

Regioselective Allene Hydrosilylation Catalyzed by N-Heterocyclic Carbene Complexes of Nickel and Palladium

Zachary D. Miller, Wei Li, Tomás R. Belderrain, and John Montgomery*, and John Montgomery

[‡]Department of Chemistry, University of Michigan, 930 North University Avenue, Ann Arbor, Michigan 48109-1055, United States [†]Laboratorio de Catálisis Homogénea, Unidad Asociada al CSIC, CIQSO-Centro de Investigación en Química Sostenible and Departamento de Química y Ciencias de los Materiales, Campus de El Carmen s/n, Universidad de Huelva, 21007-Huelva, Spain

Supporting Information

ABSTRACT: Regioselective methods for allene hydrosilylation have been developed, with regioselectivity being governed primarily by the choice of metal. Alkenylsilanes are produced via nickel catalysis with larger *N*-heterocyclic carbene (NHC) ligands, and allylsilanes are produced via palladium catalysis with smaller NHC ligands. These complementary methods allow either regioisomeric product to be obtained with exceptional regiocontrol.

lkenylsilanes and allylsilanes are useful reagent classes that Aare employed in numerous synthetic transformations. For example, the Hiyama cross-coupling² and Sakurai-type allylation and crotylation reactions³ are widely used methods for C-C bond formation. Currently, the most direct and atomeconomical routes to vinyl- and allylsilanes proceed via metalcatalyzed hydrosilylations of π -components (Scheme 1). The hydrosilylation of alkynes to afford alkenylsilanes⁴ and the hydrosilylation of 1,3-dienes to afford allylsilanes⁵ are versatile methods that have been widely employed. In both instances, control of regiochemistry is required with unsymmetrical substrates, and even the most efficient hydrosilylation procedures are often plagued by the lack of complete regioselectivity for a variety of substrates without directing group effects. Compounding these challenges is the difficulty typically seen in the separation of regioisomeric allyl- or vinylsilanes produced by these methods.

In contrast to the significant developments that have been reported in hydrosilylations of alkynes and 1,3-dienes, the hydrosilylation of allenes has rarely been examined. The hydrometalation of allenes presents considerable challenges in

Scheme 1

Previous work

$$R_3SiH + R_1 \longrightarrow R_1$$
 $R_3SiH + R_1 \longrightarrow R_3Si \longrightarrow R_1$
 $R_3SiH + R_1 \longrightarrow R_3Si \longrightarrow R_1$
 $R_1 \longrightarrow R_1$

Scheme 2

Table 1

on here	mus satalyzat	L·HX	0/ vriold	regiosel. (3:4)
entry	precatalyst	r.uv	% yield	regiosei. (3:4)
1	$Ni(COD)_2$	5a	22	33:67
2	$Ni(COD)_2$	5b	15	40:60
3	$Ni(COD)_2$	6a	58	85:15
4	$Ni(COD)_2$	6b	47	81:19
5	$Ni(COD)_2$	7	84	>98:<2
6	$Pd_2(dba)_3$	7	trace	not determined
7	$Pd_2(dba)_3$	6b	56	14:86
8	$Pd_2(dba)_3$	6a	75	12:88
9	$Pd_2(dba)_3$	5b	74	2:98
10	$Pd_2(dba)_3$	5a	80	<2:>98
_•HX =				

selectivity since 1,2- or 2,1-addition to either of the two contiguous π -systems can occur, potentially providing two regioisomeric allylsilanes and two regioisomeric vinylsilanes, in

Received: July 27, 2013

Scheme 3

some cases as mixtures of alkene stereoisomers. The hydroboration of allenes has been widely developed, but these processes are typically uncatalyzed, thus providing a single allylborane product governed by substrate steric and electronic characteristics. Significant advances have appeared in the regioselective bis-metalation of allenes, through catalytic reactions such as diboration and silaboration, but regiodivergent processes in simpler hydrometalations of allenes are lacking. Given the synthetic utility of allyl- and vinylsilanes and the potential to provide regiocontrolled access to both structure classes from a common allene substrate, we report herein our efforts to develop regioselective, catalyst-controlled methods for the hydrosilylation of allenes.

As part of our laboratory's broader interest in the development of regioselective catalytic processes, 11 we initially hoped to develop ligand-controlled regioselectivity reversal in the nickel-catalyzed hydrosilylation of allenes. However, upon screening a broad range of nickel catalysts modified with Nheterocyclic carbene or phosphine ligands, combinations that allowed high-yielding and highly regioselective access to both alkenylsilane and allylsilane products were not identified. However, in the course of these exploratory studies, we were attracted to the recent disclosures of well-defined Nheterocyclic carbene (NHC) complexes of palladium and nickel.¹² To our surprise, simple modification of the metal, while leaving the ligand and reaction conditions unchanged, provided an encouraging lead toward identifying routes to both product regioisomers. As illustrated in the hydrosilylation of cyclohexylallene with triethylsilane, the use of nickel complex 1 provided the vinylsilane 3 with 85:15 regioselectivity, whereas exactly the same procedure and same ligand, but employing the palladium complex 2 instead, provided the allylsilane 4 with 88:12 regioselectivity (Scheme 2). Reactions involving the presynthesized nickel complex 1 required an elevated temperature, likely due to the need to dissociate styrene to provide the

Table 2

SiEt ₃	conditions A	n 🔷.	conditions B	SiEt ₃
$R \checkmark $	Ni(COD) ₂ (10 mol %)	+	Pd ₂ (dba) ₃ (5 mol %)	R^{\downarrow}
8	7 (10 mol %) KO- <i>t</i> -Bu (10 mol %) THF. rt	Et ₃ SiH	5a (10 mol %) KO- <i>t</i> -Bu (10 mol %) THF. rt	9

11117,11			וחר, ת		
entry	R	conditions ^a	product (% yield)	regiosel. (8:9)	
1	Су	A	3 (84)	>98:2	
2		В	4 (80)	<2:98	
3	n-Oct	A	8a (80)	>98:2	
4		В	9a (98)	<2:98	
5	Ph	A	8b (78)	>98:2	
6		В	9b (94)	<2:98	
7	HO H	A	8c (62)	>98:2	
8		В	9c (88)	<2:98	
9	TBSO Y	A	8d (73)	>98:2	
10		В	9d (97)	<2:98	
11	N W Y Y	A	8e (78)	>98:2	
12	O	В	9e (97)	<2:98	
13	Су	B^{b}	4 (98)	<2:98	
14	Су	B ^c	4 (74)	4:96	
15	Су	B^{d}	4 (72)	<2:98	

^aUnless otherwise noted, reactions were conducted on a 0.3 mmol scale. ^bConducted on 6.0 mmol scale. ^cConducted with benchtop assembly of the catalyst and ligand followed by three vacuum/purge cycles before allene and silane introduction. ^dConducted at 2.5 mol % loading of Pd₂(dba)₃.

active catalyst form, whereas palladium complex 2 was catalytically active at rt or elevated temperature. This unusual example of regiochemistry reversal with a common ligand scaffold and within a single metal triad presented an unexpected and effective strategy for accessing both allylsilanes and alkenylsilanes from a common allene precursor.

To further boost the regioselectivity of these complementary processes, we examined ligand variation for both nickel- and palladium-catalyzed hydrosilylations. In comparing catalytic

Scheme 4

Table 3

entry	silane	product (% yield)	regiosel.
1	HSiMe ₂ Ph	12a (91)	>98:2
2	$HSiMe_2Bn$	12b (93)	>98:2
3	$HSiMe(OSiMe_3)_2$	12c (82)	>98:2
4	$H_2Si(t-Bu)_2$	12d (75)	>98:2

reactions of 1 and 2 with the corresponding complexes prepared *in situ* from Ni(COD)₂ or Pd₂(dba)₃, the chemical yields are similar and regioselectivities are identical in comparing the catalysts presynthesized or prepared *in situ* (comparing Scheme 2 to Table 1, entries 3 and 8). The use of the presynthesized complexes comes with the advantage of a higher catalyst turnover and ease of accurate weighing on small scale that becomes difficult for the *in situ* protocol. However, to facilitate more rapid screening of a variety of ligands, catalysts for the optimization studies were prepared *in situ* by treatment of Ni(COD)₂ or Pd₂(dba)₃ with the HCl or HBF₄ salt of the desired NHC ligands.

From the ligand variation studies, an interesting trend emerged that regioselectivities for alkenylsilane products using nickel were enhanced with larger ligands, whereas regioselectivities for allylsilanes using palladium were enhanced with smaller ligands. To illustrate this trend using Ni(COD)₂ as the precatalyst, IMes and SIMes slightly favored allylsilane production (Table 1, entries 1 and 2), IPr and SIPr favored alkenylsilane production with good selectivity (Table 1, entries 3 and 4), and the very bulky ligand DP-IPr afforded the alkenylsilane in 84% isolated yield with >98:2 regioselectivity (Table 1, entry 5). In comparison, using Pd₂(dba)₃ as the precatalyst, the use of DP-IPr afforded a low-yielding mixture of regioisomers (Table 1, entry 6), IPr and SIPr favored allylsilane production with good selectivity (Table 1, entries 7 and 8), and IMes and SIMes provided exceptional regioselectivities (Table 1, entries 9 and 10), with IMes providing an 80% isolated yield of the allylsilane with >98:2 regioselectivity.

The above guidelines for the in situ generated Ni(0) and Pd(0) hydrosilylation catalysts were utilized in the reactions for a variety of allene substrate classes (Table 2). In addition to the use of cyclohexyl allene (entries 1 and 2), allenes containing less hindered aliphatic groups (Table 2, entries 3 and 4), aromatic groups (Table 2, entries 5 and 6), unprotected hydroxyls (Table 2, entries 7 and 8), and silyloxy groups (Table 2, entries 9 and 10) were successfully hydrosilylated using either Ni(COD)₂ with ligand 7 (DP-IPr; conditions A) or Pd₂(dba)₃ with ligand 5a (IMes; conditions B) with consistently excellent regioselectivities (>98:2) and in high chemical yields. In addition, an allene containing a phthalimido moiety underwent smooth hydrosilylation to afford either regioisomeric product (Table 2, entries 11 and 12). To further explore the scaleup and utility in reactions not using glovebox manipulations, the palladium-catalyzed procedure with cyclohexyl allene was optimized on a 6 mmol scale, which proceeded in 98% isolated yield in >98:2 regioselectivity (Table 2, entry 13). Additionally, a procedure conducted with a standard benchtop assembly purged with a nitrogen line proceeds in 74% isolated yield with slightly eroded (96:4) regioselectivity (Table 2, entry 14). Finally, an experiment with a reduced

catalyst loading (2.5 mol % $Pd_2(dba)_3$) proceeded in 72% isolated yield (Table 2, entry 15), whereas attempts to lower the catalyst loading further resulted in more significant drops in chemical yield.

In order to evaluate the effects of introducing allene 1,1-disubstitution, we subjected vinylidene cyclohexane to both nickel- and palladium-catalyzed reaction conditions (Scheme 3). The nickel-catalyzed reaction resulted in an outcome consistent with the previous observations for monosubstituted allenes, where the silyl group is installed at the central allenic carbon to afford vinylsilane 10. Reactions with palladium, however, afforded a new observed allylsilane isomer 11, with the C—Si bond forming at the least substituted allenic carbon. This observation offers a new entry to a useful allylsilane traditionally accessed via the hydrosilylation of 1,3-dienes.

The palladium-based procedure tolerates variation of the silane structure with minimal change in regioselectivity and yield. For example, additions to cyclohexyl allene cleanly proceed with a range of synthetically versatile silane structures (Table 3).¹³ Alternatively, the nickel-based procedure is currently more limited with silane variation leading to a drop in yield and erosion of regioselectivy.

While the origin of regiocontrol is at present uncertain, we propose that a likely mechanism involves a change from a silylmetalation pathway with nickel to a hydrometalation pathway with palladium (Scheme 4). Migratory insertions to allenes typically favor addition to the central allene carbon with the formation of a metal π -allyl complex. ^{9g,10g} Through this common mechanistic feature, silylmetalation of nickel allene complex 12 to generate π -allyl intermediate 13, followed by C-H reductive elimination, would afford vinylsilane product 8. Alternatively, hydrometalation of palladium allene complex 14 to generate π -allyl intermediate 15, followed by C-Si reductive elimination, would afford allylsilane product 9. By utilizing a large ligand with nickel, disposition of the silyl group trans to the ligand in intermediate 12 reinforces the preference for silylmetalation. However, by utilizing a small ligand with palladium, disposition of the silyl group cis to the ligand in 14 reinforces the preference for hydrometalation. Detailed studies from Morken and Suginome in diborations and silaborations of allenes provide significant insight into the factors that control regiochemistry in these related processes, and the mechanistic model described in Scheme 4 draws from the insights of those studies. 9g,10g Future work will be devoted to evaluating the intriguing mechanistic basis of these regiodivergent hydrosilvlation processes.

In summary, the complementary use of metals (Ni vs Pd) results in regiochemical reversals in allene hydrosilylations. Alterations in NHC ligand structure can further improve the regioselectivities to provide regiodivergent access to a wide range of alkenylsilanes or allylsilanes in high yields and with exceptional regiocontrol. The above study demonstrates the special role that metal identity can play in governing regioselective catalytic reactions. Future studies will explore these effects in other classes of catalytic reactions.

ASSOCIATED CONTENT

Supporting Information

Experimental details and analytical data. This material is available free of charge via the Internet at http://pubs.acs.org.

AUTHOR INFORMATION

Corresponding Author

jmontg@umich.edu

Funding

No competing financial interests have been declared.

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

We thank the National Science Foundation (CHE-1265491), the National Institutes of Health (GM57014) and MINECO (CTQ2011–24502) for support of this research. NSF supports a program including a comparative analysis of base metal and precious metal catalysis. NIH supports a program in the design of NHC ligands for regioselective transformations. Professor Pedro Pérez (Universidad de Huelva) is thanked for the encouragement and helpful discussions.

REFERENCES

- (1) (a) Brook, M. A. Silicon in Organic, Organometallic, and Polymer Chemistry; Wiley: Chichester, 2000. (b) Fleming, I.; Barbero, A.; Walter, D. Chem. Rev. 1997, 97, 2063–2192. (c) Fleming, I.; Dunoguès, J.; Smithers, R. Org. React. 1989, 37, 57. (d) Ojima, I.; Li, Z. Y.; Zhu, J. W. In Chemistry of Organic Silicon Compounds; Rappoport, Z., Apeloig, Y., Eds.; John Wiley & Sons Ltd.: Chichester, U.K., 1998.
- (2) (a) Nakao, Y.; Hiyama, T. Chem. Soc. Rev. 2011, 40, 4893.
 (b) Chang, W.-T. T.; Smith, R. C.; Regens, C. S.; Bailey, A. D.; Werner, N. S.; Denmark, S. E. Org. React. 2011, 75, 213.
- (3) (a) Hosomi, A.; Endo, M.; Sakurai, H. Chem. Lett. 1976, 941.
 (b) Masse, C. E.; Panek, J. S. Chem. Rev. 1995, 95, 1293.
- (4) (a) Rooke, D. A.; Ferreira, E. M. Angew. Chem., Int. Ed. 2012, 51, 3225. (b) Trost, B. M.; Ball, Z. T. J. Am. Chem. Soc. 2001, 123, 12726. (c) Ball, Z. T.; Trost, B. M. J. Am. Chem. Soc. 2005, 127, 17644. (d) Berthon-Gelloz, G.; Schumers, J.; De Bo, G.; Markó, I. E. J. Org. Chem. 2008, 73, 4190. (e) De Bo, G.; Berthon-Gelloz, G.; Tinant, B.; Markó, I. E. Organometallics 2006, 25, 1881. (f) Denmark, S. E.; Wang, Z. G. Org. Lett. 2001, 3, 1073. (g) Kawasaki, Y.; Ishikawa, Y.; Igawa, K.; Tomooka, K. J. Am. Chem. Soc. 2011, 133, 20712. (h) Chaulagain, M. R.; Mahandru, G. M.; Montgomery, J. Tetrahedron 2006, 62, 7560. (i) Lappert, M. F.; Nile, T. A.; Takahashi, S. J. Organomet. Chem. 1974, 72, 425.
- (5) (a) Ojima, I.; Kumagai, M. J. Organomet. Chem. 1978, 157, 359.
 (b) Wu, J. Y.; Stanzl, B. N.; Ritter, T. J. Am. Chem. Soc. 2010, 132, 13214.
 (c) Onozawa, S.; Sakakura, T.; Tanaka, M. Tetrahedron Lett. 1994, 35, 8177.
 (d) Tsuji, J.; Hara, M.; Ohno, K. Tetrahedron 1974, 30, 2143.
- (6) (a) Sudo, T.; Asao, N.; Gevorgyan, V.; Yamamoto, Y. J. Org. Chem. 1999, 64, 2494. (b) Huang, G.; Isobe, M. Tetrahedron 2001, 57, 10241.
- (7) For uncatalyzed hydroborations: (a) Kister, J.; DeBaillie, A. C.; Lira, R.; Roush, W. R. J. Am. Chem. Soc. 2009, 131, 14174. (b) Chen, M.; Handa, M.; Roush, W. R. J. Am. Chem. Soc. 2009, 131, 14602. (c) Ess, D. H.; Kister, J.; Chen, M.; Roush, W. R. Org. Lett. 2009, 11, 5538. (d) Gonzalez, A. Z.; Roman, J. G.; Alicea, E.; Canales, E.; Soderquist, J. A. J. Am. Chem. Soc. 2009, 131, 1269. (e) Brown, H. C.; Liotta, R.; Kramer, G. W. J. Am. Chem. Soc. 1979, 101, 2966.
- (8) For catalyzed hydroborations: (a) Semba, K.; Shinomiya, M.; Fujihara, T.; Terao, J.; Tsuji, Y. *Chem.—Eur. J.* **2013**, *19*, 7125. (b) Yamamoto, Y.; Fujikawa, R.; Yamada, A.; Miyaura, N. *Chem. Lett.* **1999**, 1069.
- (9) (a) Ishiyama, T.; Kitano, T.; Miyaura, N. Tetrahedron Lett. 1998, 39, 2357. (b) Yang, F. Y.; Cheng, C. H. J. Am. Chem. Soc. 2001, 123, 761. (c) Pelz, N. F.; Woodward, A. R.; Burks, H. E.; Sieber, J. D.; Morken, J. P. J. Am. Chem. Soc. 2004, 126, 16328. (d) Woodward, A. R.; Burks, H. E.; Chan, L. M.; Morken, J. P. Org. Lett. 2005, 7, 5505.

- (e) Sieber, J. D.; Morken, J. P. J. Am. Chem. Soc. 2006, 128, 74.
 (f) Pelz, N. F.; Morken, J. P. Org. Lett. 2006, 8, 4557. (g) Burks, H. E.;
 Liu, S. B.; Morken, J. P. J. Am. Chem. Soc. 2007, 129, 8766.
- (10) (a) Suginome, M.; Ohmori, Y.; Ito, Y. Synlett 1999, 1567. (b) Suginome, M.; Ohmori, Y.; Ito, Y. J. Organomet. Chem. 2000, 611, 403. (c) Suginome, M.; Ohmura, T.; Miyake, Y.; Mitani, S.; Ito, Y.; Murakami, M. J. Am. Chem. Soc. 2003, 125, 11174. (d) Ohmura, T.; Suginome, M. Org. Lett. 2006, 8, 2503. (e) Ohmura, T.; Taniguchi, H.; Suginome, M. J. Am. Chem. Soc. 2006, 128, 13682. (f) Chang, K. J.; Rayabarapu, D. K.; Yang, F. Y.; Cheng, C. H. J. Am. Chem. Soc. 2005, 127, 126. (g) Abe, Y.; Kuramoto, K.; Ehara, M.; Nakatsuji, H.; Suginome, M.; Murakami, M.; Ito, Y. Organometallics 2008, 27, 1736. (h) Ohmura, T.; Oshima, K.; Taniguchi, H.; Suginome, M. J. Am. Chem. Soc. 2010, 132, 12194.
- (11) (a) Malik, H. A.; Sormunen, G. J.; Montgomery, J. J. Am. Chem. Soc. 2010, 132, 6304. (b) Liu, P.; Montgomery, J.; Houk, K. N. J. Am. Chem. Soc. 2011, 133, 6956. (c) Li, W.; Chen, N.; Montgomery, J. Angew. Chem., Int. Ed. 2010, 49, 8712.
- (12) (a) Martín, C.; Molina, F.; Alvarez, E.; Belderrain, T. R. Chem.—Eur. J. 2011, 17, 14885. (b) Iglesias, M. J.; Blandez, J. F.; Fructos, M. R.; Prieto, A.; Álvarez, E.; Belderrain, T. R.; Nicasio, M. C. Organometallics 2012, 31, 6312.
- (13) (a) Roberson, C. W.; Woerpel, K. A. J. Am. Chem. Soc. 2002, 124, 11342. (b) Tinsley, J. M.; Roush, W. R. J. Am. Chem. Soc. 2005, 127, 10818. (c) Tanino, K.; Yoshitani, N.; Moriyama, F.; Kuwajima, I. J. Ore, Chem. 1997, 62, 4206.