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Four new trace phenolic glycosides from Curculigo orchioides

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ORIGINAL ARTICLE

Four new trace phenolic glycosides from Curculigo orchioides

Ai-Xue Zuo^{ab}, Yong Shen^{ab}, Xue-Mei Zhang^a, Zhi-Yong Jiang^a, Jun Zhou^a, Jun Lü^c and Ji-Jun Chen^a*

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Four new trace phenolic glycosides named orcinosides D (1), E (2), F (3), and G (4) were isolated from the rhizomes of *Curculigo orchioides* Gaertn. Based on comprehensive spectroscopic analyses including IR, FAB-MS, HR-ESI-MS, 1D- and 2D NMR (HSQC, HMBC), their structures were elucidated as orcinol-1-*O*- β -D-xylopyranoside (1), orcinol-1-*O*- β -D-apiofuranosyl-(1 \rightarrow 2)- β -D-glucopyranoside (2), orcinol-3-*O*- β -D-apiofuranosyl-1-*O*- β -D-glucopyranoside (3), and 1-*O*- β -D-glucopyranosyl-4-ethoxyl-3-hydroxymethylphenol (4).

Keywords: orcinosides D-G; phenolic glycosides; Curculigo orchioides

1. Introduction

Curculigo orchioides Gaertn., a traditional Chinese medicine known as 'Xian-Mao' in Chinese, is widely distributed in China, India, Malaya, Japan, and Australia. Its rhizomes were reported to possess the properties of warming kidney, invigorating yang, expelling cold, and eliminating dampness, and used to cure impotence, enuresis, cold sperm, cold pain of back and knee, and numbness of the limb [1]. Previous phytochemical studies on the rhizomes revealed the presence of cycloartane triterpenes [2], phenolic glycosides [3], and chlorophenolic glucosides [4]. Lakshmi et al. [5] reported that phenols and phenolic glycosides played an important biological role, being responsible for stimulating the immune response by acting on the macrophages and the lymphocytes. Additionally, Wu *et al.* [6] claimed that phenolic glycoside showed potent antioxidative activities. To find more biologically active substances from *C. orchioides*, we further undertook the investigation to furnish four new trace phenolic glycosides. This paper deals with the isolation and structural elucidation of four new trace orcinosides D (1), E (2), F (3), and G (4) from the rhizomes of *C. orchioides* (Figure 1).

2. Results and discussion

Compound 1 was isolated as a colorless powder, $[\alpha]_D^{27.7} - 33.33$ (c = 0.15, MeOH). Its molecular formula was determined to be $C_{12}H_{16}O_6$ based on m/z291.0637 [M+Cl]⁻ in the negative HR-

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Figure 1. The structures of compounds 1-4.

ESI-MS spectrum. The negative FAB-MS exhibited a quasi-molecular ion peak at $m/z 255 [M-H]^{-}$ and a fragment-ion peak at m/z 123 [M-C₅H₉O₄]⁻, suggesting the presence of a pentose sugar moiety in compound 1. The IR absorption bands at 3424 and 1619, 1500, 1468 cm^{-1} implied the existence of a hydroxyl and an aromatic ring. The ¹H NMR spectrum (Table 1) displayed one methyl at $\delta_{\rm H}$ 2.22 (3H, s, H-7), three aromatic proton signals at $\delta_{\rm H}$ 6.31 (1H, br s, H-2), 6.37 (1H, br s, H-4), 6.29 (1H, br s, H-6), and one anomeric proton resonance at $\delta_{\rm H}$ 4.78 (1H, d, J = 7.4 Hz, H-1[']). Analyses of the NMR spectral data of compound 1 suggested that the structure of compound 1 was very similar to that of orcinol glucopyranoside isolated from C. orchioides [7] except for a different sugar unit in compound 1. The carbon signals due to the sugar moiety at $\delta_{\rm C}$ 102.8, 74.7, 77.7, 71.0, 66.9 in the ¹³C NMR spectrum (Table 2), which matched those of methyl- β -D-xylopyranoside [8], revealed that the sugar moiety was a β -Dxylopyranosyl unit. Hydrolysis of compound 1 gave orcinol and xylose by comparison with the authentic samples on TLC and PC. The β -linkage of the xylosyl unit was verified by the J value in the ¹H NMR spectrum at $\delta_{\rm H}$ 4.78 (1H, d, J = 7.4 Hz, H-1[']). The correlation of H-1['] with C-1 in the HMBC experiment (Figure 2) established that the xylopyranosyl unit was linked at C-1 of the aglycone. Consequently, compound **1** was deduced as orcinol-1-O- β -D-xylopyranoside, named orcinoside D.

Compound 2 was obtained as a colorless powder and assigned the molecular formula C₁₈H₂₆O₁₁ based on the negative HR-ESI-MS spectrum at m/z 417.1404 $[M-H]^{-}$. The IR spectrum showed absorption bands at 3431 cm^{-1} (hydroxyl) and 1627, 1506, 1464 cm^{-1} (aromatic ring). Compound 2 was hydrolyzed with 2 M H₂SO₄ in methanol to yield orcinol, glucose, and apiose identified by TLC and PC comparison with the authentic samples. In the ¹H NMR spectrum of compound **2**, a methyl at $\delta_{\rm H}$ 2.21 (3H, s, H-7), three aromatic protons at $\delta_{\rm H}$ 6.30 (1H, br s Hz, H-2), 6.37 (1H, br s, H-4), 6.27 (1H, br s, H-6), and two anomeric proton signals at $\delta_{\rm H}$ 4.89 (1H, d, J = 7.3 Hz, H-1'), 5.44 (1H, d, J = 1.2 Hz, H-1") were observed. Comparing the ¹³C NMR spectral data of compound 2 with those of orcinol glucopyranoside [7] revealed that compound 2 had one more set of apiofuranosyl signals at $\delta_{\rm C}$ 110.7, 78.0, 80.8, 75.5, 66.1 [4] than orcinol glucopyranoside. Moreover, C-2' was downfield-shifted to $\delta_{\rm C}$ 78.1 compared to $\delta_{\rm C}$ 74.8 in orcinol glucopyranoside [7], which implied that the additional apiofuranosyl moiety was located at the C-2' of compound 2. The cross-peaks between

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Table 1. ¹H NMR data of compounds $\mathbf{1}^{a}$, $\mathbf{2}^{b}$, $\mathbf{3}^{b}$, and $\mathbf{4}^{b}$ in CD₃OD, δ in ppm, J in Hz.

		δ _H		
No.	1	2	3	4
5	6.31 (1H, br s)	6.30 (1H, br s)	6.37 (1H, br s)	6.78 (1H, d, J = 3.0 Hz)
4	6.37 (1H, br s)	6.37 (1H, br s)	6.39 (1H, br s)	
5				7.06 (1H, d, J = 8.8 Hz)
9	6.29 (1H, br s)	6.27 (1H, br s)	6.29 (1H, br s)	6.66 (1H, dd, J = 8.8, 3.0 Hz)
7	2.22 (3H, s)	2.21 (3H, s)	2.22 (3H, s)	4.47 (1H, d, J = 12.5 Hz)
				4.65 (1H, d, $J = 12.5 \text{Hz}$)
OCH ₂ CH ₃				3.55 (2H, q, J = 7.0 Hz)
				1.21 (3H, $t, J = 7.0 \text{ Hz}$)
1'	4.78 (1H, d, $J = 7.4$ Hz)	4.89 (1H, d, $J = 7.3$ Hz)	4.78 (1H, d, $J = 7.3$ Hz)	4.69 (1H, d, $J = 7.6$ Hz)
2'	3.38–3.40 (m)	3.93-3.94 (m)	3.41-3.42 (m)	3.42-3.43 (m)
3/	3.38-3.40 (m)	3.56-3.58 (m)	3.91-3.92 (m)	3.35-3.36 (m)
4'	3.54-3.56 (m)	3.34–3.37 (m)	3.29 - 3.30 (m)	3.35–3.36 (m)
5'	3.90 (1H, dd, J = 11.4, 5.3 Hz)	3.36-3.58 (m)	3.91–3.92 (m)	3.42-3.43 (m)
	3.31-3.32 (m)			
6'		3.86 (1H, dd, J = 12.0, 1.5 Hz)	3.96–3.97 (m)	3.87 (1H, dd, J = 12.5, 1.5 Hz)
		3.68 (1H, dd, J = 12.0, 5.0 Hz)	3.74-3.76 (m)	3.68 (1H, dd, J = 12.5, 5.0 Hz)
$1^{\prime\prime}$		5.44 (1H, d, $J = 1.2 \text{Hz}$)	4.97 (1H, d, $J = 2.4$ Hz)	
2"		3.94 (1H, d, J = 4.0 Hz)	3.41 - 3.42 (overlapped)	
4″		4.04 (1H, d, $J = 9.5 \text{Hz}$)	3.95 (1H, d, $J = 9.7$ Hz)	
5"		3.34 (2H, s)	3.41-3.42 (overlapped)	
Notes: ^a Recorded	d at 400 MHz. ^b Recorded at 500 MHz.			

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	$\delta_{\rm C}$			
No.	1	2	3	4
1	159.9 (s)	159.3 (s)	159.9 (s)	153.9 (s)
2	102.2 (d)	102.0 (d)	102.2 (d)	116.7 (d)
3	159.3 (s)	159.3 (s)	158.9 (s)	130.7 (s)
4	109.7 (d)	109.4 (d)	109.9 (d)	150.3 (s)
5	141.2 (s)	141.3 (s)	141.1 (s)	119.2 (d)
6	111.2 (d)	111.2 (d)	111.2 (d)	116.0 (d)
7	21.7 (q)	21.7 (q)	21.7 (q)	68.7 (t)
OCH ₂ CH ₃				66.8 (t)
OCH_2CH_3				15.4 (q)
1'	102.8 (d)	100.8 (d)	102.1 (d)	104.6 (d)
2'	74.7 (d)	78.1 (s)	74.6 (d)	75.1 (d)
3'	77.7 (d)	78.7 (d)	77.7 (d)	78.2 (d)
4'	71.0 (d)	71.4 (d)	71.3 (d)	71.5 (d)
5'	66.9 (t)	78.5 (d)	76.7 (d)	78.1 (d)
6'		62.5 (t)	61.5 (t)	62.7 (t)
1″		110.7 (d)	110.7 (d)	
2"		78.0 (d)	77.8 (d)	
3″		80.8 (s)	80.3 (s)	
4″		75.5 (t)	74.9 (t)	
5″		66.1 (t)	65.6 (t)	

Table 2. ¹³C NMR data of compounds 1^a , 2^b , 3^b , and 4^b in CD₃OD, δ in ppm.

Notes: ^aRecorded at 100 MHz. ^bRecorded at 125 MHz.

H-1' and C-1, H-1" and C-2' in the HMBC (Figure 2) spectrum also implied that the inner glucopyranose was attached at C-1 of the aglycone and the apiofuranose was connected at C-2' of the inner glucose. Thus, compound **2** was determined as orcinol-1-O- β -D-apiofuranosyl-(1 \rightarrow 2)- β -D-glucopyranoside, named orcinoside E.

Compound 3, colorless powder, had the molecular formula $C_{18}H_{26}O_{11}$ as determined by the negative HR-ESI-MS spectrum at m/z 417.1407 [M-H]⁻. The IR spectrum showed absorption $3422 \, \mathrm{cm}^{-1}$ and 1615. bands at $1462 \,\mathrm{cm}^{-1}$, implying the existence of hydroxyl and aromatic groups. Hydrolysis of compound 3 with 2M H₂SO₄ in methanol liberated orcinol, glucose, and apiose identified by comparison with authentic samples on TLC and PC. The ¹H NMR spectrum (Table 1) showed three aromatic ring proton signals at $\delta_{\rm H}$ 6.37 (1H, br s, H-2), 6.39 (1H, br s, H-4), 6.29 (1H, br s, H-4), one methyl signal at $\delta_{\rm H}$ 2.22 (3H, s, H-7), together with two anomeric proton resonances at $\delta_{\rm H}$ 4.78 (1H, d, J = 7.3 Hz, H-1') and 4.97 (1H, d, J = 2.4 Hz, H-1"). The ¹³C NMR spectrum (Table 2) contained 18 carbon signals, involving an orcinol unit signal ($\delta_{\rm C}$ 159.9, 102.2, 158.9, 109.9, 141.1, 111.2, 21.7), a glucopyranosyl unit resonance ($\delta_{\rm C}$ 102.1, 74.6, 77.7, 71.3, 76.7, 61.5), and an apiofuranosyl moiety ($\delta_{\rm C}$ 110.7, 77.8, 80.3, 74.9, 65.6). Analysis of the NMR spectral data demonstrated that compound 3 had one more apiofuranosyl unit than orcinol glucopyranoside [7]. The HMBC correlations between the proton at $\delta_{\rm H}$ 4.97 (1H, d, J = 2.4 Hz, H-1'') and the carbon at $\delta_{\rm C}$ 158.9 (s, C-3) suggested that the additional apiofuranosyl moiety was located at C-3. Accordingly, compound 3 was deduced as orcinol-3-O-β-D-apiofuranosyl-1-O-β-D-glucopyranoside, named orcinoside F.



Figure 2. The key HMBC correlations of compounds 1-4.

Compound 4 was obtained as a colorless powder. Its molecular formula was deduced as C₁₅H₂₂O₈ by the negative HR-ESI-MS spectrum at m/z 365.1010 $[M+C1]^{-}$. The IR spectrum displayed absorption bands due to a hydroxyl group at 3415 cm^{-1} and an aromatic ring at 1628, 1500, 1453 cm⁻¹. Hydrolysis of compound 4 yielded glucose identified by comparing with the authentic sample on PC. The ¹H NMR spectrum of compound 4 presented proton signals ascribable to an $-OCH_2CH_3$ at δ_H 3.55 (2H, q, J = 7.0 Hz), 1.21 (3H, t, J = 7.0 Hz), a --CH₂OH at $\delta_{\rm H}$ 4.47 (1H, d, J = 12.5 Hz, H-7a), 4.65 (1H, d, J = 12.5 Hz, H-7b), and a trisubstituted aromatic ring at $\delta_{\rm H}$ 6.78 (1H, d, *J* = 3.0 Hz, H-2), 7.06 (1H, d, J = 8.8 Hz, H-5, 6.66 (1H, d, J = 8.8, 3.0 Hz, H-6), together with an anomeric proton signal at $\delta_{\rm H}$ 4.69 (1H, d,

J = 7.6 Hz, H-1'), indicating a β -linkage of the sugar moiety. In the ¹³C NMR spectrum for compound 4, the carbon signals assignable to an aromatic ring at $\delta_{\rm C}$ 153.9, 116.7, 130.7, 150.3, 119.2, 116.0, an $-OCH_2CH_3$ at δ_C 66.8, 15.4, and a -CH₂OH at $\delta_{\rm C}$ 68.7 were observed, in addition to a set of B-D-glucopyranosyl groups at $\delta_{\rm C}$ 104.6, 75.1, 78.2, 71.5, 78.1, 62.7, which were in agreement with the ¹³C NMR spectral data of methyl-β-Dglucopyranoside [8]. The NMR spectral data of compound 4 were similar to those of 4-ethoxy-3-hydroxymethylphenol [9], except that compound 4 had one more glucopyranose moiety. The long-range correlations between the anomeric proton at $\delta_{\rm H}$ 4.69 (1H, d, J = 7.6 Hz, H-1[']) and C-1 in the HMBC experiment ascertained that the glucopyranosyl unit was located at C-1. As a result, compound 4 was corroborated as 1-*O*-β-D-glucopyranosyl-4-ethoxyl-3-hydroxymethylphenol, named orcinoside G.

3. Experimental

3.1 General experimental procedures

Optical rotations were performed on a Horiba SEPA-300 polarimeter (Tokyo, Japan). IR spectra were recorded on a Bio-Rad FTS-135 spectrometer (Richmond, CA, USA) with KBr pellets, ν in cm^{-1} . UV spectra were measured on a UV-210A spectrometer (Shimadzu, Japan). NMR spectra were conducted on Bruker AV-400 or DRX-500 spectrometers (Karlsruhe, Germany) with TMS as the internal standard. Chemical shifts (δ) were expressed in ppm and coupling constants (J) in Hz. FAB-MS was recorded on a VG-Auto-spec-3000 mass spectrometer (Manchester, UK). ESI-MS and HR-ESI-MS were taken on an API Qstar-Pulsar-1 mass spectrometer (Applied Biosystems/MDS Sciex, Ontario, Canada). Column chromatography (CC) separations were performed on silica gel (200-300 mesh; Qingdao Meigao Chemical Co., Ltd, Qingdao, China), Al₂O₃ (Shanghai Wusi Chemical Reagents Company, Shanghai, China), D₁₀₁ macroporous resins (Tianjin Pesticide Chemical Company, Tianjin, China), Sephadex LH-20 (Pharmacia Fine Chemical Co. Ltd, Uppsala, Sweden), and Lichroprep RP-18 $(40-63 \,\mu m;$ Merck. Darmstadt, Germany). Fractions were monitored by TLC and visualized by spraying with 10% H_2SO_4 in EtOH followed by heating.

3.2 Plant material

The rhizomes of *C. orchioides* Gaertn. were collected in Wenshan County, Yunnan Province, China, in November 2005, and authenticated by Prof. Dr Li-Gong Lei, Kunming Institute of Botany, Chinese Academy of Sciences. A voucher specimen (No. 20051106) has been deposited in the Group of Anti-virus and Natural Medicinal Chemistry, Kunming Institute of Botany, Chinese Academy of Sciences.

3.3 Extraction and isolation

The air-dried and powdered rhizomes of C. orchioides (200 kg) were extracted with 70% EtOH (each 1000 liters, 2h) three times under reflux to yield an extract which was combined and concentrated to a small volume (600 liters) and subjected to CC (macroporous resin D101, 200 kg), with gradient elution of H₂O, 10% EtOH-H₂O, 40% EtOH-H₂O, 70% EtOH-H₂O, 90% EtOH $-H_2O$ to afford five fractions (I–V). Fraction II (10% EtOH-H₂O eluted, 800 g) was subjected to Al₂O₃ CC (8kg, 14×50 cm), subsequently eluted with EtOAc-EtOH-H₂O (9:1:0.1), EtOAc-EtOH-H₂O (8:2:0.2), and EtOAc-EtOH- H_2O (7:3:0.2) to afford fractions A–C.

Fraction A (260 g) was successively subjected to RP-18 CC $(1 \text{ kg}, 6 \times 60 \text{ cm})$ eluted with MeOH $-H_2O$ (1:9) to afford fractions 1-3. Fraction 2 (2.0 g) was performed on silica gel CC (100 g, CHCl₃- $3.4 \times 27 \,\mathrm{cm}$) eluted with MeOH-H₂O (9:1:0.1) to give a residue (1.2 g) which was subjected to Sephadex LH-20 CC (53 g, 2.2×62 cm) eluted with CHCl₃-MeOH (1:1), and further purified by silica gel CC (15 g. 1×15 cm) with EtOAc-EtOH-H₂O (9:1:0.1) as the solvent to yield compounds 4 (10 mg) and 1 (9 mg). Fraction 3 (3.0 g) was applied to a silica gel CC $(100 \text{ g}, 3.4 \times 27 \text{ cm})$ eluted with CHCl₃-MeOH $-H_2O$ (8.5:1.5:0.15) to give four portions. The second portion (2.0 g) was purified on **RP-18** CC (120 g, 2.5×33 cm) eluted with MeOH-H₂O (3:97) to afford compounds 2 (12 mg)and **3** (1.2 g).

3.3.1 Orcinoside D (1)

Colorless powder. $[\alpha]_D^{27.7} - 33.33$ (*c* = 0.15, MeOH). UV (MeOH) λ_{max}^{MeOH} (log ε): 279 (3.38) nm. IR (KBr) ν_{max} : 3424, 2921, 1619, 1500, 1468, 1325, 1073, 1043, 835, 670, 567 cm⁻¹. ¹H and ¹³C NMR spectral (CD₃OD) data are listed in Tables 1 and 2. FAB-MS (negative): *m/z* 255 [M-H]⁻, 239 [M-OH]⁻, 123 [M-H-xylose]⁻. HR-ESI-MS (negative): *m/z* 291.0637 [M+Cl]⁻ (calcd for C₁₂H₁₆O₆Cl, 291.0653).

3.3.2 Orcinoside E (2)

Colorless powder. $[\alpha]_D^{20.8} - 66.67$ (c = 0.3, MeOH). UV (MeOH) λ_{max}^{MeOH} (log ε): 273 (3.21) nm. IR (KBr) ν_{max} : 3431, 2923, 1627, 1506, 1464, 1318, 1113, 1071, 1033, 577, 563 cm⁻¹. ¹H and ¹³C NMR spectral (CD₃OD) data are listed in Tables 1 and 2. FAB-MS (negative): m/z417 [M-H]⁻. HR-ESI-MS (negative): m/z 417.1404 [M-H]⁻ (calcd for C₁₈H₂₅O₁₁, 417.1396).

3.3.3 Orcinoside F(3)

Colorless powder. $[\alpha]_{D}^{20.9} - 74.24$ (c = 0.3, MeOH). UV (MeOH) λ_{max}^{MeOH} (log ε): 279 (3.30) nm. IR (KBr) ν_{max} : 3422, 2923, 1615, 1462, 1426, 1321, 1172, 1071, 994, 828, 677, 650, 571 cm⁻¹. ¹H and ¹³C NMR spectral (CD₃OD) data are listed in Tables 1 and 2. FAB-MS (negative): m/z 417 [M-H]⁻; HR-ESI-MS (negative): m/z 417.1407 [M-H]⁻ (calcd for C₁₈H₂₅O₁₁, 417.1396).

3.3.4 Orcinoside G(4)

Colorless powder. $[\alpha]_D^{27.4} - 35.16$ (c = 0.28, MeOH). UV (MeOH) λ_{max}^{MeOH} (log ε): 286 (3.31) nm. IR (KBr) ν_{max} : 3415, 2974, 1628, 1500, 1453, 1383, 1293, 1074, 1044, 894, 803, 633, 582 cm⁻¹. ¹H and ¹³C NMR spectral (CD₃OD) data are listed in Tables 1 and 2. FAB-MS (negative): m/z 329 [M-H]⁻; HR-ESI-MS (negative): m/z 365.1010 [M+Cl]⁻ (calcd for C₁₅H₂₂O₈Cl, 365.1003).

3.4 Acid hydrolysis

Each of compounds 1-4 (2 mg) was dissolved in MeOH (1.0 ml) and 4 M H₂SO₄ (1.0 ml) solution and hydrolyzed under reflux for 2 h. The hydrolysate was allowed to cool, diluted with 2 ml H₂O, and extracted with 2 ml EtOAc. The aqueous layer was neutralized with aqueous Ba(OH)₂ and concentrated in vacuo to give a residue, in which xylose (from 1), glucose (from 2-4) and apiose (from 2 and 3) were identified by comparing with authentic samples on PC [BuOH-EtOAc-H₂O 4:1:5, upper layer, $R_{\rm f} = 0.60$ (xylose); BuOH– EtOAc $-H_2O$ 4:1:5, upper layer, $R_{\rm f} = 0.45$ (glucose); PhOH-H₂O, 4:1, $R_{\rm f} = 0.55$ (apiose) on PC, respectively]. Orcinol (from 1-3) was detected from the EtOAc layer by TLC comparison with an authentic sample (silica gel, CHCl₃–MeOH 9:1).

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