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An Enantiomerically Pure Tricyclic Isoindoline System by Cyclisation of Tricarbonyl[η6-(R)-N-cyanomethyl-4-phenyloxazolidine]chromium

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Abstract: The isolation of isoindoline 4 by oxidative trapping indicates that a tricyclic η^5 complex is a contributing form of the anion derived from the title compound 3. Copyright © 1996 Published by Elsevier Science Ltd

Over the past ten years, we have studied the asymmetric alkylation and the 1,3-dipolar cycloadditions of the chiral synthon (R)-N-cyanomethyl-4-phenyloxazolidine 1.1 In general, the stereochemical induction in alkylation reactions is modest (20-60% d.e. for 2), 2 probably due to a lack of structural rigidity in the intermediate carbanion, whose reactive centre is located on a poorly-restrained side-chain of the rigid ring system. To explain the preponderant S absolute configuration at the new chiral centre, a preferred reactive conformer I (cis relative configuration, exo cyanomethyl anion) was invoked for the intermediate carbanion. 2a

Herein we report our first observations on the corresponding tricarbonylchromium(0) complex 3. Our interest in this compound arose from the knowledge that $(\eta^6$ -arene)chromium compounds react with carbanions derived from alkyl nitriles to give negatively-charged η^5 intermediates.³ Some intramolecular examples are known, for which the bicyclic intermediates have been trapped,⁴ but no simple cases have been described for γ -cyanoarenes, nor for heterocyclic systems. Furthermore, in simple intermolecular cases, the subsequent addition of an electrophile to the η^5 adduct results in alkylation of the nitrile α -carbon with concomitant cleavage and regeneration of the neutral $(\eta^6$ -arene)chromium system.³ If such phenomena could be induced for 3, the highly-original intermediate II should allow much better stereochemical control during alkylation: temporary formation of a sigma bond effectively rigidifies the anion and completely blocks one side of the reactive centre. Thus 3 was prepared from 1 using standard procedures⁵ (49% yield) in order to address two questions: does a tricyclic anionic η^5 complex II form on deprotonation? And, if so, does it allow improved stereoselectivity during alkylation?

Efficient deprotonation of 3 was achieved at -70 °C with LDA (2 equiv) in HMPA/THF or *n*-BuLi (2 equiv) in THF, as evidenced by quantitative incorporation of deuterium in recovered 3 upon treatment with excess D₂O. Oxidative trapping-decomplexation of 3 anion solutions generated in this way was carried out with an excess of iodine, and standard work up gave compound 4 as a single stereomer in about 20% yield. 2D NOESY NMR experiments showed correlations which suggested an R configuration at the new chiral centre, precisely as would be expected from the cyclisation of a *cis-exo* conformer of 3 anion.

Under the present reaction conditions, the isolated yield of 4 remains low, which we interpret as an indication of a low concentration of II in the anion solution. In consequence, 3 has no greater ability than 1 for stereochemical control during alkylation. Indeed, as shown in the Table, yields and diastereomeric excesses of alkylation products are more or less comparable for the two compounds.

Nonetheless, intermediate II is clearly accessible, apparently with total stereochemical control, via an unprecedented intramolecular ortho-directed cyclisation of a $(\gamma$ -cyanoarene)chromium complex. The potential of this structure for asymmetric alkylation, and the novelty of the chiral isoindoline derivatives such as 4 which can be derived therefrom, justify continuing efforts towards the optimisation of its formation.⁶

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Table	1 2		3 3	
RX	Yield (%)§	d.e. (%)*	Yield (%)§	d.e. (%)*
MeI	70	36	68	40
EtI	91	58	61	30
PhCH ₂ Br	67	44	52	43
Allyl Br	63	26	64	36
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- * In all cases the major diastereomer has an S configuration at the new chiral centre
- § Isolated pure after chromatography

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Table

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- 6. All new products showed satisfactory spectral and/or analytical data.