This article was downloaded by: [University of Hawaii at Manoa] On: 06 January 2015, At: 19:33 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry

Publication details, including instructions for authors and subscription information: <u>http://www.tandfonline.com/loi/lsyc20</u>

Synthesis of Azetidin-2-Ones from Imines and Acids Using Trichloroacetonitrile-Triphenylphosphine Reagent

Vidyesh V. Govande a , M. Arun a , A. R. A. S. Deshmukh a & B. M. Bhawal a

^a Division of Organic Chemistry (Synthesis), National Chemical Laboratory, Pune, 411 008, India Published online: 04 Dec 2007.

To cite this article: Vidyesh V. Govande , M. Arun , A. R. A. S. Deshmukh & B. M. Bhawal (2000) Synthesis of Azetidin-2-Ones from Imines and Acids Using Trichloroacetonitrile-Triphenylphosphine Reagent, Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry, 30:22, 4177-4182, DOI: 10.1080/00397910008087035

To link to this article: http://dx.doi.org/10.1080/00397910008087035

PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and

are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms & Conditions of access and use can be found at http://www.tandfonline.com/page/terms-and-conditions

SYNTHESIS OF AZETIDIN-2-ONES FROM IMINES AND ACIDS USING TRICHLOROACETONITRILE-TRIPHENYLPHOSPHINE REAGENT

Vidyesh V. Govande, M. Arun, A. R. A. S. Deshmukh, B. M. Bhawal*

Division of Organic Chemistry (Synthesis), National Chemical Laboratory, Pune - 411 008, India.

Abstract: A one step synthesis of azetidin-2-ones (3a-j) has been described by the reaction of imines (1) with acids (2) in presence of trichloroacetonitrile, triphenylphosphine and triethylamine.

Among the several methods for the synthesis of β -lactams, the cycloaddition reaction of ketenes with imines (Staudinger reaction) for the construction of β -lactam ring has found wide acceptance.¹ This is mainly because of its simplicity, predictability of stereochemical outcome and proven utility of this method for the synthesis of a large number of monocyclic, bicyclic, tricyclic and spirocyclic β -lactams.²

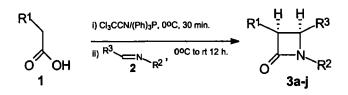
The ketenes are usually generated from acid halides (preformed or generated *in situ*) in the presence of tertiary amines. Alternatively acid activating

^{*} To whom correspondence should be addressed

reagents, like ethyl chloroformates,³ trifluoroacetic anhydride,⁴ p-toluenesulfonylchloride,⁵ phosphorus derived reagents,⁶ Mukaiyama reagent,⁷ cyanuric chloride,⁸ and several others¹ have been used. We now wish to report an application of trichloroacetonitrile-triphenylphosphine as a mild reagent for *in situ* generation of acid chlorides in the synthesis of azetidin-2-ones (β -lactams) *via* ketene-imine cycloaddition reaction.

In situ generated acid chloride from acid (1), by the reaction of trichloroacetonitrile and triphenylphosphine,⁹ was reacted with imine (2) at 0°C to give β -lactam (3) in good yields (Scheme 1). The cycloaddition reaction was found to be stereoselective and only *cis*- β -lactam formation was observed.

Scheme 1



To explore the generality of this method, several substituted β -lactams (**3a-j**) were synthesized in good yields (Table-1). This method can also be used for the synthesis of β -lactams derived from acids, which are sensitive to mineral acids or thionyl chloride (see entry 10).

Experimental

All ¹H NMR Spectra were recorded in CDCl₃ on a Brucker AC 200 and Brucker MSL 300 spectrometers and chemical shifts were reported in ppm downfield from tetramethylsilane. Infrared spectra were recorded on a Perkin-

Entry No.	R ¹	R ²	R ³	Product	Yield (%) ^a	М.р. (°С) ^ь
1	PhO	Ph	PMP	3a	61	150-151 (149-150) ¹⁰
2	PhO	PMP	Ph	3b	59	185-186
3	PhO	PMP	PMP	3c	56	166-167
4	MeO	PMP	Ph	3d	65	159-161
5	PhO	PMP	Styryl	3e	62	178-180
6	PhO	Ph	Styryl	3f	65	193-194 (193-195) ¹¹
7	PhO	m-Tolyl	Styryl	3g	71	161-162
8	PhthN	PMP	Styryl	3h	61	191-193 (192-194) ^{6e}
9	MeO	PMP	Styryl	3i	62	139-141
10	H" H	Styryl	PMP	3ј	70 ^c	232-233 ^d

Table 1 : Synthesis of β -lactams (3a-j) from acids (1) and imines (2).

^a Isolated yield of pure products. ^b The figures in parenthesis refers to the literature melting points. ^c Isolated yield of diastereomeric mixture (7:3). ^d M.p. of major diastereomer obtained in 45% yield by single crystallization from acetone-pet. ether (3:7).

Elmer Infracord spectrophotometer Model 599-B using sodium chloride optics. Melting points were determined on a Thermonik Campbell melting point apparatus and were uncorrected. Optical rotations were recorded on a JASCO-181 digital polarimeter under standard conditions. Methylene chloride was distilled over P_2O_5 under argon. Silica gel (SD's 60-120 mesh) was used for column chromatography.

A general procedure for the preparation of β -lactams (3a-j). To a solution of acid 1 (1 mmol) and trichloroacetonitrile (2 mmol) in dry CH₂Cl₂ (5 ml), a

solution of triphenylphosphine (2 mmol) in dry CH_2Cl_2 (3ml) was added at 0°C and stirred for 30 min. This mixture was then slowly added to a solution of imine 2 (0.5 mmol) and triethylamine (6 mmol) in CH_2Cl_2 (20 ml) at 0°C over a period of 15 min. The reaction mixture was allowed to warm-up to room temperature and stirred further for 12 h. It was washed successively with water (20 ml) satd. NaHCO₃ (20 ml) and brine (10 ml). The organic layer was dried (Na₂SO₄), concentrated and the product was purified by crystallization from methanol or column chromatography to give pure β -lactams (**3a-j**) in 56 to 71% yields.

4-p-Anisyl-3-phenoxy-1-phenylazetidin-2-one (3a). IR : 1739 cm⁻¹; ¹H NMR : δ 3.75 (s, 3H), 5.35 (d, J = 5.4 Hz, 1H), 5.50 (d, J = 5.4 Hz, 1H), 6.75-7.50 (m, 14H). Anal. Calcd. for C₂₂H₁₉NO₃: C, 76.49; H, 5.54; N, 4.05. Found C, 76.31; H, 5.18; N,3.85.

1-p-Anisyl-3-phenoxy-4-phenylazetidin-2-one (3b). IR : 1753 cm⁻¹; ¹H NMR : δ 3.75 (s, 3H), 5.40 (d, J = 5.4 Hz, 1H), 5.60 (d, J = 5.4 Hz, 1H), 6.70-7.50 (m, 14H). Anal. Calcd. for C₂₂H₁₉NO₃: C, 76.49; H, 5.54; N, 4.05. Found C, 76.23; H, 5.21; N, 3.93.

1,4-Bis(*p*-anisyl)-3-phenoxyazetidin-2-one (3c). IR : 1739 cm⁻¹; ¹H NMR : δ 3.85 (s, 6H), 5.40 (d, J = 5.4 Hz, 1H), 5.60 (d, J = 5.4 Hz, 1H), 6.80-7.50 (m, 13H). Anal. Calcd. for C₂₃H₂₁NO₄: C, 73.57; H, 5.64; N, 3.73. Found C, 73.20; H, 4.81; N,3.56.

1-*p***-Anisyl-3-methoxy-4-phenylazetidin-2-one (3d)**. IR : 1741 cm⁻¹; ¹H NMR : δ 3.15 (s, 3H), 3.70 (s, 3H), 4.75 (d, J = 5.5 Hz, 1H), 5.15 (d, J = 5.5 Hz, 1H), 6.70-7.50 (m, 9H). Anal. Calcd. for C₁₇H₁₇NO₃: C, 72.06; H, 6.01; N, 4.94. Found C, 71.96; H, 6.04; N, 4.75.

1-p-Anisyl-3-phenoxy-4-styrylazetidin-2-one (3e). IR : 1751 cm⁻¹; ¹H NMR : δ 3.75 (s, 3H), 5.00 (dd, J = 4.4 & 8.1 Hz, 1H), 5.50 (d, J = 4.4 Hz, 1H), 6.30 (dd, J

= 8.1 & 15.7 Hz, 1H), 6.75-7.50 (m, 15H). Anal. Calcd. for $C_{24}H_{21}NO_3$: C, 77.59; H, 5.70; N, 3.77. Found C, 77.38; H, 5.42; N, 3.80.

3-Phenoxy-1-phenyl-4-styrylazetidin-2-one (3f). IR : 1739 cm⁻¹; ¹H NMR : δ 5.60 (dd, J = 5.4 & 8.1 Hz, 1H), 6.15 (d, J = 5.4 Hz, 1H), 6.50 (dd, J = 8.1 & 15.0 Hz, 1H), 6.70 (d, J = 15 Hz, 1H), 6.75-7.80 (m, 15H). Anal. Calcd. for C₂₃H₁₉NO₂: C, 80.90; H, 5.61; N, 4.10. Found C, 80.51; H, 5.45; N, 3.87.

3-Phenoxy-4-styryl-1-*m***-tolylazetidin-2-one (3g)**. IR : 1755 cm⁻¹; ¹H NMR : δ 2.35 (s, 3H), 5.00 (dd, J = 5.2 & 8.1 Hz, 1H), 5.50 (d, J = 5.2 Hz, 1H), 6.20 (dd, J = 8.1 & 15.7 Hz, 1H), 6.75-7.50 (m, 15H). Anal. Calcd. for C₂₄H₂₁NO₂: C, 81.09; H, 5.96; N, 3.94. Found C, 81.00; H, 6.34; N, 3.77.

1-*p*-Anisyl-3-phthalimido-4-styrylazetidin-2-one (3h). IR : 1743, 1728 cm⁻¹; ¹H NMR : δ 3.75 (s, 3H), 5.00 (dd, J = 5.8 & 8.8 Hz, 1H), 5.67 (d, J = 5.8 Hz, 1H), 6.31 (dd, J = 8.8 & 16.1 Hz, 1H), 6.79 (d, J = 16.1 Hz, 1H), 6.86 (d, J = 8.8 Hz, 2H), 7.25-7.35 (m, 3H), 7.46 (d, J = 8.8 Hz, 2H), 7.71-7.85 (m, 4H).

1-*p*-Anisyl-3-methoxy-4-styrylazetidin-2-one (3i). IR : 1745 cm⁻¹; ¹H NMR : 8 3.50 (s, 3H), 3.70 (s, 3H), 4.70-4.85 (m, 2H), 6.30 (dd, J = 7 & 15 Hz, 1H), 6.75-7.00 (m, 3H), 7.20-7.50 (m, 7H). Anal. Calcd. for C₁₉H₁₉NO₃: C, 73.76; H, 6.19; N, 4.52. Found C, 73.38; H, 5.98; N, 4.35.

1-*p*-Anisyl-4-styryl-3-[3',7',7'-trimethyl-3'-methoxy]bicyclo(4.1.0)hept-4'-yloxy]azetidin-2-one (3j-major). IR : 1740 cm⁻¹; ¹H NMR : δ 0.71 (s, 3H), 0.87 (s, 3H), 1.02 (m, 1H), 1.27 (s, 3H), 1.55-2.15 (m, 6H), 3.27 (s, 3H), 3.75 (s, 3H), 4.70 (dd, J = 4.8 & 9.5 Hz, 1H), 5.42 (d, J = 4.8 Hz, 1H), 6.37 (dd, J = 8.5 & 14.4 Hz, 1H), 6.85 (m, 3H), 7.15-7.55 (m, 7H); [α]_D²⁵: -3.3 (c 1.2, CH₂Cl₂).

Acknowledgement: One of the authors (MA) thanks CSIR, New Delhi, for financial support.

References

- Georg, G.I.; Ravikumar, V. In "The Organic Chemistry of β-lactams" Georg, G.I., Ed.; VCH, New York 1993, p 295.
- a) Manhas, M.S.; Amin, S.G.; Bose, A.K. Heterocycles 1976, 5, 669. b) Caroll, R.D.; Reed, L.L. Tetrahedron Lett. 1975, 3435.
- Bose, A.K.; Manhas, M.S.; Amin, S.G.; Kapur, J.C.; Kreder, J.; Mukkavilli, L.; Ram, B.; Vincent, J.E. *Tetrahedron Lett.* 1979, 2771.
- Bose, A.K.; Kapur, J.C.; Sharma, S.D.; Manhas, M.S. Tetrahedron Lett. 1973, 2319.
- 5. Miyake, M.; Tokutake, N.; Kirisawa, M. Synthesis, 1983, 833.
- a) Cossio, F.P.; Lecca, B.; Palomo, C. J. Chem. Soc., Chem. Commun. 1987, 1743. b) Arrieta, A.; Lecca, B.; Cossio, F.P.; Palomo, C. J. Org. Chem. 1988, 53, 3784. c) Manhas, M.S.; Lal, B.; Amin, S.G.; Bose, A.K. Synth. Commun. 1976, 6, 435. d) Shridhar, D.R.; Ram, B.; Narayana, V.L. Synthesis 1982, 63.
 e) Cossio, F.P.; Ganboa, I.; Garcia, J.M.; Lecca, B.; Palomo, C. Tetrahedron Lett. 1987, 28, 1945.
- Georg, G. I.; Mashava, P. M.; Guan, Xiangming. Tetrahedron Lett. 1991, 32, 581.
- 8. Manhas, M.S.; Bose, A.K.; Khajavi, M.S. Synthesis, 1981, 209.
- 9. Jang, D. O.; Park, D. J. and Kim, J. Tetrahedron Lett. 1999, 40, 5323.
- Ahluwalia, V. K.; Mallika, N.; Singh, R. and Mehta, V. D. J. Indian Chem. Soc. 1989, 66, 200.
- 11. Sharma, S. D. and Khurana, J. P. S. Indian J. Chem. 1989, 28B, 97.

(Received in the UK 03 November 1999)