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## Direct Mannich Reaction of Glycinate Schiff Bases with *N*-(8-Quinolyl)sulfonyl Imines: A Catalytic Asymmetric Approach to *anti*- $\alpha$ , $\beta$ -Diamino Esters

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The wide-range significance of optically active  $\alpha,\beta$ -diamino acids as key structural components in bioactive compounds, as well as valuable synthetic intermediates, 1 has prompted the need for new general and efficient procedures for the asymmetric synthesis of this simple, yet polyfunctional, motif. Despite the catalytic asymmetric direct Mannich-type reaction of glycinate Schiff bases with imines being one of the most convergent routes for accessing this type of nonproteinogenic aminoacid derivatives, <sup>1a</sup> only a handful of protocols have been so far described in the literature.<sup>2</sup> Although high levels of syn-diastereoselectivity and enantiocontrol were achieved, this strategy had not been applied to the preparation of  $\alpha,\beta$ -diaminoester derivatives with *anti*-configuration<sup>3</sup> or those with an α-tetrasubstituted carbon stereocenter. To date, these limitations have been overcome with indirect approaches, mainly the catalytic asymmetric aza-Henry reaction, either between nitro compounds and  $\alpha$ -iminoesters<sup>4</sup> or between  $\alpha$ -nitro acetates and imines,<sup>5,6</sup> upon subsequent selective reduction of the nitro to amino group. Direct reaction of an  $\alpha$ -isothiocvanate N-acvl oxazolidine with N-tosvl imines has also been reported to provide protected anti- $\alpha$ ,  $\beta$ -diamino acids. As a useful complementary method, herein we describe a catalytic asymmetric approach to orthogonally protected anti-α,βdiamino esters, including those with a tetrasubstituted carbon at C-α,8 relying on the direct Mannich reaction of glycinate Schiff bases. The combined use of a Cu<sup>I</sup> complex of the commercially available Fesulphos ligand (1)9 as catalyst and readily available 8-quinolylsulfonyl-protected aldimines as substrates is key to attaining good reactivity, a broad scope, and high diastereo- and enantiocontrol.

Facing the challenge of achieving high diastereoselection, we first examined the influence of the protecting group at the iminic nitrogen in the  $Cu^{I}/(\pm)$ -Binap-catalyzed model reaction of Nbenzylideneglycine methyl ester (2a) with different imines of benzaldehyde (3a-i) in the presence of Et<sub>3</sub>N (10 mol%) as base (Table 1). Initial results were rather disappointing since the N-Bocprotected imine 3a, as well as the para-substituted arylsulfonyl imines **3b-d** and the heteroarylsulfonyl<sup>10</sup> derivatives **3e**<sup>9a,b,10b</sup> and 3f, 10a,c led to mixtures of (syn+anti)-4 with very poor diastereoselectivity (de = 0-20%, entries 1-6). The competitive formation of the 1,3-dipolar cycloaddition imidazolidine product 5 (as mixture of diastereomers) was also observed with the sulfonyl imines **3b-f**. However, the bulkier o-nosyl imine 3g led cleanly to a mixture of 4g with encouraging anti-diastereocontrol (syn/anti = 30:70, entry 7). Moreover, an excellent level of anti-diastereoselection was attained with the 8-quinolylsulfonyl-protected imine 3h<sup>10a,c</sup> (8:92 mixture of syn/anti-4h, entry 8). Interestingly, the sterically similar 1-naphthylsulfonyl imine 3i afforded a complex mixture of four products under identical conditions (entry 9), showing that the presence of the nitrogen of the 8-quinolyl moiety is crucial for good anti-diastereocontrol.

Asymmetric catalysis was then investigated by screening a number of well established chiral ligands (5 mol%) in the reaction of **2a** with imine **3h** under the previous conditions. For better chiral

**Table 1.** Influence of Imine Protecting Group in the Cu<sup>I</sup>-Catalyzed Mannich Reaction of Glycinate Schiff Base **2a** with Imines **3a**<sup>-1</sup>

entry	PG (imine)	time (h)	syn- <b>4</b> /anti- <b>4</b> / <b>5</b> <sup>b</sup>
1	Boc (3a)	6	60:40: <sup>c</sup>
2	Ts ( <b>3b</b> )	3	$43:52:5^{c,d}$
3	p-Nosyl ( <b>3c</b> )	5	$38:37:25^{c,d}$
4	$4-OMe(C_6H_4)SO_2-(3d)$	2	$40:40:20^{c,d}$
5	$(2-Thienyl)SO_{2}-(3e)$	3	17:48:35 <sup>c,d,c</sup>
6	(2-Pyridyl)SO <sub>2</sub> - ( <b>3f</b> )	2	$35:35:30^{c,d}$
7	o-Nosyl (3g)	5	30:70:
8	(8-Quinolyl)SO <sub>2</sub> - (3h)	2	8:92:
9	$(1-Naphthyl)SO_2-(3i)$	2	complex mixture

 $^a$  Conditions: **2a** (1 equiv), **3** (1.1 equiv), Cu(CH<sub>3</sub>CN)<sub>4</sub>ClO<sub>4</sub> (5 mol%), ( $\pm$ )-Binap (5 mol%), Et<sub>3</sub>N (10 mol%), CH<sub>2</sub>Cl<sub>2</sub>, rt.  $^b$  By NMR from the crude reaction mixture.  $^c$  *syn/anti* configuration not established.  $^d$  Imidazolines **5** were obtained as diastereomeric mixtures.

Table 2. Cul/Fesulphos-Catalyzed Enantioselective Direct Mannich Reaction of Iminoester 2a with Sulfonyl Imine 3h

entry	Cu <sup>I</sup> (x mol%)	solvent	t (°C)	anti/synª	yield (%) <sup>b</sup>	ee (%) <sup>c</sup>
1	Cu(CH <sub>3</sub> CN) <sub>4</sub> ClO <sub>4</sub> (5)	CH <sub>2</sub> Cl <sub>2</sub>	rt	92:8	79	91
2	CuOTf (5)	$CH_2Cl_2$	rt	94:6	77	91
3	$Cu(CH_3CN)_4PF_6$ (5)	$CH_2Cl_2$	rt	93:7	85	92
4	$Cu(CH_3CN)_4PF_6$ (5)	CH <sub>3</sub> CN	rt	91:9	62	90
5	$Cu(CH_3CN)_4PF_6$ (5)	toluene	rt	94:6	79	91
6	$Cu(CH_3CN)_4PF_6$ (5)	THF	rt	96:4	80	92
7	$Cu(CH_3CN)_4PF_6$ (5)	THF	-40	99:1	74	96
8	$Cu(CH_3CN)_4PF_6$ (5)	THF	-78	99:1	73	96
9	$Cu(CH_3CN)_4PF_6$ (3)	THF	-40	98:2	71	91
$10^d$	$Cu(CH_3CN)_4PF_6$ (1)	THF	rt	91:9	35	

 $^{\it a}$  By NMR and/or HPLC.  $^{\it b}$  Isolated yield.  $^{\it c}$  By chiral HPLC.  $^{\it d}$  12 h reaction time.

HPLC separation of products, in this study the resulting α-imino ester **4h** was reduced to its α-benzylamino derivative *anti-***6** upon addition of NaBH<sub>4</sub> (1.2 equiv)/EtOH to the crude reaction mixture. Among 12 tested ligands, <sup>11</sup> only Fesulphos (1)<sup>9,12</sup> led to a high *anti*-diastereo- (*anti/syn* = 92:8) and enantioselectivity (91% ee, Table 2, entry 1). Further refinement of reaction conditions by exploring other solvents and copper salts led to improved catalyst performance (entries 2–8). The highest diastereo- (*anti/syn* = 99: 1) and enantiocontrol (96% ee) were attained in THF at -40 or -78 °C with Cu(CH<sub>3</sub>CN)<sub>4</sub>PF<sub>6</sub>/(R)-1 (5 mol%, entries 7 and 8).

Catalyst loading could be reduced to 3 mol% while maintaining high reactivity and stereocontrol (91% ee, entry 9). Further decrease to 1 mol% led to a dramatic drop of reactivity (entry 10).

**Table 3.** Structural Variations at the  $\alpha$ -iminoester and Sulfonyl Imine

entry	R <sup>1</sup>	R <sup>2</sup> (sulfonyl imine)	product	anti/syn <sup>b</sup>	yield (%)c	ee (%) <sup>d</sup>
$1^e$	Н	4-OMeC <sub>6</sub> H <sub>4</sub> ( <b>7h</b> )	16	98:2	72	95
$2^e$	Н	$4-ClC_6H_4$ (8h)	17	98:2	92	93
$3^f$	Н	$2-\text{MeC}_6\text{H}_4$ ( <b>9h</b> )	18	98:2	80	99
$4^e$	Н	2-Naph (10h)	19	97:3	75	94
$5^e$	Н	2-Furyl ( <b>11h</b> )	20	95:5	71	85
$6^e$	Н	3-Pyridyl ( <b>12h</b> )	21	98:2	72	90
$7^e$	Н	Cinnamyl (13h)	22	85:15	64	25
$8^e$	Me	Ph ( <b>3h</b> )	23	90:10	55	89
$9^e$	Me	4-ClC <sub>6</sub> H <sub>4</sub> ( <b>8h</b> )	24	97:3	65	87
$10^e$	Me	2-thienyl ( <b>14h</b> )	25	94:6	52	96
$11^e$	Me	2-BrC <sub>6</sub> H <sub>4</sub> ( <b>15h</b> )	26	89:11	61	92

<sup>a</sup> (8-Q) = 8-Quinolyl; Cu<sup>I</sup> = Cu(CH<sub>3</sub>CN)<sub>4</sub>PF<sub>6</sub>. <sup>b</sup> By NMR and/or HPLC. <sup>c</sup> Isolated yield. <sup>d</sup> By chiral HPLC. <sup>e</sup> Reaction at -78 °C. <sup>f</sup> Reaction at −40 °C.

Table 3 summarizes the evaluation of the scope under the optimized conditions. A survey of electronically and sterically varied aryl and heteroaryl N-(8-quinolyl)sulfonyl aldimines<sup>13</sup> revealed a high degree of stereochemical fidelity in their reaction with glycinate **2a**, showing excellent diastereoselectivity (95:5–98:2) and  $\geq$ 90% ee in most cases<sup>14</sup> (entries 1–6, products 16–21). The  $\alpha,\beta$ unsaturated imine 13h, from cinnamaldeyde, was the only exception to this trend, providing moderate diastereocontrol and very low asymmetric induction (25% ee, entry 7). Pleasingly, the reaction was equally successful with the  $\alpha$ -substituted imino ester derived from L-alanine 2b, a kind of nucleophile not yet reported in this reaction even though it generates  $\alpha,\beta$ -diaminoacid derivatives tetrasubstituted at C- $\alpha$ . In the four tested reactions (entries 8–11), 2b provided the corresponding Mannich product (23-26) with similar diastereo- and enantioselectivity (87-96% ee) to that observed from 2a, albeit the chemical yields were somewhat lower (52–65%). The relative and absolute configuration of the Mannich products was established by X-ray diffraction analysis of a crystal of pure anti-6<sup>15</sup> (>99% ee) obtained by recrystallization (from CH<sub>2</sub>Cl<sub>2</sub>-hexane) of a 98:2 anti/syn mixture formed in the reaction of 2a with imine 3h (2 mmol scale) using 3 mol% of Cu<sup>I</sup>-Fesulphos.

Scheme 1. Orthogonal Deprotection of the Amino Groups and Chemical Correlation<sup>a</sup>

Scheme 1 exemplifies the sequential amino deprotection of the  $\alpha,\beta$ -diaminoester adducts under mild conditions. The optically active amino ester 27 was obtained in one pot from the reaction of 2a with 3h upon smooth acid hydrolysis of the crude resulting imino ester (2S,3S)-anti-4h. The transformation of 27 into the known enantiopure urea 28,16 confirming otherwise the stereochemical

assignment established by X-ray diffraction analysis, 15 was achieved in good yield by treatment with triphosgene (CH<sub>2</sub>Cl<sub>2</sub>, 0 °C to rt, 2 h) followed by easy cleavage of the 8-quinolylsulfonyl group with an excess of Mg turnings in MeOH, this latter step being accelerated under sonication.

In summary, we have developed a route to protected anti- $\alpha,\beta$ diamino esters compatible with the generation of a tetrasubstituted carbon at C- $\alpha$ . The choice of Fesulphos—Cu<sup>I</sup> as catalyst and N-(8quinolyl)sulfonyl as protecting group at the imine substrate are the key elements for achieving efficient control of both diastereo- and enantioselectivity (typically ≥90% ee). Further investigations to understand the exact role of the 8-quinolyl group are underway.

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Supporting Information Available: Experimental procedures and characterization data of new compounds, copies of NMR spectra. This material is available free of charge via the Internet at http://pubs.acs.org.

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