Synthesis of 2-Hydroxymethyl-1*H*-imidazoles from 1,3-Dihydroimidazole-2-thiones

Yasser M. Loksha,^a Ahmed A. El-Barbary,^b Mahmoud A. El-Badawi,^b Claus Nielsen,^c Erik B. Pedersen*^a

^a Nucleic Acid Center, Department of Chemistry, University of Southern Denmark, Campusvej 55, 5230 Odense M, Denmark Fax +4566158780; E-mail: ebp@chem.sdu.dk

^b Department of Chemistry, Faculty of Science, Tanta University, Tanta, Egypt

^c Retrovirus Laboratory, Department of Virology, State Serum Institute, Artillerivej 5, 2300 Copenhagen, Denmark

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Abstract: 1,3-Dihydroimidazole-2-thiones **1a**–**d** were desulfurized by hydrogen peroxide in either acid or neutral medium to afford 1*H*imidazoles **3a**–**d**. Reaction of **3b**,**d** with dimethylsulfamoyl chloride furnished *N*,*N*-dimethylsulfamoylimidazoles **5** and **6a**,**b**. Lithiation of **5** and **6a**,**b** followed by formylation and reduction yielded the 2-hydroxymethyl derivatives **7** and **8a**,**b**. 4-Alkyl-5-cyclohexylmethyl-1*H*-imidazol-2-yl)methanols **9a**,**b** were obtained by hydrolysis of **7** and **8a**,**b**.

Key words: 1,3-dihydroimidazole-2-thiones, desulfurization, imidazole-2-sulfinic acids, 2-hydroxymetylimidazoles

Desulfurization of sulfur-containing compounds, including thioureas, has been achieved by using a number of different reagents. Raney nickel has being the most often employed;¹ other methods involved the use of nickel-sodium hydride complexes² and transition metal compounds,³ alkali bromates and iodates,⁴ nitric acid,⁵ potassium superoxide,⁶ oxygen and potassium tert-butoxide,⁷ ozone,⁸ 3,3-dimethyldioxirine,⁹ and a photochemical reaction involving singlet oxygen.^{10,11} Alkaline hydrogen peroxide has been used for the oxidation of pyrimidine-2thiones to the corresponding sulfinic acids. Subsequent refluxing of the isolated sulfinic acids with sulfuric acid afforded the desulfinated pyrimidines.¹² Alkaline hydrogen peroxide and sodium peroxide have also been used for the conversion of thioureas into ureas.¹³ In this work, hydrogen peroxide was used for the desulfurization reaction of 1,3-dihydroimidazole-2-thiones in acid and neutral medium followed by hydroxymethylation at the 2-position. 2-(Hydroxymethyl)imidazoles are interesting starting materials and have been used in a practical synthesis of immunosuppressant compounds.14

4-Benzyl-5-ethyl-1,3-dihydroimidazole-2-thione (**1a**) has been previously synthesized by Bullerwell and Lawson,¹⁵ 4-benzyl-5-isopropyl-1,3-dihydroimidazole-2-thione (**1c**)¹⁶ and 5-alkyl-4-cyclohexyl-1,3-dihydroimidazole-2thiones **1b**,d¹⁷ have been synthesized by Loksha et al.^{16,17} In the present investigation we found that treatment of 1,3-dihydroimidazole-2-thiones **1a–d** with hydrogen peroxide in an acidic medium afforded 1*H*-imidazole-2sulfinic acids **2a–d** as the major products and the desul-

SYNTHESIS 2004, No. 1, pp 0116–0120 Advanced online publication: 09.12.2003 DOI: 10.1055/s-2003-44370; Art ID: Z13503SS © Georg Thieme Verlag Stuttgart · New York furized imidazoles **3a-d** as the minor products. Treating compounds **1a-d** with hydrogen peroxide in a neutral medium in ethanol furnished the desulfurized imidazoles 3a**d** as the major products and the ethyl esters of 1*H*-imidazole-2-sulfinic acids 4a-d as the minor products. The imidazole-2-sulfinic acids 2a-d are somewhat stable in acidic medium and only small amounts were converted to the desulfinated imidazoles 3a-d through elimination of sulfur dioxide which is further oxidized to sulfuric acid by an excess of hydrogen peroxide. Under neutral reaction conditions in ethanol, small amounts of the initially formed sulfinic acids were esterified by the solvent to compounds 4a-d, whereas the major part of the sulfinic acids were converted to compounds 3a-d (Scheme 1). 5-Benzyl-4-ethyl-1H-imidazole (3a) has been previously synthesized by Bredereck et al.¹⁸ by an independent route through condensation of 4-benzyl-5-ethyloxazole with formamide.



Scheme 1

In order to activate the imidazole ring toward lithiation reactions, 4-alkyl-5-cyclohexylmethyl-1*H*-imidazoles **3b**,**d** were reacted with dimethylsulfamoyl chloride in toluene in the presence of triethylamine. Coupling at N-1 or N-3 occurred when R¹ was an ethyl group, while coupling occurred only at N-1 when R¹ was an isopropyl group due to steric hindrance of the isopropyl group. Each of the compounds **5** and **6a**,**b** were lithiated at the 2-postion of the imidazole ring by treatment with *n*-butyl lithium at 78 °C. Subsequent formylation using dimethylformamide afforded the 2-formyl derivative which without isolation was immediately reduced by sodium borohydride to furnish 4-cyclohexylmethyl-5-ethyl-2-hydroxymethyl-imidazole-1-sulfonic acid dimethylamide (**7**) and 4-alkyl-5cyclohexylmethyl-2-hydroxymethylimidazole-1-sulfonic acid dimethylamides 8a,b. For the compounds 5 and 6 and for 7 and 8, respectively, the structural assignment was also based on R_f values from TLC on silica. Besides the above-mentioned influence of sterical hindrance on the formation of **6b**, we have previously observed smaller R_f values for 4,5-disubstituted imidazoles when the more lipophilic substituent was next to an N1 substituent and larger R_f values for the compounds with the interchanged 4- and 5-substituents.¹⁷ 5-(Cyclohexylmethyl-4-ethyl-1*H*imidazol-2-yl)methanol (9a) was obtained by either hydrolysis of 4-cyclohexylmethyl-5-ethyl-2-hydroxymethylimidazole-1-sulfonic acid dimethylamide (7) or 5cyclohexylmethyl-4-ethyl-2-hydroxymethylimidazole-1sulfonic acid dimethylamide (8a) in an acidic medium (1.5 M hydrochloric acid), while acid hydrolysis of compound **8b** afforded (5-cyclohexylmethyl-4-isopropyl-1*H*imidazol-2-yl)methanol (9b) (Scheme 2).



Scheme 2

Compounds **9a,b** are believed to exist in a rapid equilibrium between the two tautomeric forms **i** and **ii** which explains why C-4 and C-5 could not be detected in 13 C NMR.

NMR spectra were recorded on a Varian Gemini 2000 NMR spectrometer at 300 MHz for ¹H and 75 MHz for ¹³C with TMS as the internal standard. Cy denotes cyclohexyl. EIMS were recorded on a Finnigan Mat SSQ 701 spectrometer, MALDI spectra were recorded on a Fourier Transform Ion Cyclotron Resonance Mass Spectrometer. Melting points were determined on a Büchi melting point apparatus. Elementary analyses were performed at H.C. Ørsted Institute, University of Copenhagen. Silica gel (0.040–0.063 mm) used for column chromatography and analytical silica gel TLC plates 60 F_{254} were purchased from Merck. Solvents for column chromatography were distilled prior to use.

1*H*-Imidazole-2-sulfinic Acids 2a–d and 1*H*-Imidazoles 3a–d, Method A; General Procedure

To a solution of each of the compounds 1a-d (5 mmol) in HOAc (20 mL), 35% H₂O₂ (1.5 mL, 15 mmol) was added in one portion. The reaction mixture was stirred at r.t. for 2 h. The solid product formed was filtered off, and dried in vacuo affording the imidazole-2-sulfinic acids 2a-d. The filtrate was evaporated in vacuo till dryness, H₂O (20 mL) was added to the residue, the solution was neutralized with a 10% aq solution of NaHCO₃, and extracted with EtOAc (3 × 20 mL). The EtOAc extracts were collected, dried (Na₂SO₄), filtered, and evaporated under reduced pressure. The residual material was dried to afford compounds 3a-d.

5-Benzyl-4-ethyl-1*H*-imidazole-2-sulfinic Acid (2a)

Yield: 0.6 g (48%); mp 276–278 °C.

¹H NMR (300 MHz, DMSO- d_{δ}): $\delta = 1.08$ (t, 3 H, J = 7.3 Hz, CH₃CH₂), 2.61 (q, 2 H, J = 7.3 Hz, CH₃CH₂), 3.96 (s, 2 H, CH₂Ph), 7.22–7.33 (m, 5 H, Ph), 14.58 (br s, 1 H, OH).

¹³C NMR (300 MHz, DMSO- d_6): δ = 13.66 (CH_3CH_2), 16.45 (CH_3CH_2), 28.66 (CH_2Ph), 126.84 (C-4), 126.59, 128.17, 128.57, 137.88 (C_{arom}), 130.72 (C-5), 145.95 (C-2).

MALDI MS: m/z = 273 (100%, M + Na⁺)

Anal. Calcd for $C_{12}H_{14}N_2O_2S\cdot 0.25H_2SO_4$ (274.84): C, 52.44; H, 5.32; N, 10.19. Found: C, 52.66; H, 5.08; N, 10.25.

5-Cyclohexylmethyl-4-ethyl-1*H***-imidazole-2-sulfinic Acid (2b)** Yield: 0.72 g (56%); mp 290–292 °C.

¹H NMR (DMSO-*d*₆): δ = 0.86–0.98 (m, 2 H, H_{cy}), 1.13–1.50 (m, 6 H, C*H*₃CH₂ and H_{cy}), 1.55–1.65 (m, 6 H, H_{cy}), 2.45 (d, 2 H, *J* = 5.8 Hz, C*H*₂Cy), 2.56 (q, 2 H, *J* = 7.3 Hz, CH₃CH₂), 14.38 (br s, 1 H, OH).

¹³C NMR (DMSO- d_{δ}): δ = 13.79 (CH₃CH₂), 16.48 (CH₃CH₂), 25.47, 25.72, 32.04, 37.39 (C_{cy}), 30.17 (CH₂Cy), 126.68 (C-4), 130.52 (C-5), 145.68 (C-2).

Anal. Calcd for $C_{12}H_{20}N_2O_2S\cdot 0.25H_2SO_4$ (280.89): C, 51.31; H, 7.36; N, 9.97. Found: C, 51.43; H, 7.14; N, 9.92.

5-Benzyl-4-isopropyl-1*H***-imidazole-2-sulfinic Acid (2c)** Yield: 0.81 g (61%); mp 285–278 °C.

¹H NMR (DMSO- d_6): $\delta = 1.18$ [d, 6 H, J = 6.5 (CH₃)₂CH], 3.16 [hept, 1 H, J = 6.5 Hz, (CH₃)CH], 3.98 (s, 2 H, CH₂Ph), 7.20–7.32 (m, 5 H, Ph), 14.56 (br s, 1 H, OH).

¹³C NMR (DMSO- d_6): δ = 21.28 [(CH₃)₂CH], 23.69 [(CH₃)₂CH], 28.68 (CH₂Ph), 125.75 (C-4), 126.58, 128.07, 128.58, 138.03 (C_{arom}), 134.78 (C-5), 146.40 (C-2).

Anal. Calcd for $C_{13}H_{16}N_2O_2S{\cdot}0.25H_2SO_4$ (288.87): C, 54.05; H, 5.76; N, 9.70. Found: C, 54.20, H, 5.60; N, 9.73.

5-Cyclohexylmethyl-4-isopropyl-1*H*-imidazole-2-sulfinic Acid (2d)

Yield: 0.88 g (65%); mp 295–297 °C.

¹H NMR (DMSO-*d*₆): $\delta = 0.85-0.99$ (m, 2 H, H_{cy}), 1.11–1.31 [m, 9 H, (C*H*₃)₂CH and H_{cy}], 1.541.68 (m, 6 H, H_{cy}), 2.47 (d, 2H, *J* = 6.3 Hz, C*H*₂Cy), 3.05 [hept, 1 H, *J* = 6.9 Hz, (CH₃)₂CH].

¹³C NMR (DMSO-*d*₆): δ = 21.56 [(*C*H₃)₂CH], 23.73 [(CH₃)₂CH], 25.48, 25.73, 32.05, 37.51 (C_{cy}), 30.30 (*C*H₂Cy), 125.64 (C-4), 134.72 (C-5), 146.20 (C-2).

Anal. Calcd for $C_{13}H_{22}N_2O_2S \cdot 0.45H_2SO_4$ (314.53): C, 49.64; H, 7.34; N, 8.91. Found: C, 49.46; H, 7.06; N, 8.79.

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Yield: 12% (method A), 48% (method B); mp 106–108 °C (Lit.¹⁸ mp 106 °C).

¹H NMR (DMSO-*d*₆): δ = 1.07 (t, 3 H, *J* = 7.5 Hz, C*H*₃CH₂), 2.49 (q, 2 H, *J* = 7.5 Hz, CH₃CH₂), 3.79 (s, 2 H, C*H*₂Ph), 7.11–7.27 (m, 5 H, Ph), 7.40 (s, 1 H, H-2).

¹³C NMR (DMSO-*d₆*): δ = 14.59 (*C*H₃CH₂), 18.01 (CH₃*C*H₂), 31.35 (*C*H₂Ph), 125.56, 128.06, 128.14, 132.99 (C_{arom}), 141.02 (C-2).

EIMS: m/z (%) = 186 (100, M⁺).

5-Cyclohexylmethyl-4-ethyl-1*H*-imidazole (3b)

Yield: 15% (method A), 52% (method B); mp 122-124 °C.

¹H NMR (DMSO-*d*₆): $\delta = 0.82-0.93$ (m, 2 H, H_{cy}), 1.07–1.18 (m, 6 H, CH₃CH₂ and H_{cy}), 1.44–1.61 (m, 6 H, H_{cy}), 2.30 (d, 2 H, *J* = 7.0 Hz, CH₂Cy), 2.42 (q, 2 H, *J* = 7.5 Hz, CH₃CH₂), 7.39 (s, 1 H, H-2). ¹³C NMR (DMSO-*d*₆): $\delta = 14.69$ (CH₃CH₂), 18.30 (CH₃CH₂), 25.68 – 26.01 – 22.54 (CH CH₃ cr d₂), 127.82 (C 4)

25.68, 26.01, 38.08 (C_{cy}), 32.54 (CH₂Cy and C_{cy}), 127.83 (C-4), 132.54 (C-2).

EIMS: m/z (%) = 192 (32, M⁺), 109 (100).

Anal. Calcd for $\rm C_{12}H_{20}N_2 \cdot 0.4H_2O$ (199.51): C, 72.24; H, 10.51; N, 14.04. Found: C, 72.17; H, 10.06; N, 14.30

5-Benzyl-4-isopropyl-1*H*-imidazole (3c)

Yield: 12% (method A), 50% (method B); mp 152–154 °C.

¹H NMR (DMSO-*d*₆): δ = 1.12 [d, 6 H, J = 7.0 Hz, (CH₃)₂CH], 2.97 [hept, 1 H, J = 7.0 Hz, (CH₃)₂CH], 3.81 [s, 2 H, CH₂Ph], 7.11–7.27 (m, 5 H, Ph), 7.42 (s, 1 H, H-2).

¹³C NMR (DMSO-*d*): δ = 22.88 [(*C*H₃)₂CH], 24.44 [(CH₃)₂CH], 31.34 (*C*H₂Ph), 125.55, 128.05, 128.11, 135.61 (C_{arom}), 128.24 (C-4), 132.98 (C-2), 141.05 (C-5).

EIMS: m/z (%) = 200 (76, M⁺), 185 (100).

5-Cyclohexylmethyl-4-isopropyl-1*H*-imidazole (3d)

Yield: 20% (method A), 68% (method B); mp 120-122 °C.

¹H NMR (DMSO-*d*₆): δ = 0.82–0.93 (m, 2 H, H_{cy}), 1.12–1.17 [m, 8 H, (CH₃)₂CH and H_{cy}], 1.43–1.61 (m, 6 H, H_{cy}), 2.30 (d, 2 H, *J* = 7.2 Hz, CH₂Cy), 2.85 [hept, 1 H, *J* = 6.9 Hz, (CH₃)₂CH], 7.35 (s, 1 H, H-2).

¹³C NMR (DMSO-*d*₆): δ = 23.11 [(*C*H₃)₂CH], 24.57 [(CH₃)₂CH], 25.69, 26.02, 32.57, 38.07 (C_{cy}), 32.49 (*C*H₂Cy), 132.53 (C-2).

EIMS: m/z (%) = 206 (58, M⁺), 123 (100).

Anal. Calcd for $C_{13}H_{22}N_2 \cdot 0.25H_2O$ (210.84): C, 74.06; H, 10.76; N, 13.29. Found: C, 74.09; H, 10.55; N, 13.54.

1*H*-Imidazoles 3a–d and 1*H*-Imidazoles-2-Sulfinic Acids Ethyl Ester 4a–d, Method B; General Procedure

To a solution of each of compounds 1a-d (5 mmol) in EtOH (20 mL) was added 35% H₂O₂ (1.5 mL, 15 mmol) in one portion. The mixture was stirred at r.t. for 2 h. The solvent was removed under reduced pressure and H₂O (20 mL) was added to the residual oil. The reaction mixture was extracted with Et₂O (3 × 20 mL). The combined ether extracts were dried (Na₂SO₄), filtered, and evaporated in vacuo to afford the ethyl ester of imidazole-2-sulfinic acids **4a**–**d**. The H₂O layer was neutralized with10% aq NaHCO₃, extracted with EtOAc (3 × 20 mL), the EtOAc extracts were collected, dried (Na₂SO₄), filtered, and evaporated under reduced pressure. The residual material was dried to afford compounds **3a–d**. The yields of **3a–d** are given above under the products of Method A.

5-Benzyl-4-ethyl-1*H***-imidazole-2-sulfinic Acid Ethyl Ester (4a)** Yield: 0.3 g (22%); oil.

¹H NMR (DMSO-*d₆*): δ = 1.07 (t, 3 H, *J* = 7.4 Hz, C*H*₃CH₂), 1.18 (t, 3 H, *J* = 7.1 Hz, C*H*₃CH₂), 2.54 (q, 2 H, *J* = 7.4 Hz, CH₃C*H*₂), 3.78 (dq, 1 H, *J* = 10.2, 7.1 Hz, OCHHCH₃), 3.88 (s, 2 H, C*H*₂Ph), 4.10 (dq, 1 H, *J* = 10.1, 7.1 Hz, OCHHCH₃), 7.14–7.29 (m, 5 H, Ph), 13.00 (br s, 1 H, NH).

¹³C NMR (DMSO- d_6): $\delta = 14.20$ (CH₃CH₂), 15.21 (CH₃CH₂), 17.94 (br, CH₃CH₂), 31.06 (br, CH₂Ph), 61.31 (OCH₂CH₃), 125.84, 128.14, 128.19, 140.02 (C_{arom}), 144.61 (C-2).

EIMS: *m*/*z* (%) = 278 (38, M⁺), 233 (100).

5-Cyclohexylmethyl-4-ethyl-1*H*-imidazole-2-sulfinic Acid Ethyl Ester (4b)

Yield: 0.23 g (16%); mp 98–100 °C.

¹H NMR (CDCl₃): $\delta = 0.88-0.98$ (m, 2 H, H_{cy}), 1.11–1.32 (m, 9 H, CH₃CH₂, CH₃CH₂O and H_{cy}), 1.55–1.72 (m, 6 H, H_{cy}), 2.44 (d, 2 H, *J* = 6.8 Hz, CH₂Cy), 2.59 (q, 2 H, *J* = 7.4 Hz, CH₃CH₂) 3.79 (dq, 1 H, *J* = 9.8, 7.1 Hz, OCHHCH₃), 4.16 (dq, 1 H, *J* = 9.8, 7.1 Hz, OCHHCH₃), 10.70 (br s, 1 H, NH).

 ^{13}C NMR (CDCl₃): δ = 14.26 (CH₃CH₂), 15.32 (CH₃CH₂), 19.04 (br, CH₃CH₂), 26.12, 26.35, 38.49 (C_{cy}), 33.06 (CH₂Cy and C_{cy}), 61.97 (OCH₂CH₃), 143.70 (C-2).

EIMS: *m*/*z* (%) = 284 (30, M⁺), 55 (100).

Anal. Calcd for $C_{14}H_{24}N_2O_2S\cdot 0.6H_2O$ (295.23): C, 56.96; H, 8.60; N, 9.49. Found: C, 56.73; H, 8.19; N, 9.66.

5-Benzyl-4-isopropyl-1*H*-imidazole-2-sulfinic Acid Ethyl Ester (4c)

Yield: 0.15 g (10%); oil.

¹H NMR (DMSO- d_6): $\delta = 1.12-1.20$ [m, 9 H, CH₃CH₂ and (CH₃)₂CH], 2.20-3.00 [m, 1 H, (CH₃)₂CH], 3.80 (dq, 1 H, J = 10.3, 7.1 Hz, OCHHCH₃), 3.90 (s, 2 H, CH₂Ph), 4.09 (dq, 1 H, J = 10.3, 7.1 Hz, OCHHCH₃), 7.16-7.29 (m, 5 H, Ph), 12.97 (br s, 1 H, NH).

¹³C NMR (DMSO- d_{δ}): δ = 15.20 (CH₃CH₂), 22.36 [(CH₃)₂CH], 24.83 [(CH₃)₂CH], 30.59 (CH₂Ph), 61.36 (OCH₂CH₃), 125.83, 128.08, 128.23 (C_{arom}), 144.81 (C-2).

HRMS MALDI: m/z calcd for $C_{15}H_{21}NaN_2O_2S$ (MH+), 293.1323; found, 293.1317

5-Cyclohexylmethyl-4-isopropyl-1*H*-imidazole-2-sulfinic Acid Ethyl Ester (4d)

Yield: 0.2 g (13%); semisolid.

¹H NMR (DMSO-*d*₆): $\delta = 0.88-0.96$ (m, 2 H, H_{cy}), 1.06–1.19 [m, 9 H, (CH₃)₂CH and H_{cy}], 1.55–1.66 (m, 6 H, H_{cy}), 2.39 (d, 2 H, 6.8 Hz, CH₂Cy), 2.92 [hept, 1 H, 6.9 Hz, (CH₃)₂CH], 3.79 (dq, 1 H, *J* = 10.3, 7.1 Hz, OCHHCH₃), 4.08 (dq, 1 H, *J* = 10.3, 7.1 Hz, OCHHCH₃).

¹³C NMR (DMSO- d_6): δ = 15.15 (CH₃CH₂), 22.56 [(CH₃)₂CH], 24.85 [(CH₃)₂CH], 25.62, 25.92, 32.40, 37.92 (C_{cy}), 32.37 (CH₂Cy), 61.23 (OCH₂CH₃).

HRMS MALDI: m/z calcd for $C_{15}H_{27}N_2O_2S$ (MH⁺), 299.1793; found, 299.1792.

4-Cyclohexylmethyl-5-ethylimidazole-1-sulfonic Acid Dimethylamide (5) and 4-Alkyl-5-cyclohexylmethyl-imidazole-1-sulfonic Acids Dimethylamide 6a,b

To a mixture of each of the compounds **3b,d** (10 mmol) and Et_3N (1.4 mL, 10 mmol) in toluene (40 mL), dimethylsulfamoyl chloride (1 mL, 10 mmol) in toluene (10 mL) was added dropwise. The mixture was stirred at r.t. for 1 h followed by refluxing for 5 h. After cooling to r.t., H₂O (50 mL) was added to the reaction mixture and

the two layers were separated. The organic layer was dried (Na₂SO₄), filtered, and the solvent was removed under reduced pressure. The residual material was chromatographed on a silica gel column with CH_2Cl_2 -EtOAc (1:1) to afford compounds **5** and **6a,b**.

4-Cyclohexylmethyl-5-ethyl-imidazole-1-sulfonic Acid Dimethylamide (5)

Yield: 1 g (33%); mp 52–54 °C.

¹H NMR (CDCl₃): δ (ppm) = 0.88–1.00 (m, 2 H, H_{cy}), 1.08–1.31 (m, 6 H, CH₃CH₂ and H_{cy}), 1.67 –1.78 (m, 6 H, H_{cy}), 2.34 (d, 2 H, J = 6.7 Hz, CH₂Cy), 2.72 (q, 2 H, J = 7.5 Hz, CH₃CH₂), 2.87 [s, 6 H, (CH₃)₂N], 7.00 (s, 1 H, NH).

¹³C NMR (CDCl₃): δ = 14.95 (*C*H₃CH₂), 17.10 (CH₃CH₂), 26.22, 26.49, 33.28, 37.92 (C_{cy}), 34.60 (*C*H₂Cy), 37.64 [(CH₃)₂N], 129.05 (C-4), 136.65 (C-2), 139.47 (C-5).

EIMS: m/z (%) = 299 (13, M⁺), 109 (100).

Anal. Calcd for $C_{14}H_{25}N_3O_2S \cdot 0.2H_2O$ (303.04): C, 55.49; H, 8.45; N, 13.87. Found: C, 55.66; H, 8.29; N, 13.81.

5-Cyclohexylmethyl-4-ethyl-imidazole-1-sulfonic Acid Dimethylamide (6a)

Yield: 0.75 g (25%); mp 96-98 °C.

¹H NMR (CDCl₃): $\delta = 0.88-0.99$ (m, 2 H, H_{cy}), 1.11–1.26 (m, 6 H, CH₃CH₂ and H_{cy}), 1.58–1.70 (m, 6 H, H_{cy}), 2.48 (q, 2 H, J = 7.5 Hz, CH₃CH₂), 2.55 (d, 2 H, J = 6.6 Hz, CH₂Cy), 2.84 [s, 6 H, (CH₃)₂N], 7.03 (s, 1 H, H-2).

¹³C NMR (CDCl₃): δ = 13.44 (*C*H₃CH₂), 20.41 (*C*H₃CH₂), 26.21, 26.36, 33.07, 37.94 (C_{cy}), 31.36 (*C*H₂Cy), 38.47 [(*C*H₃)₂N], 125.21 (C-4), 137.07 (C-2), 142.95 (C-5).

EIMS: m/z (%) = 299 (100, M⁺).

Anal. Calcd for $C_{14}H_{25}N_3O_2S$ (299.43): C, 56.16; H, 8.42; N, 14.03. Found: C, 56.31; H, 8.43; N, 13.99.

5-Cyclohexylmethyl-4-isopropylimidazole-1-sulfonic Acid Dimethylamide (6b)

Yield: 1.72 g (55%); mp 122–124 °C.

¹H NMR (CDCl₃): $\delta = 0.88-1.02$ (m, 2 H, H_{cy}), 1.07–1.28 [m, 9 H, (CH₃)₂CH and H_{cy}], 1.59–1.82 (m, 6 H, H_{cy}), 2.56 (d, 2 H, *J* = 7.0 Hz, CH₂Cy), 2.82–2.90 [m, 7 H, (CH₃)₂CH and (CH₃)₂N], 7.84 (s, 1 H, H-2).

¹³C NMR (CDCl₃): δ = 22.41 [(*C*H₃)₂CH], 25.87 [(CH₃)₂CH], 26.26, 26.36, 33.07, 37.96 (C_{cy}), 31.56 (*C*H₂Cy), 38.36 [(CH₃)₂N], 124.17 (C-4), 137.31 (C-2), 146.90 (C-5).

EIMS: m/z (%) = 313 (100, M⁺).

Anal. Calcd for $C_{15}H_{27}N_3O_2S$ (313.45): C, 57.48; H, 8.68; N, 13.41. Found: C, 57.72; H, 8.67; N, 12.97.

4-Cyclohexylmethyl-5-ethyl-2-hydroxymethylimida-zole-1-sulfonic Acid Dimethylamide (7) and 4-Alkyl-5-cyclohexylmethyl-2-hydroxymethylimidazole-1-sulfonic Acids Dimethylamide 8a,b

Each of the compounds **5** and **6a,b** (1.5 mmol) was dissolved in THF (20 mL) under nitrogen and the solution was cooled to -78 °C. *n*-BuLi (0.81 mL, 2.2 M in *n*-hexane, 1.8 mmol) was added and the mixture was stirred for 0.5 h at -78 °C. Then DMF (0.14 mL, 1.8 mmol) was added dropwise at -78 °C. The reaction mixture was allowed to reach r.t. and quenched with H₂O (5 mL). EtOH (10 mL) was added to the reaction mixture followed by addition of NaBH₄ (0.057g, 1.5 mmol) portion-wise. After filtration, Et₂O (30 mL) was added to the filtrate, and the two layers were separated. The organic layer was dried (Na₂SO₄), filtered, and the solvent was removed under reduced pressure. The residual material was chromatographed

on a silica gel column with CH_2Cl_2 -MeOH (1:1) to afford compounds 7 and 8a,b.

4-Cyclohexylmethyl-5-ethyl-2-hydroxymethylimida-zole-1-sulfonic Acid Dimethylamide (7)

Yield: 0.15 g (30%); 100-102 °C.

¹H NMR (CDCl₃): $\delta = 0.85-0.92$ (m, 2 H, H_{cy}), 1.22–1.25 (m, 6 H, CH₃CH₂ and H_{cy}), 1.66–1.71 (m, 6 H, H_{cy}), 2.30 (d, 2 H, J = 6.8 Hz, CH₂Cy), 2.70 (q, 2 H, J = 7.4 Hz, CH₃CH₂), 3.58 (br s, 1 H, OH), 4.79 (s, 2 H, CH₂OH).

¹³C NMR (CDCl₃): δ = 15.04 (*C*H₃CH₂), 17.62 (CH₃CH₂), 26.21, 26.48, 33.28, 37.82 (C_{ey}), 34.39 (*C*H₂Cy), 37.45 [(CH₃)₂N], 59.33 (CH₂OH), 130.97 (C-4), 137.11 (C-2), 148.39 (C-5).

EIMS: m/z (%) = 329 (19, M⁺), 139 (100).

Anal. Calcd for $C_{15}H_{27}N_3O_3S$ (329.46): C, 54.68; H, 8.26; N, 12.75. Found: C, 55.12; H, 8.31; N, 12.35.

5-Cyclohexylmethyl-4-ethyl-2-hydroxymethylimida-zole-1-sulfonic Acid Dimethylamide (8a)

Yield: 0.26 g (52%); mp 84–86 °C.

¹H NMR (CDCl₃): $\delta = 0.83-0.97$ (m, 2 H, H_{cy}), 1.10–1.26 (m, 6 H, CH₃CH₂ and H_{cy}), 1.56–1.70 (m, 6 H, H_{cy}), 2.45 (q, 2 H, *J* = 7.5 Hz, CH₃CH₂), 2.54 (d, 2 H, *J* = 6.9 Hz, CH₂Cy), 2.86 [s, 6 H, (CH₃)₂N], 3.77 (br s, 1 H, OH), 4.80 (s, 2 H, CH₂OH).

¹³C NMR (CDCl₃): δ = 13.43 (CH₃CH₂), 20.27 (CH₃CH₂), 26.26, 26.38, 33.13, 37.87 (C_{cy}), 31.91 (CH₂Cy), 38.54 [(CH₃)₂N], 59.36 (CH₂OH), 127.09 (C-4), 140.59 (C-2), 148.96 (C-5).

HRMS MALDI: $\mathit{m/z}$ calcd for $\rm C_{15}H_{28}N_3O_3S$ (MH+) 330.184, found, 330.185.

5-Cyclohexylmethyl-2-hydroxymethyl-4-isopropyl-imidazole-1-sulfonic Acid Dimethylamide (8b) Yield: 0.25 g (48%); mp 118–120 °C.

¹H NMR (CDCl₃): $\delta = 0.86-0.97$ (m, 2 H, H_{cy}), 1.11–1.26 [m, 9H, (CH₃)₂CH and H_{cy}], 1.56–1.71 (m, 6 H, H_{cy}), 2.55 (d, 2 H, *J* = 6.8 Hz, CH₂Cy), 2.78–2.85 [m, 7 H, (CH₃)₂CH and (CH₃)₂N], 3.79 (br s, 1 H, OH), 4.79 (s, 2 H, CH₂OH).

¹³C NMR (CDCl₃): δ = 22.35 [(CH₃)₂CH], 25.72 [(CH₃)₂CH], 26.30, 26.38, 33.11, 37.85 (C_{cy}), 31.67 (CH₂Cy), 38.40 [(CH₃)₂N], 59.38 (CH₂OH), 126.00 (C-4), 144.45 (C-2), 149.08 (C-5).

EIMS: *m*/*z* (%) = 343 (57, M⁺), 260 (100).

Anal. Calcd for $C_{16}H_{29}N_3O_3S$ (343.48): C, 55.95; H, 8.51; N, 12.23. Found: C, 55.78; H, 8.50; N, 12.11.

(5-Cyclohexylmethyl-4-alkyl-1 H-imidazol-2-yl) methanols 9 a, b

A suspension of each of the compounds **7** and **8a,b** (3.5 mmol) in 1.5 M HCl (30 mL) was refluxed for 5 h. The solid product formed was filtered, washed with aq NaHCO₃ (20 mL), Et₂O (30 mL), and dried to afford compounds **9a,b**.

(5-Cyclohexylmethyl-4-ethyl-1*H*-imidazol-2-yl)methanol (9a)

Yield: 70% (obtained from hydrolysis of compound **7**), 75% (obtained from hydrolysis of compound **8a**).

¹H NMR (CDCl₃): $\delta = 0.82-0.92$ (m, 2 H, H_{cy}), 1.05–1.20 (m, 6 H, CH₃CH₂ and H_{cy}), 1.43–1.62 (m, 6 H, H_{cy}), 2.25 (d, 2 H, J = 6.8 Hz, CH₂Cy), 2.37 (q, 2 H, J = 7.4 Hz, CH₃CH₂), 4.34 (s, 2 H, CH₂OH).

¹³C NMR (CDCl₃): δ = 14.80 (*C*H₃CH₂), 18.38 (CH₃CH₂), 26.73, 26.05, 38.11 (C_{cy}), 32.60 (*C*H₂Cy and C_{cy}), 56.93 (CH₂OH), 145.24 (C-2).

EIMS: m/z (%) = 222 (24, M⁺), 139 (100).

Anal. Calcd for $C_{13}H_{22}N_2O\cdot 0.1H_2O$ (224.13): C, 69.67; H, 9.97; N, 12.50. Found: C, 69.55; H, 10.22; N, 12.51.

(5-Cyclohexylmethyl-4-isopropyl-1*H*-imidazol-2-yl)methanol (9b)

Yield: 0.66 g (80%); mp 128–130 °C.

¹H NMR (CDCl₃): $\delta = 0.83-0.93$ (m, 2 H, H_{cy}), 1.05–1.21 [m, 9 H, (CH₃)₂CH and H_{cy}], 1.43–1.63 (m, 6 H, H_{cy}), 2.27 (d, 2 H, *J* = 6.9 Hz, CH₂Cy), 2.81 [hept, 1 H, *J* = 6.8 Hz, (CH₃)₂CH], 4.35 (s, 2 H, CH₂OH].

¹³C NMR (CDCl₃): δ = 23.12 [(*C*H₃)₂CH], 24.66 [(CH₃)₂CH], 25.73, 26.05, 38.07 (C_{cy}), 32.06 (*C*H₂Cy and C_{cy}), 56.94 (CH₂OH), 145.27 (C-2).

EIMS: m/z (%) = 236 (26, M⁺), 153 (100).

Anal. Calcd for $C_{14}H_{24}N_2O$ (236.36): C, 71.14; H, 10.23; N, 11.85. Found: C, 70.94; H, 10.55, N, 11.74.

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References

- Belen'kii, L. I. In *Chemistry of Organosulfur Compounds: General Problems*; Belen'kii, L. I., Ed.; Ellis Horwood: Chichester, **1990**, Chap. 9.
- (2) Becker, S.; Fort, Y.; Vanderesse, R.; Caubère, P. J. Org. Chem. **1989**, 54, 4848.
- (3) Luh, T.-Y.; Ni, Z.-J. Synthesis 1990, 89.
- (4) Capps, H. H.; Dehn, W. M. J. Am. Chem. Soc. 1932, 54, 4301.

- (5) Ainsworth, C. Org. Synth., Coll Vol. V; Wiley: New York, **1973**, 1070.
- (6) Kim, Y. H.; Yon, G. H. J. Chem. Soc., Chem. Commun. 1983, 715.
- (7) Kim, Y. H.; Kim, H. J.; Yon, G. H. J. Chem. Soc., Chem. Commun. 1984, 1064.
- (8) Crestini, C.; Saladino, R.; Nicoletti, R. *Tetrahedron, Lett.* 1993, 34, 1631.
- (9) Claudia, C.; Mincione, E.; Saladino, R.; Nicoletti, R. *Tetrahedron* **1994**, *50*, 3259.
- (10) Abdou, W. M.; Sidky, M. M.; Wamhoff, H. Z. Naturforsch., B: Chem. Sci. 1987, 42, 1153.
- (11) Crank, G.; Nursyidi, A. J. Photochem. Photobiol. A: Chem. 1992, 64, 263.
- (12) Evans, R. M.; Jones, P. G.; Palmer, P. J.; Stephens, F. F. J. *Chem. Soc.* **1956**, 4106.
- (13) Kalm, M. J. J. Org. Chem. 1961, 26, 2925.
- (14) Song, Z.; DeMarco, A.; Zhao, M.; Corley, E. G.; Thompson, A. S.; McNamara, J.; Li, Y.; Rieger, D.; Sohar, P.; Mathre, D. J.; Tschaem, D. M.; Reamer, R. A.; Huntington, M. F.; Ho, G.-J.; Tsay, F.-R.; Emerson, K.; Shuman, R.; Grabowski, J. J.; Reider, P. J. J. Org. Chem. **1999**, *39*, 375.
- (15) Bullerwell, R. A. F.; Lawson, A. J. Chem. Soc. 1952, 1350.
- (16) Loksha, Y. M.; Jørgensen, P. T.; Pedersen, E. B.; El-Badawi,
 M. A.; El-Barbary, A. A.; Nielsen, C. J. Heterocycl. Chem. 2002, 39, 375.
- (17) Loksha, Y. M.; El-Badawi, M. A.; El-Barbary, A. A.; Pedersen, E. B.; Nielsen, C. Arch. Pharm. Pharm. Med. Chem. 2003, 336, 175.
- (18) Bredereck, H.; Gompper, R.; Reich, F. *Chem. Ber.* **1960**, *93*, 723.