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Letter

Total Synthesis of PF1163B

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Abstract A total synthesis of PF1163B has been achieved. The vinylogous Mukaiyama aldol reaction of ketene silyl *N*,O-acetal **6** (*ent*-**6**) mediated by excess TiCl₄ proceeded with saturated aldehydes to give adduct **10** in moderate yield with moderate stereoselectivity. The major isomer is the diastereomer that was provided by using one equivalent of TiCl₄. The Birch reduction of α,β -unsaturated imide **4**, possessing a less hindered side chain, gave **12** in good stereoselectivity by employing 2-isopropylbenzimidazole as a bulky proton source. After elongation of the carbon chain and connection with the amino acid part, we accomplished a total synthesis of PF1163B. These methods constitute a concise synthetic route to obtain polyketides as well as depsipeptides.

Key words PF1163B, total synthesis, vinylogous Mukaiyama aldol reaction, bulky proton source, depsipeptide, stereoswitching

PF1163B (Figure 1) was isolated as a potent antifungal antibiotic from a fermentation broth of *Penicilium* sp. by Sasaki and coworkers.¹ It has been discovered as a potent inhibitor of ergosterol synthesis and has an IC_{50} value of 34



ng/ml.¹ PF1163B is a depsipeptide possessing a polyketide chain with a methyl group and a hydroxy group in a separate position.¹ Despite its attractive biological activity and structure, only two precedent syntheses have been reported.^{2,3} In a series of our synthetic studies of polyketides,⁴ we have been interested in the structure and biological activities of PF1163B. Herein, we present our synthesis of PF1163B.



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Our synthetic plan of PF1163B is disclosed in Scheme 1. The macrolide was divided into known segments **2** and **3**.² Polyketide segment **3** would be synthesized from α , β -unsaturated imide **4** by stereoselective reduction and carbon elongation. α , β -Unsaturated imide **4** might be constructed by a vinylogous Mukaiyama aldol reaction.

Our previous works on vinylogous Mukaiyama aldol reactions have been developed to proceed in a stereoselective manner (Scheme 2).⁴ Treatment of *E*,*E*-vinylketene silyl *N*,*O*-acetal **7** with aldehyde in the presence of 1.1 equivalents of TiCl₄ gave anti adduct **9** in high stereoselectivity, whereas employment of an excess amount (4.0 equivalents) of TiCl₄ afforded syn adduct **8** as a predominant product.⁴ Therefore, a stereo-switching synthesis of polypropionates became possible by employing the same chiral synthon and the same Lewis acid. Additionally, the reaction of vinylketene silvl N,O-acetal ent-6, the terminal methyl group-missing substrate, in the presence of 1.1 equivalents of TiCl₄ also proceeded to yield 11 in a stereoselective manner.⁴ However, no report has described the reaction of ent-6 with an excess amount of TiCl₄. And our desired reaction is the one providing 10, the diastereomer of 11. Therefore, we first explored the conditions using vinylketene silyl N,O-acetal ent-**6** with an excess amount of TiCl₄ to develop a new stereoswitching reaction.



Scheme 2 Previous works on the vinylogous Mukaiyama aldol reaction with vinylketene silyl *N*,*O*-acetals **7** and *ent*-**6**

The vinylogous Mukaiyama aldol reaction of *ent-6* with an excess amount of TiCl₄ is disclosed in Table 1. The best results to obtain **10** were achieved with use of four equivalents of TiCl₄. The reaction with paraldehyde (**a**) gave adducts in low stereoselectivity (entry 1). However, hexanal (**b**) gave δ -hydroxyimide **10**, the diastereomer of known **11**, by using one equivalent of TiCl₄, in moderate stereoselectivity (entry 2). In the case of the reaction employing 5',5'-dimethyl derivative of *ent-6* as a chiral building block (entry 3), the reaction gave adducts **10** and **11** in low yield without stereoselectivity (entry 3). The reaction with *ent-6* and 3Downloaded by: Washington University. Copyrighted material.





^a Determined by 400 MHz ¹H NMR spectroscopy.

phenylpropanal (**c**) gave **10** in moderate yield with moderate stereoselectivity (entry 4). Branched aldehydes including β -branched aldehyde (isovaleraldehyde, **d**) and α branched aldehyde (isobutyraldehyde, **e**) also gave adducts in comparable yields and stereoselectivities (entries 5 and 6). However, unfortunately, the reactions with unsaturated aldehydes including crotonaldehyde (**f**) and benzaldehyde (**g**) gave no adducts (entries 7 and 8) and underwent decomposition to afford the corresponding α , β -unsaturated imide. Therefore, the vinylogous Mukaiyama aldol reaction with vinylketene silyl *N*,*O*-acetal *ent*-**6** in the presence of an excess amount of TiCl₄ was found to proceed with saturated aldehydes to give adduct **10** in moderate yield with moderate stereoselectivity.

On the basis of these results, we started the synthesis of the propionate segment of PF1163B (Scheme 3). The reaction of **6** with hexanal **5** in the presence of four equivalents of TiCl₄ proceeded as shown in Table 1 to give in 54% yield an inseparable 4:1 mixture of *ent*-**10b** and *ent*-**11b**. The ad-

ducts were converted into the corresponding TBS ethers, which were separated by column chromatography. The next reduction of the α , β -unsaturated imide **4** led to the C10 stereogenic center. In our total synthesis of mycocerosic acid, we found that 2-methylbenzimidazole worked as a bulky proton source to protonate the intermediate enolate in the Birch reduction of an α , β -unsaturated imide containing Evans chiral auxiliary in a stereoselective manner.⁵ Therefore, we examined the protonation with 2-alkylated benzimidazoles (Table 2).^{6,7}

When NH_4Cl was employed as a proton source, no stereoselectivity was observed (Table 2, entry 1). 2-Methylbenzimidazole, reported as an excellent proton source in the total synthesis of mycocerosic acid, gave the reduced products in 64% as a 4.4:1 mixture with the desired isomer **12** as a major isomer (entry 2). 2-Isopropylbenzimidazole gave **12** in comparable yield with 6:1 ratio (entry 3). More bulky 2-*tert*-butylbenzimidazole afforded **12** in moderate yield with low stereoselectivity (entry 4). Therefore, we employed 2-isopropylbenzimidazole as the best proton source in the Birch reduction of **4**.⁷

Synthesis of the polyketide segment **3** and completion of the total synthesis is described in Scheme 4. DIBAL reduction of **4** directly gave aldehyde **13**, which was subjected to Horner–Wadsworth–Emmons reaction⁸ with **14** gave $\alpha,\beta,\gamma,\delta$ -unsaturated ester **15**. Hydrogenation of unsaturated ester **15** gave saturated ester **16**, which was subjected to desilylation to give polyketide segment **3**. Spectral data of **3** were identical to those of reported **3**.² Thus, configurations of C10 and C13 positions were confirmed. Polyketide **3** was coupled with tyrosine derivative **2**⁹ by Yamaguchi esterification to give ester **17**. Following known procedure,² removal of the *tert*-butyl groups and the TBS group was followed by intramolecular cyclization to provide PF1163B (**1**).¹⁰

In conclusion, we have achieved a total synthesis of PF1163B. Stereogenic centers in the polyketide moiety were constructed by the vinylogous Mukaiyama aldol reaction of ketene silyl *N*,*O*-acetal **6** mediated by excess TiCl₄ and the Birch reduction proceeded with 2-isopropylbenzimidazole







^a Isolated yield. ^b Ratio of isolated compo

^b Ratio of isolated compounds.

as the proton source. Therefore, the separate stereogenic centers in PF1163B were constructed with one chiral synthon, the vinylketene silyl *N*,*O*-acetal containing the Evans chiral auxiliary. This is a short strategy to prepare mediumsize compounds. After elongation of the carbon chain and connection with the amino acid part, we accomplished a total synthesis of PF1163B (10 steps from **6**, 5.5% overall yield). These methods give a concise synthetic route to obtain polyketides as well as depsipeptides.



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Supporting Information

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- (10) Synthesis of PF1163B (1) from Acyclic 17

To a solution of compound 17 (85.9 mg, 0.114 mmol) in CH₂Cl₂ (4.0 mL) at 0 °C was added TFA (1.0 mL, 98%). The reaction mixture was stirred for 35 min at rt and then concentrated. The residue was then taken up in CH₂Cl₂ (64.0 mL) and cooled to 0 °C. Et₃N (0.19 mL, 1.368 mmol) was added dropwise, followed by the addition of BOP-Cl (174.1 mg, 0.684 mmol). The reaction mixture was stirred at 0 °C for 48 h. Then, it was concentrated and NaHCO₃ aq. (2.0 mL) was added. The aqueous layer was extracted two times with $CHCl_3$ (4.0 mL × 2) and the combined organic layers were concentrated, dried with Na₂SO₄, and purified by silica gel chromatography (n-hexane/EtOAc = 5:1) to give PF1163B(1)(27.0 mg, 0.058 mmoles, 50% over 2 steps) as a colorless oil. $R_f = 0.48$ (*n*-hexane/EtOAc = 1:1). $[\alpha]_D^{23} = -109.6$ (*c* 0.49, MeOH). ¹H NMR (400 MHz, CDCl₃): δ = 7.21–7.05 (2 H, m), 6.88-6.79 (2 H, m), 5.84-5.75 (0.6 H, m), 4.94-4.78 (1 H, m), 4.60-4.54 (0.27 H, m), 4.09-4.01 (2 H, m), 3.99-3.91 (2 H, m), 3.60-3.14 (4 H, m), 3.04-2.90 (3 H, m), 2.84-2.61 (2 H, m), 2.47-1.96 (4 H, m), 1.54-1.03 (16 H, m), 0.92-0.79 (6 H, m). 13C NMR (100 MHz, CDCl₃): δ = 173.4, 171.1, 170.2, 157.2, 130.2, 129.7, 129.1, 114.9, 114.4, 75.3, 69.1, 69.0, 61.5, 55.4, 35.1, 35.0, 34.0, 33.7, 33.6, 33.5, 31.7, 31.6, 29.4, 29.3, 28.7, 26.5, 25.0, 20.5, 14.0. HRMS-ESI: m/z [M + Na]⁺ calcd for C₂₇H₄₃O₅NNa: 484.3033; found: 484.3033. IR (KBr): 3400, 2950, 2929, 1731, 1632, 1512, 1248, 1220, 1078, 772 cm⁻¹.

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