

A Palladium-Catalyzed Method for the Synthesis of 2-(α -Styryl)-2,3dihydroquinazolin-4-ones and 3-(α -Styryl)-3,4-dihydro-1,2,4benzothiadiazine-1,1-dioxide: Access to 2-(α -Styryl)quinazolin-4(3H)-ones and $3-(\alpha-Styryl)-1,2,4$ -benzothiadiazine-1,1-dioxides

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Supporting Information

ABSTRACT: An efficient synthesis of 2-(α-styryl)-2,3-dihydroquinazolin-4-ones and 3-(α-styryl)-3,4-dihydro-1,2,4-benzothiadiazine-1,1-dioxides has been achieved in 39-94% yield through palladium-catalyzed cyclocondensation of aryl/vinyl iodides with allenamides 13-15 and 22, respectively. Base treatment of the N-tosylated products provides an easy access to $2 \cdot (\alpha - \alpha)$ styryl) quinazolin-4(3H)-ones and $3-(\alpha$ -styryl)-1,2,4-benzothiadiazine-1,1-dioxides, hitherto unknown heterocycles. The method has been tested with phenyl substituted allenamides, applied for bis-heteroannulation, and used in the preparation of analogues of the natural product Luotonin F.

■ INTRODUCTION

Quinazolin-4(3H)-one (1a, Figure 1) constitutes an important class of heterocycles being an integral part of a large number of naturally occurring alkaloids (~150) and drug candidates. It exhibits a broad range of biological activities such as anticancer, 2a,b antiviral, 2c antimalarial, 2d antifungal, 2e and human microsomal prostaglandin synthase 1 (mPGES-1) inhibitor. 2f The 2-styryl derivatives are considered as privileged structures in medicinal chemistry because of their promising pharmacological effects. For example, compounds of type 2 (Figure 1) exhibit significant cytotoxicity in L1210 murine leukemia cells and inhibitory effects on tubulin polymerizations,³ while CP-465,022 (3, Figure 1) is recognized as a noncompetitive antagonist of AMPA-selective glutamate receptors.4 Very recently, compound 4a was discovered as a potent antibiotic by high throughput virtual screening of a large number of compounds; further optimization by screening studies of analogues led to the development of an even better candidate 4b, which showed remarkable in vitro and in vivo (methicillin-resistant Staphylococus aureus mouse model) efficacy with improved pharmacokinetics.⁵

On the other hand, 1,2,4-benzothiadiazine-1,1-dioxides 1b form another class of remarkably important heterocycles because of their wide range of activities such as antihypertensive, 6a antiviral, 6b antimicrobial, 6c and ATP sensitive potassium

channel opener, 6d among others, and their uses in the development of therapeutics. For example, compound 5 has shown promising activity in the area of diabetes and obesity being a potent HCV NS5B polymerase inhibitor, while Diazoxide (compound 6, Figure 1) is a K⁺[ATP] channel agonist and has a beneficial effect in obese hyperinsulinemic adults.8

Though a number of methods using either conventional reagents 9,10 or transition metal catalysts 11,12 are known for the synthesis of 1a and 1b, only a limited number, 13 which mainly rely on acid-^{13a,b,d,e} or base ^{13f}-promoted Knoevenagel condensation of 2-methyl quinazolinones with aromatic/heteroaromatic aldehydes, exist for the synthesis of $2-\beta$ -styryl derivatives of quinazolin-4(3H)-ones (e.g., 2, 4). On the other hand, the synthetic methods available for $2-\beta$ -styryl derivatives of benzothiadiazine-1,1-dioxide are rare. ¹⁴ To the best of our knowledge, there is no reported synthesis of 2-(α styryl)quinazolin-4(3H)-ones 7 (a regioisomer of the β -styryl counterpart) and their benzothiadiazine-1,1-dioxide analogues 8, which could serve as important pharmacophores in drug discovery. This prompted us to establish convenient methods for their general synthesis. Following our efforts¹⁵ to identify

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Figure 1. General structures of quinazolin-4(3*H*)-ones, 1,2,4-benzothiadiazine-1,1-dioxides, and their 2- β -styryl derivatives, and examples of biologically active compounds belonging to these classes.

lead compounds having significant anticancer properties, we also became interested to get access to the scaffolds 7 and 8, which could enable us to explore their potency in cancer therapy.

Previous workers reported the synthesis of the N^3 -alkylated derivatives of 1a through oxidation of 2,3-dihydroquinazolin-4(1*H*)-ones using conventional oxidants [e.g., DDQ, 16a KMnO₄, 16b MnO₂, 16c I₂ 16d] or metal catalysts. Therefore, we assumed that the synthesis of 7 could also be achieved easily by oxidizing the corresponding dihydro analogues 9 (Figure 1, $R_2 = H$). Scrutiny of the literature revealed only one earlier example, that of Broggini et al. ¹⁷ for the synthesis of N-Boc-2- α styryl-1,2-dihydroquinazolin-4-ones 9 (R₂ = Boc). However, Boc deprotection of the reported compound 9 ($R_2 = Boc, R_1 =$ Ph, R = Me), followed by oxidation of the resulting product using conventional oxidants, 16a-d delivered only a trace amount of 7 $(R_1 = Ph, R = Me)$ in a few cases. This prompted us to conceive a novel strategy for the synthesis through a basepromoted reaction on 9 containing acidic aminal hydrogen (at C2). However, application of this strategy on 9 (R_2 = Boc, R_1 = Ph, R = Me) employing various bases failed to deliver any product. As N-tosylated heterocycles have been reported¹⁸ to undergo elimination of the protecting group (i.e., Ts) under basic conditions along with a neighboring proton to generate a double bond, we chose the sulphonyl moiety as an alternative N-protecting agent to establish a viable synthesis of 7 (Scheme 1). Thus, we needed to synthesize N-sulphonyl derivatives of 9 $(R_2 = SO_2R)$ as precursors, which are unknown in the literature.

Scheme 1. Our Previous Work and Envisaged Route for the Present Synthesis of 7

Encouraged by the recent spectacular advances in the area of palladium-catalyzed heteroannulations involving intramolecular nucleophilic attack of the nitrogen atom onto an allene moiety¹⁹ and in continuation of our work²⁰ on the synthesis of novel heterocycles of biological interests, we envisioned that allenamide 13, generated easily from the acetylenic substrate 10 which was used successfully in our previous synthesis 20a of (E)-2-arylmethylidene-1,4-benzodiazepinones 12, could undergo reaction with aryl/vinyl halides 11 under palladium catalysis, affording the requisite N-tosyl-2,3-dihydroquinazolin-4-one 9 (Scheme 1). Our hypothesis indeed proved to be viable after choosing an appropriate catalyst and a suitable base. We then expanded the scope of this reaction to 1,2,4-benzothiadiazine-1,1-dioxides as well. In a subsequent step, we also realized the convenient synthesis of products 9 employing a suitable base. Herein, we describe the results obtained so far in detail.

■ RESULTS AND DISCUSSION

Synthesis of anthranilic allenamide 13, the requisite starting substrate, was achieved (with 85% yield) easily upon base treatment of the N-propargylated anthranilamide 10. Thereafter, we attempted the synthesis of N-sulphonyl-2- α -styryl-2,3dihydroquinazolin-4-ones 9 at room temperature through palladium-catalyzed reactions. Initially, carrying out the reaction of 13 with phenyl iodide (11a) employing the catalyst, solvent, and base used by Broggini et al.¹⁷ required 14 h to produce the desired product 9aa (75% yield), thereby marking the process as sluggish (Table 1, entry 1). In order to find the optimal reaction conditions, we performed a series of reactions on the model substrates 11a and 13 by changing catalyst, base, solvent, etc.; selected results are summarized in Table 1. Employment of our recently reported^{20a} catalytic system [Pd/C, PPh₃, K₂CO₃, DMF] in this reaction required even more time (26 h) to complete this reaction at rt (Table 1, entry 2 vs 1). Interestingly, replacing the catalyst by Pd(dba)₂ reduced the reaction time (15 h) and improved the yield (90%) significantly (Table 1, entry 3). To our disappointment, switching to a more polar solvent like DMSO proved to be ineffective, decreasing the yield to 50% even after increasing the reaction time to 60 h (Table 1, entry 4), while an alkaline base (i.e., KOH) proved to be completely inefficient, showing no sign of product formation (Table 1, entry 5). We were encouraged to note that the use of Pd(MeCN)₂Cl₂/PPh₃ and K₂CO₃ in DMF produced the product 9aa within 8 h, albeit in moderate (47%) yield (Table 1, entry 6). Gratifyingly, use of Pd(OAc)₂ allowed the reaction to be completed within 1 h at rt, affording product 9aa in 94% yield (Table 1, entry 7). The use of other bases (e.g.,

Table 1. Optimization of the Reaction Conditions for the Synthesis of 9aa at Room Temperature^a

Sl. no.	cat. ^b	base	solvent	time (h)	yield (%) ^c
1	Pd(PPh ₃) ₄	K_2CO_3	CH ₃ CN	14	75
2	10% Pd/C	K_2CO_3	DMF	26	68
3	$Pd(dba)_2$	K_2CO_3	DMF	15	90
4	$Pd(dba)_2$	K_2CO_3	DMSO	60	50
5	$Pd(dba)_2$	KOH	DMF	2	0
6	$Pd(MeCN)_2Cl_2$	K_2CO_3	DMF	8	47
7	$Pd(OAc)_2$	K_2CO_3	DMF	1	94
8	$Pd(OAc)_2$	KO ^t Bu	DMF	1	91
9	$Pd(OAc)_2$	Cs_2CO_3	DMF	4	82
10	$Pd(OAc)_2$	Et_3N	DMF	2.5	88
11	$Pd(PPh_3)_2Cl_2$	K_2CO_3	DMF	30	65
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"Reaction conditions: 13 (1.0 equiv), 11a (1.0 equiv), palladium catalyst (0.05 equiv), PPh₃ (0.12 equiv, except for entry 1), base (4.0 equiv) in a solvent under argon. ^bPPh₃ was used as ligand except for entry 1. ^cYields of isolated pure products.

 KO^tBu , Cs_2CO_3 , Et_3N) neither improved the yield nor reduced the reaction time (Table 1, entries 8–10). Additionally, $Pd(PPh_3)_2Cl_2$ proved unsuitable (Table 1, entry 11). DMF was found to be the best solvent. Thus, the reaction conditions of entry 7 of Table 1 appeared to be optimal.

Using the optimized reaction conditions, we then extended the scope of this procedure to anthranilic allenamides 13–15 and a range aryl/heteroaryl/vinyl iodides 11a–1 (Table 2). Indeed, allenamide 13 reacted successfully with a series of

iodides 11a-l, furnishing the expected products 9aa-al with moderate to excellent yields (Table 2, entries 1-12). Heteroaryl iodides without any substitution (11b,c) also produced desired products 9ab-ac, though with somewhat lower yields (66-73%) compared to phenyl iodide (11a), proving their lower reactivity in this reaction (Table 2, entries 2-3 vs entry 1). An electron-donating group enhanced the reactivity (in 11j), delivering product 9aj with 89% yield (Table 2, entry 10). On the other hand, irrespective of the substitution, all aryl iodides 11d-i possessing either electron-donating (viz., Me, OMe) or electron-withdrawing groups (viz., CF₃, CHO, CO₂Me, NO₂) efficiently participated in this reaction to produce the products (9ad-ai) within 2.5 h with 66-86% yields (Table 2, entries 4-9). Additionally, olefinic iodides were found to be reactive under the conditions (Table 2, entries 11 and 12). Though vinyl iodide 11k furnished the product 9ak with moderate yield (39%), 2,4-dimethyl-5-iodo-uracil (111) afforded the desired compound 9al with 63% yield. Unfortunately, neither any aryl bromide/chloride nor allenamide without N-alkylation (of the amidic nitrogen) reacted.

To check the effect of other N-protecting groups, the reactivity of trifluoroacetanilide substituted allenamide 14 toward different aryl iodides was explored (Table 2, entries 13–16). Interestingly, this gave direct access to the N-deprotected products (Table 2, entries 13–16), which is similar to the observations of Cacchi and co-workers. Furthermore, the presence of an electron-withdrawing group in the iodide facilitated the reaction compared to an electron-donating group (Table 2, entry 16 vs 15).

Encouraged by these results, we became interested to test allenamide 15, which has a free amine group, in this reaction. Interestingly, this too resulted in the formation of the desired products with moderate to good yields (Table 2, entries 17–

Table 2. Palladium-Catalyzed Synthesis of 2- $(\alpha$ -Styryl)-2,3-dihydroquinazolin-4-ones 9^{α}

RI +
$$\begin{array}{c} O \\ NH \\ NH \\ R_1 \\ \end{array}$$
 $\begin{array}{c} Pd(OAc)_2 \ , PPh_3 \\ K_2CO_3 \ , DMF \\ \text{rt} \\ 9a, 13 \colon R_1 = Ts \\ 14 \colon R_1 = COCF_3 \\ 9b, 15 \colon R_1 = H \end{array}$

Entry	Allenar	mide RI (11)	Time (h)	Product	Yield (%) ^b	Entry	Allenamid	le RI (11)	Time (h)	Product	Yield (%) ^b
1	13	Ph; 11a	1	9aa	94	11	13	vinyl; 11k	1.5	9ak	39
2	13	3-pyridyl; 11b	0.5	9ab	73			O			
3	13	2-thienyl; 11c	0.75	9ac	66	12	13	N Me	0.75	9al	63
4	13	C ₆ H ₄ Me-p; 11d	0.25	9ad	73			N O			
5	13	C ₆ H ₄ CF ₃ -p; 11e	2	9ae	74			11I N Me			
6	13	C ₆ H ₄ OMe- <i>p</i> ; 11f	0.5	9af	74	13	14	11a	0.75	9ba	74
7	13	C ₆ H ₄ CO ₂ Me- <i>p</i> ; 11g	2.5	9ag	86	14	14	11e	2	9bb	58
8	13	C ₆ H ₄ CHO- <i>o</i> ; 11h	0.5	9ah	75	15	14	11f	1.5	9bc	50
9	13	C ₆ H ₄ NO ₂ -p; 11i	1	9ai	66	16	14	11g	1.5	9bd	83
		OMe				17	15	11a	0.75	9ba	60
10	13		4	9aj	89	18	15	11f	1.5	9bc	49
) N	·	,		19	15	11g	1.5	9bd	80
		N 11j OMe				20	15	11j	6.5	9be	55

^aReaction conditions: 13/14/15 (1.0 equiv), 11 (1.0 equiv), $Pd(OAc)_2$ (0.05 equiv), PPh_3 (0.12 equiv), K_2CO_3 (4.0 equiv) in DMF under argon. ^bYield of isolated pure product.

20). However, in line with our previous observation, an iodide possessing an electron-withdrawing group (e.g., CO_2Me) turned out to be more reactive compared to one having an electron-donating group like OMe (Table 2, entry 19 vs 18, 20). From these studies, it is apparent that allenamide 13 containing the N-tosyl group possessed better reactivity than other allenamides (14, 15) and delivered the products with very good yields. This underlines the crucial role of the acidic hydrogen attached to the aryl amino moiety (in allenamide), acting as a nucleophile during cyclization.

Bis-heteroannulated products 17a and 17b could also be accessed easily, though in moderate yields, when allenamide 13 was allowed to react with aryl diiodides 16a and 16b under the optimized reaction conditions, suggesting polyheteroannulation to be a viable process in one pot (Scheme 2).

Scheme 2. Synthesis of Bis-heteroannulated Products 17a-h^a

"Reaction conditions: 13 (1.0 equiv), 16a/16b (0.5 equiv), $Pd(OAc)_2$ (0.05 equiv), PPh_3 (0.12 equiv), K_2CO_3 (4.0 equiv) in DMF under argon.

To check the scope of this method further, an internal alkyne 18 was probed. It was surprising to find that even phenyl substituted crude allenamide 19 (having a propensity toward decomposition during purification through silica-gel column chromatography) reacted successfully with iodo compound 11g under the optimized reaction conditions to furnish the stereoselective product 20 in 55% yield (Scheme 3). The (E)-stereochemistry of the product was confirmed by 2D NMR. Of particular interest was the weak, but perceptible, NOE cross peaks observed between the signals for the ortho protons (with respect to the olefinic double bond, as marked with green arrow) of the two phenyl rings of the styryl unit which established the geometry shown (see the Supporting Information).

After having accomplished a general synthesis of $2-(\alpha$ -styryl)-2,3-dihydroquinazolin-4-ones 9, we planned to replace the carbonyl (CO) group of allenamide 13 by a SO₂ group and

used the allenesulfonamide derivative 22 for the synthesis of $3-(\alpha\text{-styryl})$ -3,4-dihydro-1,2,4-benzothiadiazine-1,1-dioxide 23. An initial base (KO^tBu)-promoted isomerization of the precursor acetylene compound (21) furnished the desired allenesulfonamide 22. Since this compound was found to decompose during chromatographic purifications, we directly exposed the crude allenesulfonamide (obtained after usual workup) to react with phenyl iodide (11a) and 3-iodo-pyridine (11b) separately under optimized reaction conditions; to our disappointment, conducting these reactions at rt required 30 and 45 h, respectively, to deliver the corresponding products 23a (47%) and 23b (55%), proving them inconvenient to pursue. Pleasingly, the reactions were found to be completed within 25–30 min upon heating at 60 °C, and the yield improved as well (Table 3, entries 1 and 2). Furthermore, a

Table 3. Palladium-Catalyzed Synthesis of 3-(α-Styryl)-3,4-dihydro-1,2,4-benzothiadiazine-1,1-dioxide $23^{a,b}$

SI. No.	RI (11)	time (min)	product	yield (%) ^c
1	11a	30	23a	67
2	11b	25	23b	62
3	11c	30	23c	40
4	11d	15	23d	63
5	11e	10	23e	88
6	11f	60	23f	39
7	11g	30	23g	52
8	IC ₆ H ₄ F-p; 11m	15	23m	63
9	IC ₆ H ₄ Br-p; 11n	10	23n	61

^aReaction conditions: **21** (1.0 equiv), KO^tBu (2.0 equiv) in DMF. ^bReaction conditions: Crude **22** treated with **11** (1.0 equiv), Pd(OAc)₂ (0.05 equiv), PPh₃ (0.12 equiv), and K₂CO₃ (4.0 equiv) in DMF under argon. ^cYield was calculated based on **21**.

heteroaryl iodide (Table 3, entry 3) and aryl iodides containing either electron-donating (Table 3, entries 4, 6) or electron-withdrawing (Table 3, entries 5, 7, 8, 9) groups underwent reaction successfully with crude allenesulfonamide 22 within 10–60 min to afford the expected products in moderate to very good yields (39–88%).

The structures of the products²³ were unambiguously deduced from spectroscopic (¹H, ¹³C NMR, HRMS) and

Scheme 3. Reaction of Phenyl Substituted Allenamide 19 with Iodo Compound 11g^{a,b}

"Reaction conditions: KO'Bu (2.0 equiv), in DMF at rt under an argon atmosphere. "Reaction conditions: Crude 19 treated with 11g (1.0 equiv), Pd(OAc)₂ (0.05 equiv), PPh₃ (0.12 equiv), and K₂CO₃ (4.0 equiv) in DMF under argon.

Scheme 4. Base-Promoted Synthesis of Products 7, 8^{a,b}

"Reaction conditions: For substrate **9a** (1.0 equiv), 5.0 equiv of NaH (60% dispersion in mineral oil) used; for substrate **23** (1.0 equiv), 2.0 equiv of the same base used, as excess base (5.0 equiv) was detrimental for yields. "Yield of isolated pure product.

analytical data. These were further supported by single crystal X-ray diffraction analysis (see the Supporting Information) of a few products (9aa, 9bd, and 23g).²²

After achieving the synthesis of $2-(\alpha\text{-styryl})-2,3\text{-dihydro-quinazolin-4-ones}$ 9 and their benzothiadiazine-1,1-dioxide variants 23, attention was turned toward the synthesis of the unsaturated analogues 7 through a base-promoted reaction as depicted in Scheme 4. Initially, we performed a series of reactions on model compound 13a varying the base and solvent as well; representative data are shown in Table 4. KO^tBu-

Table 4. Optimization of the Reaction Conditions for the Synthesis of $2-(\alpha-\text{Styryl})$ quinazolin-4(3H)-one 7a

entry	base	solvent	time	$yield^{c}$ (%)
1 ^a	KO^tBu	THF	30 h	28
2 ^a	KO^tBu	DMF	5 min	40
3 ^a	KOH	DMF	5 min	60
4 ^a	DBU	DMF	38 h	0
5 ^b	NaH	DMF	30 min	88
6 ^b	NaH	CH ₂ CN	38 h	60

"Reaction conditions: **9aa** (1.0 equiv), base (3.0 equiv). ^bReaction conditions: **9aa** (1.0 equiv), NaH (60% dispersion in mineral oil, 5.0 equiv). 'Yield of isolated pure product.

promoted reaction in THF at rt remained mostly incomplete (TLC), even after extending the reaction to 30 h, and yielded only 28% of the desired product 7a along with the recovery (58%) of the reactant 9aa (Table 4, entry 1). Gratifyingly, the reaction was complete within 5 min on replacing the solvent by DMF and the yield also improved somewhat (40%), though several side products (TLC) were generated (Table 4, entry 2). Thereafter, the use of KOH instead of KO^tBu increased the yield to 60% (Table 4, entry 3), but the product 7a was still accompanied by side products (TLC) in minor amounts. Though a strong organic base like DBU proved to be ineffective for this reaction (Table 4, entry 4), the use of NaH (5 equiv) was beneficial, leading to the formation of 7a (88% yield)

within 30 min (Table 4, entry 5); however, decreasing the amount of NaH (3 equiv) caused incomplete conversion of starting material even after 48 h. Further, changing the solvent to acetonitrile was not useful (Table 4, entry 6). Thus, reaction conditions of entry 5 in Table 4 proved optimal. We then checked the scope and limitations of this protocol by varying the substrates as shown in Scheme 4.

As depicted in Scheme 4, substrates 9aa–9ac containing a simple aryl/heteroaryl group in the styryl moiety afforded products (7a-c) in very good yield (74-88%). Even after introduction of an electron-donating group (viz. Me/OMe) in the aryl ring (substrates 9ad/9af), the corresponding products 7d/7f were isolated with comparable yields (78-85%). In contrast, an electron-withdrawing substituent (viz. CF_3) at the same position of the substrate (9ae) reduced the yield (52%) of the product (7e). On the other hand, employing substrates 23a and 23b having a SO_2 moiety in place of carbonyl delivered the desired products 8a and 8b, respectively, in moderate yield (57%) and 50%.

The structures of the products were confirmed by spectroscopic analysis (1 H NMR, 13 C NMR, HRMS) and further supported by single crystal X-ray diffraction of compound 7d (see the SI). 22

We next applied this methodology for synthesizing (Scheme 5) analogues of *Luotonin* F, 24 a natural alkaloid extracted from the aerial parts of *Peganum nigellastrum* Bunge, the plant which finds applications in the Chinese system of medicines for the treatment of rheumatism, inflammation, abscesses, and other diseases. 25 The total synthesis of *Luotonin* F itself applying our methodology is presently underway.

CONCLUSION

In conclusion, we have successfully developed a facile and mild method for the general synthesis of $2-(\alpha\text{-styryl})-2,3\text{-dihydro-quinazolin-4-ones 9}$ via palladium-catalyzed cyclocondensation of aryl/heteroaryl (or vinyl) iodide 11 with allenamides 13–15 at room temperature. This reaction can easily be extended for bis-heteroannulation and the use of phenyl substituted allenamide as well. Besides, the scope of this method has been extended to the synthesis of $3-(\alpha\text{-styryl})-3,4\text{-dihydro-1,2,4-benzothiadiazine-1,1-dioxides 23}$ by varying the substrate (22) and tweaking the temperature (i.e., 60 °C). A base-

Scheme 5. Application of the Methodology for the Synthesis of Luotonin F Analogues a,b,c

^aReaction conditions: 2.0 equiv of KO^tBu was used for 10/21 (1.0 equiv), but 3.0 equiv of the same base was used for 24 (1.0 equiv). ^bReaction conditions: 13/25 (1.0 equiv), 26 (1.0 equiv), Pd(OAc)₂ (0.05 equiv), PPh₃ (0.12 equiv), K₂CO₃ (4.0 equiv). ^cFor 1.0 equiv of 27a and 27c, 5.0 equiv of NaH (60% dispersion in mineral oil) was used; while, for 27b (1.0 equiv), 2.0 equiv of the same base was used.

promoted reaction of the products (9a, 23) provided an easy access to 2-(α -styryl)quinazolinones 7 and 3-(α -styryl)-1,2,4benzothiadiazine-1,1-dioxide 8, a novel class of compounds. This methodology has also been applied in the generation of a few analogues of *Luotonin F*. As $2-(\beta-\text{styryl})$ quinazolinones have already proved to be potent inducers of apoptosis and possess other remarkable phamacological activities, their 2-(α styryl) variants are also anticipated to possess potent medicinal value. Work in this direction is currently in progress and will be reported in due course.

■ EXPERIMENTAL SECTION

General. Melting points are uncorrected. Analytical thin-layer chromatography (TLC) was performed on Silica gel 60-F254 aluminum TLC sheets. Visualization of the developed chromatogram was performed by UV absorbance or iodine exposure. For purification, column chromatography was performed using 100-200 mesh silica gel. ¹H and ¹³C NMR spectra were recorded at 298 K on a 300, 400, 500, or 600 MHz spectrometer using tetramethylsilane (TMS) as internal standard. Chemical shifts (δ) are given from TMS (δ = 0.00) in parts per million (ppm) with reference to the residual nuclei of the deuterated solvent used [CDCl₃: ¹H NMR δ = 7.26 ppm (s); ¹³C NMR $\delta = 77.0$ ppm]. Coupling constants (*J*) are expressed in hertz (Hz), and spin multiplicities are given as s (singlet), d (doublet), dd (double doublet), t (triplet), m (multiplet), and br (broad). All ¹³C NMR spectra were obtained with broad-band proton decoupling. To assign the structures under consideration, the following 1D and 2D experiments were employed: 1H selective NOESY 2D: COSY and ¹H-¹³C gradient selected HSQC and HMBC in a 600 MHz NMR machine. Mass spectra were performed using ESI-TOF, EI or FAB ionization mode.

Synthesis of Allenamides 13-15 from Corresponding Acetylene Compounds 10, S1, S2. Synthesis of the Acetylenic Substrate N-Methyl-N-(prop-2-yn-1-yl)-2-(2,2,2-trifluoroacetamido)benzamide (\$1). The acetylene compounds 2-amino-Nmethyl-N-(prop-2-yn-1-yl)benzamide (S2) and N-methyl-2-[(4methylphenyl)sulfonamide]-*N*-(prop-2-yn-1-yl)benzamide (10) were prepared by a known literature procedure. ^{20a} *N*-Methyl-*N*-(prop-2-yn-1-yl)benzamide (10) were 1-yl)-2-(2,2,2-trifluoroacetamido)benzamide (S1) was prepared by adding trifluoroacetic anhydride (223 mg, 1.06 mmol) and triethylamine (81 mg, 0.80 mmol) in the solution of S2 (100 mg, 0.53 mmol) in THF (5.0 mL) at 0 °C, followed by stirring at room temperature for 3 h. After completion of reaction (TLC), the solvent was evaporated in vacuo; the residue was washed with aqueous NaHCO3 solution and partitioned between EtOAc and brine. The organic layer was dried over Na2SO4 and concentrated in vacuo. The crude product was then

purified by silica gel (100-200 mesh) column chromatography using 12% ethyl acetate-petroleum ether (v/v) as eluent.

N-Methyl-N-(prop-2-yn-1-yl)-2-(2,2,2-trifluoroacetamido)benzamidė (S1). Pale yellow amorphous solid (130 mg, 86% yield); mp 90–92 °C; ¹H NMR (CDCl₃, 300 MHz) δ 10.32 (brs, 1H), 8.29 (d, J = 8.4 Hz, 1H), 7.54-7.48 (m, 2H), 7.28-7.23 (m, 1H), 4.25 (m, 2H)2H), 3.18 (s, 3H), 2.37 (s, 1H); 13 C NMR (CDCl₃, 75 MHz) δ 169.7, 155.1 (q, J = 37.2 Hz), 134.8, 131.6, 127.9, 125.0, 124.0, 122.9, 115.6 (q, J = 286.7 Hz), 77.6, 73.4, 42.4, 36.9 ppm; HRMS (ESI⁺) m/z calculated for $C_{13}H_{11}F_3N_2O_2Na$ [M + Na]⁺ 307.0670, found 307.0672.

Procedure for Conversion of 10, S1, S2 to Allenamides 13, 14, 15. N-Methyl-2-[(4-methylphenyl)sulfonamide]-N-(propa-1,2dien-1-yl)benzamide (13). To a stirred solution of acetylene 10 (100 mg, 0.29 mmol) in dry DMF was added KO^tBu (66 mg, 0.58 mmol); the reaction was found to be completed within 5 min. (TLC). It was then quenched with water (5 mL) and extracted with EtOAc (3 × 20 mL). The combined organic extracts were dried over Na₂SO₄ and concentrated in vacuo. The resulting residue was purified by silica gel (100-200 mesh) column chromatography using 20% EtOAcpetroleum ether (v/v) as eluent to furnish the product 13 as a white amorphous solid (85 mg, 85%); mp 108-109 °C; ¹H NMR (CDCl₃, 600 MHz) δ 8.41 (brs, 1H), 7.69 (d, J = 7.8 Hz, 1H), 7.55 (brs, 2H), 7.41 (t, I = 7.2 Hz, 1H), 7.21–7.12 (m, 4H), 5.98 (s, 1H), 5.45–5.35 (m, 2H), 2.97 (s, 2H), 2.47 (s, 1H), 2.33 (s, 3H); ¹³C NMR (CDCl₃, 150 MHz) δ 201.3, 168.9, 145.6, 137.9, 137.7, 133.1, 131.0, 130.6, 129.4, 128.6, 127.2, 126.1, 104.5, 101.0, 88.9, 37.1, 32.9, 31.2, 23.0 ppm. HRMS (EI⁺) m/z calculated for $C_{18}H_{18}N_2O_3S$ [M]⁺ 342.1038, found 342.1034.

N-Methyl-N-(propa-1,2-dien-1-yl)-2-(2,2,2-trifluoroacetamido)benzamide (14). To a well-stirred solution of acetylene S1 (100 mg, 0.35 mmol) in dry DMF was added KO^tBu (316 mg, 2.82 mmol), and the reaction was continued for another 10 min until completion (TLC). Then, the reaction was quenched with water (5 mL) and extracted with EtOAc (3 × 20 mL). The combined organic extracts were dried over Na2SO4 and concentrated in vacuo; the resulting residue was purified by silica gel (100-200 mesh) column chromatography using 10% ethyl acetate-petroleum ether (v/v) as eluent to furnish the product 14 as a pale yellow amorphous solid (89 mg, 91%); mp 82–84 °C; 1 H NMR (CDCl₃, 300 MHz) δ 10.39 (brs, 1H), 8.35 (d, J = 8.4 Hz, 1H), 7.55-7.50 (m, 2H), 7.29-7.24 (m, 1H), 6.80 (brs, 1H), 5.46 (brs, 2H), 3.19 (brs, 3H); ¹³C NMR (CDCl₃, 75 MHz) δ 200.3, 167.9, 155.1 (q, J = 37.5 Hz), 135.2, 131.8, 124.8 (two carbon signals merged probably), 122.8 (two carbon signals merged probably), 115.6 (q, J = 286.9 Hz), 103.0, 87.6, 31.9 ppm. HRMS (ESI⁺) m/z calculated for $C_{13}H_{12}F_3N_2O_2$ [M + H]⁺ 285.0851, found 285.0841.

2-Amino-N-methyl-N-(propa-1,2-dien-1-yl)benzamide (15). To a well-stirred solution of acetylene S2 (100 mg, 0.53 mmol) in dry DMF was added KOH (179 mg, 3.19 mmol). The reaction was found to be

completed within 2 min (TLC). It was then guenched with water (5 mL) and extracted with EtOAc (3 × 20 mL). Combined organic extracts were dried over Na2SO4 and concentrated in vacuo. The resulting residue was purified by silica gel (100-200 mesh) column chromatography using 10% EtOAc-petroleum ether (v/v) as eluent to afford 15 as a yellow gum (55 mg, 55%); ¹H NMR (CDCl₃, 300 MHz) δ 7.22–7.17 (m, 3H), 6.75–6.71 (m, 2H), 5.41–5.39 (m, 2H), 4.43 (brs, 2H), 3.12 (s, 3H); 13 C NMR (CDCl₃, 75 MHz) δ 201.7. 169.3, 146.2, 131.2, 128.9, 118.4, 117.2, 116.8, 99.9, 86.9, 32.4 ppm. HRMS (ESI⁺) m/z calculated for C₁₁H₁₂N₂ONa [M + Na]⁺ 211.0847, found 211.0847.

General Procedure for Palladium-Catalyzed Synthesis of 2-(α -Styryl)-2,3-dihydroquinazolin-4-ones 9a,b (Table 2). To a well-stirerd solution of Pd(OAc)2 (0.05 equiv) and PPh3 (0.12 equiv) in dry DMF (4 mL) was sequentially added iodo compound 11 (1 equiv), K2CO3 (4 equiv), and allene 13/14/15 (1 equiv) under an argon atmosphere. The whole reaction mixture was allowed to stir at room temperature for the requisite time (see Table 2) until completion (TLC). DMF was removed in vacuo. The resulting mixture was extracted with EtOAc (3 × 20 mL) and washed with water (10 mL). The combined organic extracts were dried over Na₂SO₄ and concentrated in vacuo. The crude residue was then purified by (silica gel 100-200 mesh) column chromatography using 10-50% ethyl acetate-petroleum ether (v/v) as eluent to afford the desired products (9a, 9b).

3-Methyl-2-(1-phenylvinyl)-1-tosyl-2,3-dihydroquinazolin-4(1H)one (9aa). Pale brown crystalline solid (68.9 mg, 94% yield); mp 192–193 °C; ¹H NMR (CDCl₃, 300 MHz) δ 7.86 (dd, J = 7.5 Hz, 1.5 Hz, 1H), 7.49 (td, J = 7.5 Hz, 1.5 Hz, 2H), 7.42–7.33 (m, 7H), 7.26– 7.22 (m, 2H), 7.13 (d, J = 7.8 Hz, 2H), 6.45 (s, 1H), 5.31 (d, J = 1.5Hz, 1H), 4.89 (d, I = 2.1 Hz, 1H), 2.93 (s, 3H), 2.35 (s, 3H); 13 C NMR (CDCl₃, 75 MHz) δ 161.5, 144.7, 142.6, 137.4, 135.3, 133.9, 132.6, 129.5, 128.4, 128.2, 127.5, 127.0, 126.9, 125.6, 117.3, 73.8, 33.7, 21.6 ppm (two C signals seemed to overlap with other aromatic carbons); HRMS (EI⁺) m/z calculated for $C_{24}H_{22}N_2O_3S$ [M]⁺ 418.1351, found 418.1352.

3-Methyl-2-[1-(pyridin-3-yl)vinyl]-1-tosyl-2,3-dihydroquinazolin-4(1H)-one (9ab). Pale yellow solid (53.7 mg, 73% yield); mp 181–182 °C; ¹H NMR (CDCl₃, 300 MHz) δ 8.61 (d, J = 4.5 Hz, 1H), 8.47 (brs, 1H), 7.88 (d, J = 7.8 Hz, 1H), 7.84 (d, J = 7.8 Hz, 1H), 7.56–7.50 (m, 1H), 7.45-7.35 (m,3H), 7.28-7.24 (m, 2H), 7.16 (d, J = 8.4 Hz, 2H), 6.41 (s, 1H), 5.40 (s, 1H), 5.03 (d, J = 1.2 Hz, 1H), 2.97 (s, 3H), 2.38 (s, 3H); 13 C NMR (CDCl₃, 75 MHz) δ 161.3, 149.4, 148.2, 144.9, 139.9, 135.1, 134.1, 133.8, 133.2, 132.9, 129.6, 127.8, 127.7, 126.9, 126.9, 125.4, 123.1, 118.7, 73.6, 33.8, 21.6 ppm; HRMS (FAB+) m/z calculated for $C_{23}H_{22}N_3O_3S$ [M + H]⁺ 420.1382, found 420.1377.

3-Methyl-2-[1-(thiophen-2-yl)vinyl]-1-tosyl-2,3-dihydroquinazolin-4(1H)-one (9ac). White amorphous solid (49.1 mg, 66% yield); mp 182–184 °C; ¹H NMR (CDCl₃, 300 MHz) δ 7.84 (dd, J =7.6 Hz, 1.3 Hz, 1H), 7.62–7.57 (m, 2H), 7.52 (td, I = 7.6 Hz, 1.3 Hz, 1H), 7.36 (td, J = 7.5 Hz, 1.1 Hz, 1H), 7.26–7.23 (m, 3H), 7.14 (d, J =8.4 Hz, 2H), 7.09-7.06 (m, 1H), 6.39 (s, 1H), 5.47 (s, 1H), 4.73 (s, 1H), 2.89 (s, 3H), 2.36 (s, 3H); 13 C NMR (CDCl₃, 75 MHz) δ 161.4, 144.9, 139.7, 135.3, 135.1, 133.6, 132.8, 129.5, 127.8, 127.8, 127.5, 127.4, 127.1, 125.9, 125.6, 125.4, 115.6, 73.7, 33.5, 21.7 ppm; HRMS (EI⁺) m/z calculated for $C_{22}H_{20}N_2O_3S_2$ [M]⁺ 424.0915, found 424.0924.

3-Methyl-2-[1-(p-tolyl)vinyl]-1-tosyl-2,3-dihydroquinazolin-4(1H)-one (9ad). Pale brown amorphous solid (55.3 mg, 73% yield); mp 164–166 °C; ¹H NMR (CDCl₃, 300 MHz) δ 7.85 (dd, J = 7.5 Hz, 1.5 Hz, 1H), 7.51-7.46 (m, 1H), 7.43-7.40 (m, 1H), 7.34 (td, J = 7.4Hz, 1.3 Hz, 2H), 7.28-7.21 (m, 3H), 7.18-7.11 (m, 4H), 6.43 (s, 1H), 5.28 (d, J = 1.2 Hz, 1H), 4.83 (d, J = 1.5 Hz, 1H), 2.92 (s, 3H), 2.37–2.35 (m, 6H); 13 C NMR (CDCl₃, 75 MHz) δ 161.5, 144.7, 142.4, 138.1, 135.4, 134.5, 134.1, 132.7, 129.6, 129.1, 127.6, 127.5, 127.1, 127.0, 126.8, 125.7, 116.6, 73.9, 33.7, 21.7, 21.2 ppm; HRMS (EI⁺) m/z calculated for $C_{25}H_{24}N_2O_3S$ [M]⁺ 432.1508, found

3-Methyl-1-tosyl-2-{1-[4-(trifluoromethyl)phenyl]vinyl}-2,3dihydroquinazolin-4(1H)-one (9ae). Yellow amorphous solid (63.0 mg, 74% yield); mp 178–179 °C; 1 H NMR (CDCl₃, 600 MHz) δ 7.86 (dd, J = 7.8 Hz, 1.2 Hz, 1H), 7.63 (d, J = 7.8 Hz, 2H), 7.50 (td, J = 7.8 Hz, 2H), 7.50 (td, J = 7.8 Hz, 2Hz)Hz, 1.4 Hz, 1H), 7.46 (d, J = 8.4 Hz, 2H), 7.40–7.39 (m, 1H), 7.36 (td, J = 7.5 Hz, 1.0 Hz, 1H), 7.23 (d, J = 7.8 Hz, 2H), 7.14 (d, J = 8.4 Hz, 2H), 6.45 (t, I = 1.5 Hz, 1H), 5.38 (d, I = 1.2 Hz, 1H), 4.99 (d, I = 1.2 Hz, 1H), 4.90 (d, I = 1.2 Hz, 1H), 4.90 (d, I = 1.2 H 1.8 Hz, 1H), 2.95 (s, 3H), 2.36 (s, 3H); ¹³C NMR (CDCl₃, 150 MHz) δ 161.4, 144.9, 141.9, 141.0, 135.2, 133.8, 132.8, 130.3 (q, J = 32.0Hz), 129.6, 127.7, 127.7, 127.3, 127.0, 126.9, 125.5, 125.4 (q, J = 3.7)Hz), 124.0 (q, J = 270.6 Hz), 118.9, 73.7, 33.8, 21.6 ppm; HRMS (EI⁺) m/z calculated for $C_{25}H_{21}F_{3}N_{2}O_{3}S$ [M]⁺ 486.1225, found 486.1216.

2-[1-(4-Methoxyphenyl)vinyl]-3-methyl-1-tosyl-2,3-dihydroquinazolin-4(1H)-one (9af). Brown amorphous solid (58.2 mg, 74% yield); mp 150–151 °C; ¹H NMR (CDCl₃, 300 MHz) δ 7.85 (dd, J =7.8 Hz, 1.2 Hz, 1H), 7.52-7.41 (m, 2H), 7.37-7.31 (m, 3H), 7.26-7.22 (m, 2H), 7.13 (d, J = 8.4 Hz, 2H), 6.90 (d, J = 8.7 Hz, 2H), 6.42 (s, 1H), 5.26 (d, I = 0.9 Hz, 1H), 4.80 (d, I = 1.5 Hz, 1H), 3.84 (s, 3H), 2.92 (s, 3H), 2.35 (s, 3H); 13 C NMR (CDCl₃, 75 MHz) δ 161.4, 159.4, 144.6, 141.7, 135.3, 133.9, 132.5, 129.6, 129.4, 128.0, 127.4, 126.9, 126.9, 125.6, 115.8, 113.7, 73.8, 55.1, 33.6, 21.5 (one carbon signal has probably merged with other aromatic carbon signals) ppm; HRMS (EI⁺) m/z calculated for $C_{25}H_{24}N_2O_4S$ [M]⁺ 448.1457, found 448.1453.

Methyl 4-[1-(3-Methyl-4-oxo-1-tosyl-1,2,3,4-tetrahydroquinazolin-2-yl)vinyl]benzoate (9ag). White amorphous solid (71.8 mg, 86% yield); mp 198–200 °C; 1 H NMR (CDCl₃, 300 MHz) δ 8.06 (d, J = 8.7 Hz, 2H), 7.87 (d, J = 8.1 Hz, 1H), 7.54–7.48 (m, 1H), 7.43 (d, J = 8.4 Hz, 2H), 7.40-7.35 (m, 1H), 7.24 (d, J = 8.1 Hz, 2H), $7.15(d, J = 8.1 \text{ Hz}, 2H), 6.48 \text{ (s, 1H)}, 5.41 \text{ (d, } J = 0.9 \text{ Hz}, 1H), 4.99 \text{ (d, } J = 0.9 \text$ J = 1.5 Hz, 1H), 3.96 (s, 3H), 2.97 (s, 3H), 2.37 (s, 3H); ¹³C NMR (CDCl₃, 75 MHz) δ 166.7, 161.4, 144.8, 142.2, 141.9, 135.2, 133.9, 132.8, 129.8, 129.7, 129.6, 127.6, 127.6, 127.0, 126.9, 125.5, 118.6, 73.6, 52.1, 33.8, 21.6 ppm (one carbon signal is expected to merge with other aromatic carbon signals); HRMS (EI⁺) m/z calculated for $C_{26}H_{24}N_2O_5S$ [M]⁺ 476.1406, found 476.1397.

2-[1-(3-Methyl-4-oxo-1-tosyl-1,2,3,4-tetrahydroguinazolin-2-yl)vinyl]benzaldehyde (9ah). Pale yellow amorphous solid (58.7 mg, 75% yield); mp 156–157 °C; ¹H NMR (CDCl₃, 300 MHz) δ 9.80 (s, 1H), 7.91-7.87 (m, 2H), 7.58-7.50 (m, 3H), 7.44-7.35 (m, 2H), 7.21 (d, J = 8.4 Hz, 2H), 7.12 (d, J = 8.1 Hz, 2H), 6.99 (d, J = 7.2 Hz, 1H), 6.38 (s, 1H), 5.11 (d, J = 0.9 Hz, 1H), 5.06 (d, J = 1.2 Hz, 1H), 3.02 (s, 3H), 2.34 (s, 3H); 13 C NMR (CDCl₃, 75 MHz) δ 192.6, 161.3, 144.7, 142.5, 139.8, 135.5, 134.2, 133.8, 133.7, 132.8, 130.7, 129.7, 128.6, 127.7, 127.4, 126.8, 126.5, 125.1, 118.8, 73.7, 33.4, 21.6 ppm; HRMS (ESI⁺) m/z calculated for $C_{25}H_{23}N_2O_4S$ [M + H]⁺ 447.1378, found 447.1421.

3-Methyl-2-[1-(4-nitrophenyl)vinyl]-1-tosyl-2,3-dihydroquinazolin-4(1H)-one (9ai). Pale yellow amorphous solid (53.6 mg, 66% yield); mp 216–217 °C; ¹H NMR (CDCl₃, 300 MHz) δ 8.26 (d, I = 9 Hz, 2H, 7.90 - 7.87 (m, 1H), 7.56 - 7.50 (m, 3H), 7.42 - 7.37 (m, 1H)2H), 7.28–7.23 (m, 2H), 7.17 (d, *J* = 8.4 Hz, 2H), 6.49 (s, 1H), 5.47 $(d, J = 1.2 \text{ Hz}, 1\text{H}), 5.09 (d, J = 1.5 \text{ Hz}, 1\text{H}), 2.98 (s, 3\text{H}), 2.38 (s, 3\text$ 3H); 13 C NMR (CDCl₃, 75 MHz) δ 161.2, 147.5, 145.1, 143.9, 141.4, 134.9, 133.6, 132.9, 129.6, 127.9, 127.8, 127.7, 126.9, 126.8, 125.3, 123.7, 119.9, 73.5, 33.8, 21.6 ppm; HRMS (ESI⁺) m/z calculated for $C_{24}H_{21}N_3O_5SNa [M + Na]^+ \overline{486.1100}$, found 486.1096.

2-[1-(2,4-Dimethoxypyrimidin-5-yl)vinyl]-3-methyl-1-tosyl-2,3dihydroquinazolin-4(1H)-one (9aj). White amorphous solid (75.0 mg, 89% yield); mp 182–184 °C; 1 H NMR (CDCl₃, 300 MHz) δ 7.86 (d, J = 7.8 Hz, 1H), 7.61 (s, 1H), 7.54-7.49 (m, 1H), 7.41 (d, J = 8.1 (d))Hz, 1H), 7.38–7.33 (m, 1H), 7.26–7.21 (m, 2H), 7.13 (d, J = 8.1 Hz, 2H), 6.51 (s, 1H), 5.15 (d, J = 1.2 Hz, 1H), 5.01 (d, J = 1.8 Hz, 1H), 4.14 (s, 3H), 4.03 (s, 3H), 2.94 (s, 3H), 2.35 (s, 3H); ¹³C NMR (CDCl₃, 75 MHz) δ 167.8, 165.4, 161.3, 157.6, 144.7, 138.2, 135.3, 134.1, 132.8, 129.6, 127.6, 127.5, 126.9, 126.8, 125.4, 119.3, 113.1, 72.9, 54.9, 54.3, 33.8, 21.6 ppm; HRMS (EI⁺) m/z calculated for C₂₄H₂₄N₄O₅S [M]⁺ 480.1467, found 480.1469.

2-(Buta-1,3-dien-2-yl)-3-methyl-1-tosyl-2,3-dihydroquinazolin-4(1H)-one (9ak). Brown amorphous solid (25.2 mg, 39% yield); mp 152–154 °C; ¹H NMR (CDCl₃, 300 MHz) δ 7.83 (dd, J = 7.5 Hz, 1.5

Hz, 1H), 7.63 (d, J = 8.1 Hz, 1H), 7.53 (td, J = 7.6 Hz, 1.3 Hz, 1H), 7.35 (t, J = 7.2 Hz, 1H), 7.26 - 7.23 (m, 2H), 7.14 (d, J = 8.1 Hz, 2H), 6.28 (dd, J = 17.8 Hz, 11.2 Hz, 1H), 6.20 (s, 1H), 5.88 (d, J = 18 Hz, 11.2 Hz, 11.21H), 5.31 (d, J = 11.4 Hz, 1H), 5.17 (s, 1H), 4.69 (s, 1H), 2.82 (s, 3H), 2.36 (s, 3H); 13 C NMR (CDCl₃, 150 MHz) δ 163.0, 146.4, 140.6, 136.9, 135.4, 135.4, 134.2, 131.1, 129.2, 129.0, 128.9, 128.6, 127.1, 119.4, 118.9, 74.1, 34.9, 23.2 ppm; HRMS (EI⁺) m/z calculated for C₂₀H₂₀N₂O₃S [M]⁺ 368.1195, found 368.1185.

1,3-Dimethyl-5-[1-(3-methyl-4-oxo-1-tosyl-1,2,3,4-tetrahydroquinazolin-2-yl)vinyl]pyrimidine-2,4(1H,3H)-dione (9al). Brown amorphous solid (53.0 mg, 63% yield); mp 160-162 °C; ¹H NMR (CDCl₃, 300 MHz) δ 7.92 (d, J = 8.1 Hz, 1H), 7.62–7.54 (m, 2H) 7.42-7.36 (m, 1H), 7.32-7.28 (m, 2H), 7.18 (d, J = 8.1 Hz, 2H), 6.99(s, 1H), 6.58 (s, 1H), 5.28 (s, 1H), 4.93 (d, J = 1.2 Hz, 1H), 3.44-3.43(m, 6H), 2.88 (s, 3H), 2.38 (s, 3H); 13 C NMR (CDCl₃, 75 MHz) δ 161.9, 161.0, 151.2, 144.8, 140.8, 136.7, 135.2, 133.8, 132.6, 129.6, 127.6, 127.4, 126.9, 126.5, 124.9, 118.9, 110.9, 71.7, 37.0, 33.2, 27.9, 21.5 ppm; HRMS (ESI⁺) m/z calculated for $C_{24}H_{24}N_4O_5SNa$ [M + Na]+ 503.1365, found 503.1357.

3-Methyl-2-(1-phenylvinyl)-2,3-dihydroquinazolin-4(1H)-one (9ba). Yellow gum (41.0 mg, 74% yield from 14; 50.5 mg, 60% yield from 15); ¹H NMR (CDCl₃, 300 MHz) δ 7.93 (d, J = 7.5 Hz, 1H), 7.37 (brs, 5H), 7.28-7.22 (m, 1H), 6.88-6.83 (m, 1H), 6.46 (d, J =7.8 Hz, 1H), 5.59 (s, 1H), 5.43 (s, 1H), 5.24 (s, 1H), 4.30 (brs, 1H), 3.15 (s, 3H); 13 C NMR (CDCl₃, 75 MHz) δ 163.6, 144.8, 144.4, 137.9, 133.2, 128.8, 128.3, 127.1, 119.3, 116.2, 114.4, 73.4, 32.8 ppm (two carbon signals probably have merged with other aromatic carbons); HRMS (ESI⁺) m/z calculated for $C_{17}H_{17}N_2O$ [M + H]⁺ 265.1341, found 265.1371.

3-Methyl-2-{1-[4-(trifluoromethyl)phenyl]vinyl}-2,3-dihydroquinazolin-4(1H)-one (9bb). Dark brown amorphous solid (40.7 mg, 58% yield); mp 128–130 °C; 1 H NMR (CDCl₃, 300 MHz) δ 7.90 (dd, J = 7.8 Hz, 1.2 Hz, 1H), 7.60 (d, J = 8.1 Hz, 2H), 7.48 (d, J = 8.4)Hz, 2H), 7.28-7.22 (m, 1H), 6.87-6.82 (m, 1H), 6.47 (d, J = 7.8 Hz, 1H), 5.58 (d, J = 1.8 Hz, 1H), 5.48 (s, 1H), 5.34 (d, J = 0.6 Hz, 1H), 4.26 (brs, 1H), 3.11 (s, 3H); ¹³C NMR (CDCl₃, 75 MHz) δ 163.6, 144.6, 143.9, 141.6, 133.4, 130.2 (q, J = 32.2 Hz), 128.3, 127.7, 125.6 (q, J = 3.7 Hz), 123.9 (q, J = 270.5 Hz), 119.4, 118.1, 116.0, 114.5,73.6, 32.6 ppm; HRMS (EI⁺) m/z calculated for $C_{18}H_{15}F_3N_2O$ [M]⁺ 332.1136, found 332.1144.

2-[1-(4-Methoxyphenyl)vinyl]-3-methyl-2,3-dihydroquinazolin-4(1H)-one (9bc). Brown amorphous solid (31.1 mg, 50% yield from 14; 46.0 mg, 49% yield from 15); mp 113–115 °C; ¹H NMR (CDCl₃, 300 MHz) δ 7.91 (d, I = 6.3 Hz, 1H), 7.34–7.31 (m, 2H), 7.22 (dd, I= 7.6 Hz, 1.0 Hz, 1H), 6.90-6.80 (m, 3H), 6.45 (d, J = 8.1 Hz, 1H),5.56 (s, 1H), 5.38 (s, 1H), 5.16 (s, 1H), 4.27 (brs, 1H), 3.82 (s, 3H), 3.10 (s, 3H); 13 C NMR (CDCl₃, 75 MHz) δ 163.6, 159.6, 145.0, 143.7, 133.2, 130.0, 128.3, 128.2, 119.1, 116.0, 115.2, 114.4, 114.1, 73.7, 55.3, 32.5 ppm; HRMS (ESI⁺) m/z calculated for $C_{18}H_{18}$ $N_2O_2Na [M + Na]^+$ 317.1266, found 317.1258.

Methyl 4-[1-(3-Methyl-4-oxo-1,2,3,4-tetrahydroquinazolin-2-yl)vinyl]benzoate (9bd). Brown crystalline solid (56.5 mg, 83% yield from 14; 82.2 mg, 80% yield from 15); mp 156-158 °C; ¹H NMR (CDCl₃, 300 MHz) δ 8.03 (d, J = 8.1 Hz, 2H), 7.92 (d, J = 7.8 Hz, 1H), 7.45 (d, J = 8.1 Hz, 2H), 7.28–7.23 (m, 1H), 6.86 (t, J = 7.5 Hz, 1H), 6.46 (d, J = 7.8 Hz, 1H), 5.61 (s, 1H), 5.51 (s, 1H), 5.33 (s, 1H), 4.28 (brs, 1H), 3.95 (s, 3H), 3.13 (s, 3H); ¹³C NMR (CDCl₃, 75 MHz) δ 166.6, 163.5, 144.7, 144.1, 142.6, 133.3, 129.9, 129.8, 128.2, 127.2, 119.3, 117.7, 116.1, 114.5, 73.4, 52.2, 32.7 ppm; HRMS (EI⁺) m/z calculated for C₁₉H₁₈N₂O₃ [M]⁺ 322.1317, found 322.1321.

2-[1-(2,4-Dimethoxypyrimidin-5-yl)vinyl]-3-methyl-2,3-dihydroquinazolin-4(1H)-one (9be). Brown amorphous solid (57.2 mg, 55% yield); mp 150–152 °C; 1 H NMR (CDCl₃, 300 MHz) δ 7.91 (s, 1H), 7.85 (dd, J = 7.6 Hz, 1.3 Hz, 1H), 7.25–7.19 (m, 1H), 6.83–6.78 (m, 1H), 6.47–6.44 (m, 1H), 5.48 (s, 1H), 5.31 (s, 1H), 5.23 (s, 1H), 4.38 (brs, 1H), 4.01 (s, 3H), 3.98 (s, 3H), 3.15 (s, 3H); ¹³C NMR (CDCl₃, 75 MHz) δ 168.9, 166.2, 164.1, 159.2, 145.3, 140.4, 134.2, 129.1, 120.4, 119.4, 117.5, 115.4, 114.3, 73.2, 55.9, 55.3, 34.1 ppm; HRMS (ESI⁺) m/z calculated for $C_{17}H_{18}N_4O_3Na$ [M + Na]⁺ 349.1277, found 349.1257.

General Procedure for One-Pot Bis-heteroannulation (Scheme 2). To a stirred solution of diiodo compound 16a (38.6 mg, 0.12 mmol) [or 16b (39.3 mg, 0.12 mmol)] in dry DMF (3 mL) were added K₂CO₃ (129.1 mg, 0.94 mmol) and allenamide 13 (80.0 mg, 0.23 mmol) sequentially. After stirring for a few minutes, Pd(OAc)₂ (2.6 mg, 0.01 mmol) and PPh₃ (7.4 mg, 0.03 mmol) were added, and the whole reaction mixture was allowed to stir at rt under an argon atmosphere for the requisite time (see Scheme 2). After completion of the reaction (TLC), DMF was evaporated under reduced pressure and the resulting mixture was extracted with ethyl acetate (3×20 mL). The combined organic extracts were washed with water (15 mL), dried over Na2SO4, and concentrated in vacuo. The crude residue was purified by silica gel (100-200 mesh) column chromatography using 40% ethyl acetate-petroleum ether (v/v) to afford the desired product 17a (or 17b).

2,2'-[1,4-Phenylenebis(ethene-1,1-diyl)]bis[3-methyl-1-tosyl-2,3dihydroquinazolin-4(1H)-one] (17a). White crystalline solid (44.3 mg, 50% yield); mp 264–265 °C; 1 H NMR (CDCl₃, 300 MHz) δ 7.86 (d, J = 8.1 Hz, 2H), 7.56-7.51 (m, 2H), 7.47 (d, J = 7.2 Hz, 2H),7.41-7.34 (m, 6H), 7.27-7.25 (m, 4H), 7.15 (d, J = 8.1 Hz, 4H), 6.49(s, 2H), 5.40 (s, 2H), 4.93 (s, 2H), 2.96 (s, 6H), 2.37 (s, 6H); ¹³C NMR (CDCl₃, 75 MHz) δ 161.5, 144.7, 142.2, 137.2, 135.3, 133.9, 132.8, 129.5, 127.6, 127.5, 127.2, 126.9, 125.6, 117.3, 73.7, 33.8, 21.6 ppm; HRMS (ESI⁺) m/z calculated for $C_{42}H_{39}N_4O_6S_2$ [M + H]⁺ 759.2311, found 759.2304.

2,2'-[Thiophene-2,5-diylbis(ethene-1,1-diyl)]bis[3-methyl-1-tosyl-2,3-dihydroquinazolin-4(1H)-one] (17b). Brown amorphous solid (49.2 mg, 55% yield); mp 236–238 °C; ¹H NMR (CDCl₃, 300 MHz) δ 7.89 (d, J = 7.5 Hz, 2H), 7.54–7.49 (m, 8H), 7.39–7.34 (m, 4H), 7.29-7.26 (m, 4H), 6.65 (s, 2H), 4.79 (s, 2H), 4.69 (d, I = 1.2 Hz, 2H), 2.89 (s, 6H), 2.41 (s, 6H); 13 C NMR (CDCl₃, 75 MHz) δ 161.4, 144.9, 139.6, 135.2, 134.8, 133.6, 132.8, 129.5, 127.8, 127.5, 127.2, 126.7, 125.6, 116.2, 73.5, 33.5, 21.6 ppm; HRMS (ESI⁺) m/zcalculated for $C_{40}H_{36}N_4O_6S_3Na [M + Na]^+ 787.1695$, found 787.1693.

Synthesis of Internal Alkyne 18: Reaction of Phenyl Substitute Allenamide 19 with Iodo Compound 11g (Scheme 3). A mixture of iodobenzene (63 mg, 0.31 mmol), Pd(PPh₃)₂Cl₂ (4 mg, 0.006 mmol), CuI (2 mg, 0.012 mmol), and Et₃N (74 mg, 0.73 mmol) in dry DMF (3.0 mL) was stirred under an argon atmosphere for a few minutes. To a well-stirred solution of this reaction mixture was added terminal alkyne 10 (100 mg, 0.29 mmol) slowly, and stirring was continued for 1 h. After completion of reaction (TLC), the solvent was evaporated under reduced pressure and the resulting residue was extracted with EtOAc (3 × 20 mL). The combined organic extracts were washed with water (10 mL), dried over Na₂SO₄, and concentrated in vacuo. The crude residue was then purified by silica gel (100-200 mesh) column chromatography using 17% ethyl acetate-petroleum ether (v/v) as eluent to afford the internal alkyne

N-Methyl-2-[(4-methylphenyl)sulfonamide]-N-(3-phenylprop-2yn-1-yl)benzamide (18). Brown amorphous solid (93 mg, 76%); mp = 93–95 °C; ¹H NMR (CDCl₃, 300 MHz) δ 8.47 (s, 1H), 7.71–7.62 (m, 3H), 7.47–7.34 (m, 6H), 7.26–7.14 (m, 4H), 4.45 (brs, 1H), 3.65 (brs, 1H), 3.05-2.67 (m, 3H), 2.34 (brs, 3H); ¹³C NMR (CDCl₃, 75 MHz) δ 169.0, 143.5, 136.4, 136.3, 135.9, 131.7, 131.1, 129.4, 128.7, 128.3, 127.1, 125.9, 124.5, 124.3, 83.0, 82.9, 40.1, 36.7, 32.8, 21.3 ppm. HRMS (ESI⁺) m/z calculated for $C_{24}H_{22}N_2O_3SNa$ [M + Na]⁺ 441.1249, found 441.1244.

KO^tBu (47 mg, 0.42 mmol) was added to the solution of acetylene 18 (90 mg, 0.21 mmol) in dry DMF (5 mL), and the stirring was continued at rt under an argon atmosphere. The reaction was found to be completed within 15 min (TLC) and quenched with water (3 mL). It was then extracted with EtOAc (3 × 15 mL), and the combined organic extracts were washed with water (10 mL), dried over anhydrous Na2SO4, and concentrated in vacuo. Without further purification, the resulting crude compound obtained was used directly for the next palladium-catalyzed cyclization. To a stirred solution of Pd(OAc)₂ (2 mg, 0.01 mmol), and PPh₃ (7 mg, 0.02 mmol) in dry DMF were added 4-iodomethylbenzoate 11g (55 mg, 0.21 mmol), K₂CO₃ (116 mg, 0.84 mmol), and allene 19 (0.21 mmol) sequentially

at room temperature under an argon atmosphere, and the whole reaction mixture was allowed to stir at room temperature for 5 h. After completion of reaction (TLC), DMF was evaporated under reduced pressure and the resulting mixture was extracted with EtOAc (3 \times 20 mL). The combined organic extracts were washed with water, dried over Na₂SO₄, and concentrated *in vacuo*. The crude residue was purified by silica gel (100–200 mesh) column chromatography using 17% ethyl acetate–petroleum ether (v/v) as eluent to afford the desired product 20.

Methyl (*E*)-4-[1-(3-Methyl-4-oxo-1-tosyl-1,2,3,4-tetrahydro-quinazolin-2-yl)-2-phenylvinyl]benzoate (**20**). Brown amorphous solid (65.4 mg, 55% yield); mp 180–181 °C; ¹H NMR (CDCl₃, 600 MHz) δ 8.02 (dd, J = 6.6 Hz, 1.8 Hz, 2H), 7.89 (dd, J = 7.8 Hz, 1.2 Hz, 1H), 7.53–7.50 (m, 2H), 7.36–7.34 (m, 1H), 7.23 (dd, J = 6.9 Hz, 1.5 Hz, 2H), 7.14–7.11 (m, 4H), 7.06–7.05 (m, 1H), 7.02–6.99 (m, 2H), 6.73–6.71 (m, 2H), 6.28 (d, J = 1.8 Hz, 1H), 6.27 (brs, 1H), 3.94 (s, 3H), 2.99 (s, 3H), 2.35 (s, 3H); ¹³C NMR (CDCl₃, 150 MHz) δ 168.4, 162.9, 146.3, 143.2, 137.1, 136.0, 135.9, 135.7, 134.4, 132.1, 131.7, 131.2, 131.2, 130.9, 130.7, 129.6, 129.5, 129.3, 129.1, 128.5, 128.4, 126.8, 77.8, 53.7, 35.5, 23.2 ppm; HRMS (FAB+) m/z calculated for C₃₂H₂₉N₂O₅S [M + H]+ 553.1797, found 553.1804.

General Procedure for Synthesis of 3- α -Styryl-3,4-dihydro-1,2,4-benzothiadiazine-1,1-dioxide 23 (Table 3). Synthesis of Starting Acetylene Compound 21. The acetylene was synthesized in three steps from commercially available o-nitrobenzenesulfonyl chloride

Synthesis of N-Methyl-2-nitro-N-(prop-2-yn-1-yl)benzenesulfonamide (S3). To a stirred solution of o-nitrobenzenesulfonyl chloride (1 g, 4.51 mmol) and Et₃N (548 mg, 5.41 mmol) in dry dichloromethene (10 mL) was added N-methylpropargylamine (468 mg, 6.77 mmol) at 0 °C, and the whole reaction mixture was allowed to stir at room temperature for 2 h until completion (TLC). The solvent was evaporated under reduced pressure, and the resulting mixture was extracted with dichloromethane (3 × 20 mL). The combined organic extract was washed with water (15 mL), dried over anhydrous Na2SO4, and concentrated in vacuo. The crude product was purified by silica gel (100-200 mesh) column chromatography using 15% ethyl acetate-petroleum ether (v/v) as eluent to afford the product S3 as a white solid (848 mg, 74%). The spectral data of the product were in accordance with the reported²⁶ data. ¹H NMR (CDCl₃, 300 MHz) δ 8.05–8.02 (m, 1H), 7.73–7.69 (m, 2H), 7.65– 7.62 (m, 1H), 4.16 (d, J = 2.4 Hz, 2H), 3.01 (s, 3H), 2.21 (t, J = 2.5Hz, 1H); 13 C NMR (CDCl₃, 75 MHz) δ 148.2, 133.9, 131.6, 131.4, 130.8, 124.0, 76.4, 74.1, 39.6, 34.3 ppm. MS (ESI⁺) m/z 277.03 [M + Na]⁺.

2-Amino-N-methyl-N-(prop-2-yn-1-yl)benzenesulfonamide (**S4**). Saturated aqueous NH₄Cl (1.15 g, 21.50 mmol) solution was added to the solution of S3 (840 mg, 3.31 mmol) in MeOH (10 mL). Thereafter, activated Zn (1.075 g, 16.53 mmol) was added to the mixture at room temperature, and the whole reaction mixture was stirred for 2.5 h under an argon atmosphere. After completion of the reaction (TLC), the reaction mixture was then filtered through a sintered funnel and the filtrate was concentrated in vacuo. The resulting mixture obtained was diluted with 2N NaOH solution (8 mL) and extracted with EtOAc (3 × 20 mL). The combined organic extracts were washed with water (10 mL), dried over anhydrous Na₂SO₄, and concentrated in vacuo. The crude material was purified by silica gel (100-200 mesh) column chromatography using 15% ethyl acetate-petroleum ether (v/v) as eluent to afford pure S4 as a yellow liquid (422 mg, 57%); ¹H NMR (CDCl₃, 300 MHz) δ 7.60 (dd, J =8.1 Hz, 1.5 Hz, 1H), 7.33–7.26 (m, 1H), 6.76 (d, *J* = 6.9 Hz, 1H), 6.71 (d, J = 8.1 Hz, 1H), 5.05 (brs, 1H), 4.02 (d, J = 2.7 Hz, 2H), 2.87 (s, J = 2.7 Hz, 2Hz, 2H), 2.87 (s, J = 2.7 Hz, 2Hz, 2Hz), 2.87 (s, J = 2.7 Hz, 2 Hz, 2Hz), 2.87 (s, J = 2.7 Hz, 2 Hz), 2.87 (s, J = 2.7 Hz, 2 Hz), 2.87 (s, J =3H), 2.16 (t, J = 2.5 Hz, 1H); ¹³C NMR (CDCl₃, 75 MHz) δ 146.3, 134.3, 130.3, 118.4, 117.7, 117.2, 73.7, 73.7, 39.6, 34.3 ppm. HRMS (ESI⁺) m/z calculated for $C_{10}H_{12}N_2O_2SNa$ [M + Na]⁺ 247.0517, found 247.0505.

N-Methyl-2-[(4-methylphenyl)sulfonamide]-N-(prop-2-yn-1-yl)-benzenesulfonamide (21). To a stirred and cooled (0 °C) solution of S4 (400 mg, 1.79 mmol) in dry dichloromethane were added pyridine (169 mg, 2.14 mmol) and TsCl (349 mg, 1.84 mmol) sequentially,

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and the whole reaction mixture was allowed to stir at rt for overnight. The solvent was evaporated under reduced pressure, and the mixture obtained was extracted with dichloromethane. The crude residue obtained was purified by silica gel (100–200 mesh) column chromatography using 15% ethyl acetate—petroleum ether (v/v) as eluent to afford **21** as a white amorphous solid (560 mg, 83%); mp 92–94 °C; ¹H NMR (CDCl₃, 300 MHz) δ 9.23 (s, 1H), 7.86 (d, J = 8.4 Hz, 2H), 7.76 (dd, J = 8.1 Hz, 1.5 Hz, 1H), 7.71 (d, J = 8.1 Hz, 1H), 7.54–7.48 (m, 1H), 7.35–7.31 (m, 2H), 7.19–7.14 (m, 1H), 4.03 (d, J = 2.1 Hz, 2H), 2.88 (s, 3H), 2.45 (s, 3H), 2.19 (t, J = 2.5 Hz, 1H); ¹³C NMR (CDCl₃, 75 MHz) δ 144.5, 136.5, 136.2, 134.6, 130.3, 129.9, 127.3, 123.9, 123.3, 119.1, 75.6, 74.6, 39.6, 34.3, 21.6 ppm. HRMS (EI⁺) m/z calculated for C₁₇H₁₈N₂O₄S₂ [M]⁺ 378.0708, found 378.0711.

Generation of Allenesulphonamide 22 from Acetylene 21 and Subsequent Reactions with lodo Compounds 11 To Synthesize $3-(\alpha-Styryl)-3,4-dihydro-1,2,4-benzothiadiazine-$ **1,1-dioxides 23.** KO^tBu (35.6 mg, 0.32 mmol) was added to the solution of acetylene 21 (60.0 mg, 0.16 mmol) in dry DMF at rt under an argon atmosphere; the reaction was found to be complete within 2 min (TLC). The reaction was quenched with water (5 mL) and extracted with ethyl acetate. The combined organic extracts were dried over anhydrous Na₂SO₄ and concentrated in vacuo. The crude product obtained was used directly in the next step without further purification. To a stirred solution of Pd(OAc)₂ (1.8 mg, 0.008 mmol) and PPh₃ (5.0 mg, 0.02 mmol) in dry DMF (5 mL) were added iodo compound 11 (0.16 mmol), K₂CO₃ (87.6 mg, 0.64 mmol), and crude allenesulphonamide 22 (0.16 mmol) sequentially, and the whole reaction mixture was heated at 60 °C for the requisite time (see Table 3) under an argon atmosphere. After completion of reaction (TLC), DMF was evaporated under reduced pressure and the resulting residue was extracted with EtOAc (3 × 20 mL). The combined organic extracts were washed with water (15 mL), dried over anhydrous Na₂SO₄, and concentrated in vacuo. The crude mixture was purified by silica gel (100-200 mesh) column chromatography using 9-40% ethyl acetate-petroleum ether (v/v) as eluent to afford the pure product 23.

2-Methyl-3-(1-phenylvinyl)-4-tosyl-3,4-dihydro-2H-benzo[e]-[1,2,4]thiadiazine-1,1-dioxide (23a). White amorphous solid (48.3 mg, 67% yield); mp 154–155 °C; 1 H NMR (CDCl₃, 300 MHz) δ 7.77 (dd, J = 8.4 Hz, 0.9 Hz, 1H), 7.60 (dd, J = 7.6 Hz, 1.6 Hz, 1H), 7.50–7.41 (m, 4H), 7.33–7.26 (m, 3H), 7.23–7.18 (m, 2H), 7.10 (d, J = 8.1 Hz, 2H), 6.60 (d, J = 1.2 Hz, 1H), 5.62 (d, J = 0.9 Hz, 1H), 5.58 (d, J = 1.5 Hz, 1H), 2.56 (s, 3H), 2.28 (s, 3H); 13 C NMR (CDCl₃, 75 MHz) δ 144.9, 143.7, 137.8, 135.5, 134.5, 133.5, 129.9, 129.6, 128.7, 128.6, 128.2, 126.7, 126.0, 125.9, 125.7, 118.9, 77.3, 34.1, 21.7 ppm; HRMS (EI $^+$) m/z calculated for C₂₃H₂₂N₂O₄S₂ [M] $^+$ 454.1021, found 454.1016.

2-Methyl-3-[1-(pyridin-3-yl)vinyl]-4-tosyl-3,4-dihydro-2H-benzo-[e][1,2,4]thiadiazine 1,1-dioxide (23b). White amorphous solid (44.8 mg, 62% yield); mp 162–164 °C; $^1\mathrm{H}$ NMR (CDCl_3, 300 MHz) δ 8.71 (d, J=1.8 Hz, 1H), 8.60–8.59 (m, 1H), 7.83 (d, J=8.1 Hz, 2H), 7.70 (d, J=7.8 Hz, 1H), 7.59–7.52 (m, 3H), 7.34–7.26 (m, 2H), 7.22 (d, J=8.1 Hz, 2H), 6.64 (s, 1H), 5.73–5.72 (m, 2H), 2.71 (s, 3H), 2.38 (s, 3H); $^{13}\mathrm{C}$ NMR (CDCl_3, 75 MHz) δ 149.5, 147.8, 145.2, 140.8, 134.9, 134.3, 134.2, 133.6, 133.5, 129.6, 129.2, 128.1, 126.0, 125.9, 125.0, 123.4, 120.4, 76.9, 34.4, 21.7 ppm; HRMS (ESI+) m/z calculated for $\mathrm{C}_{22}\mathrm{H}_{21}\mathrm{N}_3\mathrm{O}_4\mathrm{S}_2\mathrm{Na}$ [M + Na]+ 478.0871, found 478.0866.

2-Methyl-3-[1-(thiophen-2-yl)vinyl]-4-tosyl-3,4-dihydro-2H-benzo[e][1,2,4]thiadiazine-1,1-dioxide (23c). Yellow amorphous solid (29.2 mg, 40% yield); mp 107–108 °C; ¹H NMR (CDCl₃, 600 MHz) δ 7.93 (d, J = 7.8 Hz, 1H), 7.68 (dd, J = 7.8 Hz, 1.2 Hz, 1H), 7.59–7.56 (m, 1H), 7.53 (d, J = 8.4 Hz, 2H), 7.32–7.26 (m, 3H), 7.18 (d, J = 8.4 Hz, 2H), 7.04–7.03 (m, 1H), 6.58 (s, 1H), 5.84 (s, 1H), 5.64 (s, 1H), 2.60 (s, 3H), 2.37 (s, 3H); ¹³C NMR (CDCl₃, 150 MHz) δ 144.9, 140.1, 137.2, 135.4, 133.9, 133.6, 129.8, 129.4, 128.2, 127.6, 126.0, 125.9, 125.9, 125.5, 125.1, 116.9, 76.9, 33.4, 21.6 ppm; HRMS (EI⁺) m/z calculated for C₂₁H₂₀N₂O₄S₃[M]⁺ 460.0585, found 460.0578.

2-Methyl-3-[1-(p-tolyl)vinyl]-4-tosyl-3,4-dihydro-2H-benzo[e]-[1,2,4]thiadiazine-1,1-dioxide (23d). White amorphous solid (46.8 mg, 63% yield); mp 201–202 °C; 1 H NMR (CDCl₃, 600 MHz) δ 7.88 (d, J = 8.4 Hz, 1H), 7.69 (dd, J = 7.8 Hz, 1.8 Hz, 1H), 7.57–7.53 (m, 3H), 7.42 (d, J = 8.4 Hz, 2H), 7.29 (t, J = 7.5 Hz, 1H), 7.20–7.18 (m, 4H), 6.68 (s, 1H), 5.69 (s, 1H), 5.61 (d, J = 1.2 Hz, 1H), 2.63 (s, 3H), 2.38 (s, 3H), 2.37 (s, 3H); 13 C NMR (CDCl₃, 150 MHz) δ 144.8, 143.3, 138.4, 135.5, 134.7, 134.4, 133.4, 129.6, 129.4, 129.3, 128.1, 126.4, 125.9, 125.8, 125.5, 118.0, 77.1, 33.9, 21.6, 21.2 ppm; HRMS (EI $^+$) m/z calculated for C₂₄H₂₄N₂O₄S₂ [M] $^+$ 468.1177, found 468.1172.

2-Methyl-4-tosyl-3-{1-[4-(trifluoromethyl)phenyl]vinyl}-3,4-dihydro-2H-benzo[e][1,2,4]thiadiazine-1,1-dioxide (23e). White amorphous solid (72.9 mg, 88% yield); mp 71–72 °C; ¹H NMR (CDCl₃, 300 MHz) δ 7.84 (d, J = 8.1 Hz, 1H), 7.70 (dd, J = 7.8 Hz, 1.5 Hz, 1H), 7.66–7.52 (m, 7H), 7.33–7.26 (m, 1H), 7.21 (d, J = 8.1 Hz, 2H), 6.67 (s, 1H), 5.75–5.73 (m, 2H), 2.68 (s, 3H), 2.38 (s, 3H); 13 C NMR (CDCl₃, 75 MHz) δ 145.1, 142.6, 141.2, 135.1, 134.2, 133.6, 130.4 (q, J = 32.0 Hz), 129.6, 129.2, 128.0, 127.0, 126.0, 125.9, 125.5 (q, J = 3.7 Hz), 125.1, 120.5, 76.9, 34.3, 21.6 ppm; HRMS (ESI⁺) m/z calculated for $C_{24}H_{21}F_3N_2O_4S_2Na$ [M + Na]⁺ 545.0792, found 545.0797.

3-[1-(4-Methoxyphenyl)vinyl]-2-methyl-4-tosyl-3,4-dihydro-2H-benzo[e][1,2,4]thiadiazine-1,1-dioxide (23f). Pale yellow solid (30.0 mg, 39% yield); mp 152–154 °C; ¹H NMR (CDCl₃, 300 MHz) δ 7.88 (d, J = 8.4 Hz, 1H), 7.68 (d, J = 7.8 Hz, 1H), 7.55 (d, J = 8.1 Hz, 2H), 7.47 (d, J = 8.7 Hz, 2H), 7.30 (d, J = 7.5 Hz, 2H), 7.19 (d, J = 8.1 Hz, 2H), 6.91 (d, J = 8.7 Hz, 2H), 6.65 (s, 1H), 5.66 (s, 1H), 5.59 (s, 1H), 3.84 (s, 3H), 2.62 (s, 3H), 2.37 (s, 3H); 13 C NMR (CDCl₃, 75 MHz) δ 159.8, 144.8, 142.8, 135.5, 134.3, 133.4, 133.1, 129.8, 129.4, 128.1, 127.7, 127.6, 125.9, 125.6, 117.3, 113.9, 77.2, 55.3, 33.8, 21.6 ppm; HRMS (EI⁺) m/z calculated for C₂₄H₂₄N₂O₃S₂ [M]⁺ 484.1127, found 484.1135.

Methyl 4-[1-(2-Methyl-1,1-dioxido-4-tosyl-3,4-dihydro-2H-benzo-[e][1,2,4]thiadiazin-3-yl)vinyl]benzoate (*23g*). Brown crystalline solid (42.2 mg, 52% yield); mp 142–144 °C; ¹H NMR (CDCl₃, 300 MHz) δ 8.04 (d, J = 8.4 Hz, 2H), 7.84 (d, J = 8.1 Hz, 1H), 7.68 (dd, J = 7.6 Hz, 1.3 Hz, 1H), 7.57–7.51 (m, 5H), 7.32–7.26 (m, 1H), 7.20 (d, J = 8.1 Hz, 2H), 6.69 (s, 1H), 5.78 (s, 1H), 5.73 (d, J = 1.2 Hz, 1H), 3.94 (s, 3H), 2.66 (s, 3H), 2.37 (s, 3H); ¹³C NMR (CDCl₃, 75 MHz) δ 166.7, 145.1, 142.9, 142.1, 135.1, 134.2, 133.6, 130.0, 129.9, 129.6, 129.4, 128.1, 126.7, 126.0, 125.9, 125.2, 120.4, 76.9, 52.2, 34.3, 21,7 ppm; HRMS (ESI⁺) m/z calculated for C₂₈H₂₄N₂O₆S₂Na [M + Na]⁺ 535.0973, found 535.0981.

3-[1-(4-Fluorophenyl)vinyl]-2-methyl-4-tosyl-3,4-dihydro-2H-benzo[e][1,2,4]thiadiazine-1,1-dioxide (23m). White amorphous solid (47.0 mg, 63% yield); mp 172–174 °C; ¹H NMR (CDCl₃, 300 MHz) δ 7.85 (d, J = 8.4 Hz, 1H), 7.69 (dd, J = 7.8 Hz, 1.5 Hz, 1H), 7.57–7.46 (m, SH), 7.32–7.26 (m, 1H), 7.19 (d, J = 8.1 Hz, 2H), 7.09–7.04 (m, 2H), 6.61 (s, 1H), 5.66–5.65 (m, 2H), 2.65 (s, 3H), 2.37 (s, 3H); ¹³C NMR (CDCl₃, 150 MHz) δ 162.8 (d, J = 247.5 Hz), 144.9, 142.6, 135.3, 134.3, 133.7 (d, J = 3 Hz), 133.5, 129.5, 128.4 (d, J = 7.5 Hz), 128.1, 125.9 (d, J = 15 Hz), 125.4, 118.9, 115.6, 115.4, 77.2, 33.9, 21.6 ppm (one carbon signal has probably merged with other aromatic carbon signals); HRMS (ESI⁺) m/z calculated for $C_{23}H_{21}FN_2O_4S_2Na$ [M + Na]⁺ 495.0825, found 495.0832.

3-[1-(4-Bromophenyl)vinyl]-2-methyl-4-tosyl-3,4-dihydro-2H-benzo[e][1,2,4]thiadiazine-1,1-dioxide (23n). White amorphous solid (51.5 mg, 61% yield); mp 206–208 °C; ¹H NMR (CDCl₃, 300 MHz) δ 7.84 (d, J = 8.4 Hz, 1H), 7.68 (dd, J = 7.6 Hz, 1.0 Hz, 1H),7.56–7.49 (m, 4H), 7.37 (d, J = 8.4 Hz, 2H), 7.32–7.26 (m, 2H), 7.19 (d, J = 8.1 Hz, 2H), 6.61 (s, 1H), 5.69–5.66 (m, 2H), 2.65 (s, 3H), 2.37 (s, 3H); 13 C NMR (CDCl₃, 100 MHz) δ 145.2, 142.9, 136.8, 135.4, 134.5, 133.6, 131.9, 129.8, 129.7, 128.5, 128.3, 126.1, 126.0, 125.5, 122.8, 119.5, 77.4, 34.2, 21.8 ppm; HRMS (ESI*) m/z calculated for $C_{23}H_{21}BrN_2O_4S_2Na$ [M + Na]* 555.0024, found 555.0026.

Synthesis of 2-(α-Styryl)-quinazolinones 7 and 3-(α-Styryl)-1,2,4-benzothiadiazine-1,1-dioxide 8 (Scheme 4). *Typical Procedure for the Synthesis of 3-Methyl-2-(1-phenylvinyl)-quinazolin-4(3H)-one (7a)*. To a well-stirred solution of dihydro-quinazolinone 9aa (60.0 mg, 0.14 mmol) in dry DMF (2 mL) was

added NaH (60% oil suspension) (28.0 mg, 0.70 mmol), and the whole reaction mixture was allowed to stir at rt under an argon atmosphere for 0.5 h. After complete consumption of the starting material (TLC), the reaction was quenched with water (5 mL) and extracted with ethyl acetate (3 \times 20 mL). The combined organic extracts were dried over anhydrous Na $_2$ SO $_4$ and concentrated in vacuo. The crude residue was purified by silica gel (100–200 mesh) column chromatography using 13–80% ethyl acetate–petroleum ether (v/v) to afford the pure product 7a.

3-Methyl-2-(1-phenylvinyl)quinazolin-4(3H)-one (7a). White amorphous solid (33.1 mg, 88% yield); mp 106–108 °C; ¹H NMR (CDCl₃, 600 MHz) δ 8.34–8.33 (m, 1H), 7.81–7.77 (m, 2H), 7.54–7.51 (m, 1H), 7.37–7.35 (m, 5H), 6.05 (s, 1H), 5.71 (s, 1H), 3.36 (s, 3H); ¹³C NMR (CDCl₃, 150 MHz) δ 162.4, 155.7, 147.4, 144.1, 135.9, 134.3, 129.1, 129.1, 127.6, 127.1, 126.7, 125.8, 120.8, 118.2, 32.6 ppm; HRMS (ESI⁺) m/z calculated for C₁₇H₁₅N₂O [M + H]⁺263.1184, found 263.1159.

3-Methyl-2-[1-(pyridin-3-yl)vinyl]quinazolin-4(3H)-one (7b). Yellow gum (27.8 mg, 74% yield); 1 H NMR (CDCl $_3$, 600 MHz) δ 8.73 (d, J = 1.8 Hz, 1H), 8.61 (d, J = 4.2 Hz, 1H), 8.34 (d, J = 7.8 Hz, 1H), 7.81–7.77 (m, 2H), 7.66 (d, J = 7.8 Hz, 1H), 7.56–7.53 (m, 1H), 7.32–7.30 (m, 1H), 6.13 (s, 1H), 5.83 (s, 1H), 3.39 (s, 3H); 13 C NMR (CDCl $_3$, 150 MHz) δ 162.2, 154.5, 150.1, 147.3, 147.1, 141.2, 134.5, 133.1, 131.9, 127.6, 127.4, 126.8, 123.7, 120.8, 120.1, 32.7 ppm; HRMS (ESI $^+$) m/z calculated for C $_{16}$ H $_{13}$ N $_3$ ONa [M + Na] $^+$ 286.0956, found 286.0957.

3-Methyl-2-[1-(thiophen-2-yl)vinyl]quinazolin-4(3H)-one (7c). Brown gum (33.0 mg, 87% yield); 1 H NMR (CDCl₃, 600 MHz) δ 8.34 (d, J = 8.4 Hz, 1H), 7.79–7.77 (m, 2H), 7.55–7.52 (m, 1H), 7.30 (d, J = 5.4 Hz, 1H), 6.98–6.96 (m, 1H), 6.82 (d, J = 3.0 Hz, 1H), 5.95 (s, 1H), 5.51 (s, 1H), 3.51 (s, 3H); 13 C NMR (CDCl₃, 150 MHz) δ 162.3, 154.6, 147.3, 140.1, 137.9, 134.4, 127.9, 127.7, 127.3, 126.7, 126.5, 126.3, 120.9, 116.2, 32.7 ppm; HRMS (ESI $^+$) m/z calculated for C₁₅H₁₃N₂OS [M + H] $^+$ 269.0749, found 269.0734.

3-Methyl-2-[1-(p-tolyl)vinyl]quinazolin-4(3H)-one (7d). Pale yellow crystalline solid (29.9 mg, 78% yield); mp 124–126 °C; ¹H NMR (CDCl₃, 600 MHz) δ 8.33 (d, J=7.8 Hz, 1H), 7.80–7.77 (m, 2H), 7.54–7.51 (m, 1H), 7.26–7.24 (m, 2H), 7.16 (d, J=7.8 Hz, 2H), 5.99 (s, 1H), 5.65 (s, 1H), 3.36 (s, 3H), 2.35 (s, 3H); 13 C NMR (CDCl₃, 150 MHz) δ 162.4, 155.9, 147.4, 143.9, 139.1, 134.3, 133.1, 129.8, 127.6, 127.1, 126.7, 125.7, 120.8, 117.2, 32.6, 21.2 ppm; HRMS (ESI⁺) m/z calculated for $C_{18}H_{16}N_2ONa$ [M + Na]⁺ 299.1160, found 299.1161.

3-Methyl-2-{1-[4-(trifluoromethyl)phenyl]vinyl}quinazolin-4(3H)-one (**7e**). Pale yellow amorphous solid (21.0 mg, 52% yield); mp 184–186 °C; ¹H NMR (CDCl₃, 300 MHz) δ 8.35 (d, J = 7.8 Hz, 1H), 7.81–7.80 (m, 2H), 7.64 (d, J = 8.4 Hz, 2H), 7.58–7.49 (m, 3H), 6.15 (s, 1H), 5.85 (s, 1H), 3.37 (s, 3H); 13 C NMR (CDCl₃, 75 MHz) δ 162.2, 154.8, 147.2, 142.9, 139.4, 134.5, 131.0 (q, J = 32.5 Hz), 127.6, 127.4, 126.8, 126.2–126.1 (m), 123.8 (q, J = 270.5 Hz), 120.8, 120.5, 32.6 ppm; HRMS (ESI⁺) m/z calculated for $C_{18}H_{14}F_3N_2O$ [M + H]⁺ 331.1058, found 331.1054.

2-[1-(4-Methoxyphenyl)vinyl]-3-methylquinazolin-4(3H)-one (7f). Yellow amorphous solid (33.2 mg, 85% yield); mp 156–157 °C; 1 H NMR (CDCl₃, 600 MHz) δ 8.33 (d, J = 7.8 Hz, 1H), 7.80–7.78 (m, 2H), 7.54–7.51 (m, 1H), 7.30–7.28 (m, 2H), 6.89–6.87 (m, 2H), 5.93 (s, 1H), 5.58 (s, 1H), 3.81 (s, 3H), 3.38 (s, 3H); 13 C NMR (CDCl₃, 150 MHz) δ 162.4, 160.2, 155.9, 147.4, 143.5, 134.3, 128.5, 127.6, 127.2, 127.1, 126.7, 120.8, 116.1, 114.4, 55.3, 32.6 ppm; HRMS (ESI $^+$) m/z calculated for C₁₈H₁₆N₂O₂Na [M + Na] $^+$ 315.1109, found 315.1137.

Typical Procedure for the Synthesis of 2-Methyl-3-(1-phenylvinyl)-2*H*-benzo[e][1,2,4]thiadiazine-1,1-dioxide (8a). To a well-stirred solution of 1,2,4-benzothiadiazine-1,1-dioxides 23a (60.0 mg, 0.13 mmol) in dry DMF (5.0 mL) was added NaH (60% oil suspension) (10.4 mg, 0.26 mmol), and the whole reaction mixture was allowed to stir at rt for 1 h under an argon atmosphere. On complete conversion of starting material (TLC), the reaction was quenched with water (5 mL) and extracted with EtOAc (3 × 10 mL). The combined organic extracts were dried over anhydrous Na₂SO₄

and concentrated *in vacuo*. The resulting crude residue was purified by silica gel (100-200 mesh) column chromatography using 10-50% ethyl acetate—petroleum ether (v/v) to afford the pure product 8a.

2-Methyl-3-(1-phenylvinyl)-2H-benzo[e][1,2,4]thiadiazine-1,1-dioxide (8a). White amorphous solid, mp 116–118 °C; (22.3 mg, 57% yield); ¹H NMR (CDCl₃, 300 MHz) δ 7.91 (dd, J = 7.9 Hz, 1.0 Hz, 1H), 7.74–7.65 (m, 2H), 7.53–7.48 (m, 3H), 7.46–7.39 (m, 3H), 6.02 (s, 1H), 5.93 (s, 1H), 3.11 (s, 3H); ¹³C NMR (CDCl₃, 150 MHz) δ 155.6, 143.9, 142.7, 135.6, 133.5, 129.2, 129.1, 128.2, 127.3, 126.9, 126.4, 121.3, 121.1, 31.7 ppm; HRMS (ESI⁺) m/z calculated for C₁₆H₁₅N₂O₂S [M + H]⁺ 299.0854, found 299.0848.

2-Methyl-3-[1-(pyridin-3-yl)vinyl]-2H-benzo[e][1,2,4]thiadiazine-1,1-dioxide (8b). Yellow gum (19.7 mg, 50% yield); 1 H NMR (CDCl₃, 600 MHz) δ 8.85 (d, J = 1.8 Hz, 1H), 8.65 (dd, J = 4.8 Hz, 1.2 Hz, 1H), 7.92 (dd, J = 7.8 Hz, 1.2 Hz, 1H), 7.81 (dt, J = 7.8 Hz, 2.1 Hz, 1H), 7.74–7.72 (m, 1H), 7.65 (dd, J = 8.1 Hz, 0.9 Hz, 1H), 7.55–7.52 (m, 1H), 7.38 (dd, J = 8.4 Hz, 4.8 Hz, 1H), 6.11 (s, 1H), 6.07 (s, 1H), 3.13 (s, 3H); 13 C NMR (CDCl₃, 150 MHz) δ 154.5, 150.1, 147.6, 142.4, 141.0, 133.9, 133.6, 128.3, 127.7, 127.2, 123.8, 123.3, 121.2, 32.2 ppm (one carbon merged with other aromatic carbons); HRMS (ESI⁺) m/z calculated for C₁₅H₁₄N₃O₂S [M + H]⁺ 300.0807, found 300.0800.

Synthesis of Luotonin F Analogues 28 (Scheme 5). Synthesis of N-Benzyl-2-[(4-methylphenyl)sulfonamide]-N-(propa-1,2-dien-1-yl)benzamide (25). Precursor acetylene 24 (100.0 mg, 0.24 mmol) synthesized by reported procedure 20a was treated with KO'Bu (80.5 mg, 0.72 mmol) in dry DMF (4 mL), and the whole reaction mixture was stirred at rt under an argon atmosphere for 2 h. After completion of the reaction (TLC), it was extracted with EtOAc (3 × 15 mL). The combined organic extracts were washed with water (10 mL), dried over anhydrous Na₂SO₄, and concentrated *in vacuo*. The crude residue was purified by silica gel (100–200 mesh) column chromatography using 15% ethyl acetate—petroleum ether (v/v) as eluent.

N-Benzyl-2-[(4-methylphenyl)sulfonamide]-*N*-(propa-1,2-dien-1-yl)benzamide (**25**). Yellow gum (96 mg, 96%); ¹H NMR (CDCl₃, 300 MHz) δ 8.39 (brs, 1H), 7.74 (d, J = 8.1 Hz, 1H), 7.39–7.26 (m, 9H), 7.15–7.11 (m, 1H), 6.89–6.85 (m, 2H), 5.77 (brs, 1H), 5.34–5.32 (m, 2H), 4.69 (brs, 2H), 2.23 (s, 3H); ¹³C NMR (CDCl₃, 75 MHz) δ 200.1, 167.4, 143.8, 136.5, 131.7, 129.5, 129.2, 128.9, 128.5, 127.7, 126.9, 125.8, 125.2, 124.5, 101.8, 87.8, 77.2, 47.9, 21.4 ppm. HRMS (ESI⁺) m/z calculated for C₂₄H₂₂N₂O₃SNa [M + Na]⁺ 441.1249, found 441.1235.

3-Methyl-2-[1-(quinolin-3-yl)vinyl]-1-tosyl-2,3-dihydroquinazolin-4(1H)-one (27a). Intermediate 27a was synthesized from 13 (85 mg, 0.25 mmol) adopting the general procedure of the synthesis of product 9 (Table 2) described earlier. The reaction time was 1 h. The crude product was purified by silica gel (100-200 mesh) column chromatography using 30% ethyl acetate-petroleum ether (v/ v) as eluent as a brown amorphous solid (100.2 mg, 86% yield); mp 174–176 °C; ¹H NMR (CDCl₃, 300 MHz) δ 8.88 (d, J = 2.1 Hz, 1H), 8.41 (brs, 1H), 8.14 (d, J = 8.4 Hz, 1H), 7.96–7.88 (m, 2H), 7.78 (t, J = 7.3 Hz, 1H), 7.69 - 7.63 (m, 1H), 7.52 - 7.47 (m, 1H), 7.40 - 7.35 (m, 1H)2H), 7.27-7.25 (m, 2H), 7.15 (d, J = 8.4 Hz, 2H), 6.58 (s, 1H), 5.59 (s, 1H), 5.12 (s, 1H), 2.98 (s, 3H), 2.35 (s, 3H); ¹³C NMR (CDCl₃, 75 MHz) δ 161.3, 148.8, 146.8, 145.0, 139.4, 135.1, 134.4, 133.7, 132.9, 130.3, 130.1, 129.7, 128.7, 128.5, 128.4, 127.8, 127.7, 127.5, 127.4, 127.0, 125.4, 119.3, 73.6, 33.9, 21.6 ppm; HRMS (ESI⁺) m/z calculated for C₂₇H₂₃N₃O₃SNa [M + Na]⁺ 492.1358, found 492.1353.

2-Methyl-3-[1-(quinolin-3-yl)vinyl]-4-tosyl-3,4-dihydro-2H-benzo-[e][1,2,4]thiadiazine-1,1-dioxide (27b). Intermediate 27b was synthesized from 21 adopting the general procedure of product 23 (Table 3) described earlier. The reaction time was 1.5 h. The crude product was purified by silica gel (100–200 mesh) column chromatography using 35% ethyl acetate—petroleum ether (v/v) as eluent as a white amorphous solid (66.8 mg, 50% yield); mp 184–186 °C; ¹H NMR (CDCl₃, 300 MHz) δ 9.02 (d, J = 2.4 Hz, 1H), 8.27 (d, J = 2.1 Hz, 1H), 8.12 (d, J = 8.4 Hz, 1H), 7.89–7.83 (m, 2H), 7.77–7.70 (m, 2H), 7.62–7.58 (m, 3H), 7.56–7.50 (m, 1H), 7.30 (td, J = 7.7 Hz, 1.0 Hz, 1H), 7.21 (d, J = 8.1 Hz, 2H), 6.77 (d, J = 1.2 Hz, 1H), 5.89 (d, J = 0.9 Hz, 1H), 5.83 (d, J = 1.5 Hz, 1H), 2.73 (s, 3H), 2.38

(s, 3H); $^{13}\mathrm{C}$ NMR (CDCl₃, 75 MHz) δ 149.3, 147.9, 145.3, 141.0, 135.3, 134.5, 133.7, 133.3, 130.5, 130.0, 129.8, 129.6, 129.4, 128.5, 128.3, 127.6, 127.3, 126.2, 126.1, 125.3, 120.7, 77.4, 34.5, 21.8 ppm; HRMS (ESI+) m/z calculated for $\mathrm{C_{26}H_{23}N_3O_4S_2Na}$ [M + Na]+ 528.1028, found 528.1029.

3-Benzyl-2-[1-(quinolin-3-yl)vinyl]-1-tosyl-2,3-dihydroquinazolin-4(1H)-one (27c). To a solution of $Pd(OAc)_2$ (2.6 mg, 0.01 mmol) and PPh₃ (7.2 mg, 0.03 mmol) in dry DMF (5 mL) were added 3iodoquinoline 26²⁷ (58.5 mg, 0.23 mmol), K₂CO₃ (126.8 mg, 0.92 mmol), and allene 25 (96 mg, 0.23 mmol) sequentially. The whole reaction mixture was allowed to stir at rt under an argon atmosphere, and the reaction completed after 1 h (TLC). DMF was removed under reduced pressure, and the mixture was extracted with EtOAc (3 \times 15 mL). The combined organic extracts were washed with water (10 mL), dried over anhydrous Na2SO4, and concentrated in vacuo. The crude residue was then purified by silica gel (100-200 mesh) column chromatography using 25% ethyl acetate-petroleum ether (v/v) as eluent to afford the product 27c as a white amorphous solid (75.1 mg, 60% yield); mp 103–105 °C; ¹H NMR (CDCl₃, 300 MHz) δ 8.70 (d, J = 2.1 Hz, 1H), 8.24 (d, J = 1.8 Hz, 1H), 8.11 (d, J = 8.4 Hz, 1H), 7.96-7.90 (m, 2H), 7.75 (t, J = 7.2 Hz, 1H), 7.60 (t, J = 7.2 Hz, 1H), 7.49-7.43 (m, 7H), 7.37-7.32 (m, 1H), 6.88 (d, J = 8.4 Hz, 2H), 6.81(d, J = 8.1 Hz, 2H), 6.72 (s, 1H), 5.45 (s, 1H), 5.06 (s, 1H), 5.01 (d, J= 13.8 Hz, 1H), 4.22 (d, J = 13.8 Hz, 1H), 2.27 (s, 3H); ¹³C NMR (CDCl₃, 75 MHz) δ 161.4, 149.2, 147.5, 144.3, 139.6, 135.4, 135.0, 133.6, 133.5, 133.0, 130.2, 130.2, 129.8, 129.4, 129.0, 128.9, 128.5, 128.3, 128.0, 127.3, 127.3, 127.0, 125.9, 125.3, 119.5, 71.3, 49.9, 21.5 ppm; HRMS (ESI⁺) m/z calculated for $C_{33}H_{28}N_3O_3S$ [M + H]⁺ 546.1851, found 546.1847.

3-Methyl-2-[1-(quinolin-3-yl)vinyl]quinazolin-4(3H)-one (**28a**). To a stirred solution of 27a (101.4 mg, 0.22 mmol) in dry DMF was added NaH (60% oil suspension) (43.2 mg, 1.08 mmol), and the whole reaction mixture was allowed to stir at rt under an argon atmosphere for 1 h. Upon consumption of the starting material (TLC), the reaction was quenched with water (5 mL) and extracted with EtOAc (3 \times 15 mL). The combined organic extracts were dried over anhydrous Na₂SO₄ and concentrated in vacuo. The crude product was purified by silica gel (100-200 mesh) column chromatography using 55% ethyl acetate-petroleum ether (v/v) as eluent to afford product 28a as a yellow amorphous solid (36.0 mg, 53% yield); (overall yield 39% after 3 steps); mp 180-182 °C; ¹H NMR (CDCl₃, 300 MHz) δ 9.15 (d, J = 2.1 Hz, 1H), 8.37 (d, J = 8.1 Hz, 1H), 8.13 (d, J = 8.7 Hz, 1H), 7.93 (d, J = 1.8 Hz, 1H), 7.83 (d, J = 3.6 Hz, 2H),7.77-7.70 (m, 1H), 7.58-7.53 (m, 2H), 7.49-7.46 (m, 1H), 6.29 (s, 1H), 5.91 (s, 1H), 3.42 (s, 3H); 13 C NMR (CDCl₃, 100 MHz) δ 162.5, 154.9, 148.0, 147.4, 141.6, 134.7, 132.9, 132.2, 130.5, 129.5, 128.7, 128.6, 128.4, 127.9, 127.7, 127.6, 127.0, 121.1, 120.3, 32.9 ppm; HRMS (ESI⁺) m/z calculated for $C_{20}H_{15}N_3ONa$ [M + Na]⁺ 336.1113, found 336.1117.

2-Methyl-3-[1-(quinolin-3-yl)vinyl]-2*H***-benzo[***e***][1,2,4]thiadiazine-1,1-dioxide (28b). Product 28b was synthesized adopting the procedure of 28a; the only difference was that 2.0 equiv of NaH (60% oil suspension) was used in this case instead of 5.0 equiv and the reaction time was 1.5 h. Product 28b was eluted using 40% ethyl acetate—petroleum ether (v/v) as a yellow gum (17.5 mg, 38% yield); (overall yield 19% after 3 steps); ¹H NMR (CDCl₃, 300 MHz) δ 9.16 (s, 1H), 8.26–8.14 (m, 2H), 7.85–7.54 (m, 3H), 7.25–7.20 (m, 1H), 6.91–6.82 (m, 3H), 6.26 (s, 1H), 6.13 (s, 1H), 3.15 (s, 3H); ¹³C NMR (CDCl₃, 125 MHz) δ 155.9, 148.2, 147.9, 142.6, 141.1, 134.1, 133.8, 130.7, 129.8, 129.1, 128.8, 128.5, 127.9, 127.8, 123.2, 121.4, 120.7, 115.5, 32.3 ppm; HRMS (ESI⁺) m/z calculated for C₁₉H₁₅N₃O₂SNa [M + Na]⁺ 372.0783, found 372.0785.**

3-Benzyl-2-[1-(quinolin-3-yl)vinyl]quinazolin-4(3H)-one (28c). Product 28c was prepared adopting the procedure of 28a, and the reaction time was 2 h. Product 28c was eluted using 45% ethyl acetate—petroleum ether (v/v) as a yellow gum (33.2 mg, 62% yield); (overall yield 36% after 3 steps); 1 H NMR (CDCl₃, 300 MHz) δ 9.06 (d, J = 2.1 Hz, 1H), 8.43–841 (m, 1H), 8.09 (d, J = 8.4 Hz, 1H), 7.86–7.80 (m, 3H), 7.74–7.68 (m, 1H), 7.66–7.57 (m, 2H), 7.54–7.49 (m, 1H), 7.14–7.09 (m, 2H), 7.05–7.01 (m, 3H), 6.14 (s, 1H),

5.61 (s, 1H), 5.20 (s, 2H); 13 C NMR (CDCl₃, 75 MHz) δ 162.5, 154.8, 147.7, 147.3, 140.6, 136.2, 134.9, 133.2, 130.4, 129.7, 129.2, 128.9, 128.6, 128.3, 127.9, 127.8, 127.5, 127.4, 126.7, 121.2, 120.8, 120.3, 115.6, 48.3 ppm; HRMS (ESI⁺) m/z calculated for $C_{26}H_{20}N_3O$ [M + H]⁺ 390.1606, found 390.1602.

ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.6b01242.

Experimental procedures, copies of NMR (¹H and ¹³C), HRMS of representative compounds, 2D NMR data of compound **20**, and tables of crystal data of compounds **9aa**, **9bd**, **23g**, and **7d** (PDF)

X-ray crystallographic data for compounds 7d, 9aa, 9bd, and 23g (ZIP)

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Notes

The authors declare no competing financial interest.

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