

# Silver(I)-Catalyzed Regioselective Synthesis of Triazole Fused-1,5-Benzoxazocinones

Indrajeet J. Barve,<sup>†</sup> Tushar Ulhas Thikekar,<sup>†</sup> and Chung-Ming Sun<sup>\*,†,§</sup>

<sup>†</sup>Department of Applied Chemistry, National Chiao-Tung University, 1001 Ta-Hseuh Road, Hsinchu 300-10, Taiwan <sup>§</sup>Department of Medicinal and Applied Chemistry, Kaohsiung Medical University, 100, Shih-Chuan first Road, Kaohsiung 807-08, Taiwan

**Supporting Information** 



**ABSTRACT:** An efficient and regioselective synthesis of novel 1,2,3-triazole-fused-1,5-benzoxazocinones through intramolecular cyclization of substituted ethynyl triazoyl benzoic acids was explored. A crucial precursor 5-iodo-1,2,3-triazole benzoate was obtained from substituted 2-azido benzoic acid esters in a single step through a Copper-Catalyzed Azide–Alkyne Cycloaddition (CuAAC) reaction using a CuI/NBS catalytic system. A carbon–carbon triple bond was installed through a Sonogashira coupling reaction by various terminal alkynes. Finally, the 1,4,5-substituted ethynyl triazoyl benzoic acids were cyclized by a AgOTf-mediated intramolecular cyclization to afford 8-*endo-dig* 1,2,3-triazole-fused-1,5-benzoxazocinones exclusively.

**R** ecently, triazole-fused polycyclic heterocycles have gained much attention owing to their broad spectrum of bioactivities;<sup>1</sup> 1,2,3-triazoles particularly fused with heterocycles at the 1,5-positions such as triazolo 1,5-benzodiazepin-2-one **A** were reported to inhibit serine protease.<sup>2</sup> Triazolo-benzoxazepine **B** exhibits promising anticancer activity against multiple cancer cell lines,<sup>3</sup> whereas [1,2,3]-triazolo[1,5-a][1,4]benzodiazepine **C** shows antimicrobial effects as well as potent anticancer activity against the A549 lung adenocarcinoma cancer cell line (Figure 1).<sup>4</sup> In recent years, several synthetic





strategies have been reported for the synthesis of annulated 1,2,3-triazoles at 1,5-positions with other heterocyclic motifs.<sup>5–10</sup> However, a synthetic approach to access 1,2,3-triazolefused benzoxazocinones is less explored.<sup>1b,3</sup> Therefore, the development of a new and efficient synthetic method is greatly needed to construct novel triazole-fused heterocycles. The biological significance of fused 1,2,3-triazoles prompted us to develop a simple protocol comprising minimum steps for the preparation of 1,2,3-triazole fused benzoxazocinones. The regular CuAAC reactions are limited to the synthesis of only 1,4-disubstituted 1,2,3-triazoles. In this regard, 5-iodo-1,-2,3-triazole is a useful synthetic precursor since its iodine atom can undergo further transformation into various functional groups to give corresponding trisubstituted triazoles, thereby diversification and functionalization of the 1,2,3-triazole pharmacophore can be achieved.<sup>11</sup>

Intramolecular addition reaction of carboxylic acids to proximate carbon–carbon triple bonds offers an efficient route to the construction of oxygen-containing functionalized heterocycles (Figure 2).<sup>12</sup> According to Baldwin's rules, both *exo-dig* and *endo-dig* cyclizations are possible. Hence, much attention has been focused toward the selective cyclization of an enyne carboxylic acid system.<sup>13</sup> We herein report a



Figure 2. Endo-dig and exo-dig cyclization mode of enyne carboxylic acid system.

Received: March 27, 2017

irradiation.

## Table 1. Optimization of the Reaction Conditions for Intramolecular Cyclization of Acid 1<sup>a</sup>



entry	catalyst (mol %)	base	additive	solvent	temp (°C)	time (h)	product	yield (%) <sup>b</sup>
1	PTSA (10)	_	-	CHCl <sub>3</sub>	reflux	16	N.R.	0
2	pyridine	_	_	_	100	16	N.R.	0
3	DBU (10)	_	_	CHCl <sub>3</sub>	reflux	16	N.R.	0
4	DBU (100)	_	_	CHCl <sub>3</sub>	100	0.34	2a	trace <sup>c</sup>
5	AgOTf (10)	_	_	CHCl <sub>3</sub>	reflux	16	2a	10
6	AgOTf (20)	_	_	CHCl <sub>3</sub>	reflux	16	2a	30
7	AgOTf (50)	K <sub>2</sub> CO <sub>3</sub>	_	CHCl <sub>3</sub>	reflux	16	2a	50
8	AgOTf (100)	K <sub>2</sub> CO <sub>3</sub>	_	CHCl <sub>3</sub>	100	0.34	2a	85 <sup>c</sup>
9	AgOTf (10)	NaOH	_	CHCl <sub>3</sub>	100	0.34	2a	32
10	$AgSbF_6$ (10)	_	_	CHCl <sub>3</sub>	reflux	16	N.R.	0
11	AuCl (10)	K <sub>2</sub> CO <sub>3</sub>	_	CH <sub>3</sub> CN	reflux	16	N.R.	0
12	$AuCl_3$ (10)	K <sub>2</sub> CO <sub>3</sub>	_	CH <sub>3</sub> CN	reflux	16	N.R.	0
13	AgOTf (10)	K <sub>2</sub> CO <sub>3</sub>	PPh <sub>3</sub>	CHCl <sub>3</sub>	100	0.34	2a	48
14	AuCl/AgOTf (5)	_	PPh <sub>3</sub>	CHCl <sub>3</sub>	100	0.34	2a	trace
15	$Ag_2CO_3$ (20)	_	_	CHCl <sub>3</sub>	100	0.34	2a	20
16	AgOTf (10)	K <sub>2</sub> CO <sub>3</sub>	_	toluene	130	0.25	2a	87 <sup>c</sup>
<sup>a</sup> Reaction c	onditions: 1 (0.27 mmo	1). AgOTf (10	mol %). K <sub>2</sub> CO	2 (2 equiv), sol	vent (2 mL). N.B	. = No Reactio	n. <sup>b</sup> Isolated vi	eld. <sup>c</sup> Microway

regioselective synthesis of novel 1,2,3-triazole benzoxazocinones from substituted anthranilic acids. The key step of this strategy includes preparation of a crucial intermediate 5-iodo-1,2,3-triazole benzoate by the CuI/NBS catalytic system and regioselective intramolecular cyclization under the influence of Lewis acid (AgOTf), respectively.

The required precursor acids 1 were prepared by modification of the reported procedure (see Supporting Information). With the key intermediate acid 1 in hand, we next studied the intramolecular cyclization using a variety of catalysts, and the results are summarized in Table 1. Our attempted cyclization of acid 1 under the influence of PTSA, pyridine, or DBU was unsuccessful, and only starting material 1 was recovered (Table 1, entries 1–4). When acid 1 was subjected for cyclization in the presence of 10 mol % of AgOTf in refluxing chloroform for 16 h, a pale yellow solid was obtained in 10% yield with the recovery of acid 1 (Table 1, entry 5). The structure of the isolated solid was identified as (Z)-3,5-diphenyl-7*H*-benzo[*c*][1,2,3]triazolo[1,5-*e*][1,5]-oxazocin-7-one **2a**. Its structure was confirmed by X-ray crystallographic analysis (Figure 3).

The ORTEP diagram of 2a revealed that the double bond and sp<sup>2</sup> carbon atom of the carboxylic group forces the eightmember ring to adopt a boat conformation having minimum



Figure 3. ORTEP diagram of compound 2a.

transannular nonbonded interactions within the ring. This preliminary discovery motivated us to improve the yield of cyclized compound 2a. Accordingly, an increase in the catalyst loading of AgOTf up to 20 mol %, yielded 2a in only 30% yield (Table 1, entry 6). When acid 1 was treated with AgOTf and K<sub>2</sub>CO<sub>3</sub> in refluxing chloroform, the isolated yield of the compound 2a increased up to 50% (Table 1, entry 7); a further increase in the quantity of AgOTf and K<sub>2</sub>CO<sub>3</sub> up to 100 mol % did not improve the yield. When the cyclization of 1 was carried out with AgOTf (100 mol %) and K<sub>2</sub>CO<sub>3</sub> (2 equiv) in chloroform under microwave irradiation at 100 °C for 20 min, an 85% conversion of starting material 1 into 8-endo-dig cyclized product 2a was achieved (Table 1, entry 8). Even though desired compound 2a was obtained in an excellent yield, a stoichiometric amount of AgOTf is required. Hence, effort toward the conversion of acid 1 into cyclized compound 2a under the influence of a catalytic amount of silver or gold salt was carried out. Employment of a combination of strong base NaOH and AgOTf for cyclization of 1 under microwave irradiation resulted in cyclized product 2a in only 32% yield (Table 1, entry 9). The use of  $AgSbF_6$  in chloroform under reflux conditions resulted in zero conversion (Table 1, entry 10). Surprisingly, the excellent carbophilic gold salts such as AuCl and AuCl<sub>3</sub> with a combination of K<sub>2</sub>CO<sub>3</sub> in acetonitrile by either conventional heating or microwave irradiation failed to yield compound 2a (Table 1, entries 11-12). Utilization of PPh<sub>3</sub> as an additive in combination with a catalytic amount of AgOTf and  $K_2CO_3$  under microwave conditions afforded 2a in moderate yield (Table 1, entry 13). Similarly, combining AuCl with cocatalyst AgOTf and additive PPh<sub>3</sub> yielded compound 2a in a trace amount (Table 1, entry 14). Furthermore, only 20% of product 2a was obtained when Ag<sub>2</sub>CO<sub>3</sub> was employed alone (Table 1, entry 15). Finally, the use of 10 mol % of AgOTf in the presence of  $K_2CO_3$  (2 equiv) in toluene under microwave

#### **Organic Letters**

irradiation afforded cyclized compound 2a in excellent yield (Table 1, entry 16).

To investigate the scope of this protocol, a variety of substituted acids 1 were prepared from substituted anthranilic acids S1 and alkynes S4. The corresponding 8-*endo-dig* products 2 were obtained in excellent yields and the current protocol works well with aliphatic alkynes as well as with aromatic alkynes having electron-donating and -withdrawing groups on the phenyl ring (Scheme 1).

Scheme 1. Substrate Scope for the Synthesis of 1,2,3-Triazole-Fused-1,5,-Benzoxazocinones  $2^{a}$ 



<sup>a</sup>Reaction conditions: 1 (0.27 mmol), AgOTf (10 mol %),  $K_2CO_3$  (2 equiv), solvent (2 mL).

It is interesting to note that, in all cases, only 8-*endo-dig* cyclized products **2** were obtained. The structures of the products **2b**–**t** were confirmed by the two signals at 6–6.5 and  $\sim$ 100 ppm respectively in the <sup>1</sup>H and <sup>13</sup>C NMR spectra.

The results suggested that activation of a C-C triple bond and the subsequent intramolecular addition of carboxylic acid were highly regioselective. In order to rationalize the regioselectivity in the cyclization of acid 1, we propose a plausible mechanism in Scheme 2. In the first step, carbophilic AgOTf coordinates with the triple bond of acid 1, which results in the enhancement of the electrophilicity of the alkyne.





Subsequent anti-*endo-dig* intramolecular nucleophilic attack by a carboxylate anion on Ag(I)-alkyne complex A forms B. Consequent protonolysis of the C–Ag bond of B produces 8*endo-dig* product 2 with regeneration of AgOTf. The formation of highly regioselective 8-*endo-dig* intermediate B can be rationalized through the polarization effect exerted by the electron-withdrawing moieties present at the carbon of the alkyne (Figure 4).<sup>14,15</sup> In this case, electron-deficient triazole-



Figure 4. *Endo-dig* intramolecular nucleophilic addition via triazolemediated polarization of alkyne.

mediated polarization of the alkyne increases the electrophilicity of the terminal carbon atom of the alkyne which facilitates *endo-dig* nucleophilic addition of the carboxylate anion to afford 8-*endo-dig* products **2**.

In conclusion, we have developed a synthetic protocol for the synthesis of novel 1,2,3-triazole-fused-1,5-benzoxazocinones **2** via a regioselective intramolecular cyclization of substituted ethynyl triazoyl benzoic acids **1**. The salient feature of this protocol involves the one-pot synthesis of crucial precursor 5-iodo-1,2,3-triazole benzoates **S5** from substituted 2-azido benzoic acid esters **S3** using the CuI/NBS catalytic system. The 5-iodo-1,2,3-triazole benzoates **S5** were further functionalized by a carbon–carbon triple bond through Sonogashira coupling with various terminal alkynes. Finally, the 1,4,5-substituted-1,2,3-triazole benzoic acids **1** were subjected to an intramolecular cyclization using AgOTf/K<sub>2</sub>CO<sub>3</sub> to afford exclusively 8-*endo-dig* 1,2,3-triazole fused benzoxazocinones **2** in excellent yields.

### ASSOCIATED CONTENT

#### Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.or-glett.7b00907.

Experimental procedures, characterization data, spectral data, and X-ray data for compound **2a** (PDF) Crystallographic data (CIF)

## AUTHOR INFORMATION

#### **Corresponding Author**

\*E-mail: cmsun@mail.nctu.edu.tw.

## ORCID 🔍

Chung-Ming Sun: 0000-0002-1804-1578

## Notes

The authors declare no competing financial interest.

## ACKNOWLEDGMENTS

The authors thank Ministry of Science and Technology (MOST) of Taiwan for financial assistance and the authorities of the National Chiao Tung University for providing the laboratory facilities.

#### **Organic Letters**

## REFERENCES

 (a) Mohapatra, D. K.; Maity, P. K.; Shabab, M.; Khan, M. I. Bioorg. Med. Chem. Lett. 2009, 19, 5241-5245. (b) Chandrasekhar, S.; Seenaiah, M.; Kumar, A.; Reddy, C. R.; Mamidyala, S. K.; Kumar, C. G.; Balasubramanian, S. Tetrahedron Lett. 2011, 52, 806-808.
 (c) Putapatri, S. R.; Kanwal, A.; Balasubramanian, S.; Banerjee, S. K.; Kantevari, S. Org. Biomol. Chem. 2014, 12, 8415-8421.

(2) Granger, B. A.; Kaneda, K.; Martin, S. F. Org. Lett. 2011, 13, 4542-4545.

(3) Banerji, B.; Pramanik, S. K.; Sanphui, P.; Nikhar, S.; Biswas, S. C. Chem. Biol. Drug Des. **2013**, 82, 401–409.

(4) Sudhapriya, N.; Nandakumar, A.; Arun, Y.; Perumal, P. T.; Balachandran, C.; Emi, N. RSC Adv. 2015, 5, 66260-66270.

(5) (a) Sudhir, V. S.; Kumar, N. Y. P.; Baig, R. B. N.; Chandrasekaran, S. J. Org. Chem. 2009, 74, 7588–7591. (b) Bera, S.; Panda, G. Org. Biomol. Chem. 2014, 12, 3976–3985. (c) Afraj, S. N.; Chen, C.; Lee, G. H. RSC Adv. 2014, 4, 26301–26308. (d) Ning, Y.; Wu, N.; Yu, H.; Liao, P.; Li, X.; Bi, X. Org. Lett. 2015, 17, 2198– 2201.

(6) (a) Bertelli, L.; Biagi, G.; Giorgi, I.; Livi, O.; Manera, C.; Scartoni, V.; Lucacchini, A.; Giannaccini, G.; Barili, P. L. *Eur. J. Med. Chem.* **2000**, 35, 333–341. (b) Baigi, G.; Giorgi, I.; Livi, O.; Scartoni, V.; Betti, L.; Giannaccini, G.; Trincavelli, M. L. *Eur. J. Med. Chem.* **2002**, 37, 565–571.

(7) (a) Chowdhury, C.; Mukherjee, S.; Das, B.; Achari, B. J. Org. Chem. 2009, 74, 3612–3615. (b) Mishra, K. B.; Tiwari, V. K. J. Org. Chem. 2014, 79, 5752–5762.

(8) Liu, Z.; Zhu, D.; Luo, B.; Zhang, N.; Liu, Q.; Hu, Y.; Pi, R.; Huang, P.; Wen, S. Org. Lett. **2014**, *16*, 5600–5603.

(9) (a) Donald, J. R.; Wood, R. R.; Martin, S. F. ACS Comb. Sci. 2012, 14, 135–143. (b) Guggenheim, K. G.; Toru, H.; Kurth, M. J. Org. Lett. 2012, 14, 3732–3735. (c) Hussain, M. K.; Ansari, M. I.; Kant, R.; Hajela, K. Org. Lett. 2014, 16, 560–563.

(10) Lauria, A.; Patella, C.; Dattolo, G.; Almerico, A. M. J. Med. Chem. 2008, 51, 2037-2046.

(11) (a) Hein, J. E.; Tripp, J. C.; Krasnova, L. B.; Sharpless, K. B.; Fokin, V. V. Angew. Chem., Int. Ed. 2009, 48, 8018–8021. (b) Morris, J. C.; Chiche, J.; Grellier, C.; Lopez, M.; Bornaghi, L. F.; Maresca, A.; Supuran, C. T.; Pouysségur, J.; Poulsen, S. A. J. Med. Chem. 2011, 54, 6905–6918. (c) Bogdan, A. R.; James, K. Org. Lett. 2011, 13, 4060– 4063. (d) Schulman, J. M.; Friedman, A. A.; Panteleev, J.; Lautens, M. Chem. Commun. 2012, 48, 55–57. (e) Worrell, B. T.; Hein, J. E.; Fokin, V. V. Angew. Chem., Int. Ed. 2012, 51, 11791–11794. (f) Oakdale, J. S.; Sit, R. K.; Fokin, V. V. Chem. - Eur. J. 2014, 20, 11101–11110. (g) Li, L.; Zhang, G.; Zhu, A.; Zhang, L. J. Org. Chem. 2008, 73, 3630–3633.

(12) (a) García-García, P.; Fernández-Rodríguez, M. A.; Aguilar, E. Angew. Chem., Int. Ed. 2009, 48, 5534–5537. (b) Taskaya, S.; Menges, N.; Balci, M. Beilstein J. Org. Chem. 2015, 11, 897–905. (c) Fang, G.; Bi, X. Chem. Soc. Rev. 2015, 44, 8124–8173. (d) Sekine, K.; Sadamitsu, Y.; Yamada, T. Org. Lett. 2015, 17, 5706–5709. (e) Sekine, K.; Yamada, T. Chem. Soc. Rev. 2016, 45, 4524–4532.

(13) (a) Uchiyama, M.; Ozawa, H.; Takuma, K.; Matsumoto, Y.; Yonehara, M.; Hiroya, K.; Sakamoto, T. Org. Lett. 2006, 8, 5517– 5520. (b) Yoshikawa, T.; Shindo, M. Org. Lett. 2009, 11, 5378–5381.
(c) Nagendiran, A.; Verho, O.; Haller, C.; Johnston, E. V.; Bäckvall, J. E. J. Org. Chem. 2014, 79, 1399–1405.

(14) Nolla-Saltiel, R.; Robles-Marín, E.; Porcel, S. Tetrahedron Lett. 2014, 55, 4484–4488.

(15) Jiménez-Tenorio, M.; Puerta, M. C.; Valerga, P.; Moreno-Dorado, F. J.; Guerra, F. M.; Massanet, G. M. *Chem. Commun.* **2001**, 2324–2325.