Catalytic Enantioselective 5-Hydroxyisoxazolidine Synthesis: An Asymmetric Entry to β-Amino Acids

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Abstract: The highly chemo- and enantioselective organocatalytic tandem reaction between *N*-carbamate-protected hydroxylamines and α , β -unsaturated aldehydes is presented. The reaction represents a unique entry for the asymmetric synthesis of 5-hydroxyisoxazolidines, oxazolidin-5-ones or γ -hydroxyamino alcohols in high yields and 90–99% ee. A procedure for the conversion of the oxazolidin-5-ones into the corresponding β -amino acids is also described.

Key words: organocatalysis, amination reactions, β -amino acid synthesis, isoxazolidines, asymmetric catalysis



Scheme 1

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Introduction

5-Hydroxyisoxazolidines and isoxazolidin-5-ones are important chiral building blocks¹ which are readily converted into the corresponding amino alcohols and β -amino acids.² Thus, asymmetric methods have been developed for their preparation.¹ For example, optically active 5-acetoxydihydroisoxazoles can be converted in two steps into the corresponding isoxazolidinones.^{1d,e} Moreover, utilization of chiral auxiliaries enables the asymmetric synthesis of isoxazolidin-5-ones.^{1e-j} Recently, isoxazolidin-5-ones were prepared by Lewis acid catalyzed enantioselective conjugate addition of hydroxylamines to α , β -unsaturated amide derivatives.^{1b,c,3}

In the field of organocatalysis, amine-catalyzed reactions that involve catalytic domino, tandem, or cascade reaction pathways have recently been developed.^{4–6} In this context, we have developed an asymmetric domino amine–conjugate/aldol reaction.⁷ Based on these lessons and retrosynthetic analysis, we recently discovered a chiral-amine-

catalyzed reaction between N-protected hydroxylamines and enals.⁸ The reaction is a simple asymmetric entry to 5-hydroxyisoxazolidines where the subsequent tandem intramolecular hemiacetal formation is an important driving force for product formation (procedure 1, Scheme 1).

Scope and Limitations

The mild reaction conditions for the 2-[diphenyl(trimethylsiloxy)methyl)pyrrolidine $(4)^9$ catalyzed reactions are compatible with several different types of enals. For example, the reaction between cinnamaldehyde (**2a**) and *tert*-butyl *N*-hydroxycarbamate (**1a**) in chloroform afforded the desired product **3a** in 90% yield and 99% ee at room temperature. In fact, the α , β -unsaturated aldehydes **2** reacted with *N*-hydroxycarbamates **1** leading to 5-hydroxyisoxazolidines **3** in excellent yields and enantioselectivities in the presence of chiral amine **4** (Table 1).





R ¹ OH	+ R ²	$+ R^{2}$				
1	2		3			
Entry	\mathbb{R}^1	R ²	Product	Time (h)	Yield ^a (%) ^a	ee ^b (%)
6	Boc	$4-O_2NC_6H_4$		3	75	98
7	Boc	2-naphthyl	Boc N-O	3	77	95
8	Cbz	CO ₂ Et	3g Cbz N=0 EtO_2C OH	16	85	97
9	Boc	<i>n-</i> Bu		16	94	91
10	Boc	<i>n</i> -Pr		16	93	91
11	Cbz	<i>n</i> -Pr	Cbz N=O n-Pr	16	92	95

 Table 1
 Scope of the Organocatalytic Tandem Reaction (continued)

Ph

^a Isolated yield of the pure product **3** after silica gel chromatography.

^b Determined by chiral-phase HPLC or GC analyses.

^c Reaction performed at r.t.

Thus, the reactions work well with aliphatic and aromatic α,β -unsaturated aldehydes. Moreover the reaction with *N*-Cbz-protected hydroxylamine **1b** gave the corresponding products **3** in 85–94% yield and 95–99% ee (entries 2, 8 and 11). Moreover, products **3** were formed as single anomers. The tandem reactions with a,b-unsaturated aliphatic acceptor aldehydes **2** (entries 9–11) were slower as compared to the aryl-substituted enals **2** (entries 1–7). In addition, the organocatalytic reaction is readily scaled up and gives access to a variety of 5-hydroxyisoxazolidines. Notably, the pK_a of the hydroxylamine is important since changing the carbamate group of the *N*-hydroxycarbamate **1** to an aryl group led to a complete change of chemoselectivity and 1,3-dipolar addition with the enal.¹⁰

The chiral-amine-catalyzed reaction between N-hydroxycarbamates 1 and enals 2 is also suitable for the one-pot asymmetric synthesis of oxazolidin-5-ones **5** and γ -hydroxyamino alcohols **7** (Scheme 2).

Both aromatic and aliphatic enals are suitable as substrates for these organocatalytic one-pot reactions. Thus, in situ oxidation with sodium chlorite of the 5-hydroxyisoxazilidones **3** gave the corresponding N-protected 5-oxazolidinones **5** in high overall yield with 95–99% ee. Moreover, in situ reduction with sodium borohydride gave a direct access to γ -amino alcohol **7a**. Notably, hydrogenolysis of the isolated oxazolidinones **5b** and **5c** with 10% palladium-on-carbon led to efficient N–O bond cleavage and removal of the Cbz group to quantitatively give the corresponding β -amino acids **6**. Comparison with the literature revealed that the absolute configuration of **6b** at C3 was S {[α]_D²⁵–6.9 (*c* 1, H₂O), Lit.¹¹ [α]_D²⁵–6.9 (*c* 0.8, H₂O)}.



Scheme 2 One-pot organocatalytic enantioselective synthesis of oxazolidin-5-ones 5 and amino alcohol 7a and β -amino acid 6 synthesis

In summary, we report highly chemo- and enantioselective catalytic procedures for the synthesis of 5-hydroxyisoxazolidines, oxazolidin-5-ones, and γ -amino alcohols, which are formed in high yields with 90–99% ee. Moreover, the organocatalytic tandem reaction represents a versatile asymmetric entry to different β -amino acid and γ -amino alcohol derivatives.

Procedures

Herein, we describe four typical experimental procedures demonstrating the synthetic scope of the chiral amine-catalyzed tandem reaction between hydroxylamines and enals. In Procedure 1, we report the preparation of 5-hydroxyisoxazolidine 3b (94% yield; 99% ee) starting from commercially available hydroxylamine 1b and enal 2a in the presence of chiral amine 4. The second procedure (Procedure 2) describes the in situ conversion of 3b, which is generated by 4 catalyzed reaction between 1b and 2a in one-pot, into the corresponding oxazolidin-5-one 5b in 81% yield and 99% ee. In Procedure 3, we report the one-pot synthesis of γ -hydroxyamino alcohol 7a (87% yield; 99% ee) starting from hydroxylamine 1a and enal 2a in the presence of a catalytic amount of 4 followed by in situ reduction with NaBH₄. In the last procedure (Procedure 4), the Pd/ C promoted N-O bond cleavage of oxazolidin-5-one 5b, which quantitatively gives the corresponding known β -amino acid **6b**, is described.

(-)-(3*S*,5*R*)-2-(Benzyloxycarbonyl)-3-phenylisoxazolidin-5-ol (3b); Typical Procedure

To a stirred soln of catalyst **4** (20 mol%) in CHCl₃ (0.5 mL) at 4 °C was added α , β -unsaturated aldehyde **2a** (33 mg, 0.25 mmol) and hydroxycarbamate **1b** (55 mg, 0.3 mmol). The reaction was vigorously stirred for 3 h. Next, the mixture was directly loaded upon a column and immediate subjected to chromatography (silica gel, pentane–EtOAc mixtures) to furnish the pure 5-hydroxyisoxazolidine **3b** (70 mg, 94%) as a clear oil.

The enantiomeric excess value was determined by HPLC with an Daicel Chiralpack ODH column (*n*-hexene–*i*-PrOH, 98.0:2.0), 1.0 mL/min): $t_{\rm R} = 35.9$ (major isomer), 30.8 min (minor isomer).

 $[\alpha]_D^{25}$ –22.2 (*c* 1.0, CHCl₃).

¹H NMR (400 MHz, CDCl₃): δ = 2.28–2.32 (m, 1 H), 2.78 (dd, *J* = 8.4 Hz, *J'* = 12.8 Hz, 1 H), 5.18 (s, 2 H), 5.39 (t, *J* = 8.4 Hz, 1 H), 5.84 (d, *J* = 4.4 Hz, 1 H), 7.20–7.40 (m, 10 H).

¹³C NMR (100 MHz, CDCl₃): δ = 45.2, 61.3, 68.1, 98.7, 126.0, 127.4, 127.7, 128.1, 128.4, 128.6, 135.6, 141.4, 159.3.

HRMS (ESI): $m/z [M + H]^+$ calcd for $C_{17}H_{18}NO_4$: 300.1230; found: 300.1233.

(-)-(3S)-2-(Benzyloxycarbonyl)-3-phenylisoxazolidin-5-one (5b); Typical Procedure

To a stirred soln of catalyst **4** (20 mol%) in CHCl₃ (0.5 mL) at 4 °C was added α , β -unsaturated aldehyde **2a** (33 mg, 0.25 mmol) and hydroxycarbamate **1b** (55 mg, 0.3 mmol). The reaction was vigorously stirred for 3 h. Upon completion (a small aliquot was removed for ee determination) The reaction temperature was increased to r.t., isobutene (0.1 mL), *t*-BuOH (0.4 mL), H₂O (0.2mL), KH₂PO₄ (54.4 mg, 4 mmol), and NaClO₂ (36 mg, 4mmol) were added sequentially. After 16 h, the crude product **5b** was purified by column chromatography (pentane–EtOAc mixtures) to afford isoxazolidin-5-one **5b** (60 mg, 81%) as a clear oil.

 $[\alpha]_{D}^{25} - 33.2 \ (c \ 1.0, \text{CHCl}_3).$

¹H NMR (400 MHz, CDCl₃): δ = 2.85 (dd, *J* = 5.1 Hz, *J'* = 15.9 Hz, 1 H), 3.26 (dd, *J* = 10.5 Hz, *J'* = 15.9 Hz, 1 H), 5.20–5.10 (m, 2 H), 5.60 (dd, *J* = 5.1 Hz, *J'* = 10.5 Hz, 1 H), 7.20–7.42 (m, 10 H).

¹³C NMR (100 MHz, CDCl₃): δ = 36.6, 59.1, 68.3, 127.2, 127.9, 128.1, 128.2, 128.5, 128.6, 135.5, 138.1, 157.4, 175.3.

HRMS (ESI): m/z [M + H₂O + Na]⁺ calcd for C₁₇H₁₇NNaO₅: 338.0999: found: 338.1002.

(-)-(3S)-3-[(*tert*-Butoxycarbonyl)(hydroxy)amino]-3-phenyl-propan-1-ol (7a)

To a stirred soln of the catalyst **4** (16 mg, 20 mol%) in CHCl₃ (1 mL) was added α , β -unsaturated aldehyde **2a** (33 mg, 0.25 mmol) and **1a** (40 mg, 0.3 mmol). The reaction was vigorously stirred at r.t. for 4 h. Next, the mixture was diluted with MeOH (1 mL) and cooled to 0 °C followed by addition of NaBH₄ (19 mg, 0.5 mmol). The mixture was then stirred for 10 min, quenched with 1 M HCl, and extracted with EtOAc. The organic layer was separated and dried (Na₂SO₄) and the solvent was removed. The residue was purified by column chromatography (silica gel, pentane–EtOAc, 4:1) to give **7a** (58 mg, 87%).

 $[\alpha]_{D}^{25}$ –52.0 (*c* 0.5, CHCl₃).

¹H NMR (300 MHz, CDCl₃): δ = 1.43 (s, 9 H), 2.02–2.11 (m, 1 H), 2.36–2.47 (m, 1 H), 3.76–3.81 (m, 2 H), 5.21 (dd, *J* = 10.8, 2.1 Hz, 1 H), 6.91 (br s, 1 H), 7.27–7.42 (m, 5 H).

¹³C NMR (75 MHz, CDCl₃): δ = 28.4, 34.2, 60.1, 60.3, 82.0, 127.4, 127.6, 128.5, 140.3, 157.2.

HRMS (ESI): m/z [M + Na]⁺ calcd for C₁₄H₂₁NNaO₄: 290.1363; found: 290.1355.

(-)-(3S)-3-Phenyl-2-aminopropanoic Acid (6b)¹¹

To a stirred soln of Cbz-protected isoxazolidinone **5b** (149 mg, 0.5 mmol) in MeOH (5 mL, 0.1 M), was added 10% (in weight) of Pd/C (10%). The reaction was stirred under H₂ (91 bar) overnight. Next, the crude reaction was filtered through a plug of Celite. The solvent was removed under reduced pressure to afford the pure β -amino acid **6b** (83 mg, 100%).

 $[\alpha]_{D}^{25}$ -6.9 (c 1, H₂O) {Lit.¹¹ $[\alpha]_{D}^{25}$ -6.9 (c 0.8, H₂O)}.

¹H NMR (400 MHz, D₂O/K₂CO₃): δ = 2.45–2.60 (m, 2 H), 4.27 (t, *J* = 7.2 Hz, 1 H), 7.30–7.40 (m, 5 H).

¹³C NMR (100 MHz, D₂O/K₂CO₃): δ = 46.7, 53.1, 127.6, 128.7, 128.8, 128.9, 179.4.

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