis, F.; Guénée, L.; Alexakis, A. *Angew. Chem., Int. Ed.* **2013**, *52*, 9266. (b) Xie, W.; Jiang, G.; Liu, H.; Hu, J.; Pan, X.; Zhang, H.; Wan, X.; Lai, Y.; Ma, D. *Angew. Chem., Int. Ed.* **2013**, *52*, 12924.
- (13) Banerjee, A. K.; Pineda, J. J.; Mora, H. D.; Laya, M. S. *Synth. Commun.* **2007**, *37*, 3917.
- (14) Feutrill, G. I.; Mirrington, R. N. *Tetrahedron Lett.* **1970**, *11*, 1327.
- (15) For X-ray crystal structure of **1c**, see: (a) Rueping, M.; Theissmann, T. *Chem. Sci.* **2010**, *1*, 473. (b) Xie, Y.; Zhao, Y.; Qian, B.; Yang, L.; Xia, C.; Huang, H. *Angew. Chem., Int. Ed.* **2011**, *50*, 5682.
- (16) For leading references on asymmetric transfer hydrogenation of 2-phenylquinoline (**11**) with the Hantzsch dihydropyridines (**12**), see: (a) Rueping, M.; Antonchick, A. P.; Theissmann, T. *Angew. Chem., Int. Ed.* **2006**, *45*, 3683. (b) Kundu, D. S.; Schmidt, J.; Bleschke, C.; Thomas, A.; Blechert, S. *Angew. Chem., Int. Ed.* **2012**, *51*, 5456. (c) Tu, X.-F.; Gong, L.-Z. *Angew. Chem., Int. Ed.* **2012**, *51*, 11346.
- (17) For leading reviews on asymmetric transfer hydrogenation, see: (a) Ouellet, S. G.; Walji, A. M.; MacMillan, D. W. C. *Acc. Chem. Res.* **2007**, *40*, 1327. (b) Rueping, M.; Sugiono, E.; Schoepke, F. R. *Synlett* **2010**, 852. (c) de Vries, J. G.; Mršić, N. *Catal. Sci. Technol.* **2011**, *1*, 727. (d) Zheng, C.; You, S.-L. *Chem. Soc. Rev.* **2012**, *41*, 2498. (e) Zhu, C.; Saito, K.; Yamanaka, M.; Akiyama, T. *Acc. Chem. Res.* **2015**, *48*, 388.
- (18) For references on asymmetric Friedel–Crafts reaction of trifluoromethyl ketone with indole (**14**), see: (a) Nie, J.; Zhang, G.-W.; Wang, L.; Fu, A.; Zheng, Y.; Ma, J.-A. *Chem. Commun.* **2009**, 2356. (b) Nie, J.; Zhang, G.-W.; Wang, L.; Zheng, D.-H.; Zheng, Y.; Ma, J.-A. *Eur. J. Org. Chem.* **2009**, 2009, 3145.
- (19) For leading reviews on asymmetric Friedel–Crafts reactions, see: (a) Poulsen, T. B.; Jørgensen, K. A. *Chem. Rev.* **2008**, *108*, 2903. (b) You, S.-L.; Cai, Q.; Zeng, M. *Chem. Soc. Rev.* **2009**, *38*, 2190. (c) Zeng, M.; You, S.-L. *Synlett* **2010**, 1289.