Synthesis and functionalization of 4-arylamino-2-pyridone derivatives

N. Z. Tugusheva,^a L. M. Alekseeva,^a A. S. Shashkov,^b V. V. Chernyshev,^c and V. G. Granik^a*

^aState Research Center of Antibiotics, 3a ul. Nagatinskaya, 117105 Moscow, Russian Federation. Fax: +7 (495) 231 4284. E-mail: vggranik@mail.ru
^bN. D. Zelinsky Institute of Organic Chemistry, Russian Academy of Sciences, 47 Leninsky prosp., 119991 Moscow, Russian Federation. Fax: +7 (495) 135 5328
^cDepartment of Chemistry, M. V. Lomonosov Moscow State University, 1 Leninskie Gory, 119992 Moscow, Russian Federation. Fax: +7 (495) 939 3654. E-mail: cher@biocryst.phys.msu.su

New approaches to the synthesis of the previously unknown pyrimidine, 4-amino-2pyridone, and 4-aminopyridine derivatives were developed based on the reactions of enaminoamides with dimethylformamide dimethyl acetal. New structural modifications of 4-amino-2-pyridone derivatives were performed. Numerous compounds of this type, which are of interest for biological studies, were prepared.

Key words: amide acetals, 3-cyano-2-oxo-5-R-4-(phenylamino)-1,2-dihydropyridines, 4-arylamino-3,5-dicyanopyridines, acidic hydrolysis, NMR spectroscopy, X-ray diffraction analysis.

Earlier, a procedure has been developed for the synthesis of various 1-alkyl- and 1-aryl-5-cyano- $(2-N,N-di-methylaminovinyl)-4-oxo-1,4-dihydropyrimidines^{1,2} based on the reactions of amide acetals at the primary amide amino group and at the methyl group of 3-arylamino-2-cyanocrotonamides. These pyrimidinones were found¹⁻⁴ to undergo recyclization to form 4-alkyl(aryl)amino-3-cyanopyridin-2-ones under conditions of both alkaline and acidic hydrolysis.$

The aim of the present study was to synthesize new 1-arylpyrimidin-4-one derivatives containing *ortho* substituents in the phenyl ring and investigate their behavior under conditions of acidic and alkaline hydrolysis, as well as to prepare 4-arylamino-5-formyl-2-pyridones and investigate their properties. It should be emphasized that recyclization of these pyrimidinones would be expected to give substituted 4-aminopyridines, and this class of compounds is of considerable interest in searching for compounds acting on the central nervous system. It is sufficient to note that this class includes the well-known drugs tacrine and amiridine, which are used in the therapy



of the neurodegenerative disease, such as Alzheimer's disease. $^{5-7}$

Scheme 1 presents the synthesis of pyrimidinones **4a,b** by the method developed earlier.^{1,4} In the first step, transamination of enaminoamide 1 with *ortho*-substituted anilines in acetic acid gives, *via* compounds **3a,b**, enaminoamides **2a,b** in high yields, and refluxing of the latter in DMF acetal (or in toluene in the presence of a threefold excess of DMF acetal) affords pyrimidinones **4a,b**.

Alkaline hydrolysis of compounds **4a,b** yields 3-cyano-4-(2-methoxyphenylamino)- and 3-cyano-4-(2,5dimethoxyphenylamino)pyridin-2-one **5a,b**, as observed earlier in analogous reactions with various 1-alkyl- and 1-aryl-substituted pyrimidin-4-ones.¹⁻⁴ This process involves the pyrimidine ring opening accompanied by elimination of HCOOH followed by recyclization through intramolecular transamination to form 3-cyanopyridones **5**.

Acidic hydrolysis of pyrimidin-4-ones gives rise to a mixture of compounds, different products being isolated depending on the reaction conditions and the nature of substituents in the benzene ring. For example, refluxing of pyrimidinones **4a**,**b** in glacial acetic acid affords bicyclic compounds **6a** and **6b** in 41 and 78% yields, respectively, the pyrimidine ring remaining intact. Heating of pyrimidinone **4a** in a stronger acid (1 *M* HCl, 70 °C, 4 h) gives rise to a mixture of two products, *viz.*, *N*-formyl-carbamoylpyridone **7a** and 3-carbamoylpyridone **8a**. We

Published in Russian in Izvestiya Akademii Nauk. Seriya Khimicheskaya, No. 8, pp. 1421-1432, August, 2006.

1066-5285/06/5508-1475 © 2006 Springer Science+Business Media, Inc.



Scheme 1

2: Ar = $2 \cdot \text{MeOC}_6\text{H}_4$ (**a**), 2,5-(MeO)₂C₆H₃ (**b**), 4-ClC₆H₄ (**c**) **3, 9:** R¹ = H, R² = Ph (**a**); R¹ = H, R² = 4-ClC₆H₄ (**b**); R¹ = R² = Me (**c**) **4–8:** R³ = H, R⁴ = OMe (**a**); R³ = R⁴ = OMe (**b**); R³ = R⁴ = H (**c**)

isolated compound **7a** from this mixture in individual form in 21% yield. 3-Carbamoyl-4-(2-methoxyphenylamino)-2-pyridone (**8a**) was also prepared in 82% yield by direct hydrolysis of pyrido[4,3-*d*]pyrimidine-4,5-dione **6a** with 1 *M* KOH. Apparently, heating of pyrimidinone **4a** in 1 *M* HCl first leads to cyclization giving rise to pyrido[4,3-*d*]pyrimidine **6a** with the involvement of the enamine and cyano groups followed by the pyrimidine ring opening to form compound **7a**, which is transformed into carbamoylpyridone **8a**. Such transformations have been observed earlier⁸ for 1-benzyl-substituted pyrimidinone of type **4** under both acidic and alkaline conditions.

Heating (70 °C, 1 h) of pyrimidinone **4b** containing the 2,5-dimethoxyphenyl substituent or prolonged storage of this compound at 20 °C in 1 *M* HCl also affords a mixture of products (bicyclic compound **6b**, *N*-formylcarbamoylpyridone **7b**, and an impurity of **5b** (TLC)). Storage of this mixture at 20 °C in 1 *M* KOH led to the formation of 3-carbamoylpyridone **8b** in 74% yield (see Scheme 1). *N*-Formylcarbamoylpyridone **7b**, which is an intermediate product of recyclization of pyrimidinone **4b** into carbamoylpyridone **8b**, was also isolated in individual form in 13% yield from an analogous mixture, which was prepared by moderate heating (at 70 °C for 3 h or at 40 °C for 6 h) of pyrimidinone **4b** in 99% acetic acid.

In addition, we studied the properties of 5-formylsubstituted pyridin-2-ones. Earlier, it has been found that the reactions of amide acetals with enaminoamides initially afford compounds 3 as condensation products at both the amide and methyl groups followed by cyclization to give pyridinones. It was established that it is these compounds 3 that give 4-arylamino-3-cyano-5-formylpyridin-2-ones in acidic media.⁴ The drawback of this method is that compounds 3 are difficult to prepare in individual form. However, the use of N-(p-chlorophenyl)substituted enaminoamide 2c in the reaction with DMF acetal made it possible to isolate compound $3c^{3}$, whose hydrolysis in 90% acetic acid according to the known method⁴ produced 4-(4-chlorophenylamino)-3-cyano-5formylpyridin-2-one 9b. 5-Formyl-4-phenylamino- and 4-dimethylamino-5-formylpyridones **9a,c⁴** were synthesized analogously (see Scheme 1).





Earlier,⁹ we have synthesized 4-(arylamino)-3-cyanopyridin-2-ones unsubstituted at position 5 of the pyridine ring and studied their biological properties. 5-Formylpyridinones prepared in our study allow one to synthesize compounds containing various substituents at position 5, which are of interest from the point of view of elucidation of structure-activity relationships. The reaction of 5-formylpyridone 9a with amines produced Schiff bases 10a-d (Scheme 2). The ease of the reaction is to a large extent associated with the basicity of the amine used. The reaction with γ -aminobutyric acid proceeds very slowly (product 10b was isolated in low yield), whereas the reaction with benzylamine readily affords target product 10a in high yield. The reaction of 5-formylpyridine 9a with hydrazine hydrate also proceeds readily to give compound 11, whose reactions with various aldehydes produce hydrazones 12a-e in high yields.

The aldehyde group in 5-formylpyridines 9a,b is reduced with NaBH₄ to give alcohols. These reactions afforded 4-phenylamino-substituted 5-hydroxymethylpyridones 13a,b in 70-72% yields (see Scheme 2).

The "benzyl" alcoholic group in pyridone **13a** is rather reactive. The reaction of compound **13a** with urea in AcOH produces 5-ureidomethyl derivative **14** (Scheme 3) in high yield (89%). Various ureido derivatives can exhibit anticonvulsant and antihypoxic activities.^{10,11} The transformation of 5-hydroxymethyl-4-phenylaminopyridone **13a** into the chloro derivative followed by treatment with pyridine gave pyridinium salt **15** (¹H NMR spectroscopic data, see the Experimental section). Such pyridinium salts were demonstrated¹² to form amino derivatives in the reactions with primary and secondary amines. However, in the reaction under consideration, the attack of the nucleophile (piperidine) unexpectedly does not occur at position 2 of the pyridinium fragment of salt **15** but leads to its replacement. As a result, 5-piperidinomethyl-pyridone **16** was synthesized (in 36% yield based on consumed alcohol).



Reagents and conditions. *i*. H₂NCONH₂; *ii*. 1) SOCl₂, 2) pyridine; *iii*. Piperidine.

The reactions of 5-formylpyridones 9a-c with hydroxylamine hydrochloride in pyridine under mild conditions produce oximes 17a-c (Scheme 4).

Some cyanopyridines exhibit considerable antibradykinin, analgetic, and antihypertensive effects.^{1,9} 5-Formylpyridines can be used to synthesize compounds



Scheme 4

9,17: $R^1 = H$, $R^2 = Ph(a)$; $R^1 = H$, $R^2 = 4$ -ClC₆H₄(b); $R^1 = R^2 = Me(c)$ **18–20, 22:** $R^3 = H(a)$, Cl (b) **21:** $R^3 = R^4 = H(a)$; $R^3 = H$, $R^4 = CH_2Ph(b)$; $R^3 = H$, $R^4 = Ph(c)$; $R^3 = H$, $R^4 = 4$ -MeOC₆H₄(d); $R^3 = H$, $R^4 = 4$ -ClC₆H₄(e); $R^3 = H$, $R^4 = 4$ -MeOC₆H₄(d); $R^3 = H$, $R^4 = 4$ -ClC₆H₄(e); $R^3 = H$, $R^4 = 4$ -MeOC₆H₄(d); $R^3 = H$, $R^4 = 4$ -ClC₆H₄(e); $R^3 = H$, $R^4 = 4$ -MeOC₆H₄(d); $R^3 = H$, $R^4 = 4$ -ClC₆H₄(e); $R^3 = H$, $R^4 = 4$ -MeOC₆H₄(d); $R^3 = H$, $R^4 = 4$ -ClC₆H₄(g)

containing electron-releasing substituted amino groups and two electron-withdrawing cyano groups. Alternative methods for the synthesis of such compounds are lacking, and these compounds are virtually inaccessible with the use of conventional methods of pyridine synthesis. We succeeded in preparing a series of such compounds with the use of methods for the transformation of the aldehyde moiety into the nitrile group.¹³ The treatment of 4-phenylamino-substituted 5-formylpyridones with hydroxylamine hydrochloride in pyridine followed by heating of the reaction mixture in acetic anhydride afforded 3,5-dicyanopyridones 19a,b in high yield. Apparently, the starting compounds **9a,b** are initially transformed into oximes 17a,b, which are rapidly transformed into acetyl derivatives 18a,b under the action of acetic anhydride. At higher temperature (110–115 °C), compounds **18a,b** eliminate the acetic acid molecule to give dicyanopyridones **19a,b**. We succeeded in synthesizing acetyl derivatives from both oximes 17a,b and 5-formylpyridones 9a,b. As expected, storage of these compounds in acetic anhydride at 110-115 °C afforded dicyanopyridones 19a,b, whose reactions with phosphorus oxychloride in the presence of triethylamine hydrochloride produced 2-chloro-3,5dicyano-4-(4-R-phenylamino)pyridines 20a,b in high yield (see Scheme 4).

The Cl atom in these compounds is rather labile and can be replaced by the NH_2 or RNH group. This method was used to synthesize diamino derivatives of dicyanopyridones **21a**—g. The replacement of the Cl atom by the amino group was performed by the reaction of chloropyridine **20a** with ethanolic ammonia in an autoclave (see Scheme 4).

It should be noted that heating of acetyl derivatives **18a,b** in dimethylformamide or pyridine leads to their cyclization to give 7-cyano-6-oxo-1-(4-R-phenylamino)-5,6-dihydro-1*H*-pyrazolo[4,3-*c*]pyridines **22a,b**. In the case of 4-anilino-substituted **18a**, cyclization is accompanied by the formation of dicyanopyridone **19a** (which is observed by chromatography), whereas 4-chloro-substituted 5-formylpyridone undergoes only cyclization (see Scheme 4).

This cyclization is very unusual. Hence, we studied the structures of the resulting bicyclic compounds in detail by spectroscopy (the NMR spectra of compounds **22a,b** with the complete assignment of the ¹H and ¹³C signals are given in the Experimental section).

The structure of compound **22b** was established by the HMBC (heteronuclear multiple-bond connectivity) technique. The correlations peaks at δ 8.41/112.1 (H(3)/C(3a)), 8.41/146.8 (H(3)/C(7a)), 8.73/112.1



Fig. 1. Intermolecular hydrogen bonds in the crystal structure of 22b.

(H(4)/C(3a)), 8.73/139.8 (H(4)/C(3)), 8.73/146.8 (H(4)/C(7a)), and 8.73/161.4 (H(4)/C(6)) are consistent with the proposed structure.

The NMR spectroscopic data were supported by X-ray diffraction data for bicyclic compound **22b** (see the Experimental section).

The crystal packing of compound **22b** is stabilized by intermolecular N-H...O and C-H...N hydrogen bonds (Fig. 1). The geometric characteristics of these bonds are given in Table 1.

As mentioned above, pyrimidinones **4** are transformed into 3-cyanopyridones **5** under the action of alkali. It is known¹⁴ that aromatic nitriles can be transformed into aldehydes in 75% HCOOH in the presence of a large excess of a Ni—Al₂ alloy as the catalyst. The reaction with 3-cyanopyridine was documented. The yield achieved in this reaction is substantially lower than that in the reactions with aromatic nitriles (37% and 70—97%, respectively). It was of interest to study the behavior of cyano derivatives of 4-aminophenylpyridines under these conditions. Refluxing of 3-cyanopyridine **5c**, which was prepared according to a procedure described earlier,³ in 50% HCOOH in the presence of a large excess (gram per gram of nitrile) of a Ni—Al₂ alloy as the catalyst afforded only pure formyl derivative **23** in 53% yield. The reaction

 Table 1. Geometric characteristics of hydrogen bonds in compound 22b

D-HA	Bon	id length/Å	D-HA angle		
	D—H	HA DA	/deg		
$N(7) - H(7) O(10)^{i}$	0.86	1.96 2.617(8) 133		
O(18)-H(18)O(12) ⁱⁱ	0.93	2.46 3.210(8) 138		

Note. The symmetry codes: (i) -x, 0.5 + y, 0.5 - z; (ii) 1 + x, 1 + y, z. of 3-cyanopyridine **5c** in 85% HCOOH in the presence of this catalyst produced not only 3-formylpyridone **23** but also tricyclic product **24** (Scheme 5).

Scheme 5



26: R = OH(a), $NHC_6H_4NO_2(b)$

Reagents and conditions. *i*. 63% H_2SO_4 ; *ii*. Raney Ni, HCOOH; *iii*. PrⁱOH + piperidine; *iv*. HONH₂HCl or $NH_2NHC_6H_4NO_2$.

3-Formylpyridone **23** readily undergoes cyclization to form tricyclic compound **24** upon refluxing in isopropyl alcohol in the presence of piperidine.

Refluxing of compound **5c** (40 min) in 63% H₂SO₄ leads to hydrolysis of the nitrile group to form the carboxy group followed by decarboxylation with the simultaneous sulfonation at the *para* position of the phenyl ring to give finally compound **25** in 60% yield (see Scheme 5).

3-Formylpyridone 23 is readily involved in condensation. Refluxing in ethanol with hydroxylamine hydrochloride or *p*-nitrophenylhydrazone affords oxime 26a and hydrazone 26b in 63 and 68% yields, respectively.

To summarize, we developed procedures for the synthesis of various 4-aminopyridine derivatives, performed modifications of these structures, and prepared numerous compounds, which are of interest for biological studies.

Experimental

The IR spectra were recorded on a FSM-1201 instrument in Nujol mulls. The ¹H NMR spectra were measured on Bruker AC-300 and Bruker DRX-500 spectrometers in DMSO- d_6

and DMSO- d_6 --CCl₄. The HMBC spectra were recorded on a Bruker DRX-500 spectrometer in DMSO- d_6 . The electron impact (EI) mass spectra (70 eV) were obtained on a Finnigan SSQ-710 mass spectrometer using a direct inlet system. The electrospray-ionization (ESI) mass spectra were recorded on a Waters ZQ-2000 mass spectrometer using a direct inlet system without a chromatographic column. The course of the reactions was monitored and the purity of the compounds was checked by TLC on Merck 60 F_{254} plates (ethyl acetate—ethanol, 10 : 1, as the eluent). The melting points were determined on an Electrotermal 9100 instrument (UK). The melting points and the yields of the reaction products are given in Table 2.

Table 2. Melting points and elemental analysis data for compounds 2a,b, 4a,b, 5a,b, 6a, 7a,b, 8b, 9b, 10a-c, 11, 12b,c,e, 13a,b, 14, 17a-c, 18a, 19a,b, 20a,b, 21a-g, 22b, 23, 24, and 26a,b

Com- pound	M.p./°C (solvent)	Found (%) Calculated			Molecular formula	Com- pounc	- M.p./°C d (solvent)	Found (%) Calculated			Molecular formula
		С	Н	N				С	Н	N	
2a	186—187	<u>62.65</u>	<u>5.76</u>	<u>18.05</u>	C ₁₂ H ₁₃ N ₃ O ₂	17a ^a	261-263	<u>59.73</u>	<u>4.20</u>	<u>21.66</u>	C ₁₃ H ₁₀ N ₄ O ₂ ·
	(DMF)	62.33	5.67	18.17			(MeOH)	60.35	4.06	21.24	•0.25H ₂ O
2b	177-178	<u>59.73</u>	<u>5.87</u>	<u>16.06</u>	$C_{13}H_{15}N_3O_3$	17b	293-294	<u>54.50</u>	<u>2.91</u>	<u>19.57</u>	C ₁₃ H ₈ N ₄ O ₂ Cl
	(Pr ⁱ OH)	59.76	5.79	16.08			(Pr ⁱ OH)	54.28	2.80	19.48	
4 a	259-262	<u>65.05</u>	<u>5.59</u>	<u>18.72</u>	$C_{16}H_{16}N_4O_2$	17c	203-204	<u>52.54</u>	<u>4.69</u>	<u>27.16</u>	$C_9H_{10}N_4O_2$
	(Pr ⁱ OH–DMF)	64.85	5.44	18.91			(Pr ⁱ OH)	52.42	4.89	27.17	
4b	205 - 206	<u>62.51</u>	<u>5.41</u>	<u>16.90</u>	$C_{17}H_{18}N_4O_3$	18a	286-288	<u>60.76</u>	<u>4.01</u>	<u>18.52</u>	$C_{15}H_{12}N_4O_3$
	(Pr ⁱ OH–DMF)	62.57	5.56	17.17			(DMF)	60.81	4.05	18.92	
5a	241-242	<u>64.76</u>	<u>4.64</u>	<u>17.36</u>	$C_{13}H_{11}N_3O_2$	19a	323-324	<u>65.74</u>	<u>3.47</u>	<u>23.41</u>	$C_{13}H_8N_4O$
	(Pr ⁱ OH–DMF)	64.72	4.60	17.42			(MeOH)	66.10	3.39	23.73	
5b	275-277	<u>62.06</u>	<u>4.98</u>	<u>15.15</u>	$C_{14}H_{13}N_3O_3$	19b	335	<u>57.43</u>	<u>2.75</u>	<u>20.67</u>	$C_{13}H_7N_4OCl$
	(Pr ⁱ OH–DMF)	61.99	4.83	15.49			(MeOH)	57.69	2.61	20.70	
6a	280-284	<u>61.95</u>	<u>4.07</u>	<u>14.89</u>	$C_{14}H_{11}N_3O_3$	20a ^b	192—194	<u>61.67</u>	<u>2.66</u>	<u>22.16</u>	$C_{13}H_7N_4Cl$
	(Pr ¹ OH–DMF)	62.45	4.12	15.61			(EtOH)	61.30	2.75	22.00	
7a	208-211	<u>58.72</u>	<u>4.56</u>	<u>14.44</u>	$C_{14}H_{13}N_3O_4$	20b	240-241	<u>53.89</u>	<u>2.20</u>	<u>19.31</u>	$C_{13}H_6N_4Cl_2$
	(Pr ¹ OH–DMF)	58.53	4.56	14.63			(MeOH)	54.01	2.09	19.38	
7b	208-211	<u>57.10</u>	<u>4.69</u>	<u>13.02</u>	$C_{15}H_{15}N_{3}O_{5}$	21a	225-226	<u>66.35</u>	<u>3.72</u>	<u>30.06</u>	$C_{13}H_9N_5$
	(Pr ¹ OH–DMF)	56.78	4.76	13.24			(EtOH)	66.38	3.83	29.79	
8b	280	<u>58.31</u>	<u>5.44</u>	<u>14.26</u>	$C_{14}H_{15}N_{3}O_{4}$	21b	219-220	<u>74.24</u>	<u>4.70</u>	<u>21.87</u>	$C_{20}H_{15}N_5$
	(Pr ⁱ OH–DMF)	58.13	5.23	14.53			(EtOH)	73.85	4.62	21.54	
9b	290	<u>57.02</u>	<u>2.65</u>	<u>15.39</u>	$C_{13}H_8N_3O_2Cl$	21c	234-236	<u>73.37</u>	<u>4.20</u>	<u>22.44</u>	$C_{19}H_{13}N_5$
	(Pr ⁱ OH–DMF)	57.05	2.95	15.35			(MeOH)	73.31	4.18	22.51	
10a	229-230	<u>73.27</u>	<u>4.84</u>	<u>17.21</u>	$C_{20}H_{16}N_4O$	21d	203-204	<u>70.36</u>	<u>4.40</u>	<u>20.49</u>	$C_{20}H_{15}N_5O$
	(MeOH)	73.17	4.88	17.07	a	• • •	(MeOH)	70.38	4.40	20.53	a
10b	207-209	<u>62.60</u>	<u>4.83</u>	<u>16.95</u>	$C_{17}H_{16}N_4O_3$	$21e^{c}$	218-220	<u>65.99</u>	<u>3.40</u>	<u>20.37</u>	$C_{19}H_{12}N_5CI$
10	(MeOH)	62.96	4.94	17.28	C H N C	016	(EtOH)	65.99	3.47	20.26	
luc	202-204	<u>68.51</u>	<u> 3.38</u>	13.98	$C_{23}H_{22}N_4O_3$	211	224-226	<u>/3./0</u>	4.51	21.38	$C_{20}H_{15}N_5$
11	(EtOH)	68.66	5.4/	13.93		21.	(EtOH)	/3.85	4.62	21.54	
11	>320	$\frac{61.78}{1.00}$	<u>4.54</u>	27.50	$C_{13}H_{11}N_5O$	21g	222—226 (1) E(OII)	<u>59.81</u>	$\frac{2.77}{2.02}$	18.47	$C_{19}H_{11}N_5CI_2$
121	(DMF)	61.00	4.35	2/.0/	CUNO	22L	(decomp., EtOH)	60.02	2.92	18.42	
120	290—292 (DME)	<u>00.93</u>	<u>4.49</u>	<u>19.51</u>	$C_{20}H_{15}N_5O_2$	220	>330 (DME)	57.75	$\frac{2.77}{2.01}$	$\frac{20.81}{20.70}$	$C_{13}H_7N_4OCI$
12.	(DMF)	67.23	4.20	19.01	CUNO	22	(DMF)	57.69	2.61	20.70	CUNO
120	292—293 (DME)	$\frac{04.98}{(5.12)}$	<u>4.07</u>	18.20	$C_{21}H_{17}N_5O_3$	23	241 - 242	$\frac{07.33}{(7.29)}$	<u>4.91</u> 4.70	12.00	$C_{12}H_{10}N_2O_2$
12.	(DMF)	69.12	4.39	18.09	CUNO	24	$(PnCH_3)$	07.28	4.70	13.08	
12e	290—291 (EtOU)	$\frac{08.37}{0.47}$	<u>5.72</u> 5.71	$\frac{21.10}{21.02}$	$C_{19}H_{19}N_5O$	24	330-331	$\frac{73.31}{72.40}$	4.09	14.12	$C_{12}H_8N_2O$
12-	(EIOH)	08.4/	5./1 4.70	21.02	CUNO	26-	(MeOH)	/3.40	4.11	14.28	CUNO
15a	222-223	<u>04.70</u> 64.70	<u>4.79</u>	$\frac{17.48}{17.42}$	$C_{13}H_{11}N_3O_2$	20a	234 - 230	62.80	4.85	$\frac{18.22}{18.22}$	$C_{12}H_{11}N_3O_2$
124	$(\Gamma I \cup \Pi - \bigcup M \Gamma)$	04.72 57.07	4.00	17.42	CHNOC	7 64	(MCOR)	61 61	4.04	10.00	СЦМО
130	201-204	<u>56.64</u>	<u>3.98</u> 2.66	15.38	$C_{13}\Pi_{10}N_{3}O_{2}CI$	200	301 - 302	<u>01.01</u> 61.90	4.39	<u>19.93</u> 20.05	$C_{18}\pi_{15}N_5O_3$
14	(110H-DMF)	50.04	3.00 4.00	24 24	СНМО			01.09	4.33	20.03	
14	247 - 230	<u>59.59</u>	<u>4.99</u> 1.62	$\frac{24.01}{24.72}$	$C_{14} \Pi_{13} \Pi_5 O_2$						
	$(\Gamma \Gamma \cup \Pi - D M \Gamma)$	39.30	4.03	24./2							

^a Found (%): H₂O, 1.53. Calculated (%): H₂O, 1.74.

^b Found (%): Cl, 14.08. Calculated (%): Cl, 13.95.

^c Found (%): Cl, 10.42. Calculated (%): Cl, 10.27.

1-Cyano-2-(2-methoxyphenylamino)crotonamide (2a). A mixture of enaminoamide **1** (0.3 g, 0.002 mol) and *ortho*-anisidine (0.34 g, 0.028 mol) in AcOH (5 mL) was refluxed for 6 h. The reaction mixture was concentrated *in vacuo*, the residue was triturated in water, and the precipitate that formed was filtered off and washed with water. Compound **2a** was obtained in a yield of 0.3 g (67%), m.p. 186–187 °C (PrⁱOH). IR, v/cm⁻¹: 3357, 3184 (NH); 2194 (CN); 1656 (CO); 1606 (C=C). EI MS, m/z (I_{rel} (%)): 231 [M]⁺⁺ (91), 214 [M – NH₃]⁺⁺ (72), 199 [M – MeOH]⁺⁺ (48), 185 [M – CONH₂ – 2 H]⁺⁺ (39), 171 [M – CONH₂ – CH₄]⁺⁺ (54), 148 [M – NCCH=CONH₂]⁺⁺ (100), 123 [M – C₆H₄OMe]⁺⁺ (87), 108 [Ph(OMe)]⁺⁺ (71).

1-Cyano-2-(2,5-dimethoxyphenylamino)crotonamide (2b) was prepared analogously to compound **2a** from enaminoamide **1** and 2,5-dimethoxyaniline. The yield was 55%, m.p. 177–178 °C (PrⁱOH). ¹H NMR (DMSO-d₆), δ : 2.17 (s, 3 H, Me); 3.73 and 3.77 (both s, 3 H each, 2 OMe); 6.80 (br.s, 2 H, NH₂); 6.82 (dd, 1 H, H(4), $J_o = 8.4$ Hz, $J_m = 3.0$ Hz); 6.88 (br.s, 1 H, H(6)); 7.05 (d, 1 H, H(3), $J_o = 8.4$ Hz); 12.37 (br.s, 1 H, C(1)NH). EI MS, m/z (I_{rel} (%)): 261 [M]⁺⁺ (100), 229 [M – MeOH]⁺⁺ (87), 201 [M – 2 MeO + 2 H]⁺⁺ (45), 178 [M – NCCH=CONH₂]⁺⁺ (85), 153 [H₂NC₆H₃(OMe)₂]⁺⁺ (37), 138 [C₆H₃(OMe)]⁺⁺ (84).

5-Cyano-6-(2-dimethylaminovinyl)-1-(2-methoxyphenyl)-4-oxo-1,4-dihydropyrimidine (4a). A mixture of enaminoamide **2a** (0.2 g, 0.87 mol) and DMF acetal (4 mL) was refluxed for 5 h. The reaction mixture was cooled to 5 °C, and the precipitate that formed was filtered off and washed with ethanol. Compound **4a** was obtained in a yield of 0.2 g (77%), m.p. 259–262 °C (PrⁱOH–DMF, 1 : 1). ¹H NMR (DMSO-d₆), δ: 2.78 (br.s, 6 H, NMe₂); 3.82 (s, 3 H, OMe); 4.09 (d, 1 H, H_α, J_o = 12.9 Hz); 7.15, 7.30, 7.46, and 7.59 (all m, 1 H each, H(3')–H(6')); 7.87 (d, 1 H, H_β, J_o = 12.9 Hz); 8.07 (s, 1 H, H(2)). IR, v/cm⁻¹: 3074 (NH); 2200 (CN); 1629 (CO); 1602 (C=C). EI MS, m/z (I_{rel} (%)): 296 [M]⁺⁺ (63), 265 [M – MeO]⁺⁺ (100), 238 [M – MeO – CN]⁺⁺ (81), 223 [M – MeO – CN – Me]⁺⁺ (52).

5-Cyano-1-(2,5-dimethoxyphenyl)-6-(2-dimethylaminovinyl)-4-oxo-1,4-dihydropyrimidine (4b) was prepared analogously to compound **4a** from enaminoamide **2b** and DMF acetal. The yield was 80%, m.p. 194–196 °C (PrⁱOH–DMF, 7 : 3). ¹H NMR (DMSO-d₆), & 2.81 (br.s, 6 H, NMe₂); 3.76 (s, 6 H, 2 OMe); 4.14 (d, 1 H, H_a, J_o = 12.9 Hz); 7.20–7.60 (m, 3 H, H(3'), H(4'), H(6')); 7.90 (d, 1 H, H_β, J_o = 12.9 Hz); 8.08 (s, 1 H, H(2)). IR, v/cm⁻¹: 3512, 3403, 3062 (NH); 2190 (CN); 1638 (CO). EI MS, m/z (I_{rel} (%)): 326 [M]⁺⁺ (45), 295 [M – MeO]⁺⁺ (100), 268 [M – MeO – CN]⁺⁺ (91), 253 [M – MeO – CN – Me]⁺⁺ (80).

3-Cyano-4-(2-methoxyphenyl)amino-2-oxo-1,2-dihydropyridin-2-one (5a). A mixture of pyrimidinone **4a** (0.3 g, 0.87 mol) and 1 *M* NaOH (7 mL) was refluxed for 1 h. The solution was acidified with concentrated HCl to pH 3. The precipitate that formed was filtered off and washed with water. Compound **5a** was obtained in a yield of 0.24 g (98%), m.p. 241–242 °C (PrⁱOH–DMF, 30 : 5). ¹H NMR (DMSO-d₆), 8: 3.79 (s, 3 H, OMe); 5.49 (d, 1 H, H(5), $J_o = 7.4$ Hz); 6.98, 7.11, 7.20, and 7.31 (all m, 1 H each, H(3')–H(6')); 7.24 (d, 1 H, H(6), $J_o = 7.4$ Hz); 8.60 (br.s, 1 H, C(4)NH); 11.04 (br.s, 1 H, N(1)H). EI MS, m/z (I_{rel} (%)): 241 [M]^{+*} (100), 226 [M – Me]^{+*} (47), 198 [M – Me – CO]^{+*} (45), 183 [M – MeOH – CN]^{+*} (53), 108 [Ph(OMe)]^{+*} (30). **3-Cyano-4-(2,5-dimethoxyphenyl)amino-2-oxo-1,2-dihydropyridin-2-one (5b)** was prepared analogously to compound **5a** from pyrimidinone **4b**. The yield was 76%, m.p. 275–277 °C (PrⁱOH–DMF, 2 : 1). ¹H NMR (DMSO-d₆), & 3.72 (s, 6 H, 2 OMe); 5.53 (d, 1 H, H(5), $J_o = 7.4$ Hz); 6.85 (m, 2 H, H(4'), H(6')); 7.05 (d, 1 H, H(3'), $J_o = 8.4$ Hz); 7.25 (d, 1 H, H(6), $J_o = 7.4$ Hz); 8.66 (br.s, 1 H, C(4)NH); 11.09 (br.s, 1 H, N(1)H). EI MS, m/z (I_{rel} (%)): 271 [M]^{+•} (100), 256 [M – Me]^{+•} (47), 228 [M – Me – CO]^{+•} (45).

1-(2-Methoxyphenyl)-4,5-dioxo-1,4,5,6-tetrahydropyrido[4,3-*d***]pyrimidine (6a).** A solution of compound **4a** (0.3 g, 1 mmol) in glacial AcOH (4 mL) was refluxed for 7 h. The reaction mixture was cooled to 5 °C, and the precipitate that formed was filtered off and washed with ethyl acetate. Compound **6a** was obtained in a yield of 0.11 g (41%), m.p. 280–282 °C (PrⁱOH–DMF, 1 : 1). ¹H NMR (DMSO-d₆), 8: 3.81 (s, 1 H, OMe); 5.43 (s, 1 H, H(8), $J_o = 7.2$ Hz); 7.18, 7.31, and 7.51–7.69 (all m, 1 H each, 1 H, 3 H, H(7), H(3')–H(6')); 8.22 (s, 1 H, H(2)); 11.69 (br.s, 1 H, N(6)H). IR, v/cm⁻¹: 3167 (NH); 1675, 1644 (CO). EI MS, m/z (I_{rel} (%)): 269 [M]⁺⁺ (100), 241 [M – CO]⁺⁺ (80), 213 [M – 2 CO]⁺⁺ (78).

1-(2,5-Dimethoxyphenyl)-4,5-dioxo-1,4,5,6-tetrahydropyrido[4,3-*d*]pyrimidine (6b) was prepared analogously to compound 6a from pyrimidinone 4b. Acetic acid was evaporated, the residue was triturated in water, and the crystalline precipitate that formed was filtered off and washed with water and ethyl acetate. The yield was 78%, m.p. 162–165 °C (PrⁱOH–DMF, 7:2). Found (%): C, 56.91; H, 4.59; N, 13.11. C₁₆H₉N₅O·H₂O. Calculated (%): C, 56.78; H, 4.76; N, 13.24. ¹H NMR (DMSO-d₆), &: 3.88 and 3.90 (both s, 3 H each, 2 OMe); 5.55 (br.s, 1 H, H(8)); 7.32 (dd, 1 H, H(4'), $J_o = 9.2$ Hz, $J_m =$ 2.8 Hz); 7.41 (d, 1 H, H(3'), $J_o = 9.2$ Hz); 7.43 (d, 1 H, H(6'), $J_m = 2.8$ Hz); 7.65 (br.s, 1 H, H(7)); 8.40 (s, 1 H, H(2)); 11.85 (br.s, 1 H, N(6)H). IR, v/cm⁻¹: 3472 (OH); 3380 (NH); 1685, 1654 (CO). EI MS, m/z (I_{rel} (%)): 299 [M]⁺⁺ (100), 271 [M – CO]⁺⁺ (28).

N-Formyl-4-(2-methoxyphenyl)amino-2-oxo-1,2-dihydropyridine-3-carboxamide (7a). A suspension of pyrimidinone 4a (0.3 g, 1 mmol) in 1 M HCl (3 mL) was kept at 70 °C for 4 h. The precipitate that formed was filtered off and washed with water. Compound 8a was obtained in a yield of 0.06 g (80%, ¹H NMR spectroscopic data). The aqueous mother liquor was alkalized with NaOH (sol.) to pH 8–9, and the precipitate that formed was filtered off and washed with water. A mixture of compounds 7a and 8a was obtained in a vield of 0.15 g (65 : 35. ¹H NMR spectroscopic data). The precipitate that formed was recrystallized from a 2 : 1 PrⁱOH–DMF mixture. Compound 7a was obtained in a yield 0.06 g (21%), m.p. 208-211 °C (PrⁱOH–DMF, 2 : 1). ¹H NMR (DMSO-d₆), δ: 3.82 (s, 3 H, OMe); 5.93 (d, 1 H, H(5), $J_o = 7.4$ Hz); 7.02, 7.17, and 7.31 (all m, 1 H each, 1 H, 2 H, H(3')-H(6')); 7.37 (d, 1 H, H(6), $J_o = 7.4$ Hz); 9.25 (d, 1 H, CONHC<u>H</u>O, $J_o = 9.8$ Hz); 11.44 (br.s, 1 H, N(1)H); 11.57 (br.s, 1 H, C(4)NH); 13.09 (br.d, 1 H, CON<u>H</u>CHO, $J_o = 9.8$ Hz). IR, v/cm⁻¹: 1700 (CONH₂); 1650 (CONH); 1626 (C=C, C=N). EI MS, $m/z (I_{rel} (\%))$: 287 [M]⁺ (96), 269 $[M - H_2O]^+$ (96), 259 $[M - CO]^+$ (95), 242 $[M - CO - OH]^{+}$ (97), 227 $[M - CO - MeO]^{+}$ (62), 213 $[M - 2 CO - H_2O]^+$ (100).

N-Formyl-4-(2,5-dimethoxyphenyl)amino-2-oxo-1,2-dihydropyridine-3-carboxamide (7b). A solution of pyrimidinone 4b (0.25 g, 0.77 mmol) in glacial AcOH (2 mL) was kept at 70 °C for 3 h and then at 40 °C for 6 h. The reaction mixture was concentrated in vacuo, the residue was triturated in ethyl acetate, and the precipitate that formed was filtered off and washed with ethyl acetate. A mixture of compounds 6b, 7b, and 8b (TLC) was obtained in a yield of 0.12 g. The precipitate was recrystallized from a 6 : 5 PrⁱOH–DMF mixture. Compound 7b was obtained in a yield of 0.03 g (13%), m.p. 208-211 °C (PrⁱOH–DMF, 6 : 5). ¹H NMR (DMSO-d₆), δ: 3.60 and 3.64 (both s, 3 H each, 2 OMe); 5.86 (d, 1 H, H(5), $J_o = 7.4$ Hz); 6.76 (m, 2 H, H(4'), H(6')); 6.97 (d, 1 H, H(3'), $J_o = 8.4$ Hz); 7.26 (d, 1 H, H(6), $J_o = 7.4$ Hz); 9.11 (d, 1 H, CONHC<u>H</u>O, $J_o =$ 9.8 Hz); 11.39 (br.s, 1 H, N(1)H); 11.43 (br.s, 1 H, C(4)NH); 12.98 (br.d, 1 H, CON<u>H</u>CHO, $J_0 = 9.8$ Hz). EI MS, m/z ($I_{\rm rel}$ (%)): 317 [M]^{+•} (48), 299 [M - H₂O]^{+•} (91), 289 $[M - CO]^{+}$ (85), 271 $[M - CO - H_2O]^{+}$ (90), 257 $[M - CO]^{+}$ CO - MeOH]⁺ (100).

4-(2-Methoxyphenyl)amino-2-oxo-1,2-dihydropyridine-3carboxamide (8a). A solution of bicyclic compound **6a** (0.1 g, 0.37 mmol) in 1 *M* KOH (1 mL) was kept at 20 °C for 3 h. The precipitate that formed was filtered off and washed with water. Compound **8a** was obtained in a yield of 0.09 g (82%), m.p. 285–288 °C (PrⁱOH–DMF, 2 : 1). Found (%): N, 15.86. C₁₃H₁₃N₃O₃. Calculated (%): N, 16.21. ¹H NMR (DMSO-d₆), δ : 3.80 (s, 3 H, OMe); 5.84 (d, 1 H, H(5), J_o = 7.4 Hz); 6.95–7.25 (m, 6 H, H(3')–H(6'), H(6), C(3)CONH); 9.84 (br.s, 1 H, C(3)CONH); 10.97 (br.s, 1 H, N(1)H); 12.31 (br.s, 1 H, C(4)NH). EI MS, m/z (I_{rel} (%)): 259 [M]⁺⁺ (100), 242 [M – HO]⁺⁺ (66), 213 [M – CO – H₂O]⁺⁺ (82).

4-(2,5-Dimethoxyphenyl)amino-2-oxo-1,2-dihydropyridine-3-carboxamide (8b). A suspension of pyrimidinone 4b (0.3 g, 0.9 mmol) in 1 M HCl (3 mL) was kept at 20 °C for 6 days or at 70 °C for 1 h. Then 1 M KOH (5 mL) was added to the reaction mixture, and the mixture was kept at 20 °C for 1.5 h. The precipitate that formed was filtered off and washed with water. Compound 8b containing a small impurity of compound 7b (formylcarboxamide) was obtained in a yield of 0.2 g. The precipitate was recrystallized from a 3 : 2 PrⁱOH-DMF mixture. Compound 8b was obtained in a yield of 0.07 g (27%), m.p. 208–210 °C (PrⁱOH–DMF, 7 : 5). ¹H NMR (DMSO-d₆), δ: 3.71 and 3.75 (both s, 3 H each, 2 OMe); 6.08 (d, 1 H, H(5), $J_o = 7.4 \text{ Hz}$; 7.00 (m, 2 H, H(4′), H(6′)); 7.12 (d, 1 H, H(3′), $J_o = 8.4 \text{ Hz}$; 7.22 (br.d, 1 H, C(3)CONH, $J_{gem} = 5.0 \text{ Hz}$); 7.31 (d, 1 H, H(6), $J_o = 7.4$ Hz); 10.00 (br.d, 1 H, C(3)CONH, $J_{gem} = 5.0$ Hz); 11.00 (br.s, 1 H, N(1)H); 12.49 (br.s, 1 H, $\ddot{C}(4)$ NH). EI MS, m/z (I_{rel} (%)): 289 [M]^{+•} (100), 271 $[M - H_2O]^+$ (86), 257 $[M - MeOH]^+$ (97).

4-(4-Chlorophenyl)amino-3-cyano-5-formyl-2-oxo-1,2-dihydropyridine (9b). A solution of compound **3b** (0.623 g, 1.8 mmol) in 90% AcOH (6 mL) was kept at 20 °C for 3 days. The precipitate that formed was filtered off and washed with ethyl acetate. Formylpyridone **9b** was obtained in a yield of 0.437 g (89%), m.p. 290 °C (decomp., $Pr^{i}OH-DMF$, 3 : 2). ¹H NMR (DMSO-d₆), δ : 7.41 (m, 4 H, C₆H₄Cl); 8.42 (s, 1 H, H(6)); 9.59 (s, 1 H, CHO); 10.52 (br.s, 1 H, N(1)H); 12.40 (br.s, 1 H, C(4)NH). IR, v/cm⁻¹: 3198, 3130 (NH); 2219 (CN); 1691 (CO); 1646, 1586 (C=C, C=N). EI MS, *m/z* (I_{rel} (%)): 273 [M]⁺⁺ (100), 244 [M – CHO]⁺⁺ (50).

5-(Benzylimino)methyl-3-cyano-2-oxo-4-phenylamino-1,2dihydropyridine (10a). A mixture of formylpyridone 9a (2 g, 8 mmol), benzylamine (0.9 g, 8 mol), and a catalytic amount of p-TsOH in anhydrous EtOH (40 mL) was refluxed for 1.5 h. The precipitate that formed was filtered off and washed with EtOH. Compound **10a** was obtained in a yield of 2 g (82%), m.p. 229–230 °C (MeOH). IR, v/cm⁻¹: 3315 (NH); 2208 (CN); 1675 (CO); 1644, 1635 (C=C, C=N). EI MS, m/z (I_{rel} (%)): 328 [M]^{+•} (46), 222 [M – PhCH₂NH]^{+•} (79), 106 [PhCH₂=NH]^{+•} (35), 106 [PhCH₂=NH]^{+•} (35), 91 [PhCH₂]^{+•} (100).

4-[(3-Cyano-2-oxo-4-phenylamino-1,2-dihydropyridin-5yl)methylenamino]butyric acid (10b). γ-Aminobutyric acid (1.29 g, 12.5 mmol) was added to a solution of NaOH (0.51 g) in EtOH (25 mL), and the reaction mixture was stirred at 60 °C until complete dissolution was achieved. Formylpyridone 9a (3 g, 12.6 mmol) was added to the resulting solution of the sodium salt of γ -aminobutyric acid in EtOH. The reaction mixture was refluxed for 5 h and cooled to 0 °C. The precipitate was filtered off and washed with EtOH. The sodium salt of compound 10b was obtained in a yield of 1.74 g. The salt was dissolved in warm water (20 mL) and the solution was filtered from the insoluble impurity. The filtrate was acidified with concentrated HCl to pH 2–3, and the precipitate that formed was filtered off and washed with water. Compound 10b was obtained in a yield of 0.5 g (12%), m.p. 207–208 °C (MeOH). IR, v/cm⁻¹: 3180–3230 (NH associated); 2210 (CN); 1730, 1680 (CO); 1640, 1620, 1590 (C=C, C=N). EI MS, m/z (I_{rel} (%)): 324 [M]^{+•} (65), 222 $[M - HOOC(CH_2)_3NH]^{+}$ (100).

3-Cyano-5-[2-(3,4-dimethoxyphenyl)ethyliminomethyl]-2oxo-4-phenylamino-1,2-dihydropyridine (10c) was prepared analogously to 10a from formylpyridone 9a and homoveratrylamine. The reaction time was 5.5 h. The yield was 50%, m.p. $202-204 \,^{\circ}C$ (EtOH). IR, v/cm⁻¹: 3050 (NH); 2200 (CN); 1670 (CO); 1630, 1590 (C=C, C=N). EI MS, *m/z* (*I*_{rel} (%)): 402 [M]⁺⁺ (47), 222 [M - (MeO)₂C₆H₄(CH₂)₂NH]⁺⁺ (50), 193 [(MeO)₂C₆H₃(CH₂)₂N=CH₂]⁺⁺ (43).

3-Cyano-2-oxo-4-phenylamino-5-phenyliminomethyl-1,2dihydropyridine (10d) was prepared analogously to **10a** from formylpyridone **9a** and aniline in ethylene glycol at 135 °C. The reaction time was 2 h. The yield was 38%, m.p. 300 °C (MeOH–DMF, 5 : 1). Found (%): N, 17.48. $C_{19}H_{14}N_4O$. Calculated (%): N, 17.83. ¹H NMR (DMSO-d₆), &: 7.24–7.43 (m, 10 H, 2 Ph); 8.14 (s, 1 H, H(1')); 8.67 (s, 1 H, H(6)); 12.11 (br.s, 1 H, N(1)H); 12.35 (br.s, 1 H, C(4)NH). IR, v/cm⁻¹: 3080, 3020 (NH); 2200 (CN); 1670 (CO); 1630, 1590 (C=C, C=N). EI MS, *m/z* (I_{rel} (%)): 314 [M]^{+•} (100), 222 [M – PhNH]^{+•} (91).

3-Cyano-5-hydrazinomethyl-2-oxo-4-phenylamino-1,2-dihydropyridine (11). A mixture of formylpyridone **9a** (1.2 g, 5 mmol) and hydrazine hydrate (2.5 mL) in anhydrous EtOH (5 mL) was kept at 20 °C for 1 h. The precipitate that formed was filtered off and washed with EtOH. Hydrazone **11** was obtained in a yield of 1.2 g (53%), m.p. 293–294 °C (DMF). ¹H NMR (DMSO-d₆), δ : 6.66 (br.s, 2 H, NH₂); 7.25 and 7.39 (both m, 3 H and 2 H, Ph); 7.62 (s, 1 H, H_a); 7.75 (s, 1 H, H(6)); 11.39 (br.s, 1 H, C(4)NH); 11.63 (br.s, 1 H, N(1)H). IR, v/cm⁻¹: 3400, 3200 (NH); 2200 (CN); 1690 (CO); 1620, 1570 (C=C, C=N). ES MS, *m/z*: 254 [M + H]⁺⁺, 276 [M + Na]⁺⁺, 529 [2 M + Na]⁺⁺. EI MS, *m/z* (I_{rel} (%)): 253 [M]⁺⁺ (50), 222 [M - NH₂NH]⁺⁺ (100).

5-Benzylidenehydrazonomethyl-2-oxo-4-phenylamino-1,2dihydropyridine-3-carbonitrile (12a). A mixture of hydrazone 11 (0.5 g, 1.5 mmol) and benzaldehyde (3.68 g, 0.035 mol) in anhydrous EtOH (20 mL) was kept at 20 °C for 2.5 h. The precipitate that formed was filtered off and washed with EtOH. Compound **12a** was obtained in a yield of 0.5 g (30%), m.p. 278–279 °C (EtOH). Found (%): N, 20.20. $C_{20}H_{15}N_5O$. Calculated (%): N, 20.53. ¹H NMR (DMSO-d₆), &: 7.32–7.50 and 7.83 (both m, 8 H and 2 H, 2 Ph); 8.15 (s, 1 H, H(6)); 8.72 and 8.80 (both s, 1 H each, H(1'), H(4')); 11.38 (br.s, 1 H, C(4)NH); 12.15 (br.s, 1 H, N(1)H). IR, v/cm⁻¹: 3525, 3500, 3180 (NH); 2200 (CN); 1620 (CO); 1575 (C=C, C=N). EI MS, $m/z (I_{rel} (\%))$: 341 [M]⁺ (20), 222 [M – PhCHN=NH]^{+*} (63), 120 [PhCH=NNH₂]^{+*} (100).

5-[(4-Hydroxybenzylidene)hydrazonomethyl]-2-oxo-4-phenylamino-1,2-dihydropyridine-3-carbonitrile (12b) was prepared analogously to **12a**. The mixture was allowed to stand for 53 h. The yield was 67%, m.p. 283–284 °C (DMF). IR, ν/cm^{-1} : 3400, 3200 (NH); 2200 (CN); 1690 (CO); 1620, 1570 (C=C, C=N). EI MS, m/z (I_{rel} (%)): 357 [M]⁺⁺ (35), 222 [M - HOC₆H₄CHN=NH]⁺⁺ (67), 136 [HOC₆H₄CH=NNH₂]⁺⁺ (100).

5-[(4-Hydroxy-3-methoxybenzylidene)hydrazonomethyl]-2-oxo-4-phenylamino-1,2-dihydropyridine-3-carbonitrile (12c) was prepared analogously to **12a** but with refluxing (5 h). The yield was 53%, m.p. 292–293 °C (DMF). ¹H NMR (DMSO-d₆), δ : 3.80 (s, 3 H, OMe); 6.84 (d, 1 H, H(3'), $J_o = 8.0$ Hz); 7.24 (dd, 1 H, H(2'), $J_o = 8.0$ Hz, $J_m = 2.0$ Hz); 7.28–7.46 (m, 6 H, Ph, H(6')); 8.09 (s, 1 H, H(6)); 8.57 and 8.76 (both s, 1 H each, H(1'), H(4')); 9.80 (br.s, 1 H, OH); 11.45 (br.s, 1 H, C(4)NH); 12.10 (br.s, 1 H, N(1)H). IR, v/cm⁻¹: 3300, 3180 (NH); 2200 (CN); 1660 (CO); 1580 (C=C, C=N). EI MS, m/z (I_{rel} (%)): 387 [M]^{+•} (7), 222 [M – 3-MeO,4-OHC₆H₃CHN=NH]^{+•} (100).

5-[(4-Methoxybenzylidene)hydrazonomethyl]-2-oxo-4-phenylamino-1,2-dihydropyridine-3-carbonitrile (12d) was prepared analogously to **12a**. The mixture was allowed to stand for 4 h. The yield was 69%, m.p. 280–282 °C (DMF). Found (%): C, 66.27; H, 4.57; N, 17.86; H₂O, 3.43. C₂₁H₁₇N₅O₂•0.65H₂O. Calculated (%): C, 65.84; H, 4.81; N, 18.28; H₂O, 3.06. ¹H NMR (DMSO-d₆), δ : 3.80 (s, 3 H, OMe); 7.02, 7.33–7.42, and 7.77 (all m, 2 H each, 5 H, 2 H, Ph, C₆H₄OMe); 8.10 (s, 1 H, H(6)); 8.63 and 8.75 (both s, 1 H each, H(1'), H(4')); 11.41 (br.s, 1 H, C(4)NH); 12.10 (br.s, 1 H, N(1)H). IR, v/cm⁻¹: 3560, 3350 (OH); 3180 (NH); 2216 (CN); 1651 (CO); 1635, 1616 (C=C, C=N). ES MS, *m/z*: 372 [M + H]⁺⁺, 393 [M + Na]⁺⁺, 743 [2 M + H]⁺⁺, 765 [2 M + Na]⁺⁺.

5-Cyclohexylidenehydrazonomethyl-2-oxo-4-phenylamino-1,2-dihydropyridine-3-carbonitrile (12e) was prepared analogously to **12a** but with refluxing (3 h). The yield was 91%, m.p. 290–291 °C (anhydrous EtOH). IR, v/cm⁻¹: 3190 (NH); 2200 (CN); 1690 (CO); 1620, 1580 (C=C, C=N). EI MS, m/z (I_{rel} (%)): 333 [M]⁺⁺ (50), 222 [M - C₆H₁₀N=NH]⁺⁺ (100).

3-Cyano-5-hydroxymethyl-2-oxo-4-phenylamino-1,2-dihydropyridine (13a). Sodium borohydride (0.726 g, 0.019 mol) was added portionwise with stirring to a suspension of formylpyridone **9a** (3 g, 12.6 mmol) in Pr^iOH (60 mL) at 50–55 °C for 3 h. The precipitate that formed was filtered off, washed with Pr^iOH , and dissolved in water (35 mL). The solution was acidified to pH 6 and kept at 20 °C for 2 h. The precipitate that formed was filtered off and washed with water. Compound **13a** was obtained in a yield of 2.15 g (70%), m.p. 222–223 °C (Pr^iOH –DMF, 50 : 1). ¹H NMR ($DMSO-d_6$), δ : 4.33 (s, 2 H, CH_2OH); 5.24 (br.s, 1 H, C(4)NH); 11.29 (br.s, 1 H, N(1)H). IR, v/cm⁻¹: 3341 (OH); 3300 (NH); 2216 (CN); 1644, 1631 (CO). EI MS, m/z (I_{rel} (%)): 240 [M – H]⁺⁺ (14), 221 [M – H₂O – 2 H]⁺⁺ (100), 193 [M – H₂O – 2 H – CO] (17).

4-(4-Chlorophenylamino)-3-cyano-5-hydroxymethyl-2-oxo-1,2-dihydropyridine (13b) was prepared from formylpyridone **9b** (0.5 g, 1.8 mmol) and NaBH₄ (0.069 g, 1.8 mmol) analogously to compound **13a**. The yield was 72%, m.p. 261–264 °C (PrⁱOH–DMF, 3 : 2). ¹H NMR (DMSO-d₆), δ : 4.34 (s, 2 H, CH₂OH); 5.25 (br.s, 1 H, CH₂OH); 7.16–7.46 (m, 6 H, Ph, H(6)); 8.62 (br.s, 1 H, C(4)NH); 11.35 (br.s, 1 H, N(1)H). IR, v/cm⁻¹: 3405 (OH); 3273 (NH); 2208 (CN); 1646 (CO). ES MS, *m/z*: 276 [M + H]⁺⁺, 298 [M + Na]⁺⁺, 551 [2 M + H]⁺⁺, 573 [2 M + Na]⁺.

3-Cyano-2-oxo-4-phenylamino-5-ureidomethyl-1,2-dihydropyridine (14). Urea (2.52 g, 0.042 mol) was added to a suspension of formylpyridone **9a** (0.4 g, 1.7 mmol) in AcOH (6 mL). The reaction mixture was heated with stirring at 90 °C until the precipitate was completely dissolved. The solution was refluxed for 13 h, cooled to 20 °C, and poured into water (20 mL). The precipitate that formed was filtered off and successively washed with water, a NaHCO₃ solution, and water. Compound 14 was obtained in a yield of 3.54 g (89%), m.p. 247-250 °C (MeCN–DMF, 6 : 1). ¹H NMR (DMSO-d₆), δ: 4.03 (d, 2 H, 2 HC(5), $J_0 = 6.4$ Hz); 5.82 (br.s, 2 H, NH₂); 6.55 (t, 1 H, CH_2NH , $J_0 = 6.4 Hz$; 7.13 and 7.33 (both m, 3 H and 2 H each, Ph); 7.33 (s, 1 H, H(6)); 10.26 (br.s, 1 H, C(4)NH); 11.18 (br.s, 1 H, N(1)H). IR, v/cm⁻¹: 3429 (OH); 3348, 3188 (NH); 2199 (CN); 1632 (CO). EI MS, m/z (I_{rel} (%)): 283 [M]^{+•} (1), 266 $[M - NH_3]^+$ (17), 222 $[M - NH_2CONH_2 - H]^+$ (100).

3-Cyano-2-oxo-4-phenylamino-5-piperidinomethyl-1,2-dihydropyridine (16). Thionyl chloride (0.4 mL) was added dropwise to a suspension of pyridone 13a (0.4 g, 1.66 mmol) in CH₂Cl₂ (4 mL). The reaction mixture was refluxed with stirring for 10 h, and then CH₂Cl₂ (5 mL) was added. The precipitate was triturated, filtered off, and washed with CH2Cl2. A chloro derivative was obtained in a yield of 0.353 g. The resulting chloro derivative (0.28 g) was stirred with Py (2 mL) at 90 °C for 6 h. The reaction mixture was cooled to 20 °C, the solution was removed by decantation, the residual oil was triturated with acetonitrile, and the precipitate was filtered off and washed with acetonitrile. Pyridinium salt 15 was obtained in a yield of 0.28 g (63%), m.p. 320–322 °C (decomp.). ¹H NMR (DMSO-d₆), δ: 5.97 (s, 2 H, 2 HC(6)); 7.04, 7.12, and 7.26 (all m, 2 H each, 1 H, 2 H, Ph); 7.96 (s, 1 H, H(6)); 8.08, 8.55, and 9.10 (all m, 2 H each, 1 H, 2 H, Pv): 9.51 (br.s. 1 H, C(4)NH): 11.93 (br.s, 1 H, N(1)H). ES MS, m/z: 224 $[M - C_5H_5N]^+$, 80 $[M - C_5H_5N + H]^+$. A mixture of pyridinium salt 15 (0.28 g, 0.83 mmol) and piperidine (0.19 g, 2.2 mmol) in benzene (4 mL) was refluxed with stirring for 1 h. After cooling of the reaction mixture to 20 °C, the precipitate that formed was filtered off and triturated with petroleum ether. The precipitate was filtered off and washed with water. Compound 16 was obtained in a yield of 0.145 g (57%), m.p. 229-231 °C (MeCN). Found (%): N, 18.38. C₁₈H₂₀N₄O. Calculated (%): N, 18.17. ¹H NMR (DMSO- d_6), δ : 1.41–1.55 (m, 6 H, 2 H(3')–H(5')); 2.39 (br.s, 4 H, 2 H(2'), 2 H(6')); 3.42 (s, 2 H, 2 HC(5)); 7.19-7.39 (m, 6 H, Ph, H(6)); 10.69 (br.s, 1 H, C(4)NH); 11.27 (br.s, 1 H, N(1)H). IR, v/cm⁻¹: 3250, 3150 (NH); 2210 (CN); 1630 (CO). ES MS, *m*/*z*: 309 [M + H]⁺, 331 [M + Na]⁺, $617 [2 M + H]^+, 639 [2 M + Na]^+, 224 [M - C_5H_5N]^{+*}, 86$ $[C_5H_{11}N + H]^+$.

3-Cyano-5-hydroxyiminomethyl-2-oxo-4-phenylamino-1,2dihydropyridine (17a). Formylpyridone **9a** (2 g, 8.4 mmol) was added to a solution of hydroxylamine hydrochloride (1.02 g, 14.7 mmol) in pyridine (8 mL). The reaction mixture was stirred at 80 °C for 2 h, cooled, and poured into water (50 mL). The precipitate that formed was filtered off and washed with water. Compound **17a** was obtained in a yield of 1.87 g (89%), m.p. 261–263 °C (MeOH). ¹H NMR (DMSO-d₆), 8: 7.23–7.50 (m, 5 H, Ph); 7.82 (s, 1 H, H(5')); 8.21 (s, 1 H, H(6)); 10.37 (br.s, 1 H, C(4)NH); 11.29 (br.s, 1 H, OH); 11.86 (br.s, 1 H, N(1)H). IR, v/cm⁻¹: 3660 (OH); 3420 (NH); 2220 (CN); 1680 (CO). ES MS, *m/z*: 237 [M + H – H₂O]⁺⁺, 255 [M + H]⁺⁺, 277 [M + Na]⁺⁺, 508 [2 M + H]⁺⁺, 531 [2 M + Na + H]⁺⁺. EI MS, *m/z* (*I*_{rel} (%)): 254 [M]⁺⁺ (75), 237 [M – OH]⁺⁺ (52), 236 [M – H₂O]⁺⁺ (89), 222 [M – HO – NH]⁺⁺ (100), 77 [Ph]⁺⁺ (60).

4-(4-Chlorophenyl)amino-3-cyano-5-hydroxyiminomethyl-2-oxo-1,2-dihydropyridine (17b). Formylpyridone **9b** (0.5 g, 1.8 mmol) was added to a solution of hydroxylamine hydrochloride (0.21 g, 3 mmol) in pyridine (2 mL). The reaction mixture was stirred at 115 °C for 10 min, cooled, and poured into water (10 mL). The precipitate that formed was filtered off and washed with water. Compound **17b** was obtained in a yield of 0.52 g (98%), m.p. 293–294 °C (PrⁱOH). ¹H NMR (DMSO-d₆), δ: 7.25 and 7.40 (both m, 2 H each, C₆H₄Cl); 7.67 (s, 1 H, H(5')); 8.14 (s, 1 H, H(6)); 10.34 (br.s, 1 H, C(4)NH); 11.09 (br.s, 1 H, OH); 11.80 (br.s, 1 H, N(1)H). IR, v/cm⁻¹: 3317 (OH); 2219 (CN); 1644 (CO). ES MS, *m/z*: 289 [M + H]⁺⁺, 311 [M + Na]⁺⁺, 599 [2 M + Na]⁺⁺. EI MS, *m/z* (I_{rel} (%)): 270 [M - H₂O]⁺⁺ (100), 242 [M - CO]⁺⁺ (27).

3-Cyano-4-dimethylamino-5-hydroxyiminomethyl-2-oxo-1,2-dihydropyridine (17c). Formylpyridone 9c (0.2 g, 1 mmol) was added to a solution of hydroxylamine hydrochloride (0.11 g, 1.7 mmol) in pyridine (0.9 mL). The reaction mixture was stirred at 115 °C for 0.5 h, cooled, and poured into PrⁱOH (7 mL). The inorganic precipitate was filtered off, and the filtrate was concentrated to dryness. The residue was triturated with diethyl ether and heptane and decanted. Then PriOH was added to the semicrystalline precipitate and the mixture was kept at 5 °C for 12 h. The precipitate that crystallized out was filtered off, washed with PrⁱOH, and dried. Compound 17c was obtained in a vield of 0.091 g (42%), m.p. 203–204 °C (PrⁱOH). ¹H NMR (DMSO-d₆), δ: 3.13 (s, 6 H, NMe₂); 7.44 (s, 1 H, H(5')); 7.87 (s, 1 H, H(6)); 10.85 (br.s, 1 H, OH); 11.65 (br.s, 1 H, N(1)H). IR, v/cm⁻¹: 3169 (NH); 2216 (CN); 1635 (CO). ES MS, *m/z*: 207 $[M + H]^+$, 229 $[M + Na]^+$, 413 $[2 M + H]^+$, 435 $[2 \text{ M} + \text{Na}]^+$. EI MS, m/z (I_{rel} (%)): 206 $[\text{M}]^+$ (5), 188 $[M - H_2O]$ (100), 173 $[M - H_2O - Me]$ (51).

5-Acetoxyiminomethyl-3-cyano-2-oxo-4-phenylamino-1,2dihydropyridine (18a). *A*. Formylpyridone 9a (0.5 g, 2 mmol) was added to a solution of hydroxylamine hydrochloride (0.24 g, 4 mmol) in pyridine (3 mL) at 90–95 °C. The reaction mixture was stirred for 0.5 h, and then acetic anhydride (1.3 mL) was slowly added. The reaction mixture was allowed to stand for 8 min, cooled, and poured into water. The precipitate that formed was filtered off, washed with water, and dried. Compound 18a was obtained in a yield of 0.44 g (90%), m.p. 286–288 °C (MeOH). ¹H NMR (DMSO-d₆), δ : 2.14 (s, 3 H, COMe); 7.28 and 7.40 (both m, 3 H and 2 H each, Ph); 8.04 (s, 1 H, H(5')); 8.66 (s, 1 H, H(6)); 10.09 (br.s, 1 H, C(4)NH); 12.20 (br.s, 1 H, N(1)H). IR, v/cm⁻¹: 3200, 3030 (NH); 2200 (CN); 1780 (CO); 1660 (CO, ketone); 1630, 1570 (C=C, C=N). ES MS, m/z: 237 [M - CH₃COOH + H]⁺, 297 [M + H]⁺, 319 [M + Na]⁺, 615 [2 M + Na]⁺. EI MS, m/z (I_{rel} (%)): 236 [M - CH₃COOH]⁺ (100), 209 [M - CH₃COOH - HCN]⁺. (13), 208 [M - CH₃COOH - CO]⁺ (17).

B. Compound **17a** (0.58 g, 0.002 mol) was refluxed in acetic anhydride (4 mL) for 15 min. The precipitate was filtered off and washed with water. Compound **18a** was obtained in a yield of 0.61 g (90%), m.p. $286-288 \degree C$ (DMF).

5-Acetoxyiminomethyl-4-(4-chlorophenyl)amino-3-cyano-2oxo-1,2-dihydropyridine (18b) was prepared analogously to compound 18a by the method *A* or *B* in 90 or 75% yield, respectively, m.p. 288 °C (decomp.). ¹H NMR (DMSO-d₆), δ : 2.15 (s, 3 H, COMe); 7.32 and 7.47 (both m, 2 H each, C₆H₄Cl); 8.05 (s, 1 H, H_{α}); 8.66 (s, 1 H, H(6)); 10.06 (br.s, 1 H, C(4)NH); 12.28 (br.s, 1 H, N(1)H). IR, v/cm⁻¹: 3252, 3067 (NH); 2209 (CN); 1771 (CO); 1670 (CO, ketone); 1636, 1606 (C=C, C=N). ES MS, *m*/*z*: 331 [M + H]⁺⁺, 353 [M + Na]⁺⁺, 683 [2 M + Na]⁺⁺, 271 [M + H – CH₃COOH]⁺⁺.

3,5-Dicyano-4-phenylaminopyridine (19a). Formylpyridone 9a (0.5 g, 2 mmol) was added to a solution of hydroxylamine hydrochloride (0.24 g, 4 mmol) in pyridine (3 mL) at 90–95 °C. The reaction mixture was stirred for 0.5 h, and then acetic anhydride (1.3 mL) was slowly added. After 8 min, pyridine (2 mL) was added. The reaction mixture was stirred at 115 °C for 2 h, cooled, and poured into water. The precipitate that formed was filtered off, washed with water, and dried. Compound 19a was obtained in a yield of 0.44 g (90%), m.p. 323–324 °C (MeOH). ¹H NMR (DMSO-d₆), δ: 7.25 and 7.37 (both m, 3 H and 2 H each, Ph); 8.34 (s, 1 H, H(6)); 9.62 (br.s, 1 H, C(4)NH); 12.41 (br.s, 1 H, N(1)H). IR, v/cm^{-1} : 3220 (NH); 2220 (CN); 1670 (CO); 1620, 1570 (C=C, C=N). ES MS, m/z: 237 [M + H]⁺, 259 [M + Na]⁺, 495 [2 M + Na]⁺. EI MS, m/z (I_{rel} (%)): 236 [M]⁺ (100), 209 [M - HCN]⁺ $(13), 208 [M - CO]^{+}$ (17).

4-(4-Chlorophenyl)amino-3,5-dicyanopyridine (19b) was prepared analogously to **19a** from compound **9b**. The yield was 68%, m.p. 335 °C (decomp., MeOH). ¹H NMR (DMSO-d₆), δ : 7.27 and 7.42 (both m, 2 H each, C₆H₄Cl); 8.31 (s, 1 H, H(6)); 9.58 (br.s, 1 H, C(4)NH); 12.40 (br.s, 1 H, N(1)H). IR, v/cm⁻¹: 3300 (NH); 3084, 2243, 2212 (CN); 1651 (CO). ES MS, *m/z*: 271 [M + H]⁺⁺, 293 [M + Na]⁺⁺.

2-Chloro-3,5-dicyano-4-phenylaminopyridine (20a). A mixture of compound **19a** (0.66 g, 2.8 mmol) and triethylamine hydrochloride (0.35 g, 2.5 mmol) in phosphoryl chloride (5 mL) was refluxed for 2 h, cooled, and poured into water. The precipitate that formed was filtered off and washed with water, ethanol, and diethyl ether. Compound **20a** was obtained in a yield of 0.62 g (87%), m.p. 192–194 °C (EtOH). ¹H NMR (DMSO-d₆), δ : 7.34 (m, 5 H, Ph); 8.61 (s, 1 H, H(6)); 10.20 (br.s, 1 H, C(4)NH). IR, v/cm⁻¹: 3270 (NH); 2200 (CN); 1590, 1570 (C=C, C=N).

2-Chloro-4-(4-chlorophenyl)amino-3,5-dicyanopyridine (20b) was prepared analogously to **20a** from compound **19b**. The yield was 65%, m.p. 240–241 °C (MeOH). ¹H NMR (DMSO-d₆), δ : 7.36 and 7.47 (both m, 2 H each, C₆H₄Cl); 8.64 (s, 1 H, H(6)); 10.16 (br.s, 1 H, C(4)NH). IR, v/cm⁻¹: 3280 (NH); 2226 (CN); 1568 (C=C, C=N).

2-Amino-3,5-dicyano-4-phenylaminopyridine (21a). A mixture of pyridine 20a (3.35 g, 13 mmol) and a 14% ethanolic solution of ammonia (35 mL) was heated in an autoclave at 80 °C for 6 h. The precipitate was filtered off. Pyridine **21a** was obtained in a yield of 2.7 g (87%), m.p. 225–226 °C (EtOH). ¹H NMR (DMSO-d₆), δ : 7.20 (d, 2 H, H (2'), H(6'), J = 8.8 Hz); 7.35 (t, 3 H, H(3')–H(5'), J = 8.8 Hz); 7.45 (br.s, 2 H, NH₂); 8.22 (s, 1 H, H(6)); 9.33 (br.s, 1 H, C(4)NH). IR, v/cm⁻¹: 3360, 3300, 3280, 3125 (NH, NH₂); 2220 (CN); 1660 (NH, NH₂); 1590, 1550 (C=C, C=N). EI MS, m/z (I_{rel} (%)): 235 [M]⁺⁺ (100), 208 [M – HCN]⁺⁺ (27), 208 [M – CO]⁺⁺ (17).

2-Benzylamino-3,5-dicyano-4-phenylaminopyridine (21b). A mixture of compound **20a** (1.8 g, 7.1 mmol) and benzylamine (1.5 g, 0.014 mol) in anhydrous ethanol (60 mL) was stirred at 40–45 °C for 3 h. The precipitate was filtered off, washed with ethanol and diethyl ether, and dried. Pyridine **21b** was obtained in a yield of 2.1 g (91%), m.p. 219–220 °C (EtOH). IR, v/cm⁻¹: 3360, 3280 (NH); 2200 (CN); 1590 (C=C, C=N, NH). ES MS, m/z: 326 [M + H]⁺⁺, 348 [M + Na]⁺⁺.

3,5-Dicyano-2,4-di(phenylamino)pyridine (21c) was prepared analogously to **21b** from compound **20a** (0.3 g, 1.2 mmol) and aniline (0.33 g, 3.6 mmol) in ethylene glycol (5 mL) at 100 °C. The reaction mixture was allowed to stand for 1.5 h. The yield was 91%, m.p. 234–236 °C (MeOH). IR, v/cm⁻¹: 3360, 3300 (NH); 2200 (CN); 1590 (C=C, C=N, NH). ES MS, *m/z*: 312 $[M + H]^+$, 645 $[2 M + Na]^+$.

3,5-Dicyano-2-(4-methoxyphenyl)amino-4-phenylaminopyridine (21d) was prepared analogously to **21b** from compound **20a**. The reaction mixture was allowed to stand for 2 h. The yield was 98%, m.p. 203–204 °C (MeOH). IR, v/cm⁻¹: 3360, 3260 (NH); 2200 (CN); 1600, 1560 (C=C, C=N, NH). ES MS, m/z: 342 [M + H]⁺⁺, 364 [M + Na]⁺⁺, 705 [2 M + Na]⁺⁺.

2-(4-Chlorophenyl)amino-3,5-dicyano-4-phenylaminopyridine (21e) was prepared analogously to **21b** from compound **20a** (0.35 g, 1.4 mmol) and 4-chloroaniline (0.52 g, 4.1 mmol) in ethylene glycol (5 mL) at 80 °C. The reaction mixture was allowed to stand for 2.5 h. The yield was 91%, m.p. 218–219 °C (EtOH). IR, v/cm⁻¹: 3310, 3280 (NH); 2200 (CN); 1600, 1580, 1560 (C=C, C=N, NH). ES MS, m/z: 346 [M + H]⁺⁺, 713 [2 M + Na]⁺⁺.

3,5-Dicyano-2-(4-methylphenyl)amino-4-phenylaminopyridine (21f) was prepared analogously to **21b** from compound **20a**. The reaction mixture was allowed to stand for 3 h. The yield was 42%, m.p. 224–226 °C (EtOH). IR, v/cm⁻¹: 3320, 3260 (NH); 2200 (CN); 1580, 1560 (C=C, C=N, NH). EI MS, m/z: 326 [M + H]⁺⁺, 348 [M + Na]⁺⁺, 650 [2 M + H]⁺⁺.

2,4-Di[(4-chlorophenyl)amino]-3,5-dicyanopyridine (21g) was prepared analogously to **21e** from compound **20b**. The yield was 70%, m.p. 222–226 °C (EtOH). ¹H NMR (DMSO-d₆), δ : 7.24–7.55 (m, 8 H, 2 C₆H₄Cl); 8.37 (s, 1 H, H(6)); 9.49 and 9.58 (both br.s, 1 H each, 2 NH). IR, v/cm⁻¹: 3276 (NH); 2215 (CN); 1611, 1600 (C=C, C=N, NH). ES MS, *m/z*: 380 [M + H]⁺.

7-Cyano-6-oxo-1-phenyl-5,6-dihydro-1*H***-pyrazo-lo[4,3-c]pyridine (22a).** A solution of compound **18a** (0.2 g, 0.68 mmol) in DMF (5 mL) was refluxed for 30 min, DMF was evaporated to dryness, the residue (dark-cream-colored crystals) was triturated with ethyl acetate, and the precipitate was filtered off and washed with ethyl acetate. Compound **22a** was obtained in a yield of 0.15 g (94%), m.p. 338–340 °C (decomp., $Pr^{i}OH-DMF, 1:1$). Found (%): N, 23.49. $C_{13}H_8N_4O$. Calculated (%): N, 23.72. ¹H NMR (DMSO-d₆), & 7.55 (m, 5 H, Ph); 8.40 (s, 1 H, H(4)); 8.74 (s, 1 H, H(3)); 13.00 (br.s, 1 H,

N(5)H). IR, v/cm⁻¹: 3084 (NH); 2218 (CN); 1654 (CO); 1631 (C=C, C=N). ES MS, *m/z*: 237 [M + H]⁺⁺, 259 [M + Na]⁺⁺, 473 [2 M + H]⁺⁺, 495 [2 M + Na]⁺⁺.

1-(4-Chlorophenyl)-7-cyano-6-oxo-5,6-dihydro-1H-pyrazolo[4,3-c]pyridine (22b) was prepared analogously to 22a from compound 18b. The yield was 79%, m.p. 332-335 °C (decomp., PrⁱOH–DMF, 1 : 1). ¹H NMR (DMSO-d₆), δ: 7.62 (m, 4 H, C₆H₄Cl); 8.41 (s, 1 H, H(3)); 8.73 (s, 1 H, H(4)); 13.00 (br.s, 1 H, N(5)H). ¹³C NMR (DMSO-d₆), δ: 73.7 (C(7)); 112.1 (C(3a)); 114.5 (CN); 128.4, 128.8, 133.6, 135.8 (C(1')-C(6')); 139.5 (C(3)); 140.8 (C(4)); 146.8 (C(7a)); 161.4 (C(6)). IR, v/cm⁻¹: 3230 (NH); 2212 (CN); 1682 (CO); 1604 (C=C, C=N). ES MS, m/z: 271 [M + H]⁺ · 293 [M + Na]⁺ · , 563 [2 M + Na]⁺ · The three-dimensional structure of molecule 22b was established by powder X-ray diffraction. Compound 22b crystallizes in the orthorhombic space group $P2_12_12_1$ with the unit cell parameters a = 3.879(1) Å, b = 7.633(2) Å, c = 38.83(1) Å and the unit cell volume of 1168(1) $Å^3$. The intensities of reflections were measured on an XPert PRO powder diffractometer using Cu-K α_1 radiation. The crystal structure of compound **22b** was solved by the systematic search method¹⁵ and refined by the Rietveld method using the MRIA program.¹⁶ All structural data, including the unit cell parameters, atomic coordinates, bond angles, and bond lengths, were deposited with the Cambridge Structural Database (CCDC604040).*

3-Formyl-2-oxo-4-phenylamino-1,2-dihydropyridine (23). A. A Ni-Al₂ alloy (3 g) was added to a suspension of pyridone 5c (3 g, 0.014 mol) in 50% HCOOH (40 mL). The reaction mixture was refluxed with stirring for 6 h and cooled to 20 °C. The precipitate that formed was filtered off and washed with a small amount of 50% HCOOH and water. The crude precipitate, which was a mixture of inorganic salts and the reaction product, was extracted with DMF (4×20 mL), and the precipitate was filtered off. The filtrate was concentrated to dryness, and the residue (the crystalline precipitate) was triturated with petroleum ether, filtered off, and washed with petroleum ether. Compound 23 was obtained in a yield of 1.62 g (53%), m.p. 251–254 °C (toluene). ¹H NMR (DMSO-d₆), δ: 5.88 (d, 1 H, $H(5), J_{0} = 7.4 \text{ Hz}$; 7.25–7.49 (m, 6 H, Ph, H(6)); 10.03 (s, 1 H, CHO); 11.07 (br.s, 1 H, N(1)H); 11.71 (br.s, 1 H, C(4)NH). IR, v/cm⁻¹: 3142 (OH); 1630 (CO). ES MS, m/z: 237 $[M + Na]^{+}$, 451 $[2 M + Na]^{+}$, 665 $[3 M + Na]^{+}$. EI MS, $m/z (I_{rel} (\%)): 213 [M - H]^{+} (100), 185 [M - CHO]^{+} (100).$ (Only a small amount of the starting compound was isolated from the mother filtrate of the reaction mixture.)

B. A Ni—Al₂ alloy (2 g) was added to a suspension of pyridone **5c** (2 g, 9.5 mmol) in 85% HCOOH (30 mL). The reaction mixture was refluxed with stirring for 20 h and cooled to 20 °C. The precipitate that formed (inorganic salts) was filtered off and washed with a small amount of 85% HCOOH. The filtrate was diluted with water (150 mL) and kept at 20 °C for 1 h. The precipitate that formed was filtered off and washed with water and petroleum ether. Compound **23** was obtained in a yield of 0.538 g (26%) (a mixture of compound **23** and the sample prepared by the method *A* showed no melting point depression).

^{*} These data can be obtained, free of charge, on application to the Cambridge Crystallographic Data Centre (CCDC), 12 Union Road, Cambridge CB2 1EZ, UK. Fax: +44 (0) 1223 336033. E-mail: deposit@ccdc.cam.ac.uk.

1-Oxo-1,2-dihydrobenzo[*b***]-1,6-naphthyridine (24).** *A***. A filtrate (HCOOH and washings) from the synthesis of compound 23** by the method *B* was concentrated to dryness, and the crystalline residue was triturated with PrⁱOH. Compound **24** was obtained in a yield of 0.768 g (53%), m.p. 330–331 °C (MeOH). ¹H NMR (DMSO-d₆), δ : 6.63 (d, 1 H, H(3), $J_o = 7.5$ Hz); 7.38 (d, 1 H, H(2), $J_o = 7.5$ Hz); 7.57, 7.85, 8.02, and 8.14 (all m, 1 H each, H(5)–H(8)); 9.18 (s, 1 H, H(9)); 11.07 (br.s, 1 H, N(1)H). IR, v/cm⁻¹: 3036 (NH); 1685, 1657, 1633 (CO). ES MS, *m/z*: 197 [M + H]⁺⁺. EI MS, *m/z* (I_{rel} (%)): 196 [M]⁺⁺ (100), 168 [M – CO]⁺⁺ (52), 140 [M – CO – HCN – H]⁺⁺ (27).

B. Piperidine (0.05 mL) was added to a solution of 3-formylpyridone **23** (0.05 g, 0.23 mmol) in Pr^iOH (2.5 mL). The reaction mixture was refluxed for 7 h and then cooled to 20 °C. The precipitate that formed was filtered off and washed with Pr^iOH . Compound **24** was obtained in a yield of 0.025 g (56%). A mixture of compound **24** and the sample prepared by the method *A* showed no melting point depression.

4-(2-Oxo-1,2-dihydropyridin-4-yl)aminobenzenesulfonic acid (25). A 63% H_2SO_4 solution (2 mL) was added to pyridone 5c (0.15 g, 0.7 mmol). The reaction mixture was refluxed for 1 h and then cooled to 20 °C. The precipitate that formed was filtered off, washed with water and PriOH, and triturated with a NaHCO₃ solution. The insoluble precipitate was filtered off. The filtrate was acidified with 10% HCl to pH 4. The precipitate that formed was filtered off and washed with water. Compound 25 was obtained in a yield of 0.084 g (52%), m.p. >350 °C (MeOH). Found (%): C, 49.90; H, 3.73; N, 11.04; S, 11.63. C₁₁H₁₀N₂O₄S. Calculated (%): C, 49.62; H, 3.79; N, 10.52; S, 12.04. ¹H NMR (DMSO-d₆), δ : 6.22 (d, 1 H, H(3), $J_m =$ 2.0 Hz); 6.56 (dd, 1 H, H(5), $J_o = 7.2$ Hz, $J_m = 2.0$ Hz); 7.21 and 7.68 (both m, 2 H each, $C_6H_5SO_3H$); 7.73 (d, 1 H, H(6), $J_0 =$ 7.2 Hz); 9.93 (br.s, 1 H, C(4)NH); 12.90 (br.s, 1 H, N(1)H). IR, v/cm⁻¹: 3257 (OH); 3117 (NH); 1655, 1624 (CO). EI MS, m/z (I_{rel} (%)): 185 [M - HSO₃]⁺ (100), 167 [M - HSO₃ - H_2O^{+} (13), 157 $[M - HSO_3 - CO]^{+}$ (20), 130 $[M - HSO_3 - CO]^{+}$ CO - HCN]^{+•} (43), 64 [SO₂] (48).

3-Hydroxyiminomethyl-2-oxo-4-phenylamino-1,2-dihydropyridine (26a). Formylpyridone **23** (0.185 g, 0.86 mmol) was added to a solution of hydroxylamine hydrochloride (0.06 g, 0.87 mmol) in pyridine (0.7 mL). The reaction mixture was stirred at 60 °C for 1 h and concentrated to dryness. The residue (oil) was triturated in water (7 mL) and kept at 20 °C for 40 min. The precipitate that formed was filtered off and washed with water. Compound **26a** was obtained in a yield of 0.125 g (63%), m.p. 261–262 °C (acetonitrile). ¹H NMR (DMSO-d₆), &: 5.97 (d, 1 H, H(5), $J_o = 7.5$ Hz); 7.15–7.44 (m, 6 H, Ph, H(6)); 8.50 (s, 1 H, H_a); 10.17 (br.s, 1 H, C(4)NH); 10.80 (br.s, 1 H, OH); 10.90 (br.s, 1 H, N(1)H). IR, v/cm⁻¹: 3254 (OH); 3246 (NH); 1624 (CO). ES MS, m/z: 230 [M + H]⁺⁺, 252 [M + Na]⁺⁺, 212 [M – OH] ⁺⁺. EI MS, m/z (I_{rel} (%)): 229 [M]⁺⁺ (53), 212 [M – OH]⁺⁺ (100), 197 [M – NH – OH]⁺⁺ (36).

5-(4-Nitrophenyl)hydrazonomethyl-2-oxo-4-phenylamino-1,2-dihydropyridine (26b). A mixture of 3-formylpyridone 23 (0.16 g, 0.74 mmol) and *p*-nitrophenylhydrazone (0.11 g, 0.74 mmol) in EtOH (10 mL) was refluxed in the presence of a catalytic amount of *p*-TsOH for 1.5 h. The precipitate that formed was filtered off and washed with EtOH. Compound 26b was obtained in a yield of 0.176 g (61%), m.p. 361–362 °C (DMF). ¹H NMR (DMSO-d₆), δ : 6.10 (d, 1 H, H(5), $J_o =$ 7.5 Hz); 6.91 and 8.09 (both m, 2 H each, $C_6H_4NO_2$); 7.20–7.53 (m, 6 H, Ph, H(6)); 8.69 (s, 1 H, H_{α}); 10.68 (br.s, 1 H, C(4)NH); 10.94 (br.s, 1 H, N<u>H</u> $C_6H_4NO_2$); 11.12 (br.s, 1 H, N(1)H). IR, v/cm⁻¹: 3286, 3217, 3184 (NH); 1641 (CO). EI MS, *m/z* (I_{rel} (%)): 349 [M]⁺⁺ (33), 212 [M – 4-NHC₆H₄NO₂]⁺⁺ (100), 197 [M – 4-NH–NHC₆H₄NO₂]⁺⁺ (64).

This study was financially supported by the Federal Agency for Science and Innovations of the Russian Federation (Contract No. 1/05).

References

- 1. L. V. Ershov and V. G. Granik, *Khim. Geterotsikl. Soedin.*, 1985, 646 [*Chem. Heterocycl. Compd.*, 1985, 544 (Engl. Transl.)].
- V. A. Azimov, V. G. Granik, S. I. Grizik, L. V. Ershov, N. I. Smetskaya, S. D. Yuzhakov, M. D. Mashkovskii, and L. N. Yakhontov, *Khim.-farm. Zh.*, 1985, **19**, No. 8, 947 [*Pharm. Chem. J.*, 1985, **19**, No. 8, 548 (Engl. Transl.)].
- V. G. Granik and S. I. Kaimanakova, *Khim. Geterotsikl.* Soedin., 1983, 816 [Chem. Heterocycl. Compd., 1983, 714 (Engl. Transl.)].
- 4. A. S. Ivanov, N. Z. Tugusheva, L. M. Alekseeva, and V. G. Granik, *Izv. Akad. Nauk, Ser. Khim.*, 2004, 837 [*Russ. Chem. Bull., Int. Ed.*, 2004, **53**, 873].
- M. D. Mashkovskii, *Lekarstvennye sredstva* [*Drugs*], Torgsin, Kharkov, 1997, 1235 pp. (in Russian).
- M. D. Mashkovskii, Lekarstva XX Veka [Drugs of the XX Century], Novaya Volna, Moscow, 1998, 319 pp. (in Russian).
- V. G. Granik, Lekarstva. Farmakologicheskii, biokhimicheskii i khimicheskii aspekty [Drugs. Pharmacological, Biochemical, and Chemical Aspects], Vuzovskaya Kniga, Moscow, 2001, 407 pp. (in Russian).
- V. G. Granik, S. I. Grizik, S. S. Kiselev, V. V. Chistyakov, O. S. Anisimova, and N. P. Solov´eva, *Khim. Geterotsikl. Soedin.*, 1984, 532 [*Chem. Heterocycl. Compd.*, 1984, 434 (Engl. Transl.)].
- N. Z. Tugusheva, L. V. Ershov, V. G. Granik, G. Ya. Shvarts, R. D. Syubaev, and M. D. Mashkovskii, *Khim.-farm. Zh.*, 1986, No. 7, 830 [*Pharm. Chem. J.*, 1986, No. 7, 488 (Engl. Transl.)].
- A. A. Bakibaev, V. D. Filimonov, and L. G. Tignibidina, *Khim.-farm. Zh.*, 1993, No. 3, 28 [*Pharm. Chem. J.*, 1993, No. 3, 198 (Engl. Transl.)].
- 11. A. I. Bokanov, S. Yu. Kukushkin, V. A. Parshin, L. M. Alekseeva, K. I. Kobrakov, and V. G. Granik, *Khim.-farm. Zh.*, 2004, No. 1, 11 [*Pharm. Chem. J.*, 2004, No. 1, 10 (Engl. Transl.)].
- M. Yu. Yakovlev, A. V. Kadushkin, N. P. Solov´eva, O. S. Anisimova, and V. G. Granik, *Tetrahedron*, 1998, 54, 5775.
- 13. C. H. Trabert, Arch. Pharm., 1961, 294, No. 4, 246.
- 14. T. van Es and B. Staskun, J. Chem. Soc., 1965, 5775.
- 15. V. V. Chernyshev and H. Schenk, Z. Krystallogr., 1998, 213, 1.
- V. B. Zlokazov and V. V. Chernyshev, J. Appl. Crystallogr., 1992, 25, 447.

Received May 26, 2006; in revised form July 12, 2006